Capturing Coupled Structural and Electronic Motions During Excited-State Intramolecular Proton Transfer via Computational Multi-edge Resonant Inelastic X-ray Scattering

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Abstract

Proton transfer processes build the foundation of many chemical processes. In 2 excited state intramolecular proton transfer (ESIPT) processes, the ultrafast proton 3 transfer is impulsively initiated through light. Here, we explore the time-dependent Δ coupled atomic and electronic motions during and following ESIPT through computa-5 tional time-resolved resonant inelastic X-ray scattering (RIXS). Excited-state ab initio 6 molecular dynamics simulations combined with time-dependent density functional the-7 ory calculations were performed for a model ESIPT complex, 10-hydroxybenzo[h]quinoline, 8 to obtain transient RIXS signatures. The RIXS spectra at both the nitrogen and oxy-9 gen K-edges were computed to resolve the electronic and atomic structural dynamics 10 from both the proton donor and acceptor perspective. The results demonstrate that 11

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RIXS provides unprecedented details of the local electronic structure, the coupling between different core and valence excited electronic states, and the reorganization of the electronic structure coupled to the proton transfer process. We also develop a spectroscopic ruler correlating spectral shifts of a RIXS peak to the proton transfer distance during ESIPT. This work highlights the exciting potential of time resolved RIXS experiments at newly commissioned soft X-ray free electron laser facilities in measuring coupled electronic and structural changes during ultrafast chemical processes.

Proton transfer reactions are at the heart of many chemical and biological processes. Ex-19 cited state intramolecular proton transfer (ESIPT) is a subclass of proton transfer processes 20 where the proton transfer takes place on an electronically excited state after photoexcita-21 tion. The earliest reports of ESIPT date back to the 1950s when Weller observed a Stokes 22 shift in the emission spectrum of salicylic acid, which he assigned to proton transfer from 23 the hydroxy to the carboxy group following photo-excitation.¹ The fluorescence properties 24 of ESIPT systems have led to their application in bioimaging (fluorescence markers) and 25 optoelectronics, including preparation of white-light emitting dyads or production of organic 26 light-emitting diodes.^{2–4} The variety of applications for molecules undergoing ESIPT is de-27 pendent on the wide tunability of their optoelectronic properties requiring a molecular level 28 understanding of the underlying ESIPT photophysics. 29

The intramolecular H-bonded complex, 10-hydroxybenzo[h]quinoline (HBQ), has served 30 as an excellent model system to study the photophysics of ultrafast ESIPT processes and 31 it has been the subject of intense experimental and theoretical efforts over the last two 32 decades.⁵⁻¹⁷ Fig. 1 shows an overview of the ESIPT process for the enol-keto transition in 33 HBQ which is found preferentially in its enol form on the ground electronic state when in 34 solution.¹⁸ Previous spectroscopic studies have shown that upon excitation with 400 nm, ES-35 IPT from E* to K* takes place within 13 fs.^{19,20} The fluorescence from K* is red-shifted from 36 the absorption by about 250 nm.¹⁹ Interestingly, the absorption and emission wavelengths, 37 efficiency and timescales of proton transfer and the ground state enol-keto equilibrium of 38

HBQ can be significantly altered by changing the substituents on the molecular backbone and varying the solvent environment.^{18,21} A detailed understanding of the local electronic structure on the ground and excited states of HBQ is required as a function of its molecular structure to assess the impact of synthetic tuning on the photophysics of HBQ.



Figure 1. Schematic representation of the electronic states involved in the ESIPT process of HBQ during the enol (E) to keto (K) transformation. Excitation with UV light (UV_{in}) leads to population of the E* state from where ultrafast ESIPT to the K* state takes place within 13 fs. The energy difference between E and K and E* and K* were obtained from ground-state DFT and excited-state TDDFT calculations, respectively.

X-ray spectroscopies are element-specific, giving direct information on the local electronic 43 structure near an atom of interest by tuning to its respective absorption edge. In the soft 44 X-ray regime, this allows researchers to directly probe the local electronic and structural 45 environment around carbon, oxygen or nitrogen atoms in organic molecules. With the avail-46 ability of third-generation synchrotrons and advanced sample delivery systems, soft X-ray 47 absorption near edge spectroscopy (XANES) has been used to investigate liquid and solution 48 phase samples.^{22–25} Both experimental and theoretical studies highlight the potential of soft 49 X-ray XANES to study H-bonding in a variety of systems, including water, alcohols and 50

⁵¹ aqueous solutions of organic molecules.²⁶⁻²⁹ Additionally, the intricate insights into the elec⁵² tronic structure from XANES enable a deeper understanding of solvent effects at a molecular
⁵³ level^{30,31} and of the molecular orbitals and molecular stability of organic molecules.³²

While XANES provides important insights on unoccupied electronic states, it does not 54 contain information on occupied electronic states and the coupling between core and va-55 lence excited states. To obtain additional information on these properties, resonant inelastic 56 X-ray scattering (RIXS) has received increasing attention for liquid and solution phase sam-57 ples.^{22,33,34} RIXS is a 2D X-ray spectroscopic technique that has the potential to reveal rich 58 details of the electronic structure of molecules and materials by measuring the coupling be-59 tween different electronic states.³⁵ The RIXS process is schematically explained in Fig. 2a. 60 The incoming X-ray photon $(X-ray_{in})$ leads to excitation into a core excited state (CES, 61 grey arrow) from where relaxation into a valence excited state (VES) via inelastic scatter-62 ing, or back into the ground state (GS) via elastic scattering leads to the emission of an 63 outgoing photon, X-ray_{out} (dark red arrows). The CESs are defined as states in which an 64 electron has been excited from either the nitrogen or oxygen 1s core orbital into the lowest 65 unoccupied molecular orbitals, depending on the energy of X-ray_{in}. The VESs are states 66 in which an electron has been excited from one of the highest occupied molecular orbitals 67 into one of the lowest unoccupied molecular orbitals. For inelastic scattering, the differ-68 ence between incoming and outgoing photon energy $(E_{trans} \neq 0)$ contains information on 69 the relative energies of the VESs involved in the transitions and their coupling to different 70 CESs. Theoretically, RIXS can be described using the Kramers-Heisenberg equation³⁶ as de-71 scribed elsewhere.^{37–40} The coupled electronic and structural insights from soft X-ray RIXS 72 have previously been used to experimentally and computationally investigate the H-bonding 73 structure of water in different phases,^{41–46} distinguish different bonding characters,^{32,47–49} as 74 well as to observe solvent effects on the electronic structure of solutes³¹ and more recently. 75 to gain direct structural insight into structural dynamics on a molecular level in H-bonded 76 systems⁵⁰ and single-molecule magnets.⁵¹ 77

With the advent of X-ray free electron lasers (XFELs) operating in the soft x-ray regime, 78 time-resolved RIXS and XANES experiments have become experimentally feasible and will 79 continue to expand to a wider range of systems with further improvements at XFEL facilities, 80 including the shift to higher repetition rates. Previous femtosecond RIXS experiments mon-81 itored the time-dependent changes in electronic structure in real time at the Fe L-edge^{33,52,53} 82 and the N K-edge⁵⁴ in transition metal complexes. Transient RIXS has also been used to 83 probe excited state time-dependent hydrogen bonding interactions during a photoinduced 84 thione-thiol tautomerization reaction.⁴⁹ By directly probing proton donor or acceptor sites, 85 transient soft X-ray spectroscopies can be used to investigate the hydrogen bonding dur-86 ing proton transfer reactions. The continued development of computational methods to 87 accurately model soft X-ray spectroscopy is crucial to the success of time-resolved X-ray 88 spectroscopy.^{55–58} 89

In a previous computational study from our group by Loe and co-workers, we investigated 90 the time-resolved XANES signatures of HBQ during the ESIPT process.⁸ The results showed 91 transient features of HBQ at the O and N K-edges including the appearance of an additional 92 peak and observation of time-dependent oscillations in the spectral features. The study 93 provided valuable insights into the coupled electronic and atomic motions during ultrafast 94 ESIPT in HBQ from the perspective of the H-bond donor and acceptor atoms. Here, we 95 compute transient RIXS spectra of HBQ, presenting the first RIXS study simulating the 96 time-resolved spectral signatures at the N and O K-edges during and following the ultrafast 97 ESIPT process. Our work reveals how the transient RIXS features encode time-evolving 98 electronic state couplings as a function of the proton transfer and we develop a spectroscopic 99 ruler to connect spectral energy shifts to proton transfer distance. 100

Figure 2b shows XANES spectra of HBQ in its enol form at the N (top panel) and O (bottom panel) K-edges. We note that the spectra presented here, are shifted by 11 eV to match theory and experimental data for similar compounds (see Fig. S1 and S2 in the SI). The XANES spectra at the N and O K-edges of organic complexes, like HBQ, are



Figure 2. a) Schematic showing the transitions in transient RIXS. The GS describes the system in an unperturbed state and is used as common reference for all other transitions. X-ray excitation (X-ray_{in}) at the N or O K-edge leads to population of CES_N or CES_O , respectively, (grey arrows). Relaxation from the CES to the VES causes emission of X-ray_{out} (dark red). The energy transfer (energy difference between X-ray_{in} and X-ray_{out}) describes the energy of the VES in comparison to the GS. b) Simulated XANES spectra of HBQ at the N and O K-edges. Spectra are normalized to the N K-edge spectrum.

often divided into two regions depending on the character of the CESs accessed in the 105 experiment.^{32,47,59} The contributions of the four highest occupied, and lowest unoccupied 106 molecular orbitals to the different CESs and VESs are tabulated in Table S1 and Table S2 107 and the molecular orbitals are pictured in Fig. S4 of the SI. Based on the contributions from 108 the different molecular orbitals, the lowest CESs are of π character while the CESs involved 109 in transitions above 401 eV at the N K-edge and above 534 eV at the O K-edge are of σ 110 character. In the following discussion, we will focus on the π transitions which have been 111 resolved experimentally for organic molecules.⁵⁹ 112

The 2D GS RIXS maps of the enol form of HBQ at the N and O K-edge are presented in Fig. 3c and d. The RIXS maps are plotted as energy transfer (E_{trans}) vs. incident energy (E_{inc}) where the energy transfer is the difference between incoming and emitted X-ray photon energy $E_{trans} = E_{inc} - E_{emi}$. Features at $E_{trans} = 0$ correspond to elastic scattering in the system of interest. Additional features in the RIXS spectra arise from inelastic scattering and can give insights into the coupling between the ith intermediate state or CES_{N,i} and the ¹¹⁹ f^{th} final states or VES_f involved in the RIXS process. The major peaks along the incident ¹²⁰ energy axis of the RIXS map result from excitation into the CESs and correspond to X-ray ¹²¹ absorption features. These peaks are labelled B and C following our previous study.⁸

At the N K-edge, the B (399 e V) and C (401.3 e V) peaks correspond to excitations from 122 the GS into $CES_{N,1}$ and $CES_{N,2}$ states, respectively, and are represented by the grey arrows 123 in Fig. 2a. The B peak is the strongest transition visible in the spectrum. The high oscillator 124 strength of excitation from the GS into $CES_{N,1}$ (B peak) can be explained by the strong 125 overlap between the N1s and $1\pi^*$ orbitals, which is greater than the overlap between the 126 N1s and the $2\pi^*$ orbital, which is the main contributor to $CES_{N,2}$ (see Table S1 in the SI), 127 resulting in the weaker oscillator strength of the C peak. The discussion of the cross peaks 128 in the GS RIXS map will focus on a few strong inelastic features labelled in Fig. 3c. The 129 different features will be labelled according to excitation into a CES and relaxation into a 130 VES f. For example, B_{I}^{f} corresponds to excitation into CES_{1} (B-peak) and relaxation into 131 VES f at the K-edge of atom I. For B_N^1 and for B_N^3 core excitation takes place into $CES_{N,1}$ 132 and followed by relaxation into the VES₁ or VES₃ leading to emission. B_N^3 at $E_{inc} = 399 \,\mathrm{eV}$ 133 and $E_{trans} = 4.7 \,\mathrm{eV}$ is the strongest feature in the N K-edge RIXS map. Other contributions 134 to the RIXS spectrum at the N K-edge have significantly lower oscillator strengths and will 135 not be discussed further here. Oscillator strengths for all RIXS features are listed in Table S3 136 of the SI. 137

For a more quantitative analysis of the RIXS spectrum, we will next consider different 138 spectral cuts through the RIXS maps. These spectral cuts are indicated in Fig. 3c as lines 139 in the RIXS map and plotted in Fig. 3a and Fig. 3b for the N K-edge. In Fig. 3a, partial 140 fluorescence yield (PFY) XANES spectra, obtained by integrating along the emission axis, 141 are compared to high energy resolution fluorescence detected (HERFD) XANES spectra. 142 corresponding to cuts at constant emission energies (see diagonal dashed line in Fig. 3c). In 143 transition metal systems, HERFD-XANES cuts are often used to increase spectral resolution 144 which unlike the XANES spectral resolution is not limited by the core-hole lifetime but 145

rather by the final-state lifetime.^{60,61} Due to the already small lifetime broadening in lighter atoms like N and O, HERFD-XANES does not lead to an increased resolution in our study. However, there are distinct differences between the HERFD- and PFY-XANES spectra due to their different projections on the incident energy axis. At the N K-edge, the spectral positions of the features along the E_{inc} axis do not change, however the intensity of the C peak, decreases in the HERFD-XANES spectrum relative to the PFY-XANES spectrum.

Insights into the different VESs that are coupled to excitation to a specific CES, are 152 encoded in the energy transfer transfer axis and are analyzed using constant incident energy 153 (CIE) spectral cuts. Fig. 3b shows the CIE spectra at the N K-edge along the incident 154 energies for both the B (black, $E_{inc} = 399 \text{ eV}$) and C (grey, $E_{inc} = 401.3 \text{ eV}$) peaks. The CIE 155 spectra directly report on the transition energies and coupling amplitudes between occupied 156 and unoccupied VESs, similar to UV/vis spectra, but with different selection rules.⁶² At 157 the N K-edge, the CIE spectra (see Fig. 3b) taken along the B_N and C_N peaks show a 158 signal at $4.6 \,\mathrm{eV}$ representing a transition to VES₃. For the CIE spectra along the C_N peak 159 we observe an additional feature from transition into VES₄ at $E_{trans} = 5.3 \,\mathrm{eV}$. From the 160 varying amplitudes of the spectral features in the CIE spectra we glean that there is strong 161 coupling of $CES_{N,1}$ to VES_3 and of $CES_{N,2}$ to VES_4 . The couplings of the CESs to other 162 final states are very weak as evidenced by the absence of additional peaks in the CIE spectra 163 at the N K-edge. 164

The GS RIXS map for enolic HBQ at the O K-edge is presented in Fig. 3d. The B_O and 165 C_O peaks (arising from excitation into $CES_{O,1}$ and $CES_{O,2}$) around 531.5 eV are of similar 166 intensity and are split by 0.2 eV. Fig. 3e shows the comparison of HERFD- and PFY-XANES 167 spectra at the O K-edge. The relative intensities between the B_O and C_O peaks drastically 168 change between the two spectra reflecting their different projections along the incident energy 169 axis emphasising coupling to different VESs. The CIE spectral cuts along $E_{inc} = 531.2 \text{ eV}$ 170 and $E_{inc} = 531.5 \,\mathrm{eV}$ at the O K-edge are presented in Fig. 3f. These spectra display several 171 cross-peaks highlighting the coupling of several VESs with $CES_{O,1}$ and $CES_{O,2}$. The CIE 172

¹⁷³ cuts taken at B_O and C_O show features at 3.8 eV and at 4 eV, corresponding to a transition ¹⁷⁴ into VES₁ and VES₂, respectively. Two additional features at 4.8 eV and 5.2 eV are higher ¹⁷⁵ in intensity and correspond to transition into VES₃ and VES₄, respectively.



Figure 3. Ground state RIXS of HBQ in its enol form at the N (a-c) and O K-edge (d-f). c) RIXS map of HBQ at the N K-edge. d) RIXS map of HBQ at the O K-edge. a,e) PFYand HERFD-XANES spectra. HERFD-XANES cuts are indicated by the diagonal dashed lines in the RIXS maps. b,f) Constant Incident Energy (CIE) cuts along the B peak (black) and C peak (grey). The cuts are marked by the vertical dotted lines in the 2D RIXS maps.

A comparison of the XANES spectra at the N and O K-edges (see Fig. 2) highlights the 176 small relative energy difference between the B and C peaks at the O K-edge compared to 177 the N K-edge. The difference between the two absorption edges can be explained by taking 178 a close look at the contributing molecular orbitals to the two CESs presented in Table S1 179 in the SI. From the table, we see that the transitions to $CES_{N,1}$ and $CES_{N,2}$ have strong 180 contributions by the $1\pi^*$ and $2\pi^*$ orbitals, respectively. This is in contrast to the transitions 181 to $CES_{O,1}$ and $CES_{O,2}$ where the $1\pi^*$ and $2\pi^*$ orbitals contributing equally to both states. 182 The mixed character of the CESs at the O K-edge causes the B_O and C_O peaks to be closer 183 in energy. The smaller oscillator strengths of the XANES spectral features at the O K-184 edge compared to the N K-edge are caused by the small overlap between the O1s and $1\pi^*$ 185 and $2\pi^*$ orbitals which mainly contribute to the $CES_{I,1}$, $CES_{I,2}$, in contrast the overlap of 186 the π^* orbitals with the N1s. A comparison of the HERFD-XANES and the PFY-XANES 187 spectra at the N (Fig. 3a,b) and O (Fig. 3e,f) K-edges, reveals changes in relative intensities 188

of the spectral features. These differences are strongly dependent on the chosen emission
energy at which the cut is taken, as shown in the SI (Fig. S6) and a direct comparison
between HERFD-XANES and XANES features should always be performed with caution.
However, the changes in relative intensities in HERFD spectra compared to XANES can
help in disentangling different contributions to the RIXS spectra.

¹⁹⁴ We can compare the strongest transition in energy transfer axis for the O and N K-edges ¹⁹⁵ by looking at their CIE spectral cuts along the B and C peaks (Figs. 3b and c). Due to the ¹⁹⁶ mixed character of the CESs at the O K-edge, we observe strong coupling to several VESs , ¹⁹⁷ which gives rise to the B_O^{1-4} and C_O^{1-4} features with similar spectral intensities while at the ¹⁹⁸ N K-edge we predominantly observe features from only one transition, B_N^3 . The different ¹⁹⁹ cross-peak intensities at the N and O K-edges show variation in couplings between different ²⁰⁰ electronic states depending on which atom is being investigated.

The above analysis of the calculated GS RIXS maps of HBQ at the O and N K-edges 201 highlights how we can disentangle the VESs coupled to CESs and thereby gain insights into 202 the character of the contributing molecular orbitals and into the electronic charge distribution 203 across the intramolecular H-bond. The relationship between X-ray spectral signatures and 204 H-bonding strengths has been discussed for imidazole²⁸ and acetic acid.⁵⁹ In the SI (Fig. S3). 205 we present calculated RIXS spectra of HBQ in its keto form as a limiting case for how the 206 proton transfer impacts the local electronic structure at the H-donor and acceptor atoms. We 207 also refer the reader to the discussion in the SI accompanying Fig. S3 and of the calculated 208 XANES spectrum for the GS HBQ keto form published earlier.⁸ In brief, we observe that 209 increased NH-bond strength leads to a blue-shift at the N K-edge and an increased OH-bond 210 strength leads to a blue-shift at the O K-edge. This can be attributed to a lowering in energy 211 of the corresponding 1s orbital which has previously been employed to distinguish the two 212 tautomers of 3-hydroxypyridine.⁴⁷ The cross-peaks in the RIXS maps show an overall red-213 shift of the spectral features in the energy transfer axis and larger energy differences between 214 the different transitions (see Fig. S3). The red-shift of the features along the energy transfer 215

axis can be explained by the smaller gap between GS and first VES in the keto form as
schematically presented in Fig. 1.

The position and amplitude of the peaks in the the RIXS spectra of HBQ highlight the 218 detailed electronic state coupling information that can be gained from the RIXS experiment 219 as compared to our previous XANES study. Due to the underlying processes, XANES is 220 only sensitive to CESs. In the RIXS process, the energy transfer axis contains additional 221 information on VESs and most importantly on the coupling between the CESs and VESs 222 at different atomic sites. In other words, the RIXS maps provide a complete mapping of 223 the electronic energy landscape revealing the occupied and unoccupied electronic states, 224 which dictate the (photo)chemical properties of a system. Additionally, a comparison of the 225 molecular orbitals involved in the different RIXS transitions give insights into how the local 226 electron densities impact the intramolecular H-bond in HBQ. In the following, we will show 227 how excited state RIXS can be used to directly follow the reorganization of the electronic 228 states in HBQ during and following ESIPT. 229

For computing the evolution of RIXS spectra during and following the ESIPT process, we 230 performed MD simulations on the first VES for 180 fs. Structures along the VES trajectory 231 were sampled in 1.2 fs intervals. The structural dynamics show that ESIPT takes place after 232 12 fs which is in good agreement with experimental studies.^{19,20} Transient RIXS maps were 233 calculated in 6 fs intervals on the VES_1 . We note that the analysis presented below is based 234 on a single trajectory. To confirm that the interpretation of the VES RIXS maps is not 235 biased by analysis of a single trajectory, we have compared the structural dynamics of 26 236 different trajectories, presented in the SI. The averaged results of atomic distances from the 237 calculations of the multiple trajectories confirms that vibrational coherence on the first VES 238 of HBQ following ESIPT survives averaging of 26 trajectories as presented in the SI Fig. S7. 239 Figures 4a-c show transient RIXS maps at the N K-edge calculated for structures sampled 240 at 0 fs, 7 fs, and 165 fs, respectively. Since the proton transfer takes place within 12 fs, the 241 RIXS map at 0 fs shows signals from molecules on the E^{*} state. The RIXS map at 7 fs is 242



Figure 4. Transient RIXS features at the N K-edge: a-c) Transient RIXS maps calculated for structures on the first VES after 0 fs, 7 fs and 165 fs. d) Transient CET cuts along the elastic line (indicated by horizontal lines in the 2D maps) and e) Transient CIE cuts of the A peak (vertical lines in 2D maps) for selected timepoints during the proton transfer. f) Amplitude of the anti-Stokes feature (AS) as a function of time. The inset plots the FFT amplitude of the oscillatory features.

representative of a molecule undergoing ESIPT, and the RIXS map at 165 fs is representative 243 of a molecule in the K^{*} state following proton transfer. As previously observed in the 244 calculated transient XANES spectra,⁸ an additional A_N^* peak appears to the red of the B_N^* 245 and C_N^* peaks along the incident energy. This feature corresponds to core excitation from 246 VES₁ into $CES_{N,1}^*$, simulating RIXS spectra after laser excitation (UV_{in} in Fig. 1b) and is 247 $1\,\mathrm{e\,V}$ to the red of the B^*_N peak. We also observe the appearance of a weak feature at negative 248 energy transfer energies (-4 eV to -1.5 eV) which we will refer to as the anti-Stokes feature 249 in the following. The anti-Stokes feature corresponds to excitation from the VES (VES_1) 250 into the $CES_{I,1}^*$ state and subsequent relaxation into the GS. Such features at negative energy 251 transfer have previously been experimentally observed in time-resolved RIXS studies at the 252 N K- and Fe L-edges of metal complexes.^{53,54} 253

In order to understand the time-dependence of the RIXS features during ESIPT, we look 254 at spectral cuts during 0 fs to 13 fs. The constant energy transfer (CET) spectral cuts along 255 the elastic line are presented in 4d and we see that the B_N and C_N move to higer incident 256 energies reflecting the change in electron density across the intramolecular H bond, as the 257 proton moves from the O to the N atom. Fig. 4e presents the CIE spectral cuts along the A_N^* 258 peak during ESIPT. We observe a strong blue-shift of the CIE features during ESIPT. The 259 strongest feature appears at $1.5 \,\mathrm{eV}$ and shifts towards $1.8 \,\mathrm{eV}$. This feature can be assigned 260 to the energy transfer from VES_4 to VES_1 , highlighting the time-evolving increase in energy 261 difference between the different VESs during ESIPT. The intensity of the spectral feature at 262 1.5 eV increases during ESIPT reflecting the changing overlap in electron density between 263 GS and CES orbitals as the proton transfers to the nitrogen atom. 264

Fig. 5 displays the transient RIXS maps and spectral cuts for the O K-edge. Panels a-c present the 2D RIXS maps at 0 fs, 7 fs, and 165 fs. Similiar to the N K-edge, an additional peak, A_O^* , appears at 5.5 eV to the red of the B_O peak. The B_O^* and C_O^* peaks are blueshifted by about 1 eV compared to their GS positions, emphasising the effect of valence excitation not only on the VESs but also the CESs. Fig. 5d shows the CET spectral cuts



Figure 5. Transient RIXS features at the O K-edge: a-c) Transient RIXS maps calculated for structures of the first VES after 0 fs, 7 fs and 165 fs. d) Transient CET cuts along the elastic line (indicated by horizontal lines in the 2D maps) and e) transient CIE cuts of the A peak (vertical lines in 2D maps) for selected timepoints during the proton transfer. f) Amplitude of the anti-Stokes feature (AS) as a function of time. The inset plots the FFT amplitude of the oscillatory features.

along the elastic line at the O K-edge, presenting a shift towards lower incident energies, corresponding to a decrease in energy of the CES*s during ESIPT. Fig. 5e shows the CIE spectral cuts along the A_O^* peak at the O K-edge, presenting the same trends as at the N K-edge with the strongest feature at 1.5 eV corresponding to transition from VES₄ to VES₁. This peak shifting towards 1.8 eV and increases in intensity during the first 13 fs reflecting the reorganization of the VESs and the time evolving electron density overlap between the CES and the GS during ESIPT.

Comparing the transient RIXS spectra at the N and O K-edges, the A peak has a much 277 stronger intensity at the O K-edge compared to the N K-edge. These differences in intensity 278 can be explained by the stronger overlap between VES_1 (the initial state of the RIXS process 279 after photoexcitation) and $CES_{O,1}^*$ than $CES_{N,1}^*$. The increase/decrease in energy of the 280 intermediate states observed through the shifts of the peaks in the CET spectral cuts at 281 the N and O K-edge, respectively, is linked to stabilisation of the atom-specific GS upon 282 formation of a H-bond. At the N K-edge, we additionally observe an increase in A_N^* , B_N^* 283 and C_N^* in the CET spectral cuts, while there is a slight decrease in these energy differences 284 at the O K-edge. The CIE spectral cuts (Fig. 4e and Fig. 5e) describe the same states 285 (VES) at both absorption edges, the peak positions and energy changes are the same at the 286 N and O K-edge, however the amplitudes are different. Looking at the 2D RIXS maps at 287 the N K-edge, we observe that the relative intensity of the cross peaks compared to the GS 288 increases during ESIPT and decreases following ESIPT. The opposite trend is observed at 289 the O K-edge. This highlights that the dynamic states are highly coupled at N K-edge but 290 less so at the O K-edge. 291

The dynamics of the transient RIXS signals can be investigated in more detail by integrating the anti-Stokes feature and following its intensity and position as function of time. Since the anti-Stokes feature is spectrally isolated and contains information on both CES, and VES, it is ideally suited to monitor the dynamics during and following ESIPT. The oscillations in intensity following ESIPT at both the N and O absorption edges are presented

in Fig. 4f and Fig. 5f with the Fourier transform in the insets. Even though the anti-Stokes 297 features at both absorption edges describe the coupling to the same VESs, the strength of 298 this coupling depends on the absorption edge. The insets of Figs. 4 and 5 display the Fourier 299 Transform amplitudes of the oscillations in the intensity of the anti-Stokes feature at the 300 N and O K-edge respectively. To shed light on how the electronic structure at the N and 301 O K-edges is modulated by the observed frequencies, a comparison with the excited-state 302 normal modes of the K^{*} state was made. See descriptions of the vibrations in Fig. S11 in the 303 SI. The vibrations between $1000 \,\mathrm{cm}^{-1}$ to $2000 \,\mathrm{cm}^{-1}$ correspond to backbone vibrations in-304 cluding carbonyl stretching and N–H bending. We are limited to observing vibrations within 305 the $450 \,\mathrm{cm}^{-1}$ to $2800 \,\mathrm{cm}^{-1}$ range given the sampling rate and length of the MD simulations. 306 It is worth noting that vibrational coherences have been observed in time resolved optical 307 experiments of HBQ, but those have focused on vibrations below 1000 cm⁻¹.^{11,63,64} Recently 308 we have observed high frequency vibrational coherences in HBQ during and following proton 309 transfer.⁶⁵ 310

We explore the time-dependent changes in the position of the anti-Stokes feature along 311 the E_{inc} and E_{trans} axes at the N and O K-edges in Fig. S11 in the SI. We see an increase 312 in the energy transfer energy of the anti-Stokes feature within 12 fs followed by oscillations 313 of the spectral position at the N and O K-edges. The identical spectral changes at the O 314 and N K-edges are due to the fact that the spectral shifts in the energy transfer axis are 315 given by the energy differences between the VESs. We note that these oscillations will be 316 different from those observed in an UV transient absorption due to different selection rules 317 of the RIXS process. 318

Figure 6 further explores the correlation between the movement of the proton during ESIPT and the spectral position of the anti-Stokes feature along the E_{trans} axis. The proton movement is defined as $\Delta R = R(N - H) - R(O - H)$. We observe a strong correlation between ΔR and the spectral position of the anti-Stokes feature within the first 12 fs, i.e. during ESIPT where the structural changes are dominated by the proton transfer. During

the early time points the shift in E_{trans} of the anti-Stokes feature as a function of ΔR 324 corresponds to the mapping of dynamic change in energy difference between GS and first 325 VES moving from enol to keto HBQ as presented in Fig. 1 where the difference between GS 326 and first VES decreases along the ESIPT reaction coordinate with a slope of $1.6 \,\mathrm{eV}\,\mathrm{\AA}^{-1}$. 327 The "spectral ruler" in Fig. 6 demonstrates that the spectral changes linked to the structural 328 changes are large enough to be detected with currently available experimental resolutions 329 at XFEL endstations. From Fig. 6 we see that the correlation between E_{trans} of the anti-330 Stokes feature and ΔR is not valid following ESIPT because several vibrational modes in the 331 molecule influence both structure and spectral signatures causing a break-down of the linear 332 relationship. In the SI (Fig. S11) we present the correlation between structure and spectral 333 shifts in incident energy and energy transfer for several features in the N and O K-edge RIXS 334 maps, showing that a similar trend can be observed for all of them. However, the magnitude 335 of the spectral changes depends on the investigated feature, highlighting that different states 336 have varying sensitivity to the enol-keto transition. The comparison shows that the strongest 337 spectral change is observed when looking at the changes in energy transfer of the anti-Stokes 338 feature. This work highlights the potential of using transient RIXS spectroscopy to gain 339 direct structural insight into structural dynamics on a molecular level. 340



Figure 6. E_{trans} of the AS feature as function of $\Delta R = R(N - H) - R(O - H)$ from early time-points (blue) to late time-points (red). The red line shows a linear fit of the changes in E_{trans} within the 12 fs, giving a slope of 1.6 eV Å^{-1} .

³⁴¹ In this work, we have presented a study of GS and transient RIXS signals of an H-bonded

model complex, HBQ. Investigation of the GS spectra shows the potential of RIXS to fully map the electronic structure of a given system, including both core and valence excited states. Additionally, analysis of the cross peaks in the 2D RIXS maps gives important insights into the coupling between different electronic states which contains information on the contributing molecular orbitals. By considering RIXS maps from the perspective of the proton-donor (O) and -acceptor (N) atoms, the intricate coupling between electronic and atomic structure is revealed.

The time-resolved RIXS spectra show how structural changes affect the coupling between different states. The spectral changes in the RIXS maps can directly be correlated to structural changes during the proton transfer, giving a spectral 'ruler' to measure structural changes. By using and comparing different spectral features for this ruler, we gain insights into what affects the amplitude of spectral changes in X-ray spectroscopy and the sensitivity of different states to structural changes. The dynamic changes additionally reveal the vibrations underlying the ESIPT process.

With the new capabilities of soft X-ray endstations at high repetition rate XFELs, timeresolved RIXS studies, like presented here, will soon be possible experimentally and allow for direct observation of the intricate changes in electronic structure during ESIPT.

359 Computational Methods

All calculations were performed with the open-source NWChem computational chemistry program.^{66–68} The ground-state (GS) HBQ geometry was optimized at the DFT level of theory using the PBE0 functional⁶⁹ and def2-TZVP basis set.⁷⁰ The solvent-environment of toluene was modelled using the COnductor-like Screening MOdel (COSMO)^{71,72} with a dielectric constant of $\epsilon = 2.38$. UV/Vis spectra were computed at the TDDFT level of theory,⁷³ while the XANES spectra at the O and N K-edges were computed with the restricted energy window extension of TDDFT.⁷⁴ XANES spectra were shifted relative to experimental data. Since no experimental spectra on HBQ were available, the shift was determined by performing calculations for 3-hydroxypyridine at the same level of theory and compare the results to experimental data from literature,⁴⁷ resulting in a shift of 11 eV (see SI). For this reason the spectra presented here are shifted compared to our previous work where spectra were not shifted to match experimental data.

³⁷² RIXS calculations were performed using the Kramers–Heisenberg equation, ³⁶

$$\sigma(\omega',\omega) = \frac{\omega'}{\omega} \sum_{fi} |F_{fi}(\theta)| \left[\frac{(\omega_i \omega_f \alpha)^2}{(\omega - \omega_i)^2 + \Gamma_i^2/4} \right]$$
(1)

where the amplitude σ is a function of the incoming ω' and emitted photons ω with inter-373 mediate state i and final state f where ω_j and the lifetime Γ_j of state. RIXS spectra were 374 simulated by coupling XANES and UV/Vis transitions following the protocol published ear-375 lier. 40,75 A lifetime broadening of $0.13 \,\mathrm{eV}$ and $0.27 \,\mathrm{fs}$ were applied for the core-excited state 376 at the N and O K-edge, respectively, and of 0.13 eV for the valence-excited state at both 377 absorption edges. In order to minimize the effects of elastic scattering, the angle between the 378 polarization of the incoming photon and the propagation direction of the outgoing photon, 379 θ , was set to zero. 380

For RIXS maps on the first VES, excited-state molecular dynamics simulations at the 381 TDDFT level werw performed on VES₁ using analytical gradients.⁷⁶ Excited state simula-382 tions were run using the Gaussian basis set quantum molecular dynamics (QMD) module in 383 $\rm NWChem^{77}$ and 10 au (0.24 fs) timesteps. No thermostat was used. We ran 26 excited state 384 trajectories starting from the optimised GS structure, simulating vertical excitation. For each 385 trajectory different initial velocities were randomly selected from the Maxwell–Boltzmann 386 distribution at room temperature (298.15 K). We did not employ surface hopping methods 387 since the first and second VES are energetically well separated (by $\geq 0.5 \,\mathrm{eV}$) and the proton-388 transfer occurs on VES₁. Neither nuclear quantum effects⁷⁸ nor explicit dephasing effects⁷⁹ 389 were included in our dynamics simulations. 390

We computed transient UV/vis, XANES, and RIXS spectra for geometries extracted 391 every 2.4 fs from a single excited-state trajectory. As detailed in our previous publication, 392 these computations on the first VES required three steps for each of the selected time points.⁸ 393 First, the ground-state molecular orbitals were obtained by converging the self-consistent 394 field (SCF). Second, to simulate the HOMO-LUMO excitation of an electron, occupancy 395 of the alpha HOMO and LUMO orbitals were swapped and the SCF reconverged. Third, 396 spectra were computed based on the molecular orbitals for VES_1 . The resulting broadened 397 spectra are shown in Figs. 4 and 5. 398

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⁶⁵⁰ Supporting Information Available

⁶⁵¹ Supporting information is available.

652 TOC Graphic

