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Bent Naphthodithiophenes: Synthesis and Characterisation of Isomeric Fluorophores†

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Thiophene-containing heteroarenes are one of the most well-known classes of π-conjugated building blocks for photoactive molecules. Isomeric naphthodithiophenes (NDTs) are at the forefront of this research area due to their straightforward synthesis and derivatization. Notably, NDT geometries that are bent – such as naphtho[2,1-*b*:3,4-*b'*]dithiophene (α-NDT) and naphtho[1,2-*b*:4,3-*b'*]dithiophene (β-NDT) – are seldom employed as photoactive small molecules. This report investigates how remote substituents impact the photophysical properties of isomeric α- and β-NDTs. The orientation of the thiophene units plays a critical role in the emission: in the α**(OHex)R$_2$** series conjugation from the end-caps to the NDT core is apparent, while in the β**(O*i*-Pent)R$_2$** series minimal change is observed unless strong electron acceptors, such as β**(O*i*-Pent)(PhCF$_3$)$_2$**, are employed. This push-pull Acceptor–Donor–Acceptor (A–D–A) fluorophore exhibits positive fluorosolvatochromism that correlates with increasing solvent polarity parameter, $E_T(30)$. In total, these results highlight how remote substituents are able to modulate the emission of isomeric bent NDTs.

1 Introduction

Heteroarenes, polyaromatic hydrocarbons (PAHs) bearing main group elements, are well established as building blocks for π-conjugated functional materials.[1–4] Thiophene-containing heteroarenes with benzene, naphthalene, or anthracene cores have been thoroughly investigated as electron rich moieties within both small molecule and polymer systems with applications in organic electronics.[1,5,6] Of the thiophene-based heteroarenes, naphthodithiophenes (NDTs) are of particular interest due to the plethora of isomers available and their straightforward derivatisation.[7,8] Linear and angular NDTs shown in Figure 1a represent two classes of isomeric NDTs that have found applications throughout organic electronics, most notably as organic field-effect transistors (OFETs),[9–11] organic light-emitting diodes (OLEDs),[12] organic[13,14] and dye-sensitized[15] solar cells, and more recently as hole-transporting materials (HTMs).[12,16,17] Due to the orientation of the residual α-positions (2- and 7-positions, labelled in Figure 1a), NDT-1 and NDT-2 are considered to be linear rods, while NDT-3 and NDT-4 are described as angular.[7]

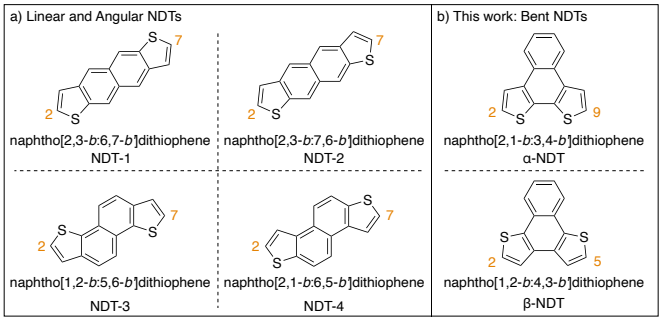

Fig. 1 Classes of Naphthodithiophenes. a) Isomeric linear and angular naphthodithiophenes. b) This work: isomeric bent naphthodithiophenes.

Alternatively bent NDTs – such as naphtho[2,1-*b*:3,4-*b'*]dithiophene (α-NDT) and naphtho[1,2-*b*:4,3-*b'*]dithiophene (β-NDT) shown in Figure 1b – are known building blocks in polymeric materials,[18] but are underrepresented as functional small molecules. At their core, unfunctionalized α- and β-NDTs are isoelectronic to triphenylene but can be easily fused to other motifs to yield electron deficient cores (e.g. phenazines[19] or imides[20]). The α-positions of α- and β-NDTs, labelled in Figure 1b, offer a facile means to modulate the emissive properties of the core via peripheral derivatisation (i.e. V-shaped[21] Acceptor–Donor–Acceptor systems). These bent push-pull fluorophores[22] are reminiscent of Höger and co-workers isomeric dithienylphenazines (termed α- and β-DTPs), whose systems have inverted electronics (i.e. the electronics are aligned to be

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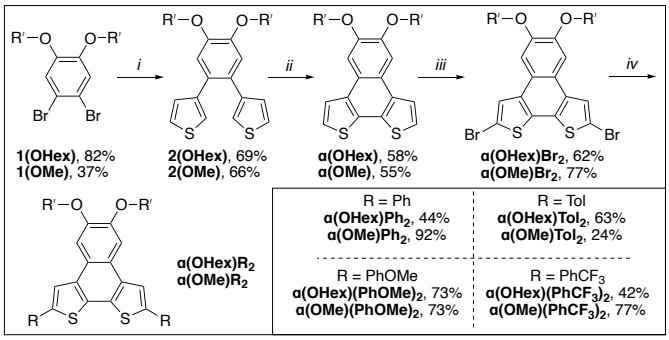
Donor–Acceptor–Donor systems). Their series of α- and β-DTPs were shown to have variable conjugation paths based on the isomer employed, ultimately yielding photoactive molecules that are fluorescent (α-DTP) or dual emitters (β-DTP).[23,24] With this in mind, probing how the peripheral functionalization of α- and β-NDTs impacts the photoactive cores, and ultimately their application as push-pull fluorophores, is of interest.

Herein we investigate the structure-property relationships in a series of isomeric bent NDTs – naphtho[2,1-*b*:3,4-*b*']dithiophene (α-NDT) and naphtho[1,2-*b*:4,3-*b*']dithiophene (β-NDT) – with variable end-caps. The initial premise of this research effort was to establish if the peripheral **R**-groups are (1) in communication with the NDT core and (2) if there are distinct differences between the isomers. NMR spectroscopy can afford preliminary evidence of communication to the core in the ground state via modulation of interior resonances far removed from the **R**-groups; while emission spectra are informative of excited state communication. Ultimately, there are clear differences in the photophysical properties between the isomers, with α-NDT being more receptive to all types of peripheral substituents and β-NDT only being responsive to electron-deficient end-caps.

2 Results and Discussion

2.1 α(OR')R_2 Series: Synthesis and Structural Analysis

With the goal of having both solution- and solid-state metrics for the **α(OR')R_2** series two monomers with differing solubility are needed. For the solution-state analysis, hexyloxy groups were used to ensure that all α-NDT derivatives are soluble across a range of concentrations, particularly with aggregation being a concern for similar PAH systems.[25] To investigate solid-state metrics methoxy groups were used to maximize the chance of growing single crystals for x-ray diffraction. Both **α(OHex)R_2** and **α(OMe)R_2** can be synthesized via a streamlined bottom-up approach. Scheme 1 details the synthetic path to the **α(OHex)R_2** series beginning with the alkylation of 4,5-dibromocatechol[26] with iodohexane to yield solubilized building block **1(OHex)**. Installation of the thiophene units is accomplished via palladium-mediated cross-coupling to afford **2(OHex)** in modest yield. Planarisation of **2(OHex)** under typical Scholl oxidation conditions[27] affords **α(OHex)** as an off-white crystalline solid after column purification. Bromination of the open α-positions with NBS yields **α(OHex)Br_2**, a versatile core molecule for attaching end-caps with variable electronic properties. Commercially available phenyl boronic acid (**R** = **Ph**, H σ_p = 0.00) and *para*-functionalized phenyl boronic acid derivatives *p*-tolyl boronic acid (**R** = **Tol**, Me σ_p = -0.17), *p*-methoxyphenyl boronic acid (**R** = **PhOMe**, OMe σ_p = -0.27), and *p*-(trifluoromethyl)phenyl boronic acid (**R** = **$PhCF_3$**, CF_3 σ_p = +0.54) were chosen as end-caps due to their variable electronic properties that are poised to be in resonance with the NDT cores. Due to this arrangement for the **R**-groups Hammett parameters[28] for σ_p, which include resonance effects, are more informative than electronically isolated σ_m. Suzuki cross-coupling of the boronic acids proved to be efficient across all derivatives, with modest isolated yields for the **α(OHex)R_2** series (Scheme 1, inset).



Scheme 1 Synthesis of α(OR')R_2 series. (i) 3-thiophene boronic acid, $Pd(PPh_3)_2Cl_2$, 2M K_2CO_3, THF, 80 °C, overnight; (ii) $FeCl_3$, $MeNO_2$, DCM, 0 °C to RT, 1 hr; (iii) NBS, $CHCl_3$, RT, 2 d; (iv) R boronic acid, $Pd(PPh_3)_2Cl_2$, 2M K_2CO_3, THF, 80 °C, overnight. Inset: isolated yields for α(OHex)R_2 and α(OMe)R_2 series.

The **α(OMe)R_2** series follows an identical synthetic path as **α(OHex)R_2** (shown in Scheme 1), with the exception of starting from 4,5-dibromoveratrole.[29] The solubility of the **α(OMe)R_2** series is demonstrably worse than the hexyloxy series, which proved challenging for purification and analysis (see SI). Due to the poor solubility of the **α(OMe)R_2** series these derivatives were used almost explicitly for crystallography. Single crystals of sufficient quality for X-ray diffraction were grown for each derivative via vapour diffusion of hexanes into saturated solutions of **α(OMe)R_2** in THF (see SI). Of particular interest is the inter-ring bond length, which can detail the level of communication the remote substituents have with the α-NDT core. The C–C linkage between the α-NDT core and the appended **R**-groups are expected to display minor elongation due to torsional strain between the ring systems (C–C $\approx$ 1.45 Å),[30] with measurements under this value being indicative of increased conjugation with the α-NDT core. Each **α(OMe)R_2** derivative has an inter-ring bond length of $\approx$ 1.46 Å, indicative of negligible bond elongation with no dependence on the **R**-group.

With these solid-state metrics established, we turned our attention to the solubilized **α(OHex)R_2** derivatives for solution-state structural analysis. Specifically of interest are resonances far removed from the **R**-groups that are part of the NDT core, as these signals are expected to shift if there is communication from the end-caps to the core. The NDT cores themselves are electron-rich, therefor **R**-groups that are electron withdrawing are expected to function as intramolecular acceptor–donor–acceptor (A–D–A) systems, while electron-rich groups are expected to yield highly polarized all donor systems. Intramolecular A–D–A species are posited to increase the contribution of the quinoidal resonance, a highly desirable characteristic in photoactive molecules.[21] Key proton resonances 2b and 3b are highlighted in Figure 2 and are matched between each derivative of the **α(OHex)R_2** series to ease comparison.

Due to the planar nature of the NDT cores aggregation is likely, thus comparison of NMR spectra at similar dilute concentrations is required. Inspection of the aromatic region of dilute ($\leq$10 mM) solutions of each **α(OHex)R_2** species reveals slight perturbations of the interior 2b proton ($\Delta\delta \approx$ 0.20 ppm, Figure

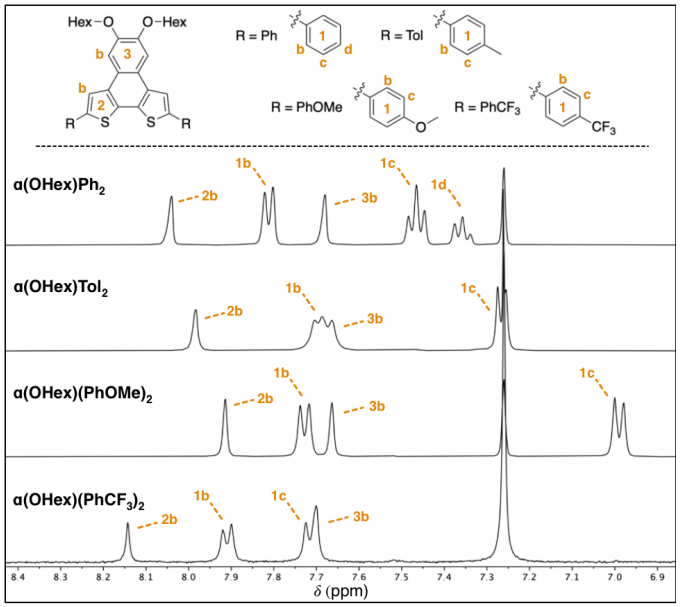


Fig. 2 Partial ^{1}H NMR spectra for the α(OHex)R_2 Series with assignment of aromatic resonances (400 MHz, $CDCl_3$). See SI for full spectra and assignments.

2) and carbon ($\Delta\delta \approx 3$ ppm, see SI) resonances. Modulation of the 2b resonances is quite unique: these signals are far removed from the functional group and indicate that there is remote communication to the NDT core upon end-capping. Relative to parent **α(OHex)Ph$_2$** the electron donating groups shift the 2b resonance upfield with **α(OHex)Tol$_2$** ($\Delta\delta$ -0.06 ppm) and **α(OHex)(PhOMe)$_2$** ($\Delta\delta$ -0.13 ppm) both being above the threshold of significance ($\Delta\delta \geq$0.05 ppm). **α(OHex)(PhCF$_3$)$_2$** is shifted downfield in comparison ($\Delta\delta$ +0.10 ppm), indicative of a shift towards a higher energy state (with respect to parent **α(OHex)Ph$_2$**). Overall, the shift differences observed for the 2b proton resonances are quite small ($\Delta\delta <$ 0.25 ppm); thus, inspection of the ^{13}C signals for the 2b resonances could be informative. Indeed, the ^{13}C resonances follow the same trend, wherein **α(OHex)Tol$_2$** ($\Delta\delta$ -0.6 ppm) and **α(OHex)PhOMe$_2$** ($\Delta\delta$ -1.1 ppm) are shifted upfield relative to **α(OHex)Ph$_2$** and **α(OHex)(PhCF$_3$)$_2$** is shifted downfield ($\Delta\delta$ +1.6 ppm). This trend fits well with σ_p for the **R**-groups for both the observed ^{1}H and ^{13}C resonances of 2b, while σ_m yields poor fits (see SI). To ensure these observations are not artefacts of aggregation, a series of variable concentration solutions were probed for the **α(OHex)R$_2$** series (see SI, 50–1 mM in $CDCl_3$). These data for **α(OHex)Tol$_2$** and **α(OHex)(PhOMe)$_2$** suggest that there is minimal propensity to aggregate in $CDCl_3$ ($\Delta\delta \leq$0.12 ppm for all resonances), with measured K_d of $\approx 3.2 \times 10^{-4}$ M and $\approx$ 1.2 M for these two derivatives respectively.[31,32]

To verify the attribution of the small perturbations observed via NMR to the remote **R**-groups, we modelled the NMR spectra of the **α(OHex)R$_2$** series computationally. To simplify simulations and clarify the assignment of NMR features, we replaced the hexyloxy groups with methoxy. To differentiate from the synthesized **α(OMe)R$_2$** series the model series is represented as *α(OMe)R$_2$'*. In our ^{1}H and ^{13}C calculated NMR spectra, we rely on the widely adopted non-relativistic density functional theory (DFT) protocol with gauge including atomic orbitals (GIAO)[33,34] as implemented in the Gaussian 16[35] programme suite. The justification of the non-relativistic approach comes from the smallness of the heavy atom effect in compounds containing sulfur and lighter elements only.[36–40] In all DFT calculations, we use a popular global hybrid exchange-correlation functional (XCF) approximation B3LYP[41,42] widely regarded in the literature as a reliable choice[43–58] even in light of recent advances.[45,49,51,59–62] Finally, we choose to work with polarization-consistent basis sets developed by Jensen for rational and systematic reduction of basis-set error specifically in DFT methods.[63–65] In particular, we use pc-2[66] for geometry optimisations and vibrational frequency calculations, and aug-pc-Sseg-2[67] for modelling NMR spectra. The latter set belongs to the pc-Sseg-*n* family of basis sets augmented with additional tight exponents for the description of the electronic density near atomic nuclei and optimised for nuclear magnetic shielding. These sets have been known to markedly outperform other available choices for core-related properties.[68,69] Due to the poor solubility of **α(OHex)(PhCF$_3$)$_2$** – and the entire **α(OMe)R$_2$** series – only experimental resonances extracted from the ^{1}H–^{13}C HSQC spectra were used for comparison to calculated parameters. Comparison of the ^{1}H and ^{13}C isotropic shieldings for the model *α(OMe)R$_2$'* series to the synthesized **α(OHex)R$_2$** (shown in Figure 3) and **α(OMe)R$_2$** (see SI) series yield quality fits ($r^2 >$ 0.9), further indicating that the modulation of the interior chemical shifts is directly related to the remote **R**-groups.

With the model *α(OMe)R$_2$'* series validated, we sought to probe how the **R**-groups impact the aromaticity of the α-NDT core. Nucleus independent chemical shift (NICS)[70] calculations are a well-known computational means to assess the aromatic (or antiaromatic) character of heteroarene systems.[71–73] Specifically, NICS-XY scans[74,75] wherein the ghost atom is set to 1.7 Å above the planar α-NDT core were performed to ascertain if there are electronic effects in the core that were not apparent via traditional spectroscopic means. In Figure 3b, we detail the results of the NICS-XY scans for the *α(OMe)R$_2$'* series referenced against a model compounds without **R**-groups, *α(OMe)'*. The *α(OMe)R$_2$'* cores are weakly paratropic no matter the **R**-group (NICS$_{1.7\pi ZZ}$ values $\approx$ 4–6.5 ppm over the path of the probe); there are minor deviations in the 'B' ring that follow the Hammett parameter (σ_p specifically) of the **R**-group (i.e. there is an increase in aromatic character as the **R**-group becomes more electron deficient).

2.2 α(OHex)R$_2$ Series: Photophysical Properties

Our focus shifted to elucidating the photophysical properties of the **α(OHex)R$_2$** series upon completion of their structural authentication. UV-visible absorption spectra for the **α(OHex)R$_2$** series, shown in Figure 3a, display defined (despite being broadened) vibronic structure in-line with known α-NDT containing materials.[76] Table 1 highlights several key metrics of these data, where there is a slight bathochromic shift when progressing from **α(OHex)Ph$_2$** to stronger electron donating groups in **α(OHex)Tol$_2$** and **α(OHex)(PhOMe)$_2$** ($\Delta\lambda_{abs,max}$ 5 nm). The

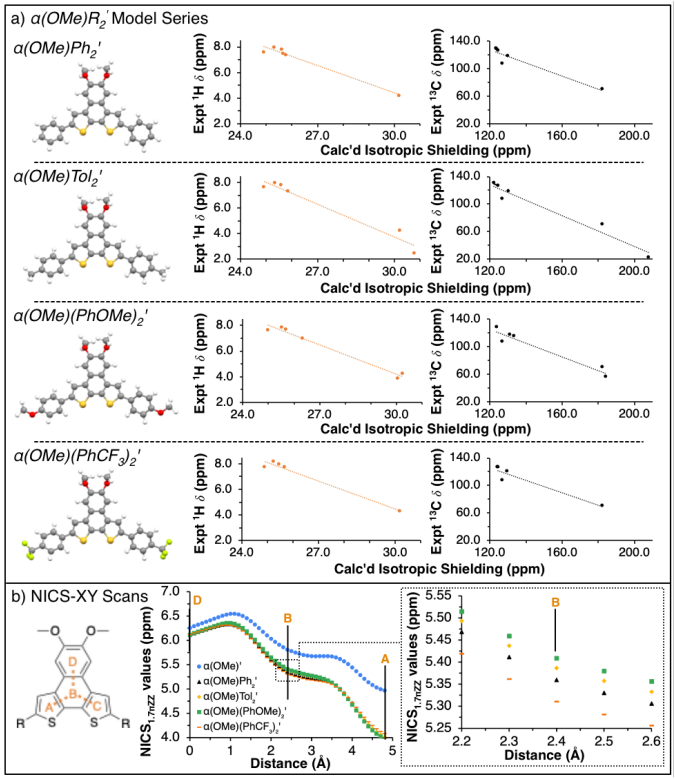


Fig. 3 Computational Analysis of $\alpha(Me)R_2'$ Series. a) Ball-and-stick representations of model compounds $\alpha(OMe)R_2'$ and their respective experimental vs calculated NMR plots (^{1}H is shown in orange, ^{13}C is given in black).; b) Partial NICS$_{1.7\pi ZZ}$ scans for $\alpha(OMe)R_2'$ with D–B–A path highlighted in orange. Inset: enhanced view of the central B ring.

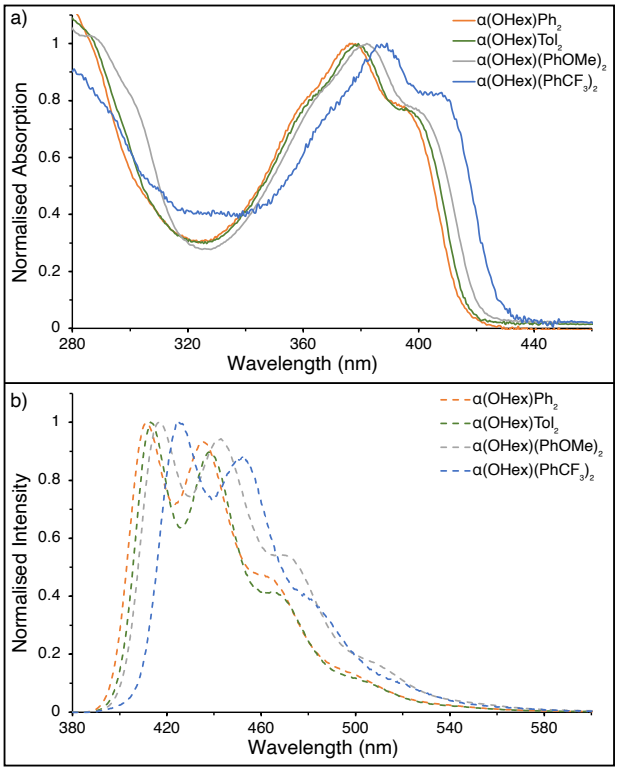


Fig. 4 Photophysical measurements for α(OHex)R$_2$ Series. a) UV-visible spectra; b) Emission spectra (λ_{ex} = 365 nm). All samples were prepared under an inert atmosphere with degassed dry THF.

Table 1 α(OHex)R$_2$ Series spectral data[a]

α(OR')R$_2$	ε M^{-1} cm^{-1}	$\lambda_{abs,max}$ nm	$E_{g,opt}$[b] eV	λ_{em}[c] nm	Φ_{PL}[c]
α(OHex)Ph$_2$	3.78×10^4	377	2.99	412	0.25
α(OHex)Tol$_2$	3.67×10^4	379	2.97	416	0.25
α(OHex)(PhOMe)$_2$	3.40×10^4	382	2.94	419	0.40
α(OHex)(PhCF$_3$)$_2$	1.18×10^4	389	2.89	427	0.24

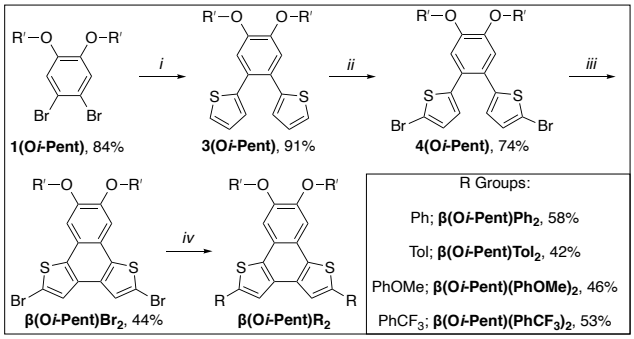
[a] UV-visible and emission spectra were measured as solutions in dry, degassed THF under inert atmosphere
[b] $E_{g,opt}$ estimated from the tangent of the absorption onset, $\lambda_{abs,onset}$
[c] λ_{ex} = 365 nm

lone A–D–A system, **α(OHex)(PhCF$_3$)$_2$**, is quite distinct with a 12 nm bathochromic shift relative to parent **α(OHex)Ph$_2$**. The optical band gap ($E_{g,opt}$) was estimated from the tangent of the absorption onset, with **α(OHex)(PhCF$_3$)$_2$** having the smallest gap ($E_{g,opt}$ 2.89 eV). The emission spectra (Figure 3b) have strong, detailed, vibronic structure.

The fine vibronic structure observed in the emission spectra are indicative of enhanced conjugation in the excited state through rigidification of the molecules (as compared to the relatively structure-less absorption spectra). For α-NDT containing molecules this implies stabilisation (and an increase in) the quinoidal resonance contribution.[77] The $E_{g,opt}$ values obtained experimentally are in agreement with the calculated band gap, $E_{g,calc}$, for each derivative (see SI). Photoluminescence quantum yield (Φ_{PL}) measurements were undertaken for each derivative under inert atmosphere in dry THF. **α(OHex)(PhOMe)$_2$** provided the highest quantum yield ($\Phi_{PL} \approx 0.40$) in THF, while **α(OHex)Ph$_2$** ($\Phi_{PL} \approx 0.25$), **α(OHex)Tol$_2$** ($\Phi_{PL} \approx 0.25$), and **α(OHex)(PhCF$_3$)$_2$** ($\Phi_{PL} \approx 0.24$) are within the error of the technique (±10%).

2.3 β(Oi-Pent)R$_2$ Series: Synthesis and Structural Analysis

The poor solubility of the **α(OHex)R$_2$** series led us to reconfigure the solubilizing groups for the β-NDT series to a short, branched alkoxy group that promotes solubility but still allows for crystallisation. Alkylation of 4,5-dibromocatechol with *iso*-pentyl bromide efficiently affords **1**(**Oi – Pent**) as a colorless oil that is converted to **3**(**Oi – Pent**) via Suzuki cross-coupling with 2-thiophene boronic acid in modest yield on the gram scale. This is quite astonishing considering this boronic acid coupling partner is known to undergo moderate protodeborylation under traditional cross-coupling conditions.[78] Blocking of the unfunctionalized α-positions of the thienyl groups is required prior to planarization to mitigate polymerization (or rearrangement).[76,79] With this in mind, bromination[80] of the residual α-positions of **3**(**Oi – Pent**) affords **4**(**Oi – Pent**) in modest yield. In our hands traditional Scholl oxidation conditions were inefficient (yields < 10%) to planarize the brominated species, but organic oxidant and acidic conditions (DDQ, methanesulfonic acid, DCM)[80,81] proved to be far superior in accessing **β(Oi-Pent)Br$_2$** in modest yield (>40% isolated yields on 500 mg scale).



Scheme 2 Synthesis of β(O*i*-Pent)R_2 series. (i) 2-thiophene boronic acid, $Pd(PPh_3)_2Cl_2$, 2M K_2CO_3, THF, 80 °C, overnight; (ii) NBS, $CHCl_3$, RT, 2 d; (iii) DDQ, $MeSO_3H$, DCM, 0 °C to RT, 1 hr; (iv) R boronic acid, $Pd(PPh_3)_2Cl_2$, 2M K_2CO_3, THF, 80 °C, overnight. Inset: isolated yields for β(Oi-Pent)R_2 series.

End-capping of **β(Oi-Pent)Br_2** with the aryl boronic acid derivatives yields the **β(Oi-Pent)R_2** series as off-white to pale yellow solids. Exchanging the solubilizing groups for *iso*-pentyloxy resulted in an appreciable improvement in solubility, with all **β(Oi-Pent)R_2** being soluble in common organic solvents (e.g. 50 mM in $CDCl_3$). The central goal of using the *iso*-pentyloxy groups is to thread the needle between solubility and the ability to grow single crystals for analysis. After several attempts with all derivatives, single crystals suitable for x-ray diffraction were grown via vapor diffusion of hexanes into a saturated solution of **β(Oi-Pent)Tol_2** in THF. The crystal obtained for **β(Oi-Pent)Tol_2** is dimeric, wherein the two units are aligned in the same direction with the β-NDT cores offset (i.e. a slip-stacked arene-arene stacking arrangement) with a distance between the dimers of ≈ 3.6 Å(see SI). The inter-ring C–C bond length for **β(Oi-Pent)Tol_2** is 1.46 Å, nearly identical to the same metrics obtained in the **α(OMe)R_2** series.

The partial aromatic region shown in Figure 4 reveals the interior 2b proton resonance (as well as the ^{13}C resonance) is responsive to remote functionalization. These perturbations of the 2b proton and carbon resonances are quite similar to those observed in the **α(OHex)R_2** series, with the **β(Oi-Pent)R_2** series having $\Delta\delta \approx 0.2$ ppm for ^{1}H and $\Delta\delta \approx 2$ ppm for ^{13}C. Fitting these chemical shifts to both σ_p and σ_m Hammett parameters reveals a quality correlation to σ_p, albeit not as pure of a fit as with the **α(OHex)R_2** series. Although the **β(Oi-Pent)R_2** derivates are quite soluble, one concern for the poor fit (r^2 0.7) for the ^{1}H spectra is aggregation. The orientation of the thiophene rings in the β-NDT isomer give the molecules a pronounced bent shape (the approximate bite angle of the terminal substituents is 110°) in comparison to the α series (bite angle ≈ 125°), which may impact their propensity to aggregate. Solutions of **β(Oi-Pent)Tol_2** and **β(Oi-Pent)(PhOMe)$_2$** in $CDCl_3$ were prepared and analyzed across a range of concentrations (50–1 mM, see SI) to probe this concern. Only chemical shifts in which $\Delta\delta \geq 0.05$ ppm were included in the fitting analysis,[31,32] revealing K_d values of 1.1 M and 1.5 M for **β(Oi-Pent)Tol_2** and **β(Oi-Pent)(PhOMe)$_2$**, respectively. These values are in-line with the **α(OHex)R_2** series, highlighting that the minor geometric differences in the bent NDTs do not impart vastly different aggregation potential.

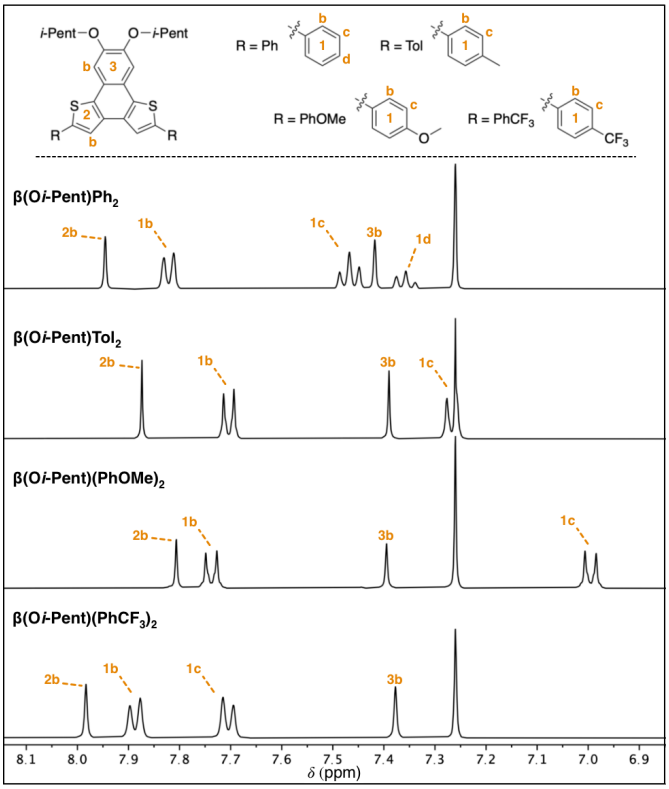


Fig. 5 Partial ^{1}H NMR spectra for the β(O*i*-Pent)R_2 Series with assignment of aromatic resonances (400 MHz, $CDCl_3$). See SI for full spectra and complete chemical shift assignments.

Following the same computational methodology as the α-NDT series, model compounds bearing methoxy groups were used to calculate both the NMR spectra and perform NICS-XY scans. The model β-NDT series, *β(OMe)R_2'*, reveals that the experimental ^{1}H and ^{13}C resonances obtained from the HSQC spectra for each derivative yield quality fits ($r^2 > 0.9$) with the calculated isotropic shielding values (see SI). NICS-XY scans across the β-NDT core yields a similiar pattern to the *α(OMe)R_2'* series, with the central differences being the $NICS_{1.7\pi ZZ}$ values of the 'A' ring (*β(OMe)R_2'* are ≈ 0.4 ppm lower than *α(OMe)R_2'*). The 'B' ring follows the same trend with σ_p, although the differences between each derivative are minute ($\Delta NICS_{1.7\pi ZZ}$ values ≈ 0.10 ppm, see SI).

2.3.1 β(Oi-Pent)R_2 Series: Photophysical Properties

The change in potential conjugation path between the α- and β-isomers opens the possibility of the β-NDT isomer to break the aromaticity of the upper aromatic ring (labelled as ring 3 in Figure 4), as is observed in Höger and co-workers β-DTP system.[23,24] Disrupting this Clar sextet[82] is unlikely, thus conjugation from the **R**-groups to solely the thiophene units is expected. Figure 5 details the UV-visible and emission spectra of the **β(Oi-Pent)R_2** series, revealing the β-isomer to be quite different than the **α(OHex)R_2** series.

First: there are no discernible differences in the photophysical properties between the **β(Oi-Pent)Ph_2**, **β(Oi-Pent)Tol_2** and **β(Oi-Pent)(PhOMe)$_2$** derivatives (Table 2). These derivatives

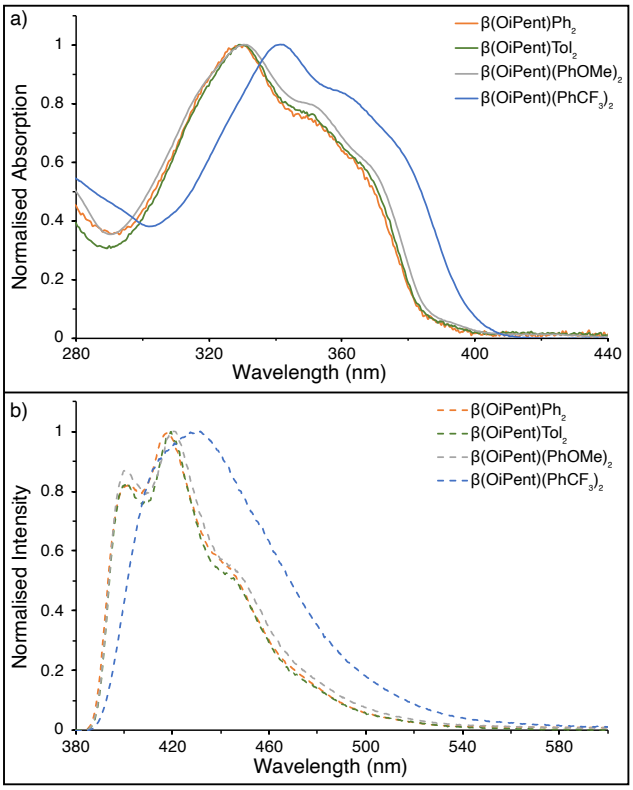


Fig. 6 Photophysical measurements for β(O*i*-Pent)R_2 Series. a) UV-visible spectra; b) Emission spectra (λ_{ex} = 365 nm). All samples were prepared under inert atmosphere with degassed dry THF.

have vibronic structure in both the emission and excitation spectra, which are similar to their α-NDT series counterparts (with larger $E_{g,opt}$ > 3.10 eV). However, **β(Oi-Pent)$(PhCF_3)_2$** is distinct in this analysis. The UV-visible spectrum is bathochromically shifted ($\lambda_{abs,max}$ = 11 nm) in comparison to the rest of the **β(Oi-Pent)R_2** series with similar vibronic structure, but the emission spectrum is broadened and ill-defined.

This is in stark contrast to the rest of the **β(Oi-Pent)R_2** series (and the **α(OHex)R_2** series), where the emission spectra have detailed structure (in THF). A series of variable polarity solvents was used to elucidate the extent, and type, of solvatochromism. The absorption spectra of **β(Oi-Pent)$(PhCF_3)_2$** display a weak hypsochromic shift ($\Delta\lambda_{abs}$= 4 nm), with a larger (albeit still weak) bathochromic shift ($\Delta\lambda_{em}$ = 9 nm) observed in the emission spectra (Figure 7). This solvatochromism is well-correlated with the solvent polarity parameter, $E_T(30)$,[83] with r^2 values of 0.98 and 0.87 for the absorption and emission spectra respectively. It should be noted that $E_T(30)$ values for binary solvent mixtures such as THF·MeCN, used here as the maximum polarity solvent, are quite complex[84] and the figure used here (based on mole fractions) is an approximation. However, it is apparent that this solvatochromism trend is due to the stabilization of the excited state in polar solvent, which in turn yields a larger Stokes shift as solvent polarity increases (Table 3). The A–D–A alignment of **β(Oi-Pent)$(PhCF_3)_2$** in tandem with the modest Stokes shift (≈ 10-20 nm) observed imply that the excited state geometry is quite rigid due to the push-pull of the intramolecular charge transfer.[85] Photoluminescence quantum yields for each solvent are given in Table 3 for **β(Oi-Pent)$(PhCF_3)_2$**, all of which are within the error of the technique (±10%).

Table 2 β(O*i*-Pent)R_2 Series spectral data[a]

β(Oi-Pent)R_2	ε M^{-1} cm^{-1}	$\lambda_{abs,max}$ nm	$E_{g,opt}$[b] eV	λ_{em}[c] nm	Φ_{PL}[c]
β(Oi-Pent)Ph_2	4.98×10^4	331	3.22	418	0.16
β(Oi-Pent)Tol_2	4.31×10^4	330	3.22	420	0.16
β(Oi-Pent)$(PhOMe)_2$	6.25×10^4	331	3.21	421	0.20
β(Oi-Pent)$(PhCF_3)_2$	4.40×10^4	342	3.10	432	0.23

[a] UV-visible and emission spectra were measured as solutions in dry, degassed THF under inert atmosphere
[b] $E_{g,opt}$ estimated from the tangent of the absorption onset, $\lambda_{abs,onset}$
[c] λ_{ex} = 365 nm

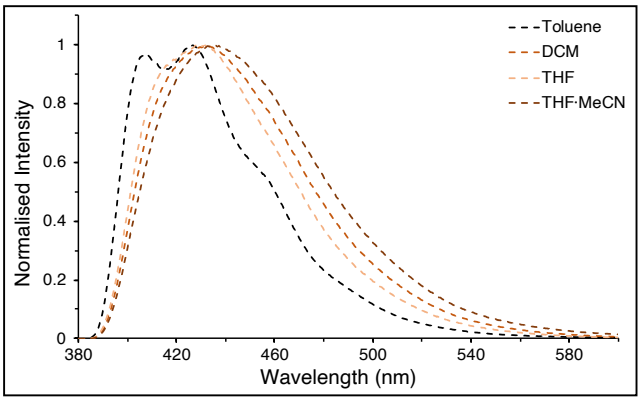


Fig. 7 Fluorosolvatochromism observed for β(O*i*-Pent)$(PhCF_3)_2$ Series.

Table 3 Solvatochromism analysis of β(O*i*-Pent)$(PhCF_3)_2$[a]

Solvent	$E_T(30)$ kcal mol^{-1}	ε M^{-1} cm^{-1}	$\lambda_{abs,max}$ nm	λ_{em}[b] nm	Φ_{PL}[b]
Tol	33.9	4.37×10^4	343	426	0.18
DCM	40.7	4.68×10^4	340	432	0.17
THF	37.4	4.35×10^4	341	432	0.23
THF·MeCN	42.4	4.93×10^4	339	435	0.15

[a] UV-visible and emission spectra were measured as solutions in dry, degassed solvents under inert atmosphere
[b] λ_{ex} = 365 nm

3 Conclusions

In summary, two isomeric sets of bent naphthodithiophenes have been synthesized and characterized as photoactive small molecules. In both the **α(OR')R_2** and **β(Oi-Pent)R_2** series structural authentication in the solid-state show that the end-caps do not impact inter-ring bond lengths, yet solution-state measurements made via NMR spectroscopy show minor perturbations of interior resonances far removed from the **R**-groups. In the **α(OHex)R_2** series spectrophotometry measurements detail fine vibronic structure across all derivatives with modest photoluminescence quantum yields (Φ_{PL} ≈ 0.24-0.40). Alternatively, the **β(Oi-Pent)R_2** series are far less luminescent due to the limited conjugation path brought on by the orientation of the thiophene

moieties. Attachment of electron donating groups has negligible impact on the properties of the fluorophores, but electron deficient groups have a distinct response to their A–D–A motif. **β(Oi-Pent)($PhCF_3$)$_2$** displays positive fluorosolvatochromism and becomes broadened as solvent polarity increases, indicative of increased intramolecular charge transfer character. The photophysical characterisation of these isomeric bent NDTs highlights how these simple fluorophores have unique applications as photoactive small molecules. Current efforts to maximize the effective conjugation path of isomeric bent NDTs in small molecule systems are underway.

Author Contributions

Z.J.K. conceptualized the project and acquired funding to carry out the experimental portion of the research. E.B.A.A, C.D.G., M.C., D.A., J.P., and Z.J.K. carried out the synthesis, characterization, and spectral analysis of compounds. V.C. and A.R. performed the computations to support the experimental findings and elucidate other parameters of interest. M.Z. carried out the crystallographic analysis. E.B.A.A. and Z.J.K wrote the original draft, with all authors participating in the review and editing process. All authors have given approval to the final version of the manuscript, their respective contributions, and have agreed to publish this work.

Conflicts of interest

There are no conflicts to declare.

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