

A Protein Cleavage Platform Based on Selective Formylation at Cysteine Residues

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Abstracts:

Site-selective cleavage of the peptide backbone in proteins is an important class of post-translational modification (PTM) in nature. However, the organic chemistry for such site-selective peptide bond cleavages has yet to be fully explored. Herein, we report cysteine *S*-formylation as a means of selective protein backbone cleavage. We developed *N*-formyl sulfonylanilide as a cysteine-selective formylation reagent for proteins. Upon *S*-formylation with the reagent, the amide bond adjacent to the *S*-formylated cysteine is cleaved by hydrolysis under neutral aqueous conditions. Formylation probes bearing a protein ligand enabled the affinity-based selective cleavage of the target proteins not only in the test tube, but also under biorelevant conditions such as in crude cell lysate and on the cell surface. These results demonstrate the high biocompatibility of this protein cleavage technology. A proof-of-concept study of the cleavage-induced protein activation further demonstrates its utility as a platform for the functional regulation of proteins by artificial PTM.

Posttranslational modifications (PTM) are used ubiquitously in nature to modulate protein structure and function. A representative class of PTM is the enzyme-catalyzed modification of amino acid side chains with various chemical groups such as phosphate, acyl, and sugar units¹. This type of PTM greatly expands the proteome diversity, which cannot be covered by genome encoding and gene transcription processes. Another important class of PTM is the covalent cleavage at specific peptide backbone sites in proteins¹⁻³. This controlled proteolysis is widely used in protein maturation, which is necessary to regulate the activity and cellular localization of proteins. In addition to these naturally occurring PTMs, artificial chemical modification is another possible approach to manipulating protein structure and function. This type of artificial PTM allows us to selectively introduce artificial molecules into target proteins and thereby regulate their function. Recent advances in a new class of chemical reactions for chemoselective protein modifications,⁴⁻⁹ which can be used for localization imaging, activity switching, and drug-transportation of proteins, demonstrate the importance of this approach. In contrast to these extensive studies, little progress has been made for decades in the development of artificial PTMs that allow site-selective cleavage of the peptide backbone in proteins.

37 To date, chemical reactions available for the cleavage of native proteins remain scarce. The conventional
38 approach is based on the cyanation of cysteine residues using reagents such as 2-nitro-5-thiocyanatobenzoic
39 acid (NTCB)^{10,11} and 1-cyano-4-dimethylaminopyridinium tetrafluoroborate (CDAP)¹², which induce the amide
40 bond cleavage adjacent to the cyanated cysteine (Figure 1a). However, these reagents require harsh pH
41 conditions (pH 9.0 for NTCB and pH 3.0 for CDAP) and a high concentration of amine (~ 1000 mM) as an
42 additive to facilitate the peptide bond cleavage, making it difficult to avoid protein denaturation. The intrinsic
43 high thiol reactivities of these reagents (reaction half-life with glutathione $t_{1/2} < 1$ min at 25 °C, Figure S1) also
44 preclude the site-selective cleavage of proteins. As a complement to the cyanation-based cleavage, Kanai *et al.*
45 have developed the stoichiometric copper-mediated oxidative cleavage of proteins at serine residues under
46 denaturing, acidic conditions (CH₃CN/H₂O/AcOH = 9:9:2) (Figure 1b).¹³ DeGrado has also developed a
47 sequence-selective nickel-assisted hydrolysis of amide bonds in proteins¹⁴ under conditions with high nickel
48 ion concentrations (~ 1 mM). However, due to their biologically incompatible reaction conditions, the
49 applications of these metal-catalyzed protein cleavages are limited to test tube experiments (Figure 1b). Very
50 recently, Davis *et al.* have reported a reductive, radical-mediated protein cleavage using a diboron reagent
51 (Figure 1c).¹⁵ Although this work marks great progress in chemical proteolysis research, this approach requires
52 the pre-formation of dehydroalanine before the site-selective cleavage. Other reported methods for chemical
53 protein cleavage require harsh conditions such as inducible protein denaturation⁶⁻¹⁸ and/or the incorporation of a
54 noncanonical amino acid to invoke the site-selective reaction.^{9,20} Furthermore, to our knowledge, there are no
55 reports of ligand-directed site-selective cleavage of proteins using a reactive affinity probe. Given the limited
56 progress in this research area, we aimed to develop a new chemical PTM with high biocompatibility which
57 allows the site-selective and single-step cleavage of protein backbones in a biological setting.

58 Here, we introduce cysteine *S*-formylation as a new artificial PTM that induces the peptide bond cleavage of
59 proteins (Figure 1d). We found that *N*-formyl sulfonylanilide serves as a cysteine-selective formylation reagent.
60 *S*-formylation by this reagent induced hydrolysis of the peptide bond adjacent to the formylated cysteine to
61 afford the cleaved protein under neutral aqueous conditions. We successfully performed the site- and target-
62 selective cleavage of proteins using liganded *S*-formylation probes not only in a buffer solution, but also in
63 crude cell lysate and on the cell surface. We also applied this chemical proteolysis to activate protein ligand
64 binding, demonstrating the utility of this cleavage platform for the functional regulation of proteins.

65

66 **Peptide cleavage by cysteine formylation**

67 We initially designed acylating agents with a sulfonylanilide scaffold (Figure 2a) based on the molecular
68 structures of *N*-acylation reagents.²¹⁻²³ and evaluated their peptide cleavage activities towards the cysteine-
69 containing peptide-**a** by fluorescence analysis (Figure 2b). This peptide was designed to recover the
70 fluorescence of 7-hydroxycoumarin upon cleavage and disassociation of the 2,4-dinitrophenyl quencher group.
71 In the time-dependent fluorescence analysis, formylation reagent **1** induces the fluorescence recovery of the
72 peptide over 12 hr under neutral aqueous conditions (100 mM phosphate buffer containing 30% CH₃CN, pH

73 7.4, 37 °C, Figure 2c and 2d). LC-MS analysis revealed rapid formation of a new peak at 20 min, which
74 corresponds to the *S*-formylated peptide-**a** ($[M+H]^+ = 972.33$) (Figure 2e). Substitution of the sulfhydryl group
75 with *N*-ethylmaleimide completely blocked the cleavage reaction, suggesting that the formylation occurs at the
76 cysteine residue (Figure S2). *S*-formylation of peptide-**a** was followed by the gradual formation of the two peaks
77 at 31 and 40 min, which were assigned as the hydrolyzed N-fragment ($[M+H]^+ = 491.18$) and the formylated
78 C-fragment ($[M+H]^+ = 500.16$) peptides, respectively. The yield of the cleavage reaction was estimated to be
79 30% at 12 h based on the quantified peak intensities of the N- and C-fragments. On the other hand, when using
80 the acetylation and butyrylation reagents **2** and **3**, the cleavage reaction proceeded with the much lower rates
81 than with **1**. This finding prompted us to further investigate the chemical properties of the *S*-formylation
82 induced-peptide bond cleavage. The aqueous stability of the *S*-acylated cysteines was evaluated by ¹H-NMR
83 under neutral aqueous conditions (100 mM phosphate buffer containing 25% CD₃CN, pD 7.4, Figure 2f, S3 and
84 S4). The data showed that the *S*-formylated cysteine **4** was gradually hydrolyzed with a half-life ($t_{1/2}$) of 6.2 hr
85 to provide the free-thiol species, indicating instability of the *S*-formyl species in neutral aqueous conditions. In
86 contrast, the corresponding *S*-acetylated cysteine **5** was stable under the same conditions ($t_{1/2} > 24$ h). Combining
87 these data analyses, we propose the reaction mechanism for the *S*-formylation-induced peptide bond cleavage
88 shown in Figure 2g. In the step of the amide bond cleavage, we speculate the formation of the *N*-formyl imide
89 as the reaction intermediate that undergoes the peptide bond cleavage, although we did not identify this species
90 in the LC-MS analysis.

91 We next evaluated the sequence dependency of the peptide bond cleavage induced by *S*-formylation using
92 reagent **1**. The initial reaction rates ($\Delta F/\text{min}$) of the series of peptides are summarized in Figure 3a. The results
93 showed that the peptide-**b** and -**c**, possessing Asp-Cys and Asn-Cys sequences, respectively, displayed much
94 higher reaction rates than peptide-**a** which has the Gly-Cys sequence (Figure 3b). The cleavage reactions of
95 peptide-**b** and -**c** proceeded almost quantitatively after 3 hr based on LC-MS analyses, which suggest the
96 formation of the corresponding N- and C-fragments from peptide-**b** and -**c** (Figure 3c, S5). In the case of peptide-
97 **c** (Xaa = Asn), we identified the succinimide derivative as the main product of the *N*-fragment (Figure S5),
98 suggesting that the intramolecular attack of the asparagine residue to the amide carbonyl group facilitates the
99 cleavage reaction. In the case of peptide-**b** (Xaa = Asp), the formation of the succinate was not identified by
100 LC-MS, probably due to its rapid hydrolysis. Other tested peptides with different amino acids adjacent to the
101 cysteine did not give significant changes in the reaction rate compared to peptide-**a** (Figure 3a and S6).

102 103 **Optimization of electrophilic reactivity of the formylation reagents**

104 We next optimized the electrophilic reactivity of the formylation reagent that allows selective formylation of
105 a target cysteine residue. Formylation reagent **1** was highly reactive toward a cysteine derivative such as
106 glutathione (GSH) ($t_{1/2} < 0.1$ hr, Table 1) and unstable by hydrolysis under the neutral aqueous conditions (pH
107 7.4, 37 °C) ($t_{1/2} = 1.0$ hr). We found that the high reactivity of **1** could be tuned by the introduction of substituents
108 on the sulfonylanilide moiety. Among the derivatives (Table 1) synthesized, the formylation reagents **9** and **10**

109 exhibited moderate reactivities toward GSH ($t_{1/2}$ = 0.15 and 1.47 hr, respectively) while maintaining sufficient
110 aqueous stability under neutral aqueous conditions ($t_{1/2}$ = 6.6 and 37.2 hr, respectively). The reactivities of **9**
111 and **10** toward other amino acids such as *N*-acetyl lysine, *N*-acetyl tyrosine, and *N*-acetyl histidine, and glycine
112 amide were further evaluated (Table 1, Table S1). The data revealed that **9** and **10** did not show noticeable
113 reactivity toward excess amounts of these amino acids except for *N*-acetyl histidine, which slightly shortened
114 the half-life of the probes. Based on these results, we designed formylation reagents **12** and **13** as the more water-
115 soluble analogs of **9** and **10**, respectively (Figure 4a). The selectivity of these formylation reagents was evaluated
116 using a small ubiquitin protein (8.5 kDa), which possesses seven lysines, one tyrosine and one histidine but
117 lacks cysteine on the surface. The MALDI-TOF mass analysis revealed that the treatment of ubiquitin (2.5 μ M)
118 with a high concentration of **13** (1 mM) resulted in scarcely any formylation even after 6 hr, while **12** provided
119 a mono-formylated ubiquitin after 6 hr (Figure S7). These data suggested the high cysteine selectivity of
120 formylation reagents **12** and **13** on the protein surface. In the peptide cleavage experiment, the reactivities of **12**
121 and **13** were not significantly decreased compared to those of probe **1** (Figure S8).

122 The rather slow cleavage rate of peptide-**a** using the formylation reagents prompted us to find additives that
123 could accelerate the reaction. Based on the proposed reaction mechanism (Figure 2g), we expected that the
124 addition of nucleophilic species would assist the amide bond cleavage. We found that a variety of thiol species
125 effectively accelerated the cleavage reaction of peptide-**a** by formylation reagent **12** (Figure 4b), whereas amine
126 derivatives such as glycine amide and propargyl amine did not show noticeable effects on the reaction rate
127 (Figure S9). Among the thiol species, glutathione (GSH) was most effective for reaction rate acceleration. The
128 yield of the cleavage reaction was evaluated to be 77% after 3 hr based on fluorescence intensity analysis. Time-
129 dependent LC-MS analysis revealed the formation of the alkynylated N-fragment (34 min) as a main product
130 in the presence of the alkynylated cysteine (Cys-alkyne) (Figure 4c). Based on this finding, we propose a
131 reaction mechanism of the thiol-assisted amide bond cleavage, which involves a nucleophilic thioester
132 formation with Cys-alkyne and a subsequent intramolecular S to N acyl transfer reaction to give stable
133 alkynylated N-fragment (Figure 4d).

134

135 **Protein cleavage induced by cysteine formylation**

136 We next applied the cysteine formylation peptide bond cleavage to proteins. As an initial attempt, we
137 employed maltose binding proteins (MBPs) possessing a single cysteine in the Xaa-Cys-Asp8 peptide tag (Xaa
138 = Asp, Asn, and Gly) at their C-termini (Figure 5a). When MBPs were treated with the formylation reagents **1**,
139 **12**, and **13** (1 or 5 mM, 2 hr) in the presence of GSH under neutral aqueous conditions (100 mM phosphate
140 buffer, pH 7.4, 37 °C), the proteins truncated at the tag site were observed by SDS-PAGE analysis (Figure 5b).
141 In the absence of GSH, the cleavage reaction proceeded less efficiently with MBP-GC compared to MBP-NC
142 and MBP-DC, as we observed in the peptide cleavage experiments (Figure S10). We next attempted the
143 cleavage of point mutated MBPs possessing a single cysteine (T32C, A207C and A269C) at the internal loop
144 regions. SDS-PAGE analysis showed the partial cleavage of the MBPs at the cysteine site upon treatment with

145 the formylation probes **1**, **12** and **13** (5 mM) in 30% acetonitrile-phosphate buffer solution (pH 7.4) (Figure 5c).
146 The cleavage efficiency of the reagents was found to be **1** > **12** > **13**, which is consistent with the order of their
147 reactivities to cysteine. Treatment of the triple cysteine mutant MBP (T32C/A207C/A269C) with the
148 formylation reagents resulted in the complete disappearance of the parent MBP and provided multiple fragments
149 of MBP. It should be noted that these formylation reagents exhibited higher cleavage efficiencies than the
150 conventional cyanation reagents CDAP and NTCB.¹⁰⁻¹² Interestingly, by harnessing the thiol-assisted cleavage
151 mechanism as shown in Figure 4d, we were able to introduce fluorescent rhodamine or biotin onto the C-
152 terminus of MBP-GC upon treatment with the corresponding cysteine derivatives (*i.e.*, Cys-TAMRA and Cys-
153 biotin) (Figure 5d). The labeling of MBP with Cys-TAMRA and Cys-biotin was confirmed by in-gel
154 fluorescence and western blot analyses, respectively (Figure 5e).

155

156 **Affinity-based cleavage of proteins by cysteine formylation**

157 To perform site-selective cleavage of proteins by cysteine *S*-formylation, we employed the *N*-terminal
158 domain of human double minute 2 (HDM2)²³ bearing a GST (glutathione *S*-transferase) (GST-HDM2) (Figure
159 6a, Figure S11). For the ligand-directed affinity-based cleavage of GST-HDM2, we designed formylation probe
160 **14** possessing a Nutlin-3 (Figure 6b), which is known as a ligand for HDM2. In GST-HDM2, Cys2 is closely
161 positioned at the binding site of Nutlin-3 and was selected as the *S*-formylation target cleavage site. When GST-
162 HDM2 (2 μM) was incubated with probe **14** (100 μM) under neutral aqueous conditions (100 mM phosphate
163 buffer, pH 7.4, 37 °C), the *N*-terminal GST domain gradually increased in the CBB stain (Figure 6c and 6d).
164 The yield of the cleavage reaction was estimated to be 79% after 6 hr based on western blot analysis using anti-
165 His tag antibody. MALDI-TOF mass analysis showed the appearance of a new peak corresponding to the *N*-
166 terminal GST fragment (Figure S12), suggesting the site-selective cleavage of GST-HDM2 at Cys2. The
167 cleavage reaction by probe **14** was effectively suppressed in the presence of Nutlin-3 (500 μM). Incubation of
168 GST-HDM2 with high concentrations of formylation reagent **13** (1 mM) induced the cleavage reaction only
169 slightly. These results exhibit the ligand-directed cleavage of GST-HDM2 by *S*-formylation with **14**.

170 We next attempted the functional regulation of HDM2 by affinity-based cleavage. For this proof-of-concept
171 experiment, we designed p53-His10-HDM2, which possesses p53 peptide analogue (L26V)^{24,25} and a His10 tag
172 at the *N*-terminus of HDM2 (Figure 6e, Figure S13). Treatment of p53-His10-HDM2 (2 μM) with the binuclear
173 Ni(II)-NTA probe **15**-2Ni(II) (20 μM) (Figure 6f)²⁶ induced the time-dependent cleavage of the *N*-terminal
174 region of HDM2 in HEPES buffer (pH 7.4) (Figure 6g). The yield of the cleavage reaction after 3 hr was
175 estimated to be 65% based on the CBB stain band intensity. The cleavage reaction did not proceed when using
176 probe **15**, which lacks nickel ions essential for binding to His10 tag. A fluorescence anisotropy experiment
177 revealed that the binding of fluorescent HDM2 probe **17** (0.4 μM) to p53-His10-HDM2 (1.5 μM) was greatly
178 suppressed, mainly due to the intramolecular interaction between the p53 peptide (L26V) and the HDM2 Nutlin-
179 binding domain. We found that the fluorescence anisotropy value gradually increased from 0.06 to 0.13 upon
180 treatment with **15**-2Ni(II) for 3 hr (Figure 6h). This time course corresponds well with the progress of the

181 cleavage reaction (Figure S14a). The analysis based on the calibration curve (Figure S14b, c) suggested that the
182 binding activity of HDM2 towards **17** was recovered from 13% to 67% of that of intact HDM2. The anisotropy
183 value did not change upon treatment of HDM2 with **16-2Ni(II)**, which lacks a reactive formyl group. These data
184 demonstrate the utility of the formylation-induced cleavage for activation of proteins.

185 We further applied the affinity-based cleavage to the kinase domain of fibroblast growth factor receptor 1
186 (FGFR1, Met399-Arg822) (Figure 7a, Figure S15). For this purpose, we designed formylation probe **18** based
187 on the structure of futibatinib²⁷, which has been developed as a covalent inhibitor of FGFR 1–4 (Figure 7b). In
188 the probe design, the acrylamide of futibatinib was replaced with N-formyl sulfonylanilide to target Cys488 for
189 formylation. When the recombinant kinase domain of FGFR1 fused with His6 tag (2 μ M) was treated with **18**
190 (50 μ M) for 3 hr, the intensity of the FGFR1 band greatly decreased in the SDS-PAGE analysis (Figure 7c) in
191 a concentration-dependent manner (10-50 μ M of **18**) (Figure S16). This decrease in the band intensity was not
192 observed with 50 μ M of the non-liganded formylation probe **1**, while the high concentration of **1** (1 mM)
193 effectively reduced the band intensity of FGFR1 (50 kDa). The western blot analysis using His tag antibody
194 suggested that *ca.* 85% of FGFR1 was degraded by **18** (Figure 7d). These results demonstrate the application
195 of ligand-directed S-formylation cleavage to the kinase domain of FGFR1 with **18**.

196

197 **Affinity-based cleavage of proteins under biorelevant conditions**

198 We performed the formylation-induced protein cleavage in crude cell lysate to assess its biocompatibility.
199 In this experiment, we employed MBP tagged with Cys-His10 at its C-terminus (Figure S17a, b) and Ni(II)-
200 NTA probe **15-2Ni(II)**. We found that the cleavage reaction of the Cys-His10 tag from MBP proceeded upon
201 treatment of MBP-Cys-His10 (0.8 μ M) with **15-2Ni(II)** (20 μ M) in the crude lysate of A431 cells (Figure 8a
202 and 8b). The yield of the cleavage reaction reached 80% after 6 hr based on western blot analysis using anti-
203 His tag antibody. This yield was comparable to those of the cleavage reaction of MBP-Cys-His10 carried out
204 in phosphate buffer (pH 7.4) (Figure S17c, d). To confirm the protein selectivity of the cleavage reaction, the
205 lysate sample treated with **15-2Ni(II)** (20 μ M) was further treated with cysteine-TAMRA to fluorescently label
206 the cleaved proteins (Figure 8c). The in-gel fluorescence analysis showed a single fluorescent band
207 corresponding to the cleaved MBP, suggesting the high target selectivity of the formylation-induced protein
208 cleavage.

209 Finally, we attempted the peptide bond cleavage of a membrane protein expressed on the cell surface. In
210 this experiment, we designed bradykinin receptor type 2 (B2R) receptor conjugated to a Cys-His10 tag and a
211 SPOT tag on the extracellular N-terminus (Figure 8d, Figure S18). Transient expression of the B2R was
212 confirmed by fluorescence imaging using the anti-SPOT nanobody and B2R antagonist peptide, which were
213 conjugated with Alexa FluorTM 568 and Cy5 dye, respectively (Figure 8e). When the cells were treated with **15-2Ni(II)**
214 (40 μ M) for 3 hr, the fluorescence of the anti-SPOT nanobody decreased on the cell surface, while the
215 fluorescence of the B2R antagonist peptide remained nearly unchanged. The quantitative ratio analysis (anti-

216 SPOT nanobody / B2R antagonist) revealed that the fluorescence of anti-SPOT nanobody decreased by 45%
217 compared to the non-treated cells (Figure 8f). On the other hand, treatments of the cells with probe **15** (without
218 nickel ions) and probe **16-2Ni(II)** (without a formyl group) gave only small changes in the fluorescence of anti-
219 SPOT nanobody. To further confirm this result, we conducted western blot analysis and evaluated the band
220 intensities of B2R detected by anti-SPOT nanobody and anti-B2R antibody (Figure 8g). The data showed that
221 the band detected by the anti-SPOT nanobody becomes weaker upon treatment of the cells with **15-2Ni(II)**,
222 while treatments with **15** (without nickel ions) and **16-2Ni(II)** gave only small changes in the band intensity.
223 The quantitative ratio analysis (anti-SPOT nanobody / B2R antibody) disclosed a significant decrease in the
224 band intensity of anti-SPOT nanobody upon treatment with **15-2Ni(II)** (Figure 8h). All these results suggest the
225 cleavage of the SPOT tag at the extracellular N-terminus region from B2R by binding-induced cysteine *S*-
226 formylation. A cell viability assay revealed that **15-2Ni(II)** did not exert significant cytotoxicity, which also
227 suggests good biocompatibility of cysteine *S*-formylation for targeting membrane surface proteins (Figure S19).

228

229 Discussion

230 Formylation of the lysine residue has been known as a protein PTM since it was first discovered in 2007,
231 although its biological function has yet to be elucidated^{28,29}. On the other hand, to our knowledge, there are no
232 reports of cysteine formylation of cellular proteins. This is probably due to the difficulty of detecting the easily
233 hydrolyzed *S*-formylated species and the apparent absence of an enzyme that catalyzes the formylation of
234 proteinaceous cysteines³⁰. For these reasons, cysteine formylation has been overlooked as a PTM of proteins.
235 In this manuscript, we show for the first time that cysteine *S*-formylation induces peptide bond cleavage by
236 hydrolysis. Unlike cysteine cyanation using NTCB and CDAP reagents^{10,12}, the formylation-induced protein
237 cleavage efficiently proceeds under mild neutral conditions. Our study revealed that cysteine *S*-acetylation and
238 *S*-butylation are much less effective for peptide bond cleavage compared to *S*-formylation (Figure 2a, 2d).
239 Degani has reported preliminary research on peptide cleavage by *p*-nitrobenzoylation of cysteine, though this
240 reaction requires harsh conditions to yield a cleaved peptide (pH 8.1, 50 °C)³¹. These data suggest the
241 exceptional ability of cysteine formylation to induce peptide bond cleavage. Considering these results, we
242 suggest that, for effective peptide bond cleavage, the cysteine residue must be modified with a small and strong
243 electron withdrawing group such as formyl and cyano groups.

244 We employed *N*-formyl sulfonylanilides as the formylation reagents for cysteine. While *N*-formylsaccharin
245 has been widely used as a formylation reagent for amines³², little is known about the utility of less reactive *N*-
246 formyl sulfonylanilide for this purpose. In this study, we revealed that *N*-formyl sulfonylanilides with tuned
247 electrophilic reactivities can work as cysteine-selective formylation reagents under neutral aqueous conditions.
248 Our affinity-based formylation probes bearing ligand units were able to selectively formylate a cysteine residue
249 on the target proteins and induced site-selective cleavage of the peptide bond. To our knowledge, this is the first
250 example of ligand-directed protein cleavage by a chemical reaction. Along with this ligand-directed selectivity,
251 the reversible hydrolytic nature of *S*-formylated cysteine also could contribute to reducing off-target formylation

252 and undesired protein cleavage. We believe that these properties contribute to the target-selective cleavage of
253 the target protein in crude cell lysate containing a multitude of proteins (Figure 8 a-c).

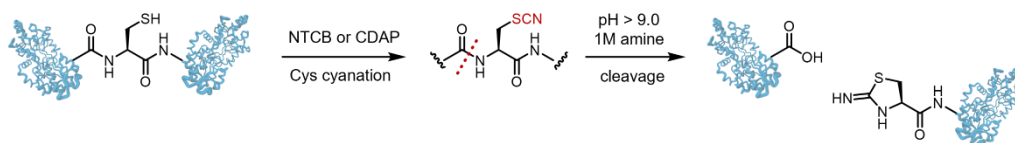
254 We are currently using this cleavage method for proteins possessing a free cysteine in a highly accessible
255 and conformationally flexible region. We anticipate that further improvement of the formylation probes (for
256 example, higher thiol selectivity and smaller molecular size) will increase the number of proteins that are
257 compatible with this cleavage approach. Global screening of cleavable cysteine sites in cellular proteins will
258 accelerate this advance. Building upon these future studies, we envision that this cleavage platform will facilitate
259 research progress in a variety of ways; targeting disease-related proteins for degradation³³ and mass
260 spectrometry-based shotgun proteomics analysis of cellular proteins³⁴ are just a few examples of a whole range
261 of possibilities.

262 **Reference**

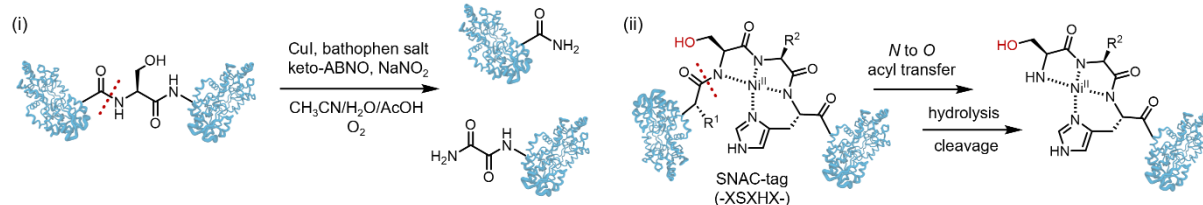
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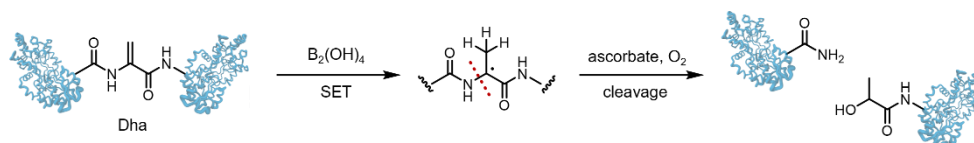
(a) Cysteine S-cyanation amide bond cleavage



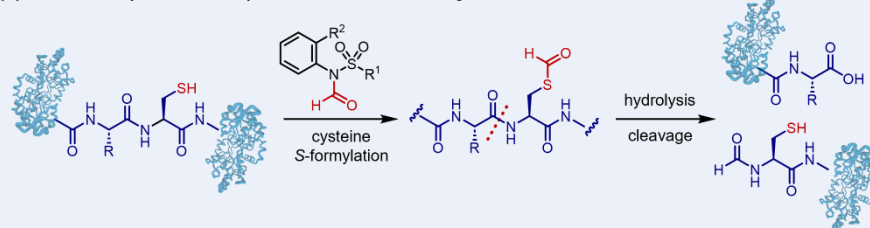
(b) Metal-catalyzed amide bond cleavage



(c) Reductive diborane dehydroalanine (Dha) cleavage



(d) **This work:** Cysteine S-formylation amide bond cleavage

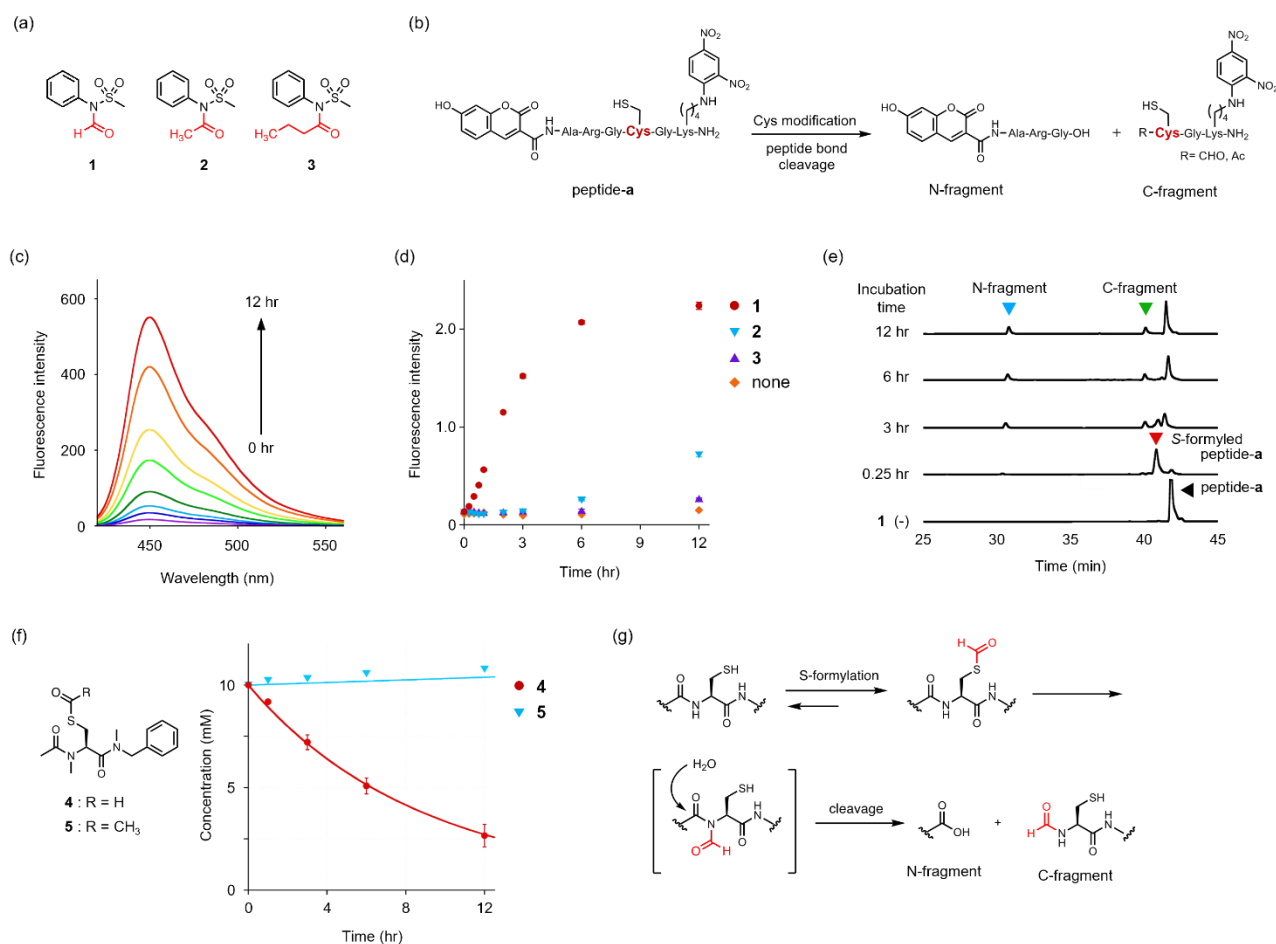


- operable under mild conditions (pH 7.4, 37 °C)
- tunable reactivity of S-formylation reagents
- applicable to affinity-based cleavage
- high biocompatibility in cell lysate and on live cell surface

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333 **Figure 1. Chemical cleavage of peptide backbone in proteins.**

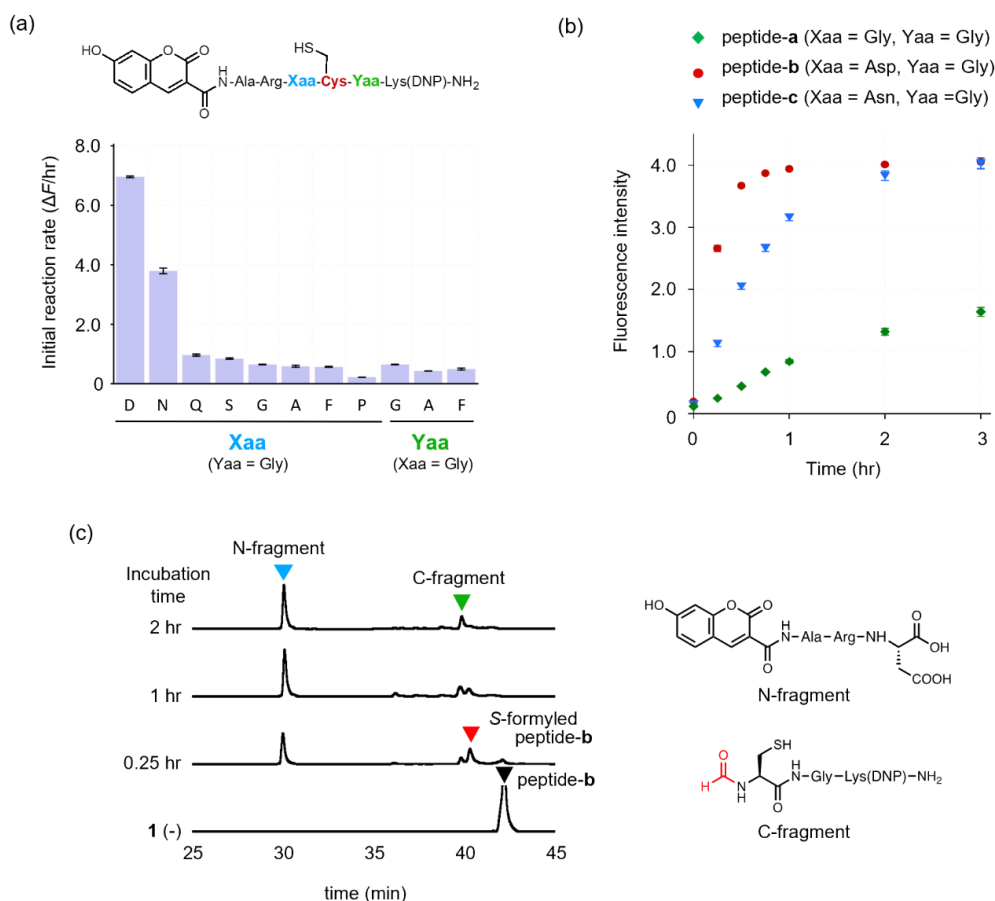
334 **a.** Protein cleavage reaction through S-Cyanation, under harsh conditions such as pH 9.0 or a large excess of amine
335 (1 M). **b.** Metal-catalyzed amide bond cleavage, i) Aerobic cleavage with Cu/keto-ABNO, ii) Ni(II)-complex
336 facilitated sequence-selective cleavage of SNAC-tag (-XSXHXs-). **c.** Reductively initiated cleavage via
337 dehydroalanine (Dha) formation. **d. This work:** amide bond cleavage induced cysteine formylation under mild
338 conditions (pH7.4), that could be useful chemical biology tools.



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340 **Figure 2. Peptide bond cleavage induced by cysteine S-formylation.**

341 **a**, Structures of the sulfonamide-based acylation reagents. **b**, Peptide bond cleavage of peptide-**a** through cysteine
 342 modification. **c**, Fluorescence spectral change of peptide-**a** (15 μM) upon addition of formylation reagent **1** (1 mM).
 343 Conditions: 100 mM sodium phosphate buffer (pH 7.4), 30% MeCN, 37 °C, λ_{ex} = 400 nm. **d**, Time trace plot of the
 344 fluorescence change of peptide-**a** (15 μM) upon treatment with acylation reagent **1-3** (1 mM). Conditions: 100 mM
 345 sodium phosphate buffer (pH 7.4), 30% MeCN, 37 °C, λ_{ex}/λ_{em} = 400/450 nm, mean ± s.d.: three independent
 346 experiments. **e**, HPLC analysis of the cleavage reaction of peptide-**a** thorough S-formylation. Peptide-**a** (30 μM) was
 347 treated with **1** (3 mM) in 100 mM sodium phosphate buffer (pH 7.4) containing 10% DMF in the presence of 0.3
 348 mM TCEP at 37 °C. The peaks were detected by UV absorbance at 370 nm. The peptides were identified by ESI-
 349 TOF-MS analysis. **f**, Time plot of the hydrolysis of S-formylated cysteine **4** and S-acetyled cysteine **5** (1 mM)
 350 analyzed by ¹H-NMR. Conditions: 100 mM sodium deuterium phosphate buffer, 25% CD₃CN, pD 7.4, 37 °C,
 351 mean ± s.d.: three independent experiments. **g**, Proposed reaction mechanism of the peptide bond cleavage induced
 352 by cysteine S-formylation.

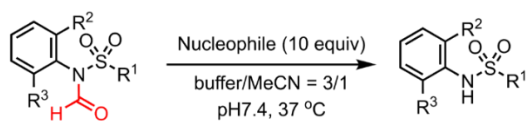


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Figure 3. Sequence dependency of the peptide cleavage reaction induced by cysteine *S*-formylation.

a, Initial reaction rates of the peptide cleavage reaction ($\Delta F/\text{min}$) upon treatment with **1**. Peptide (15 μM) was treated with **1** (1 mM) in 100 mM sodium phosphate buffer (pH 7.4) containing 30% MeCN at 37 °C. $\lambda_{\text{ex}}/\lambda_{\text{em}} = 400/450$ nm. mean \pm s.d.; three independent experiments. **b**, Time trace plot of cleavage reactions of peptide-**a-c** (Xaa-Cys-Gly, Xaa = Gly, Asp and Asn) by *S*-formylation with **1**. Peptide (15 μM) was treated with **1** (1 mM) in 100 mM sodium phosphate buffer (pH 7.4) containing 30% MeCN at 37 °C. $\lambda_{\text{ex}}/\lambda_{\text{em}} = 400/450$ nm, mean \pm s.d.: three independent experiments. **c**, HPLC analysis of the cleavage reaction of peptide-**b** by *S*-formylation. Peptide-**b** (30 μM) was treated with **1** (3 mM) in 100 mM sodium phosphate (pH 7.4) containing 10% DMF in the presence of 0.3 mM TCEP at 37 °C. The peaks were detected by UV absorbance at 370 nm. The peptides were identified by ESI-TOF-MS analysis.

364 **Table 1.** Summary of the aqueous stability and electrophilic reactivity of N-formyl sulfonylanilides.^a

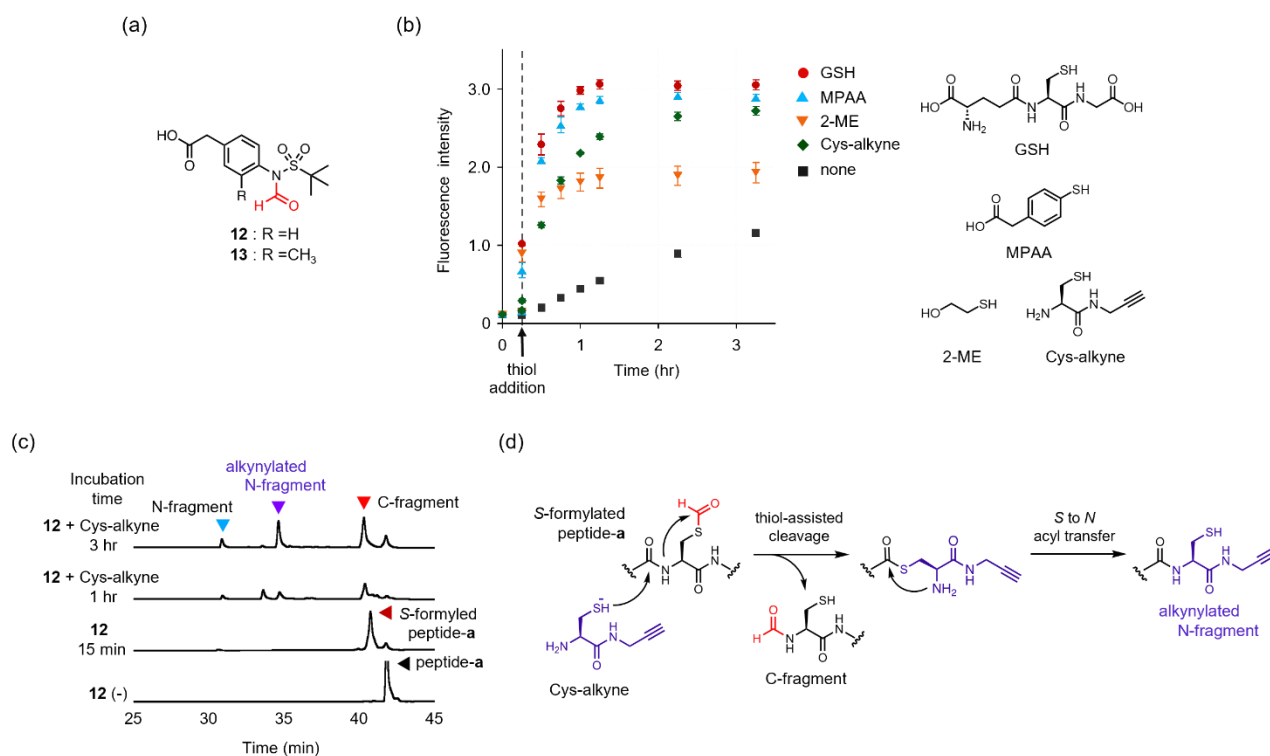


compound	Substituent			Nucleophile: $t_{1/2}$ (hr)		
	R ¹	R ²	R ³	GSH	NAc-Lys	none
1	Me	H	H	< 0.10	0.99	1.02
6	Me	Me	H	0.31	1.70	1.68
7	Me	Me	Me	2.98	16.3	16.1
8	ⁱ Pr	H	H	0.16	3.89	4.01
9	^t Bu	H	H	0.15	6.15	6.63
10	^t Bu	Me	H	1.47	37.3	37.2
11	^t Bu	Me	Me	20.9	> 48	> 48

^aData represent the mean \pm standard deviation of repeated experiments ($n = 3$).

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369 **Figure 4. Thiol-assisted cleavage of cysteine *S*-formylated peptide.**

370 **a**, Structures of the water-soluble *S*-formylation reagents **12** and **13**. **b**, Fluorescence change of peptide-a (15 μM)

371 upon treatment with **12** (1 mM) in the absence and presence of various thiol species (1 mM). Conditions: 100 mM

372 sodium phosphate (pH 7.4), 30% CH₃CN, 37 °C. mean ± s.d.: three independent experiments. **c**, HPLC analysis of

373 the cleavage reaction of peptide-a (30 μM) upon treatment with **12** (3 mM) in the presence of Cys-alkyne (3 mM).

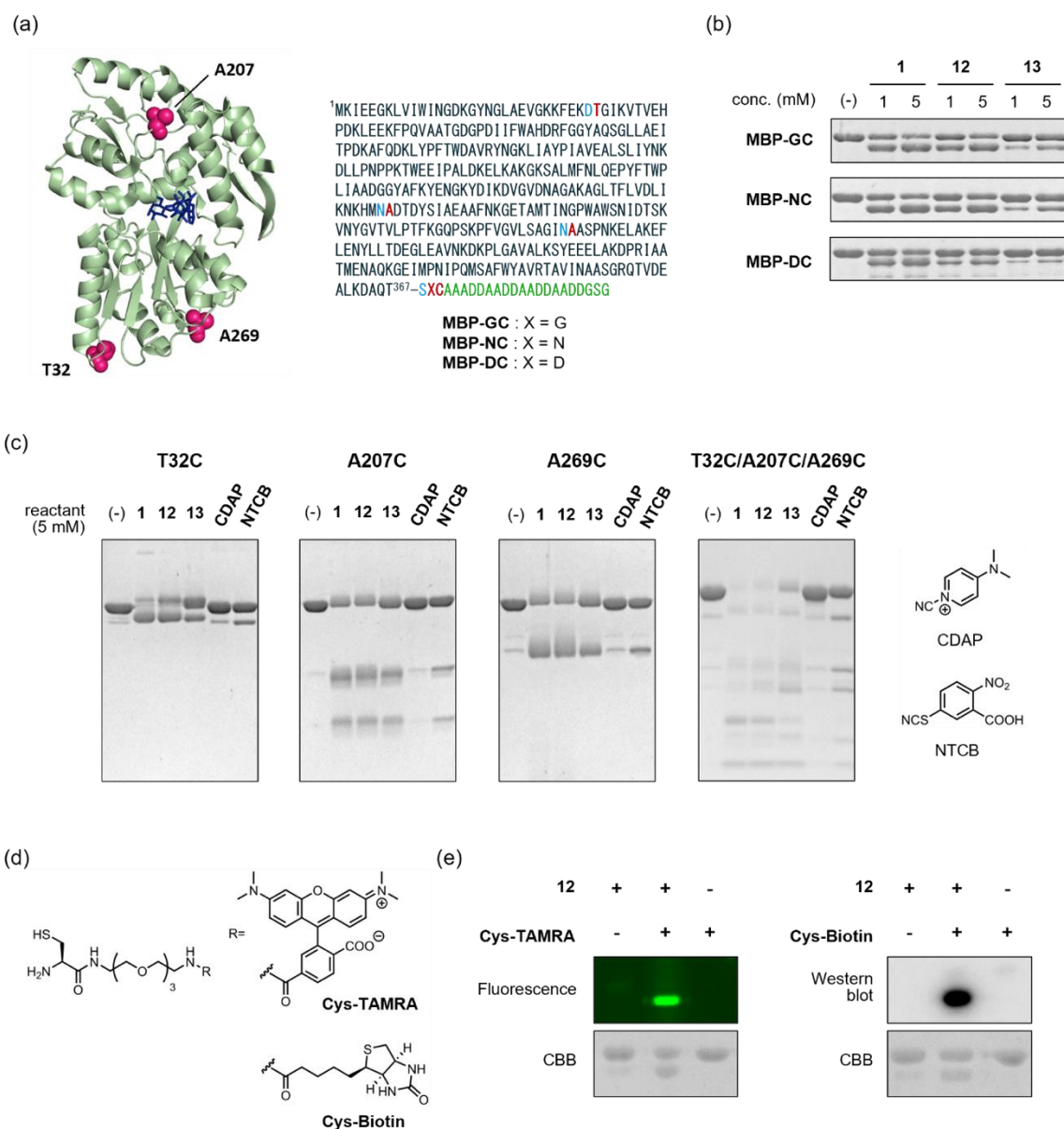
374 Conditions: 100 mM sodium phosphate (pH 7.4), 10% DMF, 3 mM TCEP, 37 °C. The peaks were detected by UV

375 absorbance at 370 nm. **d**, Proposed reaction mechanism of the thiol-assisted cleavage of *S*-formyl peptide and

376 subsequent *S* to *N* acyl transfer.

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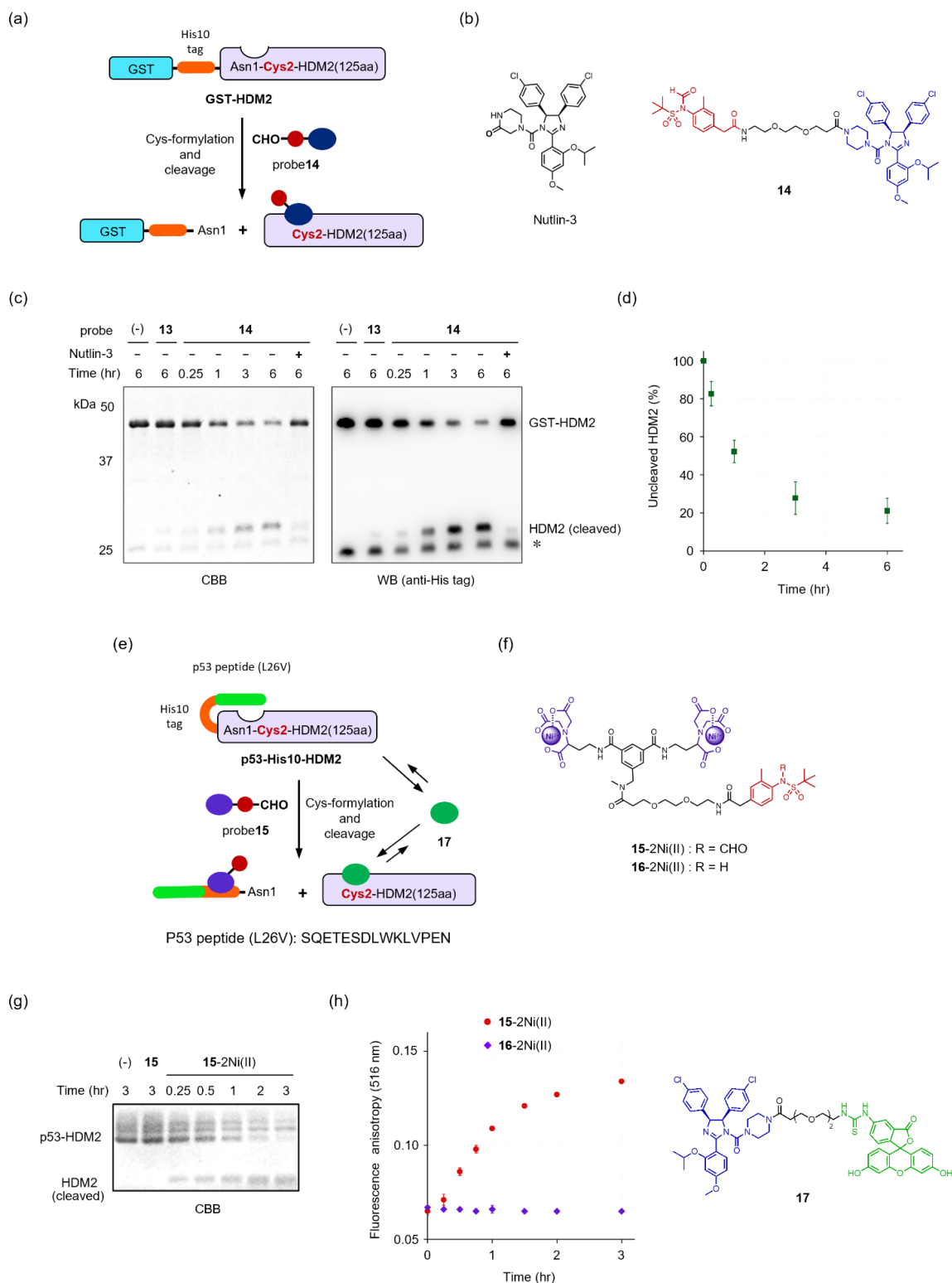
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380 **Figure 5. Protein cleavage induced by cysteine S-formylation.**

381 **a**, Crystal structure and amino acid sequence of maltose binding protein (PDB:1ANF, cysteine mutation sites are
382 shown in magenta). The peptide tag containing one cysteine residue (shown in green) is fused to the C-terminus for
383 the cleavage experiments shown in Figure 5b. **b**, Cleavage of tag-fused MBP using formylation reagents **1**, **12**, and
384 **13**. MBP (4 μ M) was first treated with the probe (1 or 5 mM) for 1 hr and then treated with GSH (5 mM) for 1 hr in
385 100 mM sodium phosphate buffer (pH 7.4) containing 40 μ M TCEP at 37 $^{\circ}$ C. **c**, Cleavage of the point mutated MBPs
386 (4 μ M) using the formylation reagents (**1**, **12**, or **13**), CDAP, or NTCB (5 mM, 1 hr) in the presence of GSH (5 mM,
387 1 hr) in 100 mM sodium phosphate containing 30% CH₃CN (pH 7.4, 37 $^{\circ}$ C). **d**, Structures of Cys-TAMRA and Cys-
388 Biotin. **e**, Cleavage of MBP-GC (30 μ M) using **12** (10 mM) and subsequent chemical labeling with Cys-TAMRA
389 and Cys-biotin (1 mM, pH 7.4, 37 $^{\circ}$ C). left panel: in-gel fluorescence analysis, right panel: western blot analysis
390 using streptavidin-horseradish peroxidase (HRP).



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Figure 6. Affinity-based cleavage and functional regulation of HDM2.

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a, Schematic illustration of the affinity-based cleavage of recombinant GST-HDM2 using formylation probe **14**.

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b, Structure of Nutlin-3 and formylation probe **14**.

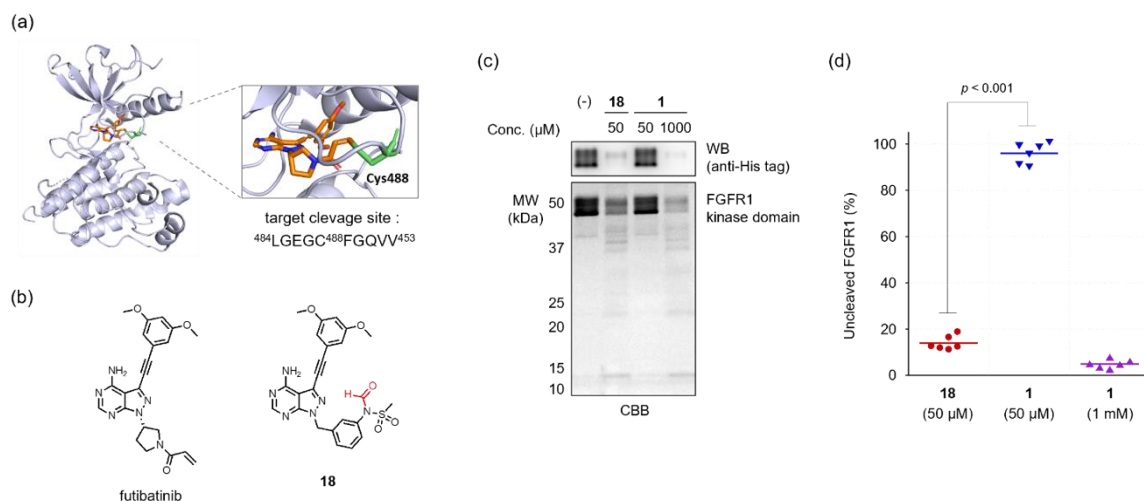
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c, Cleavage of GST-HDM2 (2 μM) using **13** (100 μM) and **14** (100 μM) in 100 mM phosphate buffer (pH 7.4) at 37 °C. Left panel: CBB staining, Right panel: western blot analysis

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using anti-His tag antibody. **d**, Time trace plot of GST-HDM2 cleavage using **14** by western blot analysis (mean ± s.d.:

397 three independent experiments). **e**, Schematic illustration of functional regulation of p53-His10-HDM2 through
398 peptide tag cleavage using probe **15**-2Ni(II). **f**, Structure of Ni(II)-NTA probe **15** targeting the oligo-histidine tag and
399 the negative control probe **16** (without a formyl group). **g**, Cleavage of p53-His10-HDM2 (2 μ M) using **15**-2Ni(II)
400 (20 μ M) or **15** (without NiCl₂) in 100 mM phosphate buffer (pH 7.4) at 37 °C. **h**, Time trace plot of the fluorescence
401 anisotropy values of probe **17** upon treatment of p53-His10-HDM2 with **15**-2Ni(II) or **16**-2Ni(II). (mean \pm s.d.: three
402 independent experiments). p53-His10-HDM2 (1.5 μ M) was treated with **15**-2Ni(II) or **16**-2Ni(II) (20 μ M, 37 °C, 3
403 hr). The fluorescence anisotropy measurement was conducted using **17** (0.4 μ M) in 100 mM phosphate buffer (pH
404 7.4) at 37 °C.



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Figure 7. Affinity-based cleavage of FGFR1 kinase domain.

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a, X-ray structure of FGFR1 in complex with futibatinib (PDB:6MZW). **b**, Structures of futibatinib and formylation

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probe **18**. **c**, Cleavage of FGFR1 kinase domain (1 μM) upon treatment with **18** (50 μM) or **1** (50 μM or 1 mM, pH

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7.4, 37 °C, 3 hr) in 100 mM phosphate buffer (pH 7.4) containing 150 μM TCEP in the presence of GSH (1 mM, 1

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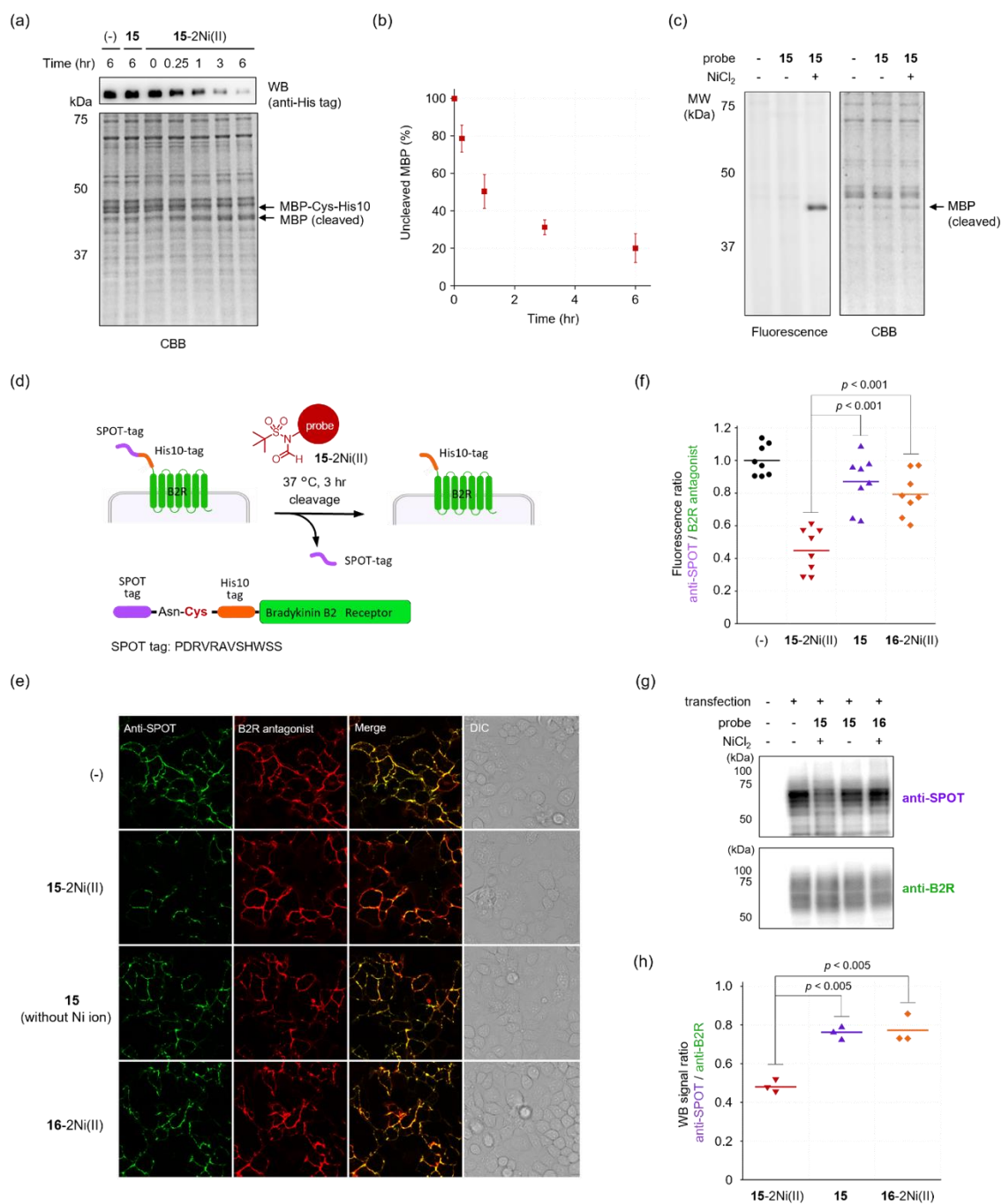
hr) at 37 °C. (upper panel: western blot analysis using anti-His tag antibody, lower panel: CBB staining). **d**, Plot of

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the relative band intensity of FGFR1 kinase domain upon treatment with **18** or **1**. The data were obtained from

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western blot analysis shown in Figure 6c (n = 6).



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Figure 8. Chemical cleavage of proteins by cysteine S-formylation under biorelevant conditions.

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a, Chemical cleavage of MBP-Cys-His10 (0.8 μM) using 15-2Ni(II) (20 μM) in A431 cell lysate (1 mg/mL in 100

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mM phosphate buffer, pH 7.4) at 37 °C. upper panel: western blot of MBP-Cys-His10 using anti-His tag antibody,

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lower panel: CBB staining. **b**, Time trace plot of the cleavage of MBP-Cys-His10 using 15-2Ni(II) in A431 Lysate

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by western blot analysis (mean ± s.d.: three independent experiments). **c**, Cleavage of MBP-Cys-His10 (0.8 μM)

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using 15-2Ni(II) (20 μM, 1 hr) in A431 lysate and subsequent fluorescence labeling with Cys-TAMRA (200 μM, 1

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hr) at 37 °C. left: in-gel fluorescence analysis, λ_{ex}/λ_{em} = 520/575 nm, right: CBB staining. **d**, Schematic illustration

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of chemical cleavage of SPOT-His10 tag fused B2R on the cell surface. **e**, Fluorescence imaging of HEK293T cells

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expressing SPOT-His10-B2R upon treatment with the formylation probes. The cells were treated with 15-2Ni(II), 15

423 (without NiCl₂) or **16-2Ni(II)** (40 μM, 3 hr) in HEPES-buffered saline (pH 7.4) at 37 °C. **f**, Plot of the fluorescence
424 intensity ratio (Alexa Fluor 568-anti-SPOT nanobody / Cy5-anti-B2R antagonist peptide) of HEK293 cells (*n* = 8).
425 **g**, Western blot analysis of SPOT-His10-B2R cleavage on the cell surface upon treatment with the formylation probes.
426 **h**, Plot of band intensity ratios (anti-SPOT/anti-B2R) in the western blot analysis (*n* = 3).