Giant Band Degeneracy *via* Orbital Engineering Enhances Thermoelectric Performance from Sb₂Si₂Te₆ to Sc₂Si₂Te₆

Wenzhen Dou, Kieran B. Spooner, Seán R. Kavanagh*, MiaoZhou* and David O. Scanlon*

W. Dou, Prof. M. Zhou

School of Physics, Beihang University, Beijing 100191, China

W. Dou, K. B. Spooner, S. R. Kavanagh, Prof. D. O. Scanlon

Department of Chemistry, University College London, London WC1H 0AJ, United Kingdom

K. B. Spooner, Prof. D. O. Scanlon

School of Chemistry, University of Birmingham, Birmingham B15 2TT, United Kingdom

S. R. Kavanagh

Department of Materials, Imperial College London, London SW7 2AZ, United Kingdom

K. B. Spooner, S. R. Kavanagh, Prof. D. O. Scanlon

Thomas Young Centre, University College London, London WC1E 6BT, United Kingdom

W. Dou, Prof. M. Zhou

Collaborative Centre for Physics and Chemistry, Institute of International Innovation,

Beihang University, Hangzhou 311115, China

Prof. M. Zhou

Tianmushan Laboratory, Hangzhou 310023, China

*E-mail: s.kavanagh19@imperial.ac.uk (S.R.K.); mzhou@buaa.edu.cn (M.Z.); d.o.scanlon@bham.ac.uk (D.O.S.)

Abstract

The complex interrelationships among thermoelectric parameters mean that a priori design of high-performing materials is difficult. However, band engineering can allow the power factor to be optimized through enhancement of the Seebeck coefficient. Herein, using layered Sb₂Si₂Te₆ and Sc₂Si₂Te₆ as model systems, we comprehensively investigate and compare their thermoelectric properties by employing density functional theory combined with semiclassical Boltzmann transport theory. Our simulations reveal that Sb₂Si₂Te₆ exhibits superior electrical conductivity compared to Sc₂Si₂Te₆ due to lower scattering rates and more pronounced band dispersion. Remarkably, despite Sb₂Si₂Te₆ exhibiting a lower lattice thermal conductivity, the introduction of Sc-d orbitals dramatically increases conduction band degeneracy in Sc₂Si₂Te₆, yielding a significantly improved Seebeck coefficient relative to Sb₂Si₂Te₆. As a result, Sc₂Si₂Te₆ is predicted to achieve an extraordinary dimensionless figure of merit (ZT) of 3.51 at 1000 K, which significantly surpasses the predicted maximum ZT of 2.76 for Sb₂Si₂Te₆ at 900 K. This work suggests that engineering band degeneracy through compositional variation is an effective strategy for improving the thermoelectric performance of layered materials.

Keywords: layered thermoelectric materials, band degeneracy, enhanced Seebeck coefficient, first-principles calculation, momentum relaxation time approximation

Introduction

With the development of electronics and ever-increasing demand for sustainable and versatile energy harvesting, thermoelectrics have become the subject of rapidly growing interest because they can directly transform thermal energy into valuable electrical power.^[1] The efficiency of thermoelectric materials is quantified by the dimensionless figure of merit ZT, defined as $ZT = S^2 \sigma T/\kappa$, where S is the Seebeck coefficient, σ represents the electrical conductivity, and T denotes the absolute temperature. The thermal conductivity κ includes both lattice thermal conductivity κ_1 and the electronic thermal conductivity κ_e . An ideal thermoelectric material should simultaneously exhibit high electrical conductivity to minimize internal energy loss, a high Seebeck coefficient for generating high voltage, and low thermal conductivity to maintain the temperature gradient. Motivated by these design principles, considerable efforts in the past decade have been devoted to exploring potential candidates that inherently possess the crucial attributes required for thermoelectric applications.^[2] Among the rapidly growing class of thermoelectric materials, layered compounds have shown advantages due to their unique crystal structure characterized by strong intralayer and weak interlayer bonds.^[3] Firstly, their strong discrepancy in chemical bonding strength is believed to induce partially-localized low-frequency phonon modes, low phonon group velocities, and large anharmonicity — all favourable for achieving low lattice thermal conductivity.^[4] Moreover, the low-dimensional structure translates to reduced dimensionality in the electrical transport properties, which can benefit thermoelectric performance.^[5] Furthermore, the anisotropic behaviour of electron and phonon transport in these layered systems provides an excellent opportunity to disentangle the interconnected thermoelectric parameters, especially when thermoelectric conversion occurs along the desired crystal direction within a single crystal.^[6]

Within the layered compounds, the $A_2B_2Q_6$ family (A = In, Cr, Bi, Sb or Sc; B = Si, Ge, and Q = Se, Te) has been intensively explored^[7] and some compounds show intriguing magnetic and magnetoelectric properties, such as pressure-driven spin-

crossover and long-range ferromagnetic order.^[7d, 7e] Interestingly, compounds based on A = Cr, Sb, In, and Bi are also reported to be promising candidates for thermoelectric applications.^[3a, 7b, 8] In particular, Sb₂Si₂Te₆ has been the focus of interest due to its remarkable thermoelectric properties.^[3a, 9] Sb₂Si₂Te₆ adopts a layered two-dimensional structure composed of Sb³⁺ cations and [Si₂Te₆]⁶⁻ units, resembling the Fe₂P₂Se₆ structure type. Owing to a low phonon group velocity and strong anharmonicity, Sb₂Si₂Te₆ exhibits a relatively low thermal conductivity, approximately 1.3 W m⁻¹ K⁻¹ at room temperature, decreasing to 0.5 W m⁻¹ K⁻¹ at 823 K.^[3a] Consequently, Sb₂Si₂Te₆ achieves an impressive ZT of 1.08 at 823 K.^[3a] Additionally, by constructing a cellular nanostructure through the deposition of a thin layer of Si₂Te₃ onto Sb₂Si₂Te₆ grains, the peak ZT at 823 K can be improved to 1.65 in the Sb₂Si₂Te₆/Si₂Te₃ cellular network.^[3a] Nevertheless, the average power factor (PF) of Sb₂Si₂Te₆ — approximately 10 µW cm⁻ 1 K⁻² within the temperature range of 400–823 K — lags behind that of conventional Pb-based thermoelectric materials (~30 μ W cm⁻¹ K⁻² for p-type PbTe and ~20 μ W cm⁻ 1 K⁻² for n-type PbTe).^[10] As a result, it is highly desirable to enhance the PF and, by extension, the overall thermoelectric performance of Sb₂Si₂Te₆-based thermoelectric devices.

Band engineering strategies are widely employed to improve the thermoelectric performance of materials.^[11] Band degeneracy, in particular, can enhance the PF by increasing the Seebeck coefficient and thus the ultimate thermoelectric performance.^[12] Band or valley degeneracy refers to the case where multiple bands have identical (degenerate) or closely matching within a few $k_{\rm B}T$ (effectively degenerate) energies. This can happen when *different bands* have minimal (or no) energy difference at the band edge (termed orbital degeneracy) and/or when multiple pockets of the *same bands* in the Brillouin zone are (effectively) degenerate, often due to symmetry equivalence *via* the combination of crystal and orbital symmetries (termed k-point degeneracy). Recently, Wang et al. reported a 22 % improvement in the *ZT* due to enhanced band degeneracy upon substituting *p*-valent Bi in BiCuOSe with *d*-valent La, predicting a *ZT* value of 1.46 for *n*-type LaCuOSe.^[13] We posit that a similar band-degeneracy induced

enhancement of the thermoelectric performance may be achievable in the $A_2B_2Q_6$ family by replacing *p*-valent Sb with a *d*-valent cation. Several members of the $A_2B_2Q_6$ family have been experimentally reported thus far, including Sc₂Si₂Te₆ in 2022^[7a] — however, the thermoelectric properties were not investigated.

In this work, we investigate the effects on band degeneracy and thermoelectric performance by modifying the conduction band orbital character in this crystal family, shifting from a Sb-*p* derived CBM in Sb₂Si₂Te₆ to a Sc-*d* CBM in Sc₂Si₂Te₆. Sc₂Si₂Te₆ shares an identical stacking fault pattern with Sb₂Si₂Te₆, and crystallizes in the same rhombohedral symmetry (space group R $\overline{3}$) with a single distinct crystallographic position each for Sc, Si and Te as revealed experimentally by high-resolution transmission electron microscopy and diffraction patterns.^[7a] We systematically investigated the thermal and electrical transport properties in layered Sb₂Si₂Te₆ and Sc₂Si₂Te₆ by solving the Boltzmann transport equation based on first-principles calculations, revealing a significant enhancement of band degeneracy and, consequently, the Seebeck coefficient in Sc₂Si₂Te₆ compared to Sb₂Si₂Te₆. We predict n-type Sc₂Si₂Te₆ to achieve an optimal *ZT* of 3.51 at 1000 K, surpassing the maximum *ZT* of 2.76 obtained for p-type Sb₂Si₂Te₆ by 27 %. These findings demonstrate that Sc₂Si₂Te₆ presents a competitive alternative to Sb₂Si₂Te₆, highlighting the validity of band degeneracy engineering in optimizing the PF and thermoelectric performance.

Results and Discussion

Equilibrium geometry and electronic structure



Figure 1 Top view of the crystal structure for (a) $Sb_2Si_2Te_6$ and (b) $Sc_2Si_2Te_6$. The 5 / 25

conventional unit cell is indicated by grey dashed lines. (c) Side view of $Sc_2Si_2Te_6$. There is a slight asymmetric distortion of the octahedra in $Sc_2Si_2Te_6$. The atom colours are Sb: orange, Sc: pink, Si: purple, and Te: blue.

Both Sb₂Si₂Te₆ (Figure 1a) and Sc₂Si₂Te₆ (Figure 1b) exhibit rhombohedral symmetry (space group R $\overline{3}$, No. 148) and consist of ABC-stacked slabs of (Sb/Sc)₂Si₂Te₆, where Sb (Sc) atoms and Si–Si dumbbells are arranged within the slabs (Figure 1c). Table 1 presents the lattice parameters of the conventional unit cell, optimized using both PBEsol and HSE06 with the D3 dispersion correction. The calculated lattice constants for Sb₂Si₂Te₆ and Sc₂Si₂Te₆ agree well with the experimental measurements^[3a, 7a] with HSE06+D3 showing improved agreement, though with a slight underestimation of the interlayer spacing as expected when neglecting temperature effects.^[14] Compared to Sb₂Si₂Te₆, Sc₂Si₂Te₆ exhibits a slight asymmetric distortion of the octahedra, with shorter Sc–Te bond lengths due to the smaller ionic radius of Sc but larger interlayer spacing.

Table 1 Calculated crystal structure information of $Sb_2Si_2Te_6$ and $Sc_2Si_2Te_6$. The % differences to the experimental structures are given in parentheses.^[3a, 7a] d_{layer} is the interlayer distance.

Functional	Compound	a (Å)	c (Å)	Sb/Sc–Te (Å)	d _{layer} (Å)
PBEsol+D3	Sb ₂ Si ₂ Te ₆	7.07 (-1.4 %)	20.10 (-5 %)	3.05	2.78
	$Sc_2Si_2Te_6$	6.85 (-2 %)	20.16 (-5 %)	2.90	2.99
HSE06+D3	Sb ₂ Si ₂ Te ₆	7.08 (-1.2 %)	20.78 (-2 %)	3.05	3.05
	$Sc_2Si_2Te_6$	6.91 (-1.2 %)	20.81 (-2 %)	2.93	3.22

Using HSE06+SOC with the *doped* package,^[15] the stability region (chemical potential limits) of Sc₂Si₂Te₆ in the Sc–Si–Te chemical space was computed (Figure S1), showing it to be thermodynamically stable with a relatively large stability window, suggesting ready synthesizability as witnessed experimentally.^[7a]

Figure 2 displays the calculated electronic band structures and DOS of $Sb_2Si_2Te_6$ and $Sc_2Si_2Te_6$ using HSE06+SOC. The difference in A-site cations markedly affects the band structure. $Sb_2Si_2Te_6$ exhibits a direct band gap of 0.48 eV, with both the CBM and VBM located at the Z point and this value is close to the experimental optical gap (~ 0.6 eV).^[3a] In contrast, Sc₂Si₂Te₆ exhibits an indirect band gap of 0.92 eV, where the VBM is located at the Γ point and the CBM is at the Σ point along the L–B path in reciprocal space. This is not a high-symmetry k-point, resulting in significantly increased k-point degeneracy at the conduction band edge.^[11b] Sb₂Si₂Te₆ has high electronic band dispersion, which will yield small carrier effective masses and higher carrier mobility. As indicated by the accompanying DOS plots in Figure 2, the CBM is composed of Sb-*5p* and Te-*5p* orbitals in Sb₂Si₂Te₆ and Sc-*3d* orbitals in Sc₂Si₂Te₆, while the VBM mainly derives from Te-*5p* orbitals. In Sb₂Si₂Te₆ there is a slight contribution to the VBM from Sb-*5s* orbitals while there is minimal hybridization with Sc-*3d* orbitals at the VBM in Sc₂Si₂Te₆ is located at Γ (rather than Z), is doubly (rather than singly) degenerate and exhibits reduced band dispersion. As expected due to the presence of heavy elements Te and (to a lesser extent) Sb, spin–orbit coupling (SOC) is found to significantly reduce the band gaps of these compounds, contributing VBM upshifts of 0.2–0.3 eV (Figure S2).



Figure 2 Calculated electronic band structures and orbital-projected DOS of (a) $Sb_2Si_2Te_6$ and (b) $Sc_2Si_2Te_6$ using HSE06+SOC, along the high-symmetry k-point path for rhombohedral crystals within the Setyawan & Curtarolo convention.^[16] The VBM is set to zero. (c) Charge density isosurface of the $Sc_2Si_2Te_6$ CBM state. The isosurface level is set to $0.02eV/Å^3$.

It is noteworthy that $Sb_2Si_2Te_6$ exhibits an energy difference of 0.3 eV between the lowest two conduction band valleys (i.e., $E_{CB1}-E_{CBM}$) and 0.6 eV between the highest two valence band valleys (i.e., $E_{VBM}-E_{VB1}$). In contrast, $Sc_2Si_2Te_6$ features two identical

CBM valleys (Σ and Σ') and a doubly orbital-degenerate Te-*p* VBM. Reduced energy differences between adjacent bands near the Fermi level result in increased effective band degeneracy, thus yielding a higher effective DOS in Sc₂Si₂Te₆. As a result of the differing locations of the band extrema in Sb₂Si₂Te₆ (both VBM and CBM at Z) and $Sc_2Si_2Te_6$ (VBM at Γ and CBM at Σ), there is a significant difference in their k-point degeneracy. For the VBM, the k-point degeneracy is 1 in both systems. However, the CBM in Sc₂Si₂Te₆ displays a notably higher k-point degeneracy of 18, as opposed to the value of 1 in Sb₂Si₂Te₆. The band-decomposed charge density of the CBM in Sc₂Si₂Te₆ can provide insights into the high band degeneracy, exhibiting hexagonal symmetry (6-fold degenerate) along the in-plane direction, as seen in Figure 2c. Due to the crystal symmetry (hexagonal ABC stacking) and orbital symmetry (hybridized d_x^2 v_{y}^{2} , d_{xz} and d_{z}^{2} , Figures 2 and S3), there are two other diagonal cross-plane directions (corresponding to the Σ' k-point) which are also degenerate and exhibit hexagonal (6fold) symmetry. Therefore, the total k-point degeneracy of the CBM state in Sc₂Si₂Te₆ is 18. To further elucidate this point, the Fermi surfaces are plotted at 0.1 eV above (below) the CBM (VBM) using the IFermi package^[17] (Figure 3). Due to the symmetry of the Brillouin zone, there is two half electron (hole) pockets at the Z point for Sb₂Si₂Te₆ and thus the k-point degeneracy is 1 for both VBM and CBM (Figure 3a). For $Sc_2Si_2Te_6$, the VBM is at Γ , and an isolated full-hole pocket is generated in the centre of the Brillouin zone — again corresponding to a k-point degeneracy of 1. Meanwhile, six relatively flat electron pockets appear close to the edges of the Brillouin zone for the CBM (Figure 3b). These electron pockets show significant anisotropy, being mostly flat along the z (cross-plane) direction but relatively dispersed along the horizontal direction. Using a dense interpolated k-point mesh and setting the Fermi surface energy to just 3 meV above CBM, we see that there are in fact three degenerate elliptical electron pockets within each of the six larger pockets observed with the higher Fermi level — corresponding to the total effective k-point degeneracy of 18 for the CBM in Sc₂Si₂Te₆ (Figure 3c). From this analysis, we also determine the CBM k-point coordinates to be $\Sigma = (0.5, 0.16, -0.16)$. The increased k-point degeneracy in Sc₂Si₂Te₆ thus yields a giant band degeneracy, dramatically enhancing the DOS effective mass. In addition, the DOS effective masses closely depend on the carrier effective masses, summarized in Table 2. Sc₂Si₂Te₆ displays larger and more anisotropic effective masses for both electrons and holes compared to Sb₂Si₂Te₆, as expected, which is also represented by the smaller Fermi velocities of Sc₂Si₂Te₆ compared to Sb₂Si₂Te₆ (Figure 3). Therefore, the heightened band degeneracy, combined with increased carrier effective masses, is anticipated to result in a significantly higher Seebeck coefficient for Sc₂Si₂Te₆.



Figure 3. Fermi surface of (a) $Sb_2Si_2Te_6$ and (b) $Sc_2Si_2Te_6$ plotted at 0.1 eV below CBM and above VBM, respectively. (c) The iso-energy Fermi surface is 3 meV above the CBM for $Sc_2Si_2Te_6$. The different colours represent the magnitude of carrier group velocities.

Table 2. Hole and electron effective masses (m_e) of Sb₂Si₂Te₆ and Sc₂Si₂Te₆ were calculated using parabolic fitting in sumo with the HSE06+SOC functional. The corresponding k-point path is given in parentheses.

System	Hole (m _e)	Electron (m _e)
Sb ₂ Si ₂ Te ₆	0.17 (Z-X) 0.28 (Z-B) 0.28 (Z-Q)	0.29 (Z-X) 0.09 (Z-B) 0.09 (Z-Q)
$Sc_2Si_2Te_6$	0.32 (L-B ₁) 0.33 (Γ-X) 3.67 (Γ-B)	0.43 (Σ-Β ₁) 0.32 (Σ-Γ)

Band alignment

The band alignment of $Sb_2Si_2Te_6$ and $Sc_2Si_2Te_6$ is calculated using the core–level alignment approach (see Supplementary Information for more details).^[18] The results are depicted in Figure 4 together with the band alignment of several other highperformance thermoelectric materials. The electron affinities (EA) and ionization potentials (IP) can provide an indication of the propensity to p- or n-type dopability. Compared to bipolar-dopable PbTe, the IP of $Sb_2Si_2Te_6$ is smaller (due to the antibonding Sb-s – Te-p interaction at the VBM), while that of $Sc_2Si_2Te_6$ is larger, indicating the formation of p-type defects may be more likely for $Sb_2Si_2Te_6$. Indeed, $Sb_2Si_2Te_6$ exhibits strong p-type behaviour experimentally, which is attributed to Sb vacancies.^[3a] The EA of $Sc_2Si_2Te_6$ is analogous to that of PbTe, n-type $Y_2Ti_2O_5S_2$ and BaSnO₃, while that of $Sb_2Si_2Te_6$ is smaller, suggesting that $Sc_2Si_2Te_6$ may favour ntype doping.



Figure 4 Band alignment of $Sb_2Si_2Te_6$ and $Sc_2Si_2Te_6$ calculated in this study and compared to the band alignments of bipolar PbTe,^[19] n-type $Y_2Ti_2O_5S_2^{[20]}$ and BaSnO₃.^[21] Ionisation potential (IP) is annotated.

Electronic transport properties

Using the calculated electronic (Figure 2), phonon (Figure S8) and electronphonon properties from DFT, the carrier mobilities were computed using the Boltzmann transport equations as implemented in the AMSET package (see Methods and Supplementary Information).^[22] The individual scattering mechanisms are plotted as functions of temperature, carrier concentration and electronic energy in Figures S4 and S5. As expected, the dominant mobility-limiting scattering mechanism varies between polar-optical phonon (POP), ionised impurity (IMP) or acoustic deformation potential (ADP) scattering, depending on the doping/carrier concentration and temperature. **10** / **25** Under temperature and doping concentrations which maximise ZT (shown later), carrier-phonon scattering dominates for Sb₂Si₂Te₆ — primarily due to the lower energy phonon modes from the heavier Sb atoms (Figure S8) — with POP scattering dominant under *p*-type doping and ADP scattering under *n*-type doping. For $Sc_2Si_2Te_6$ on the other hand, IMP scattering dominates under ZT-optimised conditions. The resultant electronic transport properties for n-type and p-type doping across a range of carrier concentrations and temperatures are shown in Figures 5 and 6, respectively. As expected, we witness decreases in electrical conductivity with increasing temperature, however the effect is relatively minor for high carrier concentrations due to the dominance of (mostly) temperature-independent impurity scattering over carrierphonon scattering in these regimes. For *n*-type Sb₂Si₂Te₆, the temperature dependence is stronger due to greater carrier-phonon interactions for this case. Overall, Sb₂Si₂Te₆ exhibits higher electrical conductivity than $Sc_2Si_2Te_6$ under both *n* and (particularly) *p*type conditions due to the lower carrier effective masses (Figure 2, Table 2) and scattering rates (Figures S4 and S5). The enhanced band degeneracy in Sc₂Si₂Te₆ may also increase the rate of inter-valley scattering, contributing to a decrease in carrier mobility.^[11b] As expected from the Wiedemann–Franz law ($\kappa_e = L\sigma T$, where L is the Lorenz number and T is temperature), the trends in electronic thermal conductivity (κ_e) mostly follow those of electrical conductivity (σ).



Figure 5 The calculated electronic transport properties as a function of temperature for 11 / 25

n-type (a, b, c, and d) $Sb_2Si_2Te_6$ and (e, f, g, and h) $Sc_2Si_2Te_6$ with four different carrier concentrations. $Sb_2Si_2Te_6$ and $Sc_2Si_2Te_6$ have demonstrated thermal stability up to 920 K and 1023 K,^[3a, 7a] respectively, and so these ranges are used in our analysis.



Figure 6 The calculated electronic transport properties as a function of temperature for p-type (a, b, c, and d) Sb₂Si₂Te₆ and (e, f, g, and h) Sc₂Si₂Te₆ with four different carrier concentrations.

The Seebeck coefficient (*S*) reflects the voltage generated in response to a temperature gradient, and its sign depends on the dominant charge carrier type (positive for holes, negative for electrons). In general, the Seebeck coefficient can be expressed as:

$$S = \frac{8\pi^2 k_B^2}{3eh^2} m_{DOS}^* T \left(\frac{\pi}{3n}\right)^{\frac{2}{3}}$$
(1)

where k_B is the Boltzmann constant, *e* represents the elementary charge, *h* denotes Plank's constant, *n* is the carrier concentration, and m_{DOS}^* represents the DOS effective mass.^[23] Thus as witnessed in Figures 5 and 6, the Seebeck coefficient typically exhibits opposite dependence on temperature and carrier concentration compared to the electrical conductivity, decreasing with carrier concentration and increasing with temperature due to their effects on carrier diffusion (and corresponding voltage) across a temperature gradient in a material. For both *n* and *p*-type conditions, we see that the Seebeck coefficient in Sc₂Si₂Te₆ consistently exceeds that of Sb₂Si₂Te₆, showing the opposite behaviour to the electrical conductivity. In particular, *n*-type Sc₂Si₂Te₆ retains

a large Seebeck coefficient (S > 300 μ V/K) even at relatively high carrier concentrations ($n = 10^{20}$ cm⁻³) and electrical conductivities ($\sigma \sim 10^5$ S/m). According to equation 1, for a given carrier concentration and temperature, the Seebeck coefficient is primarily determined by the DOS effective mass — which can be approximated by the equation $m_{DOS}^* = N_v^{\frac{2}{3}} m^*$.^[24] Here, N_v is the number of degenerate valleys in the electronic band structure and m^* is the effective mass within a single valley. As discussed previously, Sc₂Si₂Te₆ exhibits increased band degeneracy (18 and 2 for *n* and *p*-type, respectively, vs 1 for both carrier types in Sb₂Si₂Te₆) and carrier effective masses (Table 2) compared to Sb₂Si₂Te₆, elucidating the origins of increased Seebeck coefficients for Sc₂Si₂Te₆. The calculated electronic transport properties, broken down into the in-plane and cross-plane directions for these layered materials, are provided in the Supplementary Information, with Sc₂Si₂Te₆ showing pronounced anisotropy (discussed further below). Additionally, the non-monotonic behaviour for the Seebeck coefficients at low carrier concentrations is due to the bipolar conduction effect, with further discussion provided in the Supplementary Information.

The power factor (PF = $S^2\sigma$) is a key parameter in thermoelectric performance (*ZT* \simeq PF), which can have a complex dependence on carrier concentration and temperature due to their competing effects on *S* and σ . As a result, a balanced carrier concentration, striking an equilibrium between the Seebeck coefficient and electrical conductivity, is necessary to maximise PF. The optimal PF is higher for p-type (8.16 mW m⁻¹K⁻²) rather than n-type (4.67 mW m⁻¹K⁻²) Sb₂Si₂Te₆ (Figures 5 and 6). In contrast, for Sc₂Si₂Te₆, the highest PF of n-type is 19 mW m⁻¹K⁻², which significantly exceeds the value of 3.06 mW m⁻¹K⁻² for p-type doping (Figures 5 and 6). Moreover, this optimal PF for n-type Sc₂Si₂Te₆ markedly exceeds the PF predicted for various layered thermoelectric materials, such as BiCuOSe (~1.71 mW m⁻¹K⁻²),^[13] while it still lags behind that predicted for monolayer SnSe (~28 mW m⁻¹K⁻²).^[25] This ultra-high PF in *n*-type Sc₂Si₂Te₆ primarily originates from the enhanced Seebeck coefficient attributable to conduction band degeneracy, as discussed above. Using a band degeneracy strategy to enhance the PF has also been reported in other materials, such as PbTe, which achieves 13 / 25

a band degeneracy of 16 by tuning the doping and composition.^[11b]

Thermal transport properties

The calculated lattice thermal conductivity κ_1 of both compounds is shown in Figure 7. The lattice thermal conductivity decreases with temperature increases due to greater phonon–phonon scattering at higher temperatures. The lattice thermal conductivity of both compounds is found to be relatively low as expected, particularly in the inter-layer (*z*) direction, with Sc₂Si₂Te₆ exhibiting higher lattice thermal conductivity values than Sb₂Si₂Te₆ (by a factor of ~3x). This behaviour can be rationalised through the calculated phonon dispersions, which are provided in Figure S8. Given their structural and compositional similarity, the phonon dispersions are relatively similar, with many flat bands present — indicating low phonon group velocities and the likelihood of low lattice thermal conductivity — and the heavy Te atoms dominating the low-energy optical and acoustic modes in both cases. Due to the comparable mass of Sb (121.76 a.u.) and Te (127.60 a.u.) in Sb₂Si₂Te₆ (c.f. 44.96 a.u. for Sc), a greater density of phonon bands is witnessed in the low-frequency range (0–5 THz) compared to Sb₂Si₂Te₆, leading to increased phonon scattering and thus reduced lattice thermal conductivity as witnessed in Figure 7.



Figure 7 Calculated lattice thermal conductivity κ_1 of (a) Sb₂Si₂Te₆ and (b) Sc₂Si₂Te₆ as a function of temperature. Sb₂Si₂Te₆ and Sc₂Si₂Te₆ have demonstrated thermal stability up to 920 K and 1023 K,^[3a, 7a] respectively, and so these ranges are used in our analysis.

The room temperature lattice thermal conductivity (1.03 W m⁻¹ K⁻¹) of Sb₂Si₂Te₆ closely aligns with the experimental value (1.0 W m⁻¹ K⁻¹)^[3a] and is also comparable 14 / 25

to that of other layered thermoelectric materials, such as BiCuOSe $(1.0 \text{ W m}^{-1} \text{ K}^{-1})^{[13]}$ and Bi₂O₂Se $(1.1 \text{ W m}^{-1} \text{ K}^{-1})$.^[26] Importantly, the lattice thermal conductivity is highly anisotropic in both Sb₂Si₂Te₆ and Sc₂Si₂Te₆, resulting in average $\kappa_{x,y}/\kappa_z$ ratios of 1.82 and 2.24, respectively, at 300 K. The anisotropic nature of lattice thermal conductivity can be attributed to the difference between relatively weak (van der Waals) interlayer interactions along the cross-plane direction and strong (covalent) intralayer interactions along the in-plane direction. Further analysis of the atomistic origins of thermal transport behaviour in these layered compounds was performed by querying the frequency-dependent phonon group velocities, lifetimes and cumulative lattice thermal conductivity, as provided in the Supplementary Information (Figures S9 and S10). These analyses further confirmed the role of Sb/Te mass similarity in the reduced lattice thermal conductivity of Sb₂Si₂Te₆, leading to enhanced phonon–phonon scattering and thus reduced phonon lifetimes, rather than any major differences in group velocities.

Thermoelectric figure of merit

With these electrical and thermal transport properties, we can calculate the theoretical dimensionless figure of merit ZT for Sb₂Si₂Te₆ and Sc₂Si₂Te₆ as functions of temperature and carrier concentration (Figures 8 and 9). It is found that for Sb₂Si₂Te₆, the p-type system achieves a superior ZT compared to the n-type system, while the results are opposite in Sc₂Si₂Te₆. For p-type doping, Sb₂Si₂Te₆ consistently exhibits higher ZT than Sc₂Si₂Te₆ regardless of the transport direction, which can be attributed to the lower lattice thermal conductivity (Figure 7) and higher electrical conductivity (Figure 6). However, for *n*-type doping, we find an extremely large calculated ZT of 3.51 at 1000 K for the in-plane direction in Sc₂Si₂Te₆, greatly surpassing the optimal ZT observed (1.69 - 2.06) for n-type Sb₂Si₂Te₆. The exceptional ZT observed for *n*-type Sc₂Si₂Te₆ originates from the dramatically enhanced Seebeck coefficient and PF (Table 4) caused by ultra-high band degeneracy, as discussed above, in combination with the low lattice thermal conductivity (Figure 7). This value surpasses that of typical layered thermoelectric materials, such as p-type SnSe (measured ZT of ~ 2.6 at 923 K) and BiCuOSe (predicted ZT of 1.32 at 1000 K).^[3d, 27] Similar predictions of high ZT through 15 / 25

high Seebeck coefficient and PF have also been observed in other materials, such as Na₂TISb ($ZT \sim 4.81$) and InBrSe ($ZT \sim 5.12$).^[28] We note that while the in-plane ZT is extremely high for *n*-type Sc₂Si₂Te₆, the cross-plane ZT is very low (0.06 at 1000 K, the ZT without colour scale of n-type Sc₂Si₂Te₆ along the cross-plane direction as shown in Figure S11), which can be attributed to the low cross-plane *n*-type conductivity (Figure S6) — itself a consequence of the relatively localised Sc-*d* orbitals in the Sc₂Si₂Te₆ CBM, giving weak interlayer electronic interactions. Overall, these results suggest that Sc₂Si₂Te₆ holds promise as an excellent thermoelectric material for applications in the moderate-to-high temperature range, and also demonstrate the power of band engineering for enhancing thermoelectric performance within compound families. The calculated optimal ZT and associated carrier concentrations, PF, electronic thermal conductivity, and lattice thermal conductivity for Sb₂Si₂Te₆ and Sc₂Si₂Te₆ are summarized in Tables 3 and 4.



Figure 8 The predicted thermoelectric figure of merit ZT of p-type (a, c) Sb₂Si₂Te₆ and (b, d) Sc₂Si₂Te₆ against temperature and carrier concentration along the in-plane and cross-plane directions. The lightest colours indicate the highest ZT.



Figure 9 The predicted thermoelectric figure of merit ZT of n-type (a, c) Sb₂Si₂Te₆ and (b, d) Sc₂Si₂Te₆ against temperature and carrier concentration along the in-plane and cross-plane directions. The lightest colours indicate the highest ZT.

We note that the theoretically predicted maximum *ZT* for *p*-type Sb₂Si₂Te₆ exceeds the experimentally reported result (~1.08 at 823 K, increased to ~1.65 when incorporated in a cellular nanostructure with Si₂Te₃).^[3a] Notably, the calculated lattice thermal conductivity of Sb₂Si₂Te₆ matches well with the experimental measurements, suggesting the primary origin of reduced *ZT*s in experiment are due to additional scattering mechanisms, for example due to impurities and grain boundaries, and suboptimal electronic properties such as carrier concentration. This discrepancy can be explained by the following two aspects. On the one hand, in experiment, thermal radiation,^[29] air-induced thermal convection^[30] impurity phases and the effect of grain boundary scattering on carrier mobility^[31] can lead to degradation of thermoelectric performance — effects which are not included in our theoretical model. On the other hand, this experimental value corresponds to a single carrier concentration (measured to be ~5.6×10¹⁹ cm⁻³ at room temperature) which may vary with temperature, but we calculate the thermoelectric properties over a range of temperatures and carrier concentrations. Indeed, as expected, we find a high sensitivity of the overall ZT to the carrier concentration (Figure 8), with different optimal concentrations for the in-plane and cross-plane directions. Thus, given the un-optimised experimental carrier concentration and potential additional factors not accounted for in our model, a lower experimental figure of merit is expected. Our simulations therefore reflect an idealised model which can help guide experimental efforts in further improving the thermoelectric performance of this material. In particular, the anisotropy in thermoelectric properties indicates that controlled-orientation single crystals (with optimised carrier concentrations) are expected to achieve significantly higher ZT values than polycrystalline samples for these layered compounds.

Table 3 Predicted maximum ZT in Sb₂Si₂Te₆ along the in-plane and cross-plane directions together with corresponding charge carrier concentration (n), power factor (PF), electronic thermal conductivity (κ_e), and lattice thermal conductivity (κ_l).

System	Direction	ZT	n (cm ⁻³)	$PF (\mu W m^{-1} K^{-2})$	$\kappa_e(W\ m^{-1}\ K^{-1})$	$\kappa_l (W \ m^{-1} \ K^{-1})$
n-type	ху	1.69	1.47×10 ¹⁹	2160	0.70	0.42
	Z	2.06	6.81×10 ¹⁹	3520	1.32	0.22
p-type	ху	2.76	3.16×10 ¹⁹	5310	1.32	0.41
	Z	2.64	1.47×10 ¹⁹	2120	0.5	0.22

Table 4 Predicted maximum ZT in Sc₂Si₂Te₆ along the in-plane and cross-plane directions together with corresponding charge carrier concentrations (n), power factor (PF), electronic thermal conductivity (κ_e), and lattice thermal conductivity (κ_l).

System	Direction	ZT	n (cm ⁻³)	$PF (\mu W m^{-1} K^{-2})$	$\kappa_{e} (W m^{-1} K^{-1})$	$\kappa_l \left(W \ m^{-1} \ K^{-l}\right)$
n-type	ху	3.51	1.16×10 ²⁰	9110	1.40	1.21
	Z	0.06	4.64×10 ²⁰	30.2	0.014	0.54
p-type	ху	1.57	1.06×10 ²⁰	3150	0.80	1.21
	Z	1.82	1.09×10 ²⁰	1640	0.37	0.54

Summary

In conclusion, our investigation systematically explored the intrinsic thermoelectric properties of Sb₂Si₂Te₆ and Sc₂Si₂Te₆, employing a combination of semiclassical Boltzmann transport theory and first-principles calculations. While both layered compounds exhibit low intrinsic lattice thermal conductivities, the low mass difference between Sb and Te results in greater phonon-phonon scattering and shorter phonon lifetimes in Sb₂Si₂Te₆, yielding very low lattice thermal conductivity. We find substantial differences in the electronic structures of these two compounds due to the different orbital characters of the A-site cations. Specifically, Sb₂Si₂Te₆ manifests a small direct bandgap, while Sc₂Si₂Te₆ exhibits a larger, *indirect* bandgap. The combination of crystal (hexagonal ABC stacking) and orbital (hybridized $d_x^2 - y^2$, d_{xz} and d_z^2) symmetry at the Sc₂Si₂Te₆ CBM gives rise to a non-high-symmetry band-edge kpoint and dramatically enhanced band degeneracy (increasing from 1 in Sb₂Si₂Te₆ to 18). Combined with good electron mobility in the in-plane direction, this yields an ultrahigh power factor (19 mW m⁻¹ K⁻²) and thus, along with the low lattice thermal conductivity, a large predicted ZT of 3.51 for n-type Sc₂Si₂Te₆ (compared to the optimal ZT of 2.76 for p-type Sb₂Si₂Te₆). This enhanced ZT arises despite the \sim 3x increase in lattice thermal conductivity upon substituting Sb with Sc, as the band degeneracy enhancement of the PF overcomes this loss. We note that this optimal ZT requires a relatively high charge carrier concentration ($n \sim 10^{20} \text{ cm}^{-3}$), so understanding the defect (dopant) chemistry and possible engineering strategies will be key to realising this performance experimentally.^[32] This study demonstrates the potential of Sc₂Si₂Te₆ for thermoelectric applications and, more broadly, the power of band (degeneracy) engineering as a route to designing high-performance thermoelectric materials.

Supporting Information

Supporting Information is available from the Wiley Online Library or the author.

Acknowledgements

We thank Alex M. Ganose for insightful discussions regarding the electronic band structures of these compounds. W.Z.D, K.B.S, and D.O.S acknowledge support from

the European Research Council (grant 758345). S.R.K. acknowledges the EPSRC Centre for Doctoral Training in the Advanced Characterisation of Materials (EP/S023259/1) for a PhD studentship. W.Z.D and M.Z acknowledge the support of the National Key R&D Program of China (2022YFF0708800), the Natural Science Foundation of Zhejiang Province (LZ22A040004), and the National Natural Science Foundation of China (11674042). W.Z.D acknowledges the support of the international joint doctoral education fund of Beihang University. W.Z.D, K.B.S, S.R.K and D.O.S acknowledge the support of the UCL Myriad and Kathleen High-Performance Computing Facilities (Myriad@UCL, Kathleen@UCL). K.B.S, S.R.K and D.O.S acknowledge the support of the ARCHER2 UK National Supercomputing Service (https://www.archer2.ac.uk) via our membership of the UK's HEC Materials Chemistry Consortium, which is funded by the EPSRC (EP/L000202, EP/R029431 and EP/T022213), the UK Materials and Molecular Modelling (MMM) Hub (Young EP/T022213). W.Z.D and M.Z acknowledge the support of the High-Performance Supercomputing Center of Zhongfa Aviation Institute of Beihang University and the Center for High-Performance Computing of Beihang University (BHHPC). K.B.S and D.O.S acknowledge the University of Birmingham's BlueBEAR HPC service (http://www.birmingham.ac.uk/bear); Baskerville Tier 2 HPC the service (https://www.baskerville.ac.uk/), which was funded by the EPSRC and UKRI through the World Class Labs scheme (EP/T022221/1) and the Digital Research Infrastructure programme (EP/W032244/1) and is operated by Advanced Research Computing at the University of Birmingham; and the Sulis Tier 2 HPC platform hosted by the Scientific Computing Research Technology Platform at the University of Warwick, which is funded by EPSRC Grant EP/T022108/1 and the HPC Midlands+ consortium.

Computational Methods

The calculations were performed within density functional theory (DFT) as implemented in the Vienna Ab initio Simulation Package (VASP).^[33] The projector augmented wave (PAW) pseudopotential method was employed to account for the interaction between core and valence electrons.^[34] A plane-wave kinetic energy cutoff

of 400 eV and a $6 \times 6 \times 6$ Γ -centered k-point grid was used for a 10-atom primitive cell, ensuring convergence of total energy within 1 meV/atom using vaspup2.0.^[35] The cutoff was increased to 520 eV during geometry optimizations to avoid Pulay stress. To account for Van der Waals dispersion interactions, the D3 correction of Grimme et al. was applied.^[35a, 35b] The structures were fully relaxed using the Heyd–Scuseria– Ernzerhof (HSE06) hybrid DFT functional,^[36] except for lattice dynamics calculations (ionic contribution to the dielectric constant, polar optical phonon (POP) frequency and phonon force constants) for which the Perdew–Burke–Ernzerhof generalized gradient approximation (GGA) functional for solids (PBEsol) was used.^[37] There are no constraints imposed on the unit cell shape or size, and relaxation continued until the change in maximum force on the ions and electronic free energy did not exceed 0.0005 eV/Å and 1×10^{-8} eV, respectively. Electronic structure and high-frequency dielectric constant calculations used HSE06 including spin–orbit coupling (SOC) effects.

The *doped* defect calculation software^[15] was used to collate all competing phases in the Sc–Si–Te chemical space within 0.1 eV/atom of the convex hull according to the Materials Project database — which were then re-relaxed and calculated using HSE06+SOC, parse the calculation outputs and plot the calculated chemical stability region (chemical potential ranges) for $Sc_2Si_2Te_6$.

The AMSET software package was employed to calculate the electronic transport properties.^[22] Typically, the Boltzmann transport equations are solved using the constant relaxation-time approximation (CRTA), which usually results in the overestimation of ZTs.^[22, 38] Instead, AMSET improves upon the CRTA by using the momentum relaxation-time approximation (MRTA) to explicitly calculate scattering rates for the individual electronic states within the Born approximation. AMSET offers insights into the contributions to the transport properties from four types of scattering: POP scattering, acoustic deformation potential (ADP) scattering, ionized impurity (IMP) scattering, and piezoelectric (PIE) scattering. These contributions are integrated into the *ZT* equation *via* the electrical conductivity and electronic thermal conductivity. The piezoelectric constant was calculated to be negligible for these materials, and so 21/25

piezoelectric scattering is not included in the analysis. The Seebeck coefficient is also calculated by AMSET, but is not influenced by scattering.

The material parameters required as input for AMSET, such as high-frequency and static dielectric constants, elastic constants, phonon frequencies, and deformation potentials, were obtained by first-principles calculations. Ionic dielectric constant contributions and the Γ -point phonon frequencies and dipole moments required to calculate the polar-optic phonon frequency in AMSET were acquired using the finite-displacement method with the PBEsol functional. Deformation potential and high-frequency dielectric constant were determined using HSE06 + SOC. These data are available in Table S1. The transport properties have been converged with respect to the interpolation factor, as shown in Figure S12.

Lattice dynamics calculations were performed using the finite-displacement method within the Phonopy package.^[39] Convergence of the phonon dispersion with the supercell size used for second-order force constants (FCs) was explicitly checked, as illustrated in Figure S13. Second-order FCs were computed using a $3\times3\times1$ supercell derived from the 30-atom conventional cell, employing the PBEsol functional.^[40] Atom contributions to the lattice dynamics were determined from the atom-projected phonon density of states (PDOS), calculated through Fourier interpolation. The lattice thermal conductivity was calculated using the Phono3py package,^[41] which employs a supercell method to calculate the third-order FCs with the PBEsol functional. A $2\times2\times1$ supercell based on the conventional cell was used to calculate the third-order FCs. The lattice thermal conductivity was determined by solving the phonon Boltzmann transport equation within the single-mode relaxation-time approximation. The convergence of q-point mesh density was verified, and a $13\times13\times13$ mesh was selected, as illustrated in Figure S14. Plots of the electronic, phonon and thermoelectric properties were generated using Sumo,^[42] IFermi^[17] and ThermoParser.^[43]

References

- [1] G. Tan, L.-D. Zhao, M. G. Kanatzidis, Chem. Rev. 2016, 116, 12123.
- [2] a)T. Zhu, Y. Liu, C. Fu, J. P. Heremans, J. G. Snyder, X. Zhao, Adv. Mater. 2017, 29,

1605584; b)Y. Zhou, L. D. Zhao, Adv. Mater. 2017, 29, 1702676.

[3] a)Y. Luo, S. Cai, S. Hao, F. Pielnhofer, I. Hadar, Z.-Z. Luo, J. Xu, C. Wolverton, V. P. Dravid, A. Pfitzner, Q. Yan, M. G. Kanatzidis, *Joule* 2020, 4, 159; b)Q. Jin, S. Jiang, Y. Zhao, D. Wang, J. Qiu, D. M. Tang, J. Tan, D. M. Sun, P. X. Hou, X. Q. Chen, K. Tai, N. Gao, C. Liu, H. M. Cheng, X. Jiang, *Nat. Mater.* 2019, 18, 62; c)X. Chen, H. Wu, J. Cui, Y. Xiao, Y. Zhang, J. He, Y. Chen, J. Cao, W. Cai, S. J. Pennycook, Z. Liu, L.-D. Zhao, J. Sui, *Nano Energy* 2018, 52, 246; d)L. D. Zhao, S. H. Lo, Y. Zhang, H. Sun, G. Tan, C. Uher, C. Wolverton, V. P. Dravid, M. G. Kanatzidis, *Nature* 2014, 508, 373; e)L. D. Zhao, D. Berardan, Y. L. Pei, C. Byl, L. Pinsard-Gaudart, N. Dragoe, *Applied Physics Letters* 2010, 97, 092118.

[4] a)M. Samanta, K. Pal, P. Pal, U. V. Waghmare, K. Biswas, *J Am Chem Soc* 2018, 140, 5866;
b)C. Wang, Y. Chen, *npj Computational Materials* 2020, 6, 26; c)B. Qin, D. Wang, X. Liu, Y. Qin, J.-F. Dong, J. Luo, J.-W. Li, W. Liu, G. Tan, X. Tang, J.-F. Li, J. He, L.-D. Zhao, *Science* 2021, 373, 556.

[5] a)L. Wu, J. Yang, S. Wang, P. Wei, J. Yang, W. Zhang, L. Chen, *Phys. Rev. B* 2014, 90, 195210;
b)H. Ohta, S. Kim, Y. Mune, T. Mizoguchi, K. Nomura, S. Ohta, T. Nomura, Y. Nakanishi, Y. Ikuhara, M. Hirano, H. Hosono, K. Koumoto, *Nat. Mater.* 2007, 6, 129.

[6] a)A. Li, C. Hu, B. He, M. Yao, C. Fu, Y. Wang, X. Zhao, C. Felser, T. Zhu, *Nat Commun* **2021**, 12, 5408; b)A. Mehdizadeh Dehkordi, M. Zebarjadi, J. He, T. M. Tritt, *Materials Science and Engineering: R: Reports* **2015**, 97, 1.

[7] a)F. Pielnhofer, S. Bette, R. Eger, V. Duppel, J. Nuss, C. Dolle, R. E. Dinnebier, B. V. Lotsch, *Zeitschrift für anorganische und allgemeine Chemie* 2022, 648, e202200234; b)Y. Luo, Z. Ma, S. Hao, S. Cai, Z. Z. Luo, C. Wolverton, V. P. Dravid, J. Yang, Q. Yan, M. G. Kanatzidis, *J Am Chem Soc* 2022, 144, 1445; c)M. A. Korkmaz, E. Deligoz, H. Ozisik, *Journal of Electronic Materials* 2021, 50, 2779; d)I. A. Verzhbitskiy, H. Kurebayashi, H. Cheng, J. Zhou, S. Khan, Y. P. Feng, G. Eda, *Nature Electronics* 2020, 3, 460; e)C. Gong, L. Li, Z. Li, H. Ji, A. Stern, Y. Xia, T. Cao, W. Bao, C. Wang, Y. Wang, Z. Q. Qiu, R. J. Cava, S. G. Louie, J. Xia, X. Zhang, *Nature* 2017, 546, 265; f)E. Sandre, V. Carteaux, A. M. Marie, G. Ouvrard, *Journal of Alloys and Compounds* 1994, 204, 145; g)G. Ouvrard, E. Sandre, R. Brec, *J. Solid State Chem.* 1988, 73, 27; h)R. E. Marsh, *J. Solid State Chem.* 1988, 77, 190; i)J. Gopalakrishnan, K. S. Nanjundaswamy, *Materials Research Bulletin* 1988, 23, 107.

[8] a)D. Yang, W. Yao, Q. Chen, K. Peng, P. Jiang, X. Lu, C. Uher, T. Yang, G. Wang, X. Zhou, *Chem. Mater.* 2016, 28, 1611; b)R. Lefèvre, D. Berthebaud, O. Lebedev, O. Pérez, C. Castro, S. Gascoin, D. Chateigner, F. Gascoin, *Journal of Materials Chemistry A* 2017, 5, 19406.
[0] Z. Huger, L. D. Zhan, *Tunch Chem.* 2020, 2, 20

[9] Z. Huang, L.-D. Zhao, Trends Chem. 2020, 2, 89.

[10] a)Y. Pei, J. Lensch-Falk, E. S. Toberer, D. L. Medlin, G. J. Snyder, *Adv. Funct. Mater.* 2011, 21, 241; b)Q. Zhang, E. K. Chere, Y. Wang, H. S. Kim, R. He, F. Cao, K. Dahal, D. Broido, G. Chen, Z. Ren, *Nano Energy* 2016, 22, 572; c)C. C. Kang, K. A. Yamauchi, J. Vlassakis, E. Sinkala, T. A. Duncombe, A. E. Herr, *Nat Commun* 2016, 11, 1508.

[11] a)Y. Pei, H. Wang, G. J. Snyder, *Adv. Mater.* 2012, 24, 6125; b)Y. Pei, X. Shi, A. LaLonde, H. Wang, L. Chen, G. J. Snyder, *Nature* 2011, 473, 66; c)C. M. Jaworski, V. Kulbachinskii, J. P. Heremans, *Phys. Rev. B* 2009, 80, 233201; d)M. Hong, Y. Wang, T. Feng, Q. Sun, S. Xu, S. Matsumura, S. T. Pantelides, J. Zou, Z. G. Chen, *J Am Chem Soc* 2019, 141, 1742.

[12] a)W. Liu, X. Tan, K. Yin, H. Liu, X. Tang, J. Shi, Q. Zhang, C. Uher, *Phys. Rev. Lett.* 2012, 108, 166601; b)Y. Dai, W. Zhou, H.-J. Kim, Q. Song, X. Qian, T.-H. Liu, R. Yang, *npj*

Computational Materials 2022, 8, 234.

[13] N. Wang, M. Li, H. Xiao, Z. Gao, Z. Liu, X. Zu, S. Li, L. Qiao, *npj Computational Materials* **2021**, 7, 18.

[14] a)S. Jaśkaniec, S. R. Kavanagh, J. Coelho, S. Ryan, C. Hobbs, A. Walsh, D. O. Scanlon, V. Nicolosi, *NPJ 2D Mater. Appl.* 2021, 5, 27; b)I. Maity, P. K. Maiti, M. Jain, *Phys. Rev. B* 2018, 97, 161406.

[15] S. R. Kavanagh, in doped Zenodo, 2023.

[16] W. Setyawan, S. Curtarolo, Comput. Mater. Sci. 2010, 49, 299.

[17] A. M. Ganose, A. Searle, A. Jain, S. M. Griffin, *Journal of Open Source Software* **2021**, 6, 3089.

[18] S.-H. Wei, A. Zunger, Applied Physics Letters 1998, 72, 2011.

[19] J. Nishitani, K. M. Yu, W. Walukiewicz, Applied Physics Letters 2014, 105, 132103.

[20] K. Brlec, K. B. Spooner, J. M. Skelton, D. O. Scanlon, *Journal of Materials Chemistry A* **2022**, 10, 16813.

[21] W.-J. Lee, H. J. Kim, J. Kang, D. H. Jang, T. H. Kim, J. H. Lee, K. H. Kim, *Annual Review of Materials Research* 2017, 47, 391.

[22] A. M. Ganose, J. Park, A. Faghaninia, R. Woods-Robinson, K. A. Persson, A. Jain, *Nat Commun* **2021**, 12, 2222.

[23] M. Cutler, J. F. Leavy, R. L. Fitzpatrick, Phys. Rev. 1964, 133, A1143.

[24] Z. M. Gibbs, F. Ricci, G. Li, H. Zhu, K. Persson, G. Ceder, G. Hautier, A. Jain, G. J. Snyder, *npj Computational Materials* **2017**, 3, 8.

[25] F. Q. Wang, S. Zhang, J. Yu, Q. Wang, Nanoscale 2015, 7, 15962.

[26] C. Wang, G. Ding, X. Wu, S. Wei, G. Gao, New J. Phys. 2018, 20, 123014.

[27] D. Zou, S. Xie, Y. Liu, J. Lin, J. Li, Journal of Materials Chemistry A 2013, 1, 8888.

[28] a)T. Yue, Y. Zhao, J. Ni, S. Meng, Z. Dai, npj Computational Materials 2023, 9, 17; b)S.-

Z. Huang, Q.-Y. Feng, B.-Y. Wang, H.-D. Yang, B. Li, X. Xiang, X.-T. Zu, H.-X. Deng, ACS Applied Energy Materials **2022**, 5, 13042.

[29] D. Kraemer, J. Sui, K. McEnaney, H. Zhao, Q. Jie, Z. F. Ren, G. Chen, *Energy & Environmental Science* 2015, 8, 1299.

[30] Q. Zhu, S. Song, H. Zhu, Z. Ren, Journal of Power Sources 2019, 414, 393.

[31] C. Hu, K. Xia, C. Fu, X. Zhao, T. Zhu, Energy & Environmental Science 2022, 15, 1406.

[32]a)Y. Zheng, T. J. Slade, L. Hu, X. Y. Tan, Y. Luo, Z. Z. Luo, J. Xu, Q. Yan, M. G. Kanatzidis, *Chem Soc Rev* **2021**, 50, 9022; b)P. Gorai, A. Ganose, A. Faghaninia, A. Jain, V. Stevanović, *Mater. Horiz.* **2020**, 7, 1809.

[33] a)G. Kresse, J. Furthmüller, *Phys. Rev. B* **1996**, 54, 11169; b)G. Kresse, J. Furthmüller, *Comput. Mater. Sci.* **1996**, 6, 15; c)G. Kresse, J. Hafner, *Phys Rev B Condens Matter* **1994**, 49, 14251; d)G. Kresse, J. Hafner, *Phys Rev B Condens Matter* **1993**, 47, 558.

[34] a)G. K. a. D. Joubert, *Phys. Rev. B* 1999, 59, 1758; b)P. E. Blochl, *Phys Rev B Condens Matter* 1994, 50, 17953.

[35] a)S. Grimme, S. Ehrlich, L. Goerigk, *J Comput Chem* **2011**, 32, 1456; b)S. Grimme, J. Antony, S. Ehrlich, H. Krieg, *J. Chem. Phys.* **2010**, 132, 154104; c)S. R. Kavanagh, in *vaspup2.0 Zenodo*, vaspup2.0 Zenodo 2023.

[36] a)A. V. Krukau, O. A. Vydrov, A. F. Izmaylov, G. E. Scuseria, J. Chem. Phys. 2006, 125, 224106; b)J. Heyd, G. E. Scuseria, M. Ernzerhof, *The Journal of Chemical Physics* 2003, 118,

8207.

[37] a)J. P. Perdew, K. Burke, M. Ernzerhof, *Phys. Rev. Lett.* **1996**, 77, 3865; b)J. P. Perdew, A. Ruzsinszky, G. I. Csonka, O. A. Vydrov, G. E. Scuseria, L. A. Constantin, X. Zhou, K. Burke, *Phys. Rev. Lett.* **2008**, 100, 136406.

[38] K. B. Spooner, A. M. Ganose, W. W. W. Leung, J. Buckeridge, B. A. D. Williamson, R. G. Palgrave, D. O. Scanlon, *Chem. Mater.* **2021**, 33, 7441.

[39] A. Togo, I. Tanaka, Scripta Materialia 2015, 108, 1.

[40] J. M. Skelton, D. Tiana, S. C. Parker, A. Togo, I. Tanaka, A. Walsh, *The Journal of Chemical Physics* **2015**, 143, 064710.

[41] A. Togo, L. Chaput, I. Tanaka, Phys. Rev. B 2015, 91, 094306.

[42] A. M. Ganose, A. J. Jackson, D. O. Scanlon, *Journal of Open Source Software* 2018, 3, 717.

[43] K. B. Spooner, M. Einhorn, D. W. Davies, D. O. Scanlon, in *ThermoParser: Streamlined Analysis of Thermoelectric Properties*, <u>https://github.com/SMTG-Bham/ThermoParser</u> 2023.



The Sc element significantly increases the conduction band degeneracy and thus the Seebeck coefficient in $Sc_2Si_2Te_6$ compared to $Sb_2Si_2Te_6$. As a result, a compelling ZT of 3.51 is achieved in n-type $Sc_2Si_2Te_6$, exceeding the optimal ZT of 2.76 in p-type $Sb_2Si_2Te_6$ 27 %.