

# 1 **Characterization of the EcoRecover Process for Intensive Microalgal** 2 **Cultivation and Tertiary Nutrient Recovery from Wastewaters**

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21 bioeconomy; microalgal-bacterial community; microalgae; biopolymer

## 22 **Abstract**

23 Mixed community microalgal wastewater treatment technologies have the potential to advance  
24 the limit of technology for biological nutrient recovery while producing a renewable carbon  
25 feedstock, but a deeper understanding of their performance is required for system optimization  
26 and control. In Fall 2020, a 568 m<sup>3</sup>·day<sup>-1</sup> (150,000 gal·day<sup>-1</sup>) Clearas EcoRecover microalgal  
27 treatment system was installed at the Village of Roberts (Wisconsin, USA; latitude 45.0°N)  
28 wastewater treatment plant to meet their pending effluent permit of 0.04 mg·L<sup>-1</sup> total phosphorus  
29 (TP; 6-month average). Here we report data and analyses from continuous on-line system  
30 monitoring, long-term on-site monitoring, and on-site batch experiments, with particular focus on  
31 a 3.5 month winter period with limited outside influences (e.g., no major upstream process  
32 changes). Across this period of intensive monitoring, effluent TP concentrations were consistently  
33 below 0.03 mg-P·L<sup>-1</sup> for 3 months. Core microbial community taxa included *Chlorella spp.*,  
34 *Scenedesmus spp.*, and *Monoraphidium spp.*, and key indicators of stable performance included  
35 near-neutral pH, sufficient alkalinity, and a diel rhythm in dissolved oxygen. By tracking elemental  
36 and biochemical composition of the biomass, we also demonstrated the importance of  
37 carbohydrate storage (in photobioreactors) and mobilization (in a dark mix tank) in achieving  
38 reliable nutrient recovery and algal growth.

39

## 40 **Synopsis Statement**

41 This study characterizes the performance of the first full-scale installation of the EcoRecover  
42 mixed community microalgal wastewater treatment process.

## 43 TOC Graphic



44

45

## 46 1. Introduction

47 The United States Environmental Protection Agency estimates that nutrients impair 15-41% of  
48 assessed surface water area (including lakes, rivers, estuaries, etc.) in the United States.<sup>1</sup>

49 Phosphorus, specifically, is the limiting nutrient for harmful algal growth and eutrophication in  
50 many freshwater ecosystems.<sup>2</sup> To protect or restore US waters, states are adopting numeric water  
51 quality criteria for nitrogen and phosphorus by identifying impaired water bodies and adjusting  
52 effluent permits for water resource recovery facility (WRRF) to meet waterbody-specific loadings.<sup>3</sup>

53 As of 2021, eight states had gained state-wide phosphorus criteria for at least one waterbody  
54 type, while another sixteen states had added numeric criteria for select waterbodies (headwaters,  
55 Wadeable streams, reservoirs requiring algicide, etc.).<sup>4</sup> To meet increasingly prevalent and  
56 increasingly stringent effluent phosphorus limits to protect natural waterbodies, wastewater  
57 treatment plants are in need of effective and cost-efficient technologies that reliably achieve  
58 phosphorus removal or recovery.

59 To date, commercialized tertiary wastewater treatment technologies for phosphorus  
60 management have been limited to enhanced biological phosphorus removal (EBPR), chemical  
61 polishing, and membranes.<sup>5</sup> EBPR can be a lower cost among these options but cannot reliably  
62 treat below 0.3 mg·L<sup>-1</sup> total phosphorus.<sup>6</sup> Though chemical polishing with coagulants (typically

63 aluminum sulfate or ferric chloride) can achieve more stringent effluent limits, significant addition  
64 of these chemicals generates large quantities of sludge that are difficult to treat, are expensive to  
65 landfill, and that represent a recalcitrant precipitate that make phosphorus recovery challenging.<sup>7</sup>  
66 Additionally, reliably achieving very low phosphorus effluent concentrations (e.g., <0.1 mg-P·L<sup>-1</sup>)<sup>8</sup>  
67 requires coagulant dosing that is significantly higher than predicted by stoichiometric quantities  
68 due to numerous side reactions, and the stoichiometric disparity increases substantially as target  
69 effluent phosphorus concentrations decrease.<sup>9</sup> Once precipitated, phosphorus-recovery from  
70 chemical polishing sludge requires chemical extraction and/or thermal approaches at very high  
71 temperatures (1,000 to 2,000 °C).<sup>10</sup> As an alternative to bacteria-driven luxury uptake in EBPR  
72 and to chemical polishing, microalgae can achieve phosphorus recovery – including organic  
73 phosphorus that is otherwise recalcitrant in conventional WRRFs<sup>11,12</sup> – through assimilation into  
74 (i.e. recovered as) biomass. If algal treatment systems can be engineered to reliably meet effluent  
75 nutrient permits, they have the potential to leverage waste phosphorus for CO<sub>2</sub> fixation and the  
76 production of renewable bioproducts and biofuels in support of a circular economy.<sup>13</sup>

77         The EcoRecover process is an intensive (i.e., high areal productivity, small footprint)  
78 tertiary nutrient recovery process which leverages the Advanced Biological Nutrient Recovery  
79 (ABNR™, Clearas Water Recovery Inc.)<sup>14</sup> system. The process consists of a dark mix tank,  
80 photobioreactors (PBRs), and the separation of the hydraulic retention time (HRT) and solids  
81 residence time (SRT) with membranes (**Figure 1**). The EcoRecover ABNR process was first  
82 piloted as a batch system, with minimal monitoring, at the South Davis Sanitary District (South  
83 Davis, Utah, USA) beginning in August 2016. In Fall 2021, the first full-scale installation began  
84 operation at the Village of Roberts (Wisconsin, USA), with robust monitoring and a design flow of  
85 568 m<sup>3</sup>·day<sup>-1</sup>. As of Fall 2023, a 1,100 m<sup>3</sup>·day<sup>-1</sup> EcoRecover system has been constructed in  
86 Mondovi (Wisconsin, USA) and a 10,600 m<sup>3</sup>·day<sup>-1</sup> system is operating in Waupun (Wisconsin,  
87 USA). To date, however, there is no publicly available data on the performance of the EcoRecover

88 process or documentation of its full-scale performance. Broad adoption of intensive (i.e., high  
89 productivity, small footprint) microalgal treatment technologies requires data transparency and an  
90 understanding of factors governing process performance to enable mechanistic system design  
91 and control.

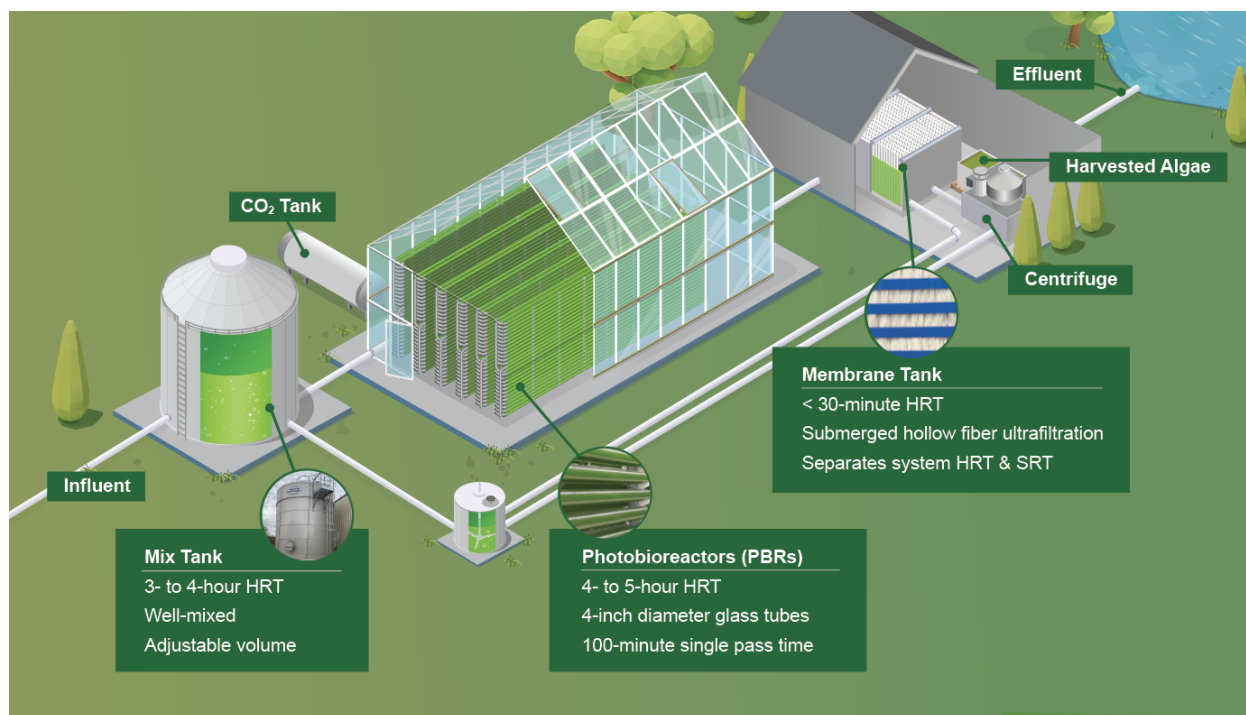
92 The objective of this work was to characterize the full-scale performance of the  
93 EcoRecover system as a tertiary treatment process for phosphorus removal and recovery from  
94 wastewater. The EcoRecover system was deployed as a tertiary treatment process to achieve  
95 effluent total phosphorus concentrations below new water quality-based permit limits of 0.12 mg-  
96 P·L<sup>-1</sup> (monthly) and 0.04 mg-P·L<sup>-1</sup> (6-month average) for the full forward (design) flow of 568  
97 m<sup>3</sup>·day<sup>-1</sup> (150,000 gal·day<sup>-1</sup>). Continuous, long-term monitoring was achieved through a network  
98 of sensors and analyzers that interfaced with a supervisory control and data acquisition system  
99 (SCADA) and were complemented by alternate-day elemental analysis of the solids and twice  
100 daily aqueous and total suspended solids (TSS) analyses during weekdays. These long-term  
101 monitoring data were supplemented with batch kinetic assays to better understand drivers of  
102 performance. Periods with effluent total phosphorus concentrations below the discharge limit were  
103 observed, stable performance indicators were identified, and carbon and nutrient dynamics within  
104 and across unit operations were characterized. Ultimately, a deeper understanding of the  
105 EcoRecover system will support further system optimization and control to advance the  
106 sustainability of microalgal wastewater treatment technologies and biological nutrient recovery.

## 107 **2. Materials and Methods**

### 108 **2.1. Full-scale Treatment System and Long-Term Operation**

109 The Roberts Wastewater Treatment Plant (WWTP; Village of Roberts, WI, USA) has an average  
110 influent flow of 410 m<sup>3</sup>·day<sup>-1</sup> and a design flow of 568 m<sup>3</sup>·day<sup>-1</sup> for a municipality of nearly 2,000  
111 residents.<sup>15</sup> The Wisconsin Department of Natural Resources decreased the Roberts WWTP's

112 Wisconsin Pollutant Discharge Elimination System's (WPDES No. 0028835) 6-month average  
113 effluent phosphorus limit from  $1 \text{ mg-P}\cdot\text{L}^{-1}$  to  $0.04 \text{ mg-P}\cdot\text{L}^{-1}$ , effective February 1, 2021, to protect  
114 and recover the water quality of the effluent receiving bodies, the East and West Twin Lakes.<sup>16</sup>  
115 The EcoRecover process, which the Village elected to implement for phosphorus-removal, was  
116 constructed in 2020 and 2021 (**Figure 1**, SI Figure S1). Secondary effluent from sequencing batch  
117 reactors (SBRs) is mixed with the microalgal community in a mix tank (average working volume  
118 of  $98 \text{ m}^3$  gal) before being sparged with  $\text{CO}_2$  and pumped through five parallel sets of PBRs ( $77.6$   
119  $\text{m}^3$  total). The PBRs are housed in a greenhouse and, in addition to daylight, receive an average  
120 of  $100 \mu\text{mol}\cdot\text{m}^{-2}\cdot\text{s}^{-1}$  photons from supplemental lighting from 54 LEDs (California Light Works  
121 MegaDrive™ Centralized Power LED Network). The separation of HRT and SRT is achieved via  
122 submerged, hollow fiber ultrafiltration modules in two parallel membrane tanks with average  
123 transmembrane pressure of  $15.4 \text{ kPa}$  (Puron ultrafiltration hollow fiber submerged membrane  
124 module, Model PHF960,  $3.8 \text{ m}^3$  working volume per train, 18 rows per module,  $0.03 \mu\text{m}$  pores,  
125 Koch Separation Solutions, Inc.). A fraction of the permeate is stored in an  $11.4\text{-m}^3$  reuse tank  
126 (averaged measured HRT of  $<10 \text{ min}$ ) for membrane backwashing while the remainder is  
127 discharged as effluent. Harvested solids are pumped from the membrane tank and dewatered via  
128 centrifugation (Disk Stack Clarifier AC1200-410, Flottweg Separation Technology, Independence,  
129 KT, USA). Centrate is mixed in a return tank ( $6.4 \text{ m}^3$ ; averaged measured HRT of  $20 \text{ min}$ ) with  
130 the PBR recycle flow and retentate from the membrane tanks before the combined flow returns  
131 to the mix tank.



132

133 **Figure 1.** EcoRecover process flow diagram. The mix tank receives secondary effluent and  
 134 recycled microalgal biomass under dark, nutrient-replete conditions. Inorganic carbon is sparged  
 135 into the mixed microbial community just prior to the PBRs where the biomass receives light, and  
 136 conditions become phosphorus-limited. Ultrafiltration in the membrane tank (membrane  
 137 bioreactor) separates the tertiary effluent from the biomass, which is either recycled back to the  
 138 mix tank or harvested.

139

140 The mix tank, a nutrient-replete, dark environment, operates as a completely stirred tank  
 141 reactor, is well-mixed through intermittent sparged aeration with resulting dissolved oxygen  
 142 generally below  $5 \text{ mg}\cdot\text{L}^{-1}$ . Mix tank effluent flows to the PBRs, where the microalgal community is  
 143 exposed to light, a nutrient-deplete environment, and a single-pass time of 100 min. The median  
 144 measured hydraulic retention times (HRTs) of the mix tank and PBRs are 3.3 and 4.3 hrs,  
 145 respectively. Batch, bench-scale kinetic experiments were conducted for durations that exceeded

146 the single pass times of the full-scale unit processes to better characterize the kinetics and  
147 stoichiometry of nutrient uptake and carbon storage and mobilization.

148 For this study we focused on monitoring data from November 1, 2022 to February 14,  
149 2023, which represented an extensively sampled period with limited outside influences (e.g., no  
150 upstream plant changes, no known chemical perturbation events). Examples of periods with  
151 significant outside influence are presented in Section 3.1.3. and SI Section S6 for transparency  
152 but are not the focus of this study.

## 153 **2.2. Bench-Scale Batch Experiments**

154 Bench-scale, batch experiments were conducted to characterize carbon and nutrient dynamics in  
155 the full-scale mix tank and PBRs. Duplicate batch experiments were conducted in cylindrical,  
156 bench-scale PBRs constructed from clear cast acrylic with the same diameter as the full-scale  
157 system (102 mm i.d., 91 cm height, 7 L working volume). The bench-scale PBRs were placed in  
158 the greenhouse with the full-scale PBRs to match light and temperature conditions. A third PBR  
159 (102 mm i.d., 69 cm height, 5 L working volume) was run in parallel, without being sampled, to  
160 ensure sufficient solids for the subsequent mix tank bench-scale experiment. Bench-scale mix  
161 tank experiments were conducted in duplicate under dark conditions in opaque, HDPE plastic  
162 containers (4-L working volume) with lids.

163 Bench-scale experiments were inoculated with biomass and process flows taken directly  
164 from the full-scale system immediately before initiation of the batch experiments. The bench-scale  
165 PBR experiments were conducted using effluent from the full-scale mix tank (20 L). To ensure  
166 the biomass in the mix tank batch experiments had adequate stored carbon (at the start of the  
167 experiment) to observe organic carbon mobilization, the mix tank experiments were conducted  
168 using secondary effluent combined with the biomass from the bench-scale PBR that was not  
169 sampled (2.2 L and 5.8 L, respectively, to match the mixing ratio in the full-scale mix tank).



170 Reactors were continuously mixed (magnetic stirrer; 300 rpm) and sampled with wide-bore 50 mL  
171 serological pipets at 0, 10, 20, 40, 60, 90, 120, 150, 180, and 240 min (PBRs were also sampled  
172 at 300 min and 360 min). The aqueous fraction of samples was immediately separated from the  
173 solids through centrifugation at 4,200 x g for 5 min at 4 °C (5804R Eppendorf centrifuge; Enfield,  
174 CT, USA) and then filtered through 0.22 µm (MF-Millipore™ Membrane Filter, 0.22 µm, item no.  
175 GSWP02500; MilliporeSigma). The solids pellet and filtered aqueous samples were stored  
176 separately at -20 °C prior to lyophilization (solids samples) and analysis (solids and aqueous  
177 samples). TSS and volatile suspended solids (VSS) were quantified at 0, 120, and 240 min (as  
178 well as 360 min for the PBRs). The reactor pH was maintained between 6.8 and 7.5 to avoid pH  
179 inhibition; adjustments were accomplished with 2 M HCl. The alkalinity of PBR samples –  
180 determined via titration of 100 mL samples to pH 4.5 (Mettler Toledo DL55 titrator) – was initially  
181 600 mg·L<sup>-1</sup> as CaCO<sub>3</sub> and maintained above 200 mg·L<sup>-1</sup> as CaCO<sub>3</sub> through NaHCO<sub>3</sub> addition to  
182 avoid carbon limitation.

### 183 **2.3. Continuous On-Line Monitoring**

184 For the continuously operating full-scale system, long-term monitoring was achieved through  
185 online sensors and analyzers for pH, dissolved oxygen (DO), TSS, PO<sub>4</sub><sup>3-</sup>, NH<sub>4</sub><sup>+</sup>, NO<sub>3</sub><sup>-</sup>, turbidity,  
186 temperature, and photosynthetically active radiation (PAR; Table S1 and Figure S2 in the SI),  
187 which interfaced with a SCADA system. Hydraulic parameters, including flowrates and tank  
188 volumes, were also collected through on-line monitoring. Most sensors were on-line by late  
189 November 2020. Following the International Water Association Good Modeling Practice Unified  
190 Protocol,<sup>17</sup> the long-term continuous online monitoring data were reconciled to ensure that  
191 systematic errors (e.g., shifts or drifts) in the data set were identified and resolved using the kernel  
192 smoothing method of a Python package for functional data analysis (scikit-fda).<sup>18</sup> In particular,  
193 pH, TSS, PO<sub>4</sub><sup>3-</sup>, NH<sub>4</sub><sup>+</sup>, and NO<sub>3</sub><sup>-</sup> were additionally corrected to match the magnitude of the daily

194 onsite laboratory measurement data (SI\_SCADA and SI\_AIMS spreadsheets, Supporting  
195 Information, SI).<sup>19</sup>

## 196 **2.4. Aqueous and Suspended Solids Analyses**

### 197 **2.4.1. Long-Term On-Site Monitoring**

198 Beginning in December 2021, long-term continuous on-line monitoring was supplemented by  
199 analyses of once to twice daily grab and 24-hour composite samples from the full-scale system.  
200 Aqueous parameters were measured with Hach kits after samples were filtered through 0.45  $\mu\text{m}$   
201 mixed cellulose ester filters. Specifically, aqueous samples were analyzed for orthophosphate  
202 and total phosphate (Hach TNT843); alkalinity (TNT870); nitrate (TNT835 or TNT836; dependent  
203 on sample concentration range); ammonium (TNT830, TNT831, or TNT832; dependent on  
204 sample concentration range); nitrite (TNT839 and TNT840); and total nitrogen (TNT827). The  
205 method detection limit (MDL) and minimum reporting level (MRL) for total phosphorus and  
206 orthophosphate were estimated according to Ripp 1996,<sup>20</sup> and were, respectively, found to be  
207 0.005 and 0.005  $\text{mg-P}\cdot\text{L}^{-1}$  (MDL) and 0.014 and 0.016  $\text{mg-P}\cdot\text{L}^{-1}$  (MRL; SI Table S2). Briefly, 9  
208 replicates of the same concentration were analyzed using TNT843, and MDL was defined as the  
209 product of the t-value for n-1 samples ( $t = 2.896$ ) and the sample standard deviation of those  
210 replicates. MRL was defined as three times the value of the MDL.

### 211 **2.4.2. Batch, Bench-Scale Analyses**

212 Batch experiment samples (solids and aqueous) were analyzed both on-site and at the University  
213 of Illinois Urbana-Champaign (UIUC). Handheld probes were used to measure pH (Orion 3-Star  
214 portable pH meter; Thermo Scientific), temperature and DO (Orion RDO dissolved oxygen probe;  
215 Thermo Scientific), and ammonium concentrations (ProDSS multiparameter digital water quality  
216 meter; YSI); each sensor was calibrated immediately prior to use in the bench-scale experiments.  
217 Solids storage and analyses for TSS and VSS were performed as in Bradley et al. 2021.<sup>21-25</sup>

218 Briefly, sample TSS was determined by filtration through 0.7  $\mu\text{m}$  glass fiber filters (Whatman GF/F).  
219 After filtration, filters were heated at 105  $^{\circ}\text{C}$  for 1 h and desiccated for 30 min prior to weighing.  
220 VSS was determined by combusting samples for 20 min at 550  $^{\circ}\text{C}$ .

221 Samples for phosphate, nitrate, and nitrite were immediately filtered with 0.22  $\mu\text{m}$  filters  
222 and frozen. After storage, aqueous samples were thawed and re-filtered prior to analysis via ion  
223 chromatography (Dionex ICS-2100 ion chromatograph, Dionex IonPac AS18 column; SI Section  
224 S2 Figures S3-S5 for calibration curves). The average MRL for phosphate was determined to be  
225 0.027  $\text{mg}\cdot\text{P}\cdot\text{L}^{-1}$  and the MRL range was 0.022 to 0.037  $\text{mg}\cdot\text{P}\cdot\text{L}^{-1}$  (SI Section S2, Table S3).

## 226 **2.5. Solids Characterization**

### 227 **2.5.1. Elemental Composition**

228 For the elemental analysis of biomass, a solids pellet was collected through centrifugation of a  
229 culture sample, immediately frozen, and then lyophilized for 48 hrs. Phosphorus content was  
230 measured through inductively coupled plasma mass spectrometry (ICP-MS; Model NexION 350D,  
231 PerkinElmer) and elemental carbon, nitrogen, and hydrogen were measured with a CHN Analyzer  
232 (Model CE440, Exeter Analytical) by the UIUC Microanalysis Laboratory. If replicate results had  
233 a greater than 5% difference, the sample was reanalyzed and the results were replaced.

### 234 **2.5.2. SEM-EDS Biomass Surface Characterization**

235 The surface of lyophilized and ground solids samples were characterized using Scanning Electron  
236 Microscopy-Energy Dispersive Spectroscopy (SEM-EDS)<sup>26-28</sup> in the Microscopy Suite at the  
237 Beckman Institute for Advanced Science and Technology at UIUC. Before imaging, the samples  
238 were mounted on a stub using carbon tape. The samples were imaged using a SEM (Model  
239 Quanta FEG 450, FEI company) operating at 15.0 kV under the low vacuum mode and at a  
240 working distance of 10 mm. The elemental compositions were measured using the EDAX light-  
241 element energy-dispersive spectroscopy system (AMETEK, Inc.) attached to the SEM.

### 242 **2.5.3. Carbohydrate, Protein, and Lipid Quantification**

243 Solids storage and analyses for protein-to-N ratio and carbohydrate content were performed as  
244 in Bradley et al. 2021.<sup>21–25</sup> Briefly, protein content was estimated by multiplying the elemental  
245 nitrogen content by a conversion factor that represents the ratio of N content to  
246 protein. Conversion factors were determined by analysis of amino acid residuals; amino acid  
247 profiling was performed by Bio-Synthesis, Inc. (Lewisville, Texas, USA). Total monomeric  
248 carbohydrate content of lyophilized solids was determined after a two-step acid hydrolysis of the  
249 complete biomass. The hydrolysate was neutralized, filtered, and the monosaccharide  
250 concentration was quantified against glucose standards. Solids crude lipid content was quantified  
251 as in Gardner-Dale et al.<sup>24,29,30</sup> Briefly, crude lipids from lyophilized solids were extracted using an  
252 adaptation of the Folch method and a 2:1 (v/v) chloroform:methanol solvent mixture. After the  
253 extraction, sodium chloride solution was added to bring the final mixture to a 8:4:3 ratio of  
254 chloroform:methanol:sodium chloride. The mixture was centrifuged, resulting in a biphasic  
255 system; the bottom phase containing the crude lipids was transferred to weighting dishes to be  
256 measured gravimetrically after the carrier solvent evaporated.

### 257 **2.5.4. Flow Imaging Microscopy**

258 Mix tank effluent samples (10 mL in a 15 mL conical tube) were collected once daily. Samples  
259 were diluted to approximately  $1 \times 10^6$  particles per milliliter prior to running on a FlowCam 5000  
260 flow imaging microscope (Yokogawa Fluid Imaging Technologies, Inc.). The resulting collection  
261 of detected particles was screened to remove background objects, then used as input data for a  
262 deep learning classification model trained on representative libraries of the dominant taxonomic  
263 groups observed in the system. Details are in SI Section S5.

## 264 **2.5.5. High Throughput 18S rRNA Sequencing**

265 1-mL samples of suspended biomass from the mix tank and PBR effluent were collected in  
266 triplicate and stored in 5-mL polypropylene transport tubes filled with 3 mL Zymo DNA/RNA  
267 Shield. Samples were kept at -20°C until they were shipped overnight on ice to the University at  
268 Buffalo (UB) SUNY. DNA extraction was performed using the DNeasy Powersoil Pro Kit (Qiagen),  
269 and extracts were stored at -20°C. PCR amplification of the eukaryotic 18S rRNA genes targeted  
270 the V8-V9 region (details in Bradley et al., 2016).<sup>31</sup> Gel electrophoresis was conducted post PCR,  
271 and bands of expected size and quality were purified using the QIAquick gel purification kit  
272 (Qiagen). The purified amplicons from each sample were pooled into a DNA library at equimolar  
273 proportions (10 ng). Sequencing was performed on the Illumina MiSeq platform with v3 chemistry  
274 (300-cycle paired-end reads) at the UB Genomics and Bioinformatics Core. Raw sequencing data  
275 are available on NCBI under BioProject accession number PRJNA1045645. The sequencing read  
276 processing (i.e., quality filtering and trimming, taxonomic assignment) and statistical analyses  
277 (including alpha and beta diversity) were conducted following the established protocol (MiSeq  
278 SOP) provided by mothur v1.48.0.<sup>32</sup>

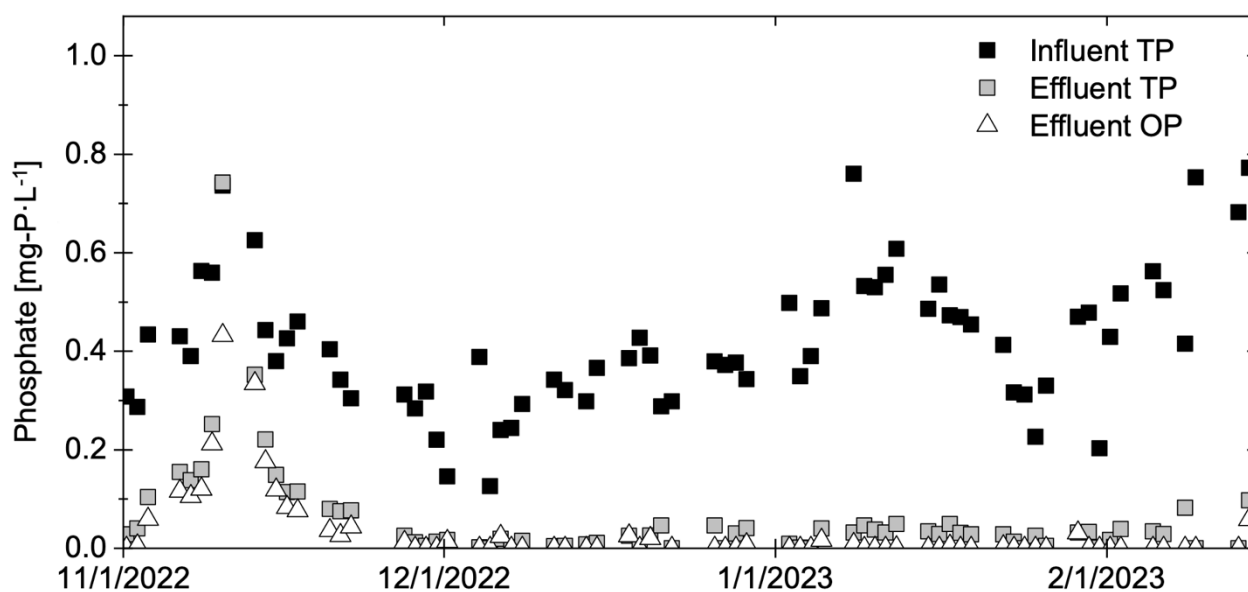
## 279 **3. Results and Discussion**

### 280 **3.1. Characterization of Long-Term Performance**

#### 281 **3.1.1. Achieving Non-Detect Effluent Phosphorus and Identifying Key Indicators of Stable** 282 **System Performance**

283 The first full-scale installation of the EcoRecover process at Roberts, Wisconsin has  
284 demonstrated phosphorus recovery from secondary effluent via microalgal biomass cultivation  
285 24-hours per day and across seasons. Periods of performance with effluent total phosphorus  
286 concentrations below the discharge limit ( $<0.04 \text{ mg-P}\cdot\text{L}^{-1}$ ) were observed for sustained periods  
287 (months at a time), frequently reaching non-detect values.

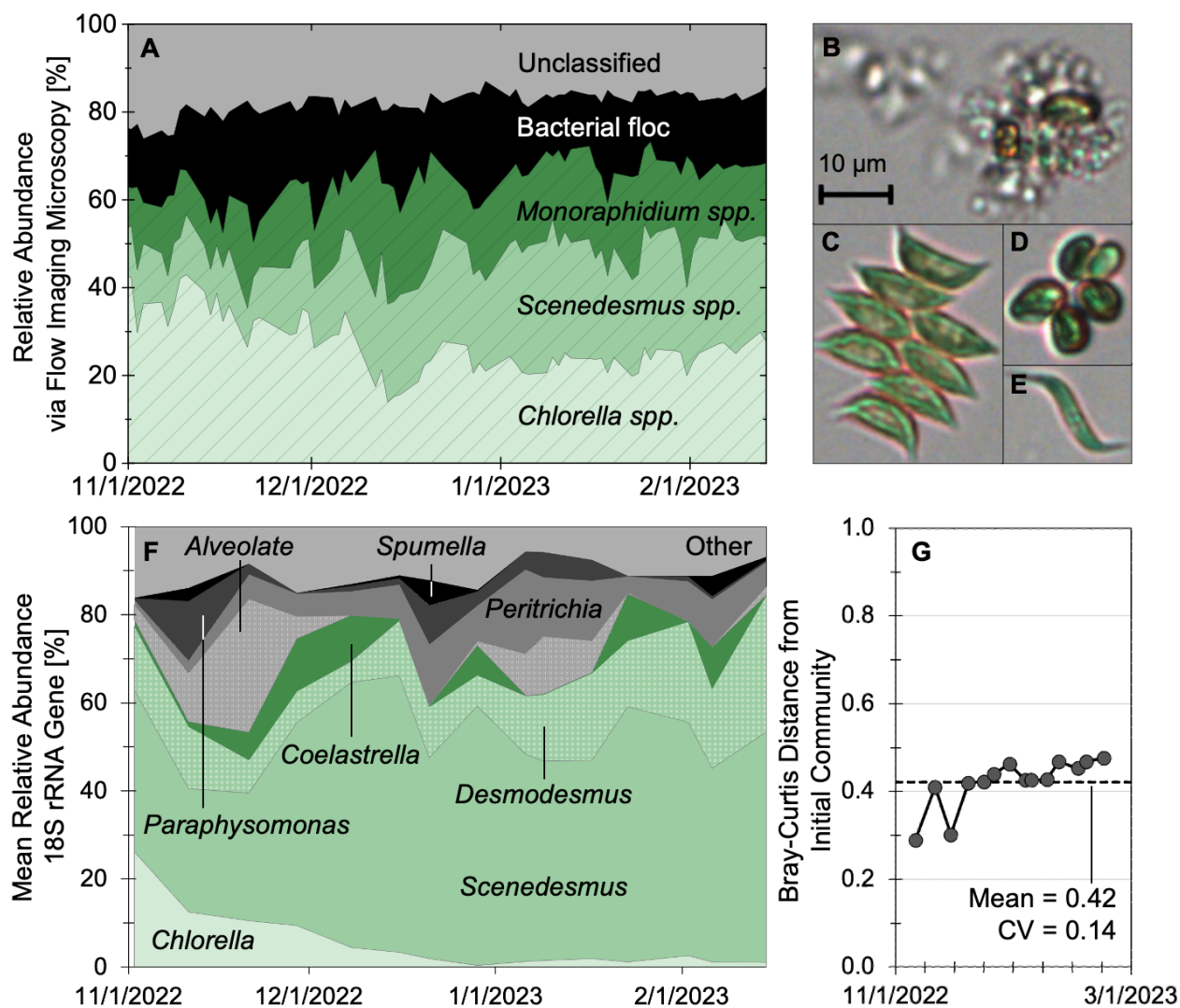
288 The focus period (November 1, 2022 through February 14, 2023; 106 days) began with a  
289 two week upset and recovery period, followed by 92 days (November 15, 2022 through February  
290 14, 2023) of superior performance in which the system continuously achieved effluent (permeate)  
291 orthophosphate concentrations below  $0.04 \text{ mg-P}\cdot\text{L}^{-1}$  (**Figure 2**, SI Figure S6). Across the full focus  
292 period, the effluent total phosphorus concentration of 24-hour composite samples averaged  $0.06$   
293  $\pm 0.11 \text{ mg-P}\cdot\text{L}^{-1}$  ( $0.03 \pm 0.08 \text{ mg-P}\cdot\text{L}^{-1}$  orthophosphate; average  $\pm$  standard deviation); within the  
294 92-day period of excellent performance, the effluent total phosphorus concentration averaged  
295  $0.03 \pm 0.03 \text{ mg-P}\cdot\text{L}^{-1}$  ( $0.01 \pm 0.02 \text{ mg-P}\cdot\text{L}^{-1}$  orthophosphate). Effluent ammonia concentrations  
296 were highly variable (Figure S7); some nitrification and minimal nitrite concentrations were  
297 observed from November 1 to December 8, 2022. (SI Figures S8 and S9). From December 8,  
298 2022 to February 14, 2023, nitrification was negligible.



300 **Figure 2.** Influent and permeate total phosphorus (TP) concentrations and permeate  
301 orthophosphate ( $\text{PO}_4^{3-}$ ) concentrations for the EcoRecover process in winter from November 1,  
302 2022, to February 14, 2023. 24-hour composite samples were collected and immediately  
303 analyzed by an on-site laboratory technician (EH).

304

305           Across the entire focus period, the algal community was relatively stable. Flow imaging  
306 microscopy identified *Chlorella spp.*, *Scenedesmus spp.*, and *Monoraphidium spp.* as dominant  
307 constituents of the microalgal community (**Figure 3A-E**). High-throughput sequencing of 18S  
308 rRNA genes confirmed the eukaryotic community was dominated by green microalgae, including  
309 *Scenedesmus* (28% – 63%), *Desmodesmus* (5% – 31%), and *Chlorella* (0.5% – 26.5%; **Figure**  
310 **3F**). The eukaryotic community remained stable during this period (**Figures 3G**), as indicated by  
311 a consistent Bray-Curtis dissimilarity among samples across the intensive sampling period from  
312 the starting community on November 2, 2022 (mean = 0.42, std. dev. = 0.06, and coeff. of variation  
313 = 0.14; **Figure 3G**). Additionally, when compared to a variable performance period (February 15,  
314 2023 to April 28, 2023) immediately following the focus period, eukaryotic communities in these  
315 two periods showed significantly different clusters ( $p < 0.01$ , AMOVA)<sup>33</sup> as illustrated by ellipses  
316 encompassing 95% of cluster assigned datapoints (SI Figure S10). Full, longer-term sequencing  
317 results and more in-depth community structure analyses are the focus of a separate study.



318  
 319 **Figure 3.** (A) Mixed microbial community taxa distribution for the EcoRecover process from  
 320 November 1, 2022 through February 14, 2023. (B-E) FlowCam images of the four dominant  
 321 categories identified via flow imaging microscopy during this period: (B) bacterial floc; (C)  
 322 *Scenedesmus spp.*; (D) *Chlorella spp.*; and (E) *Monoraphidium spp.* (F) Mean relative  
 323 abundance (MRA) of the top eukaryotic genera from 18S rRNA sequencing. (G) Bray-Curtis  
 324 distances of each sampling point from the initial eukaryotic community on November 2, 2022.

325  
 326 Key indicators of stable performance included near-neutral pH throughout the system,  
 327 sufficient alkalinity in the EcoRecover influent (>200 mg·L<sup>-1</sup> as CaCO<sub>3</sub>) and residual alkalinity in



328 the EcoRecover effluent ( $>100 \text{ mg}\cdot\text{L}^{-1}$  as  $\text{CaCO}_3$ ), and a steady daily rhythm in the DO  
329 concentration of both the mix tank effluent and PBR effluent (Figure S11). In particular, the cycling  
330 of DO concentration in the PBRs is readily apparent (often reaching  $10\text{-}20 \text{ mg}\cdot\text{L}^{-1}$ ), corresponds  
331 to the photosynthetic activity of the microalgal community, and generally follows the rhythm of  
332 natural light intensity (SI Figure S12).

### 333 **3.1.2 Solids Composition and Yield**

334 Across the full focus period, the average biomass yield was  $33 \pm 8 \text{ kg TSS}\cdot(\text{kg-P})^{-1}$ , corresponding  
335 to a phosphorus content of  $3.2 \pm 0.7\%$  ( $5^{\text{th}}/50^{\text{th}}/95^{\text{th}}$  percentiles of  $1.92\%/3.56\%/4.40\%$ ;  $n = 32$ ) in  
336 harvested biomass (SI Figure S13). The phosphorus content in algal biomass has been shown to  
337 vary across species<sup>34,35</sup> and also within species as a function of their physiological state.<sup>36–40</sup> The  
338 phosphorus content in microalgal cells is often around 1% of dry weight under limited phosphorus  
339 availability<sup>41</sup> but may be as high as 4–6% dry weight, inclusive of microalgae achieving luxury  
340 uptake of phosphorus.<sup>42–44</sup> Given that the phosphorus content of the harvested biomass is  
341 inversely proportional to the total mass harvested and revenue from biomass sales, harvested  
342 biomass was further analyzed to better understand the distribution of phosphorus in the solids.

343 Elevated phosphorus in the biomass may have resulted from both luxury uptake and  
344 precipitation within EPS. SEM-EDS was performed on a range of samples from across the entire  
345 monitoring period (April 5, 2022 to February 13, 2023; SI Section S4.2 Table S5), including  
346 samples with the highest and lowest phosphorus content, from periods with and without coagulant  
347 use, and start and end points of batch experiments. Of selected samples for SEM-EDS imaging,  
348 the February 13, 2023, sample had the greatest overall phosphorus content (5.08%).  
349 Phosphorus-rich granules were observed in biomass samples and appear to be closely  
350 associated with the microalgal cells, indicating that elevated phosphorus could be due to  
351 precipitation of inorganic phosphorus with metals in extracellular polymeric substances (EPS), as  
352 is commonly observed in EBPR biomass (Figure S13).<sup>26,28</sup> Since the dominant taxa at this date

353 was *Scenedesmus* spp. (**Figure 3**) – which is capable of polyphosphate accumulation – it is  
354 possible the granules were associated with luxury uptake and release, which can induce  
355 precipitation within EPS.<sup>38</sup> High pH at the surface of microalgal cells is also expected due to  
356 inorganic carbon fixation, which may have facilitated inorganic phosphorus precipitation on the  
357 cell surfaces or within cell EPS even at low total phosphorus concentrations.<sup>45–47</sup> Beginning in  
358 December 2021, alum (aluminum sulfate liquid #41817; Hawkins, Inc.; Roseville, Minnesota) or  
359 Aqua Hawk<sup>®</sup> 15047 (an inorganic coagulant/polymerized aluminum chlorosulfate solution;  
360 Hawkins, Inc.; Roseville, Minnesota) were frequently used at the headworks of the Roberts  
361 WWTP to manage influent quaternary ammonium (pandemic disinfectant and microalgal inhibitor)  
362 and to attenuate excess influent phosphorus concentrations (i.e., concentrations >0.6 mg-P·L<sup>-1</sup>).  
363 During periods of coagulant use, which frequently corresponded to the highest bulk solids  
364 phosphorus content, the imaged solids had elevated metal content (Al, Ca, Mg) and granules of  
365 localized elevated phosphorus and metals (SI Figure S14). In periods without coagulant use,  
366 inorganic phosphorus granules were not readily observed (SI Figure S15), but the imaged cells  
367 had a higher concentration of phosphorus relative to that of imaged cells during periods of  
368 coagulant use (SI Figure S6). In either case (with or without coagulant addition), bulk liquid  
369 equilibrium modeling would not have predicted phosphorus precipitation; an accurate prediction  
370 of phosphorus distribution (including the formation of phosphorus- and metal-rich granules) would  
371 require more detailed analysis of the fate of organic and inorganic phosphorus in algae biomass  
372 to explicitly characterize mechanism of phosphorus accumulation, which is beyond the scope of  
373 this study.

374 The nitrogen content of solids was less variable than that of phosphorus. Across the focus  
375 period, the average biomass yield was  $12.8 \pm 0.7$  kg TSS·(kg-N)<sup>-1</sup>, corresponding to a solids  
376 nitrogen content of  $7.8 \pm 0.4\%$  (average  $\pm$  standard deviation; 5<sup>th</sup>/50<sup>th</sup>/95<sup>th</sup> percentiles of

377 7.10%/7.82%/8.49%). Similar to phosphorus, nitrogen content of microalgae can vary, with  
378 reported nitrogen content ranging from 5.87% to 11.16% of total solids.<sup>48,49</sup>

379 N:P mass ratios across the focus period were  $2.5 \pm 0.6$  (SI Figure S16) but were notably  
380 higher prior to January 8, 2023 ( $3.3 \pm 0.3$ , with phosphorus content of  $2.4 \pm 0.2\%$ ) than after ( $2.1$   
381  $\pm 0.2$ ; with phosphorus content of  $3.7 \pm 0.4\%$ ). The nitrogen content remained relatively stable at  
382  $7.8 \pm 0.5\%$  to  $7.9 \pm 0.4\%$  for the periods before and after January 8, 2023, respectively. Solids  
383 N:P ratio may vary with growth rate (linked to SRT) and influent N:P ratio through interspecific  
384 stoichiometric plasticity (especially under nutrient-limitation) or microbial community composition  
385 shifts in favor of species that have a competitive advantage in a given set of environmental  
386 conditions.<sup>24,40,50</sup> The relationship between influent N:P and solids N:P was not significant (linear  
387  $R^2 < 0.001$ ,  $n = 12$ ), However, solids N:P had a negative linear correlation with SRT ( $R^2 = 0.63$ ,  
388 SI Figure S17) during the period of superior performance (November 15, 2022 to February 14,  
389 2023), varying from 2.1 to 4.2 days. In addition to SRT, the harvested solids N:P ratio of may also  
390 have been influenced by the fate of recovered phosphorus, which varied between assimilation  
391 within the cells and highly localized precipitation on the cell surfaces or within the EPS during  
392 upstream coagulant use.

### 393 **3.1.3. Susceptibility to Process Upsets**

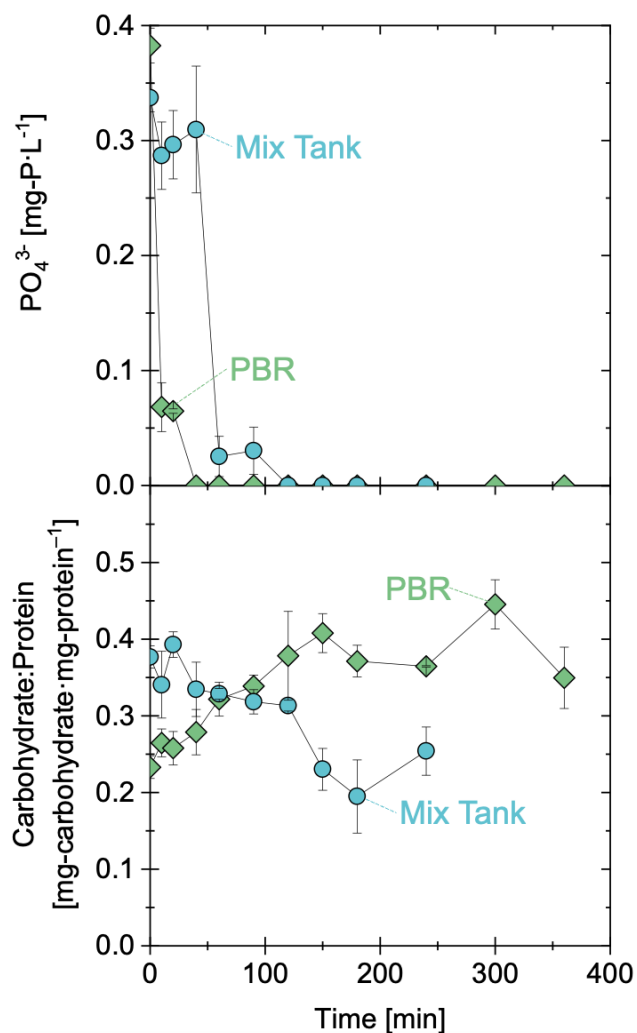
394 Periods of process upset or variable performance were defined as periods of prolonged or  
395 intermittent exceedance of the  $0.04 \text{ mg-P}\cdot\text{L}^{-1}$  effluent target, respectively. The system has been  
396 susceptible to process upsets driven by upstream changes to the wastewater treatment process  
397 (e.g., reduction in SBR settling time that resulted in TSS concentrations  $>100 \text{ mg-TSS}\cdot\text{L}^{-1}$  entering  
398 the EcoRecover process) and high influent concentrations of a disinfectant, quaternary  
399 ammonium. Periods of process upset or variable performance were often characterized by  
400 variable or basic pH, loss of DO diel rhythm, and at times included insufficient alkalinity and solids  
401 composition changes (details and images in SI Section S6 and Figure S18).

### 402 3.2. Carbon and Nutrient Dynamics across Unit Operations

403 Past work has demonstrated that carbohydrate storage and mobilization is important for the  
404 uptake of nutrients across light/dark cycles.<sup>24,25</sup> Specifically, exposure to lit, nutrient limited  
405 conditions can induce the storage of biopolymers which can then be leveraged under dark  
406 conditions to support nutrient uptake and continued metabolic activity. Thus, it was expected the  
407 storage and mobilization of carbohydrates may play a key role in the EcoRecover process. To  
408 better understand these dynamics, batch experiments were conducted on-site under mix tank  
409 (dark) and PBR (illuminated) conditions in May 2022 (prior examining the extant, full-scale carbon  
410 and nutrient dynamics) to characterize orthophosphate (**Figure 4**), ammonium (Figure S19),  
411 nitrate/nitrite (Figure S20), and biomass compositional (Figures S21, S22, S24-26) and  
412 concentration (Figure S23) trends. In the simulated mix tank (performed with simulated PBR  
413 culture mixed with EcoRecover influent), the initial orthophosphate of  $0.337 \pm 0.012 \text{ mg-P}\cdot\text{L}^{-1}$  was  
414 removed within 120 min (**Figure 4A**) and the initial ammonium concentration of  $14 \pm 2 \text{ mg-N}\cdot\text{L}^{-1}$   
415 was reduced to  $8.90 \pm 0.01 \text{ mg-N}\cdot\text{L}^{-1}$  over the course of 240 min (Figure S19). In the simulated  
416 PBR (performed with EcoRecover mix tank effluent), the initial orthophosphate of  $0.383 \pm 0.015$   
417  $\text{mg-P}\cdot\text{L}^{-1}$  was removed within 40 min (**Figure 4A**) and the initial ammonium concentration of  $36 \pm$   
418  $4 \text{ mg-N}\cdot\text{L}^{-1}$  was reduced to  $5.24 \pm 0.13 \text{ mg-N}\cdot\text{L}^{-1}$  over 360 min (Figure S19).

419 As expected, phosphorus limitation and the availability of light in the simulated PBR  
420 resulted in carbohydrate storage and an increase in the carbohydrate:protein ratio of solids and  
421 dark conditions in the simulated mix tank resulted in carbohydrate mobilization and a decrease in  
422 solids carbohydrate:protein ratio over time (**Figure 4B**). Consistent with this observation, the  
423 solids C:N and C:P mass ratios increased in the PBR and decreased in the mix tank across the  
424 bench-scale, batch experiments (SI Figures S21 and S22, respectively). PBR solids  
425 concentrations (quantified as VSS) increased from  $470 \pm 40$  to  $700 \pm 20 \text{ mg}\cdot\text{L}^{-1}$  over 360 minutes,  
426 whereas mix tank VSS concentrations did not change significantly (two-sample t-test, two-tailed,

427 p = 0.876; SI Figure S23). Oxygen was produced under PBR conditions and consumed under mix  
428 tank conditions (SI Figure S24). No significant change in the solids N:P or lipids content was  
429 demonstrated over the course of the experiments (SI Figures S25 and S26, respectively). These  
430 bench-scale results confirmed that (i) photosynthesis in the PBRs supports the uptake of residual  
431 phosphorus, the accumulation carbohydrates in cell biomass, and the production of oxygen, and  
432 (ii) a significant amount of phosphorus removal occurs in dark, nutrient-rich conditions (i.e., the  
433 mix tank) and is facilitated by consumption of dissolved oxygen and the mobilization of stored  
434 carbohydrates.



435

436 **Figure 4.** (A) Orthophosphate and (B) carbohydrate:protein ratios in bench-scale batch

437 experiments to mimic conditions in the mix tank (blue circles) and PBRs (green diamonds).

438 Batch experiments were run in duplicate. The duration of the experiments was at least twice the

439 HRT of the full-scale unit processes. Mix tank experiments were carried out with bench-scale

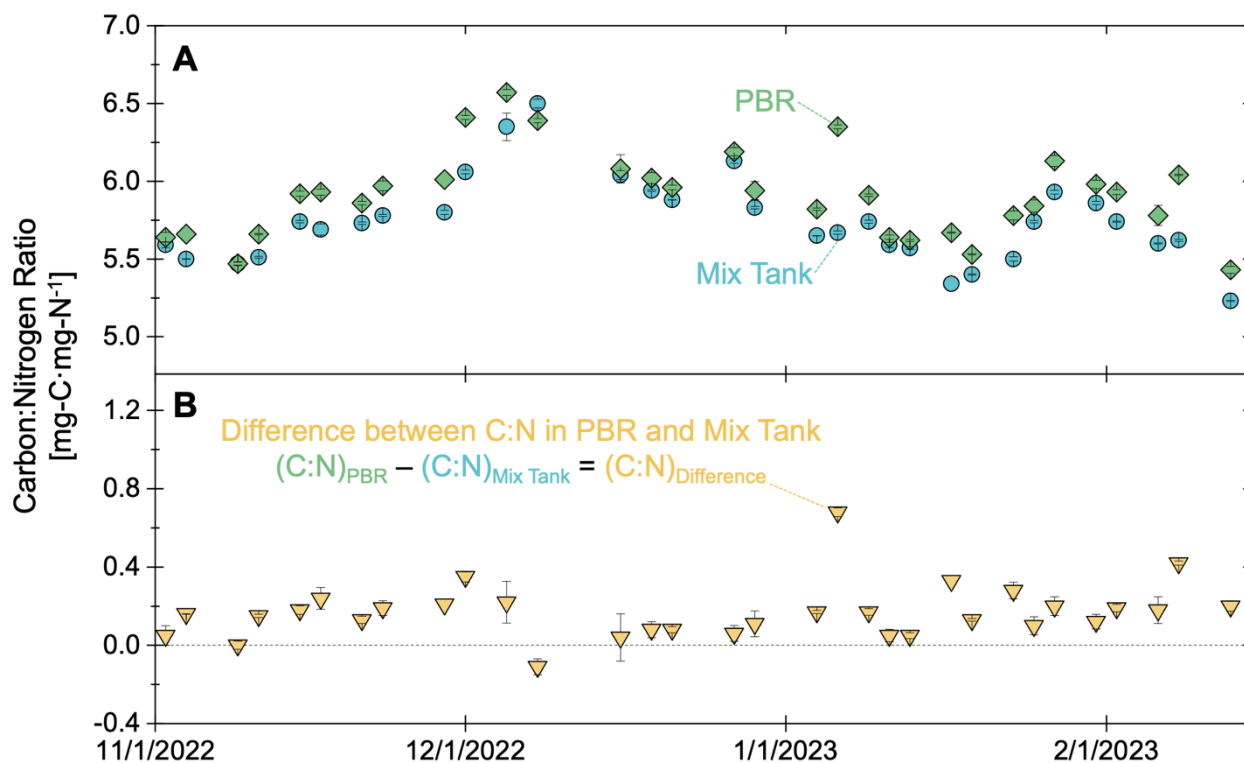
440 PBR culture combined with full-scale secondary effluent. PBR experiments were carried out with

441 full-scale mix tank culture. Symbols represent averages, error bars extending to individual

442 replicate (i.e., minimum and maximum) values.

443

444 In the full-scale system, similar trends were observed with higher biomass C:N ratios (and  
445 carbohydrate:protein ratios) in the PBR effluent than the mix tank effluent during periods of good  
446 performance (**Figure 5**, SI Figure S27). The consistency of higher C:N ratios in biomass leaving  
447 the PBRs relative to the biomass leaving the mix tank further underscores the importance of  
448 stored carbohydrates in nutrient recovery. To further elucidate the relationship between  
449 carbohydrate dynamics and nutrient recovery, the mobilized carbohydrates (consumed in the dark  
450 environment of the mix tank) were normalized to the quantity of phosphorus recovered from the  
451 system. Across the full focus period, the average amount of carbohydrate mobilized per  
452 phosphorus recovered was 19 g-carbohydrate·g-P<sup>-1</sup> (n = 30), which was approximately half the  
453 value in past bench-scale P-limited experiments.<sup>24</sup> Although orthophosphate was measured  
454 online in the system influent and effluent, it was not measured in the mix tank effluent; therefore,  
455 the carbohydrate and C:N dynamics relative to phosphorus uptake could not be quantified, and  
456 may be a focus of future work to continue to inform the design and optimization of the EcoRecover  
457 process. Overall, the storage and mobilization of carbohydrates enabled continuous nutrient  
458 removal in the mix tank with residual polishing in the PBRs.



459  
 460 **Figure 5.** (A) Solids C:N ratio (by mass) in the full-scale mix tank (blue circles) and PBR (green  
 461 diamonds) effluent from November 1, 2022 to February 14, 2023. Error bars represent relative  
 462 error from analytical duplicates. (B) Difference between the PBR and mix tank solids C:N ratios,  
 463 with positive values supporting the hypothesis that the microbial community was storing  
 464 carbohydrates in the PBRs and mobilizing (i.e., consuming) stored carbohydrates in the mix tank.  
 465 Error bars in (B) represent propagated relative error (SI Equation S1) from analytical duplicates  
 466 in (A).

467

### 468 3.3. A Path Forward for Intensive, Suspended Growth, Mixed Microalgal Wastewater 469 Treatment

470 Wastewater resource recovery facilities that support urban populations are often landlocked and  
 471 need to intensify their processes to meet more stringent effluent permits or increase their  
 472 treatment capacity.<sup>51</sup> Adoption of high productivity, small footprint (intensive) microalgal



473 technologies creates an opportunity to sustainably convert waste nutrients to marketable products  
474 and meet rigorous effluent nutrient criteria. While enhanced biological phosphorus removal  
475 (EBPR) may remove phosphorus to effluent concentrations approaching  $0.1 \text{ mg-P}\cdot\text{L}^{-1}$ , the  
476 EcoRecover process has demonstrated long-term recovery of phosphorus to achieve effluent total  
477 phosphorus concentrations below  $0.03 \text{ mg-P}\cdot\text{L}^{-1}$ , even in winter months in Wisconsin (latitude of  
478  $45^\circ \text{ N}$ ). An additional advantage of algal-based systems is the potential for organic phosphorus  
479 (and organic nitrogen) recovery,<sup>11</sup> which remain a critical challenge for conventional bacterial and  
480 precipitation-based nutrient removal technologies.<sup>52</sup> Future studies may specifically focus on  
481 organic nutrient recovery, as well as the sustainability implications (e.g., reduced chemical  
482 dosages and  $\text{CO}_2$  sequestration)<sup>53</sup> of replacing alternative tertiary treatment processes such as  
483 chemical phosphorus polishing.

484 In the algal cultivation space, technologies are often compared based on their areal  
485 productivities. In this study, the characterized EcoRecover process was intentionally designed to  
486 be phosphorus-limited to meet stringent permit requirements. As a result, the system was not  
487 designed to maximize biomass productivity and instead prioritized reliable effluent quality (with  
488 biomass production and sale serving as a secondary benefit). Nonetheless, across the focus  
489 period (November 1, 2022 to February 14, 2023), the EcoRecover's areal productivity was  $15 \pm$   
490  $4 \text{ g}\cdot\text{m}^{-2}\cdot\text{day}^{-1}$  in winter months (external temperatures from  $-27$  to  $24 \text{ }^\circ\text{C}$ , daily average of  $-7 \text{ }^\circ\text{C}$ )  
491 at a high altitude ( $45^\circ \text{ N}$ ). In other monitored periods subject to upstream upsets or other external  
492 pressures (e.g., chemical shortages), areal productivities on the order of  $45 \text{ g}\cdot\text{m}^{-2}\cdot\text{d}^{-1}$  (average  
493 from July 26 to September 6, 2022; individual timepoint estimates ranged from 36 to  $60 \text{ g}\cdot\text{m}^{-2}\cdot\text{d}^{-1}$ )  
494 were also observed.

495 Ultimately, this work represents the first full-scale characterization of the EcoRecover  
496 process for algae cultivation and tertiary nutrient recovery. Parallel studies will include more in-  
497 depth analyses of the microbial ecology, including longer-term sequencing results and more in-

498 depth community structure analyses.<sup>54</sup> Future work will continue to build off this understanding to  
499 advance our ability to optimize the design of this system, mechanistically and dynamically model  
500 its performance, and develop tailored solutions for utilities seeking to simultaneously advance  
501 goals for improved effluent quality and engagement with the circular bioeconomy.

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511

## 512 **Supplemental Information Available**

513 The following Supporting Information is available:

514 Images of the EcoRecover system at Roberts, WI, USA; Detailed descriptions of on-line  
515 monitoring equipment; Expanded continuous monitoring results for system performance;  
516 Detailed description of ion chromatography analyses; Descriptions of periods of system  
517 performance upset; Solids characterization through SEM-EDS; Batch experiment  
518 aqueous and biomass analyses. (PDF)

519

520 Cleaned long-term AIMS monitoring data from November 1, 2022 through February 14,  
521 2023 (XLSX)

522

523           Cleaned long-term SCADA monitoring data from November 1, 2022 through February 14,  
524           2023 (XLSX)

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575

## 576 **References**

- 577 (1) Nutrients Working Group. *Nutrient Redution Progress Tracker Version 1.0- 2017*;  
578 Association of Clean Water Administrators, 2018. [https://www.acwa-us.org/wp-](https://www.acwa-us.org/wp-content/uploads/2018/03/Nutrient-Reduction-Progress-Tracker-Version-1.0-2017-Report.pdf)  
579 [content/uploads/2018/03/Nutrient-Reduction-Progress-Tracker-Version-1.0-2017-Report.pdf](https://www.acwa-us.org/wp-content/uploads/2018/03/Nutrient-Reduction-Progress-Tracker-Version-1.0-2017-Report.pdf)  
580 (accessed 2022-07-19).
- 581 (2) Schindler, D. W.; Carpenter, S. R.; Chapra, S. C.; Hecky, R. E.; Orihel, D. M. Reducing  
582 Phosphorus to Curb Lake Eutrophication Is a Success. *Environ. Sci. Technol.* **2016**, *50* (17),  
583 8923–8929. <https://doi.org/10.1021/acs.est.6b02204>.
- 584 (3) US EPA. *Progress towards Adopting Total Nitrogen and Total Phosphorus Numeric Water*  
585 *Quality Standards*. US EPA Nutrient Policy and Data. [https://www.epa.gov/nutrient-policy-](https://www.epa.gov/nutrient-policy-data/progress-towards-adopting-total-nitrogen-and-total-phosphorus-numeric-water)  
586 [data/progress-towards-adopting-total-nitrogen-and-total-phosphorus-numeric-water](https://www.epa.gov/nutrient-policy-data/progress-towards-adopting-total-nitrogen-and-total-phosphorus-numeric-water)  
587 (accessed 2022-07-19).
- 588 (4) US EPA. *N/P Criteria Progress Map*. State Progress Toward Developing Numeric Nutrient  
589 Water Quality Criteria for Nitrogen and Phosphorus. [https://www.epa.gov/nutrient-policy-](https://www.epa.gov/nutrient-policy-data/state-progress-toward-developing-numeric-nutrient-water-quality-criteria#tb3)  
590 [data/state-progress-toward-developing-numeric-nutrient-water-quality-criteria#tb3](https://www.epa.gov/nutrient-policy-data/state-progress-toward-developing-numeric-nutrient-water-quality-criteria#tb3) (accessed  
591 2022-07-20).
- 592 (5) Li, X.; Shen, S.; Xu, Y.; Guo, T.; Dai, H.; Lu, X. Application of Membrane Separation  
593 Processes in Phosphorus Recovery: A Review. *Science of The Total Environment* **2021**,  
594 767, 144346. <https://doi.org/10.1016/j.scitotenv.2020.144346>.
- 595 (6) Bunce, J. T.; Ndam, E.; Ofiteru, I. D.; Moore, A.; Graham, D. W. A Review of Phosphorus  
596 Removal Technologies and Their Applicability to Small-Scale Domestic Wastewater  
597 Treatment Systems. *Frontiers in Environmental Science* **2018**, *6*.

- 598 (7) Zahed, M. A.; Salehi, S.; Tabari, Y.; Farraji, H.; Ataei-Kachooei, S.; Zinatizadeh, A. A.;  
599 Kamali, N.; Mahjouri, M. Phosphorus Removal and Recovery: State of the Science and  
600 Challenges. *Environ Sci Pollut Res* **2022**, *29* (39), 58561–58589.  
601 <https://doi.org/10.1007/s11356-022-21637-5>.
- 602 (8) Bott, C. B.; Parker, D. S. *Nutrient Management Volume II: Removal Technology*  
603 *Performance & Reliability*; Water Environment Research Foundation: Alexandria, VA, 2011.  
604 [https://www.waterrf.org/resource/nutrient-management-volume-ii-removal-technology-](https://www.waterrf.org/resource/nutrient-management-volume-ii-removal-technology-performance-reliability)  
605 [performance-reliability](https://www.waterrf.org/resource/nutrient-management-volume-ii-removal-technology-performance-reliability).
- 606 (9) HydroQual, Inc.; Stensel, D. H. *Wastewater Phosphorus Control and Reduction Initiative*;  
607 Prepared for the Minnesota Environmental Science and Economic Review Board.; 2005; pp  
608 37–38. <https://wrl.mnpals.net/islandora/object/WRLrepository%3A941/datastream/PDF/view>  
609 (accessed 2022-11-07).
- 610 (10) Petzet, S.; Peplinski, B.; Bodkhe, S. Y.; Cornel, P. Recovery of Phosphorus and  
611 Aluminium from Sewage Sludge Ash by a New Wet Chemical Elution Process (SESAL-  
612 Phos-Recovery Process). *Water Science and Technology* **2011**, *64* (3), 693–699.  
613 <https://doi.org/10.2166/wst.2011.682>.
- 614 (11) Qin, C.; Liu, H.; Liu, L.; Smith, S.; Sedlak, D. L.; Gu, A. Z. Bioavailability and  
615 Characterization of Dissolved Organic Nitrogen and Dissolved Organic Phosphorus in  
616 Wastewater Effluents. *Science of The Total Environment* **2015**, *511*, 47–53.  
617 <https://doi.org/10.1016/j.scitotenv.2014.11.005>.
- 618 (12) Sutherland, D. L.; Bramucci, A. Dissolved Organic Phosphorus Bioremediation from  
619 Food-Waste Centrate Using Microalgae. *Journal of Environmental Management* **2022**, *313*,  
620 115018. <https://doi.org/10.1016/j.jenvman.2022.115018>.
- 621 (13) Mayer, B. K.; Baker, L. A.; Boyer, T. H.; Drechsel, P.; Gifford, M.; Hanjra, M. A.;  
622 Parameswaran, P.; Stoltzfus, J.; Westerhoff, P.; Rittmann, B. E. Total Value of Phosphorus  
623 Recovery. *Environ. Sci. Technol.* **2016**, *50* (13), 6606–6620.  
624 <https://doi.org/10.1021/acs.est.6b01239>.
- 625 (14) *Clearas Water Recovery, Inc.* <https://www.clearassolutions.com/> (accessed 2022-07-15).
- 626 (15) *Village of Roberts, Wisconsin.* <https://www.robertswisconsin.com/> (accessed 2022-07-  
627 14).
- 628 (16) Wisconsin Department of Natural Resources. *2022 Water Conditions: List Appendix A-*  
629 *Impaired Waters List.* Water Conditions Lists.  
630 <https://dnr.wisconsin.gov/topic/SurfaceWater/ConditionLists.html> (accessed 2022-07-26).

- 631 (17) Rieger, L. Guidelines for Using Activated Sludge Models. *wio* **2012**, *11*.  
632 <https://doi.org/10.2166/9781780401164>.
- 633 (18) *KernelSmoother*. scikit-fda.  
634 <https://fda.readthedocs.io/en/stable/modules/preprocessing/autosummary/skfda.preprocessing.smoothing.KernelSmoother.html> (accessed 2023-11-13).
- 636 (19) Kim, G.-Y.; Molitor, H. R.; Zhang, X.; Li, Y.; Avila, N. M.; Shoener, B. D.; Schramm, S.  
637 M.; Morgenroth, E.; Snowling, S. D.; Bradley, I. M.; Pinto, A. J.; Guest, J. S. Development  
638 and Validation of a Phototrophic-Mixotrophic Process Model (PM2) and a Process Simulator  
639 for Microalgae-Based Wastewater Treatment. <https://doi.org/In preparation>.
- 640 (20) Ripp, J. *Analytical Detection Limit Guidance & Laboratory Guide for Determining Method*  
641 *Detection Limits*; Wisconsin Department of Natural Resources, Laboratory Certification  
642 Program, 1996.
- 643 (21) Pruvost, J.; Van Vooren, G.; Cogne, G.; Legrand, J. Investigation of Biomass and Lipids  
644 Production with *Neochloris Oleoabundans* in Photobioreactor. *Bioresource Technology*  
645 **2009**, *100* (23), 5988–5995. <https://doi.org/10.1016/j.biortech.2009.06.004>.
- 646 (22) Lourenço, S. O.; Barbarino, E.; Marquez, U. M. L.; Aidar, E. Distribution of Intracellular  
647 Nitrogen in Marine Microalgae: Basis for the Calculation of Specific Nitrogen-to-protein  
648 Conversion Factors. *Journal of Phycology* **1998**, *34* (5), 798–811.  
649 <https://doi.org/10.1046/j.1529-8817.1998.340798.x>.
- 650 (23) Van Wychen, S.; Laurens, L. M. L. *Determination of Total Carbohydrates in Algal*  
651 *Biomass: Laboratory Analytical Procedure*; Technical Report NREL/TP-5100-60957;  
652 National Renewable Energy Laboratory: Golden, CO, 2015.
- 653 (24) Gardner-Dale, D. A.; Bradley, I. M.; Guest, J. S. Influence of Solids Residence Time and  
654 Carbon Storage on Nitrogen and Phosphorus Recovery by Microalgae across Diel Cycles.  
655 *Water Research* **2017**, *121*, 231–239. <https://doi.org/10.1016/j.watres.2017.05.033>.
- 656 (25) Bradley, I. M.; Li, Y.; Guest, J. S. Solids Residence Time Impacts Carbon Dynamics and  
657 Bioenergy Feedstock Potential in Phototrophic Wastewater Treatment Systems. *Environ.*  
658 *Sci. Technol.* **2021**. <https://doi.org/10.1021/acs.est.1c02590>.
- 659 (26) Mañas, A.; Biscans, B.; Spérandio, M. Biologically Induced Phosphorus Precipitation in  
660 Aerobic Granular Sludge Process. *Water Research* **2011**, *45* (12), 3776–3786.  
661 <https://doi.org/10.1016/j.watres.2011.04.031>.
- 662 (27) Liu, Y.-Q.; Cinquepalmi, S. Exploration of Mechanisms for Calcium Phosphate  
663 Precipitation and Accumulation in Nitrifying Granules by Investigating the Size Effects of



- 664 Granules. *Water Research* **2021**, *206*, 117753.  
665 <https://doi.org/10.1016/j.watres.2021.117753>.
- 666 (28) Huang, W.; Huang, W.; Li, H.; Lei, Z.; Zhang, Z.; Tay, J. H.; Lee, D.-J. Species and  
667 Distribution of Inorganic and Organic Phosphorus in Enhanced Phosphorus Removal  
668 Aerobic Granular Sludge. *Bioresource Technology* **2015**, *193*, 549–552.  
669 <https://doi.org/10.1016/j.biortech.2015.06.120>.
- 670 (29) Axelsson, M.; Gentili, F. A Single-Step Method for Rapid Extraction of Total Lipids from  
671 Green Microalgae. *PLoS One* **2014**, *9* (2), e89643.  
672 <https://doi.org/10.1371/journal.pone.0089643>.
- 673 (30) Folch, J.; Lees, M.; Sloane Stanley, G. H. A Simple Method for the Isolation and  
674 Purification of Total Lipides from Animal Tissues. *J. Biol. Chem.* **1957**, *226* (1), 497–509.
- 675 (31) Bradley, I. M.; Pinto, A. J.; Guest, J. S. Design and Evaluation of Illumina MiSeq-  
676 Compatible, 18S rRNA Gene-Specific Primers for Improved Characterization of Mixed  
677 Phototrophic Communities. *Appl. Environ. Microbiol.* **2016**, *82* (19), 5878–5891.  
678 <https://doi.org/10.1128/AEM.01630-16>.
- 679 (32) Kozich, J. J.; Westcott, S. L.; Baxter, N. T.; Highlander, S. K.; Schloss, P. D.  
680 Development of a Dual-Index Sequencing Strategy and Curation Pipeline for Analyzing  
681 Amplicon Sequence Data on the MiSeq Illumina Sequencing Platform. *Appl. Environ.*  
682 *Microbiol.* **2013**, AEM.01043-13. <https://doi.org/10.1128/AEM.01043-13>.
- 683 (33) Excoffier, L.; Smouse, P. E.; Quattro, J. M. Analysis of Molecular Variance Inferred from  
684 Metric Distances among DNA Haplotypes: Application to Human Mitochondrial DNA  
685 Restriction Data. *Genetics*, 1992, *131*, 479–491.
- 686 (34) Garcia, N. S.; Sexton, J.; Riggins, T.; Brown, J.; Lomas, M. W.; Martiny, A. C. High  
687 Variability in Cellular Stoichiometry of Carbon, Nitrogen, and Phosphorus Within Classes of  
688 Marine Eukaryotic Phytoplankton Under Sufficient Nutrient Conditions. *Front. Microbiol.*  
689 **2018**, *9*, 543. <https://doi.org/10.3389/fmicb.2018.00543>.
- 690 (35) Finkel, Z. V.; Follows, M. J.; Liefer, J. D.; Brown, C. M.; Benner, I.; Irwin, A. J.  
691 Phylogenetic Diversity in the Macromolecular Composition of Microalgae. *PLoS ONE* **2016**,  
692 *11* (5), e0155977. <https://doi.org/10.1371/journal.pone.0155977>.
- 693 (36) Choi, H. J.; Lee, S. M. Effect of the N/P Ratio on Biomass Productivity and Nutrient  
694 Removal from Municipal Wastewater. *Bioprocess Biosyst Eng* **2015**, *38* (4), 761–766.  
695 <https://doi.org/10.1007/s00449-014-1317-z>.



- 696 (37) Beuckels, A.; Smolders, E.; Muylaert, K. Nitrogen Availability Influences Phosphorus  
697 Removal in Microalgae-Based Wastewater Treatment. *Water Research* **2015**, *77*, 98–106.  
698 <https://doi.org/10.1016/j.watres.2015.03.018>.
- 699 (38) Rhee, G.-Y. A Continuous Culture Study of Phosphate Uptake, Growth Rate and  
700 Polyphosphate in *Scenedesmus* Sp. <sup>1</sup>. *Journal of Phycology* **1973**, *9* (4), 495–506.  
701 <https://doi.org/10.1111/j.1529-8817.1973.tb04126.x>.
- 702 (39) Whitton, R.; Le Mével, A.; Pidou, M.; Ometto, F.; Villa, R.; Jefferson, B. Influence of  
703 Microalgal N and P Composition on Wastewater Nutrient Remediation. *Water Research*  
704 **2016**, *91*, 371–378. <https://doi.org/10.1016/j.watres.2015.12.054>.
- 705 (40) Geider, R.; La Roche, J. Redfield Revisited: Variability of C:N:P in Marine Microalgae  
706 and Its Biochemical Basis. *European Journal of Phycology* **2002**, *37* (1), 1–17.  
707 <https://doi.org/10.1017/S0967026201003456>.
- 708 (41) Grobbelaar, J. U. Inorganic Algal Nutrition. *Handbook of microalgal culture: applied*  
709 *phycology and biotechnology* **2013**, 123–133.
- 710 (42) Solovchenko, A. E.; Ismagulova, T. T.; Lukyanov, A. A.; Vasilieva, S. G.; Konyukhov, I.  
711 V.; Pogosyan, S. I.; Lobakova, E. S.; Gorelova, O. A. Luxury Phosphorus Uptake in  
712 Microalgae. *Journal of Applied Phycology* **2019**, *31* (5), 2755–2770.  
713 <https://doi.org/10.1007/s10811-019-01831-8>.
- 714 (43) Powell, N.; Shilton, A. N.; Pratt, S.; Chisti, Y. Factors Influencing Luxury Uptake of  
715 Phosphorus by Microalgae in Waste Stabilization Ponds. *Environ. Sci. Technol.* **2008**, *42*  
716 (16), 5958–5962. <https://doi.org/10.1021/es703118s>.
- 717 (44) Sanz-Luque, E.; Bhaya, D.; Grossman, A. R. Polyphosphate: A Multifunctional  
718 Metabolite in Cyanobacteria and Algae. *Front. Plant Sci.* **2020**, *11*, 938.  
719 <https://doi.org/10.3389/fpls.2020.00938>.
- 720 (45) Zerveas, S.; Mente, M. S.; Tsakiri, D.; Kotzabasis, K. Microalgal Photosynthesis Induces  
721 Alkalization of Aquatic Environment as a Result of H<sup>+</sup> Uptake Independently from CO<sub>2</sub>  
722 Concentration – New Perspectives for Environmental Applications. *Journal of Environmental*  
723 *Management* **2021**, *289*, 112546. <https://doi.org/10.1016/j.jenvman.2021.112546>.
- 724 (46) Hartley, A. M.; House, W. A.; Callow, M. E.; Leadbeater, B. S. C. Coprecipitation of  
725 Phosphate with Calcite in the Presence of Photosynthesizing Green Algae. *Water Research*  
726 **1997**, *31* (9), 2261–2268. [https://doi.org/10.1016/S0043-1354\(97\)00103-6](https://doi.org/10.1016/S0043-1354(97)00103-6).
- 727 (47) Xu, M.; Bernards, M.; Hu, Z. Algae-Facilitated Chemical Phosphorus Removal during  
728 High-Density *Chlorella Emersonii* Cultivation in a Membrane Bioreactor. *Bioresource*  
729 *Technology* **2014**, *153*, 383–387. <https://doi.org/10.1016/j.biortech.2013.12.026>.

- 730 (48) Tibbetts, S. M.; Milley, J. E.; Lall, S. P. Chemical Composition and Nutritional Properties  
731 of Freshwater and Marine Microalgal Biomass Cultured in Photobioreactors. *J Appl Phycol*  
732 **2015**, 27 (3), 1109–1119. <https://doi.org/10.1007/s10811-014-0428-x>.
- 733 (49) Molitor, H. R.; Schnoor, J. L. Using Simulated Flue Gas to Rapidly Grow Nutritious  
734 Microalgae with Enhanced Settleability. *ACS Omega* **2020**, 5 (42), 27269–27277.  
735 <https://doi.org/10.1021/acsomega.0c03492>.
- 736 (50) Valverde-Pérez, B.; Ramin, E.; Smets, B. F.; Plósz, B. G. EBP2R – An Innovative  
737 Enhanced Biological Nutrient Recovery Activated Sludge System to Produce Growth  
738 Medium for Green Microalgae Cultivation. *Water Research* **2015**, 68, 821–830.  
739 <https://doi.org/10.1016/j.watres.2014.09.027>.
- 740 (51) Sturm, B. *State of Knowledge and Workshop Report: Intensification of Resource*  
741 *Recovery (IR2) Forum*; TIRR1R15; Water Environment Research Foundation: Alexandria,  
742 VA, 2016.
- 743 (52) Clark, D. L. Nutrient Management: Regulatory Approaches to Protect Water Quality:  
744 Volume 1 – Review of Existing Practices. *wio* **2010**, 9.  
745 <https://doi.org/10.2166/9781780403465>.
- 746 (53) Falk, M. W.; Reardon, D. J.; Neethling, J. B.; Clark, D. L.; Pramanik, A. Striking the  
747 Balance between Nutrient Removal, Greenhouse Gas Emissions, Receiving Water Quality,  
748 and Costs. *Water Environment Research* **2013**, 85 (12), 2307–2316.  
749 <https://doi.org/10.2175/106143013X13807328848379>.
- 750 (54) Alam, M. M.; Hodaei, M.; Hartnett, E.; Gincley, B.; Kim, G.-Y.; Pinto, A. J.; Bradley, I. M.  
751 Community Structure and Function During Periods of High Performance and System Upset  
752 in a Full-Scale Mixed Microalgal Wastewater Resource Recovery Facility. **In preparation**.  
753