

# Organocatalytic Enantioselective Nucleophilic Addition of Indole Imine 5-Methides

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## Abstract

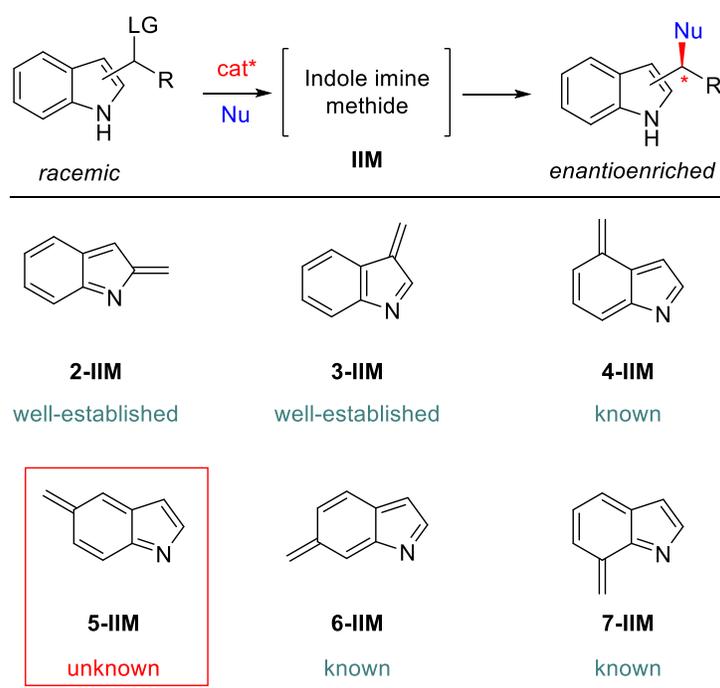
*Despite the enormous developments in the asymmetric transformations of indole imine methides (IIMs), the remote asymmetric induction involving IIMs remains challenging due to the spatial interaction requirement between the substrate and catalyst. Herein we report the first catalytic asymmetric nucleophilic addition to indole imine 5-methide (5-IIM), the only topological isomer of IIMs whose asymmetric addition remains unknown. Considering the challenging remote stereocontrol, high efficiency and respectable enantioselectivity were achieved, providing access to a range of enantioenriched indole-containing triarylmethanes.*

Indole is a privileged heterocycle with broad applications in both organic synthesis and medicinal chemistry.<sup>1</sup> Specifically, indole imine methides (IIMs) are highly versatile intermediates in the synthesis of diverse functionalized indole-containing molecules.<sup>2,3</sup> Similar to the reactivity of (aza-) quinone methides, IIMs are highly electrophilic and thus prone to nucleophilic addition at the methide position owing to their strong tendency to rearomatization.<sup>2,3</sup> In particular, in the past decade, a wide range of catalytic asymmetric processes have been achieved by convenient *in situ* generation of IIMs followed by subsequent stereocontrolled nucleophilic addition or cycloaddition.<sup>2</sup> These processes, typically operated in a one-pot manner under mild conditions, have allowed facile access to diverse enantioenriched indole-containing molecules from simple racemic/achiral precursors.

Depending on the position of exocyclic methide unit, IIMs have multiple topological isomers, namely, indole imine 2-, 3-, 4-, 5-, 6-, 7-methides (Scheme 1).<sup>2e,4-9</sup> Among them, indole imine 2- and 3-methides (2- and 3-IIMs) are most studied isomers regarding their asymmetric transformations.<sup>2,4,5</sup> This is probably partly due to the short distance between the

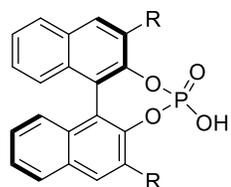
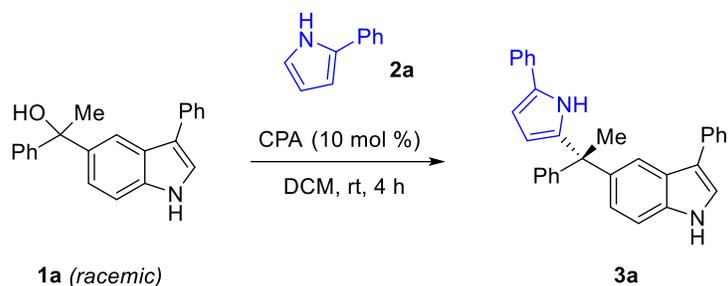
methide position and the imine motif, the typical catalytic activation site, which permits relatively easy stereocontrol. In contrast, other topological isomers, such as 4-, 6-, and 7-IIMs, have been much less-explored for asymmetric synthesis, since catalytic remote stereocontrol is often required in these cases in order to achieve high enantioselectivity.<sup>6-8</sup> While sporadic examples are known for these isomers, however, 5-IIM is the only isomer whose asymmetric transformations remain unknown, to the best of our knowledge. Indeed, among all these topological isomers, 5-IIM has the longest distance between the methide and imine motifs, which exerts substantial challenge in effective stereocontrol by the catalyst. In this context, herein we fill this gap by introducing the the first catalytic asymmetric process of 5-IIMs, leading to efficient access to enantioenriched indole-containing triarylalkanes bearing a remote benzylic chirality.

### Scheme 1. Introduction to Topological Isomers and Current State of Catalytic Asymmetric Nucleophilic Addition of IIMs.

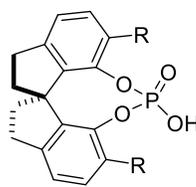


We employed the racemic indol-5-ylmethanol **1a** as the model precursor to 5-IIM and 2-phenyl-1*H*-pyrrole (**2a**) as the nucleophile for the initial study (Table 1). Chiral phosphoric acids (CPAs) were evaluated as potential catalysts for this reaction, considering their general excellent performance in the asymmetric transformations of other IIMs and the related *para*-aza-quinone methides.<sup>2,4-10</sup> With the well-known TRIP catalyst (**A1**), the reaction in DCM proceeded successfully at room temperature to form the desired product **3a** in 58% yield, albeit with only 76:24 enantiomeric ratio (er, entry 1). This result implied that the remote enantiocontrol was indeed challenging. Further considerable efforts were devoted to exhaustive screening of a wide range of other chiral phosphoric acids. Representative examples were shown (entries 2-6). Changing the 3,3'-substituents to 1-naphthyl and 1-anthryl led to improvement on both yield and enantioselectivity (entries 2-3). Further screening of SPINOL-derived CPAs identified that (*R*)-**B1** provided the highest enantioselectivity (entry 4). Further solvent screening indicated that coordinating solvents, such as diethyl ether and ethyl acetate, led to low yield and moderate enantioselectivity, likely due to competitive binding with the acid catalyst. (entries 7-8). In contrast, toluene slightly improved the enantioselective (91:9 er, entry 9).

### Table 1. Evaluation of Conditions<sup>a</sup>



(*R*)-A1: R = 2,4,6-*i*Pr<sub>3</sub>C<sub>6</sub>H<sub>2</sub>  
 (*R*)-A2: R = 1-naphthyl  
 (*R*)-A3: R = 9-anthryl



(*R*)-B1: R = 1-naphthyl  
 (*R*)-B2: R = 9-anthryl  
 (*R*)-B3: R = 9-phenanthryl

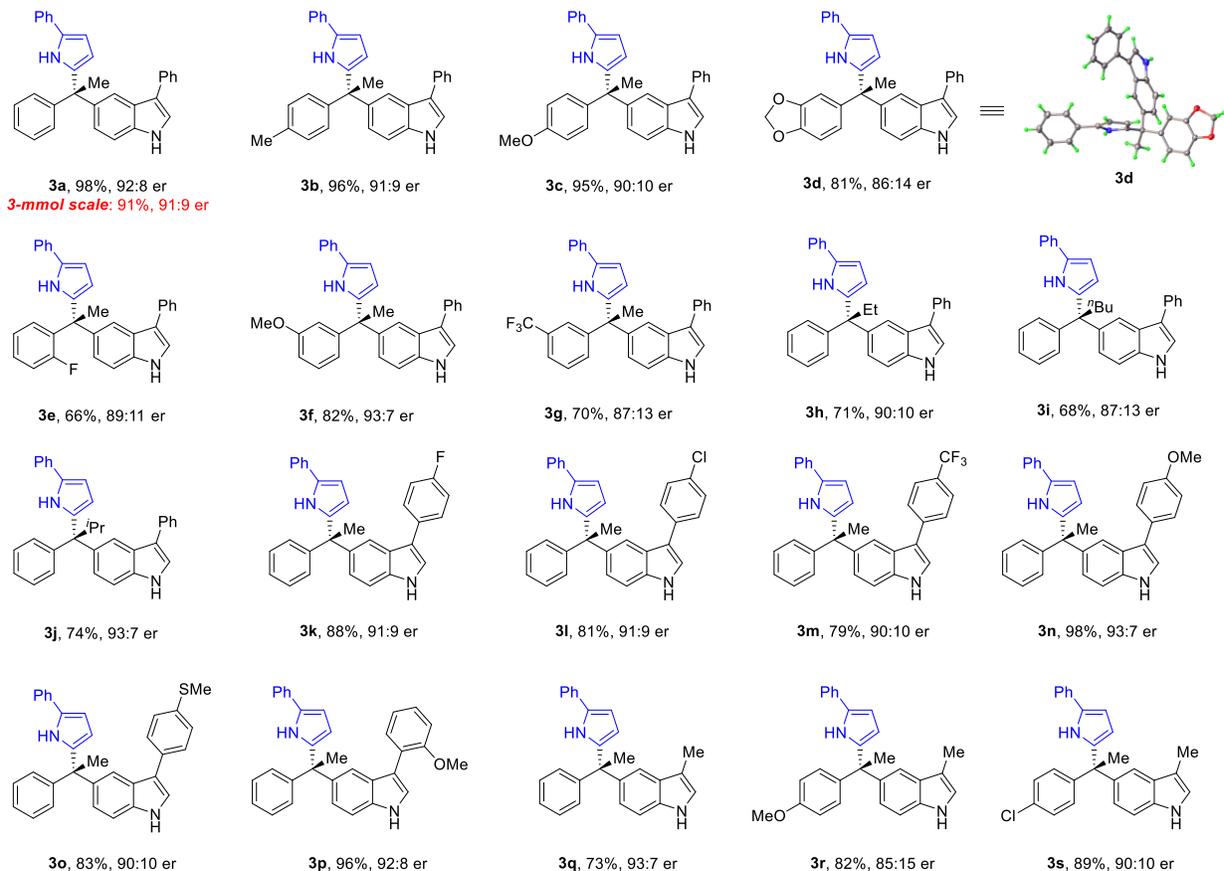
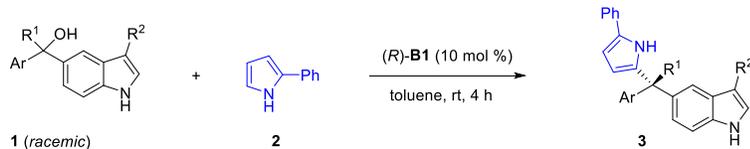
Entry	CPA	Solvent	Yield of 3a <sup>b</sup>	er of 3a <sup>b</sup>
1	( <i>R</i> )-A1	DCM	58%	76:24
2	( <i>R</i> )-A2	DCM	>95%	82:18
3	( <i>R</i> )-A3	DCM	>95%	87:13
4	( <i>R</i> )-B1	DCM	>95%	89:11
5	( <i>R</i> )-B2	DCM	74	86:14
6	( <i>R</i> )-B3	DCM	87	78:22
7	( <i>R</i> )-B1	EtOAc	22	-
8	( <i>R</i> )-B1	Et <sub>2</sub> O	46	70:30
9	( <i>R</i> )-B1	toluene	>95%	91:9

<sup>a</sup> Reaction scale: **1a** (0.05 mmol), **2** (0.075 mmol, 1.5 equiv), catalyst (5 μmol, 10 mol %), solvent (0.5 mL), 4 h. <sup>b</sup> Yield was determined by <sup>1</sup>H NMR analysis of the crude mixture using

CH<sub>2</sub>Br<sub>2</sub> as an internal standard. Er was determined by HPLC analysis on a chiral stationary phase.

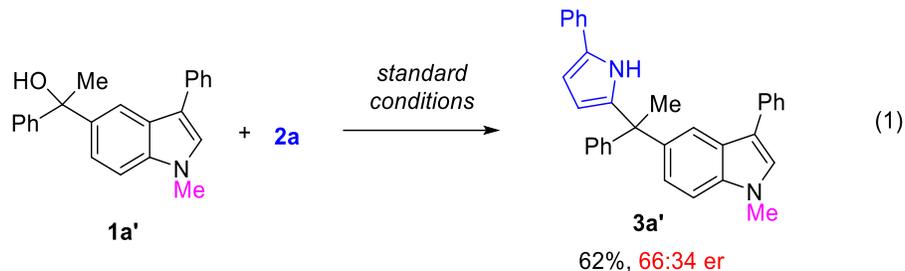
With the optimized conditions in hands, we next examined the reaction scope with various substituted indol-5-ylmethanols (Scheme 2). In general, this protocol provided efficient access to a wide range of indole-containing triarylalkanes **3a-3s** with high yield and good to high enantioselectivity. Electron-donating and electron-withdrawing substituents at different positions of the 5-benzylic aryl ring did not obviously affect the reaction efficiency or enantioselectivity (**3b-3g**). Similarly, substrates with different aliphatic substituents at 5-benzylic position with different chain length (**3h-3i**) and steric hinderance (**3j**) all reacted without deterioration in enantioselectivity. Different functional groups, including aryl halide, trifluoromethyl, thioether, acetal, and ether, were compatible with the mild conditions. It is worth noting that the enantioselectivity was respectable, particularly considering that this process involves very remote stereocontrol. Finally, the robustness of this protocol was further demonstrated by a gram-scale synthesis of **3a**, resulting in comparable yield and enantioselectivity as the small-scale reaction. The structure and absolute configuration of the product **3d** were confirmed unambiguously by X-ray crystallography (CCDC 2311114).

### Scheme 2. Reaction Scope<sup>a</sup>



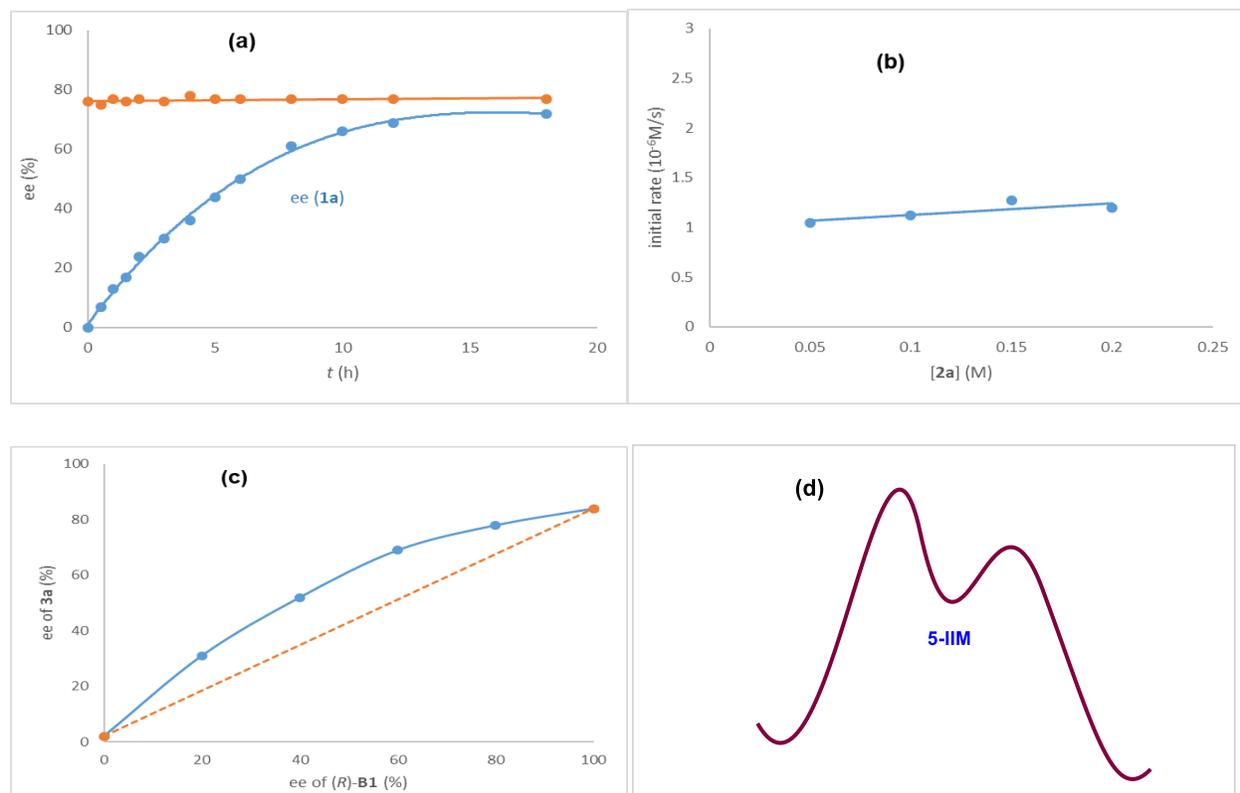
<sup>a</sup> Reaction conditions: **1** (0.36 mmol), **2** (0.3 mmol), (*R*)-**B1** (0.03mmol), toluene (3.0 mL), 4 h.

To gain insight into the role of the hypothetical indole imine methide intermediate, we carried out a control experiment with *N*-methylated indole substrate **1a'**. Under the standard conditions, the reaction resulted in lower yield and significantly decreased enantioselectivity (eq 1), indicating the importance of the N-H functionality, which is crucial for the formation of the 5-IIM intermediate.



To further probe the reaction mechanism, we investigated the time-dependence of the ee values of both the product **3a** and starting material **1a** throughout the reaction progress. The results showed that the ee value of **1a** increased over time, indicating kinetic resolution of the substrate (Figure 1a). However, the product ee remained constant during the reaction, which was consistent with the involvement of an achiral 5-IIM intermediate and thus enantioconvergent nature of this process. Taken together, these results may suggest that the direct  $S_N2$  substitution pathway of indol-5-yl methanols is not operative in this process. In addition, the presence of kinetic resolution may suggest that the first step is irreversible. Thus, the second step, nucleophilic addition, is highly likely a fast step and thus the whole reaction might exhibit zeroth order in nucleophile. Indeed, subsequent kinetic studies indicated that this reaction rate had no dependence on nucleophile concentration (Figure 1b), which further confirmed that the first step is rate-determining and irreversible. The observed remote stereocontrol prompted us to gain more information about the possible enantiodetermining transition state. Interestingly, we observed positive non-linear effects (Figure 1c).<sup>11</sup> While there is no solid evidence for the formation of higher-order catalyst aggregates responsible for asymmetric induction, it could be possible that two or more chiral phosphoric acid molecules are involved in the

enantiodetermining transition state to relay hydrogen bonding and enable better remote stereocontrol.

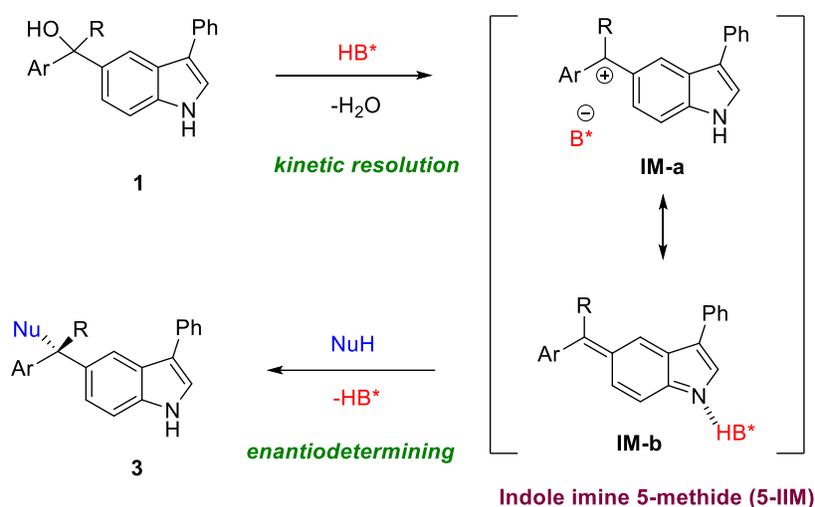


**Figure 1.** Mechanistic studies. (a) Time-dependence of substrate and product ee values. (b) Zeroth order in nucleophile. (c) Positive non-linear effects. (d) Proposed reaction coordinate diagram.

Based on the above results and analysis, a reaction diagram and a possible reaction mechanism is proposed (Figure 1d and Scheme 3). The chiral phosphoric acid catalyst initially activates the tertiary alcohol moiety, leading to the elimination of a water molecule from substrate **1** to form the ion pair **IM-a**, consisting of an indo-5-yl carbocation and a chiral phosphate anion. This intermediate is stabilized by the resonance form **IM-b**, which can be

viewed as a CPA-activated indole imine 5-methide (5-IIM). Subsequent asymmetric nucleophilic addition takes place at the 5-methide position to form the desired products **3** with enantiocontrol. While the exact structure of the enantiodetermining transition state is unknown, it can also be hypothesized that the pyrrole nucleophile may also have hydrogen bond interaction with the basic phosphoryl (P=O) motif of the catalyst in a bifunctional activation mode.

### Scheme 3. Proposed Mechanism



In summary, we have developed the first catalytic asymmetric reactions of indole imine 5-methides. With this process, all the topological isomers of indole imine methides have been successfully demonstrated for asymmetric synthesis. The ultra long distance between the imine and methide motifs in this case posed substantial challenges in remote enantiocontrol. By careful catalyst screening and condition optimization, excellent efficiency and good enantioselectivity were achieved for the intermolecular C–C bond formation with a range of indol-5-ylmethanols, which provided efficient access to diverse enantioenriched indole-containing triarylmethanes bearing a remote stereogenic center. Control experiments and kinetic studies

were consistent with the involvement of an IIM intermediate and further suggested that the initial dehydration step is rate-determining and irreversible.

## **ASSOCIATED CONTENT**

### Supporting Information

The Supporting Information is available free of charge via the Internet at <http://pubs.acs.org>.

Additional experimental and computational details, and spectroscopic data of all compounds (PDF).

X-ray crystallography data (CIF)

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### **Notes**

The authors declare no competing financial interest.

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