

**Separation and Characterization of Synthetic Cannabinoid Metabolite Isomers using SLIM High-Resolution Ion Mobility-Tandem Mass Spectrometry (HRIM-MS/MS)**

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## Supporting Tables

**Table S1.** Instrumental parameters for the MOBILion MOBIE SLIM & Agilent 6546 QTOF.

Instrument Region	Instrumental Parameter	Experimental Value
<b>Ionization Source</b>	Polarity	Positive
	Gas Temp	325 °C
	Drying Gas	12 L/min
	Nebulizer	20 psi
	Sheath Gas Temp	275 °C
	Sheath Gas Flow	10 L/min
	VCap	4000 V
	Nozzle Voltage (Expt)	1000 V
	Fragmentor	400 V
	Oct 1 RF Vpp	750 V
	<b>SLIM Conditions</b>	Funnel In
Funnel Out		100 V
Funnel Conductance Limit (CL)		95 V
SLIM Bias		90 V
SLIM Mode		HRIM (13 m)
SLIM Wave Shape		Sine
Fill Time		40 ms
Trap Time		0.3 ms
Release Time		3.2 ms
IMS Frame Length		1000 ms
Fill TW Frequency		15 kHz
Fill TW Amplitude		5 V <sub>pp</sub>
Release TW Frequency		15 kHz
Release TW Amplitude		30 V <sub>pp</sub>
Separation TW Frequency		20 kHz
Separation TW Amplitude		35 V <sub>pp</sub>
SLIM Exit CL		50 V
Quad Bias		45 V
Quad Pressure (Rough Vac)		2.5 Torr
SLIM RF Amplitude		270 V
SLIM RF Frequency	1200 kHz	

**Table S2.** Experimental  $^{DT}CCS_{N_2}$  and raw  $^{SLIM}CCS_{N_2}$  for the protonated species,  $[M+H]^+$ , for all metabolites. These  $^{SLIM}CCS_{N_2}$  values were derived based on a calibration using the Agilent Tune Mix ions. Chemical formulae and theoretical  $m/z$  are also included.

Metabolite	Formula	$[M+H]^+$ $m/z$	$^{DT}CCS_{N_2}$ ( $\text{\AA}^2$ )	$^{SLIM}CCS_{N_2}$ ( $\text{\AA}^2$ )	$\Delta CCS$ (%)
JWH 018 4-hydroxyindole			189.0 $\pm$ 0.1	187.8 $\pm$ 0.2	-0.64%
JWH 018 N-(5-hydroxypentyl)	C <sub>24</sub> H <sub>23</sub> NO <sub>2</sub>	358.181	187.0 $\pm$ 0.1	187.0 $\pm$ 0.2	0.00%
JWH 018 6-hydroxyindole			192.4 $\pm$ 0.1	191.0 $\pm$ 0.4	-0.73%
JWH 250 N-(4-hydroxypentyl)			187.6 $\pm$ 0.1	186.4 $\pm$ 0.2	-0.64%
JWH 250 N-(5-hydroxypentyl)	C <sub>22</sub> H <sub>25</sub> NO <sub>3</sub>	352.191	187.6 $\pm$ 0.1	186.7 $\pm$ 0.4	-0.48%
JWH 250 5-hydroxyindole			191.5 $\pm$ 0.1	189.9 $\pm$ 0.3	-0.84%
MDA-19 N-(4-hydroxybenzoyl)			196.1 $\pm$ 0.1	194.2 $\pm$ 0.1	-0.97%
MDA-19 N-(5-hydroxyhexyl)			189.3 $\pm$ 0.1	188.3 $\pm$ 0.1	-0.53%
4-cyano CUMYL-BUTINACA	C <sub>21</sub> H <sub>23</sub> N <sub>3</sub> O <sub>3</sub>	366.182	186.4 $\pm$ 0.1	ND	
APP-BUTINACA phenylpropanoic acid			187.3 $\pm$ 0.1	185.3 $\pm$ 0.4	-1.07%

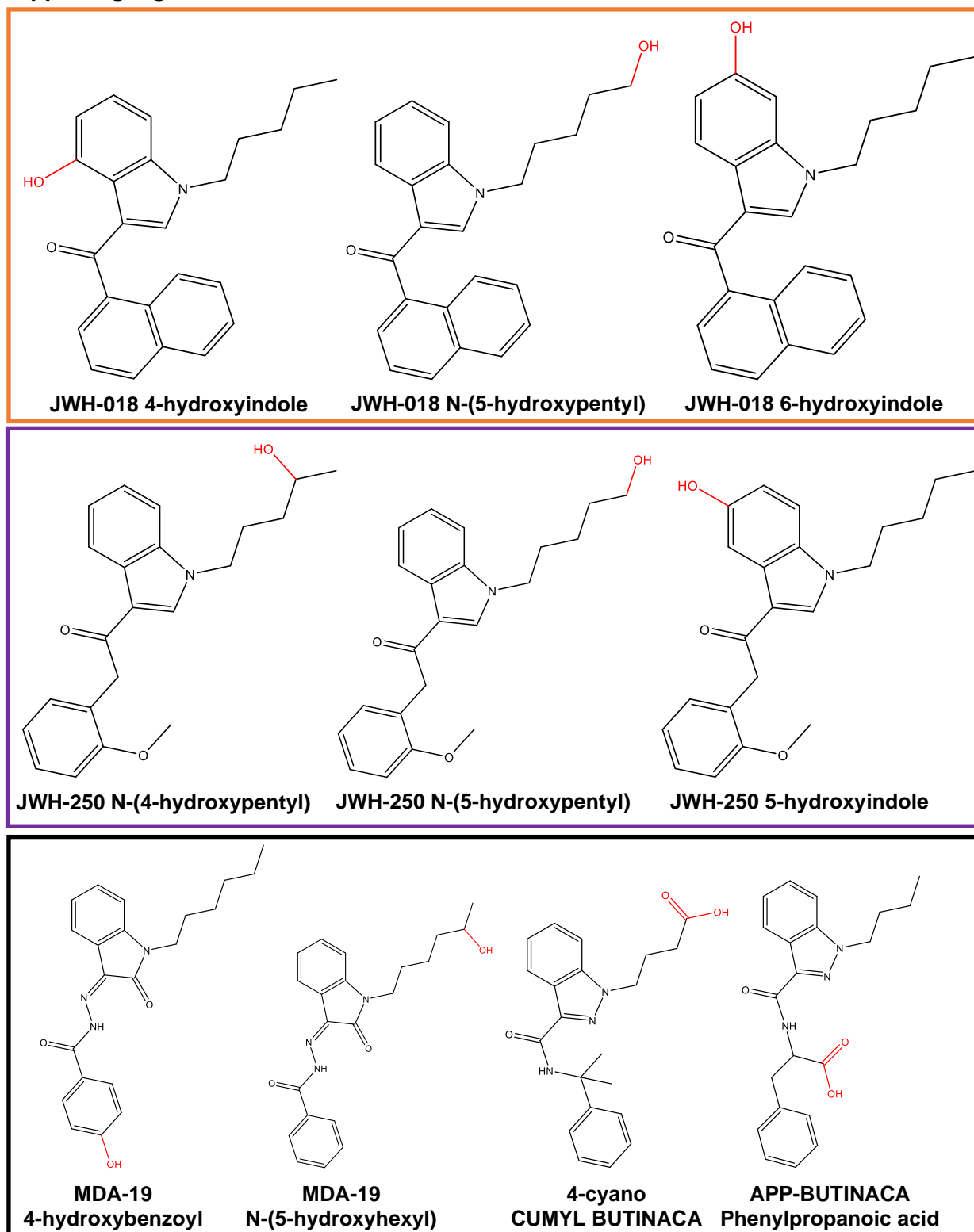
**Table S3.** Experimental  $^{DT}CCS_{N_2}$  and raw  $^{SLIM}CCS_{N_2}$  for the sodiated species,  $[M+Na]^+$ , for all metabolites. These  $^{SLIM}CCS_{N_2}$  values were derived based on a calibration using the Agilent Tune Mix ions. Chemical formulae and theoretical  $m/z$  are also included.

Metabolite	Formula	$[M+Na]^+$ $m/z$	$^{DT}CCS_{N_2}$ ( $\text{\AA}^2$ )	$^{SLIM}CCS_{N_2}$ ( $\text{\AA}^2$ )	$\Delta CCS$ (%)
JWH 018 4-hydroxyindole			206.4 $\pm$ 0.1	204.6 $\pm$ 0.1	-0.88%
JWH 018 N-(5-hydroxypentyl)	C <sub>24</sub> H <sub>23</sub> NO <sub>2</sub>	380.163	193.9 $\pm$ 0.1	191.6 $\pm$ 0.1	-1.19%
JWH 018 6-hydroxyindole			206.3 $\pm$ 0.1	207.5 $\pm$ 0.1	0.58%
JWH 250 N-(4-hydroxypentyl)			184.3 $\pm$ 0.1	186.1 $\pm$ 0.2	0.97%
JWH 250 N-(5-hydroxypentyl)	C <sub>22</sub> H <sub>25</sub> NO <sub>3</sub>	374.173	183.8 $\pm$ 0.1	181.1 $\pm$ 0.4	-1.48%
JWH 250 5-hydroxyindole			203.7 $\pm$ 0.1	200.8 $\pm$ 0.2	-1.43%
MDA-19 N-(4-hydroxybenzoyl)			211.1 $\pm$ 0.1	208.4 $\pm$ 0.3	-1.29%
MDA-19 N-(5-hydroxyhexyl)			190.8 $\pm$ 0.1	189.4 $\pm$ 0.2	-0.74%
4-cyano CUMYL-BUTINACA	C <sub>21</sub> H <sub>23</sub> N <sub>3</sub> O <sub>3</sub>	388.164	195.7 $\pm$ 0.1	193.5 $\pm$ 0.2	-1.13%
APP-BUTINACA phenylpropanoic acid			194.8 $\pm$ 0.1	192.7 $\pm$ 0.1	-1.08%

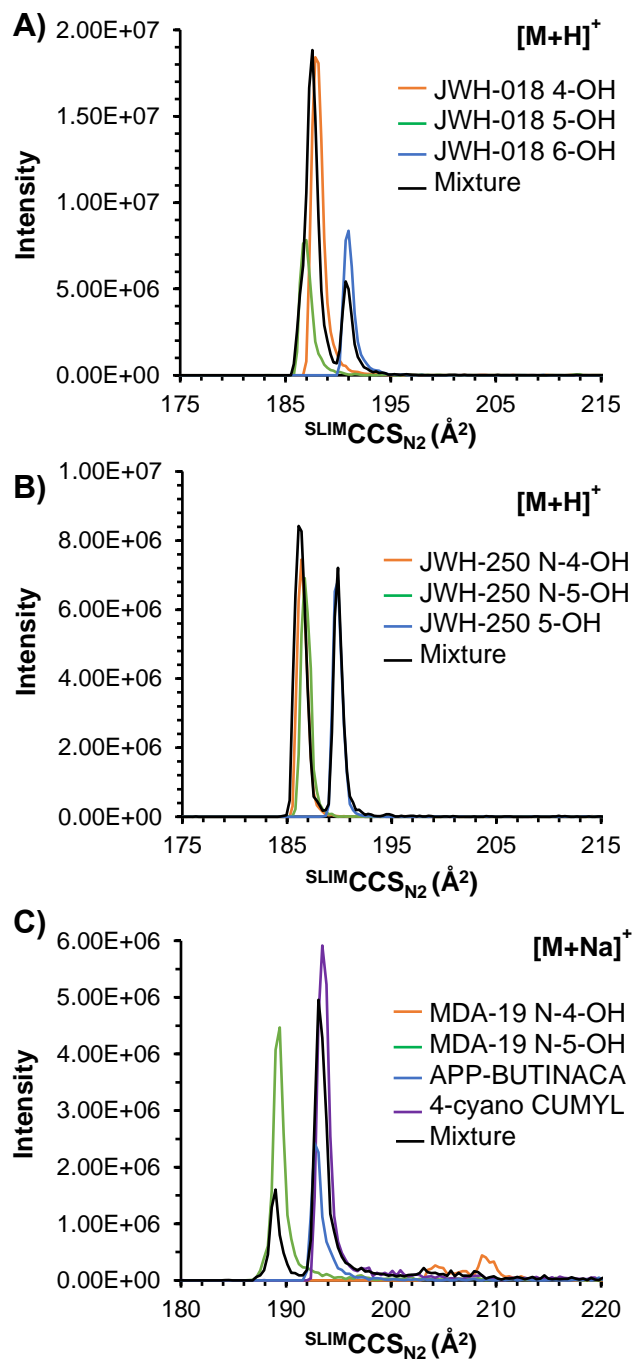
**Table S4.** Experimental  $^{DT}CCS_{N_2}$  and raw  $^{SLIM}CCS_{N_2}$  for the sodiated dimer species,  $[2M+Na]^+$ , for all metabolites. These  $^{SLIM}CCS_{N_2}$  values were derived based on a calibration using the Agilent Tune Mix ions. Chemical formulae and theoretical  $m/z$  are also included.

Metabolite	Formula	$[M+Na]^+$ $m/z$	$^{DT}CCS_{N_2}$ ( $\text{\AA}^2$ )	$^{SLIM}CCS_{N_2}$ ( $\text{\AA}^2$ )	$\Delta CCS$ (%)
<b>JWH 018 4-hydroxyindole</b>			$281.7 \pm 0.1$	$289.5 \pm 0.4$	2.73%
<b>JWH 018 N-(5-hydroxypentyl)</b>	$C_{24}H_{23}NO_2$	737.336	$261.9 \pm 0.1$	$267.9 \pm 0.1$	2.27%
<b>JWH 018 6-hydroxyindole</b>			$290.8 \pm 0.1$	$296.5 \pm 0.2$	1.94%
<b>JWH 250 N-(4-hydroxypentyl)</b>			$271.4 \pm 0.1$	$278.3 \pm 0.2$	2.51%
<b>JWH 250 N-(5-hydroxypentyl)</b>	$C_{22}H_{25}NO_3$	725.357	$259.1 \pm 0.1$	$266.4 \pm 0.2$	2.78%
<b>JWH 250 5-hydroxyindole</b>			$267.6 \pm 0.1$	$273.4 \pm 0.2$	2.14%
<b>MDA-19 N-(4-hydroxybenzoyl)</b>			$281.6 \pm 0.1$	$285.8 \pm 0.2$	1.48%
<b>MDA-19 N-(5-hydroxyhexyl)</b>			$268.4 \pm 0.1$	$272.8 \pm 0.1$	1.63%
<b>4-cyano CUMYL-BUTINACA</b>	$C_{21}H_{23}N_3O_3$	753.338	$264.5 \pm 0.1$	$268.2 \pm 0.3$	1.39%
<b>APP-BUTINACA phenylpropanoic acid</b>			$275.0 \pm 0.1$	$279.2 \pm 0.3$	1.52%

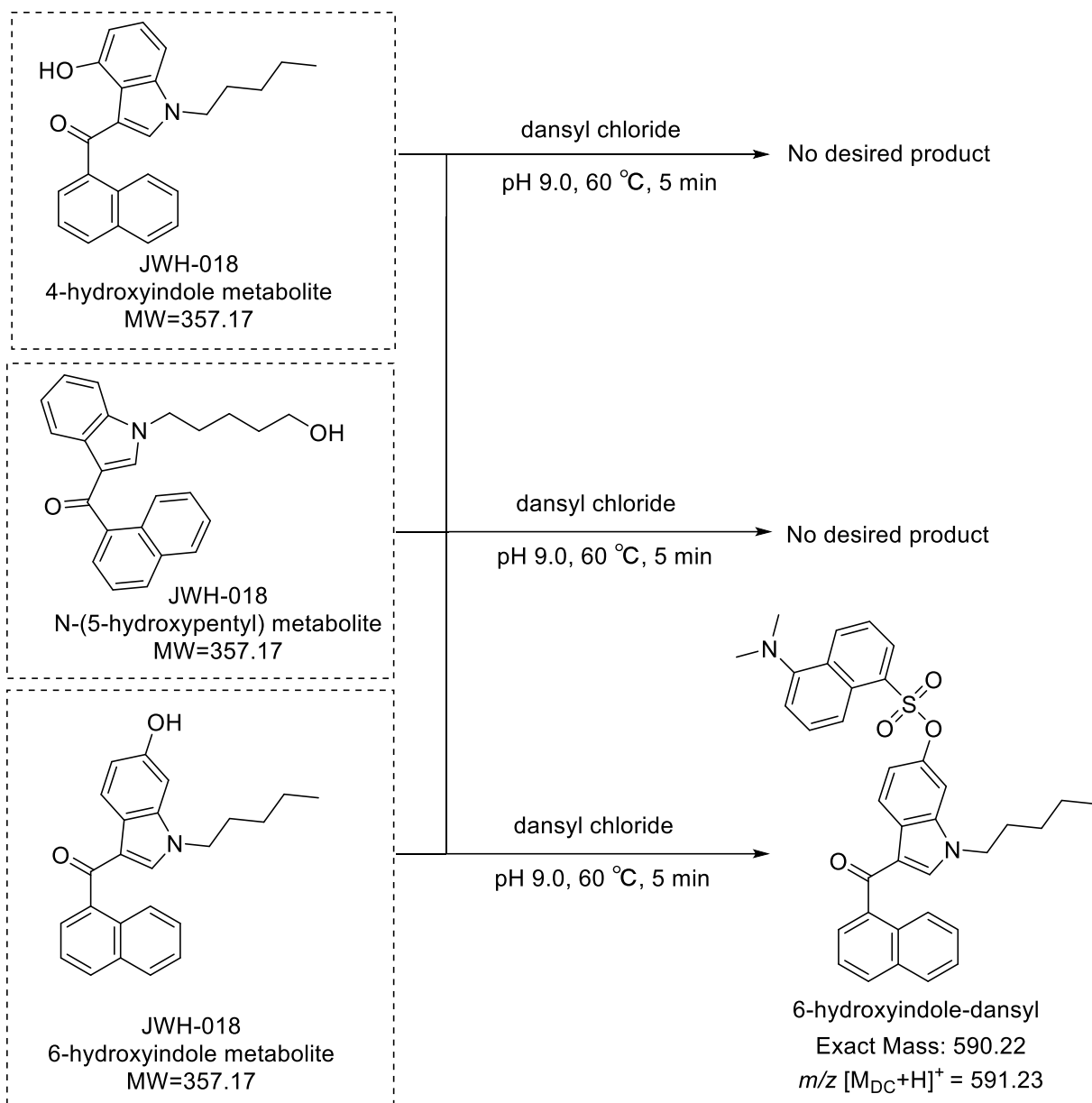
## Supporting Figures



**Figure S1.** Structures of the synthetic cannabinoid metabolites analyzed in this study, grouped by isomers with structural differences highlighted in red.



**Figure S2.** SLIM IM separations of (A) JWH-018 isomers as  $[M+H]^+$ , (B) JWH-250 isomers as  $[M+H]^+$ , and (C) MDA-19 isomers as  $[M+Na]^+$ . All groups are displayed as individual compounds and an equimolar mixture.



**Figure S3.** Reaction of dansyl chloride with JWH-018 metabolites selectively results in product formation for the 6-OH metabolite, yielding the  $[M_{DC}+H]^+$  product at  $m/z$  591.23.