### Article

Investigation of the time effect on the nanozyme-mediated oxidation of 3, 3'diaminobenzidine by hydrogen peroxide

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# Abstract

The time effect on the nanozyme-mediated oxidation of 3, 3'-diaminobenzidine by hydrogen peroxide was investigated. Silver-based nanomaterials were synthesized and then used as peroxidase mimics. The time-dependent activity of the 3, 3'-diaminobenzidine oxidation catalyzed by silver-based nanomaterials was calculated by probing the color intensity of the produced brown-colored indamine polymer during the reaction at different time intervals. The time-dependent activity curve was used as an index for evaluating the time effect on the process, showed that the concentration of the brown-colored indamine polymer was increased by increasing the oxidation time and then leveled off, revealing saturation of active nodes of nanoparticles with nanozyme-substrate.

*Keywords;* silver nanoparticles; peroxidase-mediated oxidation; 3, 3'-diaminobenzidine; indamine polymer; time effect on oxidation

# 1. Introduction

To date, several different nanomaterials with unique properties than the bulk materials have been designed and synthesized by different protocols. The above-mentioned nanomaterials have been applied for different applications in modern life due to their unique optical characteristics [1-3], catalytic activity [4, 5], anti-cancer, and medical properties [6, 7], as well as anti-bacterial characteristics [8, 9]. Some of the well-known nanoscale materials with a wide application range in science and technology are carbon/metal-based nanoparticles [10, 11], carbon dots and quantum dots [12, 13], metal oxides [14], and magnetic Fe<sub>3</sub>O<sub>4</sub> nanoparticles [15], as well as some nanoscale metalorganic frameworks [16, 17]. Some of these nanomaterials exhibit characteristic enzymelike activity and features that make them suitable as memetic materials for native enzymes. Due to the nanoscale size distribution and intrinsic enzyme-like properties of these nanomaterials, they are called "nanozymes" and have been widely applied for different practical uses in industrial, clinical, and environmental catalysis [18-21]. These nanozymes reveal some significant advantages over the native enzymes including lower cost efficiency and higher cycling stability [19, 22, 48, 49]. Due to their applicability for catalyzing enzyme-mediated reactions in harsh conditions, up to now, different nanoparticles with intrinsic peroxidase-like activity for example, Mn<sub>3</sub>O<sub>4</sub> nanozymes [23], Cu/CuFe<sub>2</sub>O<sub>4</sub> nanozymes [24], and BSA-modified manganese dioxide nanoparticles [25], as well as BSA-stabilized manganese phosphate nanoflower [26] had been designed

synthesized. Besides, the carbon-based nanozymes [27], silica-coated- Fe<sub>3</sub>O<sub>4</sub> magnetic nanoparticles [28], manganese dioxide ( $MnO_2$ ) and Fe<sub>3</sub>O<sub>4</sub> nanozymes s [29, 30], pyritebased, metal-organic frameworks-based, gold/silver-based, S/N co-doped carbon dotbased nanozymes [31, 32, 33, 34, 35]. Among the different nanomaterials with excellent peroxidase-like activity, gold/silver-based nanozymes have been widely for developing nanozyme-based sensors [36, 37], nanozyme-based cancer treatment [38], and nanozymemediated dye degradation [39]. Moreover, since the first report of patients infected with the new infection disease, COVID-19 in 2019 [40, 41], nanozyme-based methods have been developed for fast clinical diagnosis of this pandemic infection [42]. Hence, due to the importance of nanozymes, their biochemical characterization is an interesting topic. In this regard, the biochemical behavior of enzyme-like nanosilver was also investigated by our research group [43]. Besides, recently, Hormozi Jangi et al. reported some research articles on the investigation of biochemical behaviors of BSA-stabilized gold nanoparticles, silver nanoparticles, and MnO<sub>2</sub> nanoparticles [44-47]. Herein, the time effect on the nanozyme-mediated oxidation of 3, 3'-diaminobenzidine by hydrogen peroxide was investigated. Silver-based nanomaterials were synthesized and then used as peroxidase mimics. The time-dependent activity of the 3, 3'-diaminobenzidine oxidation catalyzed by silver-based nanomaterials was calculated by probing the color intensity of the produced brown-colored indamine polymer during the reaction at different time intervals. The time-dependent activity curve was used as an index for

evaluating the time effect on the process, showed that the concentration of the browncolored indamine polymer was increased by increasing the oxidation time and then leveled off, revealing saturation of active nodes of nanoparticles with nanozymesubstrate.

### 2. Experimental section

#### 2.1. Synthesis of AgNPs

The synthesis was performed based on the process reported [27]. To do this, silver ions were reduced by NaBH<sub>4</sub> in the presence of sodium citrate as a stabilizer within 3 hours. After this time, the AgNPs were collected and stored at 4 °C.

### 2.2. Oxidation reactions

To do the oxidation reactions, a suitable amount of DAB was introduced into the buffer solutions containing silver nanoparticles and hydrogen peroxide with a fixed pH of 7.0. The reaction proceeded for about 0.0000-25.0 min for DAB oxidation. Thereafter the colored products were analyzed by UV-Vis spectrophotometer at 460.0 nm for detecting the DAB-ox.

# 3. Results and discussion

#### **3.1.** Characterization of silver nanozymes

The silver-based nanoparticles were synthesized using citrate as the capping and stabilizer. The as-synthesized silver nanoparticles were then characterized via the TEM imaging method for calculation of their mean size as well as for the morphological properties. The TEM image of these nanoparticles shown in Figure 1 exhibited that the as-prepared nanozymes have a spherical morphology and uniform particles. Besides, the mean size of these nanoparticles was estimated at about 11.0 nm for the TEM image.



Figure 1. TEM image of as-prepared silver nanoparticles.

# 3.2. Time-course studies toward DAB oxidation

the peroxidase-like activity of silver nanoparticles toward DAB oxidation was also evaluated. To evaluate the peroxidase-like activity of the as-prepared AgNPs against DAB, the oxidation of DAB was performed by hydrogen peroxide in the presence of AgNPs as peroxidase mimics. In this regard, the time course studies were performed by probing the brown-colored product via spectrophotometric detection at 460.0 nm. Afterward, the plot of oxidation of DAB in the presence of AgNPs as a function of time was constructed by plotting the absorbance at 460.0 nm as a function of reaction time. The results are shown in Figure 2. As can be seen from this figure, the AgNPs can catalyze the oxidation of DAB to form a brown-colored product with a maximum absorbance at 460.0 nm.



**Figure 2.** Time course radar plot of oxidation of DAB in the presence of silver nanozymes as a function of time

To explore more precise on the nanozymatic activity of the as-prepared silver nanozymes toward oxidation of DAB at different incubation times, the relative activity of nanozymes was also calculated and used as an index for investigating the time effect on the oxidation process of DAB for producing the corresponding poly(DAB). The results are shown in Figure 3. The results showed that after a long oxidation time as long as five minutes, the nanozyme activity reached about 32% of its maximal activity toward DAB oxidation. The oxidation process slowly proceeded and the nanozyme activity reached about 54% after 12.0 min. The maximal activity of silver nanozymes was obtained after 20 min toward DAB oxidation. After this time, the incubation time could not affect the production of the poly(DAB), and therefore the relative activity of the nanozymes was leveled off. The results reveal that an incubation time over 25.0 min was enough for active nodes presented on the surface of the silver nanoparticles, to be blocked by the substrate (DAB) molecules. In fact, the active nodes on the surface of the silver nanoparticles were completely saturated in 25.0 min by the DAB molecules. Considering this fact that the active nodes (the binding sites) on the surface of the nanozymes are limited, the saturation of the DAB molecules leads to leveling off of the relative activity of silver nanzymes.



Figure 3. Relative activity of silver nanozymes for oxidation of DAB as a function of time

# 4. Conclusions

Time effect on the nanozyme-mediated oxidation of 3, 3'-diaminobenzidine by hydrogen peroxide was investigated. Silver-based nanomaterials were synthesized and then used as peroxidase mimics. The time-dependent activity of the 3, 3'-diaminobenzidine oxidation catalyzed by silver-based nanomaterials was calculated by probing the color intensity of the produced brown-colored indamine polymer during the reaction at different time intervals. The time-dependent activity curve was used as an index for evaluating the time effect on the process, showed that the concentration of the browncolored indamine polymer was increased by increasing the oxidation time and then leveled off, revealing saturation of active nodes of nanoparticles with nanozymesubstrate.

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### **Conflict of interest**

None.

# 5. References

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