Sulfamate Tethered *Aza***-Wacker Strategy for a Kasugamine Synthon**

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Supporting Information Placeholder

ABSTRACT: We present our preparation of a kasugamine synthon, which proceeds in 14 steps from a literature epoxide. We expect that this kasugamine derivative can be used for the total syntheses of kasugamycin, minosaminomycin, and analogue antibiotics. A key step in the synthesis is our laboratory's sulfamate-tethered *aza*-Wacker cyclization.

Kasugamycin and minosaminomycin are structurally related aminoglycoside antibiotics (**Figure 1**). Kasugamycin was first isolated in 1965 from a bacterial strain (*Streptomyces kasugiensis*) found in a soil sample near the Kasuga shrine in Nara, Japan.¹⁻³ Shortly after isolation, empirical testing demonstrated that kasugamycin was active against a variety of bacteria and fungi.^{4, 5} While its antibacterial activity is much less than that of other members of the aminoglycoside family, its ability to control rice blast disease caused by the fungus *Pyricularia oryzae* was so remarkable that it continues to be used commercially as an agricultural fungicide (Kasumin).^{6, 7} Minosaminomycin is a structurally related antibiotic to kasugamycin and has demonstrable activity against a variety of pathogenic bacteria, including mycobacteria. $8,9$

Our interest in kasugamycin and minosaminomycin stems from both fundamental and applied considerations. These antibiotics have been synthetically unexamined for close to fifty years.^{8, 10-16} Our laboratory has a deep interest in the development of new reactions which would enable the syntheses of complex antibiotics. Specifically, we envisioned that our laboratory's sulfamate-tethered *aza*-Wacker cyclization could serve as a key step for the assembly of the kasugamine portion of these antibiotics.17-24 Our interest in designing practical synthetic routes to antibiotics with new mechanisms of action²⁵⁻³¹ relative to approved agents is motivated by the global rise in resistance to antibacterial and antifungal agents currently in use.32-34 It is our hope that *de novo* syntheses will allow for the rapid preparation of analogues³⁵⁻³⁷ and will serve as a start of a drug-discovery effort to replenish worrisome antibacterial and antifungal pipelines. In this account, we describe our preparation of a kasugamine synthon, a common building block for both kasugamycin and minosaminomycin.

 There are two general strategies for the syntheses of unusual monosaccharides.³⁸ One is the classic

"heterocycle→heterocycle" approach; in such syntheses, chemists begin with a more readily available monosaccharide and then systematically manipulate it to the end target. A second, less explored approach is to commence with a linear precursor, build functional group complexity, and then effect a late-stage cyclization ("linear→cyclic"). The former approach generally has fewer total step counts; on balance, due to existing complexity, monosaccharides can be very challenging starting materials, whose modifications generally require lengthy protection-deprotection sequences. For this reason, we prefer the latter approach, because of the greater flexibility it affords and chose kasugamine derivative **A** as our first target (**Scheme 1**).

Figure 1. Kasugamycin and minosaminomycin are potent aminoglycoside antibiotics and are structurally related.

Scheme 1 Retrosynthetic analysis.

In accord with this plan, our retrosynthesis is depicted in **Scheme 1**. After a late stage hemiacetalization of heteroatom-rich linear precursor **C**, deoxygenation and azide reduction would furnish target **A**. **C** would be built systematically from epoxide **F**. 39, 40 We envisioned our laboratory's sulfamate-tethered aza -Wacker cyclization^{17, 21} would serve as a key step during the transformation of epoxide **F** into aldehyde **C**. The successful execution of this strategy would increase the prominence of both classical and tethered *aza*-Wacker chemistry for the stereoselective assembly of complex targets.^{19, 41}

Scheme 2 Opening sequence of reactions.

Our forward synthesis (**Scheme 2**) commenced with morpholine amide **1**, itself prepared in two steps from commercial Derythrono-1,4-lactone in a procedure developed and disclosed by our laboratory.¹⁷Morpholine amide **1** was converted into diol **2** by heating with NaBH⁴ in EtOH. Treatment of diol **2** with benzoyl chloride in ice-cold CH₂Cl₂ allowed for selective esterification of the primary alcohol. The secondary alcohol of benzoate **3** was tosylated using a standard mixture of NEt3, TsCl, and catalytic DMAP. Treatment of 4 with mixture of K_2CO_3 in

Entry	Catalyst	Additional Parameters	Solvent	Yield
ı	Pd(OAc) ₂ (10%)	$Cu(OAc)2$ (1 equiv.) O_2 (1 atm.), 60 °C, 16 h	CH3CN	36%
2	Pd ₂ (dba) ₃ (15%)	$Cu(OAc)2$ (1 equiv.) O_2 (1 atm.), 60 °C, 16 h	CH3CN	50%
3	PdCl ₂ (nbd) (15%)	$Cu(OAc)2$ (1 equiv.) O_2 (1 atm.), 60 °C, 16 h	CH3CN	30%
4	Pd (CH ₃ CN) ₂ Cl ₂ (15%)	$Cu(OAc)2$ (1 equiv.) O_2 (1 atm.), 60 °C, 16 h	CH3CN	41%
5	Pd (CH ₃ CN) ₂ Cl ₂ (15%)	$Cu(OAc)2$ (1 equiv.) O_2 (1 atm.), 60 °C, 16 h	DMSO	41%
6	Pd(CH ₃ CN) ₂ Cl ₂ (15%)	$Cu(OAc)2$ (1 equiv.) $O2$ (1 atm.) 4A molecular sieves Fmoc-Gly-OH (1 equiv.) 60 °C. 16 h	DMSO	63%

Scheme 3. Optimization of the key sulfamate-tethered aza-Wacker cyclization. MeOH allowed for the hydrolysis of the benzoate ester; the resulting alkoxide anion displaced the tosylate in an intramolecular S_{N2} reaction to form epoxide 5. CuI mediated the regioselective opening of epoxide **5** by *cis*-1-propenyl-magnesium bromide. For this transformation, we have found that it is critical to freshly prepare *cis*-1-propenylmagnesium bromide; for reasons unknown, using commercial solutions of 1-propenylmagnesium bromide did not allow for clean reactivity. Sulfamoylation of alcohol **6** proceeded smoothly with Johnson and Magolan's protocol.⁴²

As a first step to evaluating the sulfamate-tethered *aza*-Wacker cyclization with **7**, we tested our standard protocol of Pd(OAc) α /Cu(OAc) α in hot CH₃CN and a \sim 1 atm O α atmosphere. We were pleased that this delivered desired oxathiazinane **8** in a 36% yield, an excellent starting point for further optimization (**Scheme 3**, **Entry 1**). Switching to $Pd_2(dba)$ at a higher loading increased the yield to 50% (**Scheme 3**, **Entry 2**). Using PdCl₂(nbd) and Pd(CH₃CN)₂Cl₂ in place of Pd₂(dba)₃ did not help the reaction (**Scheme 3**, **Entries 3–5**). We have previously observed a dramatic improvement in sulfamate-tethered *aza*-Wacker cyclizations when mono-protected amino acid ligands and 4Å molecular sieves are used. We hypothesize, in accord with the Yu laboratory's observations, $43-45$ that mono-protected amino acids (MPAA) ligands help promote the Pd (II) – Pd (0) catalytic cycles operative during *aza*-Wacker transformations. Indeed, in line with our predictions, the yield of oxathiazinane **8** increased to 63% upon the addition of MPAA ligand Fmoc-Gly-OH and 4Å molecular sieves (**Scheme 3**, **Entry 6**). Despite the presence of molecular sieves, it was critical to use anhydrous DMSO for the best reaction yield. X-ray structural analysis of **8** (**CCDC: 2289376**) allowed us to unambiguously confirm its identity and absolute stereochemistry.

Moving forward, a Cbz group was appended to the oxathiazinane nitrogen for a subsequent ring-opening with NaN³ (**Scheme 4**).46, 47 The alkene of **10** was cleaved by dihydroxylation followed by treatment with NaIO4. Aldehyde **12** was unstable to purification and was immediately treated with TsOH to form hemiacetal **13** as a mixture of anomers. This anomeric mixture was bis-acetylated with Ac2O/pyridine, and the anomeric acetate was exchanged with thiophenol using BF₃•OEt₂.

Scheme 5 (A) Displacement of the tosylate by hydride does not give expected product under multiple conditions. (B) Upon treatment with Pd/C and H_2 gas, deiodination and azide reduction occur in one step to furnish the desired target.

For **15**, we have assigned the configuration at the anomeric carbon based on the small coupling constant of the anomeric proton (~1.5 Hz) and by analogy to the crystal structure of a later intermediate (*vide infra*). The acetate of **15** was removed using K_2CO_3 in MeOH.

Initially, it was our plan to tosylate the alcohol of **16** and to displace it with an appropriate hydride source (**Scheme 5A**). Accordingly, tosylate **17** was prepared by stirring **16** with NEt³ and TsCl. However, despite multiple trials, including with NaBH4, LiAlH4, and NaBH3CN, we were unable to cleanly obtain **18**, likely because of the presence of an azide functionality. We thus sought to develop a protocol which would allow for the clean formation of the methyl group and the primary amine of kasugamine in a single step. Knowing that there was precedent for the Pd-catalyzed hydrogenolysis of azides and alkyl iodides,⁴⁸ we prepared iodide **19** using a Finkelstein reaction (**Scheme 5B**). Indeed, stirring 19 with NEt₃ and Pd-C under an atmosphere of H² gas cleanly delivered kasugamine derivative **20**. Crystal structure analysis of **20** (**CCDC 2291427**) allowed us to unambiguously confirm its identity and absolute stereochemistry.

In summary, we have completed a concise synthesis of a kasugamine synthon, which we envision can be used in preparations of kasugamycin, minosaminomycin, and related analogues. A highlight of the synthesis is our sulfamate-tethered *aza*-Wacker cyclization. The march towards total syntheses of these antibiotics continues in our laboratory.

ASSOCIATED CONTENT

Data Availability Statement: The data underlying this study are available in the published article and its Supporting Information. Supporting information: Contains additional experimental details including NMR Spectra and X-ray Crystallographic Information.

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