# Machine Learning-Guided Discovery of Polymer Membranes for CO<sub>2</sub> Separation

1

2

Yasemin Basdogan,<sup>1</sup> Dylan R. Pollard,<sup>2</sup> Tejus Shastry,<sup>3</sup> Matthew R. Carbone,<sup>4</sup> Sanat K. Kumar,<sup>3</sup> and Zhen-Gang Wang<sup>1\*</sup>

<sup>1</sup>Chemical Engineering, California Institute of Technology, Pasadena, CA 91125, USA <sup>2</sup>Chemical Engineering, Auburn University, Auburn, AL 36830, USA <sup>3</sup>Chemical Engineering, Columbia University, New York, NY 10027, USA <sup>4</sup>Computational Science Initiative, Brookhaven National Laboratory, Upton, NY 11973, USA

\*To whom correspondence should be addressed; E-mail: zgw@caltech.edu.

Designing polymer membranes with high gas permeability and selectivity remains a grand challenge for energy, the environment, and economic sustainability. Increasing both the selec-5 tivity and permeability is a difficult multi-task constrained design problem for polymer mem-6 branes due to the trade-off between these two properties. The complexity of chemical com-7 position and morphology of polymers makes this problem especially hard to attack with trial-8 and-error or intuition-based strategies. In this work, we instead present a machine learning 9 (ML)-driven genetic algorithm to tackle the design problem of polymer membranes for  $CO_2$ 10 separation from  $N_2$  and  $O_2$ . Using literature data of permeability for three gases,  $CO_2$ ,  $N_2$ , and 11 O<sub>2</sub>, we constructed multiple ML models using different fingerprinting featurization schemes to 12 predict all three gas permeabilities as well as the CO<sub>2</sub>/N<sub>2</sub> and CO<sub>2</sub>/O<sub>2</sub> selectivity values. Then, 13 we employed a genetic algorithm to design new polymers and evaluated their performance with 14 respect to the Robeson upper bounds using our machine learning models. We were able to iden-15 tify new polymer membranes that are promising for both  $CO_2/N_2$  and  $CO_2/O_2$  separations. The 16

<sup>17</sup> top discovered polymers are predicted to have high glass transition temperatures,  $T_g$ . Similarly, <sup>18</sup> the pyridine functionality was found in  $\approx 20\%$  of the predicted polymers. Both of these facts <sup>19</sup> are well in line with currently accepted experimental wisdom for CO<sub>2</sub> based separations. The <sup>20</sup> framework developed here can be used to design polymers for any application involving con-<sup>21</sup> strained optimization. Finally, we outlined the strengths and limitations of this approach, as well <sup>22</sup> as the imminent challenges and opportunities with using machine learning guided data-driven <sup>23</sup> inverse design of polymers.

#### <sup>24</sup> Introduction

The increased concentration of CO<sub>2</sub> in the atmosphere is the single most important anthro-25 pogenic cause of global warming. Decreasing the release of CO<sub>2</sub> into the atmosphere requires 26 efficient CO<sub>2</sub> capture and separation technologies. Decades of research have been devoted to 27 improving existing gas separation technologies, but there is still an imminent need to find new 28 methodologies given the current course of climate change (1). Traditional unit operations have 29 the ability to isolate high-purity products, but they have a high carbon footprint due to the high 30 energy requirements. Membrane-based technologies are an attractive alternative because they 31 provide savings in capital and energy-related operating costs, and offer advantages related to the 32 ease of operation and compact environmental footprint (2-4). Polymer membranes have been 33 successfully investigated for H<sub>2</sub> recovery, N<sub>2</sub> generation, but there is still a significant opportu-34 nity to improve polymer membrane technology for CO<sub>2</sub> separations. Although hundreds of new 35 materials are synthesized each year, most of the commercial membranes used today are from 36 the 1990s, and they rely on a dozen or so common polymer structures. This is largely because 37 the two properties that are important for a membrane material - high flux (permeability) and 38 high gas purity (selectivity) – are inversely correlated. This inverse relationship between gas se-39 lectivity and permeability was first examined by Robeson in 1991 (5) and revisited in 2008 for 40

<sup>41</sup> pure homopolymer membranes (6) and is famously known as the Robeson Upper Bound. Since
<sup>42</sup> then, there have been considerable efforts in designing polymers that are above the empirically
<sup>43</sup> determined upper bound for a given application (7–9).

Designing polymers with targeted structural and functional properties is challenging due 44 to the practically infinite polymer chemistry design space. Trial-and-error or intuition-based 45 strategies are not efficient, and they are likely to miss optimal solutions due to the complex-46 ity of chemical composition and morphology of polymers. Furthermore, these strategies with 47 traditional experimental and computational routes are time and resource consuming. Machine 48 Learning (ML) models trained on polymer data sets can mitigate this problem, as it is possi-49 ble to predict a new material's properties instantaneously by interpolating within an existing 50 dataset (10). There have been a number of studies in the recent literature that leveraged ML 51 to predict the properties of polymers. For example, Alves *et al.* developed models to discover 52 polymeric micelle formulations for poorly soluble drugs using micellar solubilization data (11). 53 Tao et al. used ML models to predict the glass transition temperature of a polymer based on 54 its structural formulation (12). Later, these authors also did a benchmarking study to compare 55 the predictive power of numerous ML models and showed the importance of structure and fea-56 ture representations (13). Xu et al. used ML models to study swelling of polymer membranes 57 in different solvents with chemically informed molecular representations and descriptors (14). 58 Wang et al. used ML models to screen polymers for pervaporation separation (15) and devel-59 oped a data-driven approach to predict the fractional free volume of polymers (16). There are 60 also excellent review articles published in the last couple of years that summarize the recent 61 developments in ML studies of polymer properties (17–26). 62

The success of applying ML models to design new polymer membranes for gas separation has been comparatively lacking, largely owing to limitations in data availability. Barnett *et al.* used experimental gas permeability data to develop a ML model to predict gas separation

in polymer membranes (27). They have successfully identified several polymers for improved 66 CO<sub>2</sub>/CH<sub>4</sub> separation and synthesized two of them to experimentally validate the ML predic-67 tions. Yuan et al. used ML algorithms to predict the missing values for the permeability of 68 different gases in the online Polymer Gas Separation Membrane Database of the Membrane 69 Society of Australasia (28). Yang et al. used the same data set and leveraged ML models to pre-70 dict gas permeability based on the polymer chemistry (29). However, a ML model that predicts 71 polymer properties by itself does not lead to the discovery of new polymer membranes with op-72 timal properties. In principle, one can propose many candidate polymers, possibly at random, 73 and use ML to predict their performance. This is obviously not an efficient strategy. A ML 74 "forward model" needs to be coupled with an inverse design/generative algorithm to efficiently 75 explore the polymer material space. Genetic algorithms (GA) are an example of a data-driven 76 inverse design method, which can be effectively coupled with an ML model. Srinivasan et al. 77 used GA to design single-stranded DNA grafted colloids (30). These authors were able to re-78 produce the experimentally validated phase diagram and additionally identify the formation of 79 four previously unobserved crystal structures. Kim et al. demonstrated one of the first data-80 driven inverse design methods of new polymers having high band gap and high glass transition 81 temperature that is relevant for high-temperature and high-energy density dielectrics (31). They 82 successfully identified new polymer structures with the desired properties. 83

In this work, we follow a similar procedure to Kim *et al.* to design new polymer membranes with the desired selectivity and permeability for  $CO_2$  separation from  $N_2$  and  $O_2$ . First, we start by assembling a library of gas permeabilities corresponding to the experimental studies of various polymers. Next, we train multiple ML models based on various fingerprints to determine which ML model performs the best in predicting gas permeability. Then, we use our ML models to drive a GA for 100 generations and create more than 16000 new polymer structures. We also use different fitness functions to design the best possible polymers given our initial data set. Application of this combined ML-GA framework results in the discovery of more than 20 new polymers that are above both the  $CO_2/N_2$  and  $CO_2/O_2$  Robeson upper bounds, many of which contain aromatic functional groups along with oxygen- and nitrogen- motifs, aligning with experimental observations that show imines and polyethers as promising polymer membranes for  $CO_2$  separation (*32–36*).

### 96 **Results**

We compiled a literature database of permeability for three gases –  $CO_2$ ,  $N_2$ , and  $O_2$  – in a 97 variety of polymers at a temperature range of 300-330 K. The number of data points for each gas 98 is different due to the availability of data in the literature, so we only considered polymers that 90 have permeability measurements for all three gasses. This resulted in 780 different polymers 100 in our library, which represent a sizable portion of the polymers that are typically included in 101 the most up-to-date Robeson plots. The selectivity versus permeability data for CO<sub>2</sub>/N<sub>2</sub> and 102  $CO_2/O_2$  are shown in Figure 1. We see that there are only three polymers that are above the 103  $CO_2/N_2$  Robeson upper bound (37, 38). These polymers have a benzene ring and ether oxygen 104 functional groups in common, which are known to be favorable for  $CO_2$  separation. There are 105 more than ten polymers that are above the  $CO_2/O_2$  upper bound as shown in Figure 1b. The list 106 of all polymers in our library and permeability measurements are provided in the SI. 107

The first step in applying ML models to evaluate physical properties is choosing an appropriate mathematical form to be used as input. This is commonly known as featurization (or fingerprinting in the chemo-informatics literature) and it is of critical importance to the quality and interpretability of the ML models. We start with generating the simplified molecular-input line-entry system (SMILES) (*39, 40*) representations of our polymers based on their repeating units. We cap the two ends of the monomer structure with hydrogen atoms to create a consistent data set. Based on our SMILES strings, we use two common fingerprints in the literature, the



Figure 1: Robeson plot of selectivity versus permeability for (a)  $CO_2/N_2$  and (b)  $CO_2/O_2$  separations. The 2008 Robeson upper bounds (6) are shown as dashed black lines. Color code represents the different classes of polymers. Each data points represent a single polymer.

Extended Connectivity Fingerprint with bond diameter four Angstroms (ECFP4) (41, 42) and 115 the Molecular ACCess System (MACCS) (43, 44) fingerprint. MACCS is a common substruc-116 ture keys-based fingerprint consisting of a binary vector of 166 bits depending on the presence 117 of certain substructures or features from a given list of structural keys (45). ECFP4 is an ex-118 ample of a topological fingerprint that is based on analyzing all the fragments of the molecule 119 by looking at the environment of each atom up to a set radius, and then hashing every one of 120 these environments to create the fingerprint. One needs to be careful when using hashed finger-121 prints because a bit cannot be traced back to a given feature, and this may result in a given bit 122 corresponding to more than one different feature, which is called "bit collision" (46). We use 123 ECFP4, based on the Morgan algorithm (47), which is a 2048 bit fingerprint as implemented in 124 RDKit. Figure 2 shows the comparison for predicting  $CO_2$  permeability with the random forest 125 regression model using both fingerprints. We fit and plot the logarithmic permeability values to 126 better visualize the data set. Both fingerprints result in  $R^2$  value of 0.982 for the training set. 127 However,  $R^2$  for the test set is considerably higher when we use ECFP4 as shown in Figure 2. 128 We also compare the root mean square error (RMSE) of the fits for both fingerprints. The test 129

set RMSE with ECFP4 fingerprint is 0.131, and the test set RMSE with MACCS fingerprint is
0.161. Thus, we use ECFP4 to train and test our ML models for the rest of this paper.



Figure 2: Comparison of CO<sub>2</sub> permeability model predictions using (a) ECFP and (b) MACCS fingerprints with random forest regression algorithm.

We start with randomly splitting our data set into one of two categories for each gas; one 132 is used for training the ML model, while the other is initially withheld during training. The 133 training data sets were 80% of our total database for each gas, which represents more than 600 134 polymers for each gas. We then apply the trained model to the remaining 20% of the polymers 135 (test set) and use these data as verification of the model's accuracy. Then, we employ various 136 ML models on the training sets including support vector regression (SVR), k-nearest neighbors 137 (KNN), decision tree, and random forest regression. Next, we compare the predictive power of 138 these popular ML regression models on CO<sub>2</sub> permeability values. 139

First we study SVR, which has the ability to consider non-linearity in the permeability data (48). SVR results in  $R^2$  value of 0.84 and 0.203 RMSE on the test set. KNN regression, which predicts the target value by local interpolation of the targets associated with the nearest neighbors in the training set, results in  $R^2$  value of 0.822 and 0.242 RMSE on the test set. Then, we employed a decision tree regression model, which uses a tree structure and inference layer to

achieve the final decision of the modeling results (18). Decision tree regression performs better 145 than both the SVR and KNN regression with  $R^2$  value of 0.881 and 0.148 RMSE on the test set. 146 Finally, to make a more accurate prediction, we used a random forest regression model, which 147 is an ensemble learning method for regression that combines predictions from multiple decision 148 tree models. As expected, random forest predictions are better than all the other algorithms that 149 we have tried with  $R^2$  value of 0.941 and 0.135 RMSE on the test set. Figure 3 summarizes the 150 different ML regression models that we have tried. We note that Yang et al. compared random 151 forest regression models with deep neural networks (DNN) and showed DNN model performs 152 better than the random forest regression model (29). However, DNNs typically require much 153 more data than what is available for this study. 154

To determine where on the Robeson plot a polymer is located, we need to be able to predict 155 the CO<sub>2</sub>/N<sub>2</sub> and CO<sub>2</sub>/O<sub>2</sub> selectivity as well as the CO<sub>2</sub> permeability. The ideal selectivity  $\alpha_{i/i}$ 156 for the gas pair is the ratio of the permeabilities  $P_i$  and  $P_j$ . Thus, we need ML models to predict 157 N2 and O2 permeability as well. Because the random forest regression model is the best per-158 forming model for the CO<sub>2</sub> permeability, we have continued using random forest regression for 159 the  $N_2$  and  $O_2$  permeability. Figure 4 shows model predictions for the  $N_2$  and  $O_2$  permeability. 160 For both gases we can predict the gas permeability with R<sup>2</sup> values higher than 0.9. The RMSE 161 for  $N_2$  and  $O_2$  are 0.171 and 0.147, respectively. This demonstrates that we can predict all 162 three gas permeabilities accurately with the random forest regression model. Now that we have 163 established an accurate ML model to predict a polymer membrane's performance with respect 164 to the Robeson upper bounds, we start designing new polymers with a GA and evaluate their 165 performance on the fly with these ML models. 166

The first step of the GA is to construct the "gene pool" that will be used to create the initial parent polymers. We used the "Breaking of Retrosynthetically Interesting Chemical Substructures" (BRICS) algorithm as implemented in the RDKit Python package to get the chemical



Figure 3: Comparison of  $CO_2$  permeability model predictions using (a) SVR, (b) KNN regression, (c) Decision tree regression, and (d) Random forest regression models with ECFP fingerprints.

building blocks, or fragments, from our polymer library (49). A total of 79 unique fragments 170 were extracted from 780 reference polymers. Figure 5 shows six functional groups that appear 171 most frequently in our library. To initiate the GA process, 100 parent polymers consisting of 4 172 building blocks in their monomer unit were created in the first generation. The fragments were 173 chosen randomly from our gene pool of the 79 chemical fragments. Then, 15 families with the 174 smallest Tanimoto similarity score (50), with 3 parents in each family, were chosen to perform 175 crossover and mutation operations to alter their sequence of chemical building blocks, resulting 176 in 12 offspring polymers in each family. During crossover, two parents generate an offspring 177 by combining one random segment from a parent with another random segment from the other 178



Figure 4: (a)  $N_2$  and (b)  $O_2$  permeability random forest regression model predictions with ECFP fingerprints.

parent. The segmentation point of a parent polymer was chosen according to a Gaussian dis-179 tribution with a mean at the center of the sequence and standard deviation of 0.3 blocks. We 180 also applied mutation operations on 60% of the genes to increase the chemical diversity, where 181 we randomly selected the building block in the sequence and replaced it with a new building 182 block randomly chosen from the list of the 79 blocks. In each GA iteration, the top perform-183 ing offspring polymers with the highest fitness evaluation were retained as parents to create the 184 next generation offspring polymers. We also assigned 10% migration rate between different 185 families, whereby the highest-scoring polymers that were not selected as a parent migrated to 186 a different family. An essential component in this evolutionary process is the polymer property 187 estimation, which traditionally has been evaluated by experiments that are very time-consuming 188 and expensive; here, we use our ML models for on-the-fly polymer property estimation. 189

We ran the GA for 100 generations and generated more than 16000 new polymer structures as shown in Figure 6. All the new polymers generated with the GA are reported in the SI, where we highlight the top performing 100 polymers. We optimized multiple parameters in the GA framework to guide the evolutionary process towards the targeted design area. First,



Figure 5: The six most chemical functional groups that appear in our library of experimental polymers using the BRICS algorithm. "A" represents the binding sites.

we ran the GA with an ML model trained with two different fingerprints, ECFP and MACCS 194 keys, and found that the fingerprint used in the ML does not influence the top performing poly-195 mers identified from the GA framework. Next, we tried running the GA for an additional 100 196 generations to see if running the GA for longer will result in better performing polymers, but 197 found that the additional iterations did not result in any improved polymer structures. Finally, 198 we tried multiple fitness functions to optimize the evolutionary trajectory. To optimize both 199  $CO_2/N_2$  and  $CO_2/O_2$  selectivity as well as  $CO_2$  permeability, one needs a fitness function that 200 includes all three metrics. However, the functional form of the fitness function is not clear a 201 *priori*. We tried the fitness function  $\log(P_{CO_2}) \times \alpha_{CO_2/N_2} \times \alpha_{CO_2/O_2}$  and showed that GAs 202 using this fitness function failed to identify new polymer structures that are above both up-203 per bounds. We found that the fitness function based on the actual  $P_{CO_2}$  permeability values 204  $(P_{CO_2} \times \alpha_{CO_2/N_2} \times \alpha_{CO_2/O_2})$  did result in several polymers that are above both the CO<sub>2</sub>/N<sub>2</sub> 205 and  $CO_2/O_2$  upper bounds as shown in Figure 6. Because  $CO_2$  permeability values are gener-206 ally orders of magnitude larger than the selectivity values, this model favors the polymers that 207 have higher permeability, thus biasing the GA towards better performing polymers. 208

We used the BRICS algorithm on the GA-generated polymers to understand which chemical building blocks are frequently observed in the top performing polymers. The top six frequently



Figure 6: Robeson plot of selectivity versus permeability for (a)  $CO_2/N_2$  and (b)  $CO_2/O_2$  separations. The 2008 Robeson upper bounds are shown as dashed black lines (6). Colors represent experimental and GA generated polymers. Red color represents polymers generated with the GA. Each data point represents a single polymer.

observed functional groups with the 100 fittest GA-generated polymers are shown in Figure 211 7. We identified a total of 464 chemical fragments within the fittest 100 polymers, and 18% 212 of the fragments were pyridine functional groups. More than 70 polymers in the top perform-213 ing GA-generated polymers have the pyridine functional group in their repeating unit. This 214 is by far the most frequently observed functional group, which is then followed by benzoxa-215 zole with 3%. We observe benzoxazole functional group in 13 polymers within the 100 fittest 216 GA-generated polymers. Similarly, benzene, phosphonamidic acid, naphthalene, and dibro-217 mobenzene functional groups are also observed in the top performing polymers generated with 218 the GA. We show six example polymer structures that have high fitness function values in Fig-219 ure 8. We note that these polymers have pyridine, benzoxazole, benzene, and phosphonamidic 220 acid functional groups, which we identified as the most abundant functional groups with the 221 BRICS algorithm. These polymers also include oxygen-, sulfur-, and nitrogen- containing 222 motifs, similar to the three experimental polymers that are above the  $CO_2/N_2$  upper bound. 223 Oxygen- and nitrogen- containing motifs are reminiscent of imines and polyethers, which are 224

known to be high performing polymer membranes. Interestingly, our initial analysis using the 225 Polymer Genome software (51–54) suggests that most of the top 100 fittest polymers have high 226 glass transition temperature—well above the standard operating conditions (> 400 K). We note 227 that the three polymers in the experimental data set that are above the  $CO_2/N_2$  upper bound also 228 have glass transition temperatures around 400 K. We speculate that the superior performance 229 of these glassy amorphous polymers for gas separation may be due to their high fractional free 230 volume and high number of microvoids (55, 56). It remains an open question whether or not 231 these polymers are easily synthesizable and easy to implement as membranes given their com-232 plicated chemistry. Further computational and experimental studies will be required to better 233 understand these polymers and their efficacy as membrane materials. 234



Figure 7: The six functional groups that appear most in the 100 fittest polymers generated with the GA using the BRICS algorithm. "A" represents the binding sites.

#### 235 **Discussion**

We constructed an ML-driven GA to tackle the inverse design problem of polymer membranes for  $CO_2$  separation. We showed that the hashed-based ECFP4 yields lower predictive errors on the test sets than the substructure keys-based fingerprints. We presented different regressionbased ML models, where random forest regression models resulted in the lowest RMSE and highest  $R^2$  values for both the test and the training set. Although random forest regression



Figure 8: Example of polymer repeating units generated with the GA. "A" represents the binding sites.

models can successfully predict both the gas permeability and the selectivity, we used mod-241 els trained on the gas permeability, since these models have better predictive power than the 242 models trained on the selectivity. After obtaining the ML models to predict the performance 243 of any polymer membrane candidate, we implemented a data-driven inverse design algorithm 244 to efficiently explore the polymer material space. In theory, one can use any inverse design 245 algorithm for such a problem, but we have implemented a GA since it has been successfully 246 used for other polymer applications. We created the gene pool using the BRICS algorithm on 247 the experimental data and obtained 79 unique genes to initiate the GA process with 100 parent 248 polymers that have 4 genes in their monomer unit. Fitness function is a key driving parameter 249 in the GA, and as such, we used  $P_{CO_2} \times \alpha_{CO_2/N_2} \times \alpha_{CO_2/O_2}$  to determine the fitness of the 250 polymers. We performed crossover and mutation functions for 100 generations to create more 251 than 16000 polymers during the GA process. Among these 16000 polymers, we were able to 252 identify more than 20 new polymers that are above both CO<sub>2</sub>/N<sub>2</sub> and CO<sub>2</sub>/O<sub>2</sub> upper bounds. 253 While validation of the new polymer structures identified in this work requires future molecular 254 dynamics simulations and experimental measurements, this work helps identify the strengths 255

<sup>256</sup> and the weaknesses of combining ML models and GAs as discussed below.

Three key points emerged from our analysis on comparing popular fingerprints. First, 257 hashed-based fingerprints result in lower predictive errors on the test sets than the substruc-258 ture keys-based fingerprints. However, hashed-based fingerprints have one main disadvantage 259 compared to the substructure keys-based fingerprints, which is not having a one-to-one corre-260 spondence between the fingerprint vector and the chemical structure. This is not the case with 261 the substructure keys-based fingerprints, where each bit corresponds to a predetermined sub-262 structure. This disadvantage does not affect our framework since we do not go back-and-forth 263 between the fingerprint and the chemical structure, and only use the fingerprinting when evalu-264 ating the fitness function in the GA framework. Not performing crossover and mutation func-265 tions on the fingerprints makes it possible to overcome this main disadvantage associated with 266 hashed-based fingerprints. Next, we note that the top performing polymer candidates identified 267 within this framework does not depend on which fingerprint is used in the GA. The relative 268 strength of the polymers with each other are similar with the two different fingerprints. The 269 main difference with using different fingerprints is the absolute value of the fitness function, 270 since the ML models trained on the substructure keys-based fingerprints tend to underestimate 271 the gas permeability. Finally, using the proper descriptor for a given application is still an open 272 question in the polymer informatics field, but we have shown that most often it does not affect 273 the final result. However, we emphasize that more sophisticated descriptors, like physical de-274 scriptor vectors that include bulk properties of the polymer membranes, may make a difference 275 in the top performing polymers identified from the GA (57). For example, we believe the glass 276 transition temperature is a key property for polymer membranes, and including this informa-277 tion in the fingerprint can lead to better performing polymer membranes within our framework. 278 The main bottleneck in switching from popular fingerprints to physical descriptors is gathering 279 consistent measurement data from hundreds of different experimental papers. The only way to 280

overcome this bottleneck is to create our own data sets using computational simulations so that
we can consistently calculate the physical properties of interest for each polymer.

We have demonstrated that a random forest regression model performs best when predicting 283 the gas permeability and selectivity of polymer membranes. Because there is a significant dif-284 ference in the R<sup>2</sup> values and the RMSEs, we use random forest regression models in the entire 285 framework. However, random forest algorithms, like essentially all data-driven methods, are 286 intrinsically interpolative (10). They are only suited to optimize properties within the bounds 287 of the data the model was trained on. Models can still generalize, and interpolate "between 288 molecules" in some abstract design space (58), but they will not make accurate predictions 289 outside of this space. Thus, the performance of the new polymer structures identified in this 290 framework depends on the initial data set and the range of selectivity and permeability values 291 covered. One way to address this challenge is by using computer simulations combined with 292 e.g. an active learning loop to curate a polymer library will enable us to expand the number of 293 data points included in the ML framework. It is important to note that regardless of the data-294 driven approach, ML models will always have a strong dependency on the initial data that they 295 are trained on. The only way to surpass this main limitation is to move towards active learn-296 ing algorithms where we give the algorithm the ability to "learn" and draw inferences from its 297 experience to accelerate the evolutionary process (59, 60). New molecules generated as part of 298 the GA procedure could be screened by an uncertainty-quantifying algorithm, and when confi-299 dence is low, new simulations can be run to acquire new ground truth data, which can then be 300 used to retrain the model. This is only possible if we use computer simulations (or a very high-301 throughput, autonomous experiment) to curate the data since we will need to make on-the-fly 302 property estimations with an active learning framework. 303

<sup>304</sup> Finally, we identify two significant points for coupling GAs with ML models to design new <sup>305</sup> polymer membranes. First, and most importantly, the fitness function drives the evolutionary

process of the GA. It is therefore of utmost importance to select the correct fitness function 306 to direct the GA towards the targeted design area. It is customary to train ML models on the 307 logarithmic permeability values since it narrows the range of the data, hence resulting in more 308 accurate models. Thus we tried the fitness function  $\log (P_{CO_2}) \times \alpha_{CO_2/N_2} \times \alpha_{CO_2/O_2}$ , aiming to 309 maximize both the gas permeability and selectivity throughout the evolutionary process. How-310 ever, this fitness function was not able to push the GA towards the targeted design area. Even 311 though ML models trained on logarithmic permeability have slightly higher predictive power, 312 the small numerical value of the logarithmic permeability diminishes the importance of the per-313 meability contribution to the fitness function. On the other hand, with a fitness function that 314 includes the absolute value of the permeability  $(P_{CO_2} \times \alpha_{CO_2/N_2} \times \alpha_{CO_2/O_2})$ , we were able to 315 push the evolutionary process toward the targeted design area and identified more than 20 new 316 polymers that are above both CO<sub>2</sub>/N<sub>2</sub> and CO<sub>2</sub>/O<sub>2</sub> upper bounds. We attribute the superior per-317 formance of this fitness function to the fact that the absolute value of the permeability is usually 318 two orders of magnitude higher than selectivity values. This analysis also shows improving 319 the selectivity with the GA is much harder than the permeability since the fitness function be-320 comes insensitive to the selectivity values when we include the gas permeability. In the future 321 this can be avoided by normalizing the parameters where the normalization would negate this 322 effect. Next we emphasize that, our GA was able to converge within 100 generations, since 323 running the algorithm for an additional 100 generations did not result in any superior polymer 324 membranes. With 100 generations and 4 initial building blocks in the first generation, a total 325 of 17571 new unique polymer structures were created. With 79 unique genes, the number of 326 sequences that can be generated by the GA is at least 79<sup>4</sup> (since longer sequences are gener-327 ated throughout the evolutionary process). This suggests that the GA converges very fast, only 328 exploring less than 1% of the possible polymer material space. We decided to use 4 genes 329 with the initial generation because we have a relatively small gene pool. Using more building 330

<sup>331</sup> blocks with the initial generation could have created more complicated structures throughout <sup>332</sup> the evolutionary process. This can be further explored when we have a larger gene pool.

Our approach demonstrates successful implementation of an ML-driven GA to design poly-333 mer membranes for CO<sub>2</sub> separation, but more importantly, this framework can be used to design 334 polymer structure for any application (e.g. ion separation membranes and polymer electrolytes 335 for batteries), where there is a constrained optimization problem. The main limitation of the 336 current framework arises from its dependence on the initial experimental data. Curating the 337 data with computer simulations is a possible way to overcome this limitation. With better con-338 trol over the initial data set we will be in a position to explore more sophisticated descriptors 339 and switch to an active learning framework where we make on-the-fly property estimations. 340 Computational ML-driven inverse design of polymer membranes is a promising platform that 341 can be further tailored to consider functions that incorporate the sustainability and synthetic vi-342 ability of the polymers, in addition to gas selectivity and permeability, which are not yet widely 343 considered in computational studies. 344

#### 345 Acknowledgments

This work was supported in part by Resnick Sustainability Institute (YB, ZGW). Additional support was provided by Hong Kong Quantum AI Lab, AIR@InnoHK of Hong Kong Government (ZGW). DP acknowledges support by the Caltech Amgen Scholars Program. This material is also based upon work supported by the U.S. Department of Energy, Office of Science, Office of Basic Energy Sciences, under Award Numbers FWP PS-030 and DE-SC-0012704 (MRC, SKK). At Columbia financial support for this work was provided by the U.S. Department of Energy under Grants DE-SC0021272 (TS).

#### **Supplementary Material**

We provide the initial polymer library we have used to train our ML models with each polymer represented as a SMILES string as well as the GA generated polymers with permeability for three gasses ( $CO_2$ ,  $N_2$ , and  $O_2$ ).

## **357 References**

- 1. D. F. Sanders, et al., Polymer 54, 4729 (2013).
- 2. R. W. Baker, K. Lokhandwala, Ind. Eng. Chem. Res. 47, 2109 (2008).
- 360 3. R. W. Baker, B. T. Low, *Macromolecules* 47, 6999 (2014).
- <sup>361</sup> 4. D. S. Sholl, R. P. Lively, *Nature* **532**, 435 (2016).
- <sup>362</sup> 5. L. M. Robeson, J. Membr. Sci. **62**, 165 (1991).
- 363 6. L. M. Robeson, J. Membr. Sci. 320, 390 (2008).
- <sup>364</sup> 7. H. B. Park, *et al.*, *Science* **318**, 254 (2007).

- <sup>365</sup> 8. M. D. Guiver, Y. M. Lee, *Science* **339**, 284 (2013).
- <sup>366</sup> 9. N. Du, *et al.*, *Nat. Mater.* **10**, 372 (2011).
- <sup>367</sup> 10. M. R. Carbone, *MRS Bulletin* **47**, 968–974 (2022).
- <sup>368</sup> 11. V. M. Alves, *et al.*, *Sci. Adv.* **5**, eaav9784 (2019).
- <sup>369</sup> 12. L. Tao, G. Chen, Y. Li, *Patterns* 2, 100225 (2021).
- <sup>370</sup> 13. L. Tao, V. Varshney, Y. Li, J. Chem. Inf. Model 61, 5395 (2021).
- <sup>371</sup> 14. Q. Xu, J. Jiang, ACS Appl. Polym. Mater. 2, 3576 (2020).
- <sup>372</sup> 15. M. Wang, Q. Xu, H. Tang, J. Jiang, ACS Appl. Mater. Interfaces 14, 8427 (2022).
- <sup>373</sup> 16. M. Wang, J. Jiang, ACS Appl. Mater. Interfaces 14, 31203 (2022).
- 17. M. Rahimi, S. M. Moosavi, B. Smit, T. A. Hatton, Cell Rep. Phys. Sci. 2, 100396 (2021).
- <sup>375</sup> 18. Y. Liu, O. C. Esan, Z. Pan, L. An, *Energy and AI* **3**, 100049 (2021).
- <sup>376</sup> 19. T. K. Patra, ACS Polym. Au 2, 8 (2021).
- 20. A. Tayyebi, A. S. Alshami, X. Yu, E. Kolodka, J. Membr. Sci. Letters p. 100033 (2022).
- 378 21. K. Sattari, Y. Xie, J. Lin, Soft Matter (2021).
- 22. R. S. K. Valappil, N. Ghasem, M. Al-Marzouqi, J. Ind. Eng. Chem. 98, 103 (2021).
- <sup>380</sup> 23. S. Gupta, L. Li, *JOM* pp. 1–15 (2022).
- <sup>381</sup> 24. L. Chen, *et al.*, *Mater. Sci. Eng. R Rep.* **144**, 100595 (2021).
- 382 25. Y. Amamoto, *Polym. J.* pp. 1–11 (2022).

- 383 26. Q. Xu, J. Jiang, Mol. Syst. Des. Eng. (2022).
- <sup>384</sup> 27. J. W. Barnett, *et al.*, *Sci. Adv.* **6**, eaaz4301 (2020).
- 28. Q. Yuan, et al., J. Membr. Sci. 627, 119207 (2021).
- <sup>386</sup> 29. J. Yang, L. Tao, J. He, J. R. McCutcheon, Y. Li, *Sci. Adv.* 8, eabn9545 (2022).
- 387 30. B. Srinivasan, et al., Proc. Natl. Acad. Sci. 110, 18431 (2013).
- 388 31. C. Kim, R. Batra, L. Chen, H. Tran, R. Ramprasad, *Comput. Mater. Sci.* 186, 110067
   (2021).
- 390 32. S. La Cognata, et al., Eur. J. Chem. 28, e202201631 (2022).
- <sup>391</sup> 33. Y. Zu, et al., *Microporous and Mesoporous Mater.* **334**, 111779 (2022).
- <sup>392</sup> 34. H. Lin, B. D. Freeman, *Macromolecules* **39**, 3568 (2006).
- 393 35. J. Liu, X. Hou, H. B. Park, H. Lin, Eur. J. Chem. 122, 15980 (2016).
- 394 36. T. Tran, Y. Fu, D.-e. Jiang, H. Lin, *Macromolecules* 55, 9860 (2022).
- 395 37. G. Polotskaya, S. Agranova, T. Antonova, G. Elyashevich, J. Appl. Polym. Sci. 66, 1439
   (1997).
- 397 38. Y. Li, M. Ding, J. Xu, J. Appl. Polym. Sci. 63, 1821 (1997).
- <sup>398</sup> 39. D. Weininger, J. Chem. Inf. Model. 28, 31 (1988).
- <sup>399</sup> 40. M. Krenn, et al., Patterns **3**, 100588 (2022).
- 400 41. D. Rogers, R. D. Brown, M. Hahn, J. Biomol. Screen. 10, 682 (2005).
- 401 42. D. Rogers, M. Hahn, J. Chem. Inf. Model. 50, 742 (2010).

- 402 43. M. S. Keys, CA, USA (2011).
- 403 44. J. L. Durant, B. A. Leland, D. R. Henry, J. G. Nourse, J. Chem. Inf. Model. 42, 1273 (2002).
- 404 45. A. Cereto-Massagué, et al., Methods 71, 58 (2015).
- 405 46. M. Sastry, J. F. Lowrie, S. L. Dixon, W. Sherman, J. Chem. Inf. Model 50, 771 (2010).
- 406 47. H. L. Morgan, J. Chem. Doc. 5, 107 (1965).
- 407 48. X. Yu, Fibers Polym. 11, 757 (2010).
- 408 49. J. Degen, C. Wegscheid-Gerlach, A. Zaliani, M. Rarey, *ChemMedChem* 3, 1503 (2008).
- 409 50. P. Baldi, R. Nasr, Journal of chemical information and modeling 50, 1205 (2010).
- 410 51. C. Kim, A. Chandrasekaran, T. D. Huan, D. Das, R. Ramprasad, *J. Phys. Chem. C* . 122,
  411 17575 (2018).
- 412 52. C. Kuenneth, et al., Patterns 2, 100238 (2021).
- 413 53. H. Doan Tran, et al., J. Appl. Phys. 128, 171104 (2020).
- 414 54. A. Chandrasekaran, C. Kim, S. Venkatram, R. Ramprasad, *Macromolecules* 53, 4764
  415 (2020).
- 416 55. R. Mahajan, R. Burns, M. Schaeffer, W. J. Koros, J. Appl. Polym. Sci. 86, 881 (2002).
- 417 56. R. Recio, et al., J. Appl. Polym. Sci. 107, 1039 (2008).
- 418 57. R. A. Patel, C. H. Borca, M. A. Webb, Mol. Syst. Des. Eng. 7, 661 (2022).
- 419 58. R. Gómez-Bombarelli, et al., ACS central science 4, 268 (2018).

- <sup>420</sup> 59. T. D. Loeffler, S. Banik, T. K. Patra, M. Sternberg, S. K. Sankaranarayanan, *J. Phys. Com- mun.* 5, 031001 (2021).
- 422 60. T. K. Patra, V. Meenakshisundaram, J.-H. Hung, D. S. Simmons, *ACS Comb. Sci.* 19, 96
   423 (2017).