

Supporting Information

Organic Base-Mediated Wittig Reaction of Perfluorohalogenated Benzaldehydes for Designing Halogen Bond-Driven Smart Polymer Materials: Toward Digitalization as Reliable Strategy in Organic Synthesis

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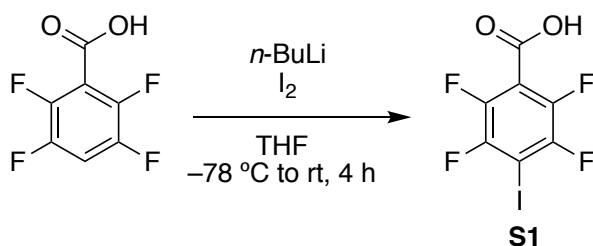
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1. General information

Unless otherwise noted, all reactions were carried out under an atmosphere of standard grade nitrogen gas (oxygen <10 ppm) in flame-dried glassware with magnetic stirring. Anhydrous THF, CH₂Cl₂, toluene and diethyl ether (Et₂O) were supplied from Kanto Chemical Co., Inc. as “Dehydrated solvent system”. Other reagents were purchased from commercial suppliers and used without further purification. Purification of reaction products was carried out by column chromatography on silica gel 60 (spherical, neutral, 100-210 μm; KANTO and Merck). Analytical thin layer chromatography (TLC) was performed on E. Merck precoated (0.25 mm) silica gel 60-F254 plates. Visualization was accomplished with UV light and phosphomolybdic acid solution in ethanol by heating. ¹H NMR spectra were recorded on a JEOL ECA-400 (400 MHz) spectrometer at ambient temperature. NMR solvent was purchased from CIL (CDCl₃). Data are reported as follows: chemical shifts are reported in ppm from tetramethylsilane on the δ scale, with solvent resonance employed as internal standard (CDCl₃ 7.26 ppm), multiplicity (br = broad, s = singlet, d = doublet, dd = doublet of doublet, t = triplet, q = quartet, and m = multiplet), integration, coupling constant (Hz) and assignment. ¹³C NMR spectra were recorded on a JEOL ECA-400 (100 MHz) spectrometer at ambient temperature. Chemical shifts are reported in ppm from tetramethylsilane on the δ scale, with solvent resonance employed as internal standard (CDCl₃ 77.0 ppm). ¹⁹F NMR spectra were recorded on a JEOL ECS-400 (376 MHz) spectrometer. Chemical shifts are reported in ppm from the α,α,α-trifluorotoluene (−63.72 ppm) resonance as external standard. Infrared (IR) spectra were recorded on a Jasco FT/IR-460 plus using ATR. High-resolution mass spectra (HRMS) analysis was performed on a JEOL JMS-700 (double-focusing magnetic sector mass analyzer: EB) with the fast atom bombardment (FAB) using 3-nitrobenzyl alcohol as the matrix at the Instrument Center, Institute for Molecular Science.

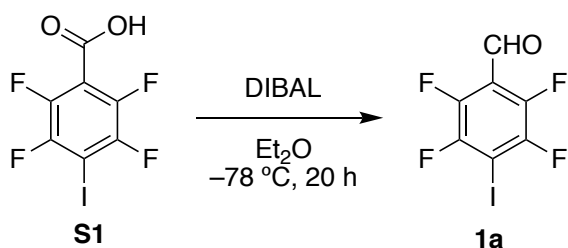
2. Synthesis of 2,3,5,6-tetrafluoro-4-halobenzaldehydes

Synthesis of 2,3,5,6-tetrafluoro-4-iodobenzoic acid (**S1**)



The **S1** was synthesized according to literature procedure.¹ To a solution of 2,3,5,6-tetrafluorobenzoic acid (2.91 g, 15.0 mmol, 1.0 equiv.) in THF (300 mL) was added *n*-BuLi (1.6 M in hexane, 18.3 mL, 29.3 mmol, 2.0 equiv.) dropwise over 15 min at -78 °C under nitrogen atmosphere, then the mixture was stirred at -78 °C for 1 h. A solution of iodine (3.70 g, 14.3 mmol, 0.95 equiv.) in THF (20 mL) was then slowly added to the reaction mixture. The mixture was allowed to slowly warm up to room temperature and stirred for 4 h. 2M HCl aq. (19 mL) was added to the mixture and the organic layer was separated. The aqueous layer was extracted with diethyl ether (3 x 50 mL). The combined organic layers were washed with sat. Na₂S₂O₃ aq. (50 mL) and brine (50 mL), dried over MgSO₄, and concentrated under reduced pressure after filtration to afford **S1** as white solid (4.56 g, 14.3 mmol, 95% yield). The **S1** was used in the next reaction without further purification.

Synthesis of 2,3,5,6-tetrafluoro-4-iodobenzaldehyde (**1a**)

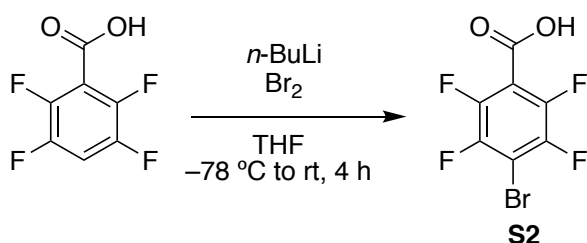


The **1a** was synthesized according to literature procedure.² To a solution of 2,3,5,6-tetrafluoro-4-iodobenzoic acid **S1** (4.64 g, 14.5 mmol, 1.0 equiv.) in diethyl ether (180 mL) was added DIBAL (1.0 M in *n*-hexane, 28.4 mL, 29.0 mmol, 2.0 equiv.) dropwise over 15 min at -78 °C under nitrogen atmosphere, then the mixture was stirred at -78 °C for 14 h. Methanol (22.7 mL) was added to the mixture, and the mixture was allowed to warm to room temperature. Then 30% Rochelle's salt aq. (28.4 mL), celite (30 g), and sea sand (50 g) were added to the mixture, and the mixture was further stirred at room temperature for 1.5 h. The resulting suspension was filtered by celite and washed with

diethyl ether (50 mL). A filtrate was extracted with diethyl ether (3 x 50 mL). The combined organic layers were washed with sat. Rochelle's salt aq. (20 mL, 15 mL) and brine (30 mL), dried over Na₂SO₄, and concentrated under reduced pressure after filtration. The obtained crude product was purified by silica gel chromatography (hexane/ethyl acetate = 30:1 to 8:1) to afford **1a** (1.36 g, 4.50 mmol, 31%) as white solid. R_f = 0.42 (hexane/ethyl acetate = 8:1)

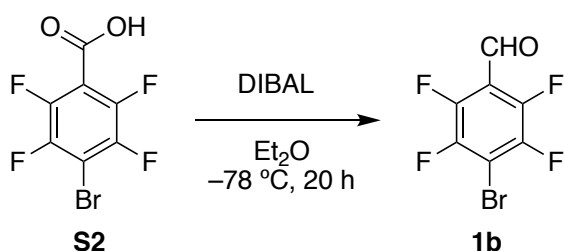
¹H NMR (CDCl₃, 400 MHz) δ 10.32 (s, 1H). ¹⁹F NMR (CDCl₃, 376 MHz) δ -118.18- -118.24 (m, 2F), -143.56- -143.65 (m, 2F). elemental analysis calcd (%) for C₇HF₄IO: C 27.66, H 0.33, F 25.00, I 41.75, found: C 27.67, H 0.56, F 24.77, I 41.70.

Synthesis of 4-bromo-2,3,5,6-tetrafluorobenzoic acid (**S2**)



The **S2** was synthesized according to **Synthesis of 2,3,5,6-tetrafluoro-4-iodobenzoic acid (S1)**. To a solution of 2,3,5,6-tetrafluorobenzoic acid (584 mg, 3.01 mmol, 1.0 equiv) in THF (64 mL) was added *n*-BuLi (1.59 M in hexane, 3.80 mL, 6.04 mmol, 2.0 equiv) dropwise over 15 min at -78 °C under nitrogen atmosphere, then the mixture was stirred at -78 °C for 1 h. Bromine (310 μL, 6.10 mmol, 2.0 equiv.) was then added to the reaction mixture. The mixture was allowed to slowly warm up to room temperature and stirred for 4 h. 2 M HCl aq. (4.0 mL) was added to the mixture and the organic layer was separated. The aqueous layer was extracted with diethyl ether (3 x 30 mL). The combined organic layers were washed with sat. Na₂S₂O₃ aq. (50 mL) and brine (50 mL) and concentrated under reduced pressure after filtration. The residual product was dissolved in CH₂Cl₂ (40 mL) then the solution was back-extracted with 5% NaOH aq. (4 x 25 mL). The aqueous layer was acidified with 2 M HCl aq. (100 mL), then extracted with CH₂Cl₂ (4 x 25 mL). The combined organic layers were washed with H₂O (15 mL) and brine (20 mL), dried over MgSO₄, and concentrated under reduced pressure after filtration to afford **S2** as white solid (728 mg, 2.67 mmol, 89% yield). The **S2** was used in the next reaction without further purification.

Synthesis of 4-bromo-2,3,5,6-tetrafluorobenzaldehyde (**1b**)³

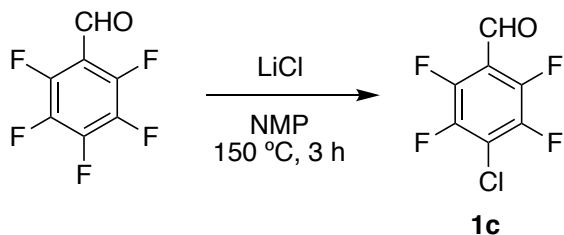


The **1b** was synthesized according to **Synthesis of 2,3,5,6-tetrafluoro-4-iodobenzaldehyde (1a)** using 4-bromo-2,3,5,6-tetrafluorobenzoic acid (**S2**) (728 mg, 2.67 mmol, 1.0 equiv.) to afford **1b** as white solid (286 mg, 1.11 mmol, 37%).

R_f = 0.42 (hexane/ethyl acetate = 8:1)

¹H NMR (400 MHz, CDCl₃) δ 10.30 (s, 1H). ¹⁹F NMR (376 MHz, CDCl₃) δ -131.3- -131.4 (m, 2F), -141.0- -141.1 (m, 2F).

Synthesis of 4-chloro-2,3,5,6-tetrafluorobenzaldehyde (**1c**)



The **1c** was synthesized according to literature procedure.⁴

A 100 mL three neck flask equipped with a magnetic stirring bar and a septum was charged with LiCl (466 mg, 11.0 mmol, 1.1 equiv.). The flask was flame-dried under vacuo then backfilled with nitrogen gas after cooling to room temperature. *N*-methylpyrrolidone (15 mL) and pentafluorobenzaldehyde (1.25 mL, 10.0 mmol, 1.0 equiv.) were added to the flask. The resulting suspension was warmed up to 150 °C and stirred for 3 h. After cooling to room temperature, the mixture was poured into ice-water (50 mL) then extracted with CH₂Cl₂ (3 x 15 mL). The combined organic layers were washed with brine (15 mL), dried over Na₂SO₄, and concentrated under reduced pressure after filtration. The residual crude product was purified by silica gel column chromatography (hexane/ethyl acetate = 10:1) to afford a **1c** as pale yellow solid (1.52 g, 7.10 mmol, 71% yield, 89% purity). The **1c** was further purified by washing with little amount of hexane to afford 99% purity of **1c** as white solid (891 mg, 4.20 mmol, 42% yield).

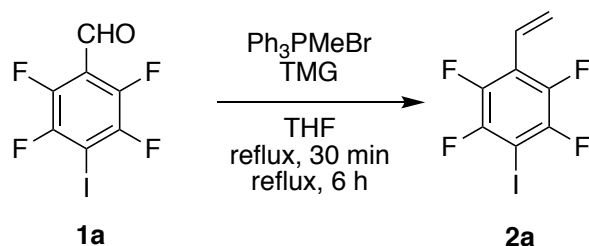
R_f = 0.20 (hexane/ethyl acetate = 10:1)

¹H NMR (400 MHz, CDCl₃) δ 10.30 (s, 1H). ¹⁹F NMR (CDCl₃, 376 MHz) δ -138.8- -138.9 (m, 2F), -144.1- -144.2 (m, 2F).

3. Synthesis and Characterization of 2,3,5,6-tetrafluoro-4-halostyrenes

General procedure for Wittig reaction:

Synthesis of 2,3,5,6-tetrafluoro-4-iodostyrene (**2a**)

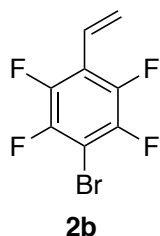


To a solution of methyltriphenylphosphonium bromide (0.643 g, 1.80 mmol, 3.6 equiv.) in THF (5 mL) was added 1,1,3,3-tetramethylguanidine (282 μL , 2.25 mmol, 4.5 equiv.) and the reaction mixture was stirred at 80 °C for 30 min. Then a solution of 2,3,5,6-tetrafluoro-4-iodobenzaldehyde **1a** (0.152 g, 0.50 mmol, 1.0 equiv.) in THF (1.5 mL) was added to the mixture. The mixture was stirred for 6 h. H_2O (10 mL) was added to the mixture and the organic layer was separated. The aqueous layer was extracted with dichloromethane (3 x 10 mL). The combined organic layers were washed with brine (10 mL), dried over MgSO_4 . Small aliquots from the organic layers were analyzed by ^{19}F NMR using α,α,α -trifluorotoluene as standard (53% NMR yield). The organic layers were concentrated under reduced pressure to afford **2a** as colorless oil. *Due to stability of product, the **1c** was characterized as mixture and full data could not be collected.*

$R_f = 0.53$ (hexane)

^1H NMR (CDCl_3 , 400 MHz) δ 6.69 (dd, $J = 11.9, 6.2$ Hz, 1H), 6.15 (d, $J = 18.1$ Hz, 1H), 5.76 (d, $J = 11.9$ Hz, 1H). ^{19}F NMR (CDCl_3 , 376 MHz) δ -121.65—121.74 (m, 2F), -141.56—141.66 (m, 2F). ^{13}C NMR $\{^{19}\text{F}\}$ (CDCl_3 , 100 MHz) δ 147.24 (dd, $J = 229, 15$ Hz), 144.11 (dd, $J = 239, 15$ Hz), 124.40 (t, $J = 160$ Hz), 122.41 (d, $J = 166$ Hz), 117.51 (t, $J = 13$ Hz), 70.40. IR (ATR) 1471, 1416, 1259, 1101, 983, 953, 794.

4-bromo-2,3,5,6-tetrafluorostyrene (**2b**)



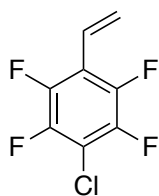
The **2b** was synthesized according to General Procedure using **1b** (128 mg, 0.500 mmol, 1.0 equiv.) to afford **2b** in 39% NMR yield. *Due to stability of product, the **2b** was*

characterized as mixture and full data could not be collected.

Rf = 0.58 (hexane)

^1H NMR (CDCl_3 , 400 MHz) δ 6.67 (dd, $J = 11.9, 18.1$ Hz, 1H), 6.14 (d, $J = 17.9$ Hz, 1H), 5.76 (d, $J = 11.9$ Hz, 1H). ^{19}F NMR (CDCl_3 , 376 MHz) δ -134.43- -134.50 (m, 2F), -142.10- -142.18 (m, 2F). ^{13}C NMR $\{^{19}\text{F}\}$ (CDCl_3 , 100 MHz) δ 145.02, 144.81, 124.42 (t, $J = 161$ Hz), 122.16 (d, $J = 160$ Hz), 116.36 (t, $J = 13.4$ Hz), 98.31. IR (ATR) 1481, 1457, 1421, 1398, 1267, 1122, 1072, 988, 957, 820, 741, 704.

4-chloro-2,3,5,6-tetrafluorostyrene (2c)



2c

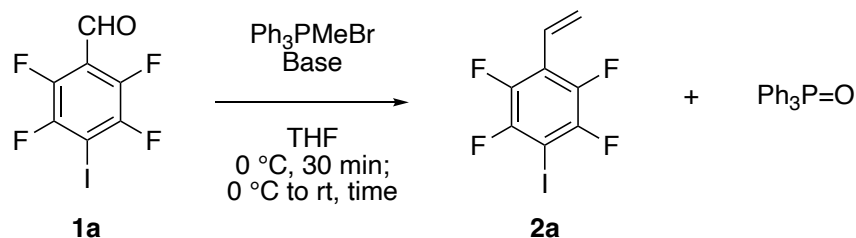
The **2c** was synthesized according to General Procedure using **1c** (106 mg, 0.500 mmol, 1.0 equiv.) to afford **2c** in 29% NMR yield. *Due to stability of product, the 2c was characterized as mixture and full data could not be collected.*

Rf = 0.60 (hexane)

^1H NMR (CDCl_3 , 400 MHz) δ 6.67 (dd, $J = 11.5, 18.6$ Hz, 1H), 6.12 (d, $J = 18.6$ Hz, 1H), 5.76 (d, $J = 12.6$ Hz, 1H). ^{19}F NMR (CDCl_3 , 376 MHz) δ -142.11- -142.19 (m, 2F), -142.65- -142.71 (m, 2F). ^{13}C NMR $\{^{19}\text{F}\}$ (CDCl_3 , 100 MHz) δ 144.84, 144.18, 124.36 (t, $J = 161$ Hz), 121.94 (d, $J = 162$ Hz), 115.70 (t, $J = 13.4$ Hz), 110.74. IR (ATR) 1485, 1465, 1425, 1402, 1279, 1108, 1000, 989, 961, 939, 901, 859.

4. Initial study of Wittig reaction

Table S1. Wittig reaction of **1a** using conventional bases



Entry	Ph_3PMeBr Equiv.	Base Equiv.	time (h)	Yield (%) ^b	
				2a	$\text{Ph}_3\text{P=O}$
1	1.2	<i>n</i> -BuLi 1.2	2	31	64
2	1.2	<i>n</i> -BuLi 1.2	24	34	71
3	1.2	<i>t</i> -BuOK 1.2	2	2	28
4	1.2	<i>t</i> -BuOK 1.2	24	7	51
5	1.2	NaH 1.2	2	<1	2
6	1.2	NaH 1.2	24	2	4
7	1.2	K_2CO_3 1.2	2	<1	6
8	1.2	K_2CO_3 1.2	24	<1	10

a) The reaction was carried out using 0.50 mmol of **1a**, Ph_3PMeBr (1.2 equiv.) and indicated base in THF (6.5 mL) for indicated time. b) determined by ^{19}F and ^{31}P NMR using α,α,α -trifluorotoluene and triphenylphosphine as internal standards.

5. DFT Calculations

5-1. Computational analysis by DFT

All molecular geometries were optimized by the M06-2X functional with Grimme's D3 dispersion correction⁵ using the 6-311+G(d,p) basis set. The SMD solvation model⁶ was used with the solvents indicated. The stationary geometries were checked by the vibration analyses after the geometry optimization procedures. The stationary geometries and their energies were refined using Gaussian 16 software package.⁷

5-2. Computational details

The table shows total energy E , enthalpy H , and Gibbs free energy G (hartree) at the SMD/M06-2X-D3/6-311+G(d,p) level.

Table S2. Organic bases

compound	solvent	temp. (°C)	E	H	G
MTBD	THF	60	-478.0685409	-477.820122	-477.874209
DBN	THF	60	-383.4162621	-383.216875	-383.264765
DBU	THF	40	-462.0245018	-461.765565	-461.814401
DBU	THF	60	-462.0245018	-461.764137	-461.817565
DBU	THF	80	-462.0245018	-461.762611	-461.820816
DBU	CH ₂ Cl ₂	55	-462.0271203	-461.767163	-461.819432
DBU	MeCN	95	-462.0265648	-461.763596	-461.825434
DBU	toluene	125	-462.0220921	-461.756142	-461.825920
DBU	Et ₂ O	45	-462.0242003	-461.764818	-461.814828
DBU	1,4-dioxane	115	-462.0179106	-461.752743	-461.819820
TMG	THF	60	-362.5440492	-362.341413	-362.394088
TMG	THF	80	-362.5440492	-362.340084	-362.397289
piperidine	THF	60	-251.8589684	-251.691224	-251.731886
quinuclidine	THF	60	-329.2536598	-329.048446	-329.092528
Et ₃ N	THF	60	-292.3478434	-292.128626	-292.180267

Table S3. (Organic base)·HBr salts

compound	solvent	temp. (°C)	<i>E</i>	<i>H</i>	<i>G</i>
MTBD·HBr	THF	60	-3052.9024750	-3052.635936	-3052.698567
DBN·HBr	THF	60	-2958.2499495	-2958.033479	-2958.091369
DBU·HBr	THF	40	-3036.8606171	-3036.584522	-3036.642610
DBU·HBr	THF	60	-3036.8606171	-3036.582886	-3036.646370
DBU·HBr	THF	80	-3036.8606171	-3036.581151	-3036.650232
DBU·HBr	CH ₂ Cl ₂	55	-3036.8639084	-3036.586625	-3036.648763
DBU·HBr	MeCN	95	-3036.8654663	-3036.584698	-3036.658308
DBU·HBr	toluene	125	-3036.8508323	-3036.567264	-3036.650314
DBU·HBr	Et ₂ O	45	-3036.8576031	-3036.581161	-3036.640606
DBU·HBr	1,4-Dioxane	115	-3036.8452591	-3036.562597	-3036.641936
TMG·HBr	THF	60	-2937.3741575	-2937.154729	-2937.217686
TMG·HBr	THF	80	-2937.3741575	-2937.153162	-2937.221512
piperidine·HBr	THF	60	-2826.6854590	-2826.500219	-2826.550472
quinuclidine·HBr	THF	60	-2904.0849020	-2903.862007	-2903.915777
Et ₃ N·HBr	THF	60	-2867.1788689	-2866.941629	-2867.001054

Table S4. HBr

compound	solvent	temp. (°C)	<i>E</i>	<i>H</i>	<i>G</i>
HBr	THF	40	-2574.7712306	-2574.761714	-2574.785546
HBr	THF	60	-2574.7712306	-2574.761493	-2574.787075
HBr	THF	80	-2574.7712306	-2574.761271	-2574.788617
HBr	CH ₂ Cl ₂	55	-2574.7716789	-2574.761997	-2574.787141
HBr	MeCN	95	-2574.7712393	-2574.761118	-2574.789796
HBr	toluene	125	-2574.7712645	-2574.760778	-2574.792136
HBr	Et ₂ O	45	-2574.7714449	-2574.761869	-2574.786137
HBr	1,4-Dioxane	115	-2574.7703863	-2574.760006	-2574.790467

6. Correlation analyses

6-1. Preparation of data sets

Data sets were prepared as csv. file using bellow descriptors.

Table S5. Descriptors for correlation analyses

Category	Descriptor	Physical meaning	Reference
Reaction	yield (%) of TFIS	Yield of 2,3,5,6-tetrafluoro-4-iodostyrene 2a	
	yield (%) of Ph ₃ P=O	Yield of triphenylphosphine oxide	
	UDR-Pro	Undesired Reaction Product: 0: Not observed. -1:observed	
	temp. (deg C)	Reaction temperature	
	time (h)	Reaction time	
Basicity	pK _{BH}	Basicity of organic base.	Ref. 8
	V _{s,min} (kJ/mol)	Electrostatic potential energy for nitrogen atom center	
	NBO (a.u.)	NBO charge for nitrogen atom center	
	delta G (kcal/mol)	ΔG: Gibbs free energy for nitrogen atom center	
	HBD	Number of hydrogen-bonded donor centers connected to or conjugated with the basicity center	
Structure	NN	Number of Nitrogen atom in organic base	Ref. 8
	Cyclic/Acyclic	Type of structure for organic base (Equal to 1 if the structure is cyclic; equal to 0 if the structure is acyclic)	
	NRing	Number of ring in organic base	
	Ring Size of N	Size of the ring containing the basicity center (“0” if the center is not part of a ring)	
	Ring Size of S	Size of the ring that is out of the basicity center (“0” if the organic base is without ring structure or no	

		ring structure in the out of active center)	
Solvent	SVI-Et(30)	Solvent index: polarity index based on molar absorption energy (kcal/mol)	Ref. 9
	SVI-n	Solvent index: Reflective index	Ref. 10
	SVI-epsilon	Solvent index: dielectric constant	
	SVI-mu	Solvent index: dipole moment	
	SVI-pi*	Solvent index: polarity index based on solvatochromism	
	SVI-PC-1	Principal components which are reported in ACS solvent selection tool	Ref. 11
	SVI-PC-2		
	SVI-PC-3		
	SVI-PC-4		
	SVI-PC-5		
	SVI-DN	Solvent index: Number of Electron pair Donor	Ref. 12
	SVI-AN	Solvent index: Number of Electron pair Acceptor	
	SVI-dD	Solvent index: The energy from dispersion forces between molecules	Ref. 13
SVI-dP	Solvent index: The energy from dipolar intermolecular forces between molecules		
SVI-dH	The energy from hydrogen bonds between molecules		

6-2. Missing values

Some missing values such as pK_{B-H} of DBU in CH_2Cl_2 , toluene, diethyl ether, MTBE, glyme, and 1,4-dioxane, ΔG of DBU in MTBE and glyme, NBO of DBU in MTBE and glyme, $V_{s,min}$ of DBU in MTBE and glyme, SVI-pi* of MTBE, SVI-DN of MTBE, and SVI-AN of MTBE, were complemented by Datachemical LAB¹⁴ using VBGMR as prediction model.

6-3. Heatmaps

Correlation analyses were performed by Datachemical LAB¹⁴ using the prepared data sets.

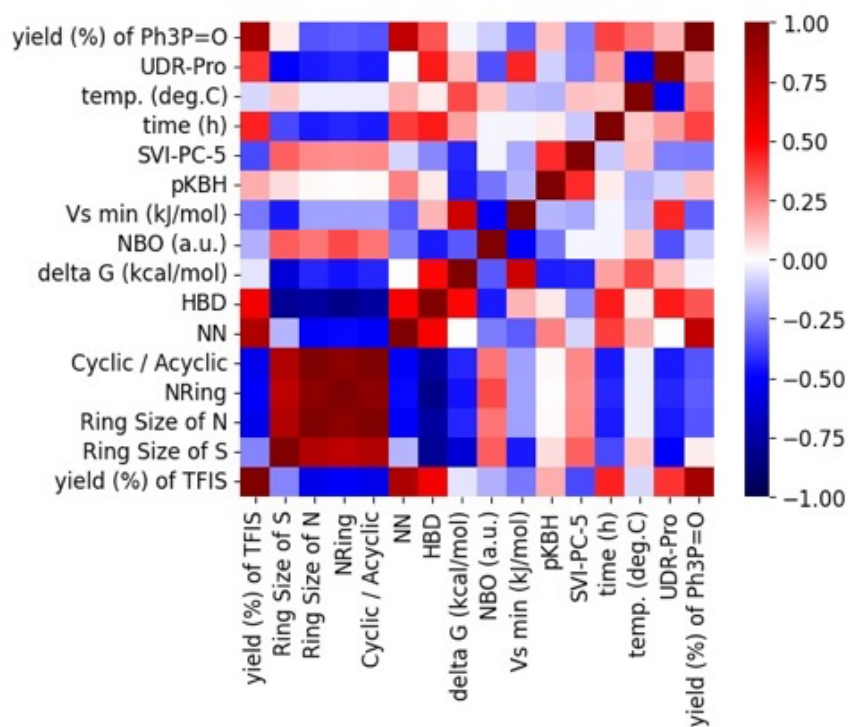


Figure S1. Heatmap for reaction, basicity, and structure.

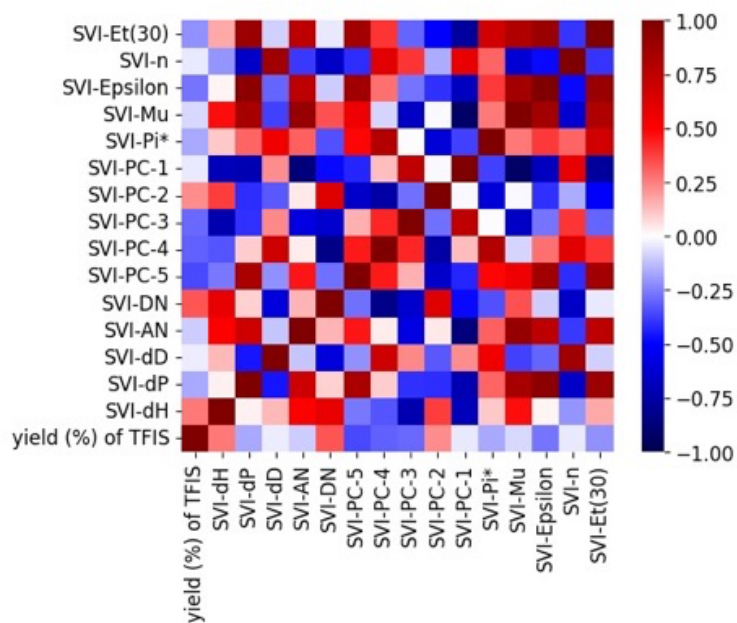


Figure S2. Heatmap for solvent index.

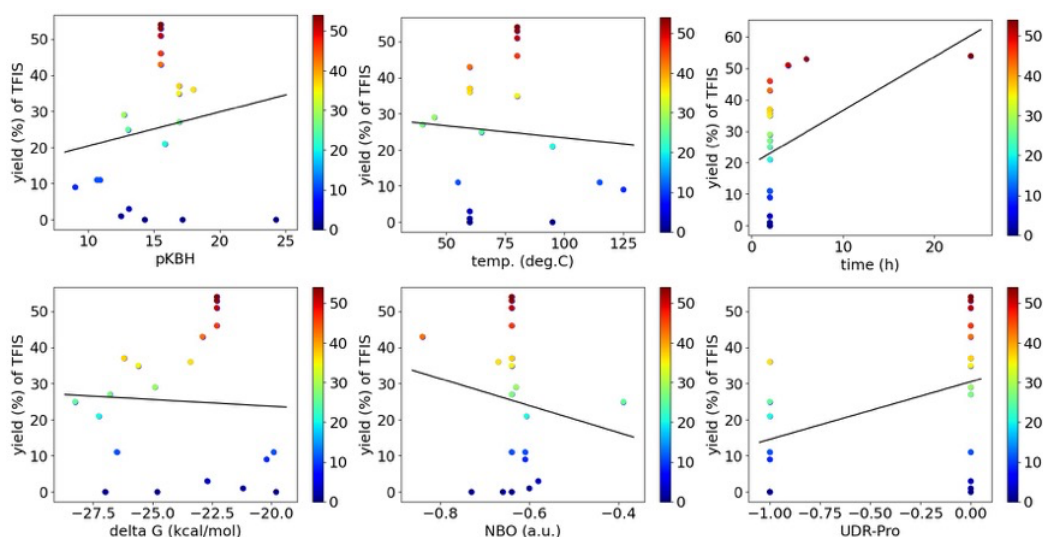
Table S6. Absolute value of coefficients $|r|$ between yield of TFIS and other descriptorsa) $|r|$ with reaction, basicity, and structure

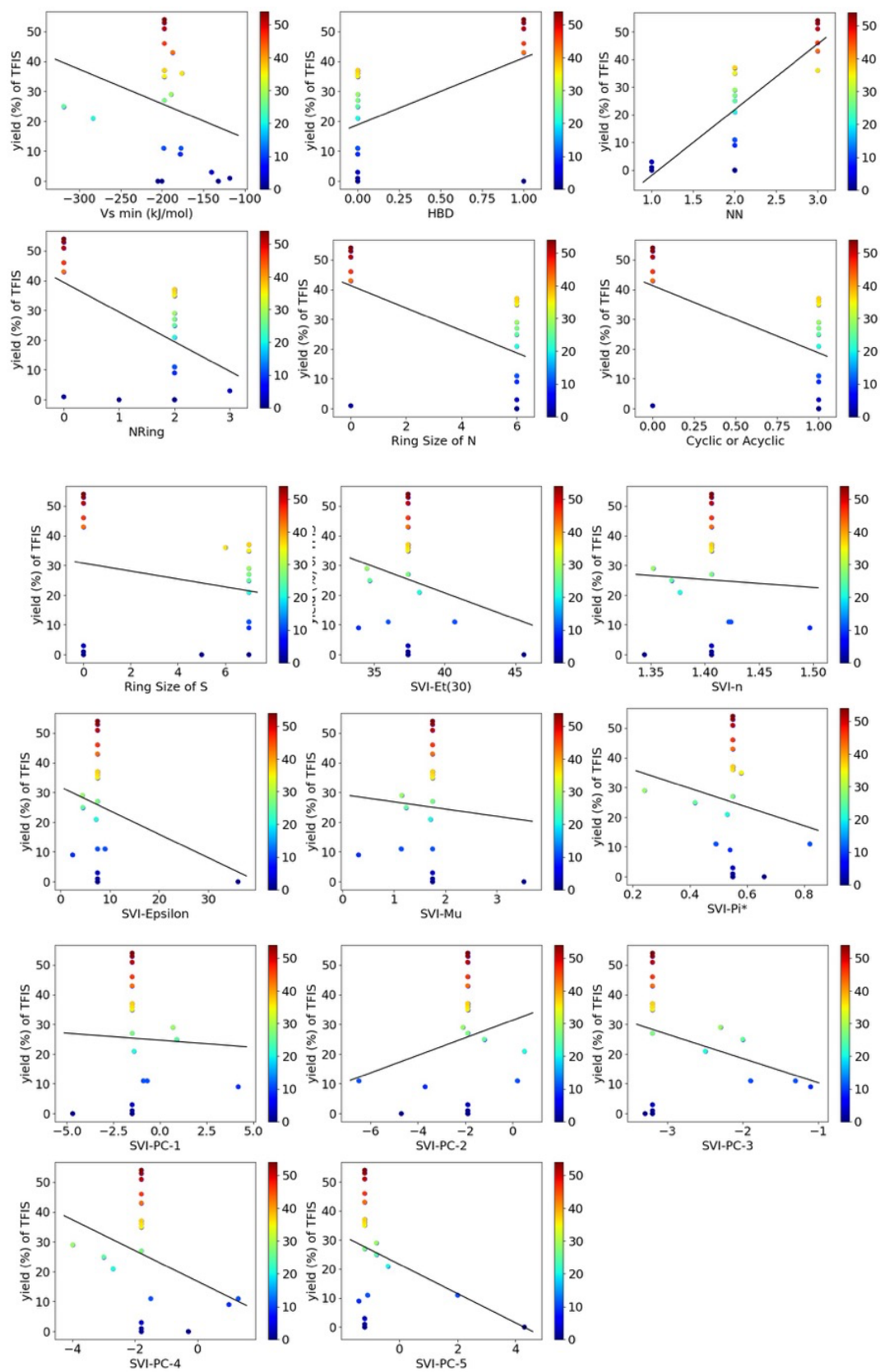
yield (%) of Ph3P=O	0.84831668
NN	0.81926259
Ring Size of N	0.55164264
Cyclic / Acyclic	0.55164264
HBD	0.54594722
NRing	0.51394616
time (h)	0.43482153
UDR-Pro	0.40571214
SVI-PC-5	0.35895377
Vs min (kJ/mol)	0.26216825
Ring Size of S	0.23785348
pKBH	0.16004195
NBO (a.u.)	0.15235514
temp. (deg.C)	0.07632792
delta G (kcal/mol)	0.05130469

b) $|r|$ with solvent index

SVI-PC-5	0.35895377
SVI-DN	0.33311941
SVI-PC-4	0.31192379
SVI-PC-3	0.29447727
SVI-epsilon	0.27164037
SVI-dH	0.26296359
SVI-PC-2	0.22569313
SVI-Et(30)	0.21588364
SVI-dP	0.16739549
SVI-pi*	0.16669091
SVI-AN	0.09911477
SVI-mu	0.07085037
SVI-n	0.04178254
SVI-PC-1	0.03975838
SVI-dD	0.03536133

6-4. Scatter plots between yield of TFIS and other descriptors



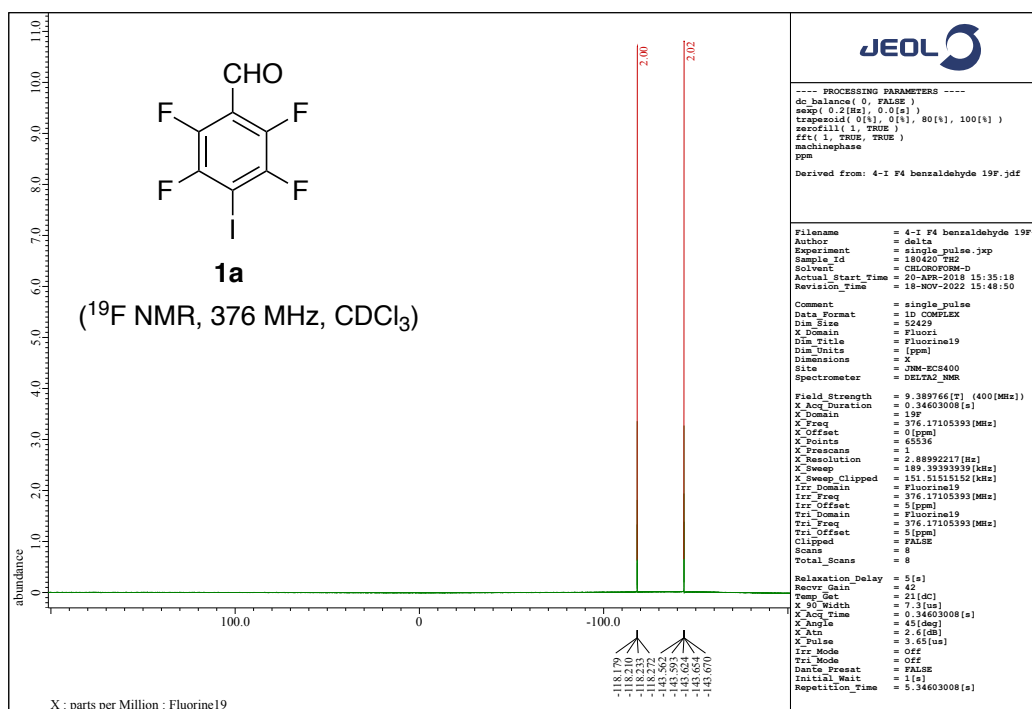
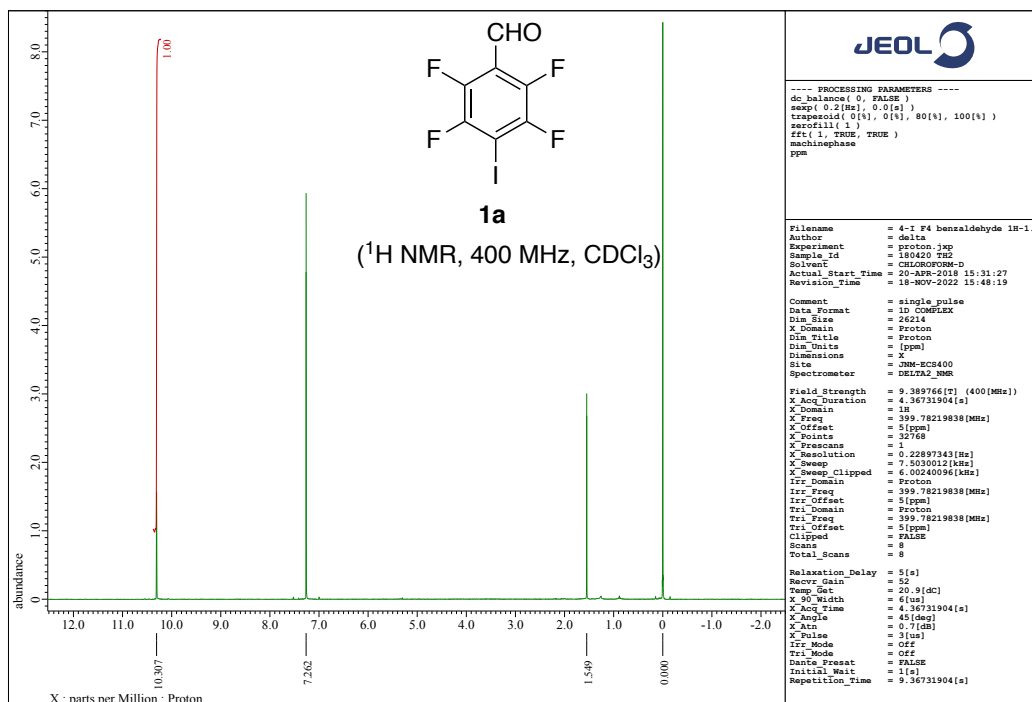


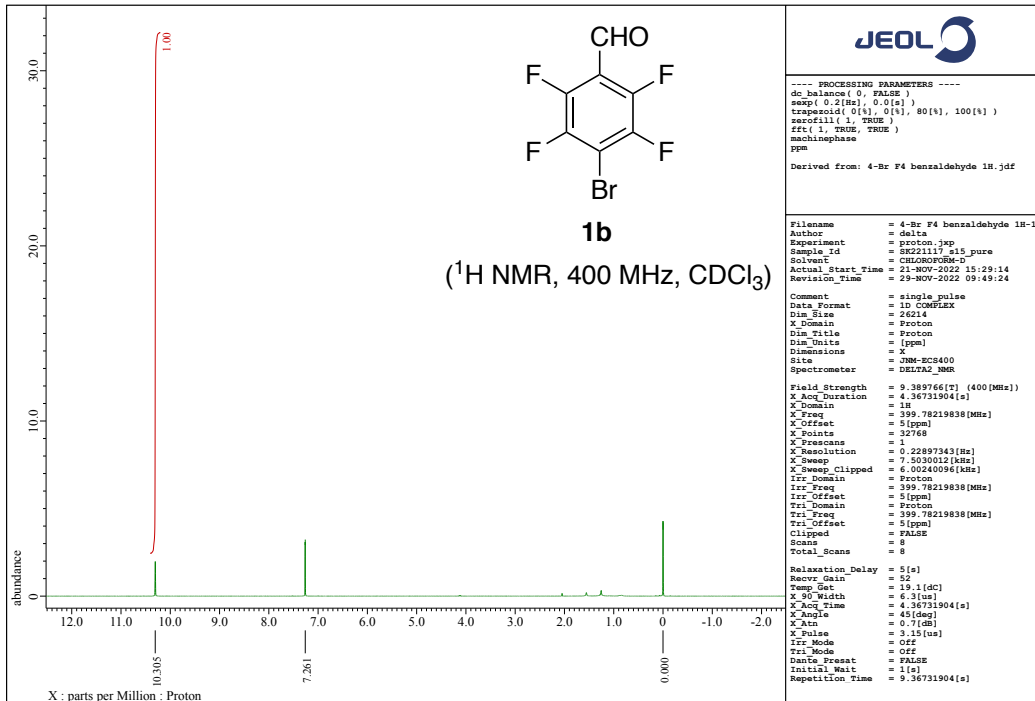
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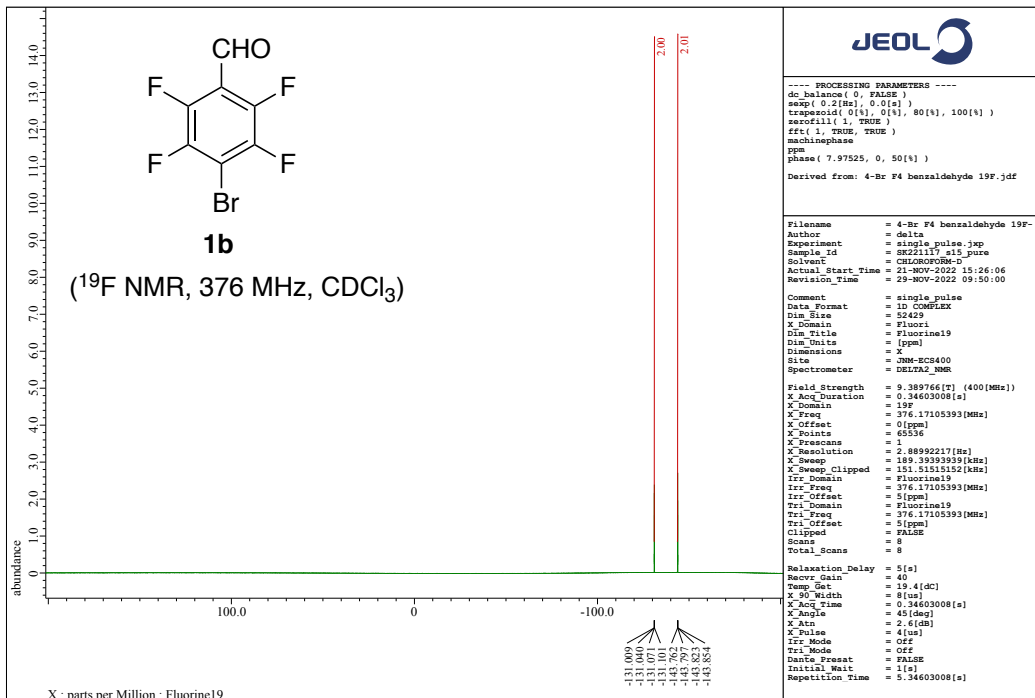
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8. NMR spectra

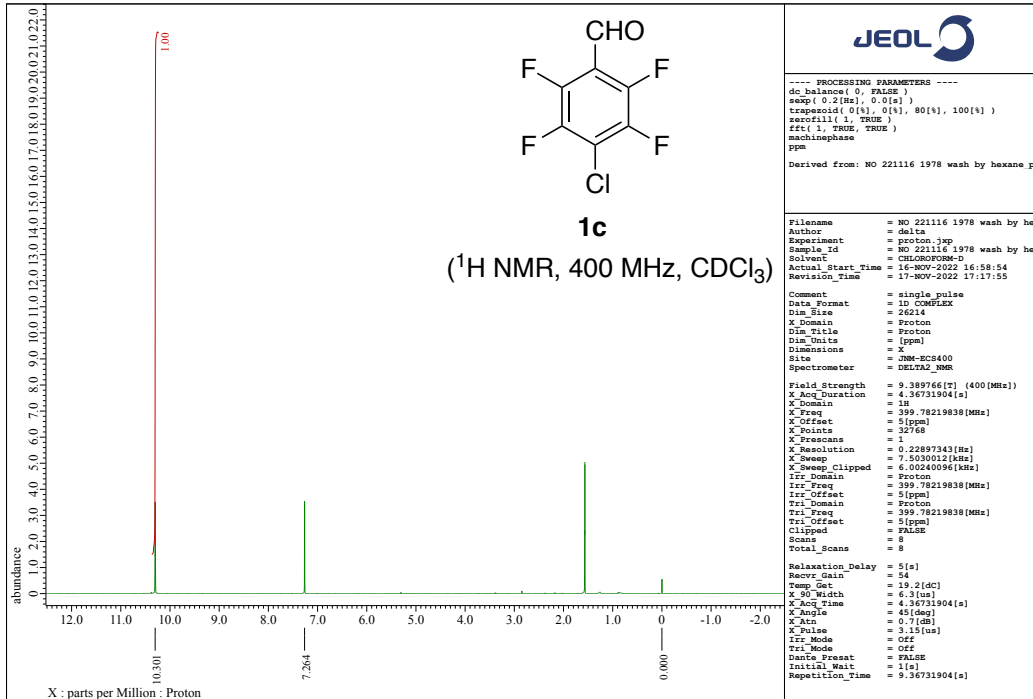




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Experiment	= proton_jmp
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Solvent	= CDCl3OBSM-D
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Revision_Time	= 29-NOV-2022 09:49:24
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Dim Title	= Proton
Dim Units	= [ppm]
Dimensions	= X
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Spectrometer	= DFXZ2_UMS
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X_Acq_Duration	= 4.36731904[s]
X_Domain	= 1H
X_Freq	= 399.78219838[MHz]
X_Offset	= 5 [ppm]
X_Points	= 32768
X_Prescans	= 1
X_Resolution	= 3.22897343[Hz]
X_Sweep	= 7.5030012[kHz]
X_Sweep_Clippped	= 6.00240096[kHz]
Irr_Domain	= Proton
Irr_Freq	= 399.78219838[MHz]
Irr_Offset	= 5 [ppm]
Tri_Domain	= Proton
Tri_Freq	= 399.78219838[MHz]
Tri_Offset	= 5 [ppm]
Clipped	= FALSE
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Total_Scans	= 8
Relaxation_Delay	= 5[s]
Recvr_Gain	= 12
Temp_Get	= 19.1[dc]
X_90_Width	= 6.31[us]
X_Acq_Time	= 4.36731904[s]
X_Angle	= 45[deg]
X_Atn	= 7.4[db]
X_Pulse	= 3.15[us]
Irr_Mode	= Off
Tri_Mode	= Off
Bank_Freat	= FALSE
Initial_Wait	= 1[s]
Repetition_Time	= 9.36731904[s]



JEOL	
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Solvent	= CDCl3OBSM-D
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Dim Title	= Fluorine19
Dim Units	= [ppm]
Dimensions	= X
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Spectrometer	= DFXZ2_UMS
Field Strength	= 9.389766[T] (400[MHz])
X_Acq_Duration	= 0.34603008[s]
X_Domain	= 19F
X_Freq	= 376.17105393[MHz]
X_Offset	= 0 [ppm]
X_Points	= 65536
X_Prescans	= 1
X_Resolution	= 2.8892217[Hz]
X_Sweep	= 189.39393939[kHz]
X_Sweep_Clippped	= 151.51515152[kHz]
Irr_Domain	= Fluorine19
Irr_Freq	= 376.17105393[MHz]
Irr_Offset	= 5 [ppm]
Tri_Domain	= Fluorine19
Tri_Freq	= 376.17105393[MHz]
Tri_Offset	= 5 [ppm]
Clipped	= FALSE
Scans	= 8
Total_Scans	= 8
Relaxation_Delay	= 5[s]
Recvr_Gain	= 40
Temp_Get	= 19.4[dc]
X_90_Width	= 8[us]
X_Acq_Time	= 0.34603008[s]
X_Angle	= 45[deg]
X_Atn	= 2.6[db]
X_Pulse	= 4[us]
Irr_Mode	= Off
Tri_Mode	= Off
Bank_Freat	= FALSE
Initial_Wait	= 1[s]
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JEOL

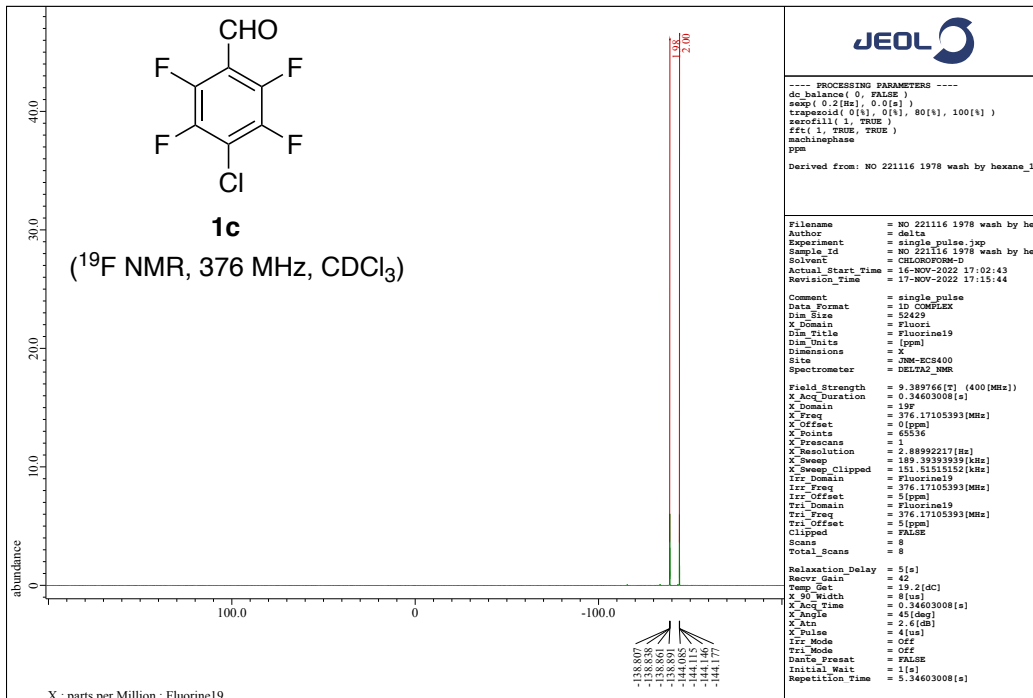
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 zero fill (1, TRUE)
 ifft (1, TRUE, TRUE)
 machine phase
 ppm
 Derived from: NO 221116 1978 wash by hexane p

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 Dim Title = Proton
 Dim Units = [ppm]
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 X_Domain = 1H
 X_Freq = 399.78219838 [MHz]
 X_Offset = 5 [ppm]
 X_Points = 32748
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 X_Sweep = 7.5030012 [kHz]
 X_Sweep_Clippped = 4.00240096 [kHz]
 Irr_Domain = Proton
 Irr_Freq = 399.78219838 [MHz]
 Irr_Offset = 5 [ppm]
 Tri_Domain = Proton
 Tri_Freq = 399.78219838 [MHz]
 Tri_Offset = 5 [ppm]
 Clipped = FALSE
 Scans = 8
 Total_Scans = 8

Relaxation_Delay = 5 [s]
 Recvr_Gain = 44
 Temp_Get = 19.2 [dC]
 X_90_Width = 6.3 [us]
 X_Acq_Time = 4.36731904 [s]
 X_Angle = 45 [deg]
 X_Atn = -7 [dB]
 X_Pulse = 3.15 [us]
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 Dnfmr_Treat = FALSE
 Initial_Wait = 1 [s]
 Repetition_Time = 9.36731904 [s]



JEOL

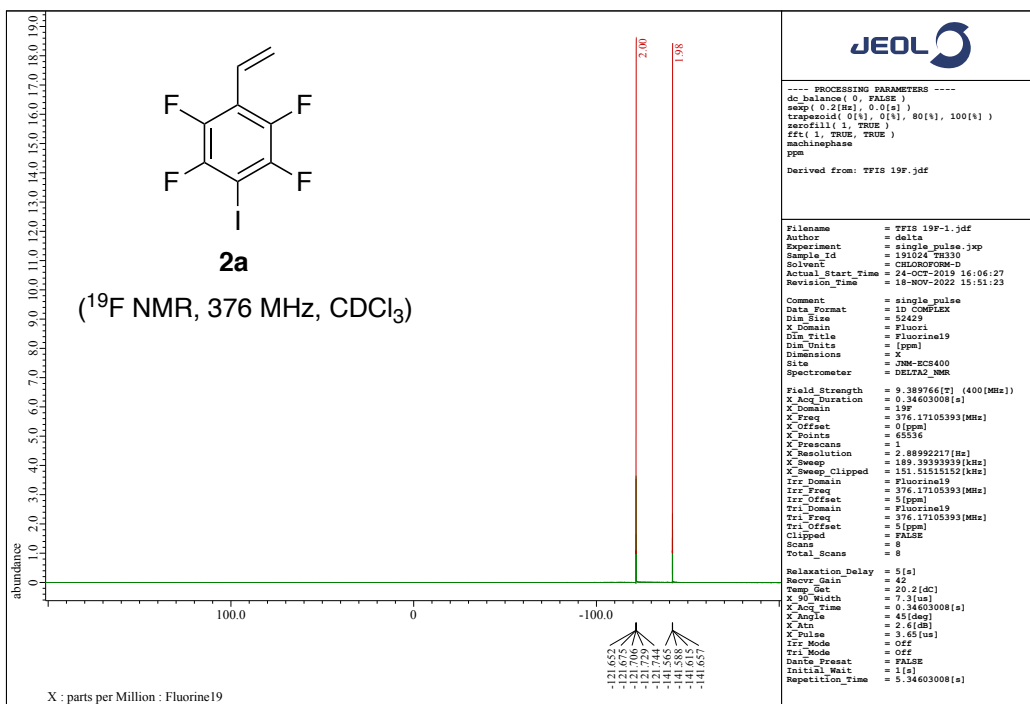
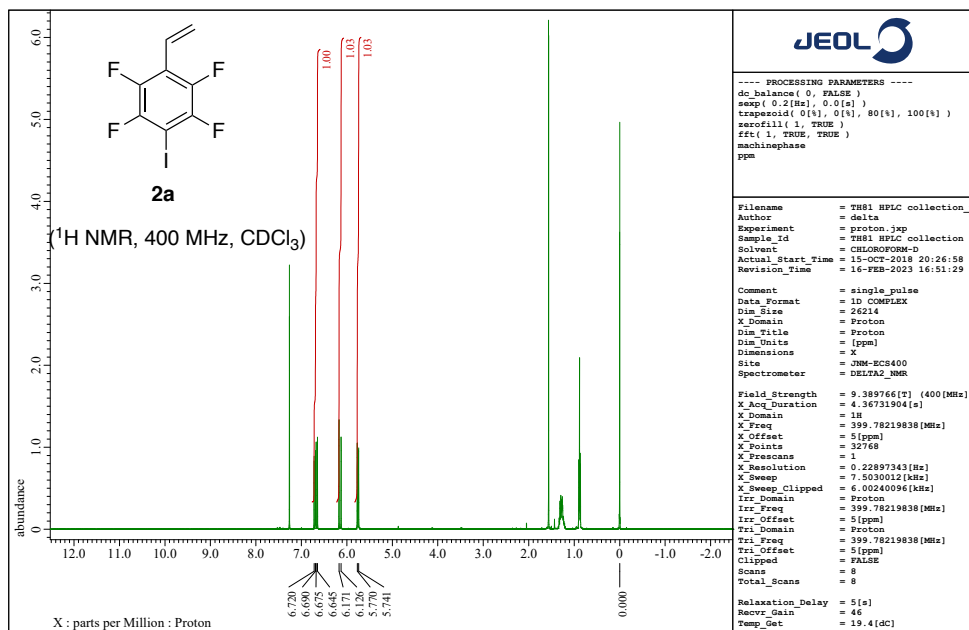
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 frequency (0 [Hz], 0 [Hz], 50 [Hz], 100 [Hz])
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 ifft (1, TRUE, TRUE)
 machine phase
 ppm
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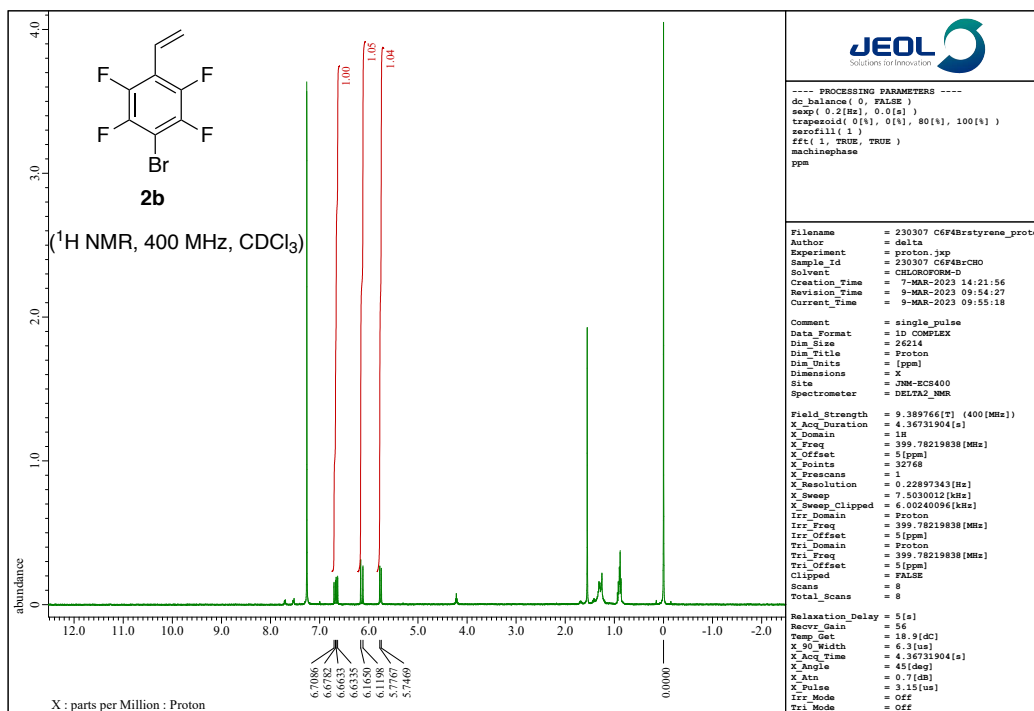
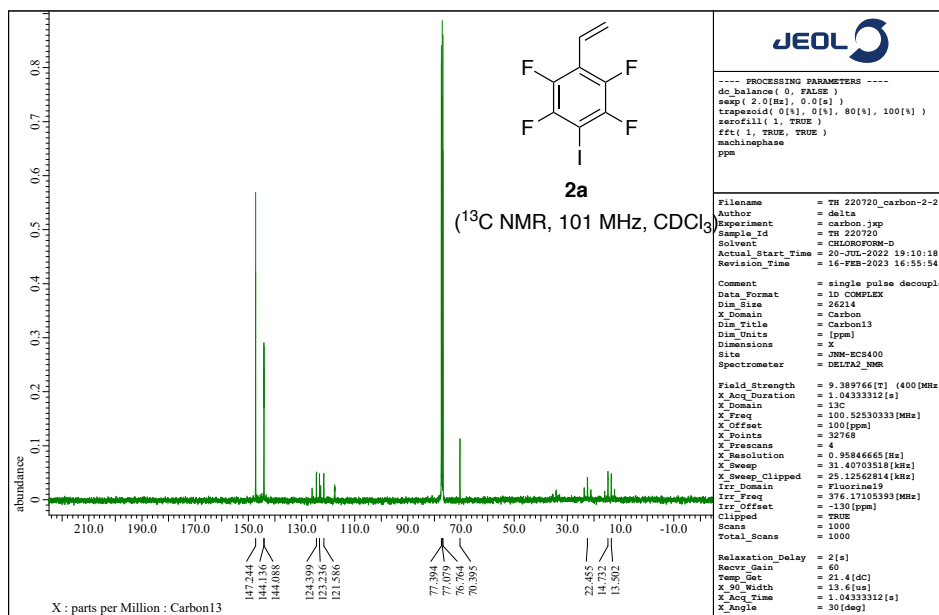
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 Author = delta
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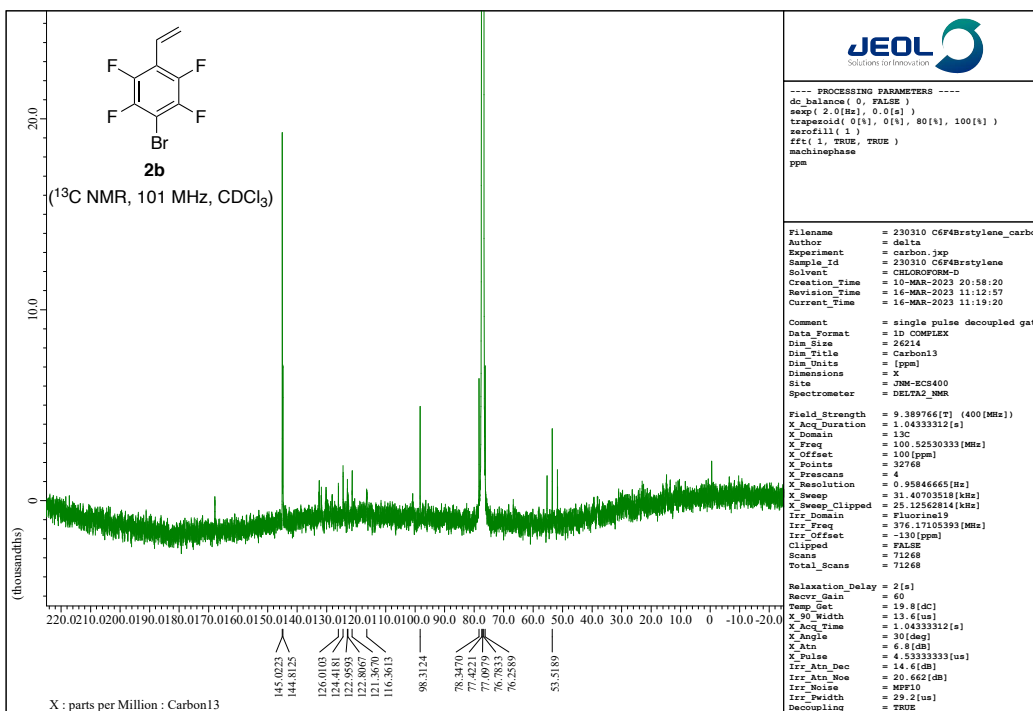
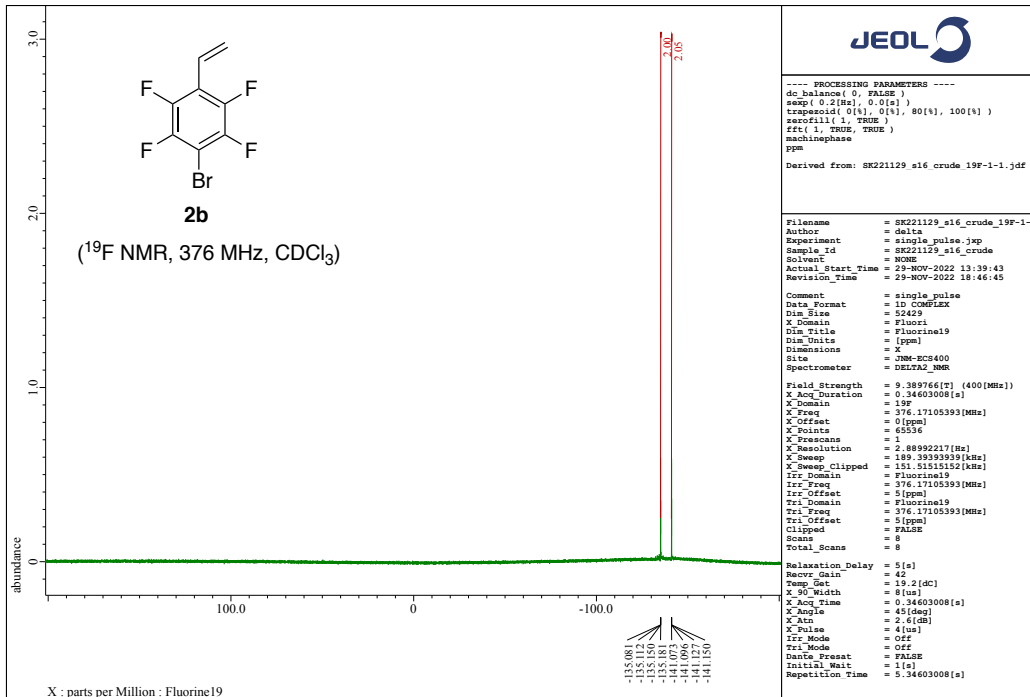
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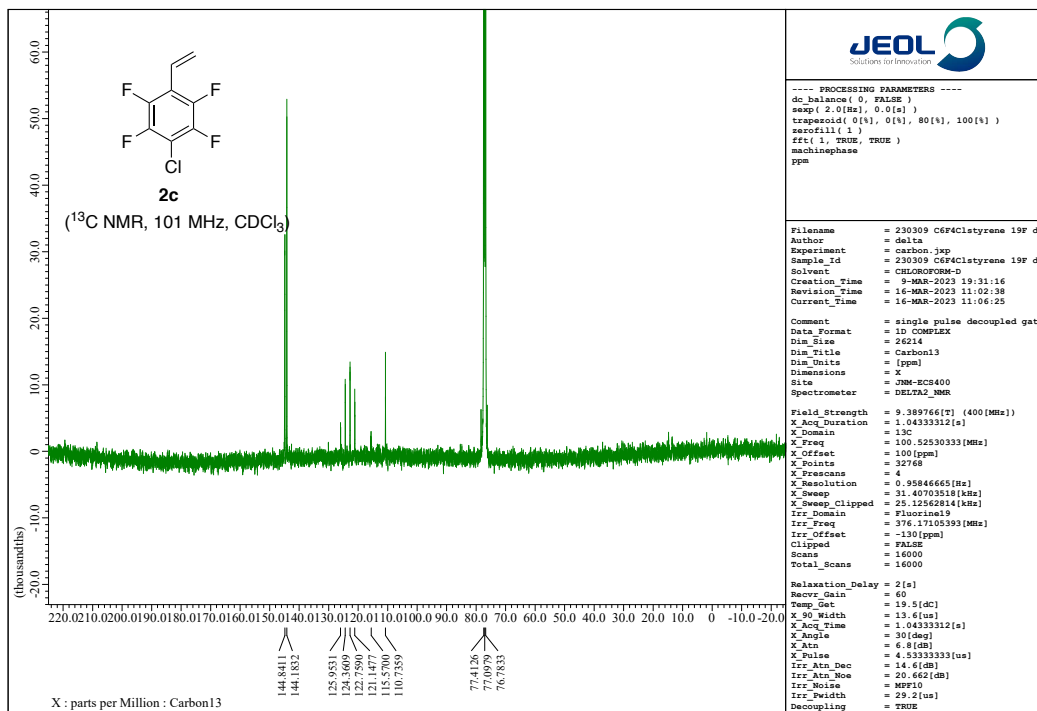
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 X_Sweep = 189.39393939 [kHz]
 X_Sweep_Clippped = 151.51515152 [kHz]
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 Irr_Freq = 376.17105393 [MHz]
 Irr_Offset = 5 [ppm]
 Tri_Domain = Fluorine19
 Tri_Freq = 376.17105393 [MHz]
 Tri_Offset = 5 [ppm]
 Clipped = FALSE
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 Total_Scans = 8

Relaxation_Delay = 5 [s]
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 X_Pulse = 4 [us]
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 Tri_Mode = Off
 Dnfmr_Treat = FALSE
 Initial_Wait = 1 [s]
 Repetition_Time = 5.34603008 [s]









9. Cartesian coordinates

SMD(THF)/M06-2X-D3/6-311+G(d,p)

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	C	2.3327910	-0.110016	0.196553
	N	1.2851200	0.822431	-0.210660
	C	-0.0534970	0.477837	-0.036416
	N	-0.3519410	-0.879547	-0.113837
	C	0.6397190	-1.893801	0.230145
	N	-0.9170250	1.424617	0.097599
	C	-2.3355060	1.103661	0.187886
	C	-2.6897980	-0.261040	-0.385280
	C	-1.7225590	-1.277964	0.188443
	C	1.6276240	2.222664	-0.026893
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	H	2.0062600	-1.521448	-1.385327
	H	2.4536920	-0.109576	1.291172
	H	3.2708990	0.240120	-0.237344
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	N	0.8327260	-1.493327	0.058036
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	C	2.1347940	0.605709	-0.413682
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	C	-1.5637440	1.174287	0.199331
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	H	-1.7471400	1.402050	1.258290
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	H	-2.5817590	0.109030	-1.375259
	H	-1.7780240	-2.055963	-0.646741
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DBU	C	-0.3996040	0.779957	-0.362430
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	N	-1.4775850	1.451142	-0.134876
	C	-2.7021520	0.673102	0.010899
	C	-2.4503380	-0.619957	0.778252
	C	-1.4234370	-1.437073	0.015015
	C	0.9033620	-1.287631	-0.811225
	C	1.9608660	-1.310327	0.294824
	C	2.7555870	-0.008392	0.385790
	C	1.9099220	1.245609	0.613626
	C	0.8950360	1.548091	-0.502897
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	H	-3.1075880	0.430495	-0.981621
	H	-3.3670780	-1.199732	0.901906
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	H	3.3210160	0.118704	-0.545832
	H	2.5840830	2.102315	0.697998
	H	1.3787460	1.172049	1.569665
	H	1.3509470	1.353609	-1.479851
	H	0.6177120	2.601354	-0.478841
TMG	C	-2.3980710	0.548984	-0.372276
	N	-1.1717170	-0.189521	-0.124589
	C	-0.0069810	0.554755	-0.007916
	N	1.1543680	-0.202238	0.114045
	C	2.3885290	0.502159	0.409433
	C	-1.3778950	-1.352177	0.734058
	N	-0.0646900	1.842669	-0.007767
	C	1.3575560	-1.363736	-0.748421
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	H	2.2455850	1.192539	1.241622
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	H	2.7712870	1.060134	-0.457015
	H	-1.9299490	-1.066957	1.639136
	H	-1.9592120	-2.107104	0.198967
	H	-0.4297760	-1.788903	1.037322

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	H	1.9554060	-1.087828	-1.626526
	H	1.8891780	-2.145219	-0.199631
piperidine	C	0.7473950	-1.210570	0.206149
	C	-0.7145130	-1.258097	-0.229077
	N	1.3803360	-0.000001	-0.321053
	C	0.7473960	1.210569	0.206149
	C	-0.7145120	1.258097	-0.229077
	C	-1.4515220	0.000001	0.236373
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	H	0.7911170	-1.240158	1.308618
	H	-0.7547310	-1.324686	-1.321773
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	H	-1.5019230	0.000001	1.332208
quinuclidine	H	1.0147230	-2.086630	0.509705
	H	-1.2959800	-1.902641	0.861284
	H	1.2028050	-1.035674	1.913119
	H	-1.0881450	-0.562444	1.987590
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	H	1.2083860	2.171486	-0.056033
	H	-1.0816300	2.005164	-0.510328
	H	1.0216360	0.603706	-2.057890
	H	-1.2882040	0.202703	-2.083600
	H	1.2121210	-1.136864	-1.848821
	H	-1.0795520	-1.442127	-1.484432
	H	2.3827590	-0.000645	0.005820
	C	1.2910160	-0.000337	0.003118
	C	0.7505480	-1.107306	0.920362
	C	-0.7915000	-0.944349	1.007013
	C	0.7523040	1.348875	0.501228
	C	-0.7891830	1.346303	0.311720
	C	0.7565800	-0.242131	-1.416200
	C	-0.7855340	-0.401336	-1.324321
	N	-1.2902720	0.000395	-0.003173
Et ₃ N	C	-0.4749240	-1.308636	0.284321
	N	-0.0001130	0.000282	-0.172069
	C	-0.8964540	1.065502	0.284894

	C	0.3407890	-2.482232	-0.246538
	C	-2.3202550	0.946142	-0.246833
	C	1.3704630	0.243746	0.285500
	C	1.9803460	1.535420	-0.247163
	H	-1.5016630	-1.428805	-0.062703
	H	-0.5020430	-1.342421	1.389589
	H	-0.9128670	1.104662	1.390189
	H	-0.4866610	2.015039	-0.060729
	H	-0.2003310	-3.413166	-0.062373
	H	1.3167000	-2.572148	0.232666
	H	0.4915930	-2.383693	-1.325019
	H	-2.8558630	1.880296	-0.062843
	H	-2.3098700	0.766353	-1.325315
	H	-2.8863380	0.146008	0.232161
	H	1.9876370	-0.586292	-0.059396
	H	1.4122490	0.239185	1.390789
	H	3.0568750	1.531742	-0.061521
	H	1.5704740	2.426804	0.229883
	H	1.8211190	1.614812	-1.325986
MTBT·HBr	C	-1.0068240	-2.460757	-1.097889
	C	-1.7306070	-1.774405	0.040751
	N	-0.7650340	-1.247760	1.011102
	C	0.4062580	-0.733036	0.596751
	N	0.8827370	-0.975494	-0.635382
	C	0.0047440	-1.499346	-1.683288
	N	1.1548020	-0.050298	1.484781
	C	2.2823050	0.787246	1.076883
	C	3.0212060	0.096342	-0.050052
	C	2.0282060	-0.217373	-1.151314
	C	-1.3291950	-0.802749	2.280801
	H	-1.7234350	-2.754074	-1.866009
	H	-0.5040110	-3.360684	-0.734225
	H	-2.3400500	-0.941727	-0.327903
	H	-2.3714760	-2.477618	0.572930
	H	-0.4953780	-0.650972	-2.165040
	H	0.6414920	-1.997481	-2.415959
	H	2.9198870	0.927184	1.949511
	H	1.9020660	1.758666	0.745763
	H	3.8037040	0.748410	-0.439786
	H	3.4868190	-0.823960	0.312189
	H	1.6474950	0.705071	-1.605293
	H	2.4879840	-0.829153	-1.928659
	H	-1.5421420	0.271577	2.251504
	H	-2.2609460	-1.341169	2.439704
	H	-0.6633080	-1.039318	3.112933
	H	0.6627570	0.276674	2.302520
	Br	-1.0461590	1.916226	-0.437244

DBN·HBr	C	-0.9097890	0.272460	-0.081282
	N	-2.2246700	0.140883	-0.073546
	N	-0.0841440	-0.740733	-0.113131
	C	-0.5681330	-2.120670	-0.184151
	C	-1.9524030	-2.212454	0.447291
	C	-2.8838530	-1.158613	-0.144643
	C	-2.9216430	1.430084	-0.137570
	C	-1.8314530	2.419138	0.295597
	C	-0.5173050	1.719672	-0.088989
	H	0.1471960	-2.750996	0.343277
	H	-0.5965900	-2.435135	-1.230932
	H	-2.3689470	-3.205234	0.277423
	H	-1.8708940	-2.056074	1.525751
	H	-3.8145700	-1.094772	0.420623
	H	-3.1271160	-1.383621	-1.187694
	H	-3.7834030	1.414837	0.530031
	H	-3.2666190	1.607135	-1.160773
	H	-1.9435430	3.389539	-0.183452
	H	-1.8723610	2.556276	1.376976
	H	0.3198740	1.912672	0.580446
H	-0.1980360	1.971357	-1.104782	
H	0.9333280	-0.540060	-0.087416	
Br	3.0603100	0.077581	0.016278	

DBU·HBr	C	-0.6348050	0.189540	-0.408120
	N	-1.8630840	0.688273	-0.394588
	N	0.4212110	0.948594	-0.207686
	C	0.3428440	2.382042	0.047829
	C	-0.9667000	2.675250	0.759743
	C	-2.1178770	2.095981	-0.044250
	C	-3.0406850	-0.137676	-0.705519
	C	-3.3900680	-1.130789	0.401117
	C	-2.5182030	-2.385837	0.385278
	C	-1.0180000	-2.129207	0.526923
	C	-0.4095390	-1.282232	-0.605115
	H	1.2061600	2.655177	0.654081
	H	0.4010720	2.924717	-0.900005
	H	-1.1080200	3.750249	0.871782
	H	-0.9477680	2.224983	1.755531
	H	-2.2804190	2.661666	-0.966575
	H	-3.0403950	2.118701	0.537863
	H	-3.8603880	0.562945	-0.862056
	H	-2.8762890	-0.656171	-1.653730
	H	-3.3166710	-0.620125	1.367716
H	-4.4354640	-1.423767	0.271050	
H	-2.8393870	-3.053804	1.189000	
H	-2.6927270	-2.919322	-0.556638	
H	-0.4970600	-3.089197	0.528909	
H	-0.8005660	-1.654143	1.489533	

	H	-0.8292450	-1.582091	-1.570372
	H	0.6691100	-1.439259	-0.650844
	H	1.3503470	0.494077	-0.168580
	Br	3.3859720	-0.402147	0.020913
TMG·HBr	C	-0.0308540	1.878926	0.683299
	N	1.1146830	1.187171	0.096466
	C	1.2237470	-0.147932	0.168299
	N	2.4058470	-0.745910	-0.068828
	C	2.4744480	-2.182813	-0.310901
	C	1.9137430	1.988357	-0.827798
	N	0.1594150	-0.878504	0.489604
	C	3.6779280	-0.115829	0.281171
	H	-0.8643890	1.940968	-0.020782
	H	-0.3673710	1.358011	1.578361
	H	0.2931740	2.883468	0.958103
	H	1.6259080	-2.510131	-0.910627
	H	3.3881460	-2.386969	-0.868950
	H	2.5041790	-2.746840	0.627605
	H	1.2313510	2.572056	-1.449708
	H	2.5708370	2.672271	-0.285650
	H	2.5081050	1.348800	-1.476260
	H	0.2766370	-1.852629	0.726987
	H	3.5083290	0.825936	0.797703
	H	4.2198720	-0.785548	0.953177
	H	4.2854190	0.060005	-0.609220
	H	-0.8078390	-0.554217	0.327230
	Br	-3.0196010	-0.236821	-0.112374
piperidine·HBr	C	1.0839030	-1.243801	0.001518
	C	2.6014360	-1.256828	0.096221
	N	0.5376680	-0.000087	0.619343
	C	1.0837410	1.243719	0.001552
	C	2.6012710	1.256919	0.096267
	C	3.2005970	0.000096	-0.536810
	H	0.6246090	-2.088369	0.514213
	H	0.7467770	-1.238533	-1.037402
	H	2.8951160	-1.319902	1.150198
	H	2.9700390	-2.158029	-0.397594
	H	-0.5159650	-0.000166	0.498557
	H	0.6243350	2.088207	0.514275
	H	0.7466190	1.238427	-1.037369
	H	2.9697690	2.158185	-0.397509
	H	2.8949360	1.319988	1.150248
	H	4.2858610	0.000166	-0.417879
	H	2.9900470	0.000105	-1.612014
	H	0.7373240	-0.000082	1.623695
	Br	-2.5473950	-0.000001	-0.081123

quinuclidine·HBr	H	2.5624050	1.814239	1.141931
	H	0.2662270	1.997477	0.623412
	H	2.8233060	2.096060	-0.580093
	H	0.5599450	1.756376	-1.107163
	H	2.5645630	0.080587	-2.140642
	H	0.2678200	-0.456959	-2.043065
	H	2.8226690	-1.552228	-1.523950
	H	0.5578690	-1.835326	-0.968315
	H	2.5608470	-1.895764	1.002097
	H	0.2635210	-1.539021	1.414804
	H	2.8190800	-0.545364	2.107801
	H	0.5550010	0.080293	2.072008
	H	3.9693150	-0.001146	0.002536
	C	2.8793960	-0.000636	0.001424
	C	2.3446700	1.432654	0.141685
	C	0.8244390	1.412121	-0.106784
	C	2.3451190	-0.594855	-1.310498
	C	0.8244220	-0.797402	-1.170316
	C	2.3424270	-0.838840	1.171375
	C	0.8219440	-0.613839	1.274532
	N	0.3455560	0.000576	-0.001348
	H	-0.7119100	0.001003	-0.002042
	Br	-2.8169730	0.000015	0.000046

Et ₃ N·HBr	C	1.4908260	-0.412473	1.378027
	N	1.0665400	-0.000564	0.001847
	C	1.5049420	1.392893	-0.327458
	C	1.0954160	-1.836691	1.727304
	C	1.1101770	2.411698	0.727306
	C	1.4975630	-0.987785	-1.038692
	C	1.1174480	-0.578221	-2.450945
	H	1.0045890	0.277667	2.065319
	H	2.5717730	-0.261131	1.436286
	H	2.5868600	1.358899	-0.478673
	H	1.0277790	1.645956	-1.272608
	H	1.2573760	-1.981673	2.796768
	H	1.6890220	-2.579859	1.194989
	H	0.0357380	-2.005732	1.520258
	H	1.2817010	3.408242	0.316825
	H	0.0484920	2.324168	0.970857
	H	1.6975240	2.320972	1.640955
	H	1.0063220	-1.926379	-0.787544
	H	2.5770290	-1.117082	-0.926888
	H	1.2805590	-1.434214	-3.107940
	H	1.7196560	0.249609	-2.824740
	H	0.0599260	-0.307750	-2.503904
	H	0.0072220	0.003600	-0.003010
	Br	-2.1204450	0.002632	-0.004090

HBr	Br	0.000000	0.000000	0.039511
	H	0.000000	0.000000	-1.382876