2	A Multi-modal Pre-training Transformer for
3	Universal Transfer Learning in Metal-Organic
4	Frameworks
5	Yeonghun Kang <sup><math>t, \perp</math></sup> , Hyunsoo Park <sup><math>t, \perp</math></sup> , Berend Smit <sup>‡</sup> , and Jihan Kim <sup><math>t, *</math></sup>
6	<sup>†</sup> Department of Chemical and Biomolecular Engineering, Korea Advanced Institute of Science
7	and Technology (KAIST), 291, Daehak-ro, Yuseong-gu, Daejeon 34141, Republic of Korea
8	<sup>‡</sup> Laboratory of molecular simulation (LSMO), Institut des Sciences et Ingénierie Chimiques,
9	Valais, Ecole Polytechnique Fédérale de Lausanne (EPFL), Rue de l'Industrie 17, CH-1951,
10	Sion, Switzerland
11	$\perp$ These authors contributed equally to this work
12	*Correspondence to : jihankim@kaist.ac.kr
13	

# 14 ABSTRACT

Metal-organic frameworks (MOFs) are a class of crystalline porous materials that exhibit a vast 15 16 chemical space due to their tunable molecular building blocks with diverse topologies. Given that 17 an unlimited number of MOFs can, in principle, be synthesized, constructing structure-property 18 relationships through a machine learning approach allows for efficient exploration of this vast 19 chemical space, resulting in identifying optimal candidates with desired properties. In this work, 20 we introduce MOFTransformer, a multi-model Transformer encoder pre-trained with 1 million 21 hypothetical MOFs. This multi-modal model utilizes integrated atom-based graph and energy-grid 22 embeddings to capture both local and global features of MOFs, respectively. By fine-tuning the 23 pre-trained model with small datasets ranging from 5,000 to 20,000 MOFs, our model achieves 24 state-of-the-art results for predicting across various properties including gas adsorption, diffusion, 25 electronic properties, and even text-mined data. Beyond its universal transfer learning capabilities, 26 MOFTransformer generates chemical insights by analyzing feature importance through attention 27 scores within the self-attention layers. As such, this model can serve as a bedrock platform for other MOF researchers that seek to develop new machine learning models for their work. 28

## 30 Introduction

31 Metal-organic frameworks (MOFs) are a class of crystalline porous materials used for various energy and environmental applications<sup>1-4</sup> due to their excellent properties such as large surface 32 area,<sup>5</sup> high chemical/thermal stability,<sup>6</sup> and tunability.<sup>7</sup> Given that MOFs are composed of 33 34 thousands of tunable molecular building blocks (i.e., metal nodes and organic linkers), an infinite number of MOFs can, in principle, be synthesized taking into all the different combinations. To 35 36 efficiently explore this vast MOF search space, it is important to identify the structure-property 37 relationship for a given application. One can then focus on MOFs that contain specific structures 38 that can lead to user-desired properties. To gain information regarding this relationship, high-39 throughput computational screening approaches has been primarily used by conducting simulations on a large dataset of MOF structures and retroactively identifying the 40 structure/property relationship.8-11 However, this can be a cumbersome process and more 41 42 importantly, one would need to conduct independent computational screenings for each of the 43 applications, which requires a vast quantity of computational resources.

44 An alternative way to discover the structure-property relationship is through a machine-learning (ML) approach, and this methodology has gained a lot of traction lately.<sup>12,13</sup> In particular, 45 46 geometric descriptors of MOF structures (e.g. void fraction and pore volume) have been used to accurately predict various gas adsorption properties.<sup>14-16</sup> Also, Bucior et al.<sup>17</sup> developed a machine 47 learning model using energy grid histograms as descriptors to predict gas uptake properties. For 48 diffusion properties, Ibrahim et al.<sup>18</sup> developed a machine-learning model to predict N<sub>2</sub>/O<sub>2</sub> 49 50 selectivity and diffusivity using geometric, atom-type, and chemical feature descriptors. For electronic properties, Rosen et al.<sup>19</sup> demonstrated that a graph neural network facilitates capturing 51 52 the underlying chemical features leading to accurate predictions in the band gap values for the

53 MOFs. Unfortunately, in all these previous studies, the developed machine-learning model cannot 54 be readily transferred from one application to another. As such, one would need to restart the 55 training process and develop a new machine-learning model from scratch for every different 56 application.

57 To remedy this issue, one can utilize transfer learning, which incorporates knowledge from one 58 machine learning application to another and, thereby, in principle, saving computational time for 59 subsequent machine learning works. Although transfer learning has been applied in a few cases 60 for MOFs, it is still limited to specific properties (e.g. transfer knowledge from gas uptake to gas diffusivity or between different gas types), limiting their utility.<sup>16,20</sup> To make transferability a 61 62 feasible solution, a universal transfer learning model that can be applied to all possible properties 63 needs to be constructed. To achieve this, machine-learning models and descriptors should capture two disparate types of features for MOFs: (1) local features (e.g., specific bonds and chemistry 64 makeup of the building blocks) and (2) global features (e.g., geometric and topological descriptors). 65 Although both the local descriptors (e.g. CGCNN,<sup>19,21</sup> chemical descriptors,<sup>18</sup> RACs,<sup>22,23</sup> and 66 building-block embedding.<sup>11,24,25</sup>) and the global features (e.g., geometric features calculated by 67 ZEO++,<sup>26</sup> the histograms of energy-grids.<sup>16,17</sup>) have been developed previously, as far as we know, 68 69 none of these works have effectively captured both the local and global features to achieve 70 universal transfer learning.

When it comes to multi-modal learning that takes in multiple inputs, the Transformer architecture<sup>27</sup> (initially proposed for sequence data such as language models) has emerged as the dominant modeling network. Given that the Transformer consists of self-attention layers, which enables handling sequences of data in parallel, it facilitates efficient training of neural networks with vast amounts of data. In 2019, Google introduced BERT, a pre-training Transformer encoder

in the language model,<sup>28</sup> and demonstrated remarkable performance in transfer learning. By fine-76 77 tuning the pre-trained BERT model, it obtained state-of-the-art performance results for many 78 Natural Language Process (NLP) tasks such as question-answering and named entity recognition. 79 Moreover, for computer vision, various vision Transformer architectures have emerged as an alternative solution to convolution neural networks (CNNs).<sup>29</sup> Recently, the pre-trained 80 Transformers' transfer learning strategy has been expanded to multi-modal learning.<sup>30</sup> And finally, 81 82 the pre-trained multi-modal Transformers achieved state-of-the-art results in vision-language models such as image captioning and vision-question answering.<sup>31-33</sup> Due to its superior 83 84 performance, the Transformer architectures have recently been adopted to predict various properties of MOFs.<sup>34,35</sup> 85

86 In this work, for the first time in MOF research, we introduce the multi-modal Transformer 87 architecture (named "MOFTransformer"), which captures both the local and global features. Our 88 MOFTransformer was pre-trained with 1 million hypothetical MOFs (hMOFs). By fine-tuning the pre-trained MOFTransformer, it showcases excellent prediction capabilities across multiple 89 90 different properties (e.g., gas uptake, gas diffusivity, electronic properties of MOFs, and text-91 mined data). Besides its superior performance, this architecture allows chemists to capture insights 92 from attention scores obtained by the attention layers of the MOFTransformer. As such, we believe 93 that this model can serve as a bedrock architecture/model for future machine learning research for 94 the MOF community.



96 97 Figure 1. (a) Overall schematics of MOFTransformer. The model takes both local and global 98 features as inputs. In a pre-training step, it is trained with three pre-training tasks. In the fine-tuning 99 step, the model is trained to predict desired properties of MOFs using the weights of the pre-trained 100 model as initial weights. (b) The architecture of the MOFTransformer. The input embedding takes 101 atom-based graph embeddings and energy-grid embeddings that serve as local and global features,

102 respectively.

# 103 **Results**

#### 104 **MOFTransformer**

105 The overall schematics of our MOFTransformer is shown in Figure 1(a). To build towards 106 universal transfer learning, both pre-training and fine-tuning strategies are implemented. The 107 objective of pre-training is to allow the MOFTransformer to learn the essential characteristics of a 108 MOF. This pre-trained model serves as a starting point for all subsequent applications. Fine-tuning 109 refers to the process of training the pre-trained models for the specific application at hand (e.g. gas 110 adsorption uptake prediction). Figure 1(b) shows the schematic of the MOFTransformer 111 architecture, which is based on a multi-layer bidirectional Transformer encoder developed by Vaswani et al.<sup>27</sup> The MOFTransformer is a multi-modal Transformer that takes two types of 112 113 embedding as inputs, each representing the local and global features: (1) atom-based graph 114 embedding (2) energy-grid embedding.

Previously, Xie et al.<sup>21</sup> devised crystal graph convolution neural networks (CGCNN) that 115 116 transforms atoms (i.e., nodes), bonds (i.e., edges), and their features (i.e., the distance between 117 atoms) into a vector space. Although CGCNN consists of convolutional layers and pooling layers 118 from the original paper, the atom-based graph embedding in the MOFTransformer uses output 119 vectors of the CGCNN without the pooling layers. It allows our model to deal with the atom-wise 120 features without losing information. It should be noted that many atoms in the unit-cell of MOFs 121 have the same embedding from the CGCNN, given that the CGCNN creates the embedding by 122 taking atom types of nodes, distances, and atom types of the neighbor nodes (see Supplementary 123 Figure S1). We grouped the topologically identical atoms and defined these sets as unique atoms 124 (the details of the algorithm are explained in Supplementary Note S1). Removing the information from the overlapping atoms enables efficient training and prevents significant memory issues that frequently appear when training with long sequences of inputs.

127 When it comes to the energy-grid embedding, the energy grids were calculated using a methane molecule probe that was selected due to its facility in modeling. Universal Force Field.<sup>36</sup> and 128 TraPPE<sup>37</sup> were used to describe adsorbate-adsorbent van der Walls interactions in MOFs and the 129 130 methane molecule, respectively. The 3D energy grids can be treated as 3D images, which means 131 that the grid points and the energy values of the energy grids serve as pixels and 1-channel colors, respectively. Similar to the Vision Transformer,<sup>29</sup> the MOFTransformer takes 1-dimensional (1D) 132 133 patches of the flattened 3D energy grids where (H, W, D) are the height, width, and depth of energy grids and (P, P, P) is the patch resolution, and  $N = HW D/P^3$  is the number of patches. Given that 134 the energy grids were interpolated to  $30 \times 30 \times 30$  Å, the height H, weight W, and depth D are 30 135 136 Å. The patch size P was set to 5 Å, so the number of patches N is 216.

The MOFTransformer model is derived from the BERT-based model<sup>28</sup> (L=12, H=768, A=12), 137 138 where L is the number of blocks, H is the hidden size, and A is the number of self-attention heads. 139 Similar to BERT's class and separate tokens, the class token [CLS] and the separate token [SEP], 140 which are learnable embedding layers, are located at the first position and between the two types 141 of embedding, respectively (see Figure 1(b)). The [CLS] token is a head token of the Transformer 142 blocks and predicts desired properties by adding a single pooling layer for the pre-training and 143 fine-tuning tasks. Apart from these, a volume token [VOL], which is the normalized cell volume, 144 is added at the final position of the input embedding because the interpolation of the energy grids 145 leads to a loss of information regarding the volume of the original energy grids. Finally, position 146 embedding and modal-type embedding, which are also learnable embedding layers, are added to 147 the input embedding by the element-wise summation. The position embedding is a vector that 148 encodes the position of the sequence, and the modal-type embedding encodes the two types of149 embedding to 0 and 1.

#### 150 Understanding MOF descriptors

151 It is important to recognize how MOF descriptors (i.e., local features and global features) 152 influence the properties of MOFs. As shown in Figure 2, H<sub>2</sub> uptake, H<sub>2</sub> diffusivity, and band gap 153 were selected as case-study applications for MOFs that represent adsorption, diffusion, and 154 electronic properties, respectively. Figure 2(a-c) shows the structure-property maps obtained from 155 the molecular simulations for each of these applications. For H<sub>2</sub> uptake and diffusivity, the data 156 was taken from our fine-tuning dataset (20,000 structures). The band gap values are obtained from 157 the QMOF database (version 13) with the PBE functional that includes a total of 20,373 structures. 158 From Figure 2(a-b), it can be seen that the H<sub>2</sub> uptake and diffusivity increase with accessible 159 volume fraction and are strongly dependent on the MOF topology due to the correlation between 160 topology and void fraction. Meanwhile, the band gap exhibits no correlation with accessible 161 volume fraction and topology, which is reasonable given that electronic properties are more 162 dependent on local chemical features as opposed to global geometric features.

163 On top of this, Figure 2(d-f) shows the correlation between the MOF properties and the types of 164 metal atoms. It can be seen that the dependence on metal atoms is lowest for H<sub>2</sub> uptake while 165 highest for the band gap energy. And similar trends can be found for the organic linkers (see 166 Supplementary Figure S2). Along with the aforementioned geometric analysis, Figure 2(d-f) 167 confirms that adsorption and diffusion properties rely more on global features, while electronic 168 properties rely more on local features. Apart from these, some properties like O<sub>2</sub> diffusivity (which 169 is more dependent on electronic effects than H<sub>2</sub> diffusivity) and CO<sub>2</sub> Henry coefficient have more 170 complex correlations between features and properties (see Supplementary Figure S3). As such,

- 171 this illustrates the importance of integrating both local and global features within the Transformer
- 172 to enable universal transferability across different applications.



174

Figure 2. (a-c) Scattered plots showing the relationship between accessible volume fraction and various properties (i.e., gas uptake, diffusivity, and bandgap). Gray dots represent the MOFs from each database, and colored dots represent MOFs with the top four topologies obtained from MOFid.<sup>38</sup> (d-f) The box plot of properties (adsorption, diffusion, and band gap) for each metal type. The dark line in the center of the box represents the median.

#### 180 **Pre-training Results**

181 The pre-training tasks play an essential role in determining the effectiveness of the transfer 182 learning performance. Three pre-training tasks were designed to capture the essential features of 183 the MOFs: (1) MOF topology prediction (MTP), (2) void fraction prediction (VFP), and (3) metal 184 cluster/organic linker classification (MOC). For the MTP task, the model was trained to predict 185 the 1,079 topologies of MOFs by adding a classification head, which consists of a single dense 186 layer to the [CLS] token. The list of topologies is summarized in Supplementary Table S1. For the 187 VFP task, the model is trained to predict accessible void fraction calculated by  $ZEO++^{26}$  by adding 188 single dense layers to the [CLS] token. Finally, the MOC task was performed as it would enable 189 the model to learn the features separately stemming from each metal node and organic linker. The 190 binary classification (determining a given MOF atom as belonging to the metal or the organic 191 linker) is conducted for the atom-wise features of atom-based embedding. The accuracies of MTP 192 and MOC were 0.97, 0.98 and the MAE of VFP was 0.01.

193 Next, we visualized the embedding vector of the pre-training model in a two-dimensional space 194 using t-SNE, and PCA methods, as shown in Figure 3. Figure 3(a) shows a result of a t-SNE plot 195 for the embedding vector of class tokens with the top 10 frequently appearing topologies in the 196 dataset. Figure 3(a) shows that MOFs with different topologies are clustered together and 197 segregated from other MOFs, indicating that proper learning has occurred. And the same pattern 198 of results was seen for all topologies (see Supplementary Figure S4). Furthermore, it is interesting 199 to note that the PCA plots exhibit the distribution of the embedding vector that gradually increases 200 according to the void fraction, as shown in Figure 3(b). This indicates that the embedding vectors 201 are clustered with similar values of void fraction. These results demonstrate that the pre-training 202 model is successfully trained to capture the critical features of the MOFs.



Figure 3. (a) For the top 10 frequently appearing topologies, the t-SNE plot embeds the class tokens of the pre-training model. (b) The class tokens of the pre-trained model are embedded by the PCA method, and a void fraction determines their colors. (c) Plots of MAE results of the finetuning model and three baseline models with datasets of H<sub>2</sub> uptake, H<sub>2</sub> diffusivity, and band gap according to dataset size from 5,000 to 20,000. The baseline models are machine learning models that were respectively used to predict gas uptake, diffusivity, and band gap values.

#### 211 **Fine-tuning Results**

212 Figure 3(c) shows the fine-tuning results for predicting H<sub>2</sub> uptake (100 bar), H<sub>2</sub> diffusivity, and 213 band gap, which were obtained from GCMC, MD, and DFT simulations, respectively. While 1 214 million hMOFs were used for the pre-training step, a relatively smaller number of MOFs (i.e., 215 5,000 to 20,000) were used for training during the fine-tuning stages. The performance of the finetuning is compared with the three baseline models (i.e., the energy histogram,<sup>17</sup> descriptor-based 216 ML model,<sup>18</sup> and CGCNN<sup>19,21</sup>) as these have shown high performance in predicting gas uptake, 217 218 diffusivity, and band gap, respectively. And from these comparisons, it can be seen that the 219 MOFTransformer outperforms all of these other models, demonstrating both its superior 220 performance as well as transferable capabilities. It is worth noting that the MatErials Graph Network (MEGNET)<sup>39</sup> outperforms the CGCNN in predicting the band gaps of MOFs<sup>40</sup>. The 221 222 MEGNET utilizes global state attributes such as system temperature as well as atomic and bond 223 attributes as inputs. However, graph network models like CGCNN and MEGNET may have 224 difficulty in effectively predicting properties that rely on global features such as gas uptake and 225 diffusivity for MOFs. This is due to the larger crystal system of MOFs, which is characterized by a larger number of atoms and defined by metal clusters and organic linkers as nodes and edges, 226 227 respectively. As a result, the MOFTransformer exhibits strong performance in universal transfer 228 learning for MOFs compared to graph network models. The ablation studies of the fine-tuning to 229 demonstrate the effect of the data size on the pre-training tasks are explained in the Supplementary 230 Note S2.

To demonstrate further transferability across different applications, the MOFTransformer was fine-tuned for various properties summarized in Table 1. Table 1 shows a performance comparison between our fine-tuned model and the machine-learning models used in other works. And it can 234 be seen that the MOFTransformer model has either similar or higher performance (i.e., higher R<sup>2</sup> 235 score or lower MAE) across all properties. In particular, it is worth noting the robustness of our 236 model across different gas types, even though the probe molecule used to generate energy grids 237 was CH<sub>4</sub>. The reason is that overall shape of energy grids is relatively insensitive to the type of 238 probe molecule which has little effect on our model to learn global features from energy-grid 239 embeddings. In addition, the MOFTransformer can accurately predict properties at ambient 240 condition, given that N<sub>2</sub>, O<sub>2</sub> uptake and diffusivity were calculated at 1 bar and 298 K. Moreover, 241 our model extends well to showcase lower MAE than the machine learning model using revised autocorrelations (RAC)<sup>41</sup> with geometric features as descriptors to predict solvent removal stability 242 243 and thermal stability collected by text-mining. It is worth highlighting that our model showcases 244 high performance in predicting the experimental data like text-minded data as well as the 245 calculated properties. This result suggests that one can easily obtain high-performance structure-246 properties relationships by using our pre-trained model and fine-tuning it without needing to 247 develop a new model from scratch.



Figure 4. The schematics for attention score of atom-based embedding and energy-grid embedding
in IRMOF-1. (left) Repeating building blocks models in IRMOF-1 with atomic size proportional
to attention score. (right) Energy-grids that represent attention scores by color. The original form
of the IRMOF-1 is shown in the "base."

# 255 **Discussion**

256 Apart from the universal transfer learning, feature importance and its interpretation can lead to 257 a better understanding of the relationship between the MOF structures and their properties. Given 258 that attention scores measure how much the model should pay attention to inputs when predicting 259 desired properties, attention layers of the Transformer were assigned high attention scores to input 260 features according to their importance. From the fine-tuning models that predict H<sub>2</sub> uptake, H<sub>2</sub> 261 diffusivity, and band gap, feature importance analysis was implemented for IRMOF-1, which is 262 one of the representative isoreticular MOFs. Figure 4 shows both the repeating building blocks 263 models, which represent the metal cluster and organic linker, of IRMOF-1 (representing local 264 features) and the  $6 \times 6 \times 6$  patches of energy-grids (representing global features). The sizes of atoms 265 in the repeating building block models are scaled according to the attention scores obtained by the 266 atom-based embeddings. And the colors of the patches are proportional to the attention scores 267 obtained from the energy-grid embeddings. As can be seen from Figure 4, the atom-based 268 embeddings are assigned with low attention scores (e.g. visualized by small atom sizes) when 269 predicting H<sub>2</sub> uptake and diffusivity. On the other hand, the energy-grid embeddings are assigned 270 with high attention scores, which is in accordance with the fact that H<sub>2</sub> uptake and diffusivity rely 271 more on the global features. Meanwhile, for the band gap prediction, there is a reversal in trend as 272 the atom-based graph embeddings have higher attention scores compared to energy-grid 273 embeddings as the band gap is more dependent on the local features. The additional feature 274 importance analysis for other properties (e.g. O<sub>2</sub> diffusivity and CO<sub>2</sub> Henry coefficient) were also 275 conducted (see Supplementary Figure S8). Note that the feature importance analysis via attention 276 scores is in line with previous findings and a chemist's intuition.

277 Beyond the case study of IRMOF-1, we implemented an in-depth analysis of feature importance 278 for the atom-based graph and the energy-grid embeddings for band gap and H<sub>2</sub> uptake, respectively. 279 Given that the band gap is defined by the difference between the conduction-band minimum (CBM) 280 and the valance-band maximum (VBM), one might think that the atoms that exhibit strong peaks 281 at the CBM and VBM play a critical role in determining its value. Interestingly, we identified that 282 the atoms with peaks at the CBM and VBM strongly correlate with the atoms having high attention 283 scores. Figure 5(a) shows the repeating building blocks models of IRMOF-1, 2, 3, and Ni-IRMOF-284 1 and their density of state (DOS) plots. IRMOF-2 and IRMOF-3 are variants of the IRMOF-1 285 structure with the BDC linker functionalized by -Br and  $-NH_2$ . For IRMOF-2 and IMROF-3, the 286 atoms that are part of the organic linkers (i.e., C, H, N, Br) have higher attention scores than those 287 from the metal clusters (i.e., Zn, O). Consistent with these results, the atoms of the organic linker 288 have peaks at the CBM and VBM compared to those of the metal clusters. Meanwhile, for the Ni-289 IRMOF-1 (which has Ni instead of Zn compared to the IRMOF-1), the atoms that belong to the 290 metal cluster have higher attention scores and stronger peaks at the CBM and VBM compared to 291 the organic linkers. These tendencies are consistent with other examples that were randomly 292 selected in the QMOF database (see Supplementary Figure S9). Apart from these, we confirmed 293 that the feature importance analysis could capture the underestimation of the band gap calculated 294 by the PBE functional (see Supplementary Note S3). Hence, these results demonstrate that the 295 fine-tuned model successfully learns the chemical features that are the more important when it 296 comes to the band gap predictions.

When it comes to the energy-grid embeddings, one could argue that the patches located near the metal atoms have an important role on determining the gas uptake  $^{42}$  Indeed, from the finetuned model to predict H<sub>2</sub> uptake, the 8 highest attention scores from the 6x6x6 energy-grid patches 300 of IRMOF-1 are located near the metal atoms as shown in Figure 5(b). The metal atoms can make 301 stronger bonding with adsorbates than other atoms such C, H, O, resulting in lower energy values 302 for energy-grid patches near the metal atoms. Based on these observations, one can infer that the 303 energy values of energy-grid patches can have an impact on determining attention score. Therefore, 304 we plotted the relationship between the energy values of energy-grid points and the attention scores 305 for each patch to further illustrate this relationship. The minimum energy values are normalized 306 by their corresponding structure (or unit cell), which is represented on the y-axis of Figure 5(c). 307 Figure 5(c) suggests that the energy-grid points with high attention scores tend to have relatively 308 low energy values, as seen in the patches near the metal atoms. It is essential to highlight the fact 309 that the scatter points within the high attention region (attention score > 0.008) exhibit a lower 310 difference of energy than 20 kJ/mol.



Figure 5. (a) Schematics of attention score for atom-based embedding, and density of state (DOS)
plots for IRMOF-1, 2, 3, and Ni-IRMOF-1. The atomic sizes of repeating building blocks model

314	are proportional to the attention score. E means the energy, and E <sub>f</sub> indicates the Fermi level.
315	Positive and negative values of DOS indicate spin-up and spin-down channels, respectively. (b)
316	Schematic of high attention score patches of energy-grid embedding for IRMOF-1. (c) Scattered
317	plot for the difference of minimum energy between patch and unit cell according to energy-grid
318	embedding. $E_{p,min}$ refers the minimum energy of the patch, and $E_{u,min}$ refers to the minimum energy
319	of the unit cell. The red line ( $x = 0.008$ ) distinguishes between high and low attention regions.

# 321 Conclusions

322 For the first time, we introduced a multi-modal pre-trained Transformer to capture both local 323 and global features of MOFs. The model facilitates capturing the chemistry of metal nodes and 324 organic linkers from the CGCNN and the information on geometric and topological features such 325 as pore volume and topology from the energy grids. By fine-tuning the MOFTransformer model, our model outperforms all of the other state-of-the-art machine learning model across various 326 327 different properties, showing its universal transferability as well as superior performance. 328 Moreover, the model can provide insights by analyzing the feature importance from attention 329 scores obtained from attention layers of the fine-tuned model. We believe that this model can be 330 used as a bedrock model for other MOF researchers who wish to start their machine learning work 331 and, as such, can help accelerate materials discovery and research within the field of porous 332 materials.

### 334 Methods

#### 335 **Construction of hMOFs**

The hMOFs used to train our MOFTransformer were constructed using PORMAKE,<sup>11</sup> a Python library that can generate MOFs by combining building blocks with different topologies. These building blocks and the topologies were obtained from ToBaCCo,<sup>43</sup> CoREMOF (with all of the solvents removed),<sup>44</sup> and RCSR database.<sup>45</sup> Altogether, 1 million and 20,000 hMOFs were generated for the pre-training, and fine-tuning dataset, respectively, and the details of building hMOFs are explained in Supplementary Note S4. All of the generated structures were geometrically optimized using the LAMMPS<sup>46</sup> package with the UFF force field.<sup>36</sup>

## 343 Computational details for molecular simulation

344 For the fine-tuning dataset, H<sub>2</sub> uptake and diffusivity (or diffusion coefficient) were selected to 345 represent adsorption and diffusive properties. H<sub>2</sub> was selected to enable facile calculation while 346 being different from the guest molecule (i.e., methane) used for the energy grid construction. The calculations were conducted using the RASPA package.<sup>47</sup> For the H<sub>2</sub> molecule, a united atom 347 348 model was adopted. Also, the pseudo-Feynma-Hibbs model was used to express the H<sub>2</sub> behavior 349 at low temperature, which leads to fitting the Lenard-Jones (LJ) potentials to Feynman-Hibbs potential at T = 77 K.<sup>48,49</sup> Except for the H<sub>2</sub> molecules, the UFF force field was used with the 350 351 Lorentz-Berthelot mixing rule and a cutoff distance of 12.8 Å.

For H<sub>2</sub> uptake calculation, the GCMC calculation was performed at 100 bar and 77 K for 10k production cycles with 5k cycles used for the initialization. Diffusivity (or diffusion coefficient) was calculated at infinite dilution at 77 K using the MD simulation. Given that the intermolecular interactions of the H<sub>2</sub> atoms are ignored for the infinite dilution simulation, it may sometimes lead to the initial configurations of the H<sub>2</sub> atoms captured within the small pores of MOFs. The initial configurations were obtained from the MC simulation without infinite dilution for 5k cycles to prevent this from happening. Then, the MD simulations were conducted by NVT ensemble with 1 fs time step.<sup>18,50</sup> The simulations were run for 3 million cycles, with 1k cycles used for the initialization and 10k cycles for equilibration. The guest molecules' mean-squared displacement (MSD) was computed every 10k cycles, and the diffusion coefficient was obtained using the slope of the MSD through Einstein's relation.<sup>51</sup>

363

### Pre-training and Fine-tuning

In the pre-training step,  $AdamW^{52}$  optimizer with a learning rate of  $10^{-4}$  and weight decay of 10<sup>-2</sup> was used in all three tasks. The model was trained with a batch size of 1,024 during 100 epochs. The pre-training dataset was randomly split into training, validation, test sets with the number of 800,000, 126,611, 100,000, respectively. The learning rate was warmed up during the first 5 % of the total epoch and then was linearly decayed to zero for the remaining epochs.

369 For fine-tuning, the MOFTransformer is trained to predict the desired properties with the model 370 initialized by the converged weights from the pre-trained model. By adding a single dense layer to 371 the class token, all model weights are fine-tuned to predict desired properties of MOFs. Given that 372 the relatively small datasets are used during the fine-tuning step, the model was trained with a 373 batch size of 32 during 20 epochs whose optimizer and learning rates are the same as those of the 374 pre-training step. The fine-tuning dataset was randomly split into training, validation, test sets in a 375 ratio of 0.8:0.1:0.1. For scaling the target properties, the standardization method was adopted. 376 Training details of the three baseline models for comparison of the fine-tuning models are 377 explained in Supplementary Note S5.

# 379 **Conflicts of interest**

380 There are no conflicts to declare.

## **381** Author Contributions

382 Y.K and H.P contributed equally to this work. Y.K and H.P developed MOFTransformer and
383 wrote the manuscript with J.K. The manuscript was written through the contributions of all authors.
384 All authors have given approval for the final version of the manuscript.

# 385 Data availability

Data used in this work are available via Figshare (<u>10.6084/m9.figshare.21155506</u>). It provides the pre-trained model and the atom-based graph embeddings and the energy-grid embeddings used as inputs of the MOFTransformer for CoREMOF, QMOF database .as well as fine-tuning data. In addition, The UFF-optimized CIF files of hypothetical MOFs used in this work are available via Figshare (<u>10.6084/m9.figshare.21810147</u>)

## **Code availability**

The MOFTransformer library is available at <u>https://github.com/hspark1212/MOFTransformer</u>. Documents for the library is available at <u>https://hspark1212.github.io/MOFTransformer</u> which provides up-to-date documentation for pre-training, fine-tuning, and feature importance analysis with the MOFTransformer. For the sake of reproducibility, all results in this manuscript are obtained from a 1.0.1 version of MOFTransformer library, which is available at https://pypi.org/project/moftransformer/1.0.1.

# 398 Acknowledgements

399 H. P., Y. K., and J. K. acknowledge funding from National Research Foundation of Korea (NRF) 400 under Project Number 2021M3A7C208974513. This work was supported by the National 401 Supercomputing Center with supercomputing resources including technical support (KSC-2021-402 CRE-0460). BS is supported by the PrISMa Project, which is funded through the ACT programme 403 (Accelerating CCS Technologies, Horizon2020 Project No 294766). Financial contributions made 404 from: BEIS together with extra funding from NERC and EPSRC, UK; RCN, Norway; SFOE, 405 Switzerland and US-DOE, USA, are gratefully acknowledged. Additional financial support from 406 TOTAL and Equinor, is also gratefully acknowledged.

# 408 **References**

- 409 1 Freund, R. *et al.* The current status of MOF and COF applications. *Angewandte Chemie*410 *International Edition* 60, 23975-24001 (2021).
- 411 2 Kumar, S. *et al.* Green synthesis of metal–organic frameworks: A state-of-the-art review
  412 of potential environmental and medical applications. *Coordination Chemistry Reviews*413 420, 213407 (2020).
- 414 3 Qian, Q. *et al.* MOF-based membranes for gas separations. *Chemical reviews* **120**, 8161-415 8266 (2020).
- 416 4 Lee, J. *et al.* Metal–organic framework materials as catalysts. *Chemical Society Reviews*417 38, 1450-1459 (2009).
- 418 5 Deng, H. *et al.* Large-pore apertures in a series of metal-organic frameworks. *science*419 336, 1018-1023 (2012).
- 420 6 Ding, M., Cai, X. & Jiang, H.-L. Improving MOF stability: approaches and applications.
  421 *Chemical Science* 10, 10209-10230 (2019).
- Wang, C., Liu, D. & Lin, W. Metal–organic frameworks as a tunable platform for
  designing functional molecular materials. *Journal of the American Chemical Society* 135,
  13222-13234 (2013).
- 425 8 Colón, Y. J. & Snurr, R. Q. High-throughput computational screening of metal–organic
  426 frameworks. *Chemical Society Reviews* 43, 5735-5749 (2014).
- Boyd, P. G. *et al.* Data-driven design of metal–organic frameworks for wet flue gas CO2
  capture. *Nature* 576, 253-256 (2019).
- Daglar, H. & Keskin, S. Recent advances, opportunities, and challenges in highthroughput computational screening of MOFs for gas separations. *Coordination Chemistry Reviews* 422, 213470 (2020).
- Lee, S. *et al.* Computational screening of trillions of metal–organic frameworks for highperformance methane storage. *ACS Applied Materials & Interfaces* 13, 23647-23654
  (2021).
- Altintas, C., Altundal, O. F., Keskin, S. & Yildirim, R. Machine learning meets with
  metal organic frameworks for gas storage and separation. *Journal of Chemical Information and Modeling* 61, 2131-2146 (2021).
- 438 13 Chong, S., Lee, S., Kim, B. & Kim, J. Applications of machine learning in metal-organic
  439 frameworks. *Coordination Chemistry Reviews* 423, 213487 (2020).
- 440 14 Ahmed, A. & Siegel, D. J. Predicting hydrogen storage in MOFs via machine learning.
  441 *Patterns* 2, 100291 (2021).
- Simon, C. M. *et al.* The materials genome in action: identifying the performance limits for methane storage. *Energy & Environmental Science* 8, 1190-1199 (2015).
- 44416Lim, Y. & Kim, J. Application of transfer learning to predict diffusion properties in445metal-organic frameworks. *Molecular Systems Design & Engineering* (2022).
- Bucior, B. J. *et al.* Energy-based descriptors to rapidly predict hydrogen storage in metal–
  organic frameworks. *Molecular Systems Design & Engineering* 4, 162-174 (2019).
- 448 18 Orhan, I. B., Daglar, H., Keskin, S., Le, T. C. & Babarao, R. Prediction of O2/N2
  449 Selectivity in Metal–Organic Frameworks via High-Throughput Computational
  450 Screening and Machine Learning. ACS Applied Materials & Interfaces 14, 736-749
  451 (2021).

452	19	Rosen, A. S. et al. Machine learning the quantum-chemical properties of metal-organic
453		frameworks for accelerated materials discovery. <i>Matter</i> <b>4</b> , 1578-1597 (2021).
454	20	Ma, R., Colon, Y. J. & Luo, T. Transfer learning study of gas adsorption in metal-
455		organic frameworks. ACS applied materials & interfaces 12, 34041-34048 (2020).
456	21	Xie, T. & Grossman, J. C. Crystal graph convolutional neural networks for an accurate
457		and interpretable prediction of material properties. <i>Physical review letters</i> <b>120</b> , 145301
458		(2018).
459	22	Moosavi, S. M. et al. Understanding the diversity of the metal-organic framework
460		ecosystem. Nature communications 11, 1-10 (2020).
461	23	Nandy, A. et al. MOFSimplify, machine learning models with extracted stability data of
462		three thousand metal-organic frameworks. Scientific Data 9, 1-11 (2022).
463	24	Yao, Z. et al. Inverse design of nanoporous crystalline reticular materials with deep
464		generative models. Nature Machine Intelligence 3, 76-86 (2021).
465	25	Lim, Y., Park, J., Lee, S. & Kim, J. Finely tuned inverse design of metal-organic
466		frameworks with user-desired Xe/Kr selectivity. Journal of Materials Chemistry A 9,
467		21175-21183 (2021).
468	26	Willems, T. F., Rycroft, C. H., Kazi, M., Meza, J. C. & Haranczyk, M. Algorithms and
469		tools for high-throughput geometry-based analysis of crystalline porous materials.
470		Microporous and Mesoporous Materials 149, 134-141 (2012).
471	27	Vaswani, A. et al. Attention is all you need. Advances in neural information processing
472		systems <b>30</b> (2017).
473	28	Devlin, J., Chang, MW., Lee, K. & Toutanova, K. Bert: Pre-training of deep
474		bidirectional transformers for language understanding. arXiv preprint arXiv:1810.04805
475		(2018).
476	29	Dosovitskiy, A. et al. An image is worth 16x16 words: Transformers for image
477		recognition at scale. arXiv preprint arXiv:2010.11929 (2020).
478	30	Hu, R. & Singh, A. in Proceedings of the IEEE/CVF International Conference on
479		Computer Vision. 1439-1449.
480	31	Zhou, L. et al. in Proceedings of the AAAI Conference on Artificial Intelligence. 13041-
481		13049.
482	32	Li, L. H., Yatskar, M., Yin, D., Hsieh, CJ. & Chang, KW. Visualbert: A simple and
483		performant baseline for vision and language. <i>arXiv preprint arXiv:1908.03557</i> (2019).
484	33	Kim, W., Son, B. & Kim, I. in International Conference on Machine Learning. 5583-
485		5594 (PMLR).
486	34	Cao, Z., Magar, R., Wang, Y. & Farimani, A. B. MOFormer: Self-Supervised
487		Transformer model for Metal-Organic Framework Property Prediction. arXiv preprint
488		arXiv:2210.14188 (2022).
489	35	Chen, P., Jiao, R., Liu, J., Liu, Y. & Lu, Y. Interpretable Graph Transformer Network for
490		Predicting Adsorption Isotherms of Metal–Organic Frameworks. Journal of Chemical
491		Information and Modeling <b>62</b> , 5446-5456 (2022).
492	36	Rappé, A. K., Casewit, C. J., Colwell, K., Goddard III, W. A. & Skiff, W. M. UFF, a full
493		periodic table force field for molecular mechanics and molecular dynamics simulations.
494		Journal of the American chemical society 114, 10024-10035 (1992).
495	37	Martin, M. G. & Siepmann, J. I. Transferable potentials for phase equilibria. 1. United-
496		atom description of n-alkanes. The Journal of Physical Chemistry B 102, 2569-2577
497		(1998).

498	38	Bucior, B. J. <i>et al.</i> Identification schemes for metal–organic frameworks to enable rapid
499	20	search and cheminformatics analysis. Crystal Growth & Design 19, 6682-6697 (2019).
500	39	Chen, C., Ye, W., Zuo, Y., Zheng, C. & Ong, S. P. Graph networks as a universal
501		machine learning framework for molecules and crystals. Chemistry of Materials 31,
502	40	3304-3572 (2019).
503	40	Nandy, A., Duan, C. & Kulik, H. J. Using Machine Learning and Data Mining to
504		Leverage Community Knowledge for the Engineering of Stable Metal–Organic
505	4.1	Frameworks. Journal of the American Chemical Society 143, 1/535-1/54/ (2021).
506	41	Janet, J. P. & Kulik, H. J. Resolving transition metal chemical space: Feature selection
507		for machine learning and structure–property relationships. The Journal of Physical
508	40	Chemistry A 121, 8939-8954 (2017).
509	42	Koizumi, K., Nobusada, K. & Boero, M. Hydrogen storage mechanism and diffusion in
510	12	metal-organic frameworks. Physical Chemistry Chemical Physics 21, 756-764 (2019).
511	43	Colon, Y. J., Gomez-Gualdron, D. A. & Snurr, R. Q. Topologically guided, automated
512		construction of metal-organic frameworks and their evaluation for energy-related
513		applications. Crystal Growth & Design 17, 5801-5810 (2017).
514	44	Chung, Y. G. <i>et al.</i> Advances, updates, and analytics for the computation-ready,
515		experimental metal-organic framework database: CoRE MOF 2019. Journal of Chemical
516	4.5	& Engineering Data 64, 5985-5998 (2019).
517	45	O'Keeffe, M., Peskov, M. A., Ramsden, S. J. & Yaghi, O. M. The reticular chemistry
518		structure resource (RCSR) database of, and symbols for, crystal nets. Accounts of
519	16	<i>chemical research</i> <b>41</b> , 1/82-1/89 (2008).
520	46	Plimpton, S. Fast parallel algorithms for short-range molecular dynamics. <i>Journal of</i>
521	47	computational physics 117, 1-19 (1995).
522	47	Dubbeldam, D., Calero, S., Ellis, D. E. & Snurr, R. Q. RASPA: molecular simulation
523		software for adsorption and diffusion in flexible nanoporous materials. <i>Molecular</i>
524	40	Simulation 42, 81-101 (2016).
525	48	Feynman, R. P., Hibbs, A. R. & Styer, D. F. Quantum mechanics and path integrals.
526	40	(Courier Corporation, 2010).
527	49	Fischer, M., Hoffmann, F. & Fröba, M. Preferred hydrogen adsorption sites in various
528	50	MOFs—a comparative computational study. <i>ChemPhysChem</i> 10, 2647-2657 (2009).
529	50	Daglar, H., Erucar, I. & Keskin, S. Exploring the performance limits of MOF/polymer
530		MMMs for O2/N2 separation using computational screening. <i>Journal of Membrane</i>
531		Science 618, 118555 (2021).
532	51	Ewald, P. P. Die Berechnung optischer und elektrostatischer Gitterpotentiale. Annalen
533	50	der physik <b>369</b> , 253-287 (1921).
534	52	Loshchilov, I. & Hutter, F. Decoupled weight decay regularization. <i>arXiv preprint</i>
535		arXiv:1/11.05101 (2017).

538	<b>Table 1.</b> A table of fine-tuning results with the publicly accessible databases of MOFs that include
539	the properties calculated by GCMC, MD, and even text-mining data. The results of the machine

540 learning models used in the paper on the databases are summarized to compare the performance.

Property	MOFTransformer	Original paper	Number of data	Remarks	Ref
N <sub>2</sub> uptake	R2:0.78	R2:0.71	5,286	CoREMOF	18
O <sub>2</sub> uptake	R2:0.83	R2:0.74	5,286	CoREMOF	18
N <sub>2</sub> diffusivity	R2:0.77	R2:0.76	5,286	CoREMOF	18
O <sub>2</sub> diffusivity	R2:0.78	R2:0.74	5,286	CoREMOF	18
CO <sub>2</sub> henry coefficient	MAE : 0.30	MAE : 0.42	8,183	CoREMOF	22
Solvent removal stability classification	ACC : 0.76	ACC : 0.76	2,148	Text-mining data	40
Thermal stability regression	R2 : 0.44 (MAE : 45°C)	R2 : 0.46 (MAE : 44°C)	3,098	Text-mining data	40