

1 **The Impact of the Different Geometrical**
2 **Restrictions on the Nonadiabatic**
3 **Photoisomerization of Biliverdin**
4 **Chromophore**

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1 **Abstract:** The photoisomerization mechanism of the chromophore of bacterial
2 biliverdin (BV) phytochromes is explored with the nonadiabatic dynamics simulation
3 by using the on-the-fly trajectory surface-hopping method at the semi-empirical
4 OM2/MRCI level. Particularly, the current study focuses on the influence of the
5 geometrical constrains on the nonadiabatic photoisomerization dynamics of the BV
6 chromophore. Here a rather simplified approach is employed in the nonadiabatic
7 dynamics to capture the features of geometrical constrains, which adds the mechanical
8 restriction on the specific moieties of the BV chromophore. This simplified method
9 provides a rather quick approach to examine the influence of the geometrical
10 restrictions on the photoisomerization. As expected, different constrains bring the
11 distinctive influences on the photoisomerization mechanism of the BV chromophore,
12 giving either strong or minor modification of both involved reaction channels and
13 excited-state lifetimes after the constrains are added in different ring moieties. These
14 observations not only contribute to the primary understanding of the role of the spatial
15 restriction caused by biological environments in photoinduced dynamics of the BV
16 chromophore, but also provide useful ideas for the artificial regulation of the
17 photoisomerization reaction channels of phytochrome proteins.

18

1 **1. Introduction**

2 Phytochromes are photosensitive proteins, which are widely found in plants, bacteria
3 and algae.¹ Plant phytochrome (PΦB) promotes the seed germination and the
4 chlorophyll synthesis in plants.² Phycocyanin (PCB) and biliverdin (BV) phytochromes
5 that are found in algae and bacteria, respectively, can synthesize biological pigments
6 and promote growth.³⁻⁵ In addition, BV was also used to design the near-infrared
7 fluorescent proteins in the research of diseases such as cancers.⁶⁻¹⁰ Owing to their
8 importance, phytochromes have been extensively studied over decades.^{9,11-18}

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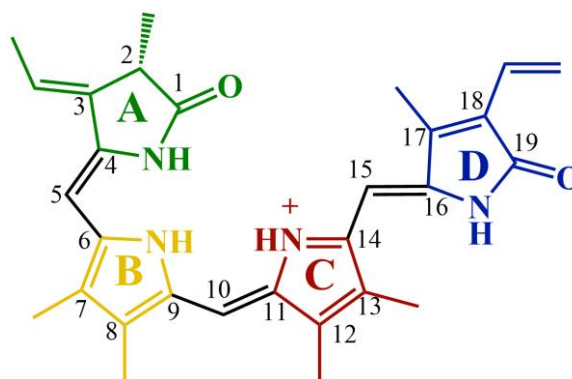


Figure 1. BV model in Pr (*ZsZsZa*) configuration

10

11 The biological functions of phytochrome are controlled by the switching between
12 two forms (Pr and Pfr).¹⁹ The physiologically inactive Pr form can absorb red light, and
13 converts to the active Pfr form. In the reversed process, the Pfr form can return to the
14 Pr form by absorbing far-red light. The switching between them drives the structural
15 rearrangement in the phytochrome proteins to realize important biological functions.

16 The central chromophore of phytochrome is a methylene-bridged linear tetrapyrrole
17 compound with several twisted carbon-carbon single and double bonds, resulting in the

1 existence of many isomers. In order to label these configurations, the bridged carbon-
2 carbon double and single bonds are defined as *Z/E* and *syn/anti(s/a)* isomerism,
3 respectively. Some studies have determined that the Pr of PΦB is in the *ZaZsZa* form,²⁰
4 while the Pr of PCB and BV has the *ZsZsZa* form (Figure 1).^{21,22} In contrast, the Pfr
5 configuration is still not fully clarified, which may be dependent on Pr → Pfr
6 photoreaction processes.²³ Therefore, the study of Pr → Pfr reaction not only deepens
7 the understanding of the photoisomerization process, but also helps to determine the
8 Pfr conformation.

9 Considerable efforts were made to understand the Pr → Pfr photoisomerization
10 mechanism. The photoisomerization process of BV and PΦB in the protein
11 environments is believed to be around the C₁₅C₁₆ double bond.²⁴⁻²⁹ Differently, Ulijasz
12 et al. proposed that the twist of C₄C₅ double bond is the key to the isomerization of
13 PCB.²³ At the same time, some studies also pointed out that the rotation of the C₁₀C₁₁
14 double bond was noticeable in the BV and PΦB isomerization,³⁰⁻³² at least in the gas
15 phase. In addition, the hydrogen-bond network patterns greatly affect the
16 photoisomerization channels of central chromophores, when different deprotonation
17 status exist in these pyrrole rings.^{33,34} These works demonstrated that the
18 photoisomerization of phytochrome chromophores may be adjusted by the distinctive
19 confinement of chromophores in the vacuum and in the surrounding environment. In a
20 vacuum, the chromophore is a "free" molecule and the rotation of any carbon-carbon
21 bond is not affected by external factors. In living organisms, nonetheless, chromophores
22 can be spatially constrained by their surrounding residuals, placing additional

1 restrictions on their photoisomerization process. For example, one common idea is that
2 Ring D in BV (Figure 1) and PΦB is not rigorously constrained due to the fact that it is
3 located in a relatively loose protein cavity.^{25,35} This leads to a high chance to realize the
4 rotation of Ring D. This view was also supported by the work of Burgie et al., who
5 showed that Ring A and Ring B are sandwiched in protein secondary domains and
6 covalently linked to the protein for PΦB.²⁵ In addition, the C₁₀ methyl bridge between
7 Ring B and Ring C is tightly wrapped by surrounding residues in BV, preventing the
8 photoisomerization at the C₁₀ position.²¹

9 Here, we are interested in several detailed aspects of the photoisomerization
10 mechanism of the BV chromophore. The BV chromophore attracts our attention due to
11 its unique properties.^{16,20,36} It is not only the chromophore of phytochrome in
12 *Deinococcus radiodurans*, but also the heme-metabolism intermediate of all aerobic
13 organisms.^{7,8,37} It can be engineered into monomeric infrared fluorescent proteins
14 (IFPs)^{38,39} that may be potentially used in disease diagnosis.^{6,9}

15 In the present work, we specifically focus on the possible influences of the restricted
16 motions in the BV chromophore. Particularly, we wish to provide a rather preliminary
17 view of the steric effects by surrounding environments on the photoisomerization
18 though a quick computational study, instead of giving a very comprehensive description
19 of the role of realistic environments in nonadiabatic dynamics. Therefore, the efficient
20 treatment of the spatial constraints becomes very essential. In principle, the dynamics
21 of molecules constrained by the biological environment can be simulated by the
22 quantum mechanical/molecular mechanics (QM/MM) method, which can include all

1 degrees of freedom of the protein environment and solution compounds explicitly.⁴⁰⁻⁴⁴
2 However, it requires rather large computational costs once performing the nonadiabatic
3 dynamics simulations. Instead, we wish to treat the geometrical restriction in the
4 nonadiabatic dynamics using a rather simplified approach with much less
5 computational cost. Previously, many efforts have been made to mimic the biological
6 environmental influence on the nonadiabatic dynamics in the simplified way. For
7 example, Warshel simulated the restriction of protein by constraining the movements
8 of terminal atoms in retinal molecules, rather than considering the motions of all
9 atoms.⁴⁵ Barbatti et al. imposed the geometrical constraints by increasing the nuclear
10 mass of the terminal hydrogen atom in the nonadiabatic dynamics simulation of the
11 protonated Schiff bases linked to proteins.⁴⁶ This idea was also successfully used to
12 study the environmental effects in the photochemistry of aminopyrimidines.⁴⁷⁻⁴⁹ The
13 employment of this “heavy-mass” approach shows many advantages: it can greatly
14 reduce the computational costs and is easy to implement in the nonadiabatic dynamics
15 simulations. Inspired by these works, we decide to take a similar idea in the trajectory
16 surface hopping (TSH) simulations. The geometrical restriction of the surrounding
17 environments is taken into account by using the very heavy atomic mass in the restricted
18 moieties of the BV chromophore. We hope to elucidate the role of the possible
19 surrounding environment on the photoinduced nonadiabatic processes, and further
20 deepen the understanding of the photoisomerization mechanism of the BV
21 chromophore in biological proteins.

22

1 **2. Computational Details**

2 **2.1 Electronic Structure Calculation**

3 The electronic structure calculations were performed with the semi-empirical
4 orthogonalized-model (OM2) method.⁵⁰ The excited-state wave function was described
5 by the multireference configuration interaction (MRCI) method within the
6 configuration interaction scheme based on the Graphical Unitary Group Approach
7 (GUGA-CI).^{51,52} The molecular orbitals were generated by using the restricted open-
8 shell Hartree-Fock (ROHF) approach. All electronic configurations were generated
9 from five reference configurations [the closed-shell, two single (HOMO-1 to LUMO
10 and HOMO to LUMO) and two double (HOMO-1 to LUMO and HOMO to LUMO)
11 excitations]. The active space (16, 12) was employed which distributes 16 electrons in
12 12 orbitals: six π orbitals, two n orbitals and four π^* orbitals. The state minima (S_0 _min
13 and S_1 _min) and minimum-energy S_0/S_1 conical intersection (CI) geometries were
14 optimized.⁵³ To get a direct view of the excited-state reaction pathways, we constructed
15 the potential energy (PE) profiles by linear interpolation of the internal coordinates
16 (LIICs) from the ground-state minimum to the minimum-energy CI structures. All
17 semi-empirical calculations were performed by using the MNDO2020 package.⁵⁴

18

19 **2.2 Nonadiabatic Dynamics**

20 The photoinduced nonadiabatic dynamics were simulated by the on-the-fly TSH
21 simulations at the OM2/MRCI level. Previous works have demonstrated that this
22 approach provides an efficient and reasonable description on the excited-state

1 nonadiabatic dynamics or polyatomic systems with low computational cost.^{32,55} A set
2 of initial conditions (geometry and velocity) were generated using Winger sampling
3 method.⁵⁶ All trajectories start from the first excited state. The time steps for the
4 propagation of the nuclear and electron motions were 0.5 fs and 0.005 fs, respectively.
5 The trajectories propagated up to 2000 fs. The nuclear motion was integrated by the
6 velocity-Verlet method. The solution of the electron motion was carried out by the
7 unitary propagation. The hopping probability was calculated with Tully's fewest
8 switches algorithm.⁵⁷ All relevant energies, gradients and nonadiabatic couplings were
9 calculated in the manner of on-the-fly along the trajectory propagation. A practical way
10 proposed by Granucci et al. with the $\gamma = 0.1$ Hartree was employed to take the
11 decoherence correction into account.⁵⁸ When hops take place, the velocity rescaling is
12 performed according to the nonadiabatic coupling vector. For frustrated hops, the
13 velocity component along the nonadiabatic coupling vector was reversed. The interface
14 between the TSH dynamics module in the JADE package⁵⁹ and the electronic-structure
15 calculations OM2/MRCI in the MNDO package was employed for the nonadiabatic
16 dynamics simulations.

17

18 **2.3 Molecular model construction**

19 All calculations were simulated using a simplified model with the *ZsZsZa* structure
20 that was identified as the main isomer of the Pr form in BV photochrome.²¹ Here,
21 several side groups (thioether bonds, propionic acid carboxyl groups on Rings B and C)
22 should have minor contribution on the skeleton motion in the excited state dynamics,

1 we simply replaced them by the methyl group to reduce computational costs.

2 We wish to gain a deep understanding of the steric effect on the photoisomerization
3 dynamics of BV molecules. Here, the BV molecules possess four five-membered rings.
4 To get a full understanding of the impacts of different geometrical constraints, we
5 considered as many restriction situations as possible, and totally constructed nine
6 models, as shown in Table 1 and Figure 1. **Model 0** represents the original simplified
7 BV chromophore model. In **Models I-IV**, a single ring is constrained. In **Models V-**
8 **VII**, two adjacent rings were constrained, along with their connecting parts. We also
9 built **Model VIII**, in which two side rings (A and D in Figure 1) were constrained. On
10 this basis, we set the atomic masses in the restricted part of the BV molecule to
11 99999999 au and the initial velocity to 0 in the nonadiabatic dynamics simulation.

12 According to the simulation results (see discussions below), nine models were
13 divided into four groups with rather different decay features: Group 1 includes **Models**
14 **0, I, IV** and **VIII**; Group 2 includes **Models II** and **V**; Group 3 includes **Models III** and
15 **VII**; and Group 4 includes **Model VI**.

16
17 **Table 1.** All BV models with different geometrical constraints. The constrained parts
18 are labelled by different colors in Figure 1.

Models	Constrained part of the BV molecule	Group
0	-	1
I	Ring A (green part)	1
II	Ring B (yellow part)	2

III	Ring C (red part)	3
IV	Ring D (blue part)	1
V	Rings A and B (green part, C ₅ methyl bridge and yellow part)	2
VI	Rings B and C (yellow part, C ₁₀ methyl bridge and red part)	4
VII	Rings C and D (red part, C ₁₅ methyl bridge and blue part)	3
VIII	Rings A and D (green part, and blue part)	1

1

2 **3. Results and Discussion**

3 **3.1 CI Structures and Channels**

4 The Pr→Pfr photoisomerization mechanism is closely related to the ultrafast
5 nonadiabatic process governed by the conical intersections. To clarify the role of the
6 CIs in the nonadiabatic dynamics, we optimized the several CI geometries (Figure 2).
7 Table S1 and S2 (in supporting information (SI)) show the key internal coordinates at
8 S₀_min, S₁_min and CIs. These four important CIs display the below geometrical
9 features:

- 10 ▪ CI₁: C₉C₁₀C₁₁N_C dihedral angle rotates to -85.4° with the elongated C₁₀C₁₁ bond and
11 the shortened C₉C₁₀ bond;
- 12 ▪ CI₂: N_BC₉C₁₀C₁₁ dihedral angle rotates to -93.4° with the elongated C₉C₁₀ bond and
13 the shortened C₁₀C₁₁ bond;
- 14 ▪ CI₃: C₁₄C₁₅C₁₆N_D dihedral angle rotates to 95.0° with the elongated C₁₅C₁₆ bond and
15 the shortened C₁₄C₁₅ bond;
- 16 ▪ CI₄: N_AC₄C₅C₆ dihedral angle rotates to 89.6° with the elongated C₄C₅ bond and the

1 shortened C₅C₆ bond.

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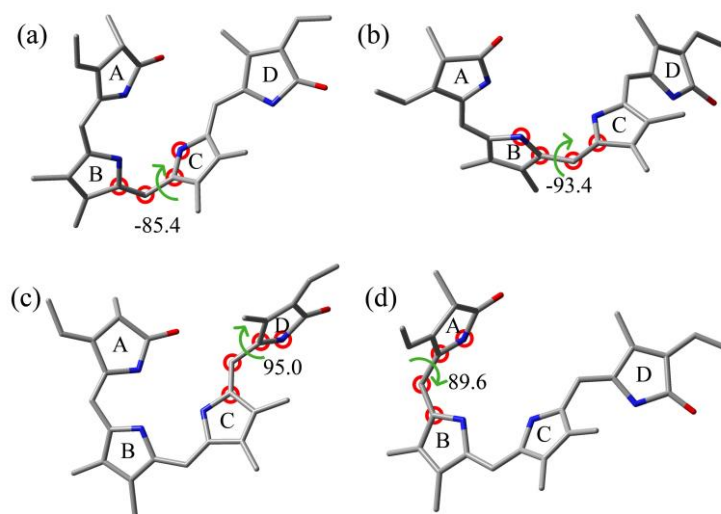


Figure 2. Four important S₀/S₁ minimum-energy CI structures and their major dihedral angles (marked by red circles) (a) CI₁; (b) CI₂; (c) CI₃; (d) CI₄.

3

4 To get a direct view of the excited-state reaction pathways, we constructed the PE
5 profiles from the ground-state minimum to the CI structures. In Figure 3a and 3b, the
6 excited-state pathways from S₀_min to both CI₁ and CI₂ are barrierless, and the former
7 one is steeper. Moreover, the energy of CI₁ is lower than that of CI₂. While the other
8 two channels towards CI₃ and CI₄ show some visible barriers (see Figure 3c and 3d),
9 and the CI₄ channel exists a slightly higher energy barrier.

10 It is necessary to point out that the torsional motions at the C₁₄C₁₅ and C₅C₆ bonds
11 also lead to the CIs, giving CI₅ and CI₆ (see Figure S1), respectively. However, as shown
12 in Figure S2, the reaction pathways towards them display very high barriers, preventing
13 the possibility of these two photoisomerization channels. Thus, we do not discuss them
14 here.

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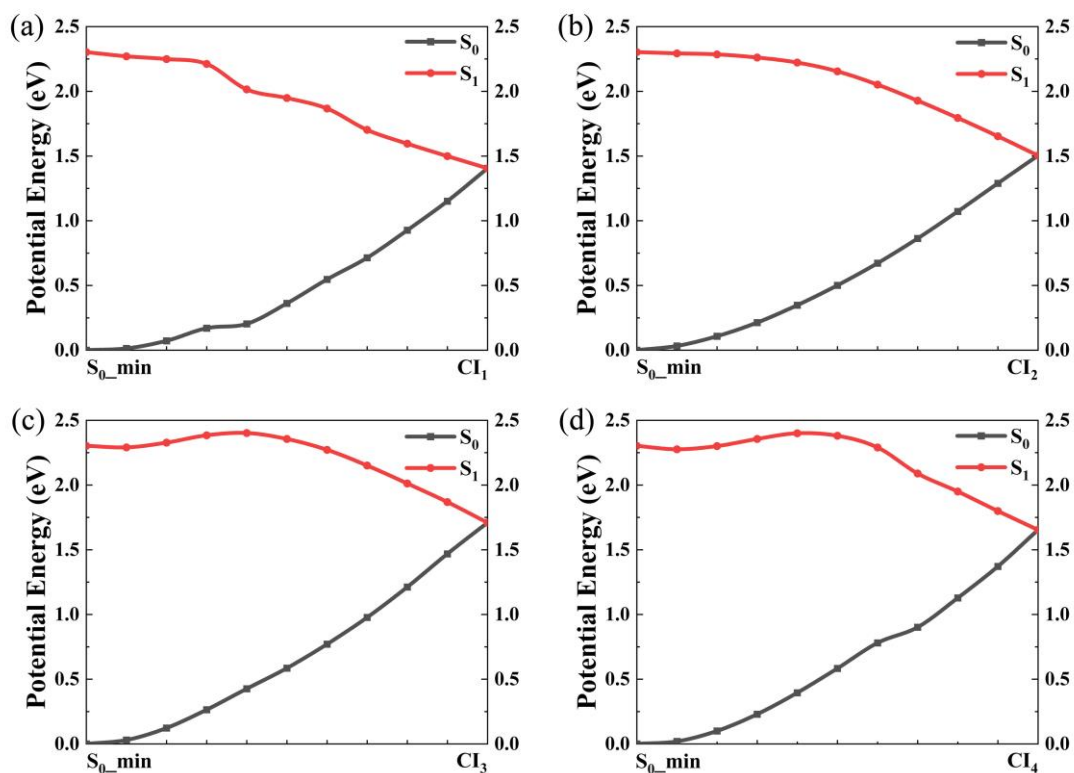


Figure 3. Potential energy curves along the linear interpolated pathway from S_0_min to the four conical intersections (a) CI_1 ; (b) CI_2 ; (c) CI_3 ; (d) CI_4 .

2

3.2 Photoisomerization Pathways

We performed the nonadiabatic dynamics simulation for all nine models mentioned above. After collecting all hop geometries and analyzing their correlations with optimized CIs, four CIs (CI_1 , CI_2 , CI_3 , and CI_4) were identified to play key roles in the nonadiabatic decay. By analyzing the hopping structures in TSH dynamics, we can divide all nine BV models into four groups according to their involved decay channels, as shown in Table 2.

10

1 **Table 2.** The branching ratio of different reaction channels in each constrained BV
 2 model, obtained from the TSH dynamics up to 2 ps.

Group	Model	CI ₁	CI ₂	CI ₃	CI ₄	No Hop
1	0	94.1%	5.9%	0.0%	0.0%	0.0%
	I (Ring A)	94.3%	1.0%	3.6%	0.0%	1.1%
	IV (Ring D)	95.7%	3.8%	0.0%	0.0%	0.5%
	VIII (Rings A and D)	87.0%	3.5%	0.0%	0.0%	9.5%
2	II (Ring B)	67.0%	1.1%	21.9%	0.0%	10.0%
	V (Rings A and B)	63.1%	1.4%	30.3%	0.0%	5.2%
3	III (Ring C)	45.5%	14.1%	4.3%	1.0%	35.1%
	VII (Rings C and D)	31.4%	13.3%	0.0%	6.4%	48.9%
4	VI (Rings B and C)	0.0%	0.0%	42.0%	8.0%	50.0%

3

4 The first group includes **Models 0, I, IV** and **VIII**. Among them, **Model 0** represents
 5 the free BV model. **Models I, II** and **VIII** refer to the situations in which the constraints
 6 are added in two terminal rings, namely Rings A, D and both, respectively. As shown
 7 in Table 2, Figure 4a and Figure S4, the CI₁ controlled pathway is the predominant
 8 decay channel in these four models. Following this channel, the trajectories experience
 9 the significant changing of the C₉C₁₀C₁₁N_C angle up to ~85.4°, as shown in Figure 2.
 10 However, the other CIs only make very small contributions. For instance, the CI₂ plays
 11 a little role here, and the CI₃ channel only appears in the TSH dynamics of **Model I**.
 12 This result can be well understood by the PE curves. As we discussed previously, there
 13 is no barrier existing in the first excited-state decay pathways from S_{0_min} to CI₁ and
 14 CI₂, and the former PE surface is steeper. At the same time, the other two channels
 15 towards CI₃ and CI₄ display visible energy barriers along the excited-state pathways.
 16 As a consequence, in **Model 0**, the channel via CI₁ becomes dominant while only a

1 small number of trajectories decay via CI_2 here. Since the motions of Rings A and D
2 are constrained in **Models I, IV and VIII**, we expect that the major torsional around the
3 carbon-carbon double bond between Rings B and C should not be largely hindered.
4 This explains why their nonadiabatic decay channels are similar to that of **Model 0**.
5 Overall, the CI_1 pathway is the dominant channel in the nonadiabatic dynamics of
6 **Models 0, I, IV and VIII** in Group 1, in which the photoisomerization mainly takes
7 place at the $C_{10}C_{11}$ bond.

8 The second group includes **Model II** and **Model V**, with Ring B and Rings A and B
9 are constrained, respectively. For these two models, most of trajectories decay via CI_1
10 while the secondary channel goes to the CI_3 (see Table 2, Figure 4b and Figure S5).
11 Compared to the models in Group 1, the contributions of both CI_1 and CI_2 become
12 decreased, while the role of CI_3 starts to be visible. The change of the branching ratio
13 to different channels can be explained as follows. Once Ring A and Ring B are
14 immobilized, the torsional motion along the C_9C_{10} bond becomes limited, resulting in
15 the reducing of the CI_2 channel. In addition, the analysis of trajectory propagation
16 clarifies that the motion of Ring B is necessary in the $C_{10}C_{11}$ isomerization channel, as
17 shown in Figure S3 in SI for details. Therefore, the constrains on Ring B leads to the
18 decreasing of the CI_1 channel. Due to the reducing of both CI_1 and CI_2 channels,
19 trajectories should move by following either CI_3 or CI_4 channel, and both show barriers
20 on the excited-state reaction pathway. Although the PE barrier in the CI_3 controlled
21 channel is only 0.06 eV lower than that of the CI_4 dominated channel, the difference in
22 the proportion of the channel is obvious. This feature may be attributed to the reasons

1 below. The nonadiabatic processes of the channels via CI_3 and CI_4 show distinctions in
2 ring motions. The decay channel via CI_3 is mainly characterized by the motion of the
3 Ring D. In the CI_4 channel, Ring A must experience the outward motion away from the
4 other three rings first, and then rotate. Therefore, the CI_3 channel is easily achieved due
5 to much less geometrical rearrangement. As a short summary, the CI_1 channel with the
6 photoisomerization site at the $C_{10}C_{11}$ bond is still dominant in the nonadiabatic
7 dynamics of Group 2. However, the branching ratio of this channel is reduced, with the
8 increasing of the CI_3 channel that displays the photoisomerization at the $C_{15}C_{16}$ bond.

9 The third group includes **Models III** and **VII**, with Ring C constrained and Rings C
10 and D constrained. As shown in Table 2, Figure 4c and Figure S6, the channel via CI_1
11 is still the most important one. Interestingly, we also see the secondary channels via CI_2 .
12 The dominant role of CI_1 and the secondary role of CI_2 are addressed in the previous
13 discussions. Compared to **Model 0**, the proportion of CI_2 (CI_1) channel has increased
14 (decreased) greatly. The analysis of trajectory evolution indicates that the torsion
15 around the $C_{10}C_{11}$ bond is strongly controlled by the Ring C (see Figure S3). As a
16 consequence, the restriction of Ring C leads to the decreasing of the CI_1 channel. The
17 reducing of the CI_1 channel certainly improves the contributions of other channels. As
18 the results, more trajectories follow the CI_2 because it is also barrierless. In addition, a
19 few of trajectories choose the CI_3 channel with the smaller potential barrier. Since
20 **Model VII** confines Rings C and D, the CI_3 channel of the $C_{15}C_{16}$ rotation between
21 Rings C and D is closed. Therefore, there is no CI_3 channel in **Model VIII**, resulting in
22 an increase in the proportion of CI_4 channels. Overall, the CI_1 channel with the

1 photoisomerization at the C₁₀C₁₁ bond is still dominant in Group 3, while its ratio is
2 largely reduced. The secondary channel is governed by the CI₂ that shows the
3 isomerization site at C₉C₁₀ bond.

4 The fourth group includes only **Model VI**, in which both two central Rings, B and C
5 are constrained. In this model, the vast majority of the trajectories decay through CI₃
6 and small number of trajectories decay via CI₄, see Table 2, Figure 4d and Figure S7.
7 Since both Rings B and C are restricted, the dihedral angle between them almost cannot
8 rotate, leading to the vanishing of the CI₁ and CI₂ channels. Due to the less geometrical
9 rearrangement discussed above, the CI₃ channel is dominant one in Group 4. In this
10 case, the photoisomerization mainly takes place at the C₁₅C₁₆ site. Nevertheless, this
11 model shows the similar geometrical constraints at the BV chromophore in protein
12 environments.

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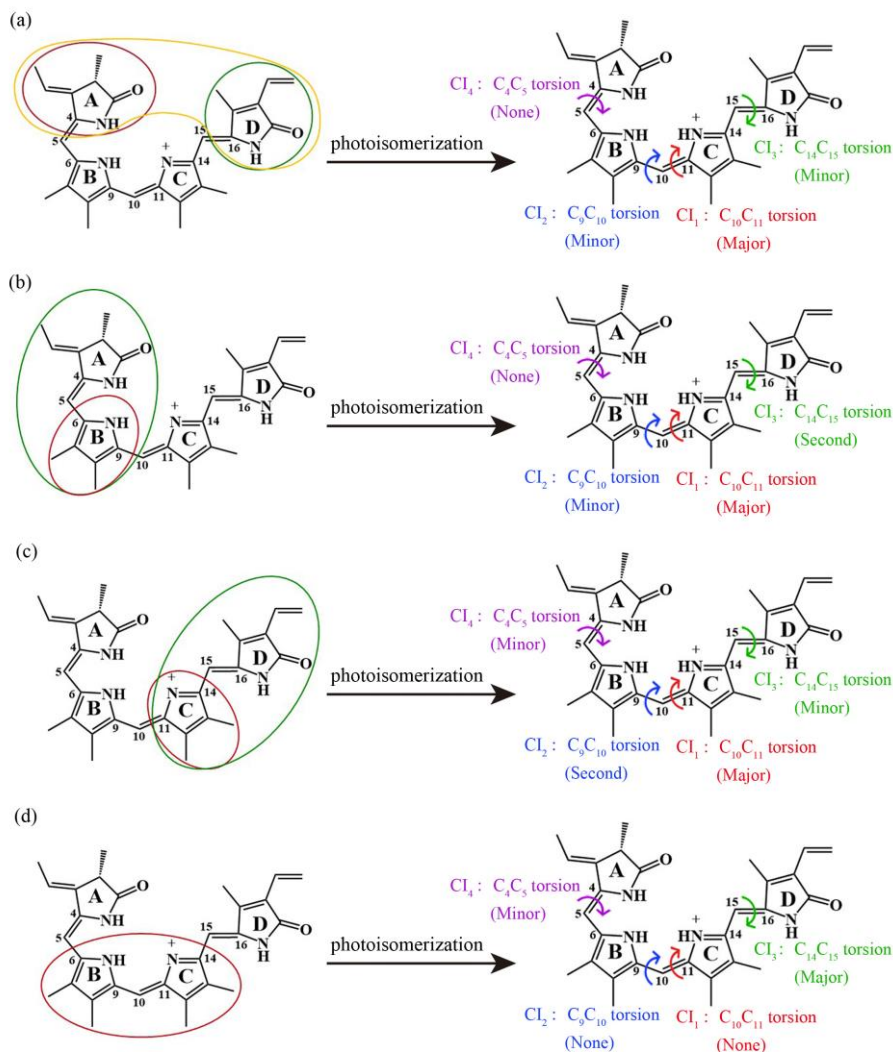


Figure 4. The branching ratios of different CI channels in each group of restricted models. The red, green and yellow circles in the left diagram indicate the constrained moieties. (a) Group 1 includes **Models 0, I, IV and VIII** (**Model 0** represents the free BV model); (b) Group 2 includes **Models II and V**; (c) Group 3 includes **Models III and VII**; (d) Group 4 includes **Model VI**.

1

2 3.3 Population

3 The above discussions clearly point out that different decay channels are responsible
 4 for the nonadiabatic dynamics of nine constrained BV models. The trajectories follow

1 distinct reaction channels with different PE profiles, we expect that the evolution of
2 electronic populations of these models should also be different. In principle, the models
3 belonging to the same group should display similar population dynamics, since their
4 underlying channels are rather similar.

5 Figure 5 shows the evolution of a few key dihedral angles with time being, from
6 starting conditions to hopping events. Figure 6 shows their population dynamics.
7 Clearly, the photoisomerization rate is strongly controlled by the dominant CI channel
8 that is significantly modified by different geometric constraints.

9 Group 1 includes **Models 0, I, IV and VIII**. In Group 1, the CI₁ is the most dominant
10 decay channel. As shown in Figure S3, the motion of Ring C is important in the CI₁
11 channel, while the motion of the Ring B is also involved necessary. The Rings B and C
12 are completely unconstrained in structure, and have no effect on the torsion of the CI₁
13 channel. Therefore, the nonadiabatic dynamics in this group display the fastest
14 population decay (see Figure 6a). For this group, most trajectories access the CI₁ around
15 400 fs (Figure 5 a–d). Only for **Model VIII**, we noticed that 9.5% of trajectories stay
16 on the first excited state at the end of simulation. As a consequence, more than 50% of
17 trajectories jump back to the S₀ state within 500 fs, and most of the trajectories return
18 to the ground state within 2000 fs.

19 Compared with Group 1, the branching ratio towards to the CI₃ channel increased
20 significantly in all models of Group 2, while the CI₁ channel is still the most important
21 one. As depicted in Figure S3, the motion of Ring B plays the visible role in the CI₁
22 channel, and thus the employment of the constrain in Ring B slows down this channel.

1 This is confirmed by Figure 5 e and f, which show that the many nonadiabatic
2 transitions at CI_1 take place around 400 fs, while some hops happen at the later time
3 $\sim 400\text{--}800$ fs. In addition, about 10.0% and 5.2% of the trajectories do not hop back to
4 the ground state for both **Model II** and **V**, respectively. As the result, a slightly longer
5 decay in the population dynamics appears for all models in Group 2. Figure 6b reflects
6 that more than 50% of the trajectories decay to the S_0 state within 700 fs, and most of
7 the trajectories back to the S_0 state at the end of simulation.

8 In Group 3 (**Models III** and **VII**), the CI_1 channels is still the most dominant one,
9 although its contribution is further decreased in all models. The Ring C is constrained
10 and greatly affects the molecular rotation through the CI_1 channel, see Figure S3.
11 Therefore, the dynamics is quite different with respect to those of the first two groups.
12 The photoisomerization via the CI_1 channel becomes much slower (Figure 5) and many
13 hops take place around 1600 fs (see Figure 5 g and h). And in **Models III** and **VII**, 35.1%
14 and 48.9% of the trajectories do not hop up to 2 ps. Overall, the population decay
15 becomes much slower.

16 Completely different from the above three groups, CI_1 and CI_2 channels do not exist
17 in the model of Group 4. Here, all trajectories decay visit CI_3 or CI_4 . The transitions
18 take place mainly through the CI_3 channel, in which the $C_{15}C_{16}$ double bond rotates
19 more slowly (see Figure 5i). In this case, most hops take place around 1200-1600 fs,
20 and 50% of the trajectories do not jump back to the ground state up to 2 ps. As a
21 consequence, Group 4 shows the slowest decay dynamics in Figure 6d.

22

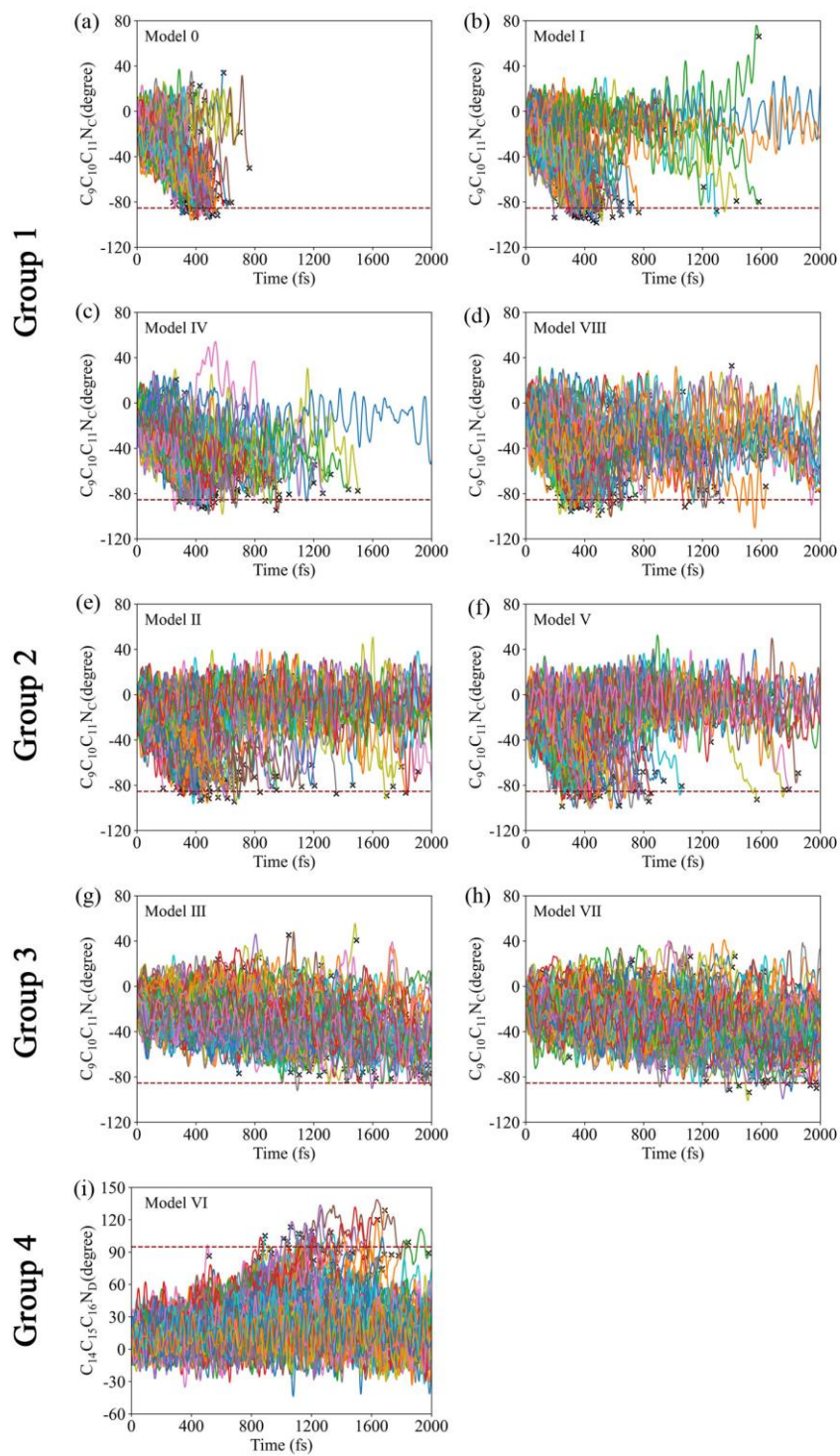


Figure 5. Time evolution of key dihedral angles ($C_9C_{10}C_{11}N_C$ for most models and $C_{14}C_{15}C_{16}N_D$ for Model VI). The red dotted line represents the dihedral angle at the dominant CI structure. The cross labels show the hopping events.

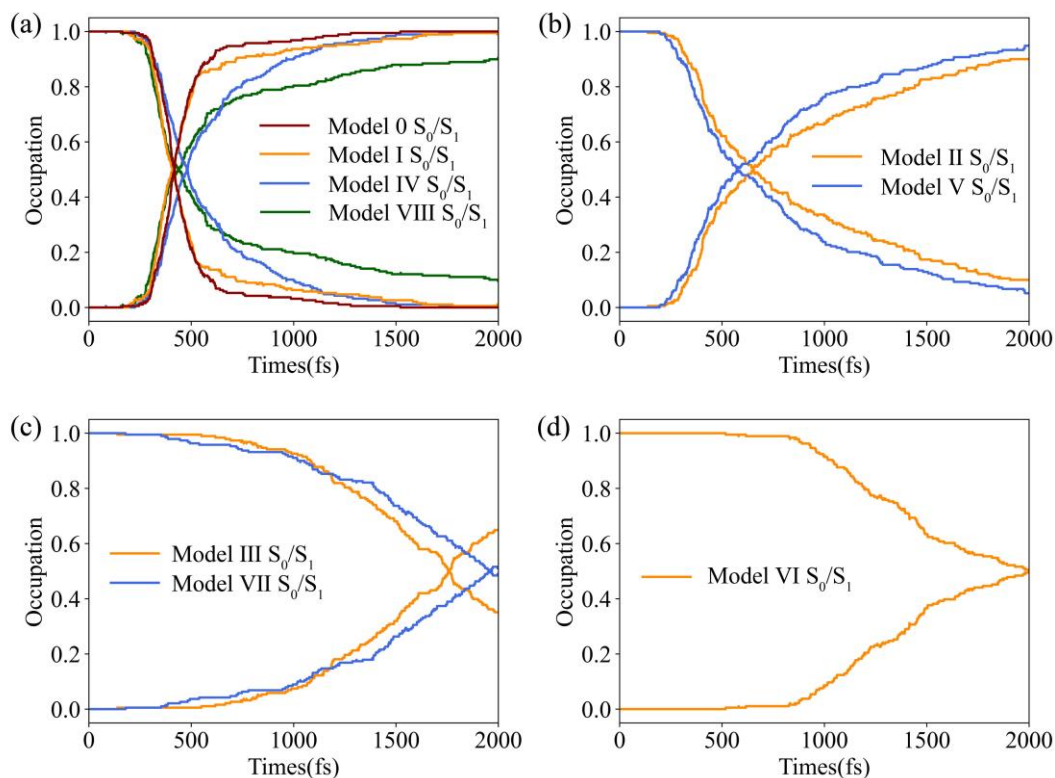


Figure 6. Time-dependent fractional occupations of the S_0 and S_1 electronic states of the nine models in the nonadiabatic dynamics starting from the S_1 state. (a) Group 1; (b) Group 2; (c) Group 3; (d) Group 4.

2

3 **4. Conclusion**

4 In this work, the photoisomerization processes of BV chromophore are investigated
 5 by employing the surface hopping nonadiabatic dynamics simulation at the OM2/MRCI
 6 level. Different geometrical constraints are considered, and the corresponding models
 7 give distinct decay channels.

8 The unconstrained BV chromophore (**Model 0**) and the constrained terminal rings
 9 (Ring A in **Model I**, Ring D in **IV**, and Rings AD in **VIII**, respectively) have similar

1 dynamics. The trajectories decay rapidly from S_1 state, with 50% of the trajectories
2 decaying within 500fs. Most trajectories choose to follow the CI_1 channel, characterized
3 by isomerization around $C_{10}C_{11}$ bond.

4 **Models II** and **V** constrain the motion of Ring B and the Rings A and B, leading to
5 the slight increasing of the excited-state lifetime with respect to **Models I, IV** and **VIII**.
6 Except the major channel via CI_1 , the secondary channel is governed by the CI_3 channel,
7 in which the photoisomerization takes place at the $C_{15}C_{16}$ bond.

8 **Models III** and **VII**, which are characterized by the restricted motion of Ring C and
9 Rings C and D, show the significantly longer excited-state decay times. Although the
10 CI_1 channel is still the primary channel, the ratio of the CI_2 channel with the
11 photoisomerization at the C_9C_{10} bond increases significantly.

12 **Model VI** with the two middle rings fixed show completely different dynamics
13 features, with 50% of the nondecay trajectories at the end of the simulation time 2000
14 fs. Such slow decay is attributed to the fact that **Model VI** decays mainly through the
15 CI_3 channel in which the $C_{15}C_{16}$ bond torsion is involved.

16 By simply mechanically restricting the motion of a specific moiety, we try to study
17 the spatial effects of surrounding environment on the photoisomerization dynamics of
18 the BV chromophore with very small computational cost. Our work clearly describes
19 the impact of various possible steric effects on the nonadiabatic dynamics processes. It
20 provides a guideline for the regulation of the photoisomerization reaction channel,
21 which can further propose some novel ideas for the design of phytochrome proteins. In
22 this sense, the current work provides the preliminary but useful understandings of this

1 important topic efficiently by using a rather simplified method, while it cannot fully
2 and accurately capture the influences of the real biological environment. In order to
3 understand the photoisomerization process of BV chromophore in the protein more
4 realistically, the advanced QM/MM method is the better choice in future research.

5

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8

9 **Supplementary Material**

10 Key bond lengths and dihedral angles of S_{0_min}, S_{1_min} and CI₁₋₄, the structure of
11 S_{0_min}, S_{1_min} and CI₅₋₆, excited-state pathways in the CI₅ and CI₆ channels, the time
12 evolution of the normal vector of the rings, the distributions of key dihedral angles at
13 the initial, hopping and CI geometries, cartesian coordinates are available.

14

15 **References**

- 16 1 N. C. Rockwell, Y. Su and J. C. Lagarias, *Annu. Rev. Plant Biol.*, 2006, **57**, 837-
17 858.
- 18 2 W. L. Butler, K. H. Norris, H. W. Siegelman and S. B. Hendricks, *Proc. Natl. Acad.*
19 *Sci. U.S.A.*, 1959, **45**, 1703-1708.
- 20 3 S. J. Davis, A. V. Vener and R. D. Vierstra, *Science*, 1999, **286**, 2517-2520.
- 21 4 Z. Jiang, L. R. Swem, B. G. Rushing, S. Devanathan, G. Tollin and C. E. Bauer,
22 *Science*, 1999, **285**, 406-409.

- 1 5 A. Blumenstein, K. Vienken, R. Tasler, J. Purschwitz, D. Veith, N. F. Dinkel and R.
2 Fischer, *Curr. Biol.*, 2005, **15**, 1833-1838.
- 3 6 E. A. Rodriguez, G. N. Tran, L. A. Gross, J. L. Crisp, X. Shu, J. Y. Lin and R. Y.
4 Tsien, *Nat. Methods*, 2016, **13**, 763-769.
- 5 7 T. Schroeder, *Nature*, 2008, **453**, 345-351.
- 6 8 R. Weissleder and M. J. Pittet, *Nature*, 2008, **452**, 580-589.
- 7 9 K. G. Chernov, T. A. Redchuk, E. S. Omelina and V. V. Verkhusha, *Chem. Rev.*,
8 2017, **117**, 6423-6446.
- 9 10 A. A. Kaberniuk, A. A. Shemetov and V. V. Verkhusha, *Nat. Methods*, 2016, **13**,
10 591-597.
- 11 11 X. Li, D. Hu, Y. Xie and Z. Lan, *J. Chem. Phys.*, 2018, **149**, 244104.
- 12 12 E. V. Gromov and T. Domratcheva, *Phys. Chem. Chem. Phys.*, 2020, **22**, 8535-8544.
- 13 13 U. Choudry, D. J. Heyes, S. J. O. Hardman, M. Sakuma, I. V. Sazanovich, J.
14 Woodhouse, E. D. L. Mora, M. N. Pedersen, M. Wulff, M. Weik, G. Schirò and N.
15 S. Scrutton, *Chembiochem*, 2018, **19**, 1036-1043.
- 16 14 D. Buhrke, *Phys. Chem. Chem. Phys.*, 2021, **23**, 20867-20874.
- 17 15 D. Buhrke, N. N. Tavraz, D. M. Shcherbakova, L. Sauthof, M. Moldenhauer, F. V.
18 Escobar, V. V. Verkhusha, P. Hildebrandt and T. Friedrich, *Sci. Rep.*, 2019, **9**, 1866.
- 19 16 R. González and M. A. Mroginski, *J. Phys. Chem. B*, 2019, **123**, 9819-9830.
- 20 17 D. Wang, X. Li, L. Wang, X. Yang and D. Zhong, *J. Phys. Chem. Lett.*, 2020, **11**,
21 8819-8824.
- 22 18 D. Wang, Y. Qin, M. Zhang, X. Li, L. Wang, X. Yang and D. Zhong, *J. Phys. Chem.*

- 1 *Lett.*, 2020, **11**, 5913–5919.
- 2 19 J. A. Ihalainen, E. Gustavsson, L. Schroeder, S. Donnini, H. Lehtivuori, L. Isaksson,
3 C. Thöing, V. Modi, O. Berntsson, B. S. Buchli, A. Liukkonen, H. Häkkänen, E.
4 Kalenius, S. Westenhoff and T. Kottke, *J. Am. Chem. Soc.*, 2018, **140**, 12396-12404.
- 5 20 C. Kneip, P. Hildebrandt, W. Schlamann, S. E. Braslavsky, F. Mark and K.
6 Schaffner, *Biochemistry*, 1999, **38**, 15185-15192.
- 7 21 J. R. Wagner, J. S. Brunzelle, K. T. Forest and R. D. Vierstra, *Nature*, 2005, **438**,
8 325-331.
- 9 22 G. Cornilescu, A. T. Ulijasz, C. C. Cornilescu, J. L. Markley and R. D. Vierstra, *J.*
10 *Mol. Biol.*, 2008, **383**, 403-413.
- 11 23 A. T. Ulijasz, G. Cornilescu, C. C. Cornilescu, J. Zhang, M. Rivera, J. L. Markley
12 and R. D. Vierstra, *Nature*, 2010, **463**, 250-254.
- 13 24 X. Yang, Z. Ren, J. Kuk and K. Moffat, *Nature*, 2011, **479**, 428-432.
- 14 25 E. S. Burgie, A. N. Bussell, J. M. Walker, K. Dubiel, and R. D. Vierstra, *Proc. Natl.*
15 *Acad. Sci. U.S.A.*, 2014, **111**, 10179-10184.
- 16 26 H. Takala, A. Björling, O. Berntsson, H. Lehtivuori, S. Niebling, M. Hoernke, I.
17 Kosheleva, R. Henning, A. Menzel, J. A. Ihalainen and S. Westenhoff, *Nature*, 2014,
18 **509**, 245-248.
- 19 27 G. Salvadori, V. Macaluso, G. Pellicci, L. Cupellini, G. Granucci and B. Mennucci,
20 *ChemRxiv*, 2022.
- 21 28 D. Morozov, V. Modi, V. Mironov and G. Groenhof, *J. Phys. Chem. Lett.*, 2022, **13**,
22 4538-4542.

- 1 29 V. Macaluso, G. Salvadori, L. Cupellini and B. Mennucci, *Chem. Sci.*, 2021, **12**,
2 5555-5565.
- 3 30 B. Durbeej, *Phys. Chem. Chem. Phys.*, 2009, **11**, 1354-1361.
- 4 31 A. Strambi and B. Durbeej, *Photochem. Photobiol. Sci.*, 2011, **10**, 569-579.
- 5 32 X. Zhuang, J. Wang and Z. Lan, *J. Phys. Chem. B*, 2013, **117**, 15976-15986.
- 6 33 H. Huang, C. Xu, K. Lin, J. Peng, F. L. Gu and Z. Lan, *ChemRxiv*, 2022.
- 7 34 X. Liu, T. Zhang, Q. Fang, W. Fang, L. González and G. Cui, *Angew. Chem. Int.*
8 *Ed.*, 2021, **60**, 18688-18693.
- 9 35 X. Yang, E. A. Stojkovic, J. Kuk and K. Moffat, *Proc. Natl. Acad. Sci. U.S.A.*, 2007,
10 **104**, 12571-1257.
- 11 36 K. Fushimi, T. Miyazaki, Y. Kuwasaki, T. Nakajima, T. Yamamoto, K. Suzuki, Y.
12 Ueda, K. Miyake, Y. Takeda, J. H. Choi, H. Kawagishi, E. Y. Park, M. Ikeuchi, M.
13 Sato and R. Narikawa, *Proc. Natl. Acad. Sci. U.S.A.*, 2019, **116**, 8301-8309.
- 14 37 X. Shu, A. Royant, M. Z. Lin, T. A. Aguilera, V. L. Ram, P. A. Steinbach and R. Y.
15 Tsien, *Science*, 2009, **324**, 804-807.
- 16 38 F. F. Jöbsis, *Science*, 1977, **198**, 1264-1267.
- 17 39 R. Weissleder and V. Ntziachristos, *Nat. Med.*, 2003, **9**, 123-128.
- 18 40 H. M. Senn and W. Thie, *Angew. Chem. Int. Ed.*, 2009, **48**, 1198-1229.
- 19 41 R. C. Otero and M. Barbatti, *Chem. Rev.*, 2018, **118**, 7026-7068.
- 20 42 E. Brunk and U. Rothlisberger, *Chem. Rev.*, 2015, **115**, 6217-6263.
- 21 43 J. Gao, *Acc. Chem. Res.*, 1996, **29**, 298-305.
- 22 44 J. Gao and D. G. Truhlar, *Annu. Rev. Phys. Chem.*, 2002, **53**, 467-505.

- 1 45 A. Warshel, *Nature*, 1976, **260**, 679-683.
- 2 46 J. J. Szymczak, M. Barbatti and H. Lischka, *J. Chem. Theory Comput.*, 2008, **4**,
3 1189-1199.
- 4 47 M. Barbatti, M. Ruckebauer, J. J. Szymczak, A. J. A. Aquino and H. Lischka, *Phys.*
5 *Chem. Chem. Phys.*, 2008, **10**, 482-494.
- 6 48 M. Barbatti and H. Lischka, *J. Phys. Chem. A*, 2007, **111**, 2852-2858.
- 7 49 M. Barbatti and H. Lischka, *J. Am. Chem. Soc.*, 2008, **130**, 6831-6839.
- 8 50 W. Weber and W. Thiel, *Theor. Chem. Acc.*, 2000, **103**, 495-506.
- 9 51 A. Koslowski, M. E. Beck and W. Thiel, *J. Comput. Chem.*, 2003, **24**, 714-726.
- 10 52 G. Granucci and M. Persico, *J. Chem. Phys.*, 2007, **126**, 134114.
- 11 53 T. W. Keal, A. Koslowski and W. Thiel, *Theor. Chem. Acc.*, 2007, **118**, 837-844.
- 12 54 W. Thiel, MNDO99 program, 2020, Max-Planck-Institut für Kohlenforschung,
13 Mülheim an der Ruhr, *Germany*, 2020.
- 14 55 S. Xia, B. Xie, Q. Fang, G. Cui and W. Thiel, *Phys. Chem. Chem. Phys.*, 2015, **17**,
15 9687-9697.
- 16 56 E. P. Wigner, in Part I: *Phys. Chem.* Part II: *Solid State Physics*, Springer, 1997,
17 110-120.
- 18 57 J. C. Tully, *J. Chem. Phys.*, 1990, **93**, 1061-1071.
- 19 58 G. Granucci and M. Persico, *J. Chem. Phys.*, 2007, **126**, 134114.
- 20 59 L. Du and Z. Lan, *J. Chem. Theory Comput.*, 2015, **11**, 1360-1374.
- 21
- 22