Mechanism of singlet fission in carotenoids from

a polyene model system

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Abstract

Singlet fission (SF) is the process of formation of multiple excitons (triplet) from

a locally excited singlet state. The mechanism of SF in polyacenes have been shown

to proceed via a charge transfer intermediate state. However, carotenoids are not

understood in the context of SF. This is possibly due to the complicated multireference

nature of the low-lying excited states of carotenoids and the presence of a dark $2^{1}A_{g}$

state below the optically bright ${}^{1}B_{u}$ state. In this work, we show that the dark A_{g}

state in polyenes / carotenoids, along with the charge transfer states, play a pivotal

role in the SF process. We notice that the relative importance of these states vary

with change in geometry and the overall presence of multiple pathways is crucial to

the success of SF process in carotenoid aggregates and disordered geometries.

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Singlet fission (SF) is the process of formation of two free triplet states residing on separated monomers from an initially excited singlet state localized on a single monomer (chromophore).^{1,2} This process proceeds via formation of coupled triplet state (coupled-TT) on multiple chromophores in near vicinity, and its subsequent diffusion to free triplets on distant chromophoric sites.³ This first step of the SF process is schematically shown in Fig. 1. SF can only occur in the presence of multiple chromophores and the SF phenomena has been observed in crystals, aggregates, thin films and in disordered states such as biological medium.⁴⁻⁶ SF is characterized by the observation of more than 100% triplet quantum efficiency, which denotes the formation of two excitons by the action of a single photon. SF process have been extensively observed in acene systems.^{4,5,7,8}

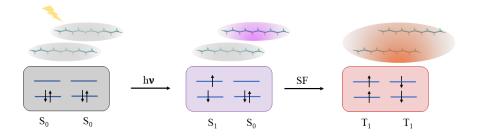


Figure 1: Schematic representation of the first step of SF process - formation of coupled-TT state.

While SF process has been known for a long time, its mechanism is only recently being studied. 8–10 In polyacenes, the coupled-TT state formation is mediated by a charge transfer (CT) excited state, where CT occurs between the two monomers where the coupled-TT state resides. 9,11,12 Since the CT state is sensitive to the relative orientation of the monomers, the efficiency of SF process depends on the crystal packing and geometry. 13–15 Therefore, many of the experimental observations in acenes have been explained within this model.

Carotenoids exhibit SF phenomena in H-aggregates and also in biological phases with significant disorder.^{6,16–19} In J-aggregates, singlet fission is observed for dipolar carotenoids, although it has not been observed for non-polar carotenoids.¹⁷ This apparent insensitivity to structural features of SF in carotenoids remains unexplained. Furthermore, the carotenoid

monomer transitions are significantly different from the acenes. The first excited state in carotenoids is an optically dark $2^{1}A_{g}$ state followed by a second optically active ${}^{1}B_{u}$ state (HOMO-LUMO excitation). 20 This unexpected ordering of the excited states is due to multireference nature of the conjugated polyenes/carotenoids. 21,22 Therefore, to address these intricacies of SF in carotenoids one has to extend the model to include both A_{g} and B_{u} states.

Carotenoids spectra and low-lying excited states are similar to conjugated polyenes due to the predominant importance of the conjugated backbone. In our work we consider a conjugated polyene model system, decapentaene to understand mechanism of SF in carotenoids. The choice of decapentaene as a model system is the position of its A_g state below the B_u state. All polyenes shorter than decapentaene have a first excited B_u state. We probe the importance of A_g states as well as CT states as effective conduits for the formation of the coupled-TT state from the initial optically active state localized on a single polyene. We will show that depending on the relative geometries of the monomers, a variety of pathways via A_g and CT states are indeed effective intermediates for SF.

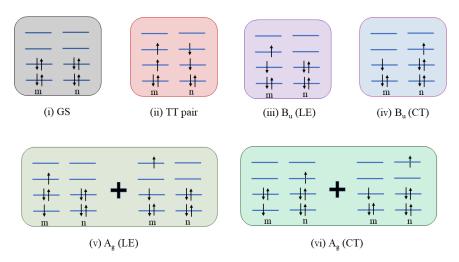


Figure 2: Electronic configuration of the prototype low-lying states of polyene / carotenoid systems. The locally excited states are denoted as LE and the charge transfer states as CT. The A_g and B_u states refer to the symmetry of the excited states on the monomers.

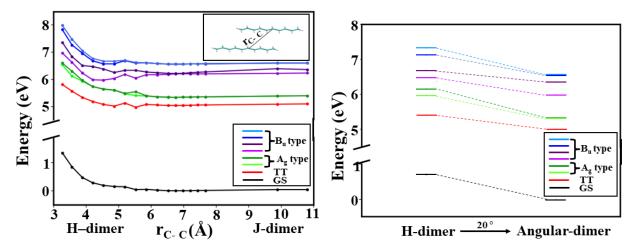
The predominant electronic configurations for the low-lying excited states of polyene dimers are shown in Figure 2. The different types of states that are possible are: (i) ground state (GS) which is predominantly closed shell in nature; (ii) coupled-TT state with two triplet states localized on monomers m and n, such that the total spin of the system is an effective singlet; (iii) optically active B_u states that are localized on single monomer (i.e., locally excited or LE) and is characterized by a single electron excitation from HOMO $_m$ to LUMO $_m$; (iv) CT state involving a HOMO $_m$ \rightarrow LUMO $_n$ excitation (we denote this as $B_u(CT)$); (v) A_g state that is localized on a single monomer (LE) and characterized by a two electron excitation $HOMO_m \rightarrow (LUMO + 1)_m$ and $(HOMO - 1)_m \rightarrow LUMO_m$; and (vi) A_g state with a CT component, i.e., two electron excitation with $HOMO_m \rightarrow (LUMO + 1)_n$ and $(HOMO - 1)_m \rightarrow LUMO_n$. It is important to note that with respect to the ground state, the B_u and $B_u(CT)$ states are one electron excitations, while the A_g , $A_g(CT)$ and TT states are two electron excitations.

The SF process has to satisfy certain energetic and coupling criteria to occur efficiently. Since SF requires the formation of two triplet excited states from one excited singlet state, the individual monomers must satisfy the basic rules of conservation of energy, ^{1,23} i.e.,

$$E(S_1 - S_0) \ge 2E(T_1 - S_0).$$
 (1)

While this is a necessary condition, it is not a sufficient one as can be suspected from the sensitivity of SF efficiency on geometry and crystal packing in some SF materials. For the energy transfer to occur effectively, the coupling term between the initial and final states, $\langle \Psi_i | \hat{H} | \Psi_f \rangle$, has to be significant. When one considers the B_u (S₁) state as the initial state and coupled-TT as the final state, this can only be viewed as a two electron transfer process and the coupling term between these states is quite small. Therefore, it is presumed that the SF process occurs predominantly through an intermediate. In Refs 25–27, using various nonadiabatic approaches, it has been proposed that the intermediate state contains large CT components in case of the polyacenes. The CT state can be viewed as an one electron transferred state from S₁ state and also connected to the coupled-TT state via one

electron transfer. Therefore, the coupling terms of CT state with both S_1 and coupled-TT states are significantly large ($\approx 100 \text{ meV}$) to allow SF process via the CT state.



(a) Excited state ordering from J-H structure of de- (b) Effect of rotation on the excited state ordering capentaene dimers.

Figure 3: (a) Potential energy surfaces of low-lying states of decapentaene dimers calculated at the (8e,8o) active space SA-CASSCF/6-31G(d) level of theory. In the inset on the upper left corner the r_{C-C} is shown, which is plotted along the x-axis. (b) Comparison of the relative state ordering in an angular geometry with H-dimer geometry.

When this same approach is considered for the carotenoids/polyenes, the energy conservation condition for the monomers is satisfied (shown in SI). In the dimer geometries, the lowest excited state is the coupled-TT state as expected (Fig. 2). A_g states reside in between the coupled-TT and B_u states. Therefore, energetically the A_g states are in the correct region to act as effective intermediates. It should be noted that the CT states also satisfy this energy criterion in the polyene/carotenoid systems.

To evaluate the possibility of A_g states as intermediates one should also calculate the coupling terms. However, before the quantitative estimates, let us look at the qualitative features of these A_g states. The excited A_g state cannot be viewed from predominantly one electron HOMO-LUMO type transition. Excited A_g state is a multireference state with two electron transitions (with respect to ground state) - one transition from HOMO \rightarrow LUMO+1 and another from HOMO-1 \rightarrow LUMO.^{21,22} Therefore, it requires at least two configurations to be qualitatively described. It can, alternatively, be viewed as two spin

flips on a 1-D Heisenberg lattice (linear chain) of antiferromagnetically coupled spins. ²⁸ The antiferromagnetic couplings ensure the formation of bond pairs between adjacent sites, i.e., p_z orbitals. Thus, the ground state has alternating up and down spins to maximize favorable interactions. When two spins are flipped at distant locations two magnons are formed at these locations, i.e., it is a bimagnon state with $\uparrow \uparrow$ and $\downarrow \downarrow$ configurations. This bimagnon A_g state looks qualitatively similar to the coupled-TT state with the noteworthy difference that the A_g state has 2 magnons located on the same polyene (or chain) while coupled-TT is on different polyenes. Due to these similarities, one can expect to obtain large coupling between the A_g state and coupled-TT state. Furthermore, it is well known that carotenoids in photosynthetic systems show fast non-radiative transitions between the B_u state and the A_g state. ²⁹ Therefore, A_g states seems to have significant transition probabilities from both B_u state as well as coupled-TT state and therefore, can act as an intermediate for SF in carotenoids / polyenes.

The dimer model structures are constructed from the decapentaene monomers. The dimer is created by forming slip stacked geometry of decapentaene and optimized at B3LYP/6-31G(d) level of theory. The H-dimer geometry and all intermediate geometries are created by reducing the slide, i.e., increasing the slip angle between the dimers, while the J-dimer structure is created by increasing the slide, i.e., reducing the slip angle. The slip angles of the H and J-dimer structures used are 70.37° and 31.66°, respectively. This is in accordance with the definitions of aggregates in Ref. 30. Fig. 3(a) shows the excited states and their nature along the H to J slipped geometries. The first excited state is a coupled-TT state, followed by 2 A_g states and finally 4 B_u states with different amounts of CT character. In H-dimer, the B_u states are all optically active, while in J-dimer the third B_u state is optically active. From the many degenerate excited states in the J-dimer, it is evident that due to large slide motion, the dimers are quite far apart and non-interacting, i.e., the states retain mostly the monomer characters. In H-dimer the degeneracies are lifted, i.e., the excited states in the monomers interact with each other giving rise to Davydov splitting. We further notice

the effect of small rotation (20°) on the H-dimer and observe that the relative ordering of the states are similar (Fig. 3(b)). There are limited interactions between the states and degeneracies are somewhat lifted at least for the B_u states.

In order to compare the different states as intermediates in the SF process, diabatic representation of each of the states are constructed by considering the spin adapted configurations as shown in Fig. 2. For the configurations, the orbitals from (8e,8o) CASSCF/6-31G(d) calculations are taken after Pipek Mezey localization on respective monomers. The couplings coefficients between these diabatic states were calculated following the procedure in Ref. 25,26 and extended to include the A_g states. The equations used for the couplings are included in SI.

The heat maps of the coupling terms between the different states are given in Fig. 3, at H-dimer, J-dimer, intermediate (H-J) and rotated geometries. On cursory inspection, one can notice the numerous high coupling terms (> 100 meV) in all geometries, except for J-dimer.

In H-dimer, there is a large coupling between the optically active local excited $B_u(LE)$ state and the A_g states with or without CT components as well as $B_u(CT)$ state. Therefore, one can expect the excitation to decay from B_u state to A_g , $A_g(CT)$ and $B_u(CT)$ states. Following this, there are large couplings between $B_u(CT)$ and A_g states with the subsequent coupled-TT state. Therefore, the SF process can occur through both $B_u(CT)$ and A_g states as intermediate in the H-dimer. The signature of A_g state as an intermediate between B_u and TT state is in good agreement with experimental observation from transient absorption studies. ¹⁶

For J-dimer, the coupling matrix looks markedly different from the other geometries. It has only few large coupling terms. There are couling terms between B_u state to A_g state but no large coupling with the TT state. Therefore, while there is a feasible pathway from B_u to A_g state, it cannot decay onwards into the TT state. Therefore, our model predicts that SF process will not occur in J aggregate geometry in non-dipolar polyenes. This is in good

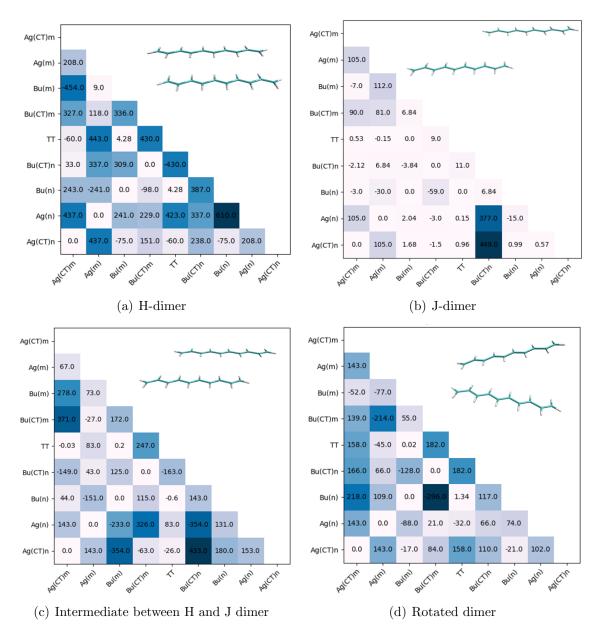


Figure 4: Off-diagonal elements of the coupling matrices for (a) H-dimer, (b) J-dimer, (c) H/J-dimer (intermediate geometry), and (d) Angular-dimer. The values of the couplings are in meV and higher coupling values are denoted by darker colors. The corresponding geometries are shown for each case.

agreement with experimental observations.

In the intermediate structure between H and J geometries, the optically active B_u state is most strongly coupled with $A_g(CT)$ and A_g states, and moderate coupling with $B_u(CT)$ state. When one considers the coupling with TT state, it is only with A_g and $B_u(CT)$.

Therefore, in this case the predominant pathway is via A_g with some contributions from $B_u(CT)$ state.

The angular geometry shows strong SF probabilities via CT states only, with both B_u and A_g characters. Therefore, it shows that disordered geometries are also capable of undergoing SF. The insensitivity of geometry on SF process in carotenoids can be explained within our model.

To summarize, from the energetic criterion in the monomer, i.e., comparison of energies between first optically active S_1 and T_1 states, it appears that SF phenomena is feasible in carotenoids and decapentaene as the model system. In order to explain the effect of geometry on SF in polyenes / carotenoids, we obtain the excited state energies at various geometries. In all the geometries, except J-dimer, significant Davydov splitting is observed, and therefore, we surmise that the excited states on the monomers are strongly interacting. We extend the diabatic model in Ref 25,26 to include all the low-lying excited states and calculate the coupling between these excited states. It shows that there are numerous efficient conduits for SF in H-dimer, e.g., A_g state and CT states with both A_g and B_u nature. In case of rotated geometry, the most important pathways are both A_g and CT state with B_u nature. The model also shows that there are no efficient pathways in J-dimer. It should be noted that J-aggregates with polar carotenoids have shown signatures of SF phenomena and to capture that one might have to include polar groups in the theoretical study.

Therefore, this model for the first time shows the importance of A_g state as the intermediate for SF process and explains the experimental observations quite succinctly. It explains the experimental observation of A_g like peaks at intermediate timescales in transient absorption spectra on the way to a coupled-TT state formation. The presence of A_g states as intermediates along with CT states for the polyene system, shows the crucial difference of SF mechanism in carotenoids as compared to acenes.

Acknowledgement

The authors thank fruitful discussions with Prof. Jyotishman Dasgupta (TIFR). D.G. thanks generous funding from SERB-POWER fellowship and IACS computational facilities. S.S. thanks DST-INSPIRE program for junior research fellowship.

Supporting Information Available

Detailed computational methods used in the work is given, along the geometries of the dimers that are referred to in the manuscript. The couplings between the different types of excited states and their formulae are given.

References

- (1) Smith, M. B.; Michl, J. Singlet fission. Chemical reviews 2010, 110, 6891–6936.
- (2) Smith, M. B.; Michl, J. Recent advances in singlet fission. *Annual review of physical chemistry* **2013**, *64*, 361–386.
- (3) Breen, I.; Tempelaar, R.; Bizimana, L. A.; Kloss, B.; Reichman, D. R.; Turner, D. B. Triplet separation drives singlet fission after femtosecond correlated triplet pair production in rubrene. *Journal of the American Chemical Society* **2017**, *139*, 11745–11751.
- (4) Korovina, N. V.; Joy, J.; Feng, X.; Feltenberger, C.; Krylov, A. I.; Bradforth, S. E.; Thompson, M. E. Linker-dependent singlet fission in tetracene dimers. *Journal of the American Chemical Society* 2018, 140, 10179–10190.
- (5) Zirzlmeier, J.; Lehnherr, D.; Coto, P. B.; Chernick, E. T.; Casillas, R.; Basel, B. S.; Thoss, M.; Tykwinski, R. R.; Guldi, D. M. Singlet fission in pentacene dimers. *Proceedings of the National Academy of Sciences* 2015, 112, 5325–5330.

- (6) Wang, C.; Schlamadinger, D. E.; Desai, V.; Tauber, M. J. Triplet excitons of carotenoids formed by singlet fission in a membrane. *ChemPhysChem* **2011**, *12*, 2891–2894.
- (7) Korovina, N. V.; Chang, C. H.; Johnson, J. C. Spatial separation of triplet excitons drives endothermic singlet fission. *Nature Chemistry* **2020**, *12*, 391–398.
- (8) Zimmerman, P. M.; Zhang, Z.; Musgrave, C. B. Singlet fission in pentacene through multi-exciton quantum states. *Nature chemistry* **2010**, *2*, 648–652.
- (9) Casanova, D. Theoretical modeling of singlet fission. *Chemical reviews* **2018**, *118*, 7164–7207.
- (10) Yost, S. R.; Lee, J.; Wilson, M. W.; Wu, T.; McMahon, D. P.; Parkhurst, R. R.; Thompson, N. J.; Congreve, D. N.; Rao, A.; Johnson, K., et al. A transferable model for singlet-fission kinetics. *Nature chemistry* 2014, 6, 492–497.
- (11) Monahan, N.; Zhu, X.-Y. Charge transfer-mediated singlet fission. *Annual review of physical chemistry* **2015**, *66*, 601–618.
- (12) Omar, Ö. H.; Padula, D.; Troisi, A. Elucidating the Relationship between Multiradical Character and Predicted Singlet Fission Activity. *ChemPhotoChem* **2020**, *4*, 5223–5229.
- (13) Bhattacharyya, K.; Datta, A. Computationally Driven Design Principles for Singlet Fission in Organic Chromophores. J. Phys. Chem. C 2019, 123, 19257–19268.
- (14) Roberts, S. T.; McAnally, R. E.; Mastron, J. N.; Webber, D. H.; Whited, M. T.; Brutchey, R. L.; Thompson, M. E.; Bradforth, S. E. Efficient singlet fission discovered in a disordered acene film. *Journal of the American Chemical Society* 2012, 134, 6388– 6400.
- (15) Nagami, T.; Ito, S.; Kubo, T.; Nakano, M. Intermolecular packing effects on singlet fission in oligorylene dimers. *ACS omega* **2017**, *2*, 5095–5103.

- (16) Kundu, A.; Dasgupta, J. Photogeneration of Long-Lived Triplet States through Singlet Fission in Lycopene H-Aggregates. The Journal of Physical Chemistry Letters 2021, 12, 1468–1474.
- (17) Musser, A. J.; Maiuri, M.; Brida, D.; Cerullo, G.; Friend, R. H.; Clark, J. The nature of singlet exciton fission in carotenoid aggregates. *Journal of the American Chemical Society* **2015**, *137*, 5130–5139.
- (18) Quaranta, A.; Krieger-Liszkay, A.; Pascal, A. A.; Perreau, F.; Robert, B.; Vengris, M.; Llansola-Portoles, M. J. Singlet fission in naturally-organized carotenoid molecules. *Physical Chemistry Chemical Physics* 2021, 23, 4768–4776.
- (19) Yu, J.; Fu, L.-M.; Yu, L.-J.; Shi, Y.; Wang, P.; Wang-Otomo, Z.-Y.; Zhang, J.-P. Carotenoid singlet fission reactions in bacterial light harvesting complexes as revealed by triplet excitation profiles. *Journal of the American Chemical Society* 2017, 139, 15984–15993.
- (20) Sashima, T.; Koyama, Y.; Yamada, T.; Hashimoto, H. The 1Bu+, 1Bu-, and 2Agenergies of crystalline lycopene, β-carotene, and mini-9-β-carotene as determined by resonance-Raman excitation profiles: dependence of the 1Bu-state energy on the conjugation length. The Journal of Physical Chemistry B 2000, 104, 5011–5019.
- (21) Ghosh, D.; Hachmann, J.; Yanai, T.; Chan, G. K.-L. Orbital optimization in the density matrix renormalization group, with applications to polyenes and β-carotene. The Journal of chemical physics 2008, 128, 144117.
- (22) Tavan, P.; Schulten, K. The low-lying electronic excitations in long polyenes: A PPP-MRD-CI study. *The Journal of chemical physics* **1986**, *85*, 6602–6609.
- (23) Zeng, T.; Hoffmann, R.; Ananth, N. The low-lying electronic states of pentacene and their roles in singlet fission. *Journal of the American Chemical Society* **2014**, *136*, 5755–5764.

- (24) Johnson, J. C.; Nozik, A. J.; Michl, J. The role of chromophore coupling in singlet fission. *Accounts of chemical research* **2013**, *46*, 1290–1299.
- (25) Berkelbach, T. C.; Hybertsen, M. S.; Reichman, D. R. Microscopic theory of singlet exciton fission. I. General formulation. The Journal of chemical physics 2013, 138, 114102.
- (26) Berkelbach, T. C.; Hybertsen, M. S.; Reichman, D. R. Microscopic theory of singlet exciton fission. II. Application to pentacene dimers and the role of superexchange. The Journal of chemical physics 2013, 138, 114103.
- (27) Li, X.; Parrish, R. M.; Martínez, T. J. An ab initio exciton model for singlet fission.

 The Journal of Chemical Physics 2020, 153, 184116.
- (28) Aryanpour, K.; Shukla, A.; Mazumdar, S. Theory of singlet fission in polyenes, acene crystals, and covalently linked acene dimers. *The Journal of Physical Chemistry C* **2015**, *119*, 6966–6979.
- (29) Polivka, T.; Frank, H. A. Molecular factors controlling photosynthetic light harvesting by carotenoids. *Accounts of chemical research* **2010**, *43*, 1125–1134.
- (30) Köhn, S.; Kolbe, H.; Korger, M.; Köpsel, C.; Mayer, B.; Auweter, H.; Lüddecke, E.; Bettermann, H.; Martin, H.-D. *Carotenoids*; Springer, 2008; Chapter 5, pp 53–98.