

Synthesis and styrene copolymerization of halogen ring-trisubstituted 2-methoxyethyl phenylcyanoacrylates

Zipporah Y. Kaffey, Tyler J. McKenna, Angelina B. Moore, Abigail T. Noble, Simoni Patel, Ana S. Rivera, Joshua Sapinsley, Elena G. Sasso, Paige Sevald, Kay A.K. Smith, Sara M. Rocus, William S. Schjerven, and Gregory B. Kharas

DePaul University, Chemistry and Biochemistry Department, 1110 West Belden Avenue, Chicago, IL 60614-3214

Contact: gkharas@depaul.edu

Abstract

Novel halogen ring-trisubstituted 2-methoxyethyl phenylcyanoacrylates, $RPhCH=C(CN)CO_2CH_2CH_2OCH_3$ (where R is 5-chloro-2,3-dimethoxy, 3-bromo-4,5-dimethoxy, 5-bromo-2,3-dimethoxy, 2-bromo-3-hydroxy-4-methoxy, 4-bromo-2,6-difluoro, 3-chloro-2,6-difluoro, 4-chloro-2,6-difluoro, 2-hydroxy-3,5-diiodo, 2,3,6-trichloro, 2,3,4-trifluoro, 2,4,6-trifluoro) were prepared and copolymerized with styrene. The acrylates were synthesized by the piperidine catalyzed Knoevenagel condensation of ring-trisubstituted benzaldehydes and 2-methoxyethyl cyanoacetate, and characterized by CHN analysis, IR, 1H and ^{13}C NMR. All the acrylates were copolymerized with styrene in solution with radical

initiation (ABCN) at 70°C. The compositions of the copolymers were calculated from nitrogen analysis.

1. Introduction

Ring trisubstituted, 3-bromo-4-ethoxy-5-methoxyphenyl 2-methoxyethyl phenylcyanoacrylate (PCA) is reported in biochemical evaluation studies of cell-active inhibitor of the cancer-promoting phosphatases of regenerating liver [1], and in dataset of log P measurements for benchmarking studies [2]. 3-Bromo-4,5-dimethoxyphenyl ring substituted PCA is mentioned in studies of 4-aryl-2-oxo-2H-chromenes as a new series of apoptosis inducers using a cell- and caspase-based high-throughput screening assay [3], and in preparation of substituted coumarins and quinolinones as caspase activators for treatment of cancer [4]. Another ring substituted PCA, 2-bromo-3-ethenyl-6-methoxyphenol is involved in the Mitsunobu reaction [5]; in studies culminating in the total synthesis of (dl)-morphine [6], and in tandem intramolecular conjugate addition/intramolecular alkylation reactions of substituted vinyl sulfones [7]. 4-Hydroxy-3,5-diiodophenyl is reported in electron transport inhibition of the cytochrome bc₁ complex of rat-liver mitochondria by phenolic uncouplers [8]. 2,3,4-Trifluoro-2-ethenyl benzene is mentioned in synthesis of isoxazolines and isoxazoles inspired by Fipronil [9], and in studies of infrared group frequency correlations for styrenes, α -methylstyrenes [10]. 1,3,4-Trifluoro-2-ethenyl benzene is used in preparation ethylene-fluorostyrene copolymer [11], and in preparation of liquid crystal composition for LC display [12]. 2,4,5-Trifluoro-2-ethenyl benzene was involved in Palladium-catalyzed intermolecular aminofluorination of styrenes [13]. 1-

Chloro-2-ethenyl-4,5-dimethoxy-benzene was involved in process for preparing styrene derivatives by a transition metal-catalyzed cross coupling of chlorostyrenes with organomagnesium compounds [14], and in process for preparing styrene derivatives by a transition metal-catalyzed cross coupling of chlorostyrenes with organomagnesium compounds [15].

In this work we have prepared novel ring-trisubstituted 2-methoxyethyl phenylcyanoacrylates, MEPA, $RPhCH=C(CN)CO_2CH_2CH_2OCH_3$, where R is 5-chloro-2,3-dimethoxy, 3-bromo-4,5-dimethoxy, 5-bromo-2,3-dimethoxy, 2-bromo-3-hydroxy-4-methoxy, 4-bromo-2,6-difluoro, 3-chloro-2,6-difluoro, 4-chloro-2,6-difluoro, 2-hydroxy-3,5-diiodo, 2,3,6-trichloro, 2,3,4-trifluoro, 2,4,6-trifluoro, and explored the feasibility of their copolymerization with styrene. To the best of our knowledge there have been no reports on either synthesis of these compounds, nor their polymerization [16].

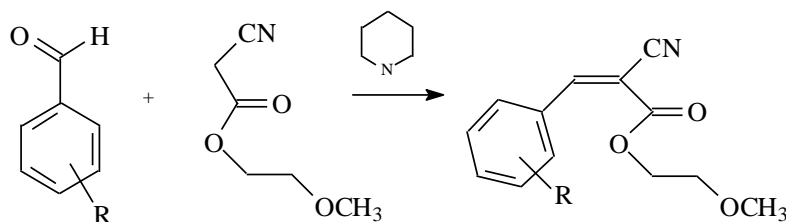
2. Experimental

5-Chloro-2,3-dimethoxy, 3-bromo-4,5-dimethoxy, 5-bromo-2,3-dimethoxy, 2-bromo-3-hydroxy-4-methoxy, 4-bromo-2,6-difluoro, 3-chloro-2,6-difluoro, 4-chloro-2,6-difluoro, 2-hydroxy-3,5-diiodo, 2,3,6-trichloro, 2,3,4-trifluoro, 2,4,6-trifluoro, 2-methoxyethyl cyanoacetate ($\geq 98.0\%$), piperidine (99%), styrene ($\geq 99\%$), 1,1'-azobis(cyclohexanecarbonitrile) (98%), (ABCN), and toluene (98%) supplied from Sigma-Aldrich Co., were used as received. Instrumentation is described in [17].

3. Results and discussion

3.1. Synthesis and characterization of 2-methoxyethyl phenylcyanoacrylates

All MEPA compounds were synthesized by Knoevenagel condensation [18] of appropriate benzaldehydes with 2-methoxyethyl cyanoacetate, catalyzed by base, piperidine (Scheme 1).



Scheme 1. Synthesis of 2-methoxyethyl phenylcyanoacrylates, where R is 5-chloro-2,3-dimethoxy, 3-bromo-4,5-dimethoxy, 5-bromo-2,3-dimethoxy, 2-bromo-3-hydroxy-4-methoxy, 4-bromo-2,6-difluoro, 3-chloro-2,6-difluoro, 4-chloro-2,6-difluoro, 2-hydroxy-3,5-diiodo, 2,3,6-trichloro, 2,3,4-trifluoro, 2,4,6-trifluoro.

The preparation procedure was essentially the same for all the MEPA compounds. In a typical synthesis, equimolar amounts of 2-methoxyethyl cyanoacetate and an appropriate benzaldehyde were mixed in equimolar ratio in a 20 mL vial. A few drops of piperidine were added with stirring. The product of the reaction was isolated by filtration and purified by crystallization from 2-propanol. The condensation reaction proceeded smoothly, yielding products, which were purified by conventional techniques. The compounds were characterized by IR, ^1H and ^{13}C NMR spectroscopies. No stereochemical analysis of the novel alkoxy ring-substituted MEPA was performed since no stereoisomers (*E* or/and *Z*) of known configuration were available.

3.1.1. 2-Methoxyethyl 5-chloro-2,3-dimethoxyphenylcyanoacrylate

Yield: 92%; mp 75.1°C; ¹H NMR: δ 8.7 (s, 1H, CH=), 7.9.-7.1 (m, 2H, Ph), 4.5 (t, 2H, OCOCH₂), 3.9 (s, 6H, PhOCH₃), 3.8 (t, 2H, OCH₂), 3.4 (s, 3H, CH₃O); ¹³C NMR: δ 162 (C=O), 153 (HC=), 152, 130, 121, 121 (Ph), 116 (CN), 104 (C=), 74 (OCH₂), 64 (OCOCH₂), 59 (OCH₃), 57 (PhOCH₃); IR: (cm⁻¹) 2982 (m, C-H), 2228 (m, CN), 1722 (s, C=O), 1599 (s, C=C), 1252 (s, C-O-CH₃), 938 (s, C-H out of plane). Anal. calcd. for C₁₅H₁₆ClNO₅: C, 55.31; H, 4.95; N, 4.30; Found: C, 53.6; H, 5.07; N, 4.70.

3.1.2. 2-Methoxyethyl 3-bromo-4,5-dimethoxyphenylcyanoacrylate

Yield 85%; mp 122.2°C; ¹H NMR δ 8.2 (s, 1H, CH=), 7.8-7.3 (m, 2H, Ph), 4.5 (t, 2H, OCOCH₂), 3.8 (s, 3H, PhOCH₃), 3.7 (t, 2H, OCH₂), 3.4 (s, 3H, CH₃O); ¹³C NMR δ 162 (C=O), 153 (HC=), 152, 151, 131, 130, 118 (Ph), 117 (CN), 102 (C=), 70 (OCH₂), 64 (OCOCH₂), 59 (OCH₃), 58 (PhOCH₃); IR (cm⁻¹): 2943 (m, C-H), 2222 (m, CN), 1723 (s, C=O), 1634 (s, C=C), 1248 (s, C-O-CH₃), 874, 848, 762 (s, C-H out of plane). Anal. calcd. for C₁₅H₁₆BrNO₅: C, 48.67; H, 4.36; N, 3.78; Found: C, 46.76; H, 4.28; N, 4.01.

3.1.3. 2-Methoxyethyl 5-bromo-2,3-dimethoxyphenylcyanoacrylate

Yield 62%; mp 112.4°C; ¹H NMR δ 8.6 (s, 1H, CH=), 8.1-7.2 (m, 2H, Ph), 4.5 (t, 2H, OCOCH₂), 3.9 (s, 6H, PhOCH₃), 3.7 (t, 2H, OCH₂), 3.4 (s, 3H, CH₃O); ¹³C NMR δ 162 (C=O), 154 (HC=), 152, 149, 129, 120 (Ph), 117 (CN), 105 (C=), 70 (OCH₂), 66 (OCOCH₂), 59 (OCH₃), 56 (PhOCH₃); IR (cm⁻¹): 2941 (m, C-H), 2224 (m, CN), 1732 (s,

C=O), 1605 (s, C=C), 1228 (s, C-O-CH₃), 849, 755 (s, C-H out of plane). Anal. calcd. for C₁₅H₁₆BrNO₅: C, 48.67; H, 4.36; N, 3.78; Found: C, 46.64; H, 4.49; N, 3.99.

3.1.4. 2-Methoxyethyl 2-bromo-3-hydroxy-4-methoxyphenylcyanoacrylate

Yield 78%; mp 123.8°C; ¹H NMR: δ 10.3 (s, 1H, OH), 8.7 (s, 2H, CH=), 8.1-6.9 (m, 2H, Ph), 4.5 (t, 2H, OCOCH₂), 4.0 (s, 6H, PhOCH₃), 3.7 (t, 2H, OCH₂), 3.4 (s, 3H, OCH₃); ¹³C NMR: δ 163 (C=O), 155 (HC=), 153, 145, 142, 122, 121, 115 (Ph), 117 (CN), 103 (C=), 74 (OCH₂), 66 (OCOCH₂), 59 (OCH₃), 57 (PhOCH₃); IR: (cm⁻¹) 2941 (m, C-H), 2222 (m, CN), 1713 (s, C=O), 1634 (s, C=C), 1234 (s, C-O-CH₃), 791 (s, C-H out of plane). Anal. calcd. for C₁₄H₁₄BrNO₅: C, 47.21; H, 3.96; N, 3.93; Found: C, 46.05; H, 4.09; N, 4.31.

3.1.5. 2-Methoxyethyl 4-bromo-2,6-difluorophenylcyanoacrylate

Yield 89%; mp 75°C; ¹H NMR: δ 8.2 (s, 1H, CH=), 7.1-7.3 (m, 2H, Ph), 4.6 (t, 2H, OCOCH₂), 3.8 (t, 2H, OCH₂), 3.4 (s, 3H, OCH₃); ¹³C NMR: δ 162 (C=O), 159 (HC=), 131, 125, 123 (Ph), 115 (CN), 100 (C=), 73 (OCH₂), 66 (OCOCH₂), 61 (OCH₃); IR: (cm⁻¹) 2937 (m, C-H), 2223 (m, CN), 1726 (s, C=O), 1602 (s, C=C), 1217 (s, C-O-CH₃), 843, 806 (s, C-H out of plane). Anal. calcd. for C₁₃H₁₀BrF₂NO₃: C, 45.44; H, 2.91; N, 4.05; Found: C, 44.88; H, 3.36; N, 4.77.

3.1.6. 2-Methoxyethyl 3-chloro-2,6-difluorophenylcyanoacrylate

Yield 76%; mp 54.1°C; ¹H NMR: δ 8.2 (s, 1H, CH=), 7.7-6.8 (m, 2H, Ph), 4.6 (t, 2H, OCOCH₂), 3.8 (t, 2H, OCH₂), 3.4 (s, 3H, OCH₃); ¹³C NMR: δ 162 (C=O), 157 (HC=), 142, 137, 136, 119, 109 (Ph), 116 (CN), 100 (C=), 71 (OCH₂), 66 (OCOCH₂), 59

(OCH₃); IR: (cm⁻¹) 2935 (m, C-H), 2226 (m, CN), 1736 (s, C=O), 1626 (s, C=C), 1225 (s, C-O-CH₃), 816, 756 (s, C-H out of plane). Anal. calcd. for C₁₃H₁₀ClF₂NO₃: C, 51.76; H, 3.34; N, 4.64; Found: C, 47.96; H, 3.16; N, 4.70.

3.1.7. 2-Methoxyethyl 4-chloro-2,6-difluorophenylcyanoacrylate

Yield 86%; mp 64.1°C; ¹H NMR: δ 8.3 (s, 1H, CH=), 7.2, 7.0 (s, 2H, Ph), 4.6 (t, 2H, OCOCH₂), 3.8 (t, 2H, OCH₂), 3.4 (s, 3H, OCH₃); ¹³C NMR: δ 161 (C=O), 159 (HC=), 18, 129, 109 (Ph), 117 (CN), 101 (C=), 70 (OCH₂), 66 (OCOCH₂), 62 (OCH₃); IR: (cm⁻¹) 2905 (m, C-H), 2230 (m, CN), 1738 (s, C=O), 1626 (s, C=C), 1257 (s, C-O-CH₃), 919, 852, 754 (s, C-H out of plane). Anal. calcd. for C₁₃H₁₀ClF₂NO₃: C, 51.76; H, 3.34; N, 4.64; Found: C, 50.62; H, 3.50; N, 5.13.

3.1.8. 2-Methoxyethyl 2-hydroxy-3,5-diiodophenylcyanoacrylates

Yield 82%; ¹H NMR: δ 9.7 (s, 1H, OH), 8.4 (s, 1H, CH=), 8.2-7.4 (s, 2H, Ph), 4.4 (t, 2H, OCOCH₂), 3.8 (t, 2H, OCH₂), 3.4 (s, 3H, OCH₃); ¹³C NMR: δ 163 (C=O), 159 (HC=), 149, 128, 107 (Ph), 116 (CN), 102 (C=), 70 (OCH₂), 65 (OCOCH₂), 62 (OCH₃); IR: (cm⁻¹) 2934 (m, C-H), 2223 (m, CN), 1718 (s, C=O), 1617 (s, C=C), 1234 (s, C-O-CH₃), 815, 743 (s, C-H out of plane). Anal. calcd. for C₁₃H₁₁I₂NO₆: C, 31.29; H, 2.22; N, 2.81; Found: C, 29.74; H, 1.97; N, 3.33.

3.1.9. 2-Methoxyethyl 2,3,6-trichlorophenylcyanoacrylate

Yield 75%; mp 85.°C; ¹H NMR: δ 8.3 (s, 1H, CH=), 7.7-7.2 (m, 2H, Ph), 4.5 (t, 2H, OCOCH₂), 3.8 (t, 2H, OCH₂), 3.5 (s, 3H, OCH₃); ¹³C NMR: δ 162 (C=O), 151 (HC=), 134, 132, 129 (Ph), 116 (CN), 101 (C=), 70 (OCH₂), 65 (OCOCH₂), 62 (OCH₃); IR: (cm⁻¹)

¹) 2937 (m, C-H), 2235 (m, CN), 1738 (s, C=O), 1630 (s, C=C), 1202 (s, C-O-CH₃), 816, 764 (s, C-H out of plane). Anal. calcd. for C₁₃H₁₀Cl₃NO₃: C, 46.67; H, 3.01; N, 4.19; Found: C, 45.14; H, 3.14; N, 4.49.

3.1.10. 2-Methoxyethyl 2,3,4-trifluorophenylcyanoacrylate

Yield 89%; mp 84.6°C; ¹H NMR: δ 8.5 (s, 1H, CH=), 8.3-7.1 (s, 2H, Ph), 4.5 (t, 2H, OCOCH₂), 3.8 (t, 2H, OCH₂), 3.5 (s, 3H, OCH₃); ¹³C NMR: δ 161 (C=O), 156 (HC=), 145, 142, 139, 114 (Ph), 116 (CN), 106 (C=), 70 (OCH₂), 65 (OCOCH₂), 62 (OCH₃); IR: (cm⁻¹) 2970 (m, C-H), 2224 (m, CN), 1736 (s, C=O), 1620 (s, C=C), 1257 (s, C-O-CH₃), 870, 762 (s, C-H out of plane). Anal. calcd. for C₁₃H₁₀F₃NO₃: C, 54.74; H, 3.53; N, 4.91; Found: C, 50.63; H, 3.52; N, 5.16.

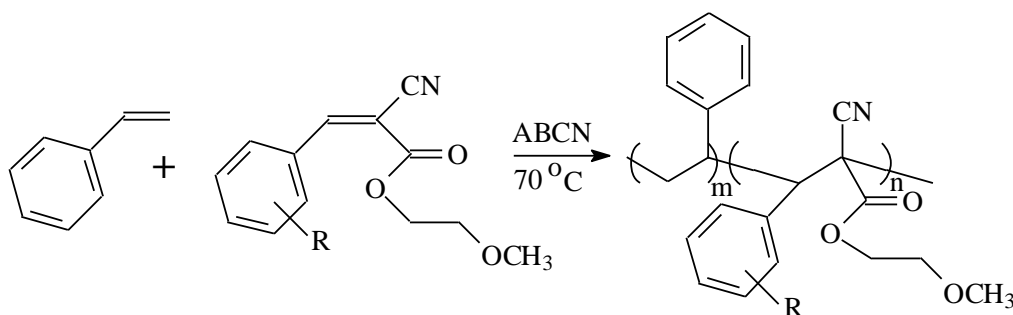
3.1.11. 2-Methoxyethyl 2,4,6-trifluorophenylcyanoacrylate

Yield 71%; mp 70.9°C; ¹H NMR: δ 8.2 (s, 1H, CH=), 7.3-6.8 (s, 2H, Ph), 4.5 (t, 2H, OCOCH₂), 3.8 (t, 2H, OCH₂), 3.5 (s, 3H, OCH₃); ¹³C NMR: δ 162 (C=O), 160 (HC=), 143, 142, 114 (Ph), 116 (CN), 102 (C=), 70 (OCH₂), 66 (OCOCH₂), 59 (OCH₃); IR: (cm⁻¹) 2914 (m, C-H), 2233 (m, CN), 176 (s, C=O), 1623 (s, C=C), 1247 (s, C-O-CH₃), 849, 758 (s, C-H out of plane). Anal. calcd. for C₁₃H₁₀F₃NO₃: C, 54.74; H, 3.53; N, 4.91; Found: C, 52.08; H, 3.83; N, 5.06.

3.3. Synthesis and characterization of styrene – MEPA copolymers

Copolymers of the ST and the MEPA compounds, P(ST-co-MEPA) were prepared in 25-mL glass screw cap vials at ST/MEPA = 3 (mol) the monomer feed using 0.12 mol/L of

ABCN at an overall monomer concentration 2.44 mol/L in 10 mL of toluene. The copolymerization was conducted at 70°C. After a predetermined time, the mixture was cooled to room temperature, and precipitated dropwise in methanol. The composition of the copolymers was determined based on the nitrogen content (cyano group in MEPA monomers). The novel synthesized MEPA compounds copolymerized readily with ST under free-radical conditions (Scheme 2) forming white flaky precipitates when their solutions were poured into methanol. The conversion of the copolymers was kept between 10 and 20% to minimize compositional drift (Table 1).



Scheme 2. Copolymerization of Styrene and ring-trisubstituted 2-methoxyethyl phenylcyanoacrylates, $R\text{PhCH} = \text{C}(\text{CN})\text{CO}_2\text{CH}_2\text{CH}_2\text{OCH}_3$, where R is 5-chloro-2,3-dimethoxy, 3-bromo-4,5-dimethoxy, 5-bromo-2,3-dimethoxy, 2-bromo-3-hydroxy-4-methoxy, 4-bromo-2,6-difluoro, 3-chloro-2,6-difluoro, 4-chloro-2,6-difluoro, 2-hydroxy-3,5-diiodo, 2,3,6-trichloro, 2,3,4-trifluoro, 2,4,6-trifluoro.

Table 1. Copolymerization of Styrene and 2-Methoxyethyl phenylcyanoacrylates.

R	Yield ^a (wt%)	N (wt%)	ST in copol. (mol%)	MEPA in copol. (mol%)
5-Chloro-2,3-dimethoxy	13.4	2.66	65.8	34.2
3-Bromo-4,5-dimethoxy	14.7	2.52	64.1	35.9
5-Bromo-2,3-dimethoxy	11.3	2.6	61.8	38.2
2-Bromo-3-hydroxy-4-methoxy	15.2	2.15	73.9	26.1
4-Bromo-2,6-difluoro	12.1	2.42	69.1	30.9
3-Chloro-2,6-difluoro	14.2	2.69	67.7	32.3
4-Chloro-2,6-difluoro	15.2	2.82	65.2	34.8
2-Hydroxy-3,5-Diiodo	13.0	1.17	87.0	13.0
2,3,6-Trichloro	12.3	1.91	79.3	20.7
2,3,4-Trifluoro	10.2	2.59	71.1	28.9
2,4,6-Trifluoro	16.3	2.65	70.1	29.9

Nitrogen elemental analysis showed that between 13.0 and 38.2 mol% of MEPA is present in the copolymers prepared at ST/MEPA = 3 (mol), which is indicative of relatively high reactivity of the MEPA monomers towards ST radical which is typical of ring-substituted phenylcyanoacrylates. Since MEPA monomers do not homopolymerize, the most likely structure of the copolymers would be isolated MEPA monomer units alternating with short ST sequences (Scheme 2).

The copolymers prepared in the present work are all soluble in ethyl acetate, THF, DMF and CHCl_3 and insoluble in methanol, ethyl ether, and petroleum ether.

4 Conclusions

Novel halogen ring-trisubstituted 2-methoxyethyl phenylcyanoacrylates,

$\text{RPhCH}=\text{C}(\text{CN})\text{CO}_2\text{CH}_2\text{CH}_2\text{OCH}_3$ (where R is 5-chloro-2,3-dimethoxy, 3-bromo-4,5-

dimethoxy, 5-bromo-2,3-dimethoxy, 2-bromo-3-hydroxy-4-methoxy, 4-bromo-2,6-difluoro,

3-chloro-2,6-difluoro, 4-chloro-2,6-difluoro, 2-hydroxy-3,5-diiodo, 2,3,6-trichloro, 2,3,4-trifluoro, 2,4,6-trifluoro) were prepared and copolymerized with styrene.

Acknowledgments

The authors are grateful to acknowledge that the project was partly supported by Chicago Society of Coatings Technology.

References

1. Biochemical evaluation of virtual screening methods reveals a cell-active inhibitor of the cancer-promoting phosphatases of regenerating liver. Hoeger, Birgit; Diether, Maren; Ballester, Pedro J.; Koehn, Maja. *European Journal of Medicinal Chemistry* (2014), 88, 89-100.
2. Large, chemically diverse dataset of log P measurements for benchmarking studies. Martel, Sophie; Gillerat, Fabrice; Carosati, Emanuele; Maiarelli, Daniele; Tetko, Igor V.; Mannhold, Raimund; Carrupt, Pierre-Alain. *European Journal of Pharmaceutical Sciences* (2013), 48(1-2), 21-29.
3. Discovery of 4-aryl-2-oxo-2H-chromenes as a new series of apoptosis inducers using a cell- and caspase-based high-throughput screening assay. Kemnitzer, William; Jiang, Songchun; Zhang, Hong; Kasibhatla, Shailaja; Crogan-Grundy, Candace; Blais, Charles; Attardo, Giorgio; Denis, Real; Lamothe, Serge; Gourdeau, Henriette; et al. *Bioorganic & Medicinal Chemistry Letters* (2008), 18(20), 5571-5575.

4. Preparation of substituted coumarins and quinolinones as caspase activators for treatment of cancer. Cai, Sui Xiong; Zhang, Hong; Kemmitzer, William E.; Jiang, Songchun; Drewe, John A.; Storer, Richard. PCT Int. Appl. (2002), WO 2002092076 A1 20021121.
5. The Mitsunobu reaction. Hughes, David L. Organic Reactions (Hoboken, NJ, United States) (1992), 42.
6. Studies culminating in the total synthesis of (dl)-morphine. Toth, J. E.; Hamann, P. R.; Fuchs, P. L. Journal of Organic Chemistry (1988), 53(20), 4694-708.
7. Tandem intramolecular conjugate addition/intramolecular alkylation reactions of substituted vinyl sulfones. Hamann, P. R.; Toth, J. E.; Fuchs, P. L. Journal of Organic Chemistry (1984), 49(20), 3865-7.
8. Electron transport inhibition of the cytochrome bc1 complex of rat-liver mitochondria by phenolic uncouplers. Tokutake, Nobuya; Miyoshi, Hideto; Fujita, Toshio. From Biochimica et Biophysica Acta, Bioenergetics (1991), 1057(3), 377-83.
9. Synthesis of Isoxazolines and Isoxazoles Inspired by Fipronil. Miller, Daniel K.; Bailey, Christopher A.; Sammelson, Robert E. Synthesis (2015), 47(18), 2791-2798.
10. Infrared group frequency correlations for styrenes, α -methylstyrenes, and related compounds. Nyquist, R. A. Applied Spectroscopy (1986), 40(2), 196-203.
11. Method for preparing ethylene-fluorostyrene copolymer. Cui, Dongmei; Wang, Tiantian; Wu, Chunji. Faming Zhuanli Shenqing (2021), CN 113136001 A 20210720.
12. Liquid crystal compound, liquid crystal composition and liquid crystal display device and their preparation. Gotoh, Yasuyuki; Kobayashi, Masahide. U.S. Pat. Appl. Publ. (2015), US 20150368272 A1 20151224.

13. Palladium-catalyzed intermolecular aminofluorination of styrenes. Qiu, Shuifa; Xu, Tao; Zhou, Juan; Guo, Yinlong; Liu, Guosheng. *Journal of the American Chemical Society* (2010), 132(9), 2856-2857.
14. Process for preparing styrene derivatives by a transition metal-catalyzed cross coupling of chlorostyrenes with organomagnesium compounds. Gotta, Matthias; Lehnemann, Bernd Wilhelm; Von Wangelin Jacobi, Axel; Guelak, Samet. U.S. Pat. Appl. Publ. (2013), US 20130324745 A1 20131205.
15. Chlorostyrenes in Iron-Catalyzed Biaryl Coupling Reactions. Guelak, Samet; von Wangelin, Axel Jacobi. *Angewandte Chemie, International Edition* (2012), 51(6), 1357-1361, S1357/1-S1357/36.
16. SciFunder, Structure search Apr 15, 2022.
17. Synthesis and styrene copolymerization of novel trisubstituted ethylenes: 1. Alkyl ring-substituted 2-methoxyethyl phenylcyanoacrylates Maddy E. Ablan, Samer A. Abuelroos, Ryan C. Arthur, Sonya Balaji, Kimberly L. Burns, Ivana A. Chychula, Kayla L. Corcoran, Yangfei Deng, Yelena Gritsaeva, Ana K. Hernandez, Sara M. Rocus, William S. Schjerven, and Gregory B. Kharas. ChemRxiv Version 1, Nov 22, 2020. <https://doi.org/10.26434/chemrxiv.13262660.v1>
18. Smith, M. B.; March, J. Addition to Carbon-Hetero Multiple Bonds, In March's *Advanced Organic Chemistry*, J. Wiley & Sons: New York, Ch.16, 1225, 2001.