Synthesis and styrene copolymerization of novel dibromo and dichloro ring-disubstituted isobutyl phenylcyanoacrylates

Daniah H. Ibrahim, Chris J. Lee, Daniel N. Madock, Marcin Marciniak, Miguel A. Montero, Juan Orozco, Viany Ortiz, Mica C. Robles, Sara M. Rocus, William S. Schjerven and Gregory B. Kharas

DePaul University, Chemistry and Biochemistry Department, 1110 West Belden Avenue, Chicago, IL 60614-3214

ABSTRACT

Novel dibromo and dichloro ring-disubstituted isobutyl phenylcyanoacrylates, RPhCH=C(CN)CO₂CH₂CH(CH₃)₂ (where R is 2,5-dibromo, 3,5-dibromo, 2,3-dichloro, 2,4dichloro, 2,5-dichloro, 2,6-dichloro, 3,4-dichloro, 3,5-dichloro) were synthesized by the piperidine catalyzed Knoevenagel condensation of ring-disubstituted benzaldehydes and isobutyl cyanoacetate and characterized by CHN analysis, IR, ¹H and ¹³C NMR. The acrylates were copolymerized with styrene in solution with radical initiation at 70°C. The compositions of the copolymers were calculated from nitrogen analysis.

^{*} Contact: gkharas@depaul.edu

1. Introduction

Dibromo ring-substituted phenylcyanoacrylates (PCA) were reported in a variety of applications. Thus, 3,5-dibromo ring-substituted PCA was mentioned in syntheses of (1) spiro[2.3]hexanes from methylenecyclopropane and cyanoalkenes catalyzed by a Tin-ate complex [1]; (2) phenylsuccinimide derivatives with anticonvulsant properties [2, 3]. 3,4dichloro ring-substituted PCA was prepared by Knoevenagel condensation at room temperature using SeO_2/ZrO_2 catalyst in water-medium and solvent-free conditions [4]. It was used in a highly efficient protocol for the regio- and stereo-selective synthesis of spiro pyrrolidine and pyrrolizidine derivatives by multicomponent reaction [5] and in Knoevenagel condensation involving a PEG bridged tertiary amine functionalized ionic liquid that exhibited thermoregulated reversible biphasic behavior with cyclohexane /isopropanol mixtures [6], as well as in synthesis of 5,6-dichloroindan-1-acids and their tetrazolyl derivatives as analgesic and anti-inflammatory agents [7]. Other applications of this PCA include synthesis of 1,8-diazabicyclo[-5.4.0]-undec-7-ene by catalyzed Knoevenagel condensation of aromatic aldehydes with active methylene compounds under solvent-free conditions [8], preparation of pyrimidine derivatives as CXCR2 receptor antagonists [9], synthesis of 2-amino-4-aryl-5,6-dihydro-4H-pyrano[3,2-c]quinolin-5-one derivatives in water [10], synthesis of new pyridine, pyridone, pyrazole, thiophene, fused pyrimidine and triazine derivatives via β -amino- β -(pyrid-4-yl)acrylonitrile [11], synthesis of dibenzo[c,f]chromenes, dibenzo[c,h]chromenes and benzo[7,8]chromeno[3,4-f]isoindoles as antimicrobial agents [12], synthesis and studies of antimicrobial activity of substituted imidazolidinediones and thioxoimidazolidinones [13], in studies of neuroleptic activity and

dopamine-uptake inhibition in 1-piperazino-3-phenylindans [14], and in synthesis of 2methoxy-5-chloro-, 3-chloro-4-methoxy-, and 3,4-dichlorophenylsuccinic acids [15]. Earlier we have reported synthesis and styrene copolymerization a number of dibromo and dichloro ring-substituted PCAs, such esters as methyl [16, 17], ethyl [18], propyl [19, 20], isopropyl [21], butyl [22], 2-methoxyethyl [23], and octyl [24].

In this work we have prepared novel dibromo and dichloro ring-disubstituted isobutyl PCA, RPhCH=C(CN)CO₂CH₂CH(CH₃)₂, where R is 2,5-dibromo, 3,5-dibromo, 2,3-dichloro, 2,4dichloro, 2,5-dichloro, 2,6-dichloro, 3,4-dichloro, 3,5-dichloro, and explored the feasibility of their copolymerization with styrene. To the best of our knowledge, there have been no reports on either synthesis of these compounds, nor their copolymerization with styrene [25].

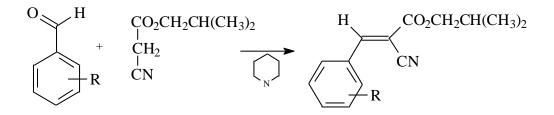
2. Experimental

2,5-Dibromo, 3,5-dibromo, 2,3-dichloro, 2,4-dichloro, 2,5-dichloro, 2,6-dichloro, 3,4dichloro, 3,5-dichloro benzaldehydes, isobutyl cyanoacetate, piperidine, styrene, 1,1'azobis(cyclohexanecarbonitrile) (ABCN), and toluene supplied from Sigma-Aldrich Co., were used as received. Instrumentation is reported in [26].

3. Results and discussion

3.1. Synthesis and characterization of isobutyl phenylcyanoacrylates

All isobutyl phenylcyanoacrylates (IPCA) compounds were synthesized by Knoevenagel condensation [27] of appropriate benzaldehydes with isobutyl cyanoacetate, catalyzed by base, piperidine (Scheme 1).



Scheme 1. Synthesis of isobutyl R-phenylcyanoacrylates, where R is 2,5-dibromo, 3,5dibromo, 2,3-dichloro, 2,4-dichloro, 2,5-dichloro, 2,6-dichloro, 3,4-dichloro, 3,5-dichloro.

The preparation procedure was essentially the same for all the monomers. In a typical synthesis, equimolar amounts of isobutyl cyanoacetate and an appropriate benzaldehyde were mixed in equimolar ratio in a 20 mL vial. A few drops of piperidine were added with stirring. The reactions was allowed to proceed 48 hrs at r.t. The product of the reaction was isolated by filtration and purified by crystallization from 2-propanol. The condensation reaction proceeded smoothly, yielding products, which were purified by conventional techniques. Melting points of the compounds in crystalline state were measured by DSC. The compounds were characterized by IR, ¹H and ¹³C NMR spectroscopies. No stereochemical analysis of the novel ring-substituted IPCA was performed since no stereoisomers (*E* or/and *Z*) of known configuration were available.

3.1.1. Isobutyl 2,5-dibromophenylcyanoacrylate

Yield: 70.3%; mp 86.4°C; ¹H NMR: δ 8.5 (s, 1H, CH=), 7.9-7.2 (m, 3H, Ph), 4.1 (d, 2H, CH₂), 2.0 (m, 1H, CH), 1.0 (d, 6H, CH₃); ¹³C NMR: δ 162 (C=O), 151 (HC=), 137, 134,

131, 124, 120 (Ph), 116 (CN), 101 (C=), 72 (CH₂), 28 (CH), 17 (CH₃); IR: (cm⁻¹) 2873 (m, C-H), 2232 (m, CN), 1734 (s, C=O), 1686 (s, C=C), 1248 (s, C-O-CH₃), 824 (s, C-H out of plane). Anal. calcd. for C₁₄H₁₃Br₂NO₂: C, 43.44; H, 3.39; N, 3.62; Found: C, 45.12; H, 3.26; N, 3.66.

3.1.2. Isobutyl 3,5-dibromophenylcyanoacrylate

Yield 71.4%; mp 122.9°C; ¹H NMR: δ 8.2 (s, 1H, CH=), 8.1-7.6 (m, 3H, Ph), 4.1 (d, 2H, CH₂), 2.1 (m, 1H, CH), 1.0 (d, 6H, CH₃); ¹³C NMR: δ 162 (C=O), 151 (HC=), 136, 134, 132, 128 (Ph), 115 (CN), 108 (C=), 73 (CH₂), 28 (CH), 19 (CH₃)₂; IR: (cm⁻¹) 3147-2868 (m, C-H), 2224 (m, CN), 1718 (s, C=O), 1612 (s, C=C), 1271 (s, C-O-CH₃), 883 (s, C-H out of plane). Anal. calcd. for C₁₄H₁₃Br₂NO₂: C, 43.44; H, 3.39; N, 3.62; Found: C, 40.91; H, 3.13; N, 2.84.

3.1.3. Isobutyl 2,3-dichlorophenylcyanoacrylate

Yield 82.8%; mp 37.2°C; ¹H NMR: δ 8.8 (s, 1H, CH=), 8.1-7.1 (m, 3H, Ph), 4.2 (d, 2H, CH₂), 2.1 (m, 1H, CH), 1.0 (d, 6H, CH₃); ¹³C NMR: δ 162 (C=O), 151 (HC=), 136, 134, 132, 128 (Ph), 115 (CN), 108 (C=), 73 (CH₂), 28 (CH), 19 (CH₃)₂; IR: (cm⁻¹) 2962 (m, C-H), 2230 (m, CN), 1730 (s, C=O), 1638 (s, C=C), 1281 (s, C-O-CH₃), 822, 751 (s, C-H out of plane). Anal. calcd. for C₁₄H₁₃Cl₂NO₂: C, 56.40; H, 4.39; N, 4.70; Found: C, 55.59; H, 4.42; N, 4.47.

3.1.4. Isobutyl 2,4-dichlorophenylcyanoacrylate

Yield 88%; mp 83.1°C; ¹H NMR δ 8.7 (s, 1H, CH=), 8.2-7.2 (m, 3H, Ph), 4.1 (d, 2H, CH₂), 2.0 (m, 1H, CH), 1.0 (d, 6H, CH₃); ¹³C NMR δ 162 (C=O), 149 (HC=), 140, 138,

131, 128 (Ph), 115 (CN), 107 (C=), 73 (CH₂), 28 (CH), 19 (CH₃); IR (cm⁻¹): 2961 (m, C-H), 2225 (m, CN), 1728 (s, C=O), 1612 (s, C=C), 1286 (s, C-O-CH₃), 782, 755 (s, C-H out of plane). Anal. Calcd. for C₁₄H₁₃Cl₂NO₂: C, 56.40; H, 4.39; N, 4.70; Found: C, 56.32; H, 4.39; N, 4.78.

3.1.5. Isobutyl 2,5-dichlorophenylcyanoacrylate

Yield 70.5%; mp 46.7°C; ¹H NMR: δ 8.5 (s, 1H, CH=), 8.2-7.2 (m, 3H, Ph), 4.1 (d, 2H, CH₂), 2.1 (m, 1H, CH), 1.0 (d, 6H, CH₃); ¹³C NMR: δ 162 (C=O), 149 (HC=), 134, 132, 130, 114 (Ph), 114 (CN), 107 (C=), 73 (CH₂), 28 (CH), 19 (CH₃); IR: (cm⁻¹) 2966 (m, C-H), 2220 (m, CN), 1728 (s, C=O), 1587 (s, C=C), 1271 (s, C-O-CH₃), 821, 758 (s, C-H out of plane). Anal. calcd. for C₁₄H₁₃Cl₂NO₂: C, 56.40; H, 4.39; N, 4.70; Found: C, 56.40; H, 4.44; N, 4.82.

3.1.6. Isobutyl 2,6-dichlorophenylcyanoacrylate

Yield 72.7%; mp 65.5°C; ¹H NMR: δ 8.3 (s, 1H, CH=), 7.2 (s, 3H, Ph), 4.1 (d, 2H, CH₂), 2.1 (m, 1H, CH), 1.0 (d, 6H, CH₃); ¹³C NMR: δ 161 (C=O), 151 (HC=), 134, 134, 131, 129 (Ph), 116 (CN), 103 (C=), 73 (CH₂), 27 (CH), 19 (CH₃)₂; IR: (cm⁻¹) 2967 (m, C-H), 2233 (m, CN), 1732 (s, C=O), 1614 (s, C=C), 1232 (s, C-O-CH₃), 781 (s, C-H out of plane). Anal. calcd. for C₁₄H₁₃Cl₂NO₂: C, 56.40; H, 4.39; N, 4.70; Found: C, 56.87; H, 4.51; N, 4.74.

3.1.7. Isobutyl 3,4-dichlorophenylcyanoacrylate

Yield 74%; mp 157.2°C; ¹H NMR δ 8.3 (s, 1H, CH=), 8.1-7.2 (m, 3H, Ph), 4.2 (d, 2H, CH₂), 2.1 (m, 1H, CH), 1.0 (d, 6H, CH₃); ¹³C NMR δ 162 (C=O), 152 (HC=), 137, 134,

133, 132, 130 (Ph), 115 (CN), 105 (C=), 73 (CH₂), 28 (CH), 19 (CH₃)₂; IR (cm⁻¹): 3532-2873 (m, C-H), 2221 (m, CN), 1718 (s, C=O), 1614 (s, C=C), 1227 (s, C-O-CH₃), 811, 743 (s, C-H out of plane). Anal. Calcd. for C₁₄H₁₃Cl₂NO₂: C, 56.40; H, 4.39; N, 4.70; Found: C, 56.27; H, 4.39; N, 4.73.

3.1.8. Isobutyl 3,5-dichlorophenylcyanoacrylate

Yield 81%; mp 117.1°C; ¹H NMR δ 8.1 (s, 1H, CH=), 8.0-7.2 (s, 3H, Ph), 4.1 (s, 2H, CH₂), 2.0 (m, 1H, CH), 1.0 (d, 6H, (CH₃)₂; ¹³C NMR δ 162 (C=O), 152 (HC=), 136, 134, 133, 129 (Ph), 114 (CN), 106 (C=), 73 (CH₂), 28 (CH), 19 (CH₃)₂; IR (cm⁻¹): 2972 (m, C-H), 2228 (m, CN), 1720 (s, C=O), 1574 (s, C=C), 1271 (s, C-O-CH₃), 841, 759 (s, C-H out of plane). Anal. Calcd. for C₁₄H₁₃Cl₂NO₂: C, 56.40; H, 4.39; N, 4.70; Found: C, 56.19; H, 4.46; N, 5.13.

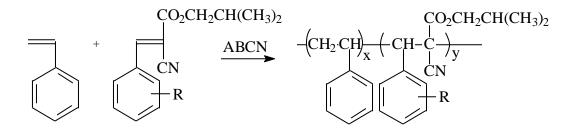
3.2. Synthesis and characterization of styrene – IPCA copolymers

Copolymers of the styrene (ST) and the IPCA compounds, P(ST-co-IPCA) were prepared in 25-mL glass screw cap vials at ST/IPCA = 3 (mol) the monomer feed using 0.12 mol/L of ABCN at an overall monomer concentration 2.44 mol/L in 10 mL of toluene. The copolymerization was conducted at 70°C. After a predetermined time, the mixture was cooled to room temperature, and precipitated dropwise in methanol. The composition of the copolymers was determined based on the nitrogen content. The novel synthesized IPCA compounds copolymerized readily with ST under free-radical conditions (Scheme 2) forming white flaky precipitates when their solutions were poured into methanol. The conversion of the copolymers was kept between 10 and 20% to minimize compositional drift (Table 1).

			ST in	IPCA in
	Yield ^a	Ν	copol.	copol.
R	(wt%)	(wt%)	(mol%)	(mol%)
2,5-Dibromo	13.6	1.20	88.2	11.8
3,5-Dibromo	10.8	1.76	79.7	20.3
2,3-Dichloro	11.5	2.54	70.9	29.1
2,4-Dichloro	13.8	2.81	65.8	34.2
2,5-Dichloro	14.5	2.55	70.7	29.3
2,6-Dichloro	11.9	1.44	86.6	13.4
3,4-Dichloro	12.5	2.46	72.3	27.7
3,5-Dichloro	11.6	2.61	69.6	30.4

Table 1. Copolymerization of isobutyl phenylcyanoacrylates with styrene.

Nitrogen elemental analysis showed that between 11.8 and 30.4 mol% of IPCA is present in the copolymers, which is indicative of relatively high reactivity of the IPCA monomers towards ST radical which is typical of different esters of PCA. Since IPCA monomers do not homopolymerize, the most likely structure of the copolymers would be isolated IPCA monomer (y = 1) units alternating with short ST sequences (x > 1) (Scheme 2).



Scheme 2. Copolymerization of ST and the ring-substituted isobutyl phenylcyanoacrylates, RPhCH = C(CN)CO₂CH₂CH(CH₃)₂, R = 2,5-dibromo, 3,5-dibromo, 2,3-dichloro, 2,4-dichloro, 2,5-dichloro, 2,6-dichloro, 3,4-dichloro, 3,5-dichloro.

The copolymers prepared in the present work are all soluble in ethyl acetate, THF, DMF and CHCl₃ and insoluble in methanol, ethyl ether, and petroleum ether.

4 Conclusions

Novel dibromo and dichloro and ring-disubstituted isobutyl phenylcyanoacrylates, RPhCH=C(CN)CO₂CH₂CH(CH₃)₂ (where R is 2,5-dibromo, 3,5-dibromo, 2,3-dichloro, 2,4dichloro, 2,5-dichloro, 2,6-dichloro, 3,4-dichloro, 3,5-dichloro) were synthesized and copolymerized with styrene. The compositions of the copolymers were calculated from nitrogen analysis.

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