

A Shelf Stable Fmoc Hydrazine Resin for the Synthesis of Peptide Hydrazides

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Abstract

C-terminal hydrazides are an important class of synthetic peptides with an ever expanding scope of applications, but their widespread application for chemical protein synthesis has been hampered due to the lack of stable resin linkers for synthesis of longer and more challenging peptide hydrazide fragments. We present a practical method for the regeneration, loading, and storage of trityl-chloride resins for the production of hydrazide containing peptides, leveraging 9-fluorenylmethyl carbazate. We show that these resins are extremely stable under several common resin storage conditions. The application of these resins to solid phase peptide synthesis (SPPS) is demonstrated through the synthesis of the 40-mer GLP-1R agonist peptide "P5". These studies support the broad utility of Fmoc-NHNH-Trt resins for SPPS of C-terminal hydrazide peptides.

Introduction

The hydrazide moiety has been ubiquitous in the field of peptide chemistry for more than fifty years,¹⁻¹⁷ primarily due to its utility as a precursor to acyl azides, used for amide couplings^{1-13, 18} and thioester formation¹⁴⁻¹⁷ to enable Native Chemical Ligation (NCL).¹⁹ Our group recently published mild conditions for formation of peptide α -thioesters from hydrazides via the Knorr pyrazole synthesis.¹⁷ This method avoids the need for strong oxidizing conditions, enabling broader compatibility with common synthetic peptide functionalities, and has been widely adopted in a broad range of applications.²⁰⁻³⁷

The increased use of C-terminal α -hydrazides in peptide chemistry creates a demand for robust synthetic tools for accessing these moieties. A number of methods have been published to allow access to C-terminal hydrazides on both synthetic^{15, 16, 38} and expressed^{15, 39, 40} peptides and proteins. The current synthetic methods suffer from one of two drawbacks. They either require a two part deprotection procedure involving hydrazinolysis followed by global deprotection,³⁸ or rely on time-of-use formation of hydrazine resins from trityl chloride resins,

where the hydrazine form is not shelf stable.¹⁶ The chloride form may also need to be regenerated prior to use, making SPPS of peptide hydrazides significantly less convenient than standard amides or acids.⁴¹⁻⁴³

Chemical protein synthesis requires the solid phase synthesis of long and hydrophobic peptide fragments, which has been addressed through the use of higher swelling polyethylene glycol (PEG) containing resins.⁴⁴⁻⁴⁸ When using PEGylated trityl chloride resins (e.g. TentaGel[®] and ChemMatrix[®]), the yields of synthetic peptides are often significantly lower than expected and decrease over time after resins are first opened.⁴⁹ In addition, while there has been

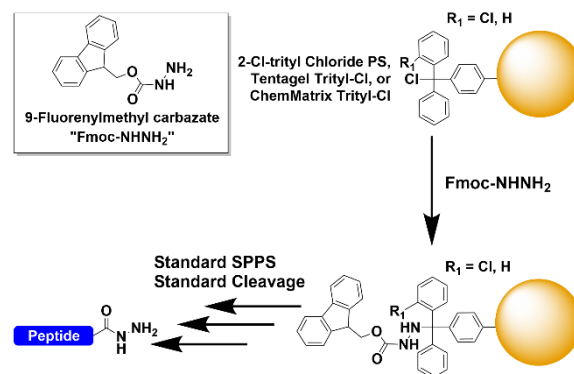


Figure 1: Proposed new strategy for preparation of peptide hydrazides using a stable Fmoc-NHNH₂ loaded resin precursor.

a move toward increased automation and real-time monitoring of peptide synthesis and purification workflows, regenerating and hydrazine loading a trityl chloride resin prior to each synthesis represents a manual process that must be performed in a fume hood, and cannot be monitored other than by initiation of synthesis. We therefore sought to develop a procedure for regenerating and loading trityl chloride resins that, upon receipt from the manufacturer, would allow for immediate loading evaluation and long term stable storage in a state that is competent for automated synthesis of peptide α -hydrazides without any extraneous steps. 9-fluorenylmethyl carbazate (Fmoc-NHNH₂) was identified as an ideal reagent for preparation and long term storage of peptide-hydrazide synthesis resins (Figure 1), and the relative stability and regenerability of TentaGel[®] resins loaded with Fmoc-NHNH₂ as compared to chloride or hydrazine was explored.

Materials and Methods

2.1 Materials

N,N-Dimethylformamide (DMF), dichloromethane (DCM), acetonitrile (ACN), Oxyma Pure, ethanedithiol (EDT), 2-methyltetrahydrofuran (2-MeTHF), and trifluoroacetic acid (TFA) were purchased from Sigma-Aldrich. Diisopropylethylamine (DIEA), 1-[bis(dimethylamino)methylene]-1H-1,2,3-triazolo[4,5-b]pyridinium-3-oxidhexafluorophosphate (HATU), diisopropylcarbodiimide (DIC), Fmoc-NHNH₂, and triisopropylsilane (TIS) were purchased from Oakwood Chemical. 2'-Chlorotrylchloride polystyrene resin (200-400 mesh) and all standard Fmoc-protected amino acids were purchased from Bachem Americas. 4-Methylpiperidine (4-MePip) was purchased from TCI America. Dimethyl sulfide (DMS) and 4-dimethylaminopyridine (DMAP) were purchased from AlfaAesar. TentaGel[®] XV trityl chloride resin (XV18130.031) was purchased from Rapp

Polymere. Trityl-OH ChemMatrix[®] (7-600-1310) was purchased from Biotage. Diethyl ether (Et₂O) and septa capped vials for storage under inert gas were purchased from Fisher Scientific. Methyl tert-butyl ether (MTBE) was purchased from VWR.

2.2 Chloride regeneration and Fmoc-NHNH₂ loading

TentaGel[®] XV Trityl Chloride was swollen in minimal DCM within a round bottom flask fitted with a rubber septum or a 40 mL glass vial fitted with a 0.125" septa cap. The headspace was purged and backfilled with nitrogen. 5 equiv of thionyl chloride, calculated from manufacturer reported resin loading, was added and the reaction was stirred under nitrogen for 2.5 h. The slurry was transferred to a fritted syringe and washed with DCM then DMF. The resin was treated with 5 equiv of Fmoc-NHNH₂ suspended in 50:50 DCM:DMF to barely cover the swelled resin and stirred for 45 min. The resin was drained and rinsed with DMF, then the Fmoc-NHNH₂ treatment was repeated. The resin was flow washed with DMF and DCM and dried by flow washing with Et₂O over a vacuum manifold before storing in a vacuum desiccator for at least 8 h prior to performing loading tests.

2.3 Fmoc Loading Tests

Fmoc Loading Tests were performed in triplicate by taking single absorbance measurement on three independently weighed and treated samples. Approximately 4 μ moles of resin was weighed into a 1.7 mL microcentrifuge tube. For example, for a TentaGel[®] XV resin with an estimated loading of 0.2 mmol/g, 20 mg of resin were used. Each sample was suspended in 1 mL of 20% 4-MePip in DMF and sonicated to ensure full resin swelling, then mixed on a rotating shaker for 20 min. The tubes were centrifuged and 10 μ L of supernatant was diluted to 1 mL with DMF. The spectrometer was blanked with 0.2% 4-MePip in DMF and A₃₀₁ measurements were taken for each sample. Loadings were

calculated according to the formula $\frac{101(A_{301})}{7.8(w)}$

where w is the weight of the resin used in milligrams.⁵⁰ Reported loadings are the arithmetic mean of the triplicate loading tests \pm one standard deviation.

2.4 Peptide synthesis and cleavage

All peptides were synthesized by conventional Fmoc/tBu chemistry using either [5equiv amino acid]:[5 equiv HATU]:[8 equiv DIEA] or [5 equiv amino acid]:[5 equiv DIC]:[5 equiv Oxyma Pure] on a CSBio model CS336X Peptide Synthesizer. Peptides were cleaved from the resin in one of four cocktails according to the peptide composition:

Cocktail N: Peptides containing neither cysteine nor methionine

- 95% TFA
- 2.5% TIS
- 2.5% H₂O

Cocktail C: Peptides containing cysteine but not methionine

- 92.5% TFA
- 2.5% TIS
- 2.5% H₂O
- 2.5% EDT

Cocktail M: Peptides containing methionine but not cysteine

- 87.5% TFA
- 5.0% TIS
- 2.5% H₂O
- 5.0% DMS

Cocktail MC: Peptides containing methionine and cysteine

- 85% TFA
- 5.0% TIS
- 2.5% H₂O
- 2.5% EDT
- 5.0% DMS.

Cleavage reactions were carried out for 1 h at room temperature and TFA removed by nitrogen

flow or rotary evaporation. Crude peptide was recovered by precipitation in diethyl ether.

2.5 Peptide characterization and purification

Peptides were characterized on a Waters Acquity I-Class UPLC equipped with a diode array detector and a single-quadrupole mass spec (Waters SQD2). The analysis was performed on a Waters Cortecs C18 column (2.1x55 mm, 1.6 μ m, 90 Å) heated to 35 C with a 0.8 mL/min flow rate (see gradient info below). Peptides masses were manually deconvoluted using the experimental mass to charge ratios (m/z) from all the observed peptide protonation states by using the onboard analyst software packages.

Alternatively, proteins and longer peptides were characterized on a Waters Acquity I-Class UPLC equipped with a diode array detector and a time of flight (ToF) mass spec (Waters G2-XS). The analysis was performed on a Waters BEH C18 column (2.1x55 mm, 1.7 μ m, 300 Å) heated to 55 °C with a 0.4 mL/min flow rate (see gradient info below). Peptides masses were deconvoluted (MaxEnt1 algorithm in Waters MassLynx software) to a monoisotopic singly-charged mass. In both cases the mobile phases were:

- Buffer A: H₂O (0.1% Formic Acid)
- Buffer B: MeCN (0.06% Formic Acid)

Peptide purification was carried out by mass directed reversed phase HPLC (RP-HPLC) on a Waters 2545 Binary Gradient Module equipped with an Acquity QDa Detector and a Waters 2767 Sample Manager. The column was a Waters XBridge Prep C18 and the mobile phases were as above, with the exception that buffer B did not contain Formic Acid.

2.6 Storage temperature

The storage temperatures reported are based on readings taken from thermometers in resin storage freezers and refrigerators. Resins stored together in the same freezer were stored in the same part of the freezer to ensure similar conditions. Room temperature varied between 18 C and 22 C.

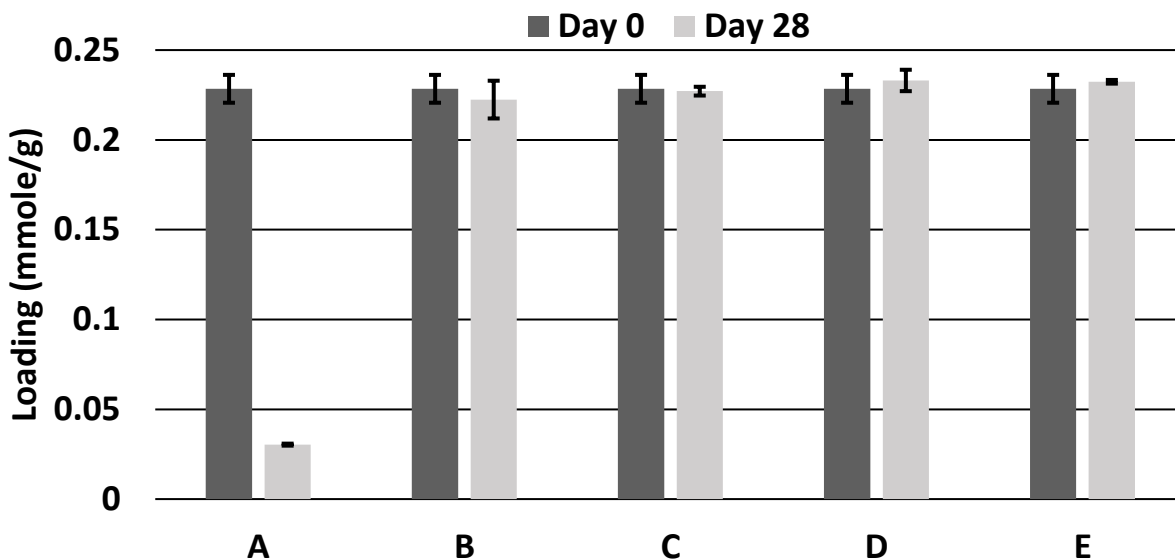


Figure 2 Retention of loading of Tentagel® XV resins stored as: (A) Cl-Trityl at -10 C sealed under N₂, (B) Fmoc-NHNH-Trityl at -10 C sealed under N₂, (C) Fmoc-NHNH-Trityl at -10 C sealed under air, (D) Fmoc-NHNH-Trityl at 3 C sealed under air, (E) Fmoc-NHNH-Trityl at RT open in a vacuum desiccator. All error bars show ± 1 standard deviation for the triplicate loading tests. Dark grey is day 0, light grey is day 28.

3 Results and Discussion

Encouraged by preliminary experiments on both Tentagel® XV Trityl and ChemMatrix® Trityl resins (Figure S1), a systematic analysis of loading retention of Fmoc-NHNH₂ on the Tentagel® XV was undertaken. To evaluate the stability of carbamate loaded trityl resins, Tentagel® XV trityl chloride with a manufacturer reported loading of 0.19 mmol/g was regenerated with thionyl chloride immediately following receipt from the manufacturer. A sample was taken for storage as the chloride, then the remainder was substituted with Fmoc-NHNH₂, and both fractions were dried thoroughly with Et₂O. The loading of this resin was measured to be 0.228 ± 0.008 mmol/g in a triplicate Fmoc loading test. The dry Fmoc-NHNH-Trityl resin was split into 4 equal fractions and stored in varying conditions for 28 days, alongside the chloride sample which was stored at -10°C under nitrogen atmosphere. The retention of loading was measured by first substituting the chloride resin with Fmoc-NHNH₂, then washing all 5 samples with DMF, DCM, and Et₂O to remove any Fmoc species not

bound to the resin. The resins were then dried from Et₂O under vacuum and stored in a vacuum desiccator for 8 h. prior to performing Fmoc-loading tests. Under all conditions tested the Fmoc-NHNH-Trityl resins had loadings that were indistinguishable from the day zero, while the chloride resin retained only $13.3\% \pm 0.3\%$ of its initial loading (Figure 2).

In order to establish whether this stability enhancement is unique to the Fmoc-NHNH₂ loaded resins as compared to previously reported hydrazide loaded resins¹⁶ sample D from Figure 1 (stored at 3 C under air for 28 days) was split in two equal portions and swelled in 1:1 DCM:DMF. One sample (D2) was then drained and deprotected with two, 15 min. treatments with 20% 4-methylpiperidine in DMF, while the other (D1) was left swollen in DCM:DMF. Both were washed thoroughly with DMF, DCM, and Et₂O before thoroughly drying and storing in a vacuum desiccator for a period of 7 days. Both resins were then swollen, D1 was deprotected, and Fmoc-glycine was coupled to each resin before thorough washing and drying. The Fmoc-protected resin (D1) retained a loading of 0.188

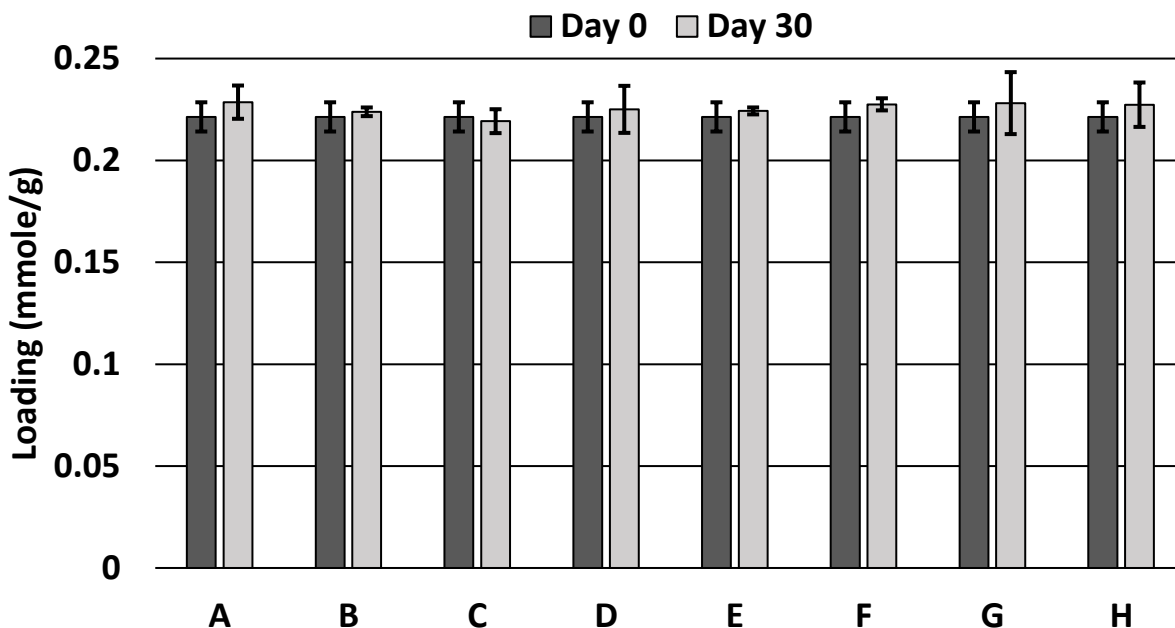


Figure 3 Retention of loading of Tentagel® XV Fmoc-NHNH-Trityl resins stored in a -10°C freezer: (A) In DCM, (B) In MTBE, (C) In 2-MeTHF, (D) In DMF, (E) In 1:1 DCM:DMF, (F) In 2:1 DCM:DMF, (G) Dry in Air, (H) Dry under N₂. All error bars show ± 1 standard deviation for the triplicate loading tests. Dark grey is day 0, light grey is day 30.

± 0.009 mmol/g, 81% of its initial value, with the losses believed to be caused by incomplete coupling or partial deprotection of the Fmoc-glycine. Meanwhile, the resin stored as the free hydrazine (D2) had a final loading of 0.0224 ± 0.001 mmol/g, just 9.6% of its initial value (Figure S2). This indicates the Fmoc group is required to maintain stable loading of the resin.

Another strategy that could help to protect resins from hydrolysis, and that would allow rapid initiation of syntheses would be to maintain pre-measured and swollen resins. To determine if this is a viable strategy for Fmoc-NHNH-Trityl resins a supply of Tentagel® XV trityl chloride with a manufacturer reported loading of 0.22 mmol/g was regenerated and substituted with Fmoc-NHNH₂. After thoroughly washing with DMF and DCM and drying with Et₂O and a vacuum desiccator the loading was measured to be 0.221 ± 0.007 mmol/g. The dry resin was split into 8 fractions, all of which were stored together in a -10 C freezer for 30 days. Prior to measuring loading retention each resin was washed with DMF and DCM and dried

thoroughly with Et₂O and 8 h of vacuum desiccation. Triplicate loading tests on each sample showed no loss of loading for any condition (Figure 3).

We further investigated the stability of the resins at room temperature without vacuum storage. Two separate samples of resin measured at

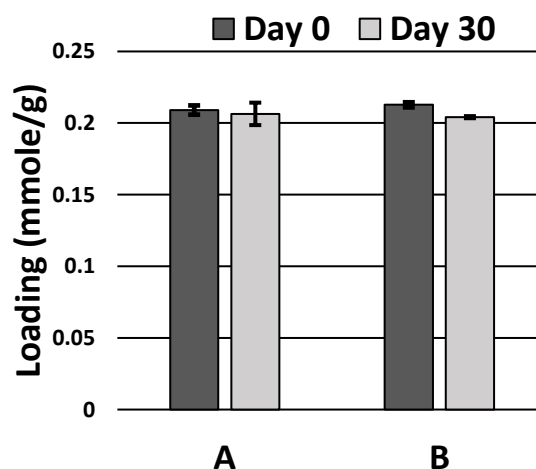


Figure 4 Retention of loading of Tentagel® XV Fmoc-NHNH-Trityl resins stored at Room Temperature, open to air. (A) and (B) represent two independent resin samples. Dark grey is day 0, light grey is day 30 (7 days in desiccator + 23 days open to air).

A

H—E—L—V—D—N—A—V—G—G—D—L—S—K—Q—M—E—E—E—A—V—R—L—F—I—E—W—L—K—N—G—G—P—S—S—G—A—P—P—P—S—NH
NH₂

Exact Mass: 4237.11
Molecular Weight: 4239.74

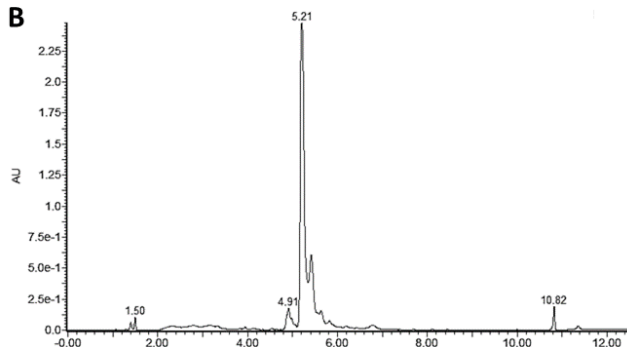
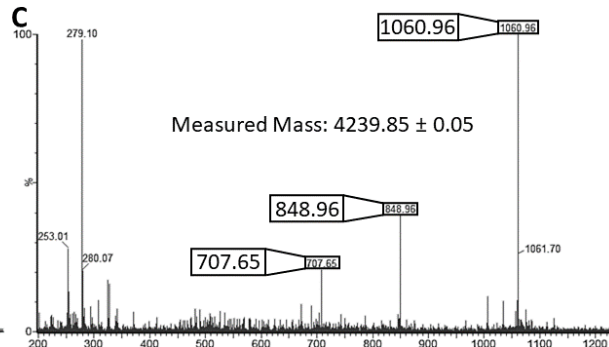
B**C**

Figure 5 Synthesis of a 40-mer peptide on Tentagel® XV Fmoc-NHNH-Trityl resin. (A) Sequence and calculated mass of the GLP-1R agonist peptide P5. (B) UV trace of crude P5 peptide. (C) Combined mass spectrum of the crude material (0-12 minutes)

0.209 ± 0.003 mmol/g and 0.213 ± 0.002 mmol/g respectively were stored in a vacuum desiccator for 7 days, then removed and stored, open to air, at room temperature, for a further 23 days. These resins were then thoroughly washed and dried with Et₂O followed by overnight desiccation, and their loadings were re-tested in triplicate, showing loading retentions of greater than 95% (Figure 4).

To demonstrate the capacity of Fmoc-NHNH-Trityl resins for the synthesis of large peptides, the 40 amino acid GLP1-R agonist peptide P5⁵¹ was synthesized at 0.1 mmol scale using HATU/DIEA conditions on a Fmoc-NHNH-Trityl Tentagel® XV (0.221 ± 0.007 mmol/g). The resulting 924 mg of resin were cleaved with 20 mL of peptide cleavage cocktail M, after precipitation from ether the peptide analyzed by LC-MS (Figure 5) and purified by automated prep-LCMS. The peptide was recovered in 17% yield from resin loading (calculated by mass based on the penta-trifluoroacetate salt) and analyzed by LC-MS TOF (Supporting Information).

4 Conclusion

9-fluorenylmethyl carbazate has been developed as a reagent for high stability loading of PEGylated trityl chloride resins pursuant to synthesis of high value peptide C-terminal hydrazides. Fmoc-NHNH-Trt Tentagel® XV resins fully retained their loadings for at least 1 month under a variety of dry and solvated storage conditions, both frozen and at room temperature, and without the need for inert atmosphere. This robust linker has enabled the development of a universal and reliable procedure for regeneration, loading, drying, and storage of Fmoc-NHNH-Trt resins, including PEGylated resins, to produce hydrazide peptides. The competence of these resins for synthesis of long peptides was demonstrated by the production of the 40-amino acid “P5” peptide in a 17% yield by room temperature SPPS without specific efforts to optimize synthetic conditions. We hope this will encourage more groups to explore the myriad applications of hydrazide containing peptides, and that resin manufacturers might make available trityl chloride resins that are preloaded with Fmoc-NHNH₂ to make these methods available to the broadest possible scope of researchers.

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Conflict of Interest

The authors declare no conflict of interest.

References

1. Honzl, J.; Rudinger, J., Amino-acids and peptides. XXXIII. Nitrosyl chloride and butyl nitrite as reagents in peptide synthesis by the azide method; Suppression of amide formation. *Collection of Czechoslovak Chemical Communications* **1961**, 26 (9), 2333-2344.
2. Greenstein, J. P.; Winitz, M., *Chemistry of the Amino Acids*. Wiley: New York, 1961; Vol. 2.
3. Denkewalter, R. G.; Veber, D. F.; Holly, F. W.; Hirschmann, R., Studies on the total synthesis of an enzyme. I. Objective and strategy. *J Am Chem Soc* **1969**, 91 (2), 503-4.
4. Hirschmann, R.; Nutt, R. F.; Veber, D. F.; Vitali, R. A.; Varga, S. L.; Jacob, T. A.; Holly, F. W.; Denkewalter, R. G., Studies on the total synthesis of an enzyme. V. The preparation of enzymatically active material. *J Am Chem Soc* **1969**, 91 (2), 507-8.
5. Jenkins, S. R.; Nutt, R. F.; Dewey, R. S.; Veber, D. F.; Holly, F. W.; Paleveda, W. J., Jr.; Lanza, T., Jr.; Strachan, R. G.; Schoenewaldt, E. F.; Barkemeyer, H.; Dickinson, M. J.; Sondey, J.; Hirschmann, R.; Walton, E., Studies on the total synthesis of an enzyme. 3. Synthesis of a protected hexacontapeptide corresponding to the 65-124 sequence of ribonuclease A. *J Am Chem Soc* **1969**, 91 (2), 505-6.
6. Strachan, R. G.; Paleveda, W. J., Jr.; Nutt, R. F.; Vitali, R. A.; Veber, D. F.; Dickinson, M. J.; Garsky, V.; Deak, J. E.; Walton, E.; Jenkins, S. R.; Holly, F. W.; Hirschmann, R., Studies on the total synthesis of an enzyme. II. Synthesis of a protected tetratetracontapeptide corresponding to the 21-64 sequence of ribonuclease A. *J Am Chem Soc* **1969**, 91 (2), 503-5.
7. Yanaihara, N.; Yanaihara, C.; Dupuis, G.; Beacham, J.; Camble, R.; Hofmann, K., Studies on polypeptides. XLII. Synthesis and characterization of seven fragments spanning the entire sequence of ribonuclease T1. *J Am Chem Soc* **1969**, 91 (8), 2184-5.
8. Felix, A. M.; Merrifield, R. B., Azide solid phase peptide synthesis. *J Am Chem Soc* **1970**, 92 (5), 1385-91.
9. Meienhofer, J., The Azide Method in Peptide Synthesis. In *Major Methods of Peptide Bond Formation*, 1979; pp 197-239.
10. Wang, P.; Layfield, R.; Landon, M.; Mayer, R. J.; Ramage, R., Transfer active ester condensation: A novel technique for peptide segment coupling. *Tetrahedron Lett* **1998**, 39 (47), 8711-8714.
11. Wang, P.; Shaw, K. T.; Whigham, B.; Ramage, R., Synthesis of peptide C-terminal derivatives using the transfer active ester condensation technique. *Tetrahedron Lett* **1998**, 39 (47), 8719-8720.
12. Lutz, J.; Musiol, H.-J.; Moroder, L., 3.1 Acid Azides. In *Houben-Weyl Methods of Organic Chemistry Vol. E 22a, 4th Edition Supplement: Synthesis of Peptides and Peptidomimetics*, Goodman, M.; Felix, A.; Moroder, L.; Toniolo, C., Eds. Georg Thieme Verlag Stuttgart: New York, 2002; Vol. E 22 a, pp 427-442.
13. Liao, Y.; Kong, Y.; Hu, N.; Jin, Z.; Wang, P., Selective Coupling at the α -Amino Group of Cysteine Using Transfer Active-ester-condensation Technology to Synthesize a Linear Octadecapeptide. *Chemistry Letters* **2010**, 39 (3), 196-197.
14. Camarero, J. A.; Hackel, B. J.; de Yoreo, J. J.; Mitchell, A. R., Fmoc-based synthesis of peptide alpha-thioesters using an aryl hydrazine support. *J Org Chem* **2004**, 69 (12), 4145-51.
15. Fang, G. M.; Li, Y. M.; Shen, F.; Huang, Y. C.; Li, J. B.; Lin, Y.; Cui, H. K.; Liu, L., Protein chemical synthesis by ligation of peptide

hydrazides. *Angew Chem Int Ed Engl* **2011**, *50* (33), 7645-9.

16. Zheng, J. S.; Tang, S.; Qi, Y. K.; Wang, Z. P.; Liu, L., Chemical synthesis of proteins using peptide hydrazides as thioester surrogates. *Nat Protoc* **2013**, *8* (12), 2483-95.
17. Flood, D. T.; Hintzen, J. C. J.; Bird, M. J.; Cistrone, P. A.; Chen, J. S.; Dawson, P. E., Leveraging the Knorr Pyrazole Synthesis for the Facile Generation of Thioester Surrogates for use in Native Chemical Ligation. *Angew Chem Int Ed Engl* **2018**, *57* (36), 11634-11639.
18. Veber, D. F.; Varga, S. L.; Milkowski, J. D.; Joshua, H.; Conn, J. B.; Hirschmann, R.; Denkwalter, R. G., Total synthesis of an enzyme. IV. Factors affecting the conversion of protected S-protein to ribonuclease S'. *Journal of the American Chemical Society* **2002**, *91* (2), 506-507.
19. Dawson, P. E.; Muir, T. W.; Clark-Lewis, I.; Kent, S. B., Synthesis of proteins by native chemical ligation. *Science* **1994**, *266* (5186), 776-9.
20. Barnes, N. G.; Nyandoro, K.; Jin, H.; Macmillan, D., Rapid access to Asp/Glu sidechain hydrazides as thioester precursors for peptide cyclization and glycosylation. *Chem Commun (Camb)* **2021**, *57* (8), 1006-1009.
21. Dao, Y.; Wang, B.; Dong, W.; Zhang, J.; Zhong, C.; Zhang, Z.; Dong, S., Facile Generation of Strained Peptidyl Thiolactones from Hydrazides and Its Application in Assembling MUC - 1 VNTR Oligomers[†]. *Chinese Journal of Chemistry* **2021**, *39* (9), 2509-2516.
22. Dardashti, R. N.; Kumar, S.; Sternisha, S. M.; Reddy, P. S.; Miller, B. G.; Metanis, N., Selenolysine: A New Tool for Traceless Isopeptide Bond Formation. *Chemistry* **2020**, *26* (22), 4952-4957.
23. Dong, H.; Meng, X.; Zheng, X.; Cheng, X.; Zheng, Y.; Zhao, Y.; Wu, C., Design and Synthesis of Cross-Link-Dense Peptides by Manipulating Regioselective Bisthioether Cross-Linking and Orthogonal Disulfide Pairing. *J Org Chem* **2019**, *84* (9), 5187-5194.
24. Eid, E.; Boross, G. N.; Sun, H.; Msallam, M.; Singh, S. K.; Brik, A., Total Chemical Synthesis of ISGylated-Ubiquitin

Hybrid Chain Assisted by Acetamidomethyl Derivatives with Dual Functions. *Bioconjug Chem* **2020**, *31* (3), 889-894.

25. Galesic, A.; Rakshit, A.; Cutolo, G.; Pacheco, R. P.; Balana, A. T.; Moon, S. P.; Pratt, M. R., Comparison of N-Acetyl-Glucosamine to Other Monosaccharides Reveals Structural Differences for the Inhibition of alpha-Synuclein Aggregation. *ACS Chem Biol* **2021**, *16* (1), 14-19.
26. Hananya, N.; Daley, S. K.; Bagert, J. D.; Muir, T. W., Synthesis of ADP-Ribosylated Histones Reveals Site-Specific Impacts on Chromatin Structure and Function. *J Am Chem Soc* **2021**, *143* (29), 10847-10852.
27. Huang, D. L.; Li, Y.; Liang, J.; Yu, L.; Xue, M.; Cao, X. X.; Xiao, B.; Tian, C. L.; Liu, L.; Zheng, J. S., The New Salicylaldehyde S,S-Propanedithioacetal Ester Enables N-to-C Sequential Native Chemical Ligation and Ser/Thr Ligation for Chemical Protein Synthesis. *J Am Chem Soc* **2020**, *142* (19), 8790-8799.
28. Li, Y.; Liu, J.; Zhou, Q.; Zhao, J.; Wang, P., Preparation of Peptide Selenoesters from Their Corresponding Acyl Hydrazides[†]. *Chinese Journal of Chemistry* **2021**, *39* (7), 1861-1866.
29. Liang, L. J.; Chu, G. C.; Qu, Q.; Zuo, C.; Mao, J.; Zheng, Q.; Chen, J.; Meng, X.; Jing, Y.; Deng, H.; Li, Y. M.; Liu, L., Chemical Synthesis of Activity-Based E2-Ubiquitin Probes for the Structural Analysis of E3 Ligase-Catalyzed Transthiolation. *Angew Chem Int Ed Engl* **2021**, *60* (31), 17171-17177.
30. Liao, P.; He, C., Chemical Synthesis of the Sec-To-Cys Homologue of Human Selenoprotein F and Elucidation of Its Disulfide-pairing Mode. *Front Chem* **2021**, *9*, 735149.
31. Lu, D.; Yin, H.; Wang, S.; Tang, F.; Huang, W.; Wang, P., Chemical Synthesis of the Homogeneous Granulocyte-Macrophage Colony-Stimulating Factor Through Se-Auxiliary-Mediated Ligation. *J Org Chem* **2020**, *85* (3), 1652-1660.
32. Mousa, R.; Hidmi, T.; Pomyalov, S.; Lansky, S.; Khouri, L.; Shalev, D. E.; Shoham, G.; Metanis, N., Diselenide crosslinks for enhanced and simplified oxidative protein folding. *Communications Chemistry* **2021**, *4* (1).

33. Tsuda, S.; Masuda, S.; Yoshiya, T., Solubilizing Trityl-Type Tag To Synthesize Asx/Glx-Containing Peptides. *Chembiochem* **2019**, *20* (16), 2063-2069.
34. Yang, X.; Miao, H.; Xiao, R.; Wang, L.; Zhao, Y.; Wu, Q.; Ji, Y.; Du, J.; Qin, H.; Xuan, W., Diverse protein manipulations with genetically encoded glutamic acid benzyl ester. *Chem Sci* **2021**, *12* (28), 9778-9785.
35. Ye, F.; Zhao, J.; Xu, P.; Liu, X.; Yu, J.; Shangguan, W.; Liu, J.; Luo, X.; Li, C.; Ying, T.; Wang, J.; Yu, B.; Wang, P., Synthetic Homogeneous Glycoforms of the SARS-CoV-2 Spike Receptor-Binding Domain Reveals Different Binding Profiles of Monoclonal Antibodies. *Angew Chem Int Ed Engl* **2021**, *60* (23), 12904-12910.
36. Zhang, Y.; Chen, J.; He, C., On Demand Attachment and Detachment of rac-2-Br-DMNPA Tailoring to Facilitate Chemical Protein Synthesis. *Org Lett* **2021**, *23* (16), 6477-6481.
37. Zhao, Z.; Metanis, N., Copper-Mediated Selenazolidine Deprotection Enables One-Pot Chemical Synthesis of Challenging Proteins. *Angew Chem Int Ed Engl* **2019**, *58* (41), 14610-14614.
38. Bello, C.; Kikul, F.; Becker, C. F., Efficient generation of peptide hydrazides via direct hydrazinolysis of Peptidyl-Wang-TentaGel resins. *J Pept Sci* **2015**, *21* (3), 201-7.
39. Li, Y. M.; Yang, M. Y.; Huang, Y. C.; Li, Y. T.; Chen, P. R.; Liu, L., Ligation of expressed protein alpha-hydrazides via genetic incorporation of an alpha-hydroxy acid. *ACS Chem Biol* **2012**, *7* (6), 1015-22.
40. Liu, J.; Ekanayake, O.; Santoleri, D.; Walker, K.; Rozovsky, S., Efficient Generation of Hydrazides in Proteins by RadA Split Intein. *Chembiochem* **2020**, *21* (3), 346-352.
41. Harre, M.; Nickisch, K.; Tilstam, U., An efficient method for activation and recycling of trityl resins. *React Funct Polym* **1999**, *41* (1-3), 111-114.
42. Redwan, I. N.; Grotli, M., Method for activation and recycling of trityl resins. *J Org Chem* **2012**, *77* (16), 7071-5.
43. Spare, L. K.; Menti, M.; Harman, D. G.; Aldrich-Wright, J. R.; Gordon, C. P., A continuous flow protocol to generate, regenerate, load, and recycle chlorotriptyl functionalised resins. *Reaction Chemistry & Engineering* **2019**, *4* (7), 1309-1317.
44. Frutos, S.; Tulla-Puche, J.; Albericio, F.; Giralt, E., Chemical Synthesis of 19F-labeled HIV-1 Protease using Fmoc-Chemistry and ChemMatrix Resin. *International Journal of Peptide Research and Therapeutics* **2007**, *13* (1-2), 221-227.
45. Camperi, S. A.; Marani, M. M.; Iannucci, N. B.; Côté, S.; Albericio, F.; Cascone, O., An efficient strategy for the preparation of one-bead-one-peptide libraries on a new biocompatible solid support. *Tetrahedron Lett* **2005**, *46* (9), 1561-1564.
46. Garcia-Ramos, Y.; Paradis-Bas, M.; Tulla-Puche, J.; Albericio, F., ChemMatrix((R)) for complex peptides and combinatorial chemistry. *J Pept Sci* **2010**, *16* (12), 675-8.
47. Bacsá, B.; Horvati, K.; Bosze, S.; Andrae, F.; Kappe, C. O., Solid-phase synthesis of difficult peptide sequences at elevated temperatures: a critical comparison of microwave and conventional heating technologies. *J Org Chem* **2008**, *73* (19), 7532-42.
48. Rapp, W., PEG Grafted Polystyrene Tentacle Polymers: Physico-Chemical Properties and Application in Chemical Synthesis. In *Combinatorial Peptide and Nonpeptide Libraries*, Jung, P. D. G., Ed. 1996; pp 425-464.
49. Rapp, W., personal communication (2018).
50. Anaspec Fmoc loading test protocol. <https://www.peptideweb.com/loading-protocols> (accessed January 2022).
51. Zhang, H.; Sturchler, E.; Zhu, J.; Nieto, A.; Cistrone, P. A.; Xie, J.; He, L.; Yea, K.; Jones, T.; Turn, R.; Di Stefano, P. S.; Griffin, P. R.; Dawson, P. E.; McDonald, P. H.; Lerner, R. A., Autocrine selection of a GLP-1R G-protein biased agonist with potent antidiabetic effects. *Nat Commun* **2015**, *6*, 8918.

