1 Mass difference matching unfolds hidden molecular structures of dissolved

2 organic matter

- 3 <u>Carsten Simon^{1,†}</u>, Kai Dührkop², Daniel Petras^{3,4}, Vanessa-Nina Roth^{1,§}, Sebastian
- 4 Böcker², Pieter C. Dorrestein³, and Gerd Gleixner^{1,*}
- ¹ Molecular Biogeochemistry, Department of Biogeochemical Processes, Max Planck Institute for Biogeochemistry,
 Hans-Knöll-Straße 10, 07745 Jena, Germany
- ² Chair for Bioinformatics, Friedrich-Schiller-University, Jena, Germany
- 8 ³ Collaborative Mass Spectrometry Innovation Center, Skaggs School of Pharmacy and Pharmaceutical Sciences,
- 9 University of California San Diego, San Diego, CA, USA
- ⁴ CMFI Cluster of Excellence, Interfaculty Institute of Microbiology and Medicine, University of Tübingen, 72076,
 Tübingen, Germany
- 12 TOC FIGURE:



13

14 ABSTRACT: Ultrahigh-resolution Fourier transform mass spectrometry (FTMS) has revealed 15 unprecedented detail of natural complex mixtures such as dissolved organic matter (DOM) on a molecular 16 formula level, but we lack approaches to access the underlying structural complexity. We here explore the 17 hypothesis that every DOM precursor is potentially linked with all emerging product ions in FTMS² 18 experiments. The resulting mass difference (Δm) matrix is deconvoluted to isolate individual precursor Δm 19 profiles and matched with structural information, which was derived from 42 Δm features from 14 in-house 20 reference compounds and a global set of 11477 Δm features with assigned structure specificities, using a 21 dataset of \sim 18000 unique structures. We show that Δm matching is highly sensitive in predicting potential 22 precursor identifies in terms of molecular and structural composition. Additionally, the approach identified 23 unresolved precursors and missing elements in molecular formula annotation (P, Cl, F). Our study provides 24 first results how Δm matching improves structural domains in Van Krevelen space, but simultaneously 25 demonstrates the wide overlap between the structural domains. We show that this effect is likely driven by 26 chemodiversity and offers an explanation for the observed ubiquitous presence of molecules in the center of the Van Krevelen space. Our promising first results suggest that ∆m matching can unfold the structural
information encrypted in DOM and assess the quality of FTMS-derived molecular formulas of complex
mixtures in general.

30 **Synopsis**: We present an approach to deconvolute and explore the structural composition of co-31 fragmented mixtures of organic molecules in environmental media.

Keywords: Natural organic matter, NOM, DI-ESI-MS/MS, FTMS, Orbitrap, tandem mass spectrometry,
 MS/MS, deconvolution

- 34
- 35

36 1. INTRODUCTION

37 Complex mixtures are key study objects in environmental and industrial applications, but their analysis remains challenging.¹⁻⁴ One of the most complex mixtures in natural ecosystems is dissolved organic matter 38 39 (DOM).^{5,6} DOM is a central intermediate of ecosystem metabolism and mirrors molecular imprints of interactions with its abiotic and biotic environment⁷⁻⁹, which form the basis for processes such as carbon 40 sequestration and nutrient recycling.^{10,11} Despite significant advances in ultrahigh-resolution mass 41 spectrometry (FTMS)^{2,4} and nuclear magnetic resonance spectroscopy¹², scientists still struggle to decode 42 43 this information on the molecular level^{13–17}, and novel approaches to identify distinct structures are required 44 to translate molecular-level information into improved process understanding.

Open and living systems promote the formation of ultra-complex mixtures of thousands to millions of individual constituents^{18,19} that mirror large environmental gradients.^{20–22}As a consequence, DOM poses significant challenges to separation, isolation, and structure elucidation. Direct infusion (DI) FTMS techniques have become indispensable tools for the molecular-level analysis of DOM as they reveal unprecedented detail of molecular formulas using the exact mass (MS¹ data, m/z) even without prior separation.²³ However, FTMS techniques are selective and do not resolve all structural detail observed at 51 the exact mass in DOM, as the presence of isobars and isomers hinders the identification of particular 52 structures from these molecular formulas.^{19,23–25} Additionally, current structural databases cover only a small 53 fraction of molecular formulas encountered, and typically lead to annotation rates < 5%.^{18,26,27}

54 One way to obtain structure information on isomers and isobars is through collision-induced dissociation 55 (CID) in fragmentation experiments (MS², or multistage MSⁿ).^{27–29} The relatively wide isolation window (\sim 56 1 Da) of mass filters applied for precursor selection commonly hinders the isolation and subsequent 57 fragmentation of single exact masses, leading to mixed "chimeric" MS² spectra of co-fragmented 58 precursors.³⁰ Even though some authors have achieved isolation of single masses or improved description 59 of chimeric tandem MS data, most studies have pointed out that fragmentation patterns were rather universal across DOM samples.^{18,19,31–35} Most of these studies, however, focused on the major product ion peaks 60 61 (fragments), which usually make up only 60 - 70 % of the total product ion abundance, and thus disregard 62 many low-abundance signals that may be more suitable to detect structural differences.^{19,31}

63 The major product ions encountered in tandem mass spectra of DOM relate to sequential neutral losses of 64 common small building blocks, mainly CO₂, H₂O, or CO units.^{14,33} A mass difference between a precursor 65 and a product ion in an MS² spectrum is herein called "delta mass" and referred to as Δm (plural Δm 's). 66 Many Δm 's such as CO₂ or H₂O are commonly observed and are thus deemed non-indicative for the identification of structural units.^{18,28,31,33,36} In contrast, other studies found recurring low m/z product ions 67 68 (e.g., at *m/z* 95, 97, 109, 111, 123, 125, 137, 139, 151, and 153) that were interpreted as a limited set of core 69 structural units substituted with a set of functional groups, yet in different amounts and configurational types that would lead to highly diverse mixtures.³⁷⁻⁴⁴ From a stochastic standpoint, the occurrence of common 70 71 neutral losses may not be surprising; for example, many structures contain hydroxyl groups that could yield 72 H₂O losses, and CO₂ could originate from ubiquitous carboxyl groups.⁴⁵ In contrast, the occurrence of two 73 molecules sharing a larger substructure would be less probable, and thus less easily detected as a major 74 peak. Signatures of DOM's structural diversity could thus prevail in the high number of low-abundance 75 fragments usually detected below m/z 200-300, as opposed to the higher abundance of fragments connected 76 to losses of small substructures such as CO2 or H2O. Given the large number of estimated isomers and 77 isobars underlying usual DOM data^{18,19,31,32,39,45-48}, we here build upon the hypothesis that every co-78 fragmented precursor potentially contributes to every emerging product ion signal. We interpret the resulting chimeric MS² data as a structural fingerprint that can be deconvoluted to obtain individual precursor Δm 79 80 matching profiles. The analysis of Δm 's that link precursor and product ions, in contrast to indicative product 81 ions (fragments) alone, is independent of the masses of the unknown precursors and known reference 82 compounds in databases of annotated Δm features. Although this approach will sacrifice the identification 83 of true knowns, it allows for the identification of potential structural analogs via indicative Δm 's and is 84 suited best when annotation rates are as low as in the case of DOM, i.e., when most compounds are yet 85 unknown.18,26,27

86 Despite the unknown identity of most of the molecules present in DOM, its potential sources can be 87 constrained reasonably well. Plants produce most of the organic matter that sustains heterotrophic food webs 88 in natural ecosystems. Plant metabolites such as polyphenols and polyaromatic structures thus represent a 89 major source of DOM. Therefore, an early decomposition phase likely exists when the imprint of 90 soluble/solubilized plant metabolites is still detectable by MS² experiments using current FTMS technology. 91 For example, lignin-related compounds show indicative methoxyl and methyl radical losses^{18,49,50}; 92 glycosides indicate the loss of a sugar unit^{51,52} and hydrolyzable tannins are expected to lose galloyl units.⁵² 93 Even related compounds such as flavon-3-ols and flavan-3-ols could potentially be distinguished by their 94 indicative retro-cyclization products.^{51,53} Mass differences related to atoms such as N, S, P, Cl, Br, I and F 95 could also help to identify unknown organic nutrient species or disinfection byproducts, thereby widening the applicability of the approach.^{1,54} Lastly, indicative Δm fingerprints could provide constraints to putative 96 97 compound group annotations derived from molecular formula data alone (Van Krevelen diagrams), or allow 98 for a more precise annotation.^{55–57}

99 We hypothesized that DOM from swamps and topsoil, in close contact to plant inputs and active microbial 100 communities, would reflect recognizable plant-related source imprints that can be revealed by Orbitrap 101 tandem mass spectrometry. Specifically, we explored links between precursor Am matching profiles and 102 precursor characteristics such as nominal mass, mass defect, initial ion abundance, fragmentation sensitivity, 103 oxygen-to hydrogen ratio (O/C), heteroatom content, and structure suggestions. These properties are in part 104 predictable from the assigned molecular formula, and thus allow for an evaluation of the approach ("proof-105 of-concept") while also revealing potential non-assigned molecules of special interest (e.g., P-, Cl-, Br-, I-106 and F-containing molecular formulas). Lastly, we hypothesized that indicative Δm features of plant phenols, 107 e.g., lignin- and tannin-related losses, would match their yet unknown structural analogs in DOM and that 108 these patterns would reflect commonly applied structural domain distributions.^{56,58}

109 2. EXPERIMENTAL SECTION

110 A detailed experimental procedure is provided in the Supplemental Information of this article (Note S-111 1). In short, we chose a set of 14 aromatic reference compounds as representative plant metabolites in DOM 112 (Figure S-1, Table S-1) and a forest topsoil pore water isolate⁵⁹ and Suwannee River Natural Organic Matter (SRNOM)⁶⁰ as exemplary DOM samples. All reference and sample solutions were directly infused 113 114 into the ESI (electrospray) source of an Orbitrap Elite (Thermo Fisher Scientific, Bremen) at negative 115 ionization mode (Table S-2) and fragmented by collision-induced dissociation (CID, MS²). We chose four 116 nominal masses within the average mass range typically observed in terrestrial DOM samples (m/z 200 – 117 500) for fragmentation (m/z 241, 301, 361, and 417, herein referred to as isolated precursor ion mixtures, 118 "IPIMs") as a first set of data to test the approach.⁶¹ Soil DOM was analyzed at three normalized collision 119 energy (NCE) levels (15, 20, and 25%). MS³ spectra of selected key product ions (aglycons of flavonoids 120 and demethylated dimethoxy-methyl-benzoquinone) were acquired as well at NCE 20 or 25. After 121 recalibration with known (Table S-3) or predicted product ions (losses of CO₂, H₂O, etc.), all major product 122 ions were annotated with a molecular formula in reference compounds (Figure S-2, Table S-4, Table S-5) 123 and DOM. Formula annotation was conducted with a Matlab routine recently incorporated into an open 124 FTMS data processing pipeline.⁶²

125 For MS² data analysis, we generated Δm matrices of every pairwise combination of precursor and 126 product ions (" Δm fingerprints"). Every value in this matrix is referred to as a Δm feature or simply Δm . We

127 compared the <u>unknown Δm features</u> in DOM to three lists of <u>known Δm features</u>:

- 128 a) 54 Δ m features ubiquitously found in DOM (**Table S-6**),
- b) 55 Δm features from the set of 14 reference compounds (Table S-7), and

130 c) 11477 Δ m features from a negative ESI MS² library with 249916 reference spectra of 17994 unique

131 molecular structures annotated by SIRIUS⁶³ (list available in the supporting datasets). Reference spectra

132 were collected from from GNPS, MassBank, MoNA, and NIST.^{64,65}

133 The detection of a known Δm feature in DOM is herein called " Δm matching", and detected Δm features 134 are called Δm matches. Matching was conducted at a mass tolerance of ± 0.0002 Da (2 ppm at 200 Da). The 135 array of Δm matches of a single precursor is called the Δm matching profile, and all precursor profiles of an 136 IPIM form the subset of matched Δm 's of the Δm matrix introduced above. The decomposition of the MS² 137 spectrum into a Δm matrix and therefore, individual Δm matching profiles is what we define as the 138 deconvolution step in this study. Δm 's of the literature- and reference-compound derived lists showed some 139 overlap and were largely part of the SIRIUS list as well (see details in SI). The specificity of any Δm feature in the SIRIUS list was checked by their association to compound classes as defined by ClassvFire.⁶⁶ The 140 141 top 15 significantly associated classes were then obtained for each Δm feature in list c) and included into 142 analyses using the reference-compound derived list (list b) as well.

143 We assessed the probability of false positive matches and accounted for molecular formula constraints 144 (numbers of elements in the formula), ion abundance and measures of fragmentation sensitivity to validate 145 our approach. The matching data was combined for each NCE level and transformed into a binary format. 146 We classified Δm matching profiles of DOM precursors and reference compounds by two-way hierarchical 147 clustering using Ward's method and Euclidean distance, as well as Principal Components Analysis (PCA) 148 in PAST (v3.10).⁶⁷ We visualized numbers of individual Δm matches and Δm cluster matches in Van 149 Krevelen space to analyze patterns in Δm matching frequency ("structural domains"). We chose the 150 structural domains reprinted in the 2014 review by Minor et al. for reference, because this represents the 151 general level of detail and type of classes distinguished in recent DOM studies (Figure S-3).^{58,68–70} In a 152 separate analysis, lignin-like and N- and S-containing formulas were also classified with a more general 153 Van Krevelen scheme besides the reference one.⁷¹

154 Finally, we assessed the agreement between structures predicted by Δm matching and those suggested 155 in natural product structural databases. We combined structure suggestions from different databases, 156 including Dictionary of Natural Products⁷², KNApSAcK⁷³, Metacyc⁷⁴, KEGG⁷⁵, and HMDB⁷⁶ as well as 157 their expanded in-silico annotations based on predicted enzymatic transformations in the MINEs database.⁷⁷ 158 Although the MINEs database covers 198 generalized chemical reaction rules it may not include all potential 159 environmental reactions because those are not necessarily only driven by enzymes. The InChi-Key of 160 structures was used to exclude stereoisomers and classify suggested structures into compound classes by 161 ClassyFire.⁶⁶

162 3. RESULTS AND DISCUSSION

163 3.1. Tandem MS fragmentation of reference compounds and construction of Am lists. The 14 164 aromatic reference compounds (Figures S-1, S-2 and S-3) yielded 42 new Δm features (i.e., not covered in 165 the list of common Δm 's, **Table S-6**) but also eight that were described in DOM. These eight Δm features 166 (namely: H₂O, 18.0106; CO, 27.9949; C₂H₄, 28.0313; C₂H₂O, 42.0106; CO₂, 43.9898; CH₂O₃, 62.0004; 167 C₂O₃, 71.9847; and C₃O₅, 115.9746) were kept in the list to compare DOM and reference compounds (**Table** 168 S-7). Besides precursor formulas #2 (Hydroxy-cinnamic acid, or p-coumaric acid; C₉H₈O₃, 164.0473), #3 169 (Gallic acid; $C_7H_6O_5$, 170.0215) and #5 (m-Guaiacol; $C_7H_8O_2$, 124.0524), which were found among the 42 170 Δm 's as potential structural equivalents, five Δm 's of potential substructures likely to be found in DOM 171 were added to the list, namely the ones of precursors #1 (Vanillic acid; C₈H₈O₄, 168.04226), #4 (Creosol, 172 $C_8H_{10}O_2$, 138.0681), **#8** (Ellagic acid; $C_{14}H_6O_8$, 302.0063) and **#10** (Catechin; $C_{15}H_{14}O_6$, 290.0790), and the 173 neutral aglycon of compounds #12 and #13 (flavonol core of Spiraeoside and Isoquercitin; C15H10O7, 174 302.0427). More details on reference compound fragmentation are given in the SI (Note S-2).

3.2. Fragmentation behavior of soil DOM. DOM precursors were isolated and fragmented to obtain
Δm data (Figure S-4). To find the best collision energy to fragment DOM, we analyzed soil DOM at three

NCE levels (15, 20 and 25). All IPIMs showed similar fragmentation properties (Note S-3, Table S-8).
Highest numbers of product ions were found at the highest NCE (Figure S-5). Product ion spectra did not
indicate abrupt structural changes upon increasing fragmentation energy, showing no separation of
isomers/isobars but a continuous increase in fragmentation across all precursors. Based on the above results,
NCE of 25 was chosen to fragment SRNOM as a second DOM sample for comparison.



182

Figure 1. Links between selected DOM precursor properties (upper panels, initial ion abundance at NCE 0; mid panels, half-life normalized collision energy (NCE) at which ion abundance has dropped by 50%; lower panels, matches of delta masses (Δm 's) of measured precursor and product ion masses (delta masses, Δm) with a list of 11477 known Δm features from SIRIUS) and each precursor's (**a**, **b**, **c**) O/C ratio or (**d**, **e**, **f**) mass defect. O/C ratios can only be shown for precursors with an annotated molecular formula. Additional data from reference compounds (red diamonds, see also **Figure S-3**) and SRNOM (orange crosses) is shown in mid and lower panels, respectively. Statistical data was derived from linear fits; asterisks (***) denote p-value < 0.001.

190 Despite common differences between precursor ion abundance and O/C ratio or mass defect (**Figure 1a**,

191 d), we found a significant positive link between both metrics and fragmentation sensitivity independent of

- 192 nominal mass, ranging from half-life NCE (i.e., the NCE level causing 50% decrease in ion abundance) of
- 193 10-35 under our instrumental settings (calculated from linear fits). Remarkably, this trend was not observed
- 194 in reference compounds (Figure 1b, e). Such a discrepancy has been observed also by Zark et al. for the
- 195 common CO₂ loss, and was interpreted as a result of intrinsic averaging.^{31,45} In contrast, Dit Foque et al.
- 196 described potential separation of less complex isomer mixtures by ramped fragmentation.²⁹ Bearing the

197 limitation in mind that we only analyzed four IPIMs here, our results support the intrinsic averaging 198 hypothesis and indicate that fragmentation sensitivity may be an additional property shaped by DOM 199 complexity.^{18,20,45} It also supports our assumption of a high number of isomers and isobars "hidden" beneath 200 each precursor molecular formula, which also increases the probability to detect meaningful links between 201 precursor and product ions. A minor group of oxygen-poor formulas was non-responsive (Note S-3). 202 Matching to the list of all SIRIUS Δm 's showed no significant relation to O/C ratio but to mass defect 203 (Figure 1c, f). In contrast to mass defect, initial ion abundance showed no link to fragmentation sensitivity 204 but was significantly correlated to higher numbers of Δm matches (r = 0.41, R² = 0.17, n = 157, p < 0.001; 205 see also Tables S-9, S-10, S-11, S-12, and Figure S-6). DOM precursors with an average O/C ratio matched more often than low O/C, fragmentation-resistant precursors (Figure 1c; Figure S-7, Note S-3)^{18,19,35} or 206 207 high O/C, easily fragmented precursors (Figure 1b). These observations together show that fragmentation 208 sensitivity and Δm matching seem to be independent DOM precursor properties and that Δm matching could 209 be driven by ion abundance. SRNOM and the soil water sample shared many molecular formulas (n=107; 210 84% of soil DOM and 74% of SRNOM formulas) which accounted for most of the precursor ion abundance 211 at NCE 25 (96,5% and 97.2%, respectively). Despite this high similarity, SRNOM precursors showed higher 212 numbers of Δm matches (Figure 1c, f) which could indicate that the same molecular formula is more 213 chemodiverse, i.e. has more underlying structural formulas, in SRNOM compared to soil DOM (further 214 discussion in section 3.5).



216

217 Figure 2. Δm matching in chemical space for soil (porewater) DOM (panels $\mathbf{a} - \mathbf{f}$) and SRNOM (panels $\mathbf{g} - \mathbf{l}$). 218 Exemplary reference compound structures with marked indicative Δm units are shown in lower panels (m – q). Grey 219 boxes refer to anticipated structural domains (Figure S-3).⁶⁴ Panels a - l show precursors with an annotated molecular 220 formula by their atomic H/C and O/C ratios (Van Krevelen plot; soil DOM, n = 127; SRNOM, n = 144); grey boxes 221 indicate representative structural domains that are commonly used (see Figure S-3 for details). Dot size encodes the 222 number of matches to non-indicative $(\mathbf{a} - \mathbf{c}, \mathbf{g} - \mathbf{i})$ vs. indicative Δm 's $(\mathbf{d} - \mathbf{f}, \mathbf{j} - \mathbf{l})$; see legends in every plot. Colored 223 boxes in indicative VK plots mark the expected structural region of formulas that would yield the respective Δm , and 224 colors refer to the structural motifs marked in panels \mathbf{m} - \mathbf{q} . Calculations based on $\Delta \mathbf{m}$ data are presented in more detail

in Table S-13. Highlighted red open diamonds in panels e and k indicate loss of up to three gallic acid equivalents
(size not drawn to scale).

227 **3.3. Evaluation of the \Delta m matching approach.** We used the matching data of molecular formulas in 228 DOM for a proof-of-concept evaluation of our Δm matching approach. Specifically, we aimed to test the 229 hypothesis that all precursors are potentially linked to all product ions in chimeric MS² spectra of 230 ultracomplex DOM. Our analysis was congruent with previous observations, showing ubiquitous losses of 231 non-indicative oxygen-containing functionalities (Table S-6) while also revealing more detail (Figure S-232 4c, Table S-7). Details are given in the Supporting Information (Note S-4); in short, we found expected 233 trends in losses of CO₂, CO, and CH₂ in both samples (Figure 2a - c, g - i, Table S-13). The predicted 234 heteroatom content (O, N, S) of assigned molecular formulas and a widened tolerance window were used 235 for further analysis of the uncovered structural information. Random Δm matching would be expected if the 236 calculated Δm values were affected by low resolution, low sensitivity, or artifacts such as reactions in the 237 instrument (e.g., between the collision and Orbitrap $cell^{78}$). Instead, we found that 1) precursors with low 238 ion abundance matched to less Δm features (Figure S-6), 2) non-fragmented precursors matched to less or 239 no Δm 's (Figure S-7), and 3) identity of Δm matches agreed with molecular formula prediction (e.g., loss 240 of S-containing Δm 's from S-containing precursors; $\leq 3 \text{ CO}_2$ losses from precursors containing seven O, 241 etc.; Figures S-8 and S-9). Our evaluation also shows that Δm matching not only helps in recalibration⁷⁹ 242 but also serves to check formula annotation routines, as it revealed unresolved precursor compositions 243 interfering especially with CHOS precursors (related to Cl, P and F). This means 1) that these atoms should 244 be included for better coverage of elemental composition (i.e., prioritization) in our specific sample context 245 and that 2) higher resolution power may be required to resolve S-, Cl-, P-, or F-containing precursor 246 compositions.¹ In summary, Δm matching revealed an inherently structured biogeochemical signal of 247 precursors that seem to fragment individually and was highly sensitive in detecting precursor-product ion 248 links. This suggests that chimeric DOM data can be deconvoluted to reveal differences in molecular 249 composition not visible from MS¹ inspection.^{23,80} It should be stressed that these results will need further 250 evaluation due to the small number of DOM precursors, m/z values and samples analyzed here (159 in soil

DOM, 221 in SRNOM), and that deconvolution should be further tested with better-characterized mixtures,
 including, e.g., structural analogs, artificial mixtures or standard additions (spiking).^{14,19,27,42,81}

3.4. Structural domains emerge from clustering with reference compound Am's. DOM precursors from both samples were compared based on Δm matching as an indicator of structural information (**Table S-7**, see section **3.1**). We grouped DOM precursors, reference compounds and Δm features by two-way hierarchical clustering (**Table S-14**). In the following, precursor clusters will be referred to by letter (A - H) and Δm clusters by number (1 – 7; **Table S-15**). Based on the specificity of SIRIUS Δm features (**Table S-14**) and clustering with 14 reference compounds we defined five of the Δm clusters found herein as being structure-specific (**Table S-15**, some shown in **Figure 3d, e, j** and **k**; for details, see also **Table S-13**).

260 Δm features C₄H₈O₄ (120.0423 Da, equivalent to tetrose loss) and C₆H₁₀O₅ (162.0528 Da, equivalent to 261 hexose loss), both members of cluster 2, were found to be specific for alcohols and polyols, carbohydrates, 262 and carbohydrate conjugates, as well as ethers (Table S-14). Reference compounds containing a polyol 263 (quinic acid in #7) or a sugar (glucose in #12 and #13, mannose in #14) contributed Δm 's to this cluster 264 (Table S-15).^{51,52} Am equivalents of these losses matched to 18 soil DOM and 24 SRNOM precursors in the 265 central Van Krevelen plot despite the absence of "carbohydrate-like" precursors (lilac square in Figure 2d, 266 **i** and **o**, **q**). The anticipated shift towards higher O/C and H/C ratios was nonetheless apparent in both 267 samples, especially compared to precursors associated with clusters 3, 4 and 7 (Figure 2e, f and k, l).

268 Δm features of clusters 3 and 4, partly specific to phenylpropanoid and benzenoid structures, were 269 contributed by flavan-3-ols (#10, #11) and aglycones of flavon-3-ols (#12 and #13) and those containing cinnamic, coumaric or gallic acid units (#7, #9, #11).^{28,33,52} Precursors that matched to clusters 3 and 4 (soil 270 271 DOM: n = 27 and n = 12; SRNOM: n = 29, n = 21) were found in the "lignin-like" domain (orange square in 272 **Figure 2e**, **k**; orange circles in panels **o**, **p**, **q**). These C- or H-rich Δm 's (e.g., $C_8H_{10}O_2$ or $C_7H_4O_4$) are likely 273 no combinations of common O-rich losses (CO, H₂O, or CO₂) due to their low O/C and O/H ratios, but this 274 requires further testing with model mixtures. Aliphatic chains prevail as O-poor substructures in substituted 275 cyclic core structures and could contribute.^{82,83} Similar to detection of polyol-equivalent Δm matches outside 276 the expected carbohydrate domain, gallate-equivalent losses were not matched to precursors in the anticipated "tannic" domain (red diamonds and turquoise square in Figure 2e, k; turquoise circle in panel
p).

279 Among the most prominent features was the methyl radical loss^{35,49,50} which matched to oxygen-poor 280 DOM precursors and was one of three Δm features in cluster 7 (soil DOM: n = 18, average O/C = 0.33, 281 SRNOM: n = 25, average O/C = 0.32, Figure 2f, I). The distribution of CH₃•-yielding precursors was 282 paralleled by CH₂ (soil DOM: r = 0.60, $R^2 = 0.35$, n = 127, p < 0.001; SRNOM: r = 0.63, $R^2 = 0.39$, n = 0.001283 144, p < 0.001) and CO losses (r = 0.55, $R^2 = 0.30$, n = 127, p < 0.001; r = 0.58, $R^2 = 0.34$, n = 144, p < 284 0.001) and implied structural similarities (Figure 2f, l), e.g., condensed structures with aliphatic, lactone, 285 or quinone moieties.³⁴ CO and CH₃• were both indicative of benzenoid structures in the SIRIUS-annotated 286 Δm data (**Table S-14**). The methyl radical loss is an expected diagnostic Δm of methoxylated aromatic rings 287 such as present in lignin (orange square in Figure 2f, l; orange circles in panels m, n; see Note S-5), but 288 was also matched to DOM precursors not classified as "lignin-like".^{18,31,35,49} The Δm features CH₃•, CO and 289 C_2H_4 were also linked to CH_4 vs. O series. These describe a regularity in DOM data explained by increments of 0.0364 Da, and are formally annotated as an exchange of CH₄ for O (Figure S-10).^{37,38} Concurrent losses 290 291 of CO and C₂H₄ explained the presence of CH₄ vs. O increments on the product ion level and were paralleled 292 by losses of CH₃[•]. This finding could also explain the ubiquitous presence of CH₄ vs O series in non-293 fragmented DOM; for example, concurrent β -oxidation and de-carbonylation could be enzymatic analogues 294 of the patterns seen in MS² data.²⁶

Matching to Δm features derived from a small set of reference compounds revealed emerging clusters of precursor and Δm feature families that may prove more indicative if constrained with further DOM and reference compound data.¹⁴ Anticipated structural domains were apparent but showed clear overlap, which means that the same precursor was part of more than one Δm -predicted structural domain. An extended analysis using the set of compound class-associated SIRIUS Δm features showed similar trends (**Figure S**-**11**, compare **Figure 2**). These findings must however be taken with caution for four reasons: SIRIUS Δm features were not obtained on the same instrument and thus may include features that,
 although correlated with certain compound classes, may not appear in DOM under the same
 instrumental settings.

- 304 2) SIRIUS Δm features may be biased towards certain classes of compounds, as is our reference set of 305 14 aromatic compounds. Here, we only considered negative ESI mode data which is commonly 306 employed for DOM analysis, and thus, adding positive ESI or other ionizations would extend the 307 range of Δm features and structural classes covered and likely decrease bias.^{14,16,23,84} The same 308 applies to other fragmentation techniques than CID.
- 309 3) Product ion abundance was disregarded in our analysis, but could be used to weigh probabilities of
 310 potential precursor-product ion links in future, potentially in combination with fragmentation energy
 311 gradients (fragmentation tree analysis)⁸⁵, moving m/z isolation windows, or variations in ion
 312 accumulation times that influence MS¹ ion abundance.⁸⁶
- 313 4) Despite a seemingly improved separation of extreme classes (high H/C ratios in fatty acids, high
 314 O/C ratios in carbohydrates, etc.), potential overlap in domain boundaries remained considerable
 315 (Figure S-11).

316 Categorization of precursors into multiple Δ m-defined structural domains was also reflected by large 317 differences in Δm matching between members of the same a-priori defined structural domains or classes 318 (i.e., only based on molecular formula). Twenty-seven precursors shown in Figure 3 were classified as 319 "lignin-like" formulas and were part of seven precursor clusters (B – H; Table S-16), thereby showing clear 320 differences in potential structural composition. Likewise, CHNO and CHOS precursors matched with many 321 of the S- and N-containing SIRIUS Δm features (spanning 3 – 78 S- and 4 – 251 N-containing Δm 's in soil 322 DOM and 0 - 154/0 - 350 in SRNOM; Tables S-17, S-18, S-19 and S-20). These represented on average 323 $79 \pm 19\%$ (63 ± 31% in SRNOM) of all Δm matches per CHOS precursor or $91 \pm 7\%$ (79 ± 28%) of all 324 CHNO precursor matches (detailed analysis, see Note S-6). Many Δm features were also associated to 325 compound classes, revealing potential structural detail (Table S-21). For example, CHNO precursor 326 matching indicated the absence of nitrate esters, but indicated the presence of reduced forms of N partly 327 explained in the literature^{87,88}, including specific Δm 's related to aralkylamines, amino acids, 328 carboximidamides, and dicarboximides/ urea-containing compounds. S-containing Δm matches indicated 329 the potential presence of sulfonic, thiol, thioether or aromatic S precursors.⁸⁴ Taken together, our results 330 show a wide potential diversity of N and S compounds in DOM that contradicts with earlier reports of mainly aromatic N and sulfonic S.34,89,90 As most of these studies analyzed marine DOM, the detection of 331 332 potentially more diverse sets of CHOS and CHNO precursors could relate to the terrestrial, less degraded 333 DOM analyzed here.^{16,91-93} Further tests with N- and S-containing reference compounds and DOM samples 334 are warranted to reveal the hidden diversity and identity of dissolved organic nitrogen and sulfur and confirm 335 potential structures, e.g., by NMR.

All in all, our results show that it may be possible to refine Van Krevelen domains by deconvoluted MS^2 data, and that complementary precursor information could be used to assess false or biased Δ m-based class assignments (e.g., elemental composition, DBE, ionization, fragmentation sensitivity, ion mobility, polarity index, etc.).^{13,56} Fluorescence or NMR spectroscopy could add valuable information if DOM would be fractionated before MS^2 data acquisition.^{21,94}

341 Data-dependent and data-independent acquisition (DDA, DIA) techniques could be used to cover the 342 whole mass range of precursors in DOM mass spectra in future, and are widely employed in LC-MS of complex mixtures.^{16,27,95,96} For example, Ludwig et al. presented a DIA scheme (SWATH-MS) that employs 343 344 one precursor scan and 32 isolation windows of 25 Da width, covering 800 Da within 3.3 seconds; similar schemes are likely transferable to acquire full mass range data of directly-injected DOM.⁹⁷ Kurek et al. 345 346 recently presented such data, covering product ions generated from similar isolation window (m/z 392 – 347 408).¹⁶ Smaller isolation windows as used herein were also employed by Leyva et al. to discern fragmentation pathways and structural families in DOM (mass range m/z 261 - 477)¹⁴; this approach could 348 349 be extended to include the diversity of Δm features shown here. Together, this shows that practicable tandem 350 MS acquisition strategies are in reach and will enable deeper analyses of Δm features in DOM soon.

351 3.5. Drivers of differences in Δm matching between soil DOM and SRNOM. Although matching
 among the two samples was largely consistent, slight differences were apparent from Van Krevelen

353 distributions (Figure 2). We therefore tested the separation of precursor clusters by ordination (Figure 3). 354 Precursor clusters were clearly separated on PC1 and PC2 which together held about 47% of variation. Most 355 considered precursors were shared among samples (64%, 38 out of 59), only a small number was samplespecific (SRNOM = 14, Soil DOM = 7). Sample-specific precursors were found in clusters A (linked to 356 357 carboxylic acids), B (phenols, polyols) and C (benzenoids, **Table S-15**), the remaining clusters D – H were 358 dominated by the shared precursors. Out of the 38 shared precursors, 30 (79%) grouped in the same 359 precursor cluster despite a general trend to higher numbers of matches in SRNOM, but eight grouped 360 differently (bold precursors in Figure 3a). These differences in matching could be related to different chemistries, i.e., different isomeric/ isobaric composition.⁸⁴ For example, based on the correlation of 361 precursor properties (Figure 3b), the cluster "switch" in C₁₁H₁₄O₆ was largely explained by higher ion 362 363 abundance and Δm matches in SRNOM, while in C₂₃H₂₂O₄, the effect was partly linked to higher 364 fragmentation resistance in SRNOM. Unfortunately, we only have data on initial ion abundance and 365 fragmentation sensitivity from the soil DOM isolate; other precursor properties, however, showed very 366 similar trends in both samples (Figure 3b).



368

369 Figure 3. Separation of DOM precursors based on Δm matching. a) Principal Components analysis of all precursors 370 with more than one match to indicative Δm features of the 14 reference compounds (i.e., Δm features shown in **Table** 371 S-7 that are not part of Table S-6, see section 3.1). Colors of dots distinguish precursors from both samples and 372 reference compounds (see legend). Precursors detected in both samples are connected by dotted black lines. Precursor 373 clusters (A - H) are marked by envelops and letters (compare Tables S-14 and S-15). Eight shared precursors that 374 switched precursor clusters are highlighted by bold molecular formula ($C_{12}H_{14}O_{9}$, A in soil DOM \rightarrow H in SRNOM; 375 $C_{19}H_{26}O_3$, $B \rightarrow C$; $C_{26}H_{26}O_5$ and $C_{23}H_{22}O_4$, $B \rightarrow D$; $C_{17}H_{14}O_9$, $G \rightarrow E$; $C_{19}H_{22}O_7$ and $C_{22}H_{26}O_8$, $H \rightarrow E$; $C_{11}H_{14}O_6$, $H \rightarrow G$). 376 b) Correlations of selected precursor properties with scores of PC axes (only DOM precursors with assigned molecular 377 formula included in the correlation). PC axes 3 and 4 are shown in addition. Correlations are indicated for all precursors 378 (n=94) and those detected in each sample (Column "Sets"). For each combination (PC = x, property = y), Pearson's r and significance are given $(0.05 \ge p > 0.01, "*"; 0.01 \ge p > 0.001, "**"; p \le 0.001, "**")$. Negative/ positive 379 380 correlation is indicated also by color (blue, red); non-significant correlations are shown in lighter color or no color if 381 no direction dominated. Matches, matches against the global list of Δm features; Structures, number of hits in natural 382 product and in-silico databases.

383 Similar clustering and Δm -predicted structural classes (Figure S-11) in shared precursors could indicate

a conserved structural composition. Likewise, Kurek et al. observed high similarity in APPI-ionized and

385 IMPRD-fragmented DOM samples, but observed clear differences in CHOS fragmentation.¹⁶ High

367

386 similarities between DOM samples would be in line with stoichiometric principles (i.e., due to a large share 387 in precursors between DOM samples) and could suggest that DOM processing diversifies, but also 388 "randomizes" the molecular composition of each precursor ("universal" signal).^{31,98,99} High congruence of 389 fragmentation patterns (and thus, Δm matching) among DOM precursors has also been interpreted as a sign 390 of similarly substituted but slightly differing core structures.^{35,37} The clusters devised here were small due 391 to the relatively small number of precursors and m/z values analyzed, and thus may not detect significant 392 differences between samples yet. However, even with our small set of precursors, the clustering by Δm 393 matching showed conserved differences in fragmentation between precursor clusters, and in part, even the 394 same precursor in different samples. The fact that this could relate to differences in ion abundance (and 395 therefore, possibly also ionization efficiency) or fragmentation sensitivity is intriguing and should be 396 investigated across a wider range of DOM chemotypes using improved classification approaches as applied 397 here (see also section 3.4).¹⁴

398 **3.6.** Ion abundance is linked to Δm matching frequency and structural diversity. Ion abundance was 399 the most important driver for Δm matching in both samples and highest in the structural domain usually assigned to ubiquitous lignin structures or carboxyl-rich aromatic molecules.^{59,83} This domain also parallels 400 401 with a maximum in potential underlying chemodiversity^{30,100}, which could explain why these signals are 402 ubiquitously found and especially dominant in reworked DOM. $^{92,101}\Delta m$ matching showed potential to reveal 403 this underlying chemodiversity effect and was therefore compared to numbers of structure suggestions and 404 Δ m-predicted compound classes per precursor (**Figure 4**). Numbers of Δ m matches were significantly and 405 positively related to the number of structure suggestions in absolute terms and for specific compound classes 406 (Table S-22). The correlation between Δm -predicted and suggested compound classes was surprisingly 407 similar in both samples and significant for almost all benzenoid-type (benzopyrans, methoxybenzenes, 408 anisoles, phenols, etc.) and most phenylpropanoid-type structures (flavonoids, linear 1,3-diarylpropanoids). 409 Among the organic acids, only vinylogous acids stood out (i.e., containing carboxylic acid groups with 410 insertions of C=C bond(s)). Significant correlations were also found for pyrans, acryloyl compounds, 411 carbohydrates, aryl ketones and alkyl aryl ethers (fatty acids and analogues only in SRNOM).



412

413 Figure 4. Agreement between chemodiversity estimates based on molecular formula (structure suggestions) and 414 precursor-product ion links (Δm matches). Panels **a**, **b**) Correlations between numbers of SIRIUS Δm matches vs. 415 structure suggestions (note log scale, incl. in-silico hits); a) soil DOM, b) SRNOM. Panels c, d) Number of SIRIUS 416 Am matches in Van Krevelen space (scales are similar but legends show different dot sizes); c) soil DOM, d) SRNOM; 417 grey boxes refer to domains defined in Figure S-3. Panels e, f) Number of predicted classes per precursor based on 418 SIRIUS Δm matches (color scale similar in both panels). Structural classes are associated to SIRIUS-annotated Δm 419 features through correlation analysis of host structures and their Δm features (classification based on Classyfire); e) 420 soil DOM, f) SRNOM.

421 The positive link between ion abundance and numbers of Δm matches on the one hand and predicted 422 and suggested structures on the other indicates that ion abundance may be linked to the number of structural 423 isomers and isobars per molecular formula in FTMS spectra of DOM and explains why Am-defined 424 structural domains showed strong overlap in this study. It also provides additional support to our assumption 425 that all precursors potentially contribute to all product ions in DOM: The patterns revealed through Δm 426 matching were largely congruent with the independent estimate of structural composition by natural product 427 databases. The fact that only some classes of compounds (mainly benzenoids and phenylpropanoids) 428 showed significant correlations could point to bias towards plant natural products in the databases employed 429 here; this means that the inclusion of other structure databases and the additional assignment of Δm 's not only to their host structures but also to host organisms (e.g., in GNPS⁶⁵) could reveal further clues about the
potential sources of molecular formulas in DOM.

432 We propose that the number of Δm matches could be interpreted as a novel, relatively easily accessible 433 measure to account for a precursor's underlying potential structural diversity. Such information could help 434 to better understand mechanisms of DOM formation and persistence in the environment. Our results 435 encourage further studies on the Δm matching behavior of synthetic mixtures of known structures and across 436 DOM chemotypes, and the improved bioinformatic exploitation of chimeric (LC-) FTMSⁿ data of complex organic mixtures.^{14,102–104} We acknowledge that natural product and in-silico databases are far from being 437 438 complete, same as the database of annotated Δm matches we used here, despite its large coverage of ~18000 439 unique structures and $\sim 11500 \text{ }\Delta\text{m}$'s. For example, precursors with low mass defects showed exceptionally 440 few structural hits, indicating bias in natural product databases (Figure S-12).¹⁸ These structures were easily 441 fragmented and yielded few Δm matches in our analysis; N- and S-containing precursors were double as 442 likely to show no suggestion compared to CHO precursors. This shows that DOM contains unique molecular 443 structures to be identified in future.

444 4. IMPLICATIONS

445 Tandem MS data of complex samples such as dissolved organic matter (DOM) is impeded by the co-446 fragmentation of precursors with similar nominal mass, and further complicated by the contribution of 447 potential isomers and isobars of a precursor. We employed an approach that analyzes the pairwise mass 448 differences between all precursor and product ions as a whole (Δm matrix). Using a very limited set of 449 precursor features from two samples, we found potential signs of structural imprints related to benzenoids, 450 phenylpropanoids, carbohydrates, sulfonic acids, thiols, thioethers and amino acids, amongst others. The 451 successful matching of indicative Δm features and precursor clustering suggests a remaining – and 452 recognizable - source imprint of primary or recycled plant remains in DOM. Tests with more DOM samples 453 and artificial/ treated mixtures (e.g., DOM with spiked known compounds, or DOM degraded by specific 454 enzymes) are required to test the assumptions employed here and to improve classifications of DOM 455 precursors by Δm clusters. Our first results indicate that FTMS² data may be useful to differentiate molecular

456 composition on the molecular formula level, and that ion abundance and fragmentation sensitivity are two 457 key variables that explain differences in MS^2 data within and among samples. This is intriguing because a 458 shared molecular formula could harbor a completely different set of structures but must be assessed with 459 larger sets of DOM data which would improve detection of such differences. Generally, our findings support 460 the view that Van Krevelen domains are associated with indicative mass losses that relate to stoichiometric 461 differences between compound classes. The most abundant precursors however showed a mixed MS² signal 462 that caused boundary overlap of these Δm -defined domains (Figure 4e, f). While this finding is in line with 463 known patterns of structural diversity and partly explains the ubiquitous presence of abundant DOM signals, 464 it introduces a new paradigm to the interpretation of DOM FTMS data by assigning unknown precursors to 465 multiple structural categories instead of just one. Further evaluation of both natural and spiked/ treated 466 complex mixtures, constantly growing MS databases, and comprehensive decomplexation methods (LC-467 MS, IMS) will together provide fundamental insights into the deconvolution of chimeric spectra from 468 complex samples, and ultimately show the potential to unfold the hidden molecular diversity and identity 469 of DOM.

470 ASSOCIATED CONTENT

471 Data and Software Code Accessibility

All tandem MS data can be found in on the Mass Spectrometry Interactive Virtual Environment (MassIVE) under the
following links: <u>ftp://massive.ucsd.edu/MSV000087117/</u> (soil DOM data), <u>ftp://massive.ucsd.edu/MSV000088869/</u>
(SRNOM data) <u>ftp://massive.ucsd.edu/MSV000087133/</u> (reference compound data) (**Data Set S-1**, raw peak data,
*.mzML files). All other data associated to this manuscript (extensive tables, Δm feature lists, Δm specificity, twoway clustering table, processed data used to create figures, etc.) is available as online free of charge from the
PANGAEA Data Publisher under the following link: <u>https://doi.pangaea.de/10.1594/PANGAEA.932592</u> (**Data Set S-1**,
2, processed data as *.xlsx files).

479 Supporting Information

480 The supporting information contains 22 tables and twelve figures, six additional notes, and 69 references.

481 Table S-1: Information on reference compounds and solutions used in this study. Table S-2: Instrument settings for

- 482 fragmentation experiments. Table S-3: Recalibration peaks used for reference compound in FTMS measurements.
- 483 Table S-4: Precursor and major product ions of the 14 reference compounds. Table S-5: Results of reference

484 compound's tandem MS data analysis with CSI:FingerID. Table S-6: List of reported DOM Δm features from MS1 485 and MS2 studies. Table S-7: List of all 50+5 Δ m features extracted from the reference compound dataset. Table S-8: 486 Properties of four isolated nominal masses (IPIMs) at different NCE levels. Table S-9: Overview of correlations 487 between key properties of the IPIM 241. Table S-10: Overview of correlations between key properties of the IPIM 488 301. Table S-11: Overview of correlations between key properties of the IPIM 361. Table S-12: Overview of 489 correlations between key properties of the IPIM 417. Table S-13: Lists of Δm values used for analysing matching 490 patterns in Van Krevelen space. Table S-14: Matching behavior of precursor clusters against Δm features (Table S-7). 491 Table S-15: Summary of two-way clustering of DOM precursors and reference compounds. Table S-16: Lignin-like 492 precursor formulas and their molecular properties and clustering. Table S-17: S-containing precursor formulas in soil 493 porewater DOM. Table S-18: N-containing precursor formulas in soil porewater DOM. Table S-19: S-containing 494 precursor formulas in SRNOM. Table S-20: N-containing precursor formulas in SRNOM. Table S-21: Structural 495 class-correlated Δm features matched to CHOS or CHNO precursors. Table S-22: Correlations between structure hits 496 and specific Am features in CHO precursors. Figure S-1: Overview of reference compounds used in the study. Figure 497 S-2: Mass accuracy assessment based on reference compound Δm 's. Figure S-3: Distribution of exemplary known 498 structures in chemical space. Figure S-4: Orbitrap tandem MS of soil porewater DOM. Figure S-5: Comparison of 499 matches to the two short Δm lists in relation to m/z and NCE. Figure S-6: Δm matches in relation to precursor ion 500 abundance in soil DOM. Figure S-7: Am matches in relation to precursor fragmentation sensitivity in soil DOM. Figure 501 S-8: Matching assessment with SIRIUS Δm 's (Molecular formula check). Figure S-9: Changes in Δm matching 502 frequency upon widening of tolerance window. Figure S-10: Link between matches to CH₃, CO and C₂H₄ and CH₄ 503 vs. O exchange series. Figure S-11: Structural VK domains based on class-correlated SIRIUS Am features. Figure S-504 12: Effect of mass defect on the number of structure suggestions. Note S-1: Supplementary experimental details. Note 505 S-2: Detailed description of reference compound fragmentation behavior. Note S-3: Behavior of non-responsive DOM 506 precursor ions. Note S-4: Am matching: Proof-of-concept data and key findings. Note S-5: Potential esterification of 507 DOM by methanol during SPE and storage. Note S-6: Structural insight into N- and S-containing DOM precursors. 508 The Supporting Information is available free of charge on the ACS Publications website.

509 Supporting information (PDF)

510 AUTHOR INFORMATION

511 Corresponding Author

512 *Email: gerd.gleixner@bgc-jena.mpg.de

513 **Present Addresses**

- 514 † C.S.: Institute for Biogeochemistry and Pollutant Dynamics, ETH Zürich, Zurich, Switzerland & Swiss Federal
- 515 Institute of Aquatic Science and Technology (Eawag), Department Water Resources and Drinking Water, Duebendorf,
- 516 Switzerland
- 517 § V.-N. R.: Thüringer Landesamt für Umwelt, Bergbau und Naturschutz (TLUBN), Jena, Germany

518 Author Contributions

519 CS performed the measurements. DP, VNR, PD and GG were involved in planning and supervised the work. KD and 520 SB compiled global Δ m feature data, analyzed its specificity, and performed structural classifications of Δ m host 521 structures as well as structure suggestions of DOM precursors. CS processed the experimental data, performed the 522 downstream analyses, drafted the manuscript, and designed the figures. The manuscript was revised through the 523 contributions of all authors. All authors have approved the final version of the manuscript. 524 Notes

525 The authors declare no competing financial interest.

526 ACKNOWLEDGMENT

527 We thank all members of the Dorrestein lab for helpful discussions and insights. We also like to acknowledge Vivian

528 Stefanow for initial structure database surveys. We acknowledge the International Max Planck Research School for

- 529 Global Biogeochemical Cycles (IMPRS-gBGC) for sponsoring CS' research stay in the Dorrestein lab at UCSD, CA,
- 530 USA. The authors acknowledge the financial support from the Max-Planck-Gesellschaft (MPG) and the German
- 531 Research Foundation (DFG, Deutsche Forschungs-Gemeinschaft) as part of the CRC 1076 "Aqua Diva" and a
- 532 Postdoctoral Research Fellowship PE 2600/1, and support through the CMFI Cluster of Excellence (EXC 2124) to
- 533 D.P. V.-N.R. received additional funding by the Zwillenberg-Tietz Stiftung. C.S. received a Ph.D. stipend from the
- 534 International Max Planck Research School for Global Biogeochemical Cycles (IMPRS-gBGC). The authors are
- 535 grateful for critical comments and helpful suggestions received from the three anonymous reviewers.

536 **REFERENCES**

(1) Hollender, J.; Schymanski, E. L.; Singer, H. P.; Ferguson, P. L. Nontarget Screening with
High Resolution Mass Spectrometry in the Environment: Ready to Go? *Environ. Sci. Technol.* 2017, *51*, 11505–11512. https://doi.org/10.1021/acs.est.7b02184.

540 (2) D'Andrilli, J.; Fischer, S. J.; Rosario-Ortiz, F. L. Advancing Critical Applications of High
541 Resolution Mass Spectrometry for DOM Assessments: Re-Engaging with Mass Spectral
542 Principles, Limitations, and Data Analysis. *Environ. Sci. Technol.* 2020, *54*, 11654–11656.
543 https://doi.org/10.1021/acs.est.0c04557.

- 544 (3) Kruve, A. Strategies for Drawing Quantitative Conclusions from Nontargeted Liquid
 545 Chromatography–High-Resolution Mass Spectrometry Analysis. *Anal. Chem.* 2020, *92*,
 546 4691–4699. https://doi.org/10.1021/acs.analchem.9b03481.
- 547 (4) Bahureksa, W.; Tfaily, M. M.; Boiteau, R. M.; Young, R. B.; Logan, M. N.; McKenna, A.
 548 M.; Borch, T. Soil Organic Matter Characterization by Fourier Transform Ion Cyclotron
 549 Resonance Mass Spectrometry (FTICR MS): A Critical Review of Sample Preparation,
 550 Analysis, and Data Interpretation. *Environ. Sci. Technol.* 2021, 55, 9637–9656.
 551 https://doi.org/10.1021/acs.est.1c01135.
- (5) Zsolnay, Á. Dissolved Organic Matter: Artefacts, Definitions, and Functions. *Geoderma* 2003, 113, 187–209. https://doi.org/10.1016/S0016-7061(02)00361-0.
- Wells, M. J. M.; Stretz, H. A. Supramolecular Architectures of Natural Organic Matter. *Sci. Total Environ.* 2019, 671, 1125–1133. https://doi.org/10.1016/j.scitotenv.2019.03.406.
- Prescott, C. E.; Grayston, S. J.; Helmisaari, H. S.; Kaštovská, E.; Körner, C.; Lambers, H.;
 Meier, I. C.; Millard, P.; Ostonen, I. Surplus Carbon Drives Allocation and Plant–Soil
 Interactions. *Trends Ecol. Evol.* 2020, 35, 1110–1118.
 https://doi.org/10.1016/j.tree.2020.08.007.
- Leinemann, T.; Preusser, S.; Mikutta, R.; Kalbitz, K.; Cerli, C.; Höschen, C.; Mueller, C.
 W.; Kandeler, E.; Guggenberger, G. Multiple Exchange Processes on Mineral Surfaces
 Control the Transport of Dissolved Organic Matter through Soil Profiles. *Soil Biol. Biochem.* **2018**, *118*, 79–90. https://doi.org/10.1016/j.soilbio.2017.12.006.
- Lange, M.; Roth, V. N.; Eisenhauer, N.; Roscher, C.; Dittmar, T.; Fischer-Bedtke, C.;
 González Macé, O.; Hildebrandt, A.; Milcu, A.; Mommer, L.; et al. Plant Diversity Enhances
 Production and Downward Transport of Biodegradable Dissolved Organic Matter. *J. Ecol.*2020, No. February, 1–14. https://doi.org/10.1111/1365-2745.13556.
- (10) Lehmann, J.; Hansel, C. M.; Kaiser, C.; Kleber, M.; Maher, K.; Manzoni, S.; Nunan, N.;
 Reichstein, M.; Schimel, J. P.; Torn, M. S.; et al. Persistence of Soil Organic Carbon Caused
 by Functional Complexity. *Nat. Geosci.* 2020, *13*, 529–534. https://doi.org/10.1038/s41561020-0612-3.
- 572 (11) Bünemann, E. K.; Bongiorno, G.; Bai, Z.; Creamer, R. E.; De Deyn, G.; de Goede, R.;
 573 Fleskens, L.; Geissen, V.; Kuyper, T. W.; Mäder, P.; et al. Soil Quality A Critical Review.
 574 Soil Biol. Biochem. 2018, 120, 105–125. https://doi.org/10.1016/j.soilbio.2018.01.030.
- 575 (12) Simpson, A. J.; Simpson, M. J.; Soong, R. Environmental Nuclear Magnetic Resonance
 576 Spectroscopy: An Overview and a Primer. *Anal. Chem.* 2018, 90, 628–639.
 577 https://doi.org/10.1021/acs.analchem.7b03241.
- 578 (13) Lu, K.; Li, X.; Chen, H.; Liu, Z. Constraints on Isomers of Dissolved Organic Matter in
 579 Aquatic Environments: Insights from Ion Mobility Mass Spectrometry. *Geochim.*580 Cosmochim. Acta 2021, 308, 353–372. https://doi.org/10.1016/j.gca.2021.05.007.
- (14) Leyva, D.; Tariq, M. U.; Jaffé, R.; Saeed, F.; Lima, F. F. Unsupervised Structural
 Classification of Dissolved Organic Matter Based on Fragmentation Pathways. *Environ. Sci.*

- 583 *Technol.* **2022**, *56*, 1458–1468. https://doi.org/10.1021/acs.est.1c04726.
- 584 Wegley, L.; Nelson, C. E.; Petras, D.; Koester, I.; Quinlan, Z. A.; Arts, M. G. I.; Nothias, (15)585 L.; Comstock, J.; White, B. M.; Hopmans, E. C. Distinguishing the Molecular Diversity, 586 Nutrient Content, and Energetic Potential of Exometabolomes Produced by Macroalgae and 587 **Reef-Building** Corals. PNAS 2022, 119, e2110283119. 588 https://doi.org/10.1073/pnas.2110283119/-/DCSupplemental.Published.
- (16) Kurek, M. R.; Poulin, B. A.; Mckenna, A. M.; Spencer, R. G. M. Deciphering Dissolved
 Organic Matter: Ionization, Dopant, and Fragmentation Insights via Fourier Transform-Ion
 Cyclotron Resonance Mass Spectrometry. *Environ. Sci. Technol.* 2020, *54*, 16249–16259.
 https://doi.org/10.1021/acs.est.0c05206.
- (17) Arakawa, N.; Aluwihare, L. I.; Simpson, A. J.; Soong, R.; Stephens, B. M.; Lane-Coplen,
 D. Carotenoids Are the Likely Precursor of a Significant Fraction of Marine Dissolved
 Organic Matter. *Sci. Adv.* 2017, *3* (9), e1602976. https://doi.org/10.1126/sciadv.1602976.
- 596 (18) Brown, T. A.; Jackson, B. A.; Bythell, B. J.; Stenson, A. C. Benefits of Multidimensional
 597 Fractionation for the Study and Characterization of Natural Organic Matter. *J. Chromatogr.* 598 A 2016, 1470, 84–96. https://doi.org/10.1016/j.chroma.2016.10.005.
- (19) Hawkes, J. A.; Patriarca, C.; Sjöberg, P. J. R.; Tranvik, L. J.; Bergquist, J. Extreme Isomeric
 Complexity of Dissolved Organic Matter Found across Aquatic Environments. *Limnol. Oceanogr. Lett.* 2018, *3* (2), 21–30. https://doi.org/10.1002/lol2.10064.
- 602 Mostovaya, A.; Hawkes, J. A.; Koehler, B.; Dittmar, T.; Tranvik, L. J. Emergence of the (20)603 Reactivity Continuum of Organic Matter from Kinetics of a Multitude of Individual 604 Molecular Constituents. Environ. Sci. Technol. 2017, 51. 11571-11579. 605 https://doi.org/10.1021/acs.est.7b02876.
- Murphy, K. R.; Timko, S. A.; Gonsior, M.; Powers, L. C.; Wünsch, U. J.; Stedmon, C. A.
 Photochemistry Illuminates Ubiquitous Organic Matter Fluorescence Spectra. *Environ. Sci. Technol.* 2018, *52*, 11243–11250. https://doi.org/10.1021/acs.est.8b02648.
- 609 (22) Benner, R.; Amon, R. M. W. The Size-Reactivity Continuum of Major Bioelements in the
 610 Ocean. Ann. Rev. Mar. Sci. 2014, No. July 2014, 1–21. https://doi.org/10.1146/annurev611 marine-010213-135126.
- 612 (23) Hertkorn, N.; Frommberger, M.; Witt, M.; Koch, B. P.; Schmitt-Kopplin, P.; Perdue, E. M.
 613 Natural Organic Matter and the Event Horizon of Mass Spectrometry. *Anal. Chem.* 2008,
 614 80, 8908–8919.
- (24) van Agthoven, M. A.; Lam, Y. P. Y.; O'Connor, P. B.; Rolando, C.; Delsuc, M. A. TwoDimensional Mass Spectrometry: New Perspectives for Tandem Mass Spectrometry. *Eur. Biophys. J.* 2019, No. 48, 213–229. https://doi.org/10.1007/s00249-019-01348-5.
- (25) Leyva, D.; Jaffe, R.; Fernandez-Lima, F. Structural Characterization of Dissolved Organic
 Matter at the Chemical Formula Level Using TIMS-FT-ICR MS/MS. *Anal. Chem.* 2020, *92*,
 11960–11966. https://doi.org/10.1021/acs.analchem.0c02347.
- 621 (26) Zhang, F.; Harir, M.; Moritz, F.; Zhang, J.; Witting, M.; Wu, Y.; Schmitt-Kopplin, P.;

- Fekete, A.; Gaspar, A.; Hertkorn, N. Molecular and Structural Characterization of Dissolved
 Organic Matter during and Post Cyanobacterial Bloom in Taihu by Combination of NMR
 Spectroscopy and FTICR Mass Spectrometry. *Water Res.* 2014, 57C, 280–294.
 https://doi.org/10.1016/j.watres.2014.02.051.
- (27) Petras, D.; Minich, J. J.; Cancelada, L. C.; Torres, R. E.; Kunselman, E.; Wang, M.; White,
 M. E.; Allen, E. E.; Prather, K. A.; Aluwihare, L. I.; et al. Non-Targeted Tandem Mass
 Spectrometry Enables the Visualization of Organic Matter Chemotype Shifts in Coastal
 Seawater. *Chemosphere* 2021, 271, 129450.
 https://doi.org/10.1016/j.chemosphere.2020.129450.
- (28) Leenheer, J. A.; Rostad, C. E.; Gates, P. M.; Furlong, E. T.; Ferrer, I. Molecular Resolution and Fragmentation of Fulvic Acid by Electrospray Ionization/ Multistage Tandem Mass Spectrometry. *Anal. Chem.* 2001, *73* (7), 1461–1471. https://doi.org/10.1021/ac0012593.
- 634 (29) Dit Fouque, D. J.; Maroto, A.; Memboeuf, A. Purification and Quantification of an Isomeric
 635 Compound in a Mixture by Collisional Excitation in Multistage Mass Spectrometry
 636 Experiments. *Anal. Chem.* 2016, 88 (22), 10821–10825.
 637 https://doi.org/10.1021/acs.analchem.6b03490.
- 638 Petras, D.; Koester, I.; Da Silva, R.; Stephens, B. M.; Haas, A. F.; Nelson, C. E.; Kelly, L. (30)639 W.; Aluwihare, L. I.; Dorrestein, P. C. High-Resolution Liquid Chromatography Tandem Mass Spectrometry Enables Large Scale Molecular Characterization of Dissolved Organic 640 641 Front. 2017, (December), 406. Matter. Mar. Sci. 4 642 https://doi.org/10.3389/fmars.2017.00405.
- (31) Zark, M.; Dittmar, T. Universal Molecular Structures in Natural Dissolved Organic Matter.
 Nat. Commun. 2018, 9 (1), 3178. https://doi.org/10.1038/s41467-018-05665-9.
- 645 (32)Lu, K.; Gardner, W. S.; Liu, Z. Molecular Structure Characterization of Riverine and Coastal 646 Dissolved Organic Matter with Ion Mobility Quadrupole Time-of-Flight LCMS (IM Q-TOF 647 Environ. Technol. LCMS). Sci. 2018. 52 (13), 7182–7191. 648 https://doi.org/10.1021/acs.est.8b00999.
- (33) Witt, M.; Fuchser, J.; Koch, B. P. Fragmentation Studies of Fulvic Acids Using Collision
 Induced Dissociation Fourier Transform Ion Cyclotron Resonance Mass Spectrometry.
 Anal. Chem. 2009, *81* (7), 2688–2694. https://doi.org/10.1021/ac802624s.
- (34) Reemtsma, T.; These, A.; Linscheid, M.; Leenheer, J.; Spitzy, A. Molecular and Structural
 Characterization of Dissolved Organic Matter from the Deep Ocean by FTICR-MS,
 Including Hydrophilic Nitrogenous Organic Molecules. *Environ. Sci. Technol.* 2008, 42,
 1430–1437. https://doi.org/10.1021/es7021413.
- 656 (35) Capley, E. N.; Tipton, J. D.; Marshall, A. G.; Stenson, A. C. Chromatographic Reduction of 657 Isobaric and Isomeric Complexity of Fulvic Acids to Enable Multistage Tandem Mass Spectral 2010. 658 Characterization. Anal. Chem. 82 (19), 8194-8202. 659 https://doi.org/10.1021/ac1016216.
- (36) Cortés-Francisco, N.; Caixach, J. Fragmentation Studies for the Structural Characterization
 of Marine Dissolved Organic Matter. *Anal. Bioanal. Chem.* 2015, 407, 2455–2462.

- 662 https://doi.org/10.1007/s00216-015-8499-3.
- (37) These, A.; Winkler, M.; Thomas, C.; Reemtsma, T. Determination of Molecular Formulas and Structural Regularities of Low Molecular Weight Fulvic Acids by Size-Exclusion Chromatography with Electrospray Ionization Quadrupole Time-of-Flight Mass Spectrometry. *Rapid Commun. Mass Spectrom.* 2004, *18* (16), 1777–1786. https://doi.org/10.1002/rcm.1550.
- (38) Stenson, A. C.; Marshall, A. G.; Cooper, W. T. Exact Masses and Chemical Formulas of Individual Suwannee River Fulvic Acids from Ultrahigh Resolution Electrospray Ionization Fourier Transform Ion Cyclotron Resonance Mass Spectra Molecular Formulas Have Been Assigned for 4626 Indi- Mass Measurements Fr. *Anal. Chem.* 2003, 75, 1275–1284. https://doi.org/10.1021/ac026106p.
- (39) Nimmagadda, R. D.; McRae, C. Characterisation of the Backbone Structures of Several
 Fulvic Acids Using a Novel Selective Chemical Reduction Method. *Org. Geochem.* 2007,
 38 (7), 1061–1072. https://doi.org/10.1016/j.orggeochem.2007.02.016.
- (40) Perdue, E. M.; Hertkorn, N.; Kettrup, A. Substitution Patterns in Aromatic Rings by Increment Analysis. Model Development and Application to Natural Organic Matter. *Anal. Chem.* 2007, *79* (3), 1010–1021. https://doi.org/10.1021/ac061611y.
- 679 Kunenkov, E. V.; Kononikhin, A. S.; Perminova, I. V.; Hertkorn, N.; Gaspar, A.; Schmitt-(41) 680 kopplin, P.; Popov, I. A.; Garmash, A. V.; Nikolaev, E. N. Total Mass Difference Statistics 681 Algorithm: A New Approach to Identification of High-Mass Building Blocks in Electrospray Ionization Fourier Transform Ion Cyclotron Mass Spectrometry Data of 682 683 Natural Organic Matter. Anal. Chem. 2009, 81 (24), 10106-10115. 684 https://doi.org/10.1021/ac901476u.
- (42) Zherebker, A. Y.; Airapetyan, D.; Konstantinov, A. I.; Kostyukevich, Y. I.; Kononikhin, A.
 S.; Popov, I. A.; Zaitsev, K. V.; Nikolaev, E. N.; Perminova, I. V. Synthesis of Model Humic
 Substances: A Mechanistic Study Using Controllable H/D Exchange and Fourier Transform
 Ion Cyclotron Resonance Mass Spectrometry. *Analyst* 2015, *140* (13), 4708–4719.
 https://doi.org/10.1039/c5an00602c.
- 690 (43) Bell, N. G. A.; Michalchuk, A. A. L.; Blackburn, J. W. T.; Graham, M. C.; Uhrín, D. Isotope691 Filtered 4D NMR Spectroscopy for Structure Determination of Humic Substances. *Angew.*692 *Chemie Int. Ed.* 2015, *54* (29), 8382–8385. https://doi.org/10.1002/anie.201503321.
- 693 McIntyre, C.; McRae, C.; Jardine, D.; Batts, B. D. Identification of Compound Classes in (44)694 Soil and Peat Fulvic Acids as Observed by Electrospray Ionization Tandem Mass 695 Spectrometry. Rapid Commun. Mass Spectrom. 2002. 16. 1604–1609. 696 https://doi.org/10.1002/rcm.761.
- (45) Zark, M.; Christoffers, J.; Dittmar, T. Molecular Properties of Deep-Sea Dissolved Organic
 Matter Are Predictable by the Central Limit Theorem: Evidence from Tandem FT-ICR-MS.
 Mar. Chem. 2017, 191, 9–15. https://doi.org/10.1016/j.marchem.2017.02.005.
- (46) Stenson, A. C.; Ruddy, B. M.; Bythell, B. J. Ion Molecule Reaction H/D Exchange as a
 Probe for Isomeric Fractionation in Chromatographically Separated Natural Organic Matter.

- 702 Int. J. Mass Spectrom. 2014, 360, 45–53. https://doi.org/10.1016/j.ijms.2013.12.026.
- (47) Kostyukevich, Y.; Kononikhin, A.; Zherebker, A.; Popov, I.; Perminova, I.; Nikolaev, E.
 Funderation of Non-Labile Oxygen Atoms in Dissolved Organic Matter by Use of 16O/18O
 Exchange and Fourier Transform Ion-Cyclotron Resonance Mass Spectrometry. *Anal. Bioanal. Chem.* 2014, 406 (26), 6655–6664. https://doi.org/10.1007/s00216-014-8097-9.
- 707 (48) Zherebker, A.; Kostyukevich, Y.; Kononikhin, A.; Kharybin, O.; Konstantinov, A. I.;
 708 Zaitsev, K. V.; Nikolaev, E.; Perminova, I. V. Enumeration of Carboxyl Groups Carried on
 709 Individual Components of Humic Systems Using Deuteromethylation and Fourier
 710 Transform Mass Spectrometry. *Anal. Bioanal. Chem.* 2017, 409, 2477–2488.
 711 https://doi.org/10.1007/s00216-017-0197-x.
- 712 (49) Liu, Z.; Sleighter, R. L.; Zhong, J.; Hatcher, P. G. The Chemical Changes of DOM from 713 Black Waters to Coastal Marine Waters by HPLC Combined with Ultrahigh Resolution 714 Mass Spectrometry. Estuar. Coast. Shelf Sci. 2011, 92, 205-216. 715 https://doi.org/10.1016/j.ecss.2010.12.030.
- (50) Dier, T. K. F.; Egele, K.; Fossog, V.; Hempelmann, R.; Volmer, D. A. Enhanced Mass
 Defect Filtering to Simplify and Classify Complex Mixtures of Lignin Degradation
 Products. *Anal. Chem.* 2016, 88, 1328–1335.
 https://doi.org/10.1021/acs.analchem.5b03790.
- (51) Fabre, N.; Rustan, I.; De Hoffmann, E.; Quetin-Leclercq, J. Determination of Flavone,
 Flavonol, and Flavanone Aglycones by Negative Ion Liquid Chromatography Electrospray
 Ion Trap Mass Spectrometry. J. Am. Soc. Mass Spectrom. 2001, 12 (6), 707–715.
 https://doi.org/10.1016/S1044-0305(01)00226-4.
- (52) Engström, M. T.; Pälijärvi, M.; Salminen, J. P. Rapid Fingerprint Analysis of Plant Extracts
 for Ellagitannins, Gallic Acid, and Quinic Acid Derivatives and Quercetin-, Kaempferoland Myricetin-Based Flavonol Glycosides by UPLC-QqQ-MS/MS. *J. Agric. Food Chem.*2015, 63 (16), 4068–4079. https://doi.org/10.1021/acs.jafc.5b00595.
- Miketova, P.; Schram, K. H.; Whitney, J.; Li, M.; Huang, R.; Kerns, E.; Valcic, S.;
 Timmermann, B. N.; Rourick, R.; Klohr, S. Tandem Mass Spectrometry Studies of Green
 Tea Catechins. Identification of Three Minor Components in the Polyphenolic Extract of
 Green Tea. J. Mass Spectrom. 2000, 35 (7), 860–869. https://doi.org/10.1002/10969888(200007)35:7<860::AID-JMS10>3.0.CO;2-J.
- (54) Luek, J. L.; Schmitt-kopplin, P.; Mouser, P. J.; Petty, W. T.; Richardson, S. D.; Gonsior, M. Halogenated Organic Compounds Identified in Hydraulic Fracturing Wastewaters Using Ultrahigh Resolution Mass Spectrometry. *Environ. Sci. Technol.* 2017, *51*, 5377–5385.
 https://doi.org/10.1021/acs.est.6b06213.
- (55) Reemtsma, T. The Carbon versus Mass Diagram to Visualize and Exploit FTICR-MS Data
 of Natural Organic Matter. J. Mass Spectrom. 2010, 45 (4), 382–390.
 https://doi.org/10.1002/jms.1722.
- (56) Rivas-Ubach, A.; Liu, Y.; Bianchi, T. S.; Tolić, N.; Jansson, C.; Paša-Tolić, L. Moving
 beyond the van Krevelen Diagram: A New Stoichiometric Approach for Compound

- 742
 Classification in Organisms. Anal. Chem. 2018, 90, 6152–6160.

 743
 https://doi.org/10.1021/acs.analchem.8b00529.
- (57) Davies, N. W.; Sandron, S.; Nesterenko, P.; Paull, B.; Wilson, R.; Haddad, P.; Shellie, R.;
 Rojas, A. Comment on "Structural Characterization of Dissolved Organic Matter: A Review
 of Current Techniques for Isolation and Analysis" by E. C. Minor, M. M. Swenson, B. M.
 Mattson, and A. R. Oyler, Environ. Sci.: Processes Impacts, 2014, 16, 2064. *Environ. Sci. Process. Impacts* 2015, *17* (2), 495. https://doi.org/10.1039/C4EM00631C.
- Minor, E. C.; Swenson, M. M.; Mattson, B. M.; Oyler, A. R. Structural Characterization of Dissolved Organic Matter: A Review of Current Techniques for Isolation and Analysis. *Environ.* Sci. Process. Impacts 2014, 16, 2064–2079. https://doi.org/10.1039/C4EM00062E.
- (59) Roth, V.-N.; Dittmar, T.; Gaupp, R.; Gleixner, G. Ecosystem-Specific Composition of
 Dissolved Organic Matter. Vadose Zo. J. 2014, 13.
 https://doi.org/http://dx.doi.org/10.2136/vzj2013.09.0162.
- (60) Green, N. W.; Mcinnis, D.; Hertkorn, N.; Maurice, P. A.; Perdue, M. E. Suwannee River
 Natural Organic Matter : Isolation of the 2R101N Reference Sample by Reverse Osmosis. *Environ. Eng. Sci.* 2014, *32*, 38–44. https://doi.org/10.1089/ees.2014.0284.
- (61) Simon, C.; Roth, V.-N.; Dittmar, T.; Gleixner, G. Molecular Signals of Heterogeneous
 Terrestrial Environments Identified in Dissolved Organic Matter: A Comparative Analysis
 of Orbitrap and Ion Cyclotron Resonance Mass Spectrometers. *Front. Earth Sci.* 2018, *6*, 1–
 16. https://doi.org/10.3389/feart.2018.00138.
- Merder, J.; Freund, J. A.; Feudel, U.; Hansen, C. T.; Hawkes, J. A.; Jacob, B.; Klaproth, K.;
 Niggemann, J.; Noriega-Ortega, B. E.; Osterholz, H.; et al. ICBM-OCEAN: Processing
 Ultrahigh-Resolution Mass Spectrometry Data of Complex Molecular Mixtures. *Anal. Chem.* 2020, *92*, 6832–6838. https://doi.org/10.1021/acs.analchem.9b05659.
- 767 (63) Dührkop, K.; Fleischauer, M.; Ludwig, M.; Aksenov, A. A.; Melnik, A. V.; Meusel, M.;
 768 Dorrestein, P. C.; Rousu, J.; Böcker, S. SIRIUS 4: A Rapid Tool for Turning Tandem Mass
 769 Spectra into Metabolite Structure Information. *Nat. Methods* 2019, *16*, 299–302.
 770 https://doi.org/10.1038/s41592-019-0344-8.
- 771 Horai, H.; Arita, M.; Kanaya, S.; Nihei, Y.; Ikeda, T.; Suwa, K.; Ojima, Y.; Tanaka, K.; (64)772 Tanaka, S.; Aoshima, K.; et al. MassBank: A Public Repository for Sharing Mass Spectral 773 for Sciences. Mass Spectrom. 703-714. Data Life J 2010, 45. 774 https://doi.org/10.1002/jms.1777.
- (65) Wang, M.; Carver, J. J.; Phelan, V. V.; Sanchez, L. M.; Garg, N.; Peng, Y.; Nguyen, D. T.
 D. D.; Watrous, J.; Kapono, C. A.; Luzzatto-Knaan, T.; et al. Sharing and Community Curation of Mass Spectrometry Data with Global Natural Products Social Molecular Networking. *Nat. Biotechnol.* 2016, *34*, 828–837. https://doi.org/10.1038/nbt.3597.
- (66) Djoumbou Feunang, Y.; Eisner, R.; Knox, C.; Chepelev, L.; Hastings, J.; Owen, G.; Fahy,
 E.; Steinbeck, C.; Subramanian, S.; Bolton, E.; et al. ClassyFire: Automated Chemical
 Classification with a Comprehensive, Computable Taxonomy. J. Cheminform. 2016, 8, 61.

- 782 https://doi.org/10.1186/s13321-016-0174-y.
- (67) Hammer, Ø.; Harper, D. A.; Ryan, P. D. PAST: Paleontological Statistics Software Package
 for Education and Data Analysis. *Palaeontol. Electron.* 2001, *4*, 9.
- (68) Milstead, R. P.; Remucal, C. K. Molecular-Level Insights into the Formation of Traditional and Novel Halogenated Disinfection Byproducts. *ACS ES&T Water* 2021, *1*, 1966–1974. https://doi.org/10.1021/acsestwater.1c00161.
- 788 (69) Wilson, R. M.; Tfaily, M. M.; Kolton, M.; Johnston, E. R.; Petro, C.; Zalman, C. A.; Hanson, 789 P. J.; Heyman, H. M.; Kyle, J. E.; Hoyt, D. W.; et al. Soil Metabolome Response to Whole-790 Ecosystem Warming at the Spruce and Peatland Responses under Changing Environments 791 Experiment. Proc. Natl. Acad. Sci. U_{\cdot} A. S. 2021. 118. 1–11. 792 https://doi.org/10.1073/pnas.2004192118.
- (70) Wu, S.; You, F.; Boughton, B.; Liu, Y.; Nguyen, T. A. H.; Wykes, J.; Southam, G.;
 Robertson, L. M.; Chan, T. S.; Lu, Y. R.; et al. Chemodiversity of Dissolved Organic Matter
 and Its Molecular Changes Driven by Rhizosphere Activities in Fe Ore Tailings Undergoing
 Eco-Engineered Pedogenesis. *Environ. Sci. Technol.* 2021, 55, 13045–13060.
 https://doi.org/10.1021/acs.est.1c04527.
- (71) Hawkes, J. A.; D'Andrilli, J.; Agar, J. N.; Barrow, M. P.; Berg, S. M.; Catalán, N.; Chen,
 H.; Chu, R. K.; Cole, R. B.; Dittmar, T.; et al. An International Laboratory Comparison of
 Dissolved Organic Matter Composition by High Resolution Mass Spectrometry: Are We
 Getting the Same Answer? *Limnol. Oceanogr. Methods* 2020, *18*, 235–258.
- (72) Chassagne, F.; Cabanac, G.; Hubert, G.; David, B.; Marti, G. The Landscape of Natural
 Product Diversity and Their Pharmacological Relevance from a Focus on the Dictionary of
 Natural Products[®]. *Phytochem. Rev.* 2019, 1–22. https://doi.org/10.1007/s11101-01909606-2.
- 806 (73) Nakamura, Y.; Mochamad Afendi, F.; Kawsar Parvin, A.; Ono, N.; Tanaka, K.; Hirai
 807 Morita, A.; Sato, T.; Sugiura, T.; Altaf-Ul-Amin, M.; Kanaya, S. KNApSAcK Metabolite
 808 Activity Database for Retrieving the Relationships between Metabolites and Biological
 809 Activities. *Plant Cell Physiol.* 2014, 55, e7. https://doi.org/10.1093/pcp/pct176.
- (74) Caspi, R.; Billington, R.; Keseler, I. M.; Kothari, A.; Krummenacker, M.; Midford, P. E.;
 Ong, W. K.; Paley, S.; Subhraveti, P.; Karp, P. D. The MetaCyc Database of Metabolic
 Pathways and Enzymes a 2019 Update. *Nucleic Acids Res.* 2019, 48, D455–D453.
 https://doi.org/10.1093/nar/gkz862.
- 814 (75) Okuda, S.; Yamada, T.; Hamajima, M.; Itoh, M.; Katayama, T.; Bork, P.; Goto, S.;
 815 Kanehisa, M. KEGG Atlas Mapping for Global Analysis of Metabolic Pathways. *Nucleic*816 *Acids Res.* 2008, *36*, 423–426. https://doi.org/10.1093/nar/gkn282.
- (76) Wishart, D. S.; Tzur, D.; Knox, C.; Eisner, R.; Guo, A. C.; Young, N.; Cheng, D.; Jewell,
 K.; Arndt, D.; Sawhney, S.; et al. HMDB: The Human Metabolome Database. *Nucleic Acids Res.* 2007, *35*, 521–526. https://doi.org/10.1093/nar/gkl923.
- 820 (77) Jeffryes, J. G.; Colastani, R. L.; Elbadawi-Sidhu, M.; Kind, T.; Niehaus, T. D.; Broadbelt,

- L. J.; Hanson, A. D.; Fiehn, O.; Tyo, K. E. J.; Henry, C. S. MINEs: Open Access Databases
 of Computationally Predicted Enzyme Promiscuity Products for Untargeted Metabolomics. *J. Cheminform.* 2015, 7, 44. https://doi.org/10.1186/s13321-015-0087-1.
- 824 (78) Baumeister, T. U. H.; Ueberschaar, N.; Pohnert, G. Gas-Phase Chemistry in the GC Orbitrap
 825 Mass Spectrometer. J. Am. Soc. Mass Spectrom. 2018. https://doi.org/10.1007/s13361-018826 2117-5.
- (79) Smirnov, K. S.; Forcisi, S.; Moritz, F.; Lucio, M.; Schmitt-Kopplin, P. Mass Difference
 Maps and Their Application for the Re-Calibration of Mass Spectrometric Data in Non-Targeted Metabolomics. *Anal. Chem.* 2019. https://doi.org/10.1021/acs.analchem.8b04555.
- (80) Adair, E.; Afonso, C.; Bell, N. G. A.; Davies, A. N.; Delsuc, M.-A.; Godfrey, R.; Goodacre,
 R.; Hawkes, J. A.; Hertkorn, N.; Jones, D.; et al. High Resolution Techniques: General
 Discussion. *Faraday Discuss.* 2019, *218*, 247–267. https://doi.org/10.1039/c9fd90045d.
- (81) Novotny, N. R.; Capley, E. N.; Stenson, A. C. Fact or Artifact: The Representativeness of
 ESI-MS for Complex Natural Organic Mixtures. J. Mass Spectrom. 2014, 49 (4), 316–326.
 https://doi.org/10.1002/jms.3345.
- (82) Lam, B.; Baer, A.; Alaee, M.; Lefebvre, B.; Moser, A.; Williams, A.; Simpson, A. J. Major
 Structural Components in Freshwater Dissolved Organic Matter. *Environ. Sci. Technol.*2007, 41, 8240–8247. https://doi.org/10.1021/es0713072.
- (83) Hertkorn, N.; Benner, R.; Frommberger, M.; Schmitt-Kopplin, P.; Witt, M.; Kaiser, K.;
 Kettrup, A.; Hedges, J. I. Characterization of a Major Refractory Component of Marine
 Dissolved Organic Matter. *Geochim. Cosmochim. Acta* 2006, 70, 2990–3010.
 https://doi.org/10.1016/j.gca.2006.03.021.
- (84) Liu, L.; Song, C.; Tian, S.; Zhang, Q.; Cai, X.; Liu, Y.; Liu, Z.; Wang, W. Structural
 Characterization of Sulfur-Containing Aromatic Compounds in Heavy Oils by FT-ICR
 Mass Spectrometry with a Narrow Isolation Window. *Fuel* 2019, 240, 40–48.
 https://doi.org/10.1016/j.fuel.2018.11.130.
- 847 (85) Böcker, S.; Dührkop, K. Fragmentation Trees Reloaded. J. Cheminform. 2016, 8, 5.
 848 https://doi.org/10.1007/978-3-319-16706-0_10.
- (86) Cao, D.; Lv, J.; Geng, F.; Rao, Z.; Niu, H.; Shi, Y.; Cai, Y.; Kang, Y. Ion Accumulation Time Dependent Molecular Characterization of Natural Organic Matter Using Electrospray Ionization-Fourier Transform Ion Cyclotron Resonance Mass Spectrometry. *Anal. Chem.*2016, 88, 12210–12218. https://doi.org/10.1021/acs.analchem.6b03198.
- (87) Lemr, K.; Holčapek, M.; Jandera, P.; Lyka, A. Analysis of Metal Complex Azo Dyes by
 High-Performance Liquid Chromatography/Electrospray Ionization Mass Spectrometry and
 Multistage Mass Spectrometry. *Rapid Commun. Mass Spectrom.* 2000, 14, 1881–1888.
- 856 (88) Piraud, M.; Vianey-Saban, C.; Petritis, K.; Elfakir, C.; Steghens, J. P.; Morla, A.; Bouchu,
 857 D. ESI-MS/MS Analysis of Underivatised Amino Acids: A New Tool for the Diagnosis of
 858 Inherited Disorders of Amino Acid Metabolism. Fragmentation Study of 79 Molecules of
 859 Biological Interest in Positive and Negative Ionisation Mode. *Rapid Commun. Mass*

- 860 Spectrom. 2003, 17, 1297–1311. https://doi.org/10.1002/rcm.1054.
- (89) Pohlabeln, A. M.; Dittmar, T. Novel Insights into the Molecular Structure of Non-Volatile
 Marine Dissolved Organic Sulfur. *Mar. Chem.* 2015, *168*, 86–94.
 https://doi.org/10.1016/j.marchem.2014.10.018.
- Wagner, S.; Dittmar, T.; Jaffé, R. Molecular Characterization of Dissolved Black Nitrogen via Electrospray Ionization Fourier Transform Ion Cyclotron Resonance Mass Spectrometry. Org. Geochem. 2015, 79, 21–30. https://doi.org/10.1016/j.orggeochem.2014.12.002.
- 868 (91) Poulin, B. A.; Ryan, J. N.; Nagy, K. L.; Stubbins, A.; Dittmar, T.; Orem, W.; Krabbenhoft,
 869 D. P.; Aiken, G. R. Spatial Dependence of Reduced Sulfur in Everglades Dissolved Organic
 870 Matter Controlled by Sulfate Enrichment. *Environ. Sci. Technol.* 2017, *51*, 3630–3639.
 871 https://doi.org/10.1021/acs.est.6b04142.
- (92) Roth, V.-N.; Lange, M.; Simon, C.; Hertkorn, N.; Bucher, S.; Goodall, T.; Griffiths, R. I.;
 Mellado-Vázquez, P. G.; Mommer, L.; Oram, N. J.; et al. Persistence of Dissolved Organic
 Matter Explained by Molecular Changes during Its Passage through Soil. *Nat. Geosci.* 2019, *12*, 755–761. https://doi.org/10.1038/s41561-019-0417-4.
- Warren, C. R. High Diversity of Small Organic N Observed in Soil Water. Soil Biol. *Biochem.* 2013, 57, 444–450. https://doi.org/10.1016/j.soilbio.2012.09.025.
- Woods, G.; Simpson, M.; Koerner, P. J.; Napoli, A.; Simpson, A. HILIC-NMR: Toward the Identification of Individual Molecular Components in Dissolved Organic Matter. *Environ. Sci. Technol.* 2011, 45 (13), 5910. https://doi.org/10.1021/es201716u.
- 881 (95) Geiger, T.; Cox, J.; Mann, M. Proteomics on an Orbitrap Benchtop Mass Spectrometer
 882 Using All-Ion Fragmentation. *Mol. Cell. Proteomics* 2010, 9, 2252–2261.
 883 https://doi.org/10.1074/mcp.M110.001537.
- 884 (96) Naz, S.; Gallart-Ayala, H.; Reinke, S. N.; Mathon, C.; Blankley, R.; Chaleckis, R.;
 885 Wheelock, C. E. Development of a Liquid Chromatography-High Resolution Mass
 886 Spectrometry Metabolomics Method with High Specificity for Metabolite Identification
 887 Using All Ion Fragmentation Acquisition. *Anal. Chem.* 2017, *89*, 7933–7942.
 888 https://doi.org/10.1021/acs.analchem.7b00925.
- (97) Ludwig, C.; Gillet, L.; Rosenberger, G.; Amon, S.; Collins, B. C.; Aebersold, R. Dataindependent Acquisition-based SWATH - MS for Quantitative Proteomics: A Tutorial. *Mol. Syst. Biol.* 2018, 14, 1–23. https://doi.org/10.15252/msb.20178126.
- (98) Lechtenfeld, O. J.; Hertkorn, N.; Shen, Y.; Witt, M.; Benner, R. Marine Sequestration of Carbon in Bacterial Metabolites. *Nat. Commun.* 2015, 6, 6711.
 https://doi.org/10.1038/ncomms7711.
- (99) Mentges, A.; Feenders, C.; Seibt, M.; Blasius, B.; Dittmar, T. Functional Molecular Diversity of Marine Dissolved Organic Matter Is Reduced during Degradation. *Front. Mar. Sci.* 2017, *4*, 194. https://doi.org/10.3389/fmars.2017.00194.
- 898 (100) Hertkorn, N.; Ruecker, C.; Meringer, M.; Gugisch, R.; Frommberger, M.; Perdue, E. M.;

- Witt, M.; Schmitt-Kopplin, P. High-Precision Frequency Measurements: Indispensable
 Tools at the Core of the Molecular-Level Analysis of Complex Systems. *Anal. Bioanal. Chem.* 2007, 389, 1311–1327. https://doi.org/10.1007/s00216-007-1577-4.
- (101) Lechtenfeld, O. J.; Kattner, G.; Flerus, R.; McCallister, S. L.; Schmitt-Kopplin, P.; Koch, B.
 P. Molecular Transformation and Degradation of Refractory Dissolved Organic Matter in the Atlantic and Southern Ocean. *Geochim. Cosmochim. Acta* 2014, *126*, 321–337. https://doi.org/10.1016/j.gca.2013.11.009.
- 906 (102) Dührkop, K.; Nothias, L. F.; Fleischauer, M.; Reher, R.; Ludwig, M.; Hoffmann, M. A.;
 907 Petras, D.; Gerwick, W. H.; Rousu, J.; Dorrestein, P. C.; et al. Systematic Classification of
 908 Unknown Metabolites Using High-Resolution Fragmentation Mass Spectra. *Nat.*909 *Biotechnol.* 2020. https://doi.org/10.1038/s41587-020-0740-8.
- (103) Rogers, S.; Wei Ong, C.; Wandy, J.; Ernst, M.; Ridder, L.; van der Hooft, J. J. J. Deciphering
 Complex Metabolite Mixtures by Unsupervised and Supervised Substructure Discovery and
 Semi-Automated Annotation from MS/MS Spectra. *Faraday Discuss.* 2019, *218*, 284–302.
 https://doi.org/10.1039/c8fd00235e.
- (104) Wolfender, J.-L.; Nuzillard, J.-M.; Van Der Hooft, J. J. J.; Renault, J.-H.; Bertrand, S.
 Accelerating Metabolite Identification in Natural Product Research: Toward an Ideal Combination of Liquid Chromatography-High-Resolution Tandem Mass Spectrometry and NMR Profiling, in Silico Databases, and Chemometrics. *Anal. Chem.* 2019, *91*, 704–742. https://doi.org/10.1021/acs.analchem.8b05112.

919