Asymmetric Defluoroallylation of 4-Trifluoromethylpyridines Enabled by Umpolung C-F Bond Activation

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Center of Basic Molecular Science, Department of Chemistry, Tsinghua University, Beijing 10084, China *C-F bond activation, N-boryl pyridiyl anion, asymmetric defluoroallylation, Ir catalyzed asymmetric allylation*

ABSTRACT: Carbon-fluorine bond activation reaction of the trifluoromethyl group represent an important approach to fluorine-containing molecules. While selective defluorofunctionalization reactions of CF₃-containing substrates have been achieved by invoking difluorocarbocation, difluorocarboradical, or difluoroorganometallic species as the key intermediate, the transformations via fluorocarbanion mechanism remained a limited success. Furthermore, the enantioselective defluorotransformation of CF₃ group has not yet been realized. Herein, we report a defluorofunctionalization reaction of 4-trifluoromethylpyridines involving pyridyldifluoromethyl anion as the key intermediate, which was developed based upon our previous studies on the *N*-boryl pyridyl anion chemistry. When combined with Ir-catalysis, asymmetric defluoroallylation of 4-trifluoromethylpyridines could be achieved to forge a difluoroalkyl-substituted chiral center. The present work opens up a new opportunity for the defluorofunctionalization of CF₃ group, and provides new insights into the *N*-boryl pyridyl anion chemistry.

Fluorine-containing compounds usually exhibit unique properties¹ (e.g., m.p., b.p., pKa, lipophilicity, etc.) and are widely used in pharmaceuticals and agrochemicals.² Among them, the molecules bearing fluoroalkyl motifs constitute a major category, which have attracted a broad research interest and thus enhanced the demand for efficient synthetic protocols.³ In this line, a well-established approach is the introduction of a fluoroalkyl group into a substrate molecule, referred to as the fluoroalkylation strategy, which is empowered by a series of fluoroalkylating reagents.⁴ On the other hand, the modification of a preexisting fluoroalkyl group in a molecule by C-F bond activation represents a more flexible yet less developed strategy.⁵ In particular, the defluorofunctionalization of a trifluoromethyl group has been attracting an increasing attention,6 due to its common existence in fluorinecontaining compounds and the challenge associated with the activation of its C-F bond.

Recently, a series of CF₃ defluorofunctionalization reactions have been developed, with different difluorocarbon species as the key intermediate (Scheme 1). Lewis-acid was found to promote difluorocarbocation formation from simple trifuromethyl arenes, and subsequent nucleophilic substitution enabled the monoselective C-F functionalization (Scheme 1A).⁷ Single electron transfer (SET) could induce the C-F bond cleavage in CF₃ to afford difluorocarboradicals, enabling radical-type defluorofunctionalization (Scheme 1B).⁸ Recently, Wang and Houk discovered that boron radical participated in the formation of difluorocarboradical species from trifluoromethylcarbonyl compounds via a spin-center shift (SCS) process, which allowed for progressive multiple C-F bond activation (Scheme 1C).⁹ Transition-metal catalysis has also been involved in defluorofunctionalization of the CF₃ group. Palladium could undergo a formal C-F bond oxidative addition with trifluoroarenes to generate aryldifluoromethylpalladium species amenable for cross-coupling, as reported by the Zhang group (Scheme 1D).¹⁰ Very recently, the Bi group demonstrated that the in-situ generated rhodium trifluoromethylcarbenoid was able to produce difluoromethyl ketone via O-H bond insertion with water, opening an avenue to CF_3 (deuteron)hvdrodefluorination.¹¹ In contrast to the aforementioned advances, CF₃-activation via a difluorocarbanion intermediate remained a limited success. Although several electrochemical- and base-metal-induced protocols were reported, they usually suffered from poor chemoselectivity and limited functionalization type.¹² Meanwhile, to the best of our knowledge, enantioselective defluorofunctionalization of the CF₃ group has not yet been realized. To date, the selective formation of difluorocarbanion intermediate as well as its asymmetric transformation is still a formidable challenge.

In our previous study, we have discovered that pyridine, diboron, and alkoxide could react to generate an *N*-boryl pyridyl anion intermediate via heterolytic cleavage of the B-B bond. This species is highly nucleophilic and reductive in nature, based on which a series of transformations involving redox catalysis and pyridine derivatization have been developed.¹³ Bearing this knowledge in mind, we sought to generate *N*-boryl pyridyl anion from 4-trifluoromethylpyridine, which may undergo an in-situ fluoride elimination due to the nucleophilicity of the pyridyl anion. A subsequent alkoxide-induced deborylation

Scheme 1. Functionalization of the Trifluoromethyl Group



might afford a 4-pyridyldifluoromethyl anion intermediate (Scheme 1E). In this way, umpolung of the CF₃ group could be achieved, enabling a new approach to fluoroalkyl anion. We further envisioned that an enantioselective transformation of this anionic species might be achieved by transition metal-catalyzed asymmetric allylic alkylation (AAA), as inspired by the successful deprotonative AAA reaction of α -fluoro 2-pyridylacetates reported independently by Hartwig and You.¹⁴ Herein, we report the development of an Ir-catalyzed asymmetric defluoroallylation reaction of 4-trifluoropyridines, which enables the enantioselective transformation of the CF₃ group via difluoromethyl anion.

To realize this design, several challenges remain to be overcome: (a) to incorporate 4-trifluoromethylpyridines into the *N*-boryl pyridyl anion chemistry; (b) to avoid decomposition of the difluoromethyl anion by alphaelimination and over-defluorination;¹⁵ and (c) to ensure the compatibility of the fluoroalkyl anion formation with the AAA reaction. Therefore, we set out to test the formation of 4-pyridyl difluoromethyl anion from 4trifluoromethylpyridine (**1a**) and trapping of this anion with several simple electrophiles (Scheme 2). To our delight, when benzaldehyde, paraformaldehyde, and *N*phenylbenzaldimine were employed as the electrophile, the corresponding addition products **2a-c** were produced in good yields. Alternatively, when 10 equivalents of CD₃OD was used instead, deuterodefluorination product **2d** was obtained with a high level of deuterium incorporation. These findings strongly supported the formation of the desired difluoromethyl anion intermediate or its equivalent, which served as a proof-of-concept of the designed reaction and showed that the mono defluorofunctionalization could be achieved selectively.





With this encouraging result in hand, we sought to merge the formation of the pyridyldifluoromethyl anion with the Ir-catalyzed AAA reaction (Table 1), although the Ir-catalyzed asymmetric defluoroallylation reaction was still unprecedented. Pyridine 1a and cinnamyl carbonate **3a** were tested as template substrates. It was found that, under the optimal reaction conditions (5 mol% iridium complex (*S*,*S*,*S*_{*a*})-**[Ir]** as the catalyst, EtONa as the base, THF as the solvent, and performing the reaction at 30 °C), the branched defluoroallylation product 4a was generated in a good vield and excellent enantioselectivity (entry 1). Altering the base from EtONa to MeONa led to a lower efficiency (entry 2), and increasing or decreasing the catalyst loading resulted in diminished yields (entries 3-7). Using the insitu formed catalyst¹⁶ from ligand L1 and [Ir(cod)Cl]₂ instead of the pre-formed one was proved inferior (entries 8 and 9). Other ligands (L2-L4) were found not competent for this reaction (entries 10-12).

Table 1. Optimization Study^a



^{*a*}Reaction conditions: **1a** (0.2 mmol), **3a** (2.0 equiv), EtONa (3.0 equiv), B₂pin₂ (3.0 equiv), (*S*,*S*,*S*₀)-**[Ir]** (5 mol%), THF (2 mL), Ar atmosphere, 30 °C, 4 h. ^{*b*}MeONa-B₂pin₂ was used instead of EtONa and B₂pin₂ in entries 3-12. ^{*c*}The Ir catalysts were pre-activated by a known preocedure for **L1**, **L3**, and **L4**.¹⁶ ^{*d*}Determined by ¹H NMR using 1,3,5-trimethoxybenzene as the internal standard. ^{*c*}Determined by HPLC analysis. ^{*f*}Yield of the isolated product.



With the optimal reaction conditions in hand, we started to explore the scope of this asymmetric difluoroallylation reaction (Table 2). The reactions of 4-trifluoropyridine (1a) with various allyl carbonates were first examined. The cinnamyl carbonates bearing electron donating groups on the phenyl ring were suitable reactants, delivering the target products 4b-d in good yields and excellent enantioselectivities. 3,4,5-Trimethoxy- and 3-methylsubstituted cinnamyl carbonates afforded the corresponding products 4e and 4f in moderate yields with excellent enantioselectivities. 4-Halogen substituents on the phenyl ring of the cinnamyl carbonate were tolerated in this reaction, albeit with diminished yield and selectivity. Allyl carbonates bearing other aromatic rings, such as thiophene and naphthalene, were also compatible substrates to afford 4i and 4j in moderate yields and excellent selectivities. It is noteworthy that an ortho-substitution on the phenyl group led to a decreased enantioselectivity (4k). It was found that, alkyl substituted allylic carbonate was not a suitable substrate, which exhibited greatly diminished reactivity.

Subsequently, the scope of the fluoroalkyl anion source was examined (Table 2). 4-Trifluoromethylpyridines bearing various 3-substituents were found compatible for this defluoroallylation reaction. Aryl groups of different electronic properties were tolerated (41-40), and in particular, another pyridine core remained intact during the reaction (4p). Further investigation showed that 3-vinyl and benzyl substituted 4-trifluoromethylpyridines also behaved well to furnish 4q and 4r in good yields and enantioselectivity. Notably, 4-trifluoromethylpyrimidines also underwent the desired reaction smoothly under identical reaction conditions to give a satisfactory outcome (4s and 4t), exhibiting the potential of the present method for adapting other N-heterocycles. Pyridine substrates bearing electron-withdrawing substitutions at the 3-position participated in the reaction smoothly to afford the corresponding defluoroalkylation products 4u-4z in high yields and excellent selectivities, among which functional groups such as ester, amide, tertary amine, and geraniol ester were well tolerated. The present protocol also allowed for reaction at a larger scale without deterioration of yield and enantioselectivity, as demonstrated by 2 mmol scale synthesis of 4c.

As for the limitation of this reaction, we found that both 2-substituted 4-trifluoromethylpyridine and 2-trifluoromethylpyridine exhibited no reactivity, implying that the formation of the *N*-boryl pyridyl anion intermediate is sensitive to steric hindrance.

To explore the synthetic utility of this reaction, we investigated the derivatization of the defluoroallylation products (Scheme 3). First, derivatization of the vinyl group was attempted, which could be transformed to a hydroxyethylene group by hydroboration-oxidation sequence to produce alcohol **5** in 71% yield and 97% ee. Ozonation of the double bond followed by reduction furnished the alcohol product **6** in 52% yield and 98% ee. Second, we tested whether a second defluorofunctionalization could be achieved by applying the *N*-boryl pyridyl anion chemistry. Gratifyingly, it was found that product **4c** could undergo defluorination and nucleophilic addition with 2-



^{*a*}Reaction conditions: **1** (0.5 mmol, 1 equiv), **3** (1.0 mmol, 2 equiv), B_2pin_2 (1.5 mmol, 3 equiv), EtONa (1.5 mmol, 3 equiv) and (*S*,*S*,*S*_{*a*})-[**Ir**] (0.025 mmol, 5 mol%) in 5 mL of THF, Ar atmosphere, 30 °C, 4 h. ^{*b*}Reaction was carried out on 2.0 mmol scale.

phenylbenzaldehyde under slightly more forcing reaction conditions, and the addition product could be oxidized to afford ketone **7** in 38% overall yield and 2:1 dr. Meanwhile, cinnamyl chloride could also be employed as the electrophile in this transformation to produce allylated product **8** in 2:1 dr, which afforded cyclized product **9** after ringclosing olefin metathesis to serve as an indicator for stereochemistry. Although the yield and diastereoselectivity was not perfect, these results showed that sequential monoselective C-F activation of 4-trifluoropyridine is viable with the present protocol.

Scheme 3. Derivatization of Products^a

A. Derivatization of the vinyl group



B. Second C-F bond functionalization:



9 (34% crude yield, 2:1 dr)

^aDMP = Dess-Martin periodane; DCM = dichloromethane, EA = ethyl acetate; rt = room temperature

Interestingly, in addition to trifluoromethylpyridines, methyl pyrid-4-ylmethyl ethers were also compatible for this umpolung process. Ethers **10** and **11** were found to undergo a demethoxyalkylation reaction smoothly with alkyl halides under the established conditions to afford the alkylated products **12** and **13**, albeit an AAA reaction was not successful at the moment. Nevertheless, this result indicated that the *N*-boryl pyridyl anion chemistry could serve as a generic strategy for umpolung of the pyrid-4ylmethyl motif bearing a leaving group to enable its nucleophilic reactivity.

Scheme 4. Demethoxylalkylation reactions



In conclusion, we have developed a monoselective defluorofunctionalization reaction of 4-trifluoromethylpyridines by invoking the diboron-base-pyridine system. This protocol leads to the umpolung of the fluoroalkyl unit on the pyridine ring, and allows for an asymmetric transformation of the CF₃ group. The present work opens up a new opportunity for the functionalization of trifluoromethylpyridines, and provides new insights into the *N*boryl pyridyl anion chemistry. We expect this reaction system to become a versatile platform for the umpolung and functionalization of pyridylmethyl moiety, and the corresponding studies are ongoing.

ASSOCIATED CONTENT

Supporting Information. Experimental procedures, characterization data, and copies of NMR spectra.

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Notes

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