

Unlocking the hydrolytic mechanism of GH92 α -1,2-mannosidases: computation inspires using C-glycosides as Michaelis complex mimics

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Abstract: The conformational changes in a sugar moiety along the hydrolytic pathway are key to understand the mechanism of glycoside hydrolases (GHs) and to design new inhibitors. The two predominant itineraries for mannosidases go via ${}^0S_2 \rightarrow B_{2,5} \rightarrow {}^1S_5$ and ${}^3S \rightarrow {}^3H_4 \rightarrow {}^1C_4$. For the CAZy family 92, the conformational itinerary was unknown. Published complexes of *Bacteroides thetaiotaomicron* GH92 catalyst with a S-glycoside and mannoimidazole indicate a ${}^4C_1 \rightarrow {}^4H_5/{}^1S_5 \rightarrow {}^1S_5$ mechanism. However, as observed with the GH125 family, S-glycosides may not act always as good mimics of GH's natural substrate. Here we present a cooperative study between computations and experiments where our results predict the $E_5 \rightarrow B_{2,5}/{}^1S_5 \rightarrow {}^1S_5$ pathway for GH92 enzymes. Furthermore, we demonstrate the Michaelis complex mimicry of a new kind of C-disaccharides, whose biochemical applicability was still a chimera.

The regulation of N-glycan modifications is crucial to maintain protein quality control and the functional development of glycoproteins in the endoplasmic reticulum and the Golgi apparatus.¹ Glycoside hydrolases (GHs) are biocatalysts dedicated to cleave glycosidic bonds connecting individual monosaccharides. Several families of GHs are also involved in N-glycan biosynthesis. Some of these enzymes that are involved in the protein folding and the function regulation are essential for the control of congenital disorders of glycosylation – also known

as carbohydrate-deficient glycoprotein syndromes.^{2,3} Among those catalysts, α -mannosidases are irreplaceable in the regulation of the mannose-containing *N*-glycan processing. Following the CAZy classification⁴, GH families 38,⁵ 47,⁶ 76,⁷ 92,⁸ 99⁹ and 125¹⁰ are responsible for *N*-glycan cleavage.

The inhibition of a GH relates to its enzymatic mechanism. Once the substrate (a saccharide) joins the active site of the hydrolase, the sugar in the -1 subsite (-1 sugar)¹¹ undergoes a conformational change during hydrolysis. When the process overcomes the highest energy state, also known as the transition state (TS), the -1 sugar becomes an oxocarbenium ion-like moiety.¹² One of the strategies to inhibit a GH is based on designing TS mimics whose conformation is similar to the oxocarbenium ion-like ring.¹³⁻¹⁶

In parallel, developing chemical derivatives of the natural substrate of GHs has emerged as a procedure to inhibit glycosidases. The most common technique utilizes sulfuration of the glycosidic bond (*S*-glycoside).¹⁷ In some experiments, the crystallization of GHs in complex with *S*-glycosides shed light on the conformation of the natural substrate in the Michaelis complex (MC, reactant).^{18,19} However, their MC mimicry was not observed in GH125. The natural substrate-related *S*-glycoside predicted a non-distorted -1 sugar moiety, while computations and further experiments showed a distorted one.²⁰ Comparing the available GH125 complexes reveals (*S*-glycoside – PDB 3QT9,¹⁰ and natural substrate – PDB 5M7I¹⁹) that the substitution of the glycosidic oxygen by sulfur leads to a lengthening-of the glycosidic bond (from 1.45 Å to 1.87 Å) and a C-X-C angular readjustment (from 108.4° to 103.7°). Due to the topology of the GH125 active site, this structural change resulted in a hydrogen bond rearrangement between the -1 sugar and the enzyme, and a consequent conformational change. Since we talk about ring mimicry, the conformational study of 6-membered rings (6-rings) is the base to classify and quantify the enzymatic itineraries of GHs. In 1975, Cremer and Pople developed a mathematical expression based on puckering coordinate(s) allowing the graphical representation of the conformational space of a ring.²¹ As depicted in **Figure 1**, a Mercator representation projects the puckering space in a two-dimensional surface where all the conformations of a pyranose ring are presented. By the principle of least nuclear motion,²² the conformations of the reactants (MC) and products (PC) must surround the TS, following an ideal linear pathway. In α -mannosidases, we

Connecting both regions (**Figure 1** – green arrow), the experiments show an unexpected ${}^4C_1 \rightarrow {}^4H_5/{}^1S_5 \rightarrow {}^1S_5$ pathway. Furthermore, analyzing the available MC mimic complexes (PDB 2WW1 and 2WW3), the catalytic water is not kept in a position allowing the nucleophilic attack ($> 4 \text{ \AA}$), while the TS mimic (PDB 2WZS) shows a water molecule at a distance of $\sim 3 \text{ \AA}$ from the anomeric center. This fact and the resulting pathway disrespecting the least nuclear motion principle indicate a possible conformational mismatch between the S-glycoside **3** and the natural substrate **1** (as observed in GH125).

Our most burning question in this work solved the structural differences between the S-glycoside **3** and MVL **4** complexes, and why is the catalytic water not kept close to the S-glycoside. The answer could be related to the abrupt conformational change of the -1 sugar. After the structures inspection, we observed the most relevant difference in the vicinity of the Ca^{2+} cation. In the presence of MVL **4**, the catalytic water is coordinated to the calcium cation. In the presence of S-glycoside **3**, the hydroxyl group (OH) in carbon 2 (C2) is occupying the position of the catalytic water. With MVL **4**, the hydroxyl at C2 interacts with the proton donor (E533).

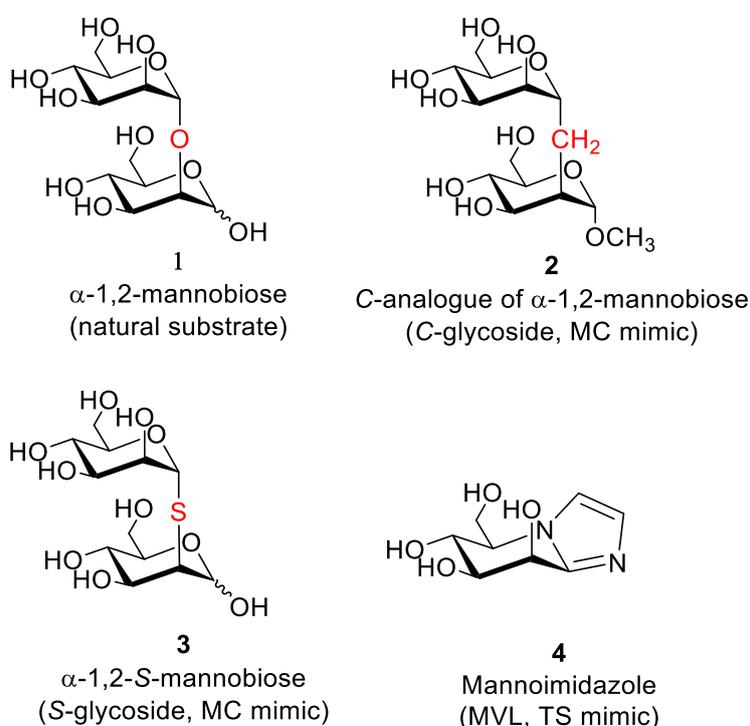


Figure 2. Schematic representation of the molecular structures of the substrates used to study the hydrolytic mechanism of GH92 α -mannosidases.

Considering all aforementioned evidence, quantum-mechanics (QM) calculations could shed light on the conformational itinerary followed by the natural substrate in GH92 enzymes. Starting from the BtGH92-MVL crystal structure (PDB 2WZS), we constructed a full-QM cluster model of the active site by manual exchange of the MVL moiety for the natural substrate (α -1,2-mannobiose **1**). Keeping the Ca^{2+} -glycoside coordination and hydrogen bond interactions observed in the presence of MVL, we observed a significant conformational change after the geometry optimization (computational details available in the SI).

The optimized cluster model (depicted in **Figure 3** as MC) presents a distorted $E_5/B_{2,5}$ conformation for the -1 sugar ring. This result differs significantly from the 4C_1 conformation observed in the *S*-glycoside **3**. Since no experimental evidence of the ring distortion existed, this calculation would have appeared just as a theoretical hypothesis. However, two recent works in organic chemistry and biochemistry inspired a strategy to demonstrate distortion of α -1,2-mannobiose **1** when forming a Michaelis complex with GH92 enzymes.²³⁻²⁵

On the one hand, the synthesis and conformational analysis of *C*-oligosaccharides^{22,23} provided an evidence of a group of molecules capable to mimic natural carbohydrates. On the other hand, Li et al. were capable to characterize and crystallize a Ca^{2+} -dependent α -1,2-mannosidase from *Enterococcus faecalis* (EfGH92), which has two domains like other GH92 mannosidases.²⁵ Structural superposition between BtGH92 and EfGH92 is provided in the SI.

C-analogue of α -1,2-mannobiose **2** (**Figure 2**, compound **2**) was received via previously published procedure using Ni-catalyzed Suzuki-Miyaura cross-coupling reaction followed by stereoselective transformations of obtained *C*-pseudodisaccharide.²⁶ Furthermore, we performed the cloning and expression of the wild-type (WT) and an acid catalyst mutant (E494Q) of EfGH92 (see SI for details). Once obtained the different catalysts, we crystalized EfGH92 (WT) with *C*-disaccharide **2** and E494Q (mutant) with the natural α -1,2-mannobiose **1**, and determined their crystal structures at high resolutions. Further synthetic details, conformational analysis of *C*-disaccharide **2** in water solution, and results of the experiments are presented in the SI.

X-ray crystallography showed that WT and the mutant form a tetramer where ligands and Ca^{2+} ions are allocated in each active site cavity (-1 and +1 subsites).

On the one hand, three of the four C-disaccharides **2** attached to WT exist in a conformation corresponding to the QM-optimized cluster model ($E_5/B_{2,5}$), while one outlier presents a ${}^1S_5/{}^1,4B$ structure (**Figure 3** – green dots). On the other hand, the natural substrate **1** attached to the mutant presents one of the four structures in the E_5 region, two of them in the $E_5/{}^4H_5$ region, and the last one in the ${}^4H_5/{}^1S_5$ space (**Figure 3** – purple dots). This evidence together with the conformations found in the BtGH92-MVL complex **4** indicate a simplified conformational itinerary crossing the $E_5 \rightarrow {}^1S_5$ region (**Figure 3** – green arrow). In terms of reactivity, the reported kinetic experiments with BtGH92 and α -1,2-mannobiose **1** show a k_{cat} of 87 s^{-1} .⁸ Applying the transition state theory,²⁷ the activation barrier is equivalent to $14.9 \text{ kcal}\cdot\text{mol}^{-1}$. To decipher the conformational itinerary followed by the -1 sugar in the cluster model, we employed a scan calculation. The proton in E533 (equivalent to E494 of EfGH92) was transferred to the glycosidic oxygen, the cleavage of the glycosidic bond, and the nucleophilic attack of the catalytic water were activated by constraining the involved distances. The optimized structures of MC, TS (“TS-like”) and PC are presented in **Figure 3**. We obtained the potential energy barriers of $\Delta E^\ddagger = 14.1 \text{ kcal}\cdot\text{mol}^{-1}$ and $\Delta E_0 = -2.7 \text{ kcal}\cdot\text{mol}^{-1}$. This result is in good agreement with the available kinetics parameters.⁸

Following the -1 sugar conformation along the reaction pathway (**Figure 3** – red dots and black dashed line), we can conclude that the itinerary in our cluster model is $E_5/B_{2,5} \rightarrow B_{2,5}/E_5 \rightarrow {}^1S_5/B_{2,5}$. Being aware of the subtle conformational changes observed in the -1 sugar both experimentally and computationally, and the dynamic nature of the biochemical systems, the accuracy of results is limited. The experimental results lead to an MC in the vicinity of E_5 and a TS crossing ${}^4H_5/{}^1S_5$. While the calculations suggest an MC in $E_5/B_{2,5}$ and a TS in $B_{2,5}/E_5$. In any case, investigated changes are only minor. Nevertheless, when we slightly simplify the nomenclature, we can conclude that GH92 α -1,2-mannosidase follows an $E_5 \rightarrow B_{2,5}/{}^1S_5 \rightarrow {}^1S_5$ conformational itinerary.

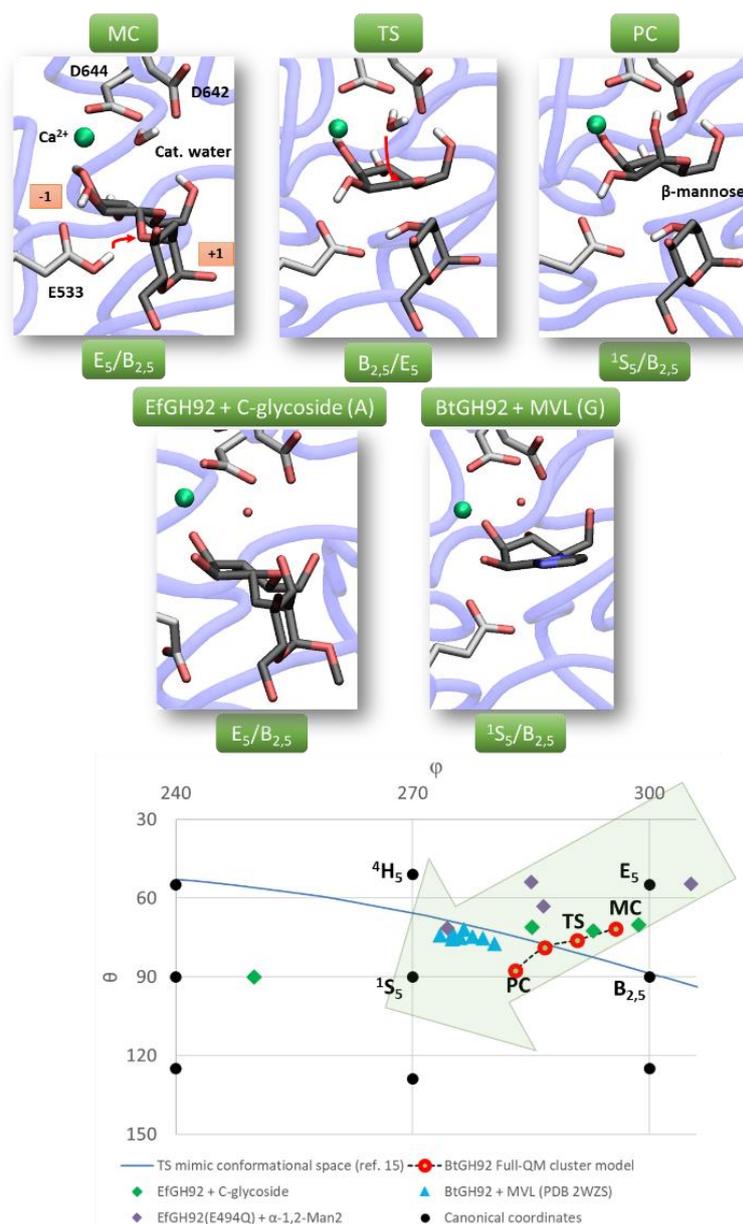


Figure 3. Simplified representative structures of MC, TS and PC of the QM-optimized cluster model of BtGH92 with **1**, the chain A of the EfGH92 enzyme in complex with the C-disaccharide **2** (MC mimic), and chain G of the BtGH92 enzyme in complex with MVL (**4**, TS mimic). Zoom view of the pucker coordinates space of the observed -1 sugar moieties in the available experiments and calculations (further details in the SI).

The analysis of crystal structures of active sites revealed (**Figure 4**) that the distorted 1- α -mannose moiety occupies the -1 subsite, while the +1 subsite is occupied by an undistorted 2-CH₂- and 2-O-mannosyl leaving group. Comparing the EfGH92-C-glycoside complex and the PDB 2WW3 (thioglycoside) structure, we observe that the hydroxyl group at C2 of the S-glycoside occupies the same position as the catalytic water in the C-glycoside complex. Due to the hydrophobic

nature of the pseudo-glycosidic methyl group, the E494 catalytic acid residue changes its orientation, and a water molecule interacts with the OH group at C2 of the C-glycoside. The closest water to the anomeric carbon is further than 4 Å in case of the S-glycoside, while a well-oriented catalytic water is coordinated to Ca²⁺ at ~3 Å from the anomeric center of the C-glycoside (**Figure 4 – C**).

In **Figure 4 – D**, we also depict the main hydrogen bond interactions present in the E494Q EfGH92-mannobiose complex. As observed in the QM-optimized BtGH92 cluster model, the OH group at C2 interacts with the oxygen of the mutated glutamine residue Q494. The NH₂ group of the mutated residue interacts with the glycosidic oxygen, properly oriented for a hypothetical proton transfer. The hydroxyls at C3 and C4 interact with aspartic acid D313. The catalytic water interacts with two aspartic acid residues, D602 and D604. Finally, the hydroxymethyl arm interacts with a water molecule and serine S66. These observations are in good agreement with the interactions observed in our calculations, strengthening the viability of the model.

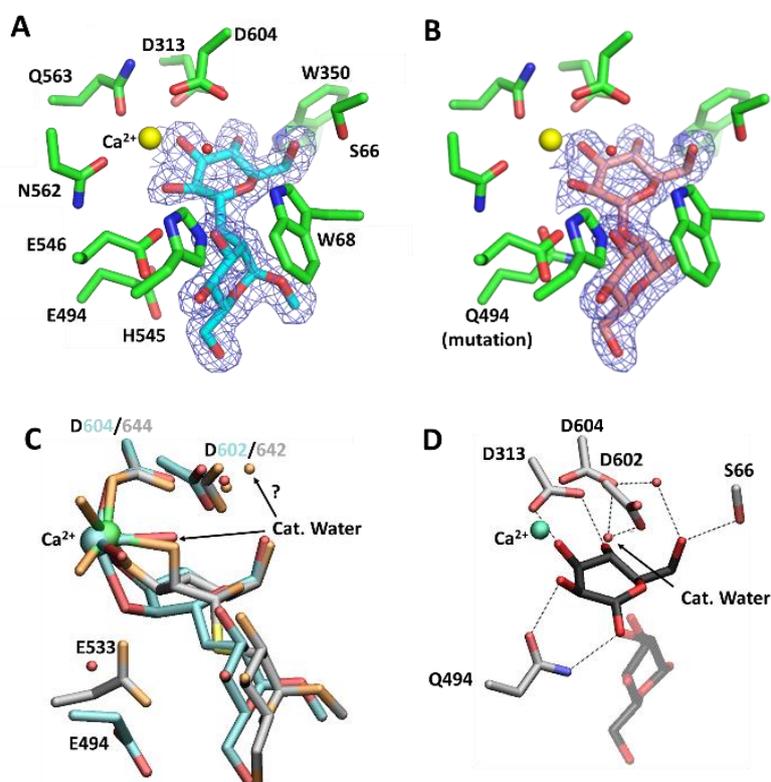


Figure 4. Active sites and observed electron densities ($2F_o - F_c$, 1σ) of (A) WT EfGH92 in complex with C-glycoside **2**, and (B) E494Q EfGH92 in complex with α -1,2-mannobiose. (C) Structural superposition of the active sites of BtGH92 in complex with 1-thio- α -1,2-mannobiose (PDB 2WW3, grey and orange), and EfGH92 in complex with **2** (this work, cyan and red). The OH in

C2 of the thioglycoside occupies the same position as the catalytic water in presence of the C-glycoside. (D) Main hydrogen bond interactions observed in the active site of E494Q EfGH92 in complex with α -1,2-mannobiose. The OH in C2 interacts with the free oxygen of the mutated catalytic acid Q494 residue.

In conclusion, computations in this study have newly been a powerful tool to decrypt the reaction mechanism of an enzyme-substrate biosystem. Both experimental results and calculations confirm that the catalytic mechanism of GH92 α -1,2-mannosidases follows the $E_5 \rightarrow B_{2,5}/^1S_5 \rightarrow ^1S_5$ itinerary (Figure S10). As observed in GH125 enzymes, S-glycosides may not act as good MC mimics, due to different interaction patterns with the active site residues, and, in this case, also, with a metal cation. We identify a new class of MC mimic through the substitution of glycosidic oxygen by the CH₂ group. This chemical change keeps the conformation of the sugar, but the glycosidic position becomes hydrophobic, suppressing thus the interaction with the catalytic residue.

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