

Computer-aided design, synthesis, and biological evaluation of [4.3.0] bicyclic prolyl oligopeptidase and fibroblast activation protein- α dual inhibitors

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Abstract. We have previously described several different chemical series of bicyclic prolyl oligopeptidase (POP) inhibitors as probes for neurodegenerative diseases that demonstrated nanomolar activity *in vitro* and submicromolar activity *in cellulo*. The more recent implication of POP in cancer, together with homologous fibroblast activation protein α (FAP), implicated in tumor growth, led us to consider developing POP/FAP dual inhibitors as a promising strategy for the development of cancer therapeutics. We report herein docking-guided design of a new bicyclic scaffold and synthesis of both covalent and non-covalent bicyclic inhibitors. Biological evaluation of first-of-their-kind [4.3.0] bicyclic compounds confirmed that reactive groups, or covalent warheads, are required for inhibitor activity. This work ultimately led to a dual inhibitor equipotent to the only anti-POP/FAP drug that ever-reached clinical trials.

Introduction.

Prolyl oligopeptidase (POP, also referred to as PREP) and fibroblast activation protein- α (FAP, also referred to as seprase) are homologous serine proteases whose function consists of cleaving short peptides at the C-terminal end of proline residues. POP was discovered in the mid-70's, and its high concentration in the central nervous system (CNS) immediately drew attention;¹⁻³ early studies associated POP protease activity to neuropeptides and peptide hormones. Inhibition of this protease activity was first investigated with the reversible covalent inhibitor **1** over thirty years ago (Figure 1). However, after significant targeted research and unsuccessful clinical trials, this

progress reached a plateau (Figure 1). In more recent years, a second boost in the development of POP inhibitors has occurred, and their potential in Alzheimer's disease (AD) and Parkinson's disease (PD) has been under further investigation.¹⁻⁴ More specifically, about 10 years ago, Lambeir and co-workers first linked POP to α -synuclein (aSyn) aggregation, a hallmark of PD,⁵ and Myöhänen and co-workers revealed the colocalization of POP with aSyn, amyloid beta (A β), and the tau protein in brain samples from patients with PD or AD.⁶ The link between POP and aSyn aggregation was further supported by extensive studies by Myöhänen and co-workers who first showed that aggregation is induced by POP-aSyn protein-protein interaction and is unrelated to the protease activity of POP.^{7,8} The same group also demonstrated the effect of KYP-2047, a POP inhibitor (Figure 2), on aSyn aggregate clearance^{9,10} and the restoration of motor behavior in mouse models,¹¹ while Lee and co-workers identified POP inhibitors and confirmed their effect on aSyn expression.¹² In 2011 then in 2015, work by Lambeir then Savolainen *et al.* suggested that POP interacts with aSyn serving as nucleation point, hence inducing aggregation.^{13,14} More unexpectedly, they observed that KYP-2047 modulates the shape of POP and reduces its ability to interact with aSyn and reduce its aggregation.¹⁴ Using a combination of NMR and small angle X-ray scattering, Giralt and co-workers revealed that POP inhibitors significantly impact this protein dynamics.¹⁵ In agreement with Savolainen *et al.*'s observations,¹⁴ they propose that while POP exists as an equilibrium of open and close conformations necessary to induce aSyn aggregation, POP bound to an inhibitor is more rigid adopting primarily the closed conformation which does not induce aggregation.

In addition to these recent advances in neurodegenerative research linking POP to AD and PD, this enzyme's proteolytic activity was also recently found to contribute to the release of acetyl-SDKP, a potent tetrapeptide that stimulates angiogenesis.¹⁶ It has since been reported that POP inhibition blocks the growth of human gastric cancer cells¹⁷ and the proliferation of breast cancer cells.¹⁸ We have also demonstrated that our own inhibitors (series based on **2** and **3**, Figure 1) can block POP protease activity in various cancer cell lines.^{19,20} The endopeptidase activity of POP is shared with FAP, the latter of which is suggested to be a key modulator of the tumor microenvironment (TME)²¹⁻²³ and is thus a promising target for novel anticancer therapeutics.²⁴ Discovered over 10 years after POP, FAP is overexpressed in most human epithelial-derived cancers²⁵ and has also been suggested to promote tumor growth.^{21,26} In fact, its inhibition significantly affects stromal growth *in vivo*.²² Most importantly, FAP is not detectable in normal

tissues,²⁷ making it an extremely valuable target for therapeutic intervention against refractory tumors, and inhibitor development has in fact already started (Figure 2).²⁸⁻³⁰

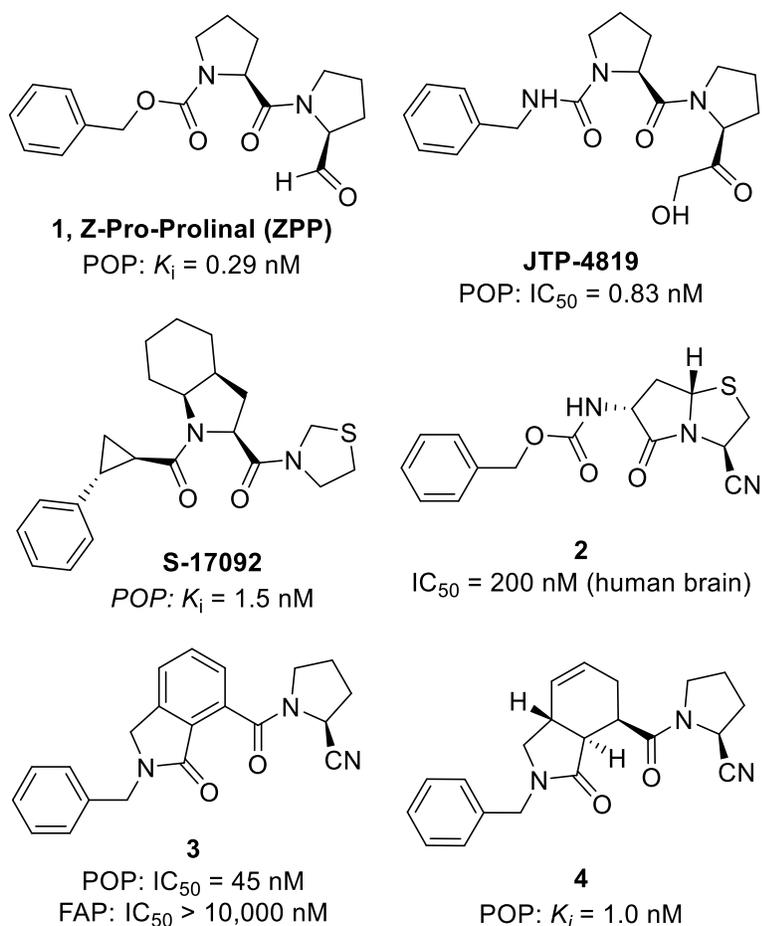


Figure 1. Selected POP and FAP inhibitors

In recent medicinal chemistry endeavors, selective inhibition of one enzyme over the other has been pursued. As illustrated in Figure 2 selectivity is very sensitive to minor structural changes. For example, while compound **5** is highly FAP selective, the analogue **6** is highly POP selective. This D-Ala-induced selectivity for FAP has been further observed recently.³¹ Nevertheless, Christiansen *et al.* suggested that targeting both FAP and POP blocks stromal invasion and angiogenesis, respectively, and may alter cancer growth.³² They designed a pseudopeptide dual inhibitor which was found to block tumor growth in mice. These findings suggest that dual inhibition is a promising strategy, though this large, non-drug-like molecule was unsuitable for further consideration.³³ Consequently, although an overview of the literature suggests that

selectivity may be easier to achieve than dual inhibition, the latter may be an ideal strategy for designing and developing anti-cancer therapeutics.

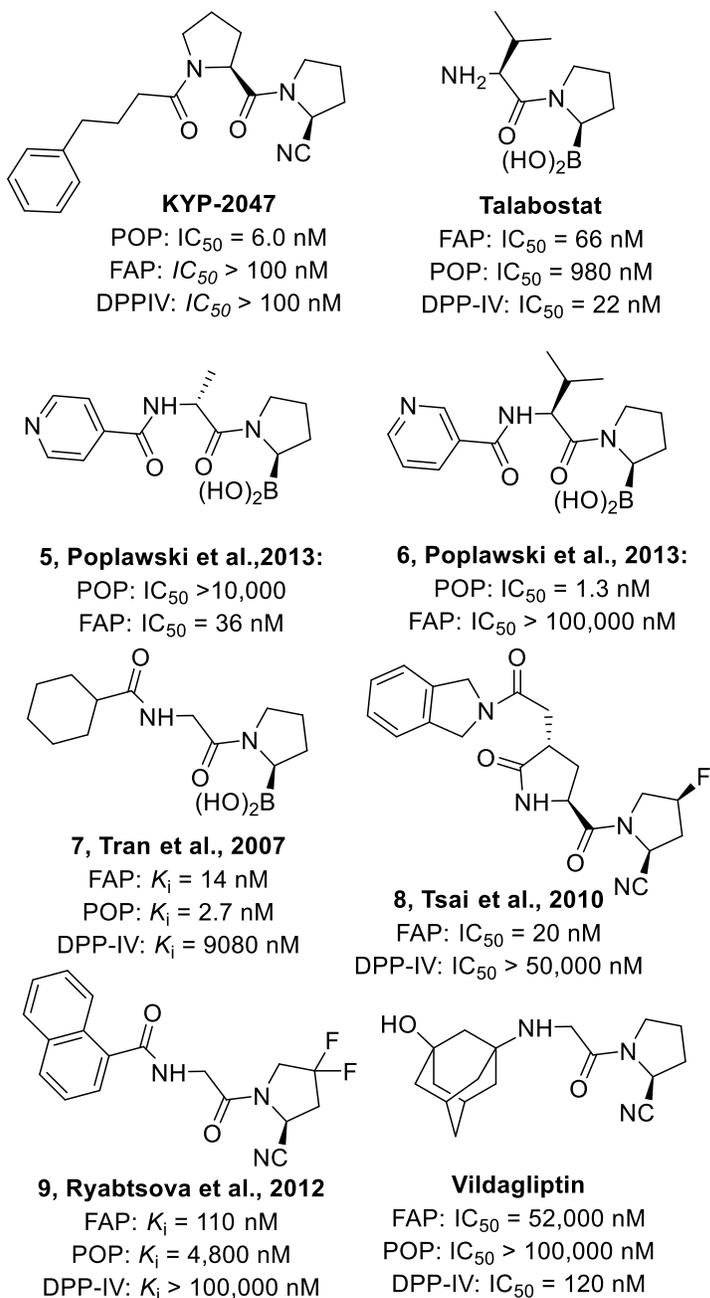


Figure 2. Selected known POP and FAP inhibitors and their selectivity profiles^{28,30,34,35}

In 2009, our own group reported a series of [3.3.0] bicyclic POP inhibitors based on compound **2** which were found to be cell-permeant and potent in the sub-micromolar range. This series of nitrile-containing compounds were designed to act as covalent inhibitors targeting the catalytic

serine in the POP active site.¹⁹ Interestingly, a few years later, **KYP-2047** was co-crystallized with POP, demonstrating the covalent nature of the binding of nitrile derivatives in the active site of POP.³⁶ However, the series of inhibitors based on compound **2** was halted after metabolism studies revealed it to be metabolized into complex mixtures via oxidation of the sulfur.³⁷ A few years later, compound **3** was discovered via virtual screening and docking-guided optimization, This inhibitor exhibited a POP inhibitory activity five times more potent than that of our first hit **2** and was active in low-micromolar concentrations on human glioblastoma and endothelial cancer cells.²⁰ In addition, we found that the introduction of the [4.3.0] bicyclic molecular scaffold improved the metabolic stability of our inhibitors.²⁰

Five years ago, we also reported the structure-based design and synthesis of a novel class of POP inhibitors based on a hexahydroisindole scaffold, such as **4** (Figure 1). A docking study guided the selection of structures (both in terms of stereo- and regiochemistry) for synthesis. Following the synthesis of the best virtual candidates, *in vitro* assays revealed that one member of this chemical series, compound **4**, was more active than any of our previous inhibitors, exhibiting a K_i of 1.0 nM. Additional assays also showed that the scaffold of this potent inhibitor, in contrast to the series based on compound **2**, is highly metabolically stable.³⁸ However, upon *in vitro* testing of **3** and **4** against recombinant FAP, they were completely inactive. Analysis of docking poses revealed a lack of stabilizing interactions with the two glutamic acid residues in the active site of FAP (Glu203 and Glu204).

With this information in hand, we became interested in the design of dual POP/FAP inhibitors. We report herein our successful efforts in the development of dual inhibitors based on an improved bicyclic core.

Results and discussion

Computer-aided design. With our first three series of POP inhibitors illustrated by compounds **2**, **3** and **4**, we have demonstrated the accuracy of our docking program FITTED³⁹⁻⁴¹ in predicting binding modes of POP covalent inhibitors. When **2** and stereoisomers of **2** (adhering to the [3.3.0] bicyclic system) were evaluated, we found that the stereochemistry corresponding to that of D-amino acids was optimal (hydrogen atom highlighted in blue in Figure 3, compound **2**). The resultant stereochemistry upon cyclization (hydrogen atom highlighted in green in Figure 3) at the

ring junction fortunately imposed a shape that fit best in the binding site. In this previous report,¹⁹ computational studies also indicated that this [3.3.0] bicyclic system was less optimal for binding to POP, and that a [4.3.0]-ring system with a specific stereochemistry should exhibit better affinity (Figure 3, compound **2a**).²² Unfortunately, our synthetic efforts were vain, as the epimer at the ring junction (hydrogen atom highlighted in green in Figure 3, compound **2a**) was the only isomer observed experimentally but was not predicted to bind optimally in the active site of POP.

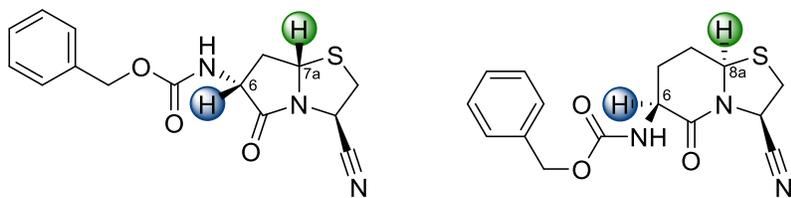


Figure 3. POP inhibitors designed by our group in the past, including required stereochemistry for optimal inhibitor stabilization in the active site of POP (highlighted with blue and green hydrogens).

Further computational predictions indicated that the affinity for POP could be improved by increasing the size of the western ring and inverting two stereocenters, both the carbon at the cyclic fusion (C_{7a} in **2**, C_{8a} in **2a**) and the carbon alpha to the cyclic amide (C_6). To do so, we decided to prepare a first series of analogues built around a [4.3.0]-ring system similar to that of **2a** but which could be accessible synthetically. After several rounds of virtual modifications and docking predictions, inhibitor structure **10a** was discovered. As can be seen in Figure 4, the predicted binding mode of nitrile **10a** is highly favored, featuring the same key interactions as potent aldehyde **1**.

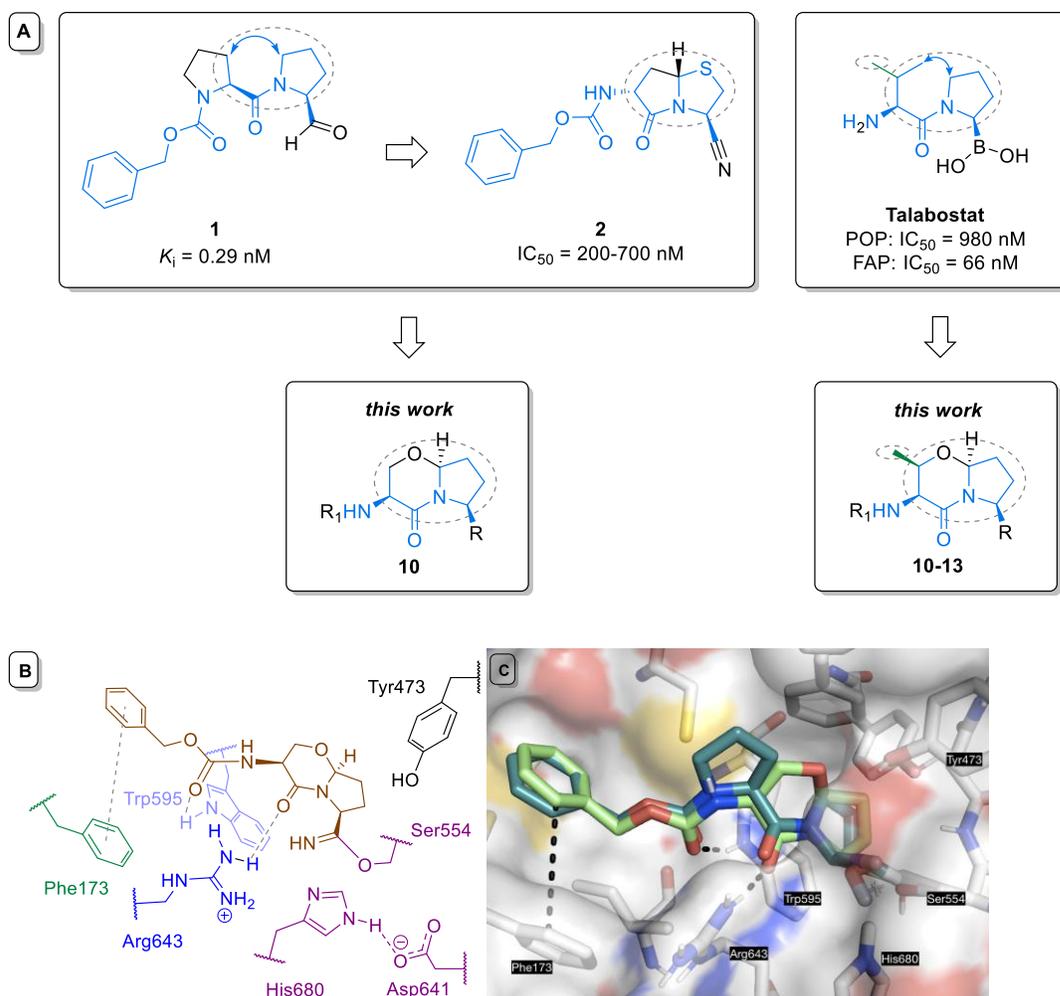
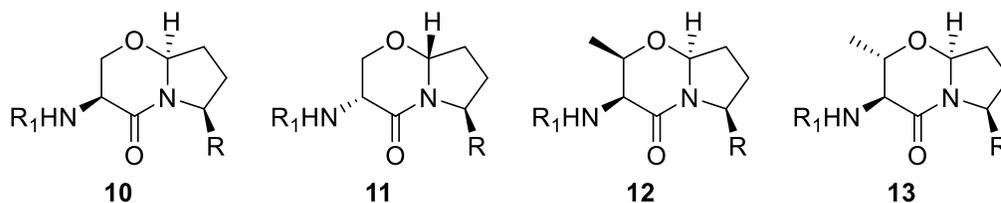


Figure 4. *In silico* design of a new series of bicycles. (A) Previously designed POP inhibitors and newly designed series of potential dual inhibitors. (B) Schematic representation of the predicted binding pose of **10a** (brown) in the POP active site, catalytic triad in purple, key residues in blue/green; (C) predicted binding mode of **10a** (green), overlaid with the predicted binding mode of **1** (teal) (docked to POP using FITTED, pdb code: 2xdw)

This prediction encouraged us to pursue the synthesis of compound **10a** and other analogues. Our previous inhibitor **2** and this newly-designed scaffold resemble previously-reported potent inhibitor **1** (Figure 4). The bicyclic scaffolds **2** and **10** were introduced by virtually rigidifying **1** and introducing heterocyclic alkanes to both optimize the docking pose and ensure synthetic feasibility. The valine-based side chain of Talabostat (Figure 2), a POP-FAP inhibitor that reached Phase III clinical trials,⁴² inspired the introduction of methyl groups into **10a** and incorporation of

the boronic acid warhead, leading to **12c** and **13b**. The complete list of new analogues selected for synthesis is provided in Table 1.

Table 1. Newly designed bicyclic analogues.



Entry	Compound #	R	R ₁
1	10a	CN	Cbz
2	10b	H	Cbz
3	10c	Bpnd	Cbz
4	10d	Bpnd	Ac
5	10e	Bpnd	Boc
6	10f	B(OH) ₂	H
7	11a	Bpnd	Boc
8	11b	B(OH) ₂	H
9	12a	Bpnd	Cbz
10	12b	Bpnd	Boc
11	12c	B(OH) ₂	H
12	13a	Bpnd	Boc
13	13b	B(OH) ₂	H

Many POP inhibitors feature nitriles, activated nitriles (*i.e.*, with proximal fluorine atoms), or boronic acids, the latter two of which are more electrophilic and lead to more potent FAP inhibition. Our current version of our docking program FITTED does not consider either the reactivity of the catalytic residue nor the reactive warhead. Nevertheless, a computational study from our group on the reactivity of the catalytic serine residues in both POP and FAP suggests that the catalytic serine in POP is more nucleophilic than that in FAP and that, as a result, nitrile derivatives are unlikely to act as potent covalent inhibitors of FAP, while boronic acids are promising alternatives.^{43,44} As a result, in our quest to develop dual POP/FAP inhibitors, the boronic ester or acid derivatives were also considered.

Boronic acids have been widely used in medicinal chemistry, notably as warheads of reversible covalent inhibitors of proteases,^{45,46} including two approved drugs (Bortezomib and Ixazomib for the treatment of relapsed multiple myeloma and mantle cell lymphoma).⁴⁷ In addition, boronic acids are remarkably stable despite their high reactivity and consistently display very low toxicology profiles.^{48,49} Consequently, we designed our bicyclic boropeptides to be structurally close to Talabostat (Figure 4), a multi-target inhibitor, which entered Phase III clinical trials for the treatment of advanced non-small cell lung cancer.⁴² However, Talabostat displayed a loss in efficacy *in vivo* believed to be a result of a reversible intramolecular cyclization into an inactive cyclic adduct.⁵⁰ The constrained scaffold of our designed boronic acids **10f**, **11b**, **12c**, and **13b** would circumvent this cyclization.

Furthermore, in both POP and FAP, the boronic acid motif may act as a transition state analogue, forming both hydrogen bonds (with His680 and Tyr473 and with His734 and Tyr571, respectively) and covalent bonds with the catalytic triad (Ser554 and Ser624, respectively) in a tetrahedral configuration, as opposed to the trigonal planar configuration conferred by nitrile-containing inhibitors (Figure 4). Nonetheless, the design of FAP/POP dual inhibitors remains challenging due to the difference in polarity between the active sites. While three hydrophobic or hydrogen bond donor residues contribute the necessary interactions for high inhibition of POP (aromatic interactions with Phe173 and hydrogen bonding with Trp595 and Arg643), high inhibition of FAP relies on interaction with hydrogen bond acceptors Glu203 and Glu204 in the hydrophilic pocket (Figure 5).

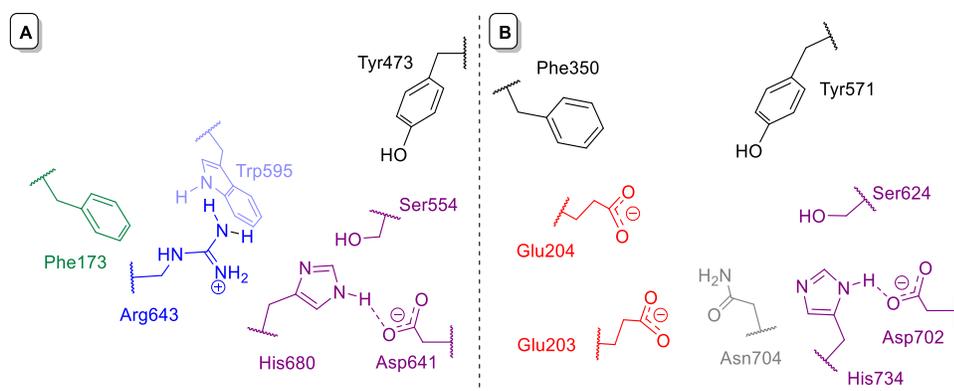


Figure 5. Schematic representation of the active sites of POP and FAP. Catalytic triads shown in purple. (A) POP: positively-charged pocket shown in blue, aromatic interaction residue shown in green; (B) FAP: negatively-charged pocket shown in red

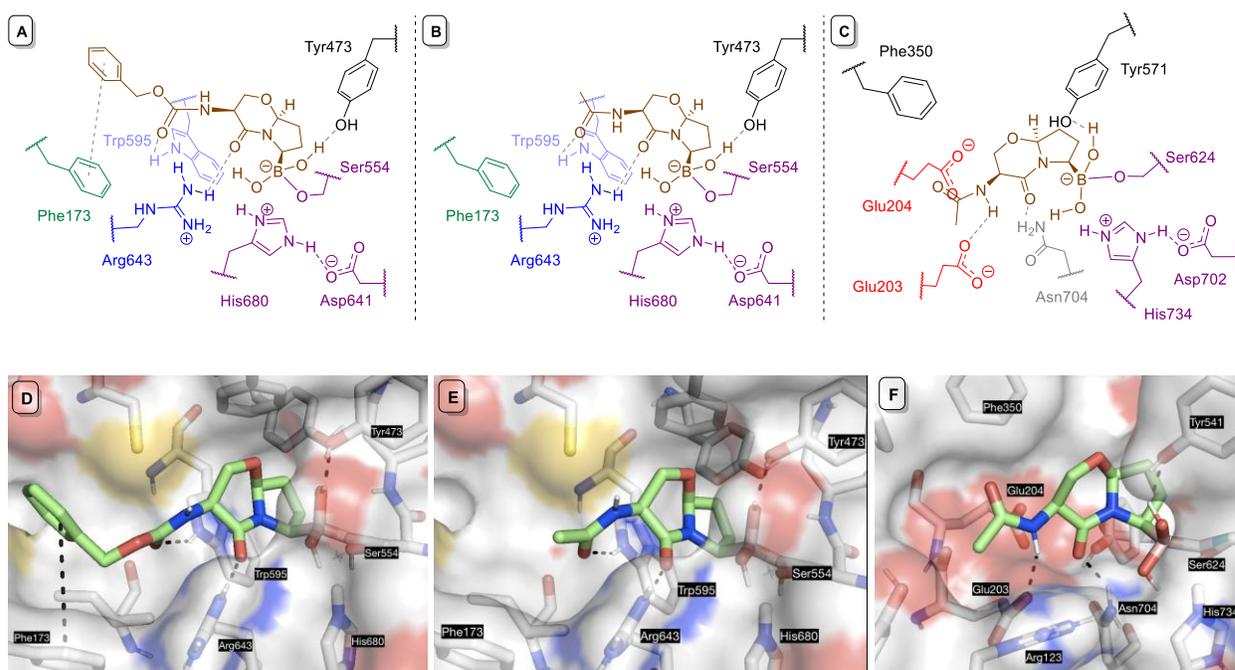


Figure 6. Predicted poses of compounds **10c** and **10d** in the active sites of POP and FAP. (A) schematic representation of the predicted binding mode of **10c** (brown) in POP; (B) schematic representation of the predicted binding mode of **10d** (brown) in POP; (C) Schematic representation of the predicted binding mode of **10d** (brown) in FAP; (D) predicted binding mode of **10c** (green) (pdb code: 2xdw); (E) predicted binding mode of **10d** (green) in POP; (F) predicted binding mode of **10d** (green) in FAP. All compounds were docked using FITTED. For schematic representations:

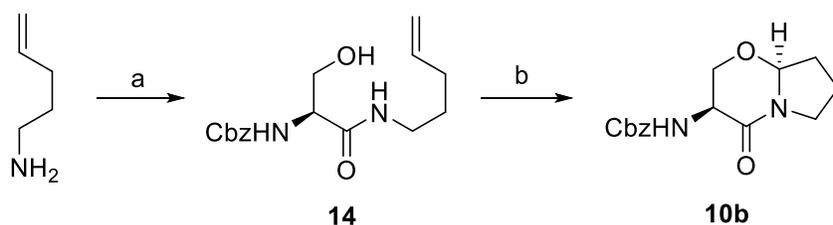
catalytic triads are shown in purple, key residues are shown in blue/green (POP) and red (FAP).
n.b. the hydrolyzed boronic esters (boronic acids) were docked.

Upon docking to POP, the *N*-Cbz boronic ester derivative **10c** was observed to fit very well into the active site of POP (Figure 6A), while docking to FAP gave unfavorable proposed binding modes, as the carboxybenzyl group is too large to fit into the active site (not shown). This compound is therefore expected to be selective for POP. After virtual optimization of the amide side chain, the acetyl group turned out to be an excellent compromise for the design of potent dual inhibitors, as key interactions were conserved. The *N*-acetyl group may act as a hydrogen bond donor in FAP (with Glu203 or 204) and as a hydrogen bond acceptor in POP (with Trp595). The docking-predicted binding mode of the *N*-acetyl analog in both POP and FAP is shown in Figure 6. Furthermore, in order to evaluate the impact of the covalent warhead, the non-covalent analog **10b** was also prepared.

Synthesis

Non-covalent series. The synthesis of this new [4.3.0] series started with the simplest of the analogues, the non-covalent inhibitor. Compound **10b** was synthesized in 3 overall steps starting with coupling *N*-Cbz-L-Ser to readily available 5-amino-1-pentene, followed by a telescoped acid-catalyzed oxidative cyclization (Scheme 1). Through the course of condition optimization, it was determined that performing the ozonolysis in presence of triphenylphosphine significantly increased the isolated yields and diastereoselectivity. In addition, a convenient procedure using resin-supported triphenylphosphine was developed in order to facilitate the purifications.

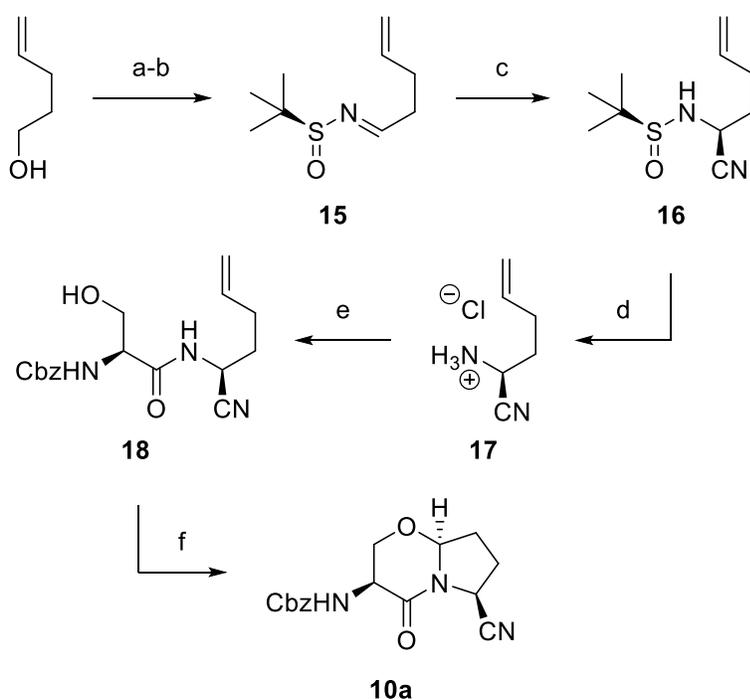
Scheme 1. Synthesis of the non-covalent series^a



^aa) *N*-Cbz-L-Ser, EDC·HCl, HOBt·H₂O, Et₃N, DCM, 0°C→rt, 18h, 78%; b) 1) O₃, Sudan III, PPh₃, DCM, -78 °C→rt, 20h; 2) TFA, DCM, rt, 2h, 40% over 2 steps.

Carbonitrile series. The synthesis of **10a** was unfortunately much more complex than that of the non-covalent analogue **10b**. Many attempts to obtain stereopure α -amino nitrile were unsuccessful; syntheses were long and yielded racemic mixtures. The synthesis was therefore redesigned, adapting chemistry from the Ellman group to obtain enantiopure sulfinylimine **15**,⁵¹ followed by a modified Strecker reaction adapted from Mabic *et. al.*⁵² to obtain sulfinamide **16**, which was deprotected in HCl to obtain stereopure α -amino carbonitrile **17** (Scheme 2). This amine was subsequently coupled to *N*-Cbz-L-Ser with good yield to give peptide **18**. Subsequent acid-catalyzed oxidative cyclization gave desired diastereopure inhibitor **10a**.

Scheme 2. Synthesis of the carbonitrile series^a



^aa) $(\text{COCl})_2$, DMSO, Et_3N , $-78^\circ\text{C} \rightarrow \text{rt}$, 2h; b) *(S)*-(-)-*tert*-butyl-sulfinylamide, CuSO_4 , DCM, rt, 18h, 62% over two steps; c) TMS-CN, $\text{Gd}(\text{OTf})_3$, $0^\circ\text{C} \rightarrow \text{rt}$, 48h, 60%, d.r. 97:3; d) HCl, Et_2O , 0°C , 1h, quant.; e) *N*-Cbz-L-Ser, EDC·HCl, HOBt·H₂O, Et_3N , DCM, $0^\circ\text{C} \rightarrow \text{rt}$, 18h, 80%; f) O_3 , Sudan III, PPh_3 , DCM, $-78^\circ\text{C} \rightarrow \text{rt}$, 20h; 2) TFA, DCM, rt, 2h, 33% over 2 steps

Boronic ester series. This series was prepared following a similar diastereoselective synthetic strategy, starting with the synthesis of sulfinylimine intermediate **15a**⁵¹ (

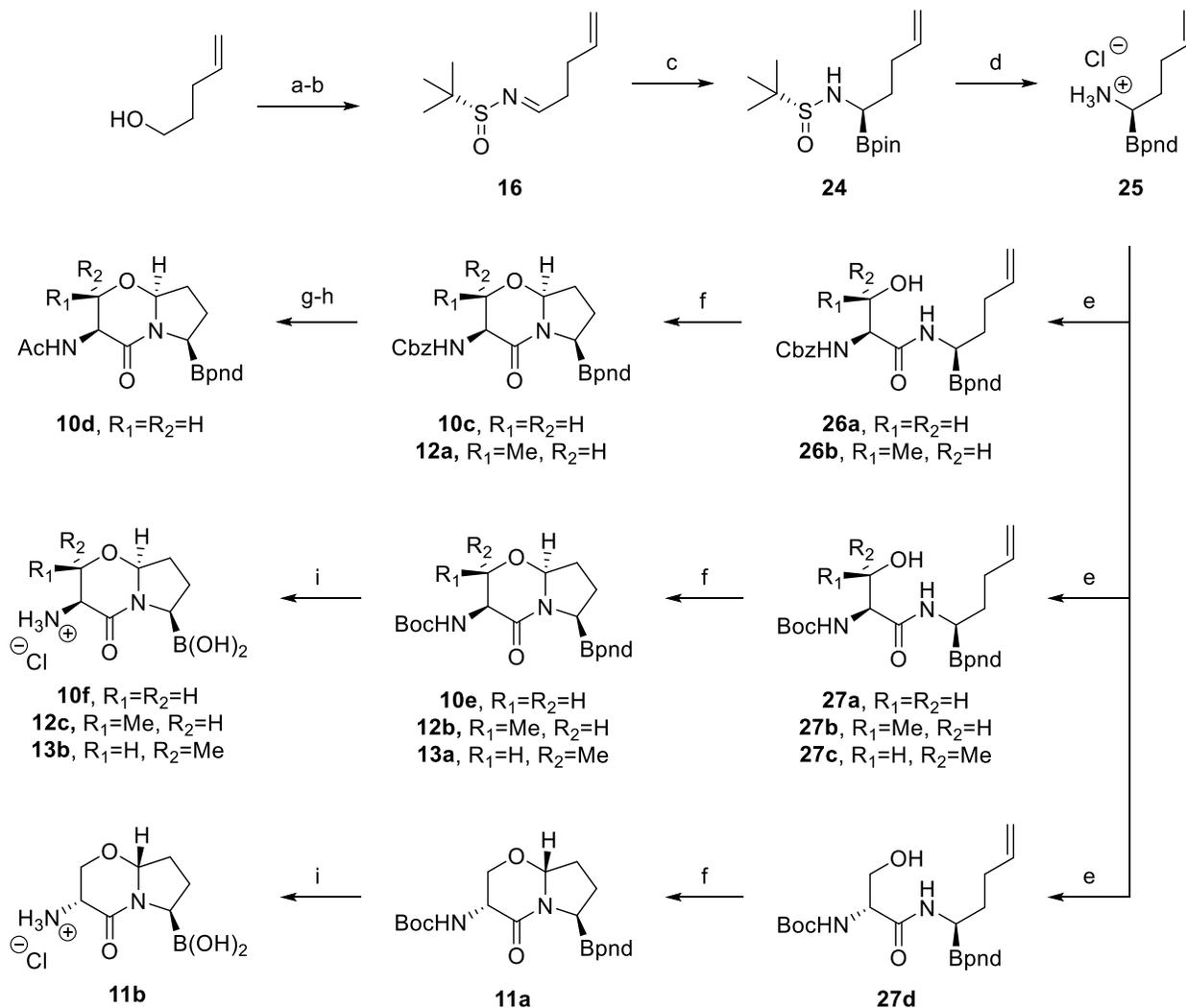
Scheme 3). The imine reacted under modified Ellman copper-catalyzed hydroboration conditions to afford the desired α -sulfinamidoboronic ester **19** with a good isolated yield and high diastereoselectivity.⁵³ A subsequent transesterification of the pinacol protecting group with the chiral (+)-pinanediol, followed by the deprotection of the sulfinamide group gave the highly diastereopure α -aminoboronic ester hydrochloride salt **20**. Peptide coupling of **20** provided the boropeptides, which were subjected to oxidative cleavage and dehydrative cyclization to obtain the corresponding bicycles as the sole diastereomers, confirmed by 1D nOe experiments (Figure 7).

To obtain the acetyl-protected bicyclic boronic ester, the *N*-Cbz-protected bicycle derivative **10c** was subjected to hydrogenation conditions, giving the free amine intermediate, which was subsequently reacted with AcCl to give the *N*-acetyl derivative **10d** (

Scheme 3). Unfortunately, attempts to purify the free amine boronic ester intermediate for biological testing were unsuccessful, as purification conditions affected the boronic ester group. This led us to another route, coupling amine **20** to *N*-Boc-protected amino acids, followed by acid-catalyzed oxidative cyclization and subsequent simultaneous removal of the Boc and (+)-pinanediol protecting groups, obtaining the free amine boronic acid bicycles with no necessary purification. Our group has previously demonstrated that boronic esters are quickly hydrolyzed to their respective boronic acids in the basic buffer used in the *in vitro* assays.^{43,44} The difference in covalent warhead within this series should therefore have negligible effect on the biological activity of these compounds. The complete synthesis of the boronic esters and acids is detailed in

Scheme 3.

Scheme 3. Synthesis of boronic ester series^a



^aa) $(\text{COCl})_2$, DMSO, Et_3N , $-78^\circ\text{C} \rightarrow \text{rt}$, 2h; b) *(R)*-*(+)*-*tert*-butyl-sulfinylamide, CuSO_4 , DCM, rt, 18h, 75% over two steps; c) B_2pin_2 , $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$, $\text{PCy}_3 \cdot \text{HBF}_4$, BnNH_2 , Toluene- H_2O 5:1, rt, 18h, 71%, d.r. > 98:2; d) 1) *(+)*-pinanediol, Et_2O , rt, 24h; 2) HCl , Et_2O , 0°C , 2h, 44% over 2 steps; e) PyBOP, L-AA (see Experimental Section), DIPEA, $0^\circ\text{C} \rightarrow \text{rt}$, 18h, 63% (**21a**), 48% (**21b**), 77% (**22a**), 69% (**22b**), 61% (**22c**), 57% (**22d**); f) 1) O_3 , DCM, PPh_3 , $-78^\circ\text{C} \rightarrow \text{rt}$, 20h; 2) TFA, DCM, rt, 2h, 58% (**10c**), 59% (**12a**), 52% (**10e**), 56% (**12b**), 60% (**13a**), 53% (**11a**); g) H_2 , Pd/C, AcOH, EtOAc, rt, 15h; h) AcCl, Et_3N , DMAP, $0^\circ\text{C} \rightarrow \text{rt}$, 2h, 63% over 2 steps; i) BCl_3 , DCM, -78°C , 1h, 34% (**10f**), 41% (**12c**), 52% (**13b**), 39% (**11b**). *pnd refers to *(+)*-pinanediol, pin refers to pinacol.

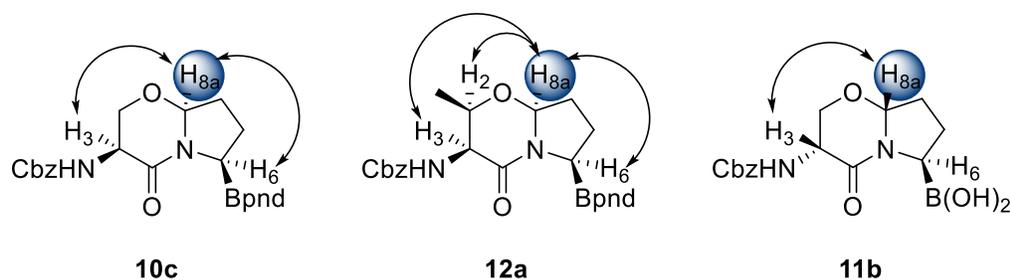
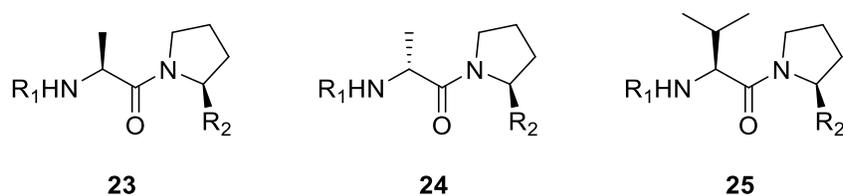


Figure 7. Selected nOe signals of the boron-containing bicycles.

Linear dipeptide series. As mentioned earlier, we expect that through the loss of flexibility of the dipeptide scaffold afforded by the constrained bicyclic core, this [4.3.0] alkane series would solve the problem of the cyclization of Talabostat *in vivo*. However, the bicycles are expected to be slightly less active than their linear counterparts, as they cannot adjust their shape to the binding site. Alternatively, reduced entropy penalty may improve their binding affinity over the more flexible Talabostat. To determine the effect of rigidification of the scaffold on biological activity, several linear dipeptide probes were also synthesized, starting from protected L- or D-Ala and L-Val, the latter of which gives inhibitors resembling Talabostat. The complete list of synthesized probes can be found in Table 2.

Table 2. Selected linear dipeptide analogues of the bicyclic series.



Entry	Compound	R ₁	R ₂
1	23a	Cbz	H
2	23b	Cbz	CN
3	23c	Cbz	Bpnd
4	23d	Boc	Bpnd
5	24a	Boc	Bpnd
6	25a	Cbz	Bpnd
7	25b	Boc	Bpnd

Table 3.

Table 3. *In vitro* activity of bicyclic and linear inhibitors against POP.

Entry	Compound	POP K_i (μM)
1	10a	0.0016 ± 0.0001
2	10b	4.4 ± 1.2
3	10c	0.0024 ± 0.0002
4	10d	1.1 ± 0.1
5	10e	0.0068 ± 0.0002
6	10f	2.2 ± 0.5
7	11a	0.34 ± 0.03
8	11b	6.2 ± 1.4
9	12a	0.0021 ± 0.0001
10	12b	0.0049 ± 0.0003
11	12c	0.84 ± 0.03
12	13a	0.12 ± 0.01
13	13b	7.2 ± 1.2
14	23a	53 ± 1
15	23b	0.00092 ± 0.00004
16	23c	0.00095 ± 0.00004
17	23d	0.0013 ± 0.00004
18	24a	1.0 ± 0.2
19	25a	0.0015 ± 0.0002
20	25b	0.0016 ± 0.00005
21	1 ^a	0.00029 ± 0.00004

^aCompound **1** was used as a positive control in the assay

The bicyclic boronic ester pro-drugs showed very potent *in vitro* activity against POP, with *N*-Cbz bicycles **10c** and **12a** exhibiting low single-digit nanomolar potency. This high inhibitor activity was predicted by the promising docking pose (Figure 6A/D) in which all three key ligand-protein interactions are fulfilled. Unexpectedly, the *N*-Boc derivative **10e** displayed similar

potency. Although the benzyl group is missing for aromatic interactions with Phe173, the large, greasy Boc protecting group might be compensating for this loss. The acetyl derivative **10d** exhibited low micromolar activity, likely attributed to the lack of a large *N*-bound group to stabilize the inhibitor in the active site. The free amine boronic acid derivative **10f** exhibited submicromolar potency in POP. This activity can likely be attributed to the assay conditions; the pH 8.0 basic buffer likely renders the amine neutral and allows it to act as a hydrogen bond acceptor to interact with Arg643. While, in general, nitrile and boronic acids are somewhat equipotent, boronic acids are likely to exhibit longer residence times in the active sites,^{43,54} making them more suitable drug candidates. As observed previously by our group,⁴⁴ it is likely that the nitrile is not properly oriented to react covalently with the catalytic serine in POP, and therefore binds non-covalently. Dose-response curves of the most potent POP bicyclic inhibitors can be found in Figure 8A.

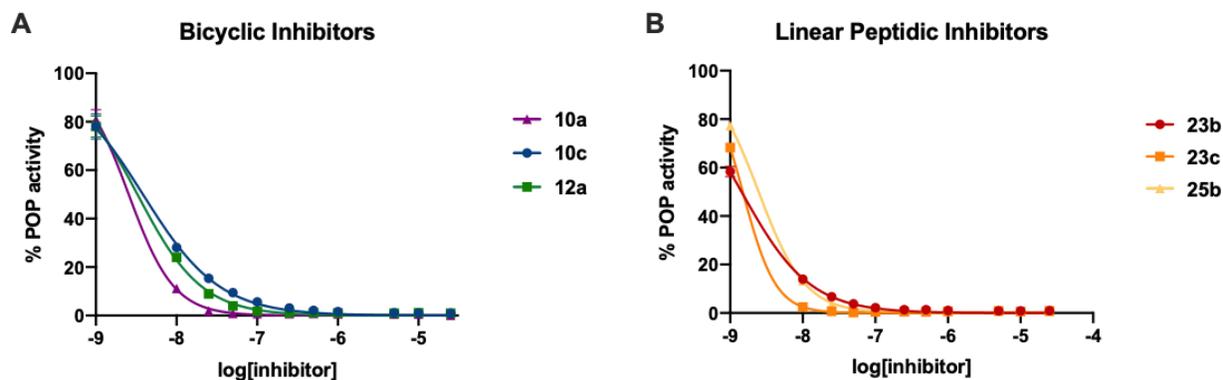


Figure 8. Dose response curves of the most potent POP inhibitors (A) the three most potent bicyclic inhibitors; (B) the three most potent linear peptidic inhibitors

The linear peptides displayed very high potencies against POP, demonstrating single-digit nanomolar potency or sub-nanomolar activity, the latter of which are on the same order of magnitude as aldehyde control inhibitor **1** and are our most potent POP inhibitors to date. The boron-containing dipeptides' trends in activity match those of their bicyclic counterparts, with D-amino acids exhibiting much lower potency than their L-amino acid analogues. Fortunately, their bicyclic analogues did not lose significant potency; L-amino acid cyclic analogues were generally in the single-digit nanomolar range. As predicted by our docking program, the designated [4.3.0] stereochemistry was optimal to inhibit the enzymes. Furthermore, the bicyclic compounds are likely to be more metabolically stable^{20,38} and more specific to our enzymatic targets *in vivo*.⁵⁵ Dose-response curves comparing of our top linear peptidic inhibitors can be found in Figure 8B.

The compounds predicted to be the most promising against FAP by docking, the *N*-acetyl bicyclic derivative and three of the free amines, were next tested against FAP. One of the Cbz-containing bicycles was also tested on FAP and displayed no inhibitory activity (see Supporting Info), confirming the need for smaller side chains in FAP inhibitors we proposed previously.¹ The results of the FAP assay are displayed in Table 4 and Figure 9. *In vitro* results indicate that free amine boronic acid **10f** exhibits nanomolar activity in FAP and low micromolar activity in POP, making it a promising dual inhibitor for future development. However, *N*-acetyl boronic ester derivative **12c** exhibits submicromolar activity in both enzymes and comparable potency to failed clinical trial candidate Talabostat against POP (Figure 1), making it a very promising drug candidate for future studies.

Table 4. *In vitro* activity of bicyclic dual inhibitors

Entry	Compound	POP K_i (μM)	FAP K_i (μM)
1	10d	1.1 ± 0.1	1.3 ± 0.5
2	10f	2.2 ± 0.5	0.20 ± 0.05
3	11b	6.2 ± 1.4	14 ± 2
4	12c	0.84 ± 0.03	0.72 ± 0.09
5	Talabostat ^a	0.98 ± 0.06	0.066 ± 0.011

^aValues are reported as IC_{50} concentrations by Jansen *et al.*⁵⁶

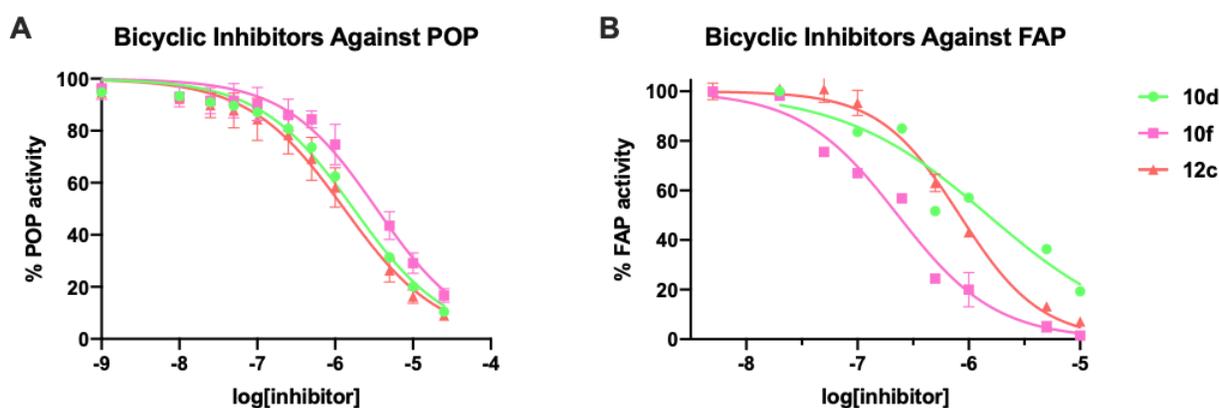


Figure 9. Dose response curves of our most potent dual inhibitors. (A) POP (B) FAP

Conclusion.

Our group's research has previously led to potent bicycle-based POP inhibitors, revealing that the introduction of bicyclic scaffolds can enhance the metabolic stability of these inhibitors.^{20,38} In the shift toward POP-FAP dual inhibitors, we then aimed to improve these bicyclic scaffolds while simultaneously constraining the known inhibitor Cbz-Pro-Prolinal **1** and failed drug candidate Talabostat. Our results indicate that we were not only able to obtain potent compounds using our computationally guided optimizations of known inhibitors, but that we were able to use this method along with synthetic developments to produce an inhibitor with comparable potency to a drug that reached Phase III clinical trials. Currently, we are carrying out cell-based assays to assess the activity of our leads *in cellulo*, as well as performing further experiments to optimize the activity and pharmacokinetic properties of our leads including **10a**, **10c**, and **12a** as selective POP inhibitors and **10f** and **12c** as dual inhibitors.

Experimental Section.

In Vitro Assays. POP *in vitro* assays were performed as previously published by our group.^{20,38,43} The POP batch used in these assays exhibited a K_m of 141.2 μM and k_{cat} of 21.2 s^{-1} . The FAP assay was performed using the FAP Assay Kit from BPSBioscience.⁵⁷ The FAP batch used in these assays exhibited a K_m of 33 μM .

Synthesis. All commercially available reagents were used without further purification. All reactions, unless otherwise indicated, were carried out in flame-dried flasks under argon atmosphere with anhydrous solvents. FTIR spectra were recorded using a Perkin-Elmer Spectrum One FT-IR. ^1H , ^{13}C , and ^{11}B NMR spectra were recorded on a Bruker 400 or 500 MHz spectrometer. Chemical shifts are reported in ppm using the residual of deuterated solvents as an internal standard. Thin layer chromatography visualization was performed by UV or by development using KMnO_4 , Curcumin, ninhydrin, or *p*-anisaldehyde. Chromatography was performed on silica gel 60 (230–240 mesh). High resolution mass spectrometry was performed by ESI on a Bruker Maxis Impact API QqTOF or by ESI or APCI on a ThermoFisher Exactive Plus Orbitrap-API at McGill University. Prior to biological testing, reverse-phase HPLC was used to verify the purity of compounds on an Agilent 1100 series instrument, equipped with VWD-detector, using a C18 reverse column (Agilent, Eclipse -C18 150 mm \AA ~ 4.6 mm, 5 μm) or Zorbax

(ZORBAX Bonus-RP, 80Å, 4.6 x 150 mm, 5 µm) with UV detection at 220 or 215 nm. All tested compounds were at least 95% pure. All compounds were stored at -20°C.

(S)-2-methyl-N-(pent-4-en-1-yl)propane-2-sulfinamide (15) and (R)-2-methyl-N-(pent-4-en-1-ylidene)propane-2-sulfinamide (15a) (*general procedure for both; the two sulfinimines are spectrally identical*) Oxalyl chloride (1.2 eq) was dissolved in DCM (1.5 M) under Ar, and the solution was cooled to -78°C. DMSO (2.5 eq) in DCM (7 M) was added *slowly*. The solution stirred for 5 minutes. 4-penten-1-ol (1 eq) in DCM (3 M) was added *slowly*, and the reaction stirred for 15 minutes. Triethylamine (3 eq) was added slowly, and the reaction stirred for 2 hours at room temperature. Water was added, and the product was extracted with DCM. The combined organic layers were washed with 1M HCl, saturated NaHCO₃, and brine, dried over Na₂SO₄, filtered and concentrated *in vacuo* at 650 mbar, 40°C. (Some solvent remains; product is volatile.) The resultant 4-pentenal (assume 100% yield) was dissolved in anhydrous DCM (0.5 M), and (*R*)-(+)-2-methyl-2-propanesulfinamide (1 eq) [*or the (S)-(-) enantiomer for the synthesis of the (S)-sulfinimine*] and anhydrous CuSO₄ (3 eq) were added. The reaction stirred at room temperature overnight. The mixture was then filtered through a pad of Celite®, and the filter cake was rinsed with DCM. The filtrate was concentrated *in vacuo* to give a brown liquid, which was purified by flash chromatography on a silica gel column (85:15 hexanes-EtOAc) to give a yellow liquid (**15** 62%, **15a** 75%). *R_f* = 0.45 (85:15 hexanes-EtOAc); ¹H NMR (500 MHz, Chloroform-*d*) δ 8.07 (t, *J* = 4.4 Hz, 1H), 5.83 (ddt, *J* = 16.8, 10.2, 6.5 Hz, 1H), 5.07 (dq, *J* = 17.2, 1.7 Hz, 1H), 5.02 (dq, *J* = 10.2, 1.5 Hz, 1H), 2.62 (td, *J* = 7.3, 4.4 Hz, 2H), 2.42 – 2.36 (m, 2H), 1.18 (s, 9H). ¹³C NMR (126 MHz, Chloroform-*d*) δ 168.94, 136.80, 115.97, 56.69, 35.41, 29.49, 22.49. Spectral and physical data were in accordance with the literature.^{58,59}

(S)-N-((S)-1-cyanopent-4-en-1-yl)-2-methylpropane-2-sulfinamide (16) Imine **15** (462 mg, 1 eq) was dissolved in DCM (24 mL), and Gd(OTf)₃ (298 mg, 0.2 eq) and TMSCN (489 mg, 0.62 mL, 2 eq) were added. The reaction stirred for 48h at room temperature and was quenched with saturated NaHCO₃. The product was extracted with DCM, and the combined organic layers were washed with brine, dried over Na₂SO₄, and concentrated *in vacuo* to give the crude product as a brown oil, which was purified by flash chromatography on a silica gel column (eluent 70:30 hexanes-EtOAc) to give the product as a yellow oil (318 mg, 60%). *R_f* = 0.50 (70:30 EtOAc-hexanes); IR (film) cm⁻¹ 3187, 3083, 2960, 2238, 1641, 1391, 1366, 1062, 911; ¹H NMR (500

MHz, Chloroform-*d*) δ 5.84 – 5.69 (m, 1H), 5.16 – 5.06 (m, 2H), 4.23 – 4.09 (m, 1H), 3.93 – 3.63 (m, 1H), 2.37 – 2.21 (m, 2H), 2.15 – 1.89 (m, 2H), 1.24 (s, 9H); ^{13}C NMR (126 MHz, Chloroform-*d*) δ 22.61 (3C), 29.47, 34.01, 45.69, 57.21, 117.28, 119.19, 135.69; HRMS (ESI+) *m/z* calcd for $[\text{C}_{10}\text{H}_{18}\text{ON}_2\text{S} + \text{Na}]^+$ 237.1032, found 237.1035.

(S)-1-cyanopent-4-en-1-aminium chloride (17) Sulfinamide **16** (298 mg, 1 eq) was dissolved in Et₂O (12 mL), and the solution was cooled to 0°C. HCl (2 M in Et₂O, 2.1 mL, 3 eq) was added dropwise, and the resultant mixture stirred for 1h at 0°C. The solvent was removed in vacuo to give the product as a white solid, which was taken to the next step without purification (136 mg, quant.) *R_f*= does not elute on silica-backed TLC plates; mp = 94-97°C; IR (neat) cm⁻¹ 3071, 2956, 1643, 1483, 1185, 926; ^1H NMR (400 MHz, Methanol-*d*₄) δ 5.86 (dddd, *J* = 17.2, 10.2, 7.1, 6.1 Hz, 1H), 5.18 (dq, *J* = 17.1, 1.6 Hz, 1H), 5.11 (dq, *J* = 10.2, 1.3 Hz, 1H), 4.56 – 4.35 (m, 1H), 2.45 – 2.22 (m, 2H), 2.11 – 2.00 (m, 2H); ^{13}C NMR (101 MHz, Methanol-*d*₄) δ 30.23, 31.26, 42.23, 116.62, 117.59, 136.33; HRMS (ESI+) *m/z* calcd for $[\text{C}_6\text{H}_{11}\text{N}_2]^+$ 111.0917, found 111.0922.

(R)-2-methyl-N-((R)-1-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)pent-4-en-1-yl)propane-2-sulfinamide (19) Tricyclohexylphosphonium tetrafluoroborate (63 mg, 0.1 eq) was dissolved in toluene (2.1 mL), and the solution was stirred rapidly. Copper (II) sulfate pentahydrate (43 mg, 0.1 eq) and water (0.9 mL) were added, turning the reaction light blue. Benzylamine (37 mg, 0.2 eq) was added, turning the mixture dark blue. The mixture stirred at room temperature for 10 minutes, and was then cooled to 0°C. The sulfinylimine **15a** (320 mg, 1 eq) in toluene (2.1 mL) was added, followed by B₂pin₂ (651 mg, 1.5 eq), and the reaction mixture turned turquoise. The reaction was kept at 0°C for 15 minutes, then was warmed to room temperature and stirred overnight, after which the reaction turned dark brown. The mixture was diluted with ethyl acetate and quenched with saturated NaHCO₃. The biphasic mixture stirred for 30 minutes. The product was then extracted from the aqueous layer with ethyl acetate, and the organic layer was washed with saturated NH₄Cl, saturated NaHCO₃ (copiously) and brine, dried over Na₂SO₄, and concentrated *in vacuo* to give the crude product as a brown oil, which was clarified with charcoal to give a clear oil (385 mg, 71%). *R_f*= streaks on regular silica; IR (film) cm⁻¹ 3206, 3079, 2976, 1639, 1380, 1370, 1332, 1142, 1058, 910; ^1H NMR (500 MHz, Chloroform-*d*) δ 5.73 (ddt, *J* = 16.9, 10.2, 6.7 Hz, 1H), 4.96 (dq, *J* = 17.1, 1.7 Hz, 1H), 4.88 (dq, *J* = 10.1, 1.4 Hz, 1H), 3.16 (d, *J* = 7.0 Hz, 1H), 2.97 (q, *J* = 7.0 Hz, 1H), 2.07 (m, 2H), 1.77 – 1.64 (m, 2H), 1.18 (s, 6H), 1.17 (s,

6H), 1.11 (s, 9H); ^{13}C NMR (126 MHz, Chloroform-*d*) δ 22.52 (3C), 24.52 (2C), 24.92 (2C), 31.01, 32.66, 42.75, 55.97, 84.00 (2C), 115.07, 138.01; ^{11}B NMR (161 MHz, Chloroform-*d*) δ 32.41; HRMS (ESI+) *m/z* calcd for $[\text{C}_{15}\text{H}_{30}\text{O}_3\text{NSB} + \text{Na}]^+$ 338.1932, found 338.1931.

(R)-1-((3a*S*,4*S*,6*S*,7a*R*)-3a,5,5-trimethylhexahydro-4,6-methanobenzo[*d*][1,3,2]dioxaborol-2-yl)pent-4-en-1-aminium chloride (20) The aminoboronic ester **19** (11.51 g, 1 eq) was dissolved in Et₂O (120 mL), and (+)-pinanediol (6.22 g, 1 eq) was added. The solution stirred at room temperature for 24 hours. The solvent was removed *in vacuo*, the residue was re-dissolved in Et₂O (75 mL), and the solution was cooled to 0°C. HCl (2M in Et₂O, 24 mL, 1.3 eq) was added dropwise, and the argon balloon was removed. After 2h of stirring at room temperature, the solvent was removed *in vacuo* to give a white solid, which was triturated at 0°C in 2:1 *n*-hexane/Et₂O and filtered, rinsing with cold 2:1 *n*-hexane/Et₂O, to give the boroamine salt **20** as a fluffy white solid (4.85 g, 44% over 2 steps). *R_f* = does not elute on silica-backed TLC plates; mp = 168–171°C; IR (film) cm⁻¹ 3130, 2921, 1605, 1405, 1389, 1076, 1029; ^1H NMR (500 MHz, Chloroform-*d*) δ 8.31 (s, br, 3H), 5.78 (ddt, *J* = 16.6, 11.1, 6.0 Hz, 1H), 5.13 (d, *J* = 16.9 Hz, 1H), 4.99 (d, *J* = 10.0 Hz, 1H), 4.38 (d, *J* = 8.2 Hz, 1H), 2.97 (s, br, 1H), 2.43 – 2.14 (m, 4H), 2.16 – 1.95 (m, 3H), 1.98 – 1.88 (m, 2H), 1.41 (s, 3H), 1.27 (s, 3H), 1.17 (d, *J* = 10.9 Hz, 1H), 0.81 (s, 3H); ^{13}C NMR (126 MHz, CDCl₃) δ 24.08, 26.71, 27.13, 28.61, 29.04, 30.71, 35.17, 37.12, 38.25, 39.57, 51.21, 77.16, 78.88, 87.74, 116.27, 137.12; ^{11}B NMR (161 MHz, CDCl₃) δ 32.35; HRMS (APCI+) *m/z* calcd for $[\text{C}_{15}\text{H}_{26}\text{O}_2\text{NB} + \text{H}]^+$ 264.2129, found 264.2129.

General peptide coupling procedure A. The protected amino acid (1 eq) was suspended in DCM (0.1 M), and HOBt•H₂O (1.2 eq) was added, followed by EDC•HCl (1.2 eq). The reaction stirred at 0°C for 1h. The amine salt (1 eq) was then added, followed by Et₃N (3 eq). The reaction stirred at 0°C for 1h, then at room temperature overnight. Water was added, and the product was extracted with DCM. The combined organic layers were washed with saturated NH₄Cl, saturated NaHCO₃, and brine, dried over Na₂SO₄, filtered, and concentrated *in vacuo* to give the crude product. The crude material was purified by flash chromatography on a silica gel column to give the corresponding dipeptide.

General peptide coupling procedure B. The protected amino acid (1 eq) was suspended in DCM (0.1 M), and Et₃N (5 eq) was added, followed by Piv-Cl (1.1 eq). The reaction stirred at 0°C for 1h. The amine salt (1 eq) was then added. The reaction stirred at 0°C for 1h, then at room

temperature overnight. Water was added, and the product was extracted with DCM. The combined organic layers were washed with saturated NH_4Cl , saturated NaHCO_3 , and brine, dried over Na_2SO_4 , filtered, and concentrated *in vacuo* to give the crude product. The crude material was purified by flash chromatography on a silica gel column to give the corresponding dipeptide.

General peptide coupling procedure C. The protected amino acid (1 eq) was suspended in DCM (0.1 M), and PyBOP (1.2 eq) was added, followed by the amine (1 eq), then DIPEA (3 eq). The reaction stirred at 0°C for 1h, then at room temperature overnight. Water was added, and the product was extracted with DCM. The combined organic layers were washed with saturated NH_4Cl , saturated NaHCO_3 , and brine, dried over Na_2SO_4 , filtered, and concentrated *in vacuo* to give the crude product. The crude material was purified by flash chromatography on a silica gel column to give the corresponding dipeptide.

General peptide coupling procedure D. The protected amino acid (1 eq) was dissolved in DMF (0.3 M), and the solution was cooled to 0°C . HATU (1.2 eq) was added, followed by the amine (1 eq), then Et_3N (10 eq). The reaction stirred at room temperature overnight. Water was added, and the product was extracted with DCM. The combined organic layers were washed with saturated NH_4Cl , saturated NaHCO_3 , and brine, dried over Na_2SO_4 , filtered, and concentrated *in vacuo* to give the crude product. The crude material was purified by flash chromatography on a silica gel column to give the corresponding dipeptide.

Benzyl (S)-(3-hydroxy-1-oxo-1-(pent-4-en-1-ylamino)propan-2-yl)carbamate (14). Dipeptide **14** was prepared following general peptide coupling procedure A, using *Z*-L-Ser as the amino acid and readily available 5-amino-1-pentene trifluoroacetate (Supporting Information) as the amine. The crude product was purified by flash chromatography on a silica gel column (eluent 70:30 EtOAc-hexanes) to give the product as a white solid (78%). $R_f = 0.43$ (80:20 EtOAc-hexanes); mp = $143\text{--}145^\circ\text{C}$; IR (film) cm^{-1} 3316, 3068, 2937, 1709, 1651, 1532, 1239, 1060, 913; ^1H NMR (500 MHz, Chloroform-*d*) δ 7.39 – 7.30 (m, 5H), 6.64 (s, 1H), 5.87 (d, $J = 7.6$ Hz, 1H), 5.77 (ddt, $J = 16.9, 10.2, 6.6$ Hz, 1H), 5.13 (s, 2H), 5.02 (dq, $J = 17.2, 1.7$ Hz, 1H), 4.98 (dq, $J = 10.2, 1.4$ Hz, 1H), 4.17 (ddd, $J = 7.8, 4.9, 3.2$ Hz, 1H), 4.15 – 4.05 (m, 1H), 3.71 – 3.60 (m, 1H), 3.25 (q, $J = 6.7$ Hz, 2H), 3.20 (s, 1H), 2.06 (q, $J = 7.2$ Hz, 2H), 1.59 (p, $J = 7.3$ Hz, 2H); ^{13}C NMR (126 MHz, Chloroform-*d*) δ 28.58, 31.09, 39.15, 55.29, 62.91, 67.51, 115.55, 128.23 (2C), 128.50,

128.74 (2C), 136.06, 137.64, 156.93, 171.00; HRMS (ESI+) m/z calcd for $[C_{16}H_{22}O_4N_2 + Na]^+$ 329.1472, found 329.1464.

Benzyl ((S)-1-(((S)-1-cyanopent-4-en-1-yl)amino)-3-hydroxy-1-oxopropan-2-yl)carbamate (18). The product was synthesized following general coupling procedure D, using **17** as the amine. The crude product was purified by silica gel (eluent 60:40 EtOAc-hexanes) to give the product as a white foam (80%). $R_f = 0.27$ (60:40 EtOAc-hexanes); mp = 87-91°C; IR (film) cm^{-1} 3409, 3321, 3278, 3020, 2940, 1707, 1671, 1516, 1217, 1058; 1H NMR (500 MHz, Acetone- d_6) δ 8.20 – 7.95 (m, 1H), 7.47 – 7.18 (m, 5H), 6.48 (d, $J = 8.7$ Hz, 1H), 5.82 (ddt, $J = 17.0, 10.1, 6.7$ Hz, 1H), 5.15 – 5.04 (m, 3H), 5.01 (dq, $J = 10.2, 1.3$ Hz, 1H), 4.33 – 4.18 (m, 2H), 3.87 (dt, $J = 10.3, 5.0$ Hz, 1H), 3.81 (dt, $J = 10.7, 5.2$ Hz, 1H), 2.23 (q, $J = 7.4$ Hz, 2H), 2.01 – 1.91 (m, 2H); ^{13}C NMR (126 MHz, Acetone- d_6) δ 30.09, 32.56, 40.53, 57.85, 62.99, 67.01, 116.59, 119.55, 128.70 (2C), 128.71, 129.24 (2C), 137.34, 137.98, 157.01, 171.22; HRMS (ESI+) m/z calcd for $[C_{17}H_{21}O_4N_3 + Na]^+$ 354.1424, found 354.1417.

Benzyl ((S)-3-hydroxy-1-oxo-1-(((R)-1-((3aS,4S,6S,7aR)-3a,5,5-trimethylhexahydro-4,6-methanobenzo[d][1,3,2]dioxaborol-2-yl)pent-4-en-1-yl)amino)propan-2-yl)carbamate (21a). The product was synthesized following general coupling procedure C, using **20** as the amine. The crude product was purified by silica gel (eluent 60:40 EtOAc-hexanes) to give the product as a white foam (63%). $R_f = 0.23$ (50:50 EtOAc-hexanes); IR (neat) cm^{-1} 3310, 3071, 2929, 1722, 1701, 1522, 1385, 1372, 1247, 1078, 1052, 906; 1H NMR (500 MHz, Acetone- d_6) δ 8.04 (s, 1H), 7.46 – 7.21 (m, 5H), 6.63 – 6.43 (m, 1H), 5.82 (ddt, $J = 17.0, 10.3, 6.7$ Hz, 1H), 5.10 (s, 2H), 4.99 (dq, $J = 17.1, 1.8$ Hz, 1H), 4.91 (ddt, $J = 10.2, 2.2, 1.2$ Hz, 1H), 4.44 – 4.35 (m, 1H), 4.31 (dt, $J = 7.9, 5.1$ Hz, 1H), 4.24 (dd, $J = 8.8, 2.2$ Hz, 1H), 3.85 (dt, $J = 10.8, 5.3$ Hz, 1H), 3.77 (dt, $J = 11.3, 5.8$ Hz, 1H), 2.84 (ddd, $J = 7.8, 6.3, 4.3$ Hz, 1H), 2.31 (ddt, $J = 13.9, 8.7, 2.5$ Hz, 1H), 2.22 – 2.09 (m, 3H), 1.94 (t, $J = 5.6$ Hz, 1H), 1.84 (tt, $J = 5.9, 3.0$ Hz, 1H), 1.78 (ddd, $J = 14.2, 3.2, 2.1$ Hz, 1H), 1.73 – 1.64 (m, 1H), 1.64 – 1.57 (m, 1H), 1.48 – 1.40 (m, 1H), 1.34 (s, 3H), 1.27 (s, 3H), 0.86 (s, 3H); ^{13}C NMR (126 MHz, Acetone- d_6) δ 24.35, 26.98, 27.56, 29.25, 31.32, 32.08, 36.74, 38.77, 39.73, 40.62, 52.70, 56.55, 63.10, 66.99, 77.75, 85.09, 114.89, 128.62 (2C), 128.65, 129.19 (2C), 137.96, 139.69, 156.93, 173.39; ^{11}B NMR (161 MHz, Acetone- d_6) δ 26.64; HRMS (APCI+) m/z calcd for $[C_{26}H_{37}O_6N_2B + Na]^+$ 507.2637, found 507.2652.

Benzyl ((2*S*,3*R*)-3-hydroxy-1-oxo-1-(((*R*)-1-((3*aS*,4*S*,6*S*,7*aR*)-3*a*,5,5-trimethylhexahydro-4,6-methanobenzo[*d*][1,3,2]dioxaborol-2-yl)pent-4-en-1-yl)amino)butan-2-yl)carbamate

(21b). The product was synthesized following general coupling procedure C, using 20 as the amine. The crude product was purified by silica gel (eluent 60:40 EtOAc-hexanes) to give the product as a white solid (48%). $R_f = 0.38$ (50:50 EtOAc-hexanes); mp = 44–47°C; IR (film) cm^{-1} 3322, 3068, 2922, 1722, 1699, 1606, 1515, 1383, 1373, 1247, 1120, 1070, 908; ^1H NMR (500 MHz, Acetone- d_6) δ 7.91 (s, 1H), 7.43 – 7.28 (m, 5H), 6.25 (d, $J = 8.2$ Hz, 1H), 5.82 (ddt, $J = 17.0, 10.2, 6.7$ Hz, 1H), 5.11 (s, 2H), 4.99 (dq, $J = 17.2, 1.7$ Hz, 1H), 4.91 (ddt, $J = 10.1, 2.3, 1.2$ Hz, 1H), 4.30 (d, $J = 4.3$ Hz, 1H), 4.25 (dd, $J = 8.8, 2.2$ Hz, 1H), 4.22 – 4.13 (m, 2H), 2.87 – 2.81 (m, 1H), 2.31 (ddt, $J = 14.0, 8.8, 2.5$ Hz, 1H), 2.21 – 2.07 (m, 3H), 1.95 (dd, $J = 11.1, 5.6$ Hz, 1H), 1.84 (tq, $J = 6.0, 3.1$ Hz, 1H), 1.81 – 1.74 (m, 1H), 1.74 – 1.65 (m, 1H), 1.65 – 1.44 (m, 1H), 1.45 – 1.37 (m, 1H), 1.34 (s, 3H), 1.27 (s, 3H), 1.22 – 1.15 (m, 3H), 0.86 (s, 3H); ^{13}C NMR (126 MHz, Acetone- d_6) δ 19.59, 24.36, 26.99, 27.56, 29.25, 31.32, 32.14, 36.72, 38.80, 39.44, 40.62, 52.69, 59.82, 67.00, 67.96, 77.85, 85.23, 114.93, 128.57 (2C), 128.66, 129.21 (2C), 138.04, 139.64, 157.12, 173.17; ^{11}B NMR (161 MHz, Acetone- d_6) δ 27.32; HRMS (ESI+) m/z calcd for $[\text{C}_{27}\text{H}_{39}\text{O}_6\text{N}_2\text{B} + \text{Na}]^+$ 521.2793, found 521.2804.

***tert*-Butyl ((*S*)-3-hydroxy-1-oxo-1-(((*R*)-1-((3*aS*,4*S*,6*S*,7*aR*)-3*a*,5,5-trimethylhexahydro-4,6-methanobenzo[*d*][1,3,2]dioxaborol-2-yl)pent-4-en-1-yl)amino)propan-2-yl)carbamate**

(22a). The product was synthesized following general coupling procedure C, using Boc-L-Ser as the amino acid and 20 as the amine. The crude product was purified by silica gel (eluent 70:30 EtOAc-hexanes) to give the product as a white solid (77%). $R_f = 0.23$ (60:40 hexanes-EtOAc); mp = 49–52°C; IR (film) cm^{-1} 3425, 3345, 3079, 3020, 2980, 1709, 1653, 1607, 1504, 1389, 1368, 1215, 1167, 1054; ^1H NMR (400 MHz, Acetone- d_6) δ 7.87 (s, 1H), 6.04 (d, $J = 8.4$ Hz, 1H), 5.83 (ddt, $J = 17.0, 10.2, 6.7$ Hz, 1H), 5.00 (dq, $J = 17.1, 1.7$ Hz, 1H), 4.91 (ddt, $J = 10.2, 2.4, 1.2$ Hz, 1H), 4.25 (dd, $J = 8.7, 2.2$ Hz, 1H), 4.23 – 4.15 (m, 2H), 3.82 (dt, $J = 11.0, 5.5$ Hz, 1H), 3.77 – 3.68 (m, 1H), 2.88 – 2.78 (m, 1H), 2.31 (ddt, $J = 13.9, 8.8, 2.5$ Hz, 1H), 2.21 – 2.07 (m, 3H), 2.00 – 1.91 (m, 1H), 1.85 (tt, $J = 5.7, 3.0$ Hz, 1H), 1.78 (ddd, $J = 14.2, 3.3, 2.2$ Hz, 1H), 1.75 – 1.65 (m, 1H), 1.65 – 1.56 (m, 1H), 1.42 (s, 9H), 1.34 (s, 3H), 1.27 (s, 3H), 0.86 (s, 3H); ^{13}C NMR (101 MHz, Acetone- d_6) δ 23.48, 26.13, 26.70, 27.64 (3C), 28.41, 30.55, 31.22, 35.89, 37.92, 38.45, 39.76, 51.85, 55.31, 62.30, 76.95, 78.75, 84.29, 113.99, 138.87, 155.39, 172.64; ^{11}B NMR (128

MHz, Acetone- d_6) δ 26.33; HRMS (ESI+) m/z calcd for $[C_{23}H_{39}O_6N_2B + Na]^+$ 473.2793, found 473.2794.

***tert*-Butyl ((2*S*,3*R*)-3-hydroxy-1-oxo-1-(((*R*)-1-((3*aS*,4*S*,6*S*,7*aR*)-3*a*,5,5-trimethylhexahydro-4,6-methanobenzo[*d*][1,3,2]dioxaborol-2-yl)pent-4-en-1-yl)amino)butan-2-yl)carbamate (22b).** The product was synthesized following general coupling procedure C, using Boc-L-Thr as the amino acid and 20 as the amine. The crude product was purified by silica gel (eluent 60:40 EtOAc-hexanes) to give the product as a white solid (69%). R_f = 0.50 (60:40 EtOAc-hexanes); mp = 54-57°C; IR (film) cm^{-1} 3492, 3349, 3198, 3079, 2976, 1720, 1695, 1607, 1500, 1391, 1368, 1238, 1167, 1078, 1052, 884; 1H NMR (500 MHz, Acetone- d_6) δ 7.99 (s, 1H), 5.94 – 5.72 (m, 2H), 5.00 (dq, J = 17.1, 1.7 Hz, 1H), 4.91 (ddd, J = 10.1, 2.3, 1.2 Hz, 1H), 4.38 (d, J = 4.7 Hz, 1H), 4.24 (dd, J = 8.8, 2.2 Hz, 1H), 4.18 (dq, J = 11.1, 6.2, 5.0 Hz, 1H), 4.15 – 3.93 (m, 1H), 2.82 (td, J = 6.9, 4.1 Hz, 1H), 2.31 (ddt, J = 14.0, 8.7, 2.5 Hz, 1H), 2.25 – 2.07 (m, 3H), 2.00 – 1.91 (m, 1H), 1.91 – 1.83 (m, 1H), 1.79 (ddd, J = 14.1, 3.2, 2.3 Hz, 1H), 1.75 – 1.67 (m, 1H), 1.66 – 1.48 (m, 2H), 1.43 (s, 9H), 1.35 (s, 3H), 1.27 (s, 3H), 1.16 (d, J = 6.2 Hz, 3H), 0.86 (s, 3H); ^{13}C NMR (126 MHz, Acetone- d_6) δ 19.71, 24.38, 27.04, 27.59, 28.51 (3C), 31.44, 32.18, 35.24, 36.85, 38.79, 39.85, 40.66, 52.79, 59.24, 67.95, 77.72, 79.66, 85.00, 114.87, 139.72, 156.48, 173.95; ^{11}B NMR (161 MHz, Acetone- d_6) δ 26.18; HRMS (ESI+) m/z calcd for $[C_{24}H_{41}O_6N_2B + H]^+$ 465.3130, found 465.3140.

***tert*-Butyl ((2*S*,3*S*)-3-hydroxy-1-oxo-1-(((*R*)-1-((3*aS*,4*S*,6*S*,7*aR*)-3*a*,5,5-trimethylhexahydro-4,6-methanobenzo[*d*][1,3,2]dioxaborol-2-yl)pent-4-en-1-yl)amino)butan-2-yl)carbamate (22c).** The product was synthesized following general coupling procedure C, using Boc-L-*allo*-Thr as the amino acid and 20 as the amine. The crude product was purified by silica gel (eluent 60:40 EtOAc-hexanes) to give the product as a white solid (61%). R_f = 0.38 (70:30 hexanes-EtOAc); mp = 49-52°C; IR (film) cm^{-1} 3424, 3310, 3075, 2976, 1720, 1697, 1641, 1607, 1500, 1451, 1389, 1368, 1218, 1167, 1080, 1020, 908; 1H NMR (500 MHz, Acetone- d_6) δ 7.87 (s, 1H), 6.05 (d, J = 8.4 Hz, 1H), 5.82 (ddt, J = 17.0, 10.2, 6.7 Hz, 1H), 5.00 (dq, J = 17.1, 1.8 Hz, 1H), 4.91 (ddd, J = 10.2, 2.2, 1.2 Hz, 1H), 4.27 – 4.22 (m, 2H), 4.08 (t, J = 7.1 Hz, 1H), 4.00 (h, J = 6.3 Hz, 1H), 2.86 – 2.79 (m, 1H), 2.31 (ddt, J = 14.0, 8.7, 2.4 Hz, 1H), 2.22 – 2.08 (m, 3H), 1.94 (t, J = 5.6 Hz, 1H), 1.85 (tt, J = 5.7, 2.9 Hz, 1H), 1.78 (ddd, J = 14.3, 3.3, 2.2 Hz, 1H), 1.74 – 1.65 (m, 1H), 1.65 – 1.57 (m, 1H), 1.42 (s, 9H), 1.34 (s, 3H), 1.27 (s, 3H), 1.19

(d, $J = 6.3$ Hz, 3H), 0.86 (s, 3H); ^{13}C NMR (126 MHz, Acetone- d_6) δ 19.85, 24.36, 26.98, 27.57, 28.52 (3C), 29.29, 31.30, 32.11, 36.74, 38.80, 39.21, 40.63, 52.70, 59.42, 68.79, 77.89, 79.60, 85.24, 114.90, 139.70, 156.47, 173.34; ^{11}B NMR (161 MHz, Acetone- d_6) δ 27.43; HRMS (ESI+) m/z calcd for $[\text{C}_{24}\text{H}_{41}\text{O}_6\text{N}_2\text{B} + \text{H}]^+$ 465.3130, found 465.3142.

***tert*-Butyl ((2*R*)-3-hydroxy-1-oxo-1-(((1*R*)-1-((3*aS*,4*S*,6*S*)-3*a*,5,5-trimethylhexahydro-4,6-methanobenzo[*d*][1,3,2]dioxaborol-2-yl)pent-4-en-1-yl)amino)propan-2-yl)carbamate (22d).**

The product was synthesized following general coupling procedure C, using Boc-D-Ser as the amino acid and 20 as the amine. The crude product was purified by silica gel (eluent 60:40 EtOAc-hexanes) to give the product as a white solid (57%). $R_f = 0.31$ (70:30 hexanes-EtOAc); mp = 47–51°C; IR (film) cm^{-1} 3412, 3302, 3075, 2929, 1701, 1657, 1607, 1452, 1391, 1368, 1219, 1167, 1054, 910; ^1H NMR (500 MHz, Acetone- d_6) δ 8.01 – 7.20 (m, 1H), 6.05 (d, $J = 7.2$ Hz, 1H), 5.82 (ddt, $J = 16.9, 10.2, 6.7$ Hz, 1H), 5.00 (dq, $J = 17.1, 1.7$ Hz, 1H), 4.91 (ddt, $J = 10.2, 2.3, 1.2$ Hz, 1H), 4.25 (dd, $J = 8.7, 2.2$ Hz, 1H), 4.22 – 4.19 (m, 1H), 4.14 (s, 1H), 3.82 (dd, $J = 11.3, 4.8$ Hz, 1H), 3.73 (dd, $J = 10.8, 5.6$ Hz, 1H), 2.86 (ddt, $J = 8.5, 6.3, 3.1$ Hz, 1H), 2.31 (ddt, $J = 13.9, 8.8, 2.5$ Hz, 1H), 2.21 – 2.08 (m, 3H), 1.95 (t, $J = 5.6$ Hz, 1H), 1.85 (tt, $J = 5.8, 2.9$ Hz, 1H), 1.81 – 1.74 (m, 1H), 1.69 (ddt, $J = 13.1, 9.2, 6.5$ Hz, 1H), 1.66 – 1.57 (m, 1H), 1.42 (s, 10H), 1.34 (s, 3H), 1.27 (s, 3H), 0.86 (s, 3H); ^{13}C NMR (126 MHz, Acetone- d_6) δ 24.35, 26.98, 27.56, 28.52 (3C), 29.27, 31.38, 32.05, 36.71, 38.81, 39.34, 40.61, 52.68, 56.00, 63.23, 77.87, 79.63, 85.28, 114.89, 139.73, 156.29, 173.32; ^{11}B NMR (161 MHz, Acetone- d_6) δ 27.28; HRMS (ESI+) m/z calcd for $[\text{C}_{23}\text{H}_{39}\text{O}_6\text{N}_2\text{B} + \text{H}]^+$ 451.2974, found 451.2985

Benzyl (*S*)-(1-oxo-1-(pyrrolidin-1-yl)propan-2-yl)carbamate (23a). The product was synthesized following general coupling procedure B, using *Z*-L-Ala as the amino acid and pyrrolidine as the amine, to give a white solid (49%). $R_f = 0.49$ (50:50 Hexanes-EtOAc); ^1H NMR (500 MHz, Acetone- d_6) δ 7.44 – 7.27 (m, 5H), 6.34 (d, $J = 7.9$ Hz, 1H), 5.08 (s, 2H), 4.43 (p, $J = 7.1$ Hz, 1H), 3.65 (dt, $J = 9.9, 6.7$ Hz, 1H), 3.51 (dt, $J = 9.9, 6.9$ Hz, 1H), 3.42 (dt, $J = 11.8, 7.1$ Hz, 1H), 3.33 (dt, $J = 11.7, 6.8$ Hz, 1H), 1.98 (p, $J = 6.8$ Hz, 2H), 1.85 (p, $J = 6.9$ Hz, 2H), 1.28 (d, $J = 6.9$ Hz, 3H); ^{13}C NMR (126 MHz, Acetone- d_6) δ 18.29, 24.71, 26.76, 46.51, 46.71, 49.16, 66.57, 128.61 (3C), 129.19 (2C), 138.31, 156.33, 171.19. Spectral and physical data were in accordance with the literature.⁶⁰

benzyl ((S)-1-((S)-2-cyanopyrrolidin-1-yl)-1-oxopropan-2-yl)carbamate (23b). The product was synthesized following general coupling procedure B, using Z-L-Ala as the amino acid and prolinonitrile PTSA salt as the amine, to give the product as a clear oil (59%). $R_f = 0.50$ (50:50 Hexanes-EtOAc); $^1\text{H NMR}$ (500 MHz, Chloroform- d) δ 7.39 – 7.28 (m, 5H), 5.56 (d, $J = 8.2$ Hz, 1H), 5.15 – 5.02 (m, 2H), 4.82 – 4.72 (m, 1H), 4.48 (dt, $J = 13.8, 7.2$ Hz, 1H), 3.74 – 3.59 (m, 2H), 2.36 – 2.08 (m, 4H), 1.39 (d, $J = 6.9$ Hz, 3H). Spectral and physical data were previously published by our group.¹⁹

Benzyl ((S)-1-oxo-1-((R)-2-((3aS,4S,6S,7aR)-3a,5,5-trimethylhexahydro-4,6-methanobenzo[d][1,3,2]dioxaborol-2-yl)pyrrolidin-1-yl)propan-2-yl)carbamate (23c). The product was synthesized following general coupling procedure C, using Z-L-Ala as the amino acid and 26 as the amine, to give a white foam (85%). $R_f = 0.42$ (50:50 Hexanes-EtOAc); IR (film) cm^{-1} 3302, 3036, 2921, 1719, 1625, 1498, 1453, 1387, 1374, 1240, 1054, 1029, 741; $^1\text{H NMR}$ (500 MHz, Acetone- d_6) δ 8.08 – 7.10 (m, 10H), 6.62 – 6.30 (m, 1H), 6.29 – 5.90 (m, 1H), 5.15 – 4.97 (m, 4H), 4.51 – 4.35 (m, 2H), 4.26 (ddd, $J = 8.9, 7.2, 2.3$ Hz, 2H), 4.00 – 3.68 (m, 1H), 3.66 – 3.28 (m, 3H), 3.04 (dd, $J = 10.1, 6.8$ Hz, 1H), 2.96 (dd, $J = 9.8, 7.0$ Hz, 1H), 2.43 – 2.24 (m, 2H), 2.11 (ddtt, $J = 10.8, 6.3, 4.6, 2.1$ Hz, 4H), 2.03 – 1.91 (m, 6H), 1.90 – 1.75 (m, 4H), 1.74 – 1.65 (m, 2H), 1.64 – 1.45 (m, 2H), 1.44 – 1.36 (m, 2H), 1.35 (s, 3H), 1.27 (d, $J = 5.0$ Hz, 10H), 1.26 – 1.20 (m, 3H), 0.85 (d, $J = 5.3$ Hz, 6H); $^{13}\text{C NMR}$ (126 MHz, Acetone- d_6) δ 17.99, 18.37, 24.27 (2C), 26.72, 26.79, 27.39, 27.46, 27.48, 27.80, 28.11, 28.18, 28.99, 29.06, 36.26, 36.34, 38.83, 38.87, 40.43 (2C), 45.41 (2C), 46.73, 46.79, 48.50, 48.68, 52.23, 52.38, 66.56, 66.58, 78.26, 78.26, 85.83, 85.98, 128.56 (2C), 128.58 (2C), 128.60 (2C), 129.17 (4C), 138.28 (2C), 156.20, 156.32, 171.04, 171.17; $^{11}\text{B NMR}$ (161 MHz, Acetone- d_6) δ 31.39; HRMS (ESI+) m/z calcd for $[\text{C}_{25}\text{H}_{35}\text{O}_5\text{N}_2\text{B} + \text{H}]^+$ 455.2712, found 455.2711.

tert-Butyl ((S)-1-oxo-1-((R)-2-((3aS,4S,6S,7aR)-3a,5,5-trimethylhexahydro-4,6-methanobenzo[d][1,3,2]dioxaborol-2-yl)pyrrolidin-1-yl)propan-2-yl)carbamate (23d). The product was synthesized following general coupling procedure C, using Boc-L-Ala as the amino acid and 26 as the amine, to give a white foam (64%). $R_f = 0.41$ (60:40 Hexanes-EtOAc); IR (film) cm^{-1} 3321, 2924, 1710, 1627, 1451, 1389, 1366, 1242, 1167, 1054, 1029; $^1\text{H NMR}$ (500 MHz, Acetone- d_6) δ 5.87 (d, $J = 7.8$ Hz, 1H), 5.78 (d, $J = 8.2$ Hz, 1H), 4.34 (p, $J = 7.0$ Hz, 2H), 4.27 (ddd, $J = 11.3, 8.9, 2.3$ Hz, 2H), 3.73 – 3.67 (m, 1H), 3.65 – 3.52 (m, 2H), 3.49 – 3.39 (m, 1H),

3.03 (dd, $J = 10.1, 6.7$ Hz, 1H), 2.94 (dd, $J = 9.8, 7.0$ Hz, 1H), 2.42 – 2.25 (m, 2H), 2.17 – 2.07 (m, 4H), 2.03 – 1.91 (m, 6H), 1.89 – 1.82 (m, 2H), 1.82 – 1.76 (m, 1H), 1.76 – 1.57 (m, 3H), 1.52 – 1.44 (m, 2H), 1.42 – 1.38 (m, 18H), 1.35 (s, 3H), 1.30 (s, 3H), 1.27 (s, 6H), 1.25 (d, $J = 4.4$ Hz, 3H), 1.22 (d, $J = 4.2$ Hz, 3H), 0.85 (s, 6H); ^{13}C NMR (126 MHz, Acetone- d_6) δ 18.12, 18.35, 24.26 (2C), 26.71, 26.78, 27.38, 27.46, 27.48, 27.77, 28.11, 28.19, 28.56 (3C), 28.59 (3C), 28.98, 29.12, 36.25, 36.38, 38.83, 38.86, 40.42, 40.45, 45.35 (2C), 46.67, 46.76, 47.96, 48.12, 52.23, 52.41, 78.21, 78.23, 79.00, 79.01, 85.80, 85.97, 155.70, 155.72, 171.33, 171.56; ^{11}B NMR (161 MHz, Acetone- d_6) δ 31.17; HRMS (ESI+) m/z calcd for $[\text{C}_{22}\text{H}_{37}\text{O}_5\text{N}_2\text{B} + \text{H}]^+$ 421.2868, found 421.2864.

tert-Butyl ((R)-1-oxo-1-((R)-2-((3aS,4S,6S,7aR)-3a,5,5-trimethylhexahydro-4,6-methanobenzo[d][1,3,2]dioxaborol-2-yl)pyrrolidin-1-yl)propan-2-yl)carbamate (24a). The product was synthesized following general coupling procedure C, using Boc-D-Ala as the amino acid and 26 as the amine, to give a mixture of diastereomers as a white foam (40%). $R_f = 0.39$ (60:40 Hexanes-EtOAc); IR (film) cm^{-1} 3317, 2928, 1713, 1629, 1451, 1389, 1366, 1243, 1165, 1031; ^1H NMR (500 MHz, Acetone- d_6) δ 5.86 (d, $J = 7.8$ Hz, 1H), 5.77 (d, $J = 7.8$ Hz, 1H), 4.35 (td, $J = 7.3, 3.1$ Hz, 2H), 4.29 (dd, $J = 8.9, 2.3$ Hz, 1H), 4.24 (dd, $J = 8.8, 2.2$ Hz, 1H), 3.76 – 3.67 (m, 1H), 3.65 – 3.51 (m, 2H), 3.44 (td, $J = 9.4, 6.7$ Hz, 1H), 3.03 (dd, $J = 10.3, 6.8$ Hz, 1H), 2.97 – 2.89 (m, 1H), 2.39 – 2.26 (m, 2H), 2.20 – 2.07 (m, 4H), 2.03 – 1.90 (m, 6H), 1.85 (tq, $J = 5.8, 2.9$ Hz, 2H), 1.82 – 1.64 (m, 4H), 1.43 – 1.41 (m, 10H), 1.41 – 1.39 (m, 10H), 1.36 (s, 3H), 1.30 (s, 3H), 1.28 (s, 3H), 1.27 (s, 3H), 1.26 – 1.23 (m, 3H), 1.22 (s, 3H), 0.85 (s, 3H), 0.85 (s, 3H).; ^{13}C NMR (126 MHz, Acetone- d_6) δ 18.25, 18.60, 24.26, 24.28, 26.80 (2C), 27.46, 27.49, 27.55, 27.70, 28.18 (2C), 28.58 (3C), 28.59 (3C), 28.94, 29.06, 36.35 (2C), 38.85 (2C), 40.40, 40.44, 45.38 (2C), 46.65, 46.83, 47.99, 48.16, 52.33, 52.37, 78.05, 78.33, 78.96, 79.06, 85.84, 86.00, 155.54, 155.68, 171.18, 171.43; ^{11}B NMR (161 MHz, Acetone- d_6) δ 31.25; HRMS (ESI+) m/z calcd for $[\text{C}_{22}\text{H}_{37}\text{O}_5\text{N}_2\text{B} + \text{H}]^+$ 421.2868, found 421.2869.

Benzyl ((2S)-3-methyl-1-oxo-1-(2-((3aS,4S,6S,7aR)-3a,5,5-trimethylhexahydro-4,6-methanobenzo[d][1,3,2]dioxaborol-2-yl)pyrrolidin-1-yl)butan-2-yl)carbamate (25a). The product was synthesized following general coupling procedure C, using Z-L-Val as the amino acid and 26 as the amine, to give a white foam (60%). $R_f = 0.31$ (65:45 Hexanes-EtOAc); IR (film) cm^{-1} 3250, 3067, 2968, 1715, 1619, 1502, 1451, 1389, 1376, 1368, 1217, 1078, 1028; ^1H NMR (500 MHz, Acetone- d_6) δ 7.91 – 6.90 (m, 10H), 6.26 (d, $J = 9.0$ Hz, 1H), 6.15 (d, $J = 9.2$ Hz, 1H), 5.18

– 4.96 (m, 4H), 4.44 – 4.12 (m, 4H), 4.01 – 3.78 (m, 1H), 3.73 – 3.40 (m, 3H), 3.32 – 2.93 (m, 2H), 2.41 – 2.20 (m, 2H), 2.15 – 2.07 (m, 3H), 2.03 – 1.90 (m, 6H), 1.90 – 1.66 (m, 6H), 1.63 – 1.45 (m, 2H), 1.35 – 1.33 (m, 3H), 1.27 (s, 12H), 1.00 – 0.92 (m, 12H), 0.86 – 0.82 (m, 6H); ¹³C NMR (126 MHz, Acetone-*d*₆) δ 18.30 (2C), 19.42, 19.66, 24.28 (2C), 26.82 (2C), 27.30, 27.47, 27.50, 27.92, 28.17 (2C), 29.05, 29.09, 31.58, 31.60, 36.24, 36.42, 38.82, 38.86, 40.41, 40.47, 45.28 (2C), 47.03, 47.20, 52.22, 52.42, 58.12, 58.30, 66.63, 66.69, 78.20, 78.26, 85.67, 86.08, 128.42 (2C), 128.56 (4C), 129.17 (4C), 138.29, 138.35, 157.07, 157.10, 170.48, 170.79; ¹¹B NMR (161 MHz, Acetone-*d*₆) δ 30.84; HRMS (ESI+) *m/z* calcd for [C₂₇H₃₉O₅N₂B + H]⁺ 483.3025, found 483.3027.

***tert*-Butyl ((*S*)-3-methyl-1-oxo-1-((*R*)-2-((3*aS*,4*S*,6*S*,7*aR*)-3*a*,5,5-trimethylhexahydro-4,6-methanobenzo[*d*][1,3,2]dioxaborol-2-yl)pyrrolidin-1-yl)butan-2-yl)carbamate (25b).** The product was synthesized following general coupling procedure C, using Boc-L-Val as the amino acid and 26 as the amine, to give a white foam (65%). *R*_f = 0.36 (70:30 Hexanes-EtOAc; IR (film) cm⁻¹ 3321, 2968, 1715, 1619, 1449, 1389, 1366, 1169, 1076, 1032; ¹H NMR (500 MHz, Acetone-*d*₆) δ 5.72 (d, *J* = 8.8 Hz, 1H), 5.64 (d, *J* = 9.2 Hz, 1H), 4.32 – 4.22 (m, 2H), 4.20 – 3.91 (m, 2H), 3.77 (ddd, *J* = 10.6, 8.5, 2.8 Hz, 1H), 3.67 (dt, *J* = 10.0, 7.6 Hz, 1H), 3.59 (ddd, *J* = 10.2, 8.1, 4.2 Hz, 1H), 3.48 (td, *J* = 9.7, 6.8 Hz, 1H), 3.01 (dd, *J* = 10.7, 6.8 Hz, 1H), 2.96 (dd, *J* = 9.5, 7.1 Hz, 1H), 2.38 – 2.26 (m, 2H), 2.18 – 2.07 (m, 4H), 2.03 – 1.90 (m, 8H), 1.88 – 1.82 (m, 2H), 1.82 – 1.58 (m, 4H), 1.48 (dd, *J* = 15.0, 10.7 Hz, 2H), 1.40 (s, 18H), 1.34 (s, 3H), 1.30 (s, 3H), 1.27 (s, 3H), 1.27 (s, 3H), 0.97 – 0.92 (m, 6H), 0.92 – 0.88 (m, 6H), 0.85 (s, 6H). ¹³C NMR (126 MHz, Acetone-*d*₆) δ 17.14, 17.32, 18.66, 18.84, 23.41 (2C), 25.93, 25.95, 26.42 (2C), 26.60, 26.63, 27.02 (2C), 27.31 (2C), 27.68 (3C), 27.72 (3C), 28.18, 28.29, 30.71, 30.80, 35.38, 35.62, 37.96, 37.99, 39.54, 39.62, 44.36 (2C), 46.08, 46.24, 51.36, 51.62, 56.54, 56.61, 77.28, 77.38, 84.77, 85.19, 155.56, 155.62, 169.76, 170.19; ¹¹B NMR (161 MHz, Acetone-*d*₆) δ 30.70; HRMS (ESI+) *m/z* calcd for [C₂₂H₄₁O₅N₂B + H]⁺ 449.3181, found 449.3180.

General acid-catalyzed oxidative cyclization procedure A. The dipeptide (1 eq) was diluted in anhydrous DCM (0.02 M), and 2-3 drops of Sudan III solution (1mg/mL in DCM) were added (enough to reach a pink color). The solution was cooled to –78°C, and N₂ gas was bubbled into the solution for 5 minutes, followed by ozone (~80% ozone output). When the solution turned dark blue, ozone addition was stopped, and N₂ was bubbled until the solution was colorless. Polymer-

bound triphenylphosphine (1.5 eq, ~3mmol/g loading, CAS 39319-11-4) was added, and the mixture stirred for 5 minutes at -78°C , then at room temperature overnight under argon atmosphere, after which the mixture became slightly opaque. TFA (1.5 eq) was added at room temperature, and the mixture was stirred for 2h. The mixture was filtered through Celite®. The solid was rinsed with DCM, and the filtrate was concentrated *in vacuo*. The residue was redissolved in EtOAc and washed with saturated NH_4Cl and brine. The organic layer was dried over Na_2SO_4 , filtered, and concentrated *in vacuo* to give the crude product, which was purified by flash chromatography on a silica gel column to give the pure product.

General acid-catalyzed oxidative cyclization procedure B. The dipeptide (1 eq) was diluted in anhydrous DCM (0.02 M) and cooled to -78°C . N_2 gas was bubbled into the solution for 5 minutes, followed by ozone (~80% output). When the solution turned a deep blue, ozone addition was immediately stopped, and N_2 was bubbled until the solution was colorless. Polymer-bound triphenylphosphine (1.5 eq, ~3mmol/g loading, CAS 39319-11-4) was added, and the mixture stirred for 5 minutes at -78°C , then at room temperature overnight, after which the mixture became slightly opaque. TFA (1.5 eq) was added at room temperature, and the mixture was stirred for 2h. The mixture was filtered through Celite®. The solid was rinsed with DCM, and the filtrate was concentrated *in vacuo*. The residue was redissolved in EtOAc and washed with saturated NH_4Cl and brine. The organic layer was dried over Na_2SO_4 , filtered, and concentrated *in vacuo* to give the crude product, which was purified by flash chromatography on a silica gel column to give the pure product.

Benzyl ((3*S*,6*S*,8*aS*)-6-cyano-4-oxohexahydro-2*H*-pyrrolo[2,1-*b*][1,3]oxazin-3-yl)carbamate (10a). The product was synthesized according to general acid-catalyzed oxidative cyclization procedure B from peptide **18** and purified by flash chromatography on a silica gel column (eluent 80-20 EtOAc-hexanes) to give the product as a white solid (33%). $R_f = 0.56$ (100% EtOAc); mp = $167\text{--}170^{\circ}\text{C}$; IR (film) cm^{-1} 3361, 3031, 2952, 1719, 1685, 1532, 1433, 1257, 1064, 1064, 1018, 698; ^1H NMR (500 MHz, Acetone- d_6) δ 7.59 – 7.23 (m, 5H), 6.71 (d, $J = 8.4$ Hz, 1H), 5.41 (t, $J = 6.5$ Hz, 1H), 5.12 (s, 2H), 4.69 (dd, $J = 5.3, 3.5$ Hz, 1H), 4.62 (q, $J = 8.6$ Hz, 1H), 4.38 (t, $J = 9.6$ Hz, 1H), 3.78 (t, $J = 8.5$ Hz, 1H), 2.52 (ddt, $J = 13.2, 5.6, 3.3$ Hz, 1H), 2.33 (dq, $J = 9.9, 3.0$ Hz, 2H), 2.08 – 1.97 (m, 1H); ^{13}C NMR (126 MHz, Acetone- d_6) δ 28.41, 32.01, 45.59, 50.42,

67.02, 68.85, 88.52, 118.79, 128.64 (2C), 128.69, 129.22 (2C), 138.03, 157.29, 167.37; HRMS (ESI+) m/z calcd for $[C_{16}H_{17}O_4N_3 + Na]^+$ 338.1111, found 338.1102.

Benzyl ((3S,8aS)-4-oxohexahydro-2H-pyrrolo[2,1-b][1,3]oxazin-3-yl)carbamate (10b) The product was synthesized according to general acid-catalyzed oxidative cyclization procedure A from peptide **14** and purified by flash chromatography on a silica gel column (eluent 80:20 EtOAc:hexanes) to give the product as a white solid (40%). R_f = 0.39 (100% EtOAc); mp = 118-122°C; IR (film) cm^{-1} 3305, 3063, 2980, 1717, 1669, 1530, 1443, 1217, 1064, 1018, 695; 1H NMR (500 MHz, Acetone- d_6) δ 7.48 – 7.25 (m, 5H), 6.50 (s, 1H), 5.20 (t, J = 5.3 Hz, 1H), 5.10 (s, 2H), 4.44 (q, J = 7.4 Hz, 1H), 4.28 (dd, J = 10.2, 8.1 Hz, 1H), 3.76 – 3.66 (m, 2H), 3.26 (ddd, J = 11.9, 7.3, 5.1 Hz, 1H), 2.23 (td, J = 10.4, 10.0, 4.9 Hz, 1H), 1.96 – 1.87 (m, 1H), 1.87 – 1.76 (m, 2H); ^{13}C NMR (126 MHz, Acetone- d_6) δ 21.89, 33.03, 45.26, 50.22, 66.92, 68.81, 88.01, 128.65 (3C), 129.19 (2C), 138.06, 157.14, 166.64; HRMS (ESI+) m/z calcd for $[C_{15}H_{18}O_4N_2 + Na]^+$ 313.1159, found 313.1163.

Benzyl ((3S,6R,8aS)-4-oxo-6-((3aS,4S,6S,7aR)-3a,5,5-trimethylhexahydro-4,6-methanobenzo[d][1,3,2]dioxaborol-2-yl)hexahydro-2H-pyrrolo[2,1-b][1,3]oxazin-3-yl)carbamate (10c). The product was synthesized according to general acid-catalyzed oxidative cyclization procedure B from peptide 21a and purified by flash chromatography on a silica gel column (eluent 50:50 hexanes-EtOAc) to give the product as a white foam (58%). R_f = 0.47 (50:50 hexanes-EtOAc); IR (film) cm^{-1} 3325, 3067, 2921, 1720, 1673, 1586, 1451, 1391, 1376, 1219, 1029; 1H NMR (500 MHz, Acetone- d_6) δ 7.44 – 7.26 (m, 5H), 6.25 (d, J = 6.9 Hz, 1H), 5.20 (dd, J = 6.1, 4.0 Hz, 1H), 5.11 (s, 2H), 4.41 (q, J = 6.6 Hz, 1H), 4.29 (dd, J = 8.8, 2.1 Hz, 1H), 4.24 (dd, J = 10.4, 7.4 Hz, 1H), 3.72 (dd, J = 10.5, 6.2 Hz, 1H), 3.05 (t, J = 7.3 Hz, 1H), 2.33 (ddt, J = 14.1, 8.9, 2.5 Hz, 1H), 2.26 – 2.18 (m, 1H), 2.14 (ddt, J = 10.2, 7.7, 3.9 Hz, 1H), 2.01 – 1.93 (m, 3H), 1.91 – 1.83 (m, 2H), 1.78 (ddd, J = 14.4, 3.3, 2.2 Hz, 1H), 1.39 (s, 3H), 1.34 – 1.29 (m, 1H), 1.28 – 1.19 (m, 3H), 0.85 (s, 3H); ^{13}C NMR (126 MHz, Acetone- d_6) δ 24.25, 24.96, 26.82, 27.43, 28.75, 33.71, 36.17, 38.86, 40.38, 43.46, 50.00, 52.21, 66.97, 68.80, 78.36, 86.48, 88.29, 128.66 (3C), 129.20 (2C), 138.04, 157.03, 166.39; ^{11}B NMR (161 MHz, Acetone- d_6) δ 31.58; HRMS (ESI+) m/z calcd for $[C_{25}H_{33}O_6N_2B + Na]^+$ 491.2324, found 491.2331.

***tert*-Butyl ((3S,6R,8aS)-4-oxo-6-((3aS,4S,6S,7aR)-3a,5,5-trimethylhexahydro-4,6-methanobenzo[d][1,3,2]dioxaborol-2-yl)hexahydro-2H-pyrrolo[2,1-b][1,3]oxazin-3-**

yl)carbamate (10e). The product was synthesized according to general acid-catalyzed oxidative cyclization procedure B from peptide 22a and purified by flash chromatography on a silica gel column (eluent 60:40 hexanes-EtOAc) to give the product as a white solid (52%). $R_f = 0.39$ (60:40 hexanes-EtOAc); mp = 144–147°C; IR (film) cm^{-1} 3337, 2924, 1715, 1673, 1449, 1391, 1368, 1165, 1078, 1030; ^1H NMR (500 MHz, Acetone- d_6) δ 5.75 (d, $J = 6.6$ Hz, 1H), 5.18 (t, $J = 5.1$ Hz, 1H), 4.34 – 4.26 (m, 2H), 4.20 (dd, $J = 10.3, 7.2$ Hz, 1H), 3.67 (dd, $J = 10.3, 6.2$ Hz, 1H), 3.06 (t, $J = 7.2$ Hz, 1H), 2.35 (ddt, $J = 14.1, 8.8, 2.5$ Hz, 1H), 2.26 – 2.18 (m, 1H), 2.18 – 2.13 (m, 1H), 2.02 – 1.92 (m, 3H), 1.92 – 1.85 (m, 2H), 1.79 (ddd, $J = 14.4, 3.3, 2.2$ Hz, 1H), 1.43 (s, 9H), 1.40 (s, 3H), 1.34 – 1.31 (m, 1H), 1.29 (s, 3H), 0.87 (s, 3H); ^{13}C NMR (126 MHz, Acetone- d_6) δ 24.25, 24.95, 26.82, 27.45, 28.49 (3C), 28.76, 33.73, 36.18, 38.88, 40.39, 43.62, 49.63, 52.21, 69.00, 78.39, 79.61, 86.48, 88.20, 156.28, 166.63; ^{11}B NMR (161 MHz, Acetone- d_6) δ 31.59; HRMS (ESI+) m/z calcd for $[\text{C}_{22}\text{H}_{35}\text{O}_6\text{N}_2\text{B} + \text{H}]^+$ 435.2661, found 435.2667.

tert-Butyl ((3R,6R,8aR)-4-oxo-6-((3aS,4S,6S,7aR)-3a,5,5-trimethylhexahydro-4,6-methanobenzo[d][1,3,2]dioxaborol-2-yl)hexahydro-2H-pyrrolo[2,1-b][1,3]oxazin-3-yl)carbamate (11a). The product was synthesized according to general acid-catalyzed oxidative cyclization procedure B from peptide 22d and purified by flash chromatography on a silica gel column (eluent 70:30 hexanes-EtOAc) to give the product as a white foam (53%). $R_f = 0.31$ (70:30 hexanes-EtOAc); IR (film) cm^{-1} 3349, 2925, 1709, 1671, 1449, 1389, 1378, 1370, 1215, 1163, 1076; ^1H NMR (500 MHz, Acetone- d_6) δ 6.02 (d, $J = 6.5$ Hz, 1H), 5.22 (t, $J = 6.1$ Hz, 1H), 4.33 (dt, $J = 8.8, 1.6$ Hz, 2H), 4.26 (dd, $J = 11.5, 7.2$ Hz, 1H), 3.67 (dd, $J = 10.3, 6.1$ Hz, 1H), 3.36 (dd, $J = 10.3, 6.9$ Hz, 1H), 2.41 – 2.27 (m, 2H), 2.21 – 2.11 (m, 1H), 2.09 (td, $J = 6.1, 5.4, 2.8$ Hz, 1H), 1.98 (t, $J = 5.7$ Hz, 1H), 1.95 – 1.81 (m, 2H), 1.83 – 1.76 (m, 2H), 1.78 – 1.69 (m, 1H), 1.42 (s, 9H), 1.36 (s, 3H), 1.28 (s, 3H), 0.86 (s, 3H); ^{13}C NMR (126 MHz, Acetone- d_6) δ 166.41, 156.43, 88.87, 86.78, 79.50, 78.62, 69.94, 52.08, 49.77, 43.44, 40.26, 38.86, 36.09, 34.28, 28.93, 28.50 (3C), 27.38, 26.83, 25.11, 24.19; ^{11}B NMR (161 MHz, Acetone- d_6) δ 31.87; HRMS (ESI+) m/z calcd for $[\text{C}_{22}\text{H}_{35}\text{O}_6\text{N}_2\text{B} + \text{Na}]^+$ 457.2480, found 457.2468.

Benzyl ((2R,3S,6R,8aS)-2-methyl-4-oxo-6-((3aS,4S,6S,7aR)-3a,5,5-trimethylhexahydro-4,6-methanobenzo[d][1,3,2]dioxaborol-2-yl)hexahydro-2H-pyrrolo[2,1-b][1,3]oxazin-3-yl)carbamate (12a). The product was synthesized according to general acid-catalyzed oxidative cyclization procedure B from peptide 21b and purified by flash chromatography on a silica gel

column (eluent 50:50 hexanes-EtOAc) to give the product as a clear oil (59%). $R_f = 0.58$ (50:50 hexanes-EtOAc); IR (film) cm^{-1} 3409, 1720, 1675, 1504, 1454, 1391, 1376, 1217, 1056, 1028; ^1H NMR (500 MHz, Acetone- d_6) δ 7.46 – 7.30 (m, 5H), 6.06 (d, $J = 8.0$ Hz, 1H), 5.23 (t, $J = 5.1$ Hz, 1H), 5.12 (s, 2H), 4.43 (dd, $J = 7.9, 5.5$ Hz, 1H), 4.34 (p, $J = 6.2$ Hz, 1H), 4.29 (dd, $J = 8.9, 2.2$ Hz, 1H), 3.08 (dd, $J = 8.2, 6.0$ Hz, 1H), 2.32 (ddt, $J = 14.1, 8.9, 2.5$ Hz, 1H), 2.25 – 2.15 (m, 1H), 2.11 (dtd, $J = 10.9, 6.4, 2.3$ Hz, 1H), 2.02 – 1.83 (m, 5H), 1.78 (ddd, $J = 14.4, 3.3, 2.2$ Hz, 1H), 1.37 (s, 3H), 1.36 – 1.31 (m, 1H), 1.24 (s, 3H), 1.12 (d, $J = 6.3$ Hz, 3H), 0.85 (s, 3H); ^{13}C NMR (126 MHz, Acetone- d_6) δ 17.05, 24.26, 24.57, 26.80, 27.43, 28.80, 33.52, 36.15, 38.87, 40.39, 43.07, 52.18, 53.55, 67.00, 73.39, 78.46, 86.50, 87.46, 128.59 (2C), 128.67, 129.22 (2C), 138.12, 157.11, 166.11; ^{11}B NMR (161 MHz, Acetone- d_6) δ 31.95; HRMS (ESI+) m/z calcd for $[\text{C}_{26}\text{H}_{35}\text{O}_6\text{N}_2\text{B} + \text{Na}]^+$ 505.2480, found 505.2489.

***tert*-Butyl ((2*R*,3*S*,6*R*,8*aS*)-2-methyl-4-oxo-6-((3*aS*,4*S*,6*S*,7*aR*)-3*a*,5,5-trimethylhexahydro-4,6-methanobenzo[*d*][1,3,2]dioxaborol-2-yl)hexahydro-2*H*-pyrrolo[2,1-*b*][1,3]oxazin-3-yl)carbamate (12b).** The product was synthesized according to general acid-catalyzed oxidative cyclization procedure B from peptide **22b** and purified by flash chromatography on a silica gel column (eluent 60:40 hexanes-EtOAc) to give the product as a white solid (56%). $R_f = 0.34$ (60:40 hexanes-EtOAc); mp = 131-134°C; IR (film) cm^{-1} 3333, 1717, 1675, 1453, 1391, 1368, 1165, 1058, 1030; ^1H NMR (400 MHz, Acetone- d_6) δ 5.57 (d, $J = 7.0$ Hz, 1H), 5.23 (dd, $J = 5.5, 4.4$ Hz, 1H), 4.39 – 4.26 (m, 3H), 3.09 (dd, $J = 8.3, 5.7$ Hz, 1H), 2.35 (ddt, $J = 14.0, 8.8, 2.5$ Hz, 1H), 2.26 – 2.11 (m, 2H), 2.03 – 1.96 (m, 2H), 1.95 – 1.89 (m, 2H), 1.89 – 1.84 (m, 1H), 1.80 (ddd, $J = 14.3, 3.3, 2.2$ Hz, 1H), 1.44 (s, 9H), 1.39 (s, 3H), 1.37 – 1.32 (m, 1H), 1.29 (s, 3H), 1.09 (d, $J = 6.1$ Hz, 3H), 0.87 (s, 3H); ^{13}C NMR (101 MHz, Acetone- d_6) δ 165.37, 155.28, 86.26, 85.53, 78.50, 77.54, 72.36, 52.06, 51.18, 42.17, 39.41, 37.92, 35.15, 32.59, 27.83, 27.51 (3C), 26.48, 25.80, 23.66, 23.26, 16.02; ^{11}B peak reported for major rotamer; full spectrum provided as Supporting Information. ^{11}B NMR (161 MHz, Acetone- d_6) δ 31.80; HRMS (ESI+) m/z calcd for $[\text{C}_{23}\text{H}_{37}\text{O}_6\text{N}_2\text{B} + \text{H}]^+$ 449.2817, found 449.2814.

***tert*-Butyl ((2*S*,3*S*,6*R*,8*aS*)-2-methyl-4-oxo-6-((3*aS*,4*S*,6*S*,7*aR*)-3*a*,5,5-trimethylhexahydro-4,6-methanobenzo[*d*][1,3,2]dioxaborol-2-yl)hexahydro-2*H*-pyrrolo[2,1-*b*][1,3]oxazin-3-yl)carbamate (13a).** The product was synthesized according to general acid-catalyzed oxidative cyclization procedure B from peptide **22c** and purified by flash

chromatography on a silica gel column (eluent 60:40 hexanes-EtOAc) to give the product as a white foam (60%). R_f = 0.43 (70:30 hexanes-EtOAc); IR (film) cm^{-1} 3329, 2929, 1717, 1671, 1449, 1389, 1376, 1368, 1165, 1054, 1031; ^1H NMR (500 MHz, Acetone- d_6) δ 5.89 (d, J = 8.6 Hz, 1H), 5.40 (dd, J = 6.3, 2.3 Hz, 1H), 4.29 (dd, J = 8.8, 2.1 Hz, 1H), 4.19 (t, J = 9.0 Hz, 1H), 3.74 – 3.56 (m, 1H), 2.99 (t, J = 8.1 Hz, 1H), 2.34 (ddt, J = 14.2, 8.8, 2.5 Hz, 1H), 2.25 – 2.12 (m, 2H), 2.01 – 1.96 (m, 2H), 1.96 – 1.90 (m, 2H), 1.90 – 1.85 (m, 1H), 1.78 (ddd, J = 14.5, 3.4, 2.1 Hz, 1H), 1.43 (s, 9H), 1.41 (s, 3H), 1.32 (d, J = 6.2 Hz, 3H), 1.28 (d, J = 3.0 Hz, 4H), 0.87 (s, 3H); ^{13}C NMR (126 MHz, Acetone- d_6) δ 167.88, 156.81, 86.47, 84.11, 79.51, 78.29, 73.80, 55.75, 52.25, 44.12, 40.40, 38.84, 36.15, 34.36, 28.67, 28.49 (3C), 27.44, 26.86, 26.35, 24.25, 19.50; ^{11}B NMR (161 MHz, Acetone- d_6) δ 31.74; HRMS (ESI+) m/z calcd for $[\text{C}_{23}\text{H}_{37}\text{O}_6\text{N}_2\text{B} + \text{H}]^+$ 449.2817, found 449.2811.

***N*-((3*S*,6*R*,8*aS*)-4-oxo-6-((3*aS*,4*S*,6*S*,7*aR*)-3*a*,5,5-Trimethylhexahydro-4,6-methanobenzo[*d*][1,3,2]dioxaborol-2-yl)hexahydro-2*H*-pyrrolo[2,1-*b*][1,3]oxazin-3-yl)acetamide (10d).** Bicycle **10c** (175 mg) was dissolved in EtOAc (15 mL), and AcOH was added (~5 drops). The solution was purged with Ar (bubbled into the solution) for 15 minutes. The Pd/C catalyst was then added, and the mixture was purged with Ar (bubbled into the solution) for 5 minutes. The Ar balloon was replaced with an H₂ balloon, and the mixture stirred overnight. The solid was then filtered through Celite® and rinsed with EtOAc. The filtrate was concentrated *in vacuo* to give a yellow oil, which was dissolved in DCM (10 mL), and the solution was cooled to 0°C. Et₃N (113 mg, 0.16 mL, 3 eq) was added, followed by acetyl chloride (35 mg, 0.031 mL, 1.2 eq) and DMAP (5 mg, 0.1 eq). The solution was warmed to room temperature and stirred for 2h. Water was added, and the product was extracted with DCM. The combined organic layers were washed with brine, dried over Na₂SO₄, and concentrated *in vacuo*. The crude product was purified by flash chromatography on a silica gel column (eluent 100% EtOAc) to give the product as a white solid (89 mg, 63% over 2 steps). R_f = 0.17 (eluent 100% EtOAc); ^1H NMR (500 MHz, Acetone- d_6) δ 7.16 (s, 1H), 5.18 (dd, J = 6.0, 4.0 Hz, 1H), 4.60 (dtd, J = 7.3, 6.4, 0.9 Hz, 1H), 4.29 (dd, J = 8.8, 2.2 Hz, 1H), 4.22 (dd, J = 10.4, 7.5 Hz, 1H), 3.57 (dd, J = 10.4, 6.4 Hz, 1H), 3.04 (t, J = 7.2 Hz, 1H), 2.35 (ddt, J = 14.2, 8.8, 2.5 Hz, 1H), 2.27 – 2.19 (m, 1H), 2.16 (dtd, J = 10.7, 6.1, 2.2 Hz, 1H), 2.01 – 1.96 (m, 2H), 1.96 – 1.92 (m, 4H), 1.92 – 1.85 (m, 2H), 1.78 (ddd, J = 14.4, 3.3, 2.2 Hz, 1H), 1.40 (s, 3H), 1.35 – 1.29 (m, 1H), 1.28 (s, 3H), 0.87 (s, 3H); ^{13}C NMR (126 MHz, Acetone- d_6) δ 170.20, 166.57, 88.20, 86.43, 78.34, 68.98, 52.24, 48.43, 43.39, 40.39, 38.86, 36.21,

33.74, 28.76, 27.43, 26.84, 25.07, 24.25, 22.65; ¹¹B peak reported for major rotamer; full spectrum provided as Supporting Information. ¹¹B NMR (161 MHz, Acetone-*d*₆) δ 31.51, 22.62; HRMS (ESI+) *m/z* calcd for [C₁₉H₂₉O₅N₂B + H]⁺ 377.2242, found 377.2244.

General procedure for the deprotection of *N*-Boc protected boronic esters. The boronic ester (1 eq) was dissolved in DCM (0.1 M), and the solution was cooled to –78°C. BCl₃ (1M in DCM, 3.5 eq) was added dropwise, and the solution was stirred at –78°C for 1h. MeOH (12 eq) was added *slowly* at –78°C. The solution was concentrated *in vacuo*, and the residue was dissolved in DCM. The product was extracted with water, and the combined aqueous phases were washed with Et₂O and concentrated *in vacuo* to give the product as a solid with no further purification necessary.

(3*S*,6*R*,8*aS*)-6-Borono-4-oxohexahydro-2*H*-pyrrolo[2,1-*b*][1,3]oxazin-3-aminium chloride (10*f*). The product was synthesized from bicycle **10e** according to the general procedure for the deprotection of *N*-Boc protected boronic esters, giving a sticky white solid (34%). *R_f* = does not elute on silica-backed TLC plates; IR (film) cm⁻¹ 3206, 3194, 2845, 1647, 1477, 1022; ¹H NMR (500 MHz, DMSO-*d*₆) δ 8.68 (s, 3H), 5.02 (dd, *J* = 6.9, 5.5 Hz, 1H), 4.14 (dd, *J* = 12.1, 6.2 Hz, 1H), 4.08 (dd, *J* = 12.1, 3.5 Hz, 1H), 3.91 – 3.85 (m, 1H), 2.98 (dd, *J* = 9.4, 4.1 Hz, 1H), 2.14 (dtd, *J* = 10.7, 5.6, 3.3 Hz, 1H), 1.93 – 1.80 (m, 2H), 1.79 – 1.73 (m, 1H); ¹³C NMR (126 MHz, DMSO) δ 22.65, 31.19, 45.20, 46.71, 66.09, 88.62, 161.45; ¹¹B NMR (161 MHz, DMSO) δ 20.23; HRMS (ESI+) *m/z* calcd for [C₇H₁₃O₄N₂B + H]⁺ 201.1041, found 201.1039.

(3*R*,6*R*,8*aR*)-6-Borono-4-oxohexahydro-2*H*-pyrrolo[2,1-*b*][1,3]oxazin-3-aminium chloride (11*b*). The product was synthesized from bicycle **11a** according to the general procedure for the deprotection of *N*-Boc protected boronic esters, giving a sticky white solid (39%). *R_f* = does not elute on silica-backed TLC plates; IR (film) cm⁻¹ 3206, 2980, 1653, 1447, 1191, 1056; ¹H NMR (500 MHz, DMSO-*d*₆) δ 8.60 (s, 3H), 5.05 (dd, *J* = 7.9, 5.4 Hz, 1H), 4.19 (dd, *J* = 12.1, 7.1 Hz, 1H), 4.05 (dd, *J* = 12.1, 3.4 Hz, 1H), 3.97 – 3.86 (m, 1H), 3.21 (dd, *J* = 10.9, 7.5 Hz, 1H), 2.21 (dtd, *J* = 11.3, 5.8, 1.6 Hz, 1H), 2.00 (dtd, *J* = 12.3, 7.1, 1.5 Hz, 1H), 1.66 (tt, *J* = 11.5, 7.3 Hz, 1H), 1.56 (qd, *J* = 11.8, 5.9 Hz, 1H); ¹³C NMR (126 MHz, DMSO) δ 23.05, 32.39, 46.12, 46.81, 66.38, 89.33, 161.95; ¹¹B NMR (161 MHz, DMSO) δ 20.15; HRMS (ESI+) *m/z* calcd for [C₇H₁₃O₄N₂B + H]⁺ 201.1041, found 201.1041.

(2*R*,3*S*,6*R*,8*aS*)-6-Borono-2-methyl-4-oxohexahydro-2*H*-pyrrolo[2,1-*b*][1,3]oxazin-3-aminium chloride (12c). The product was synthesized from bicycle **12b** according to the general procedure for the deprotection of *N*-Boc protected boronic esters, giving a sticky white solid (41%). *R_f* = does not elute on silica-backed TLC plates; IR (film) cm⁻¹ 3194, 2892, 1651, 1445, 1193, 1046; ¹H NMR (500 MHz, DMSO-*d*₆) δ 8.54 (d, *J* = 5.4 Hz, 3H), 5.12 (t, *J* = 5.9 Hz, 1H), 4.34 (p, *J* = 6.4 Hz, 1H), 3.98 (p, *J* = 5.6 Hz, 1H), 2.97 (dd, *J* = 9.1, 4.5 Hz, 1H), 2.18 – 2.03 (m, 1H), 1.93 – 1.73 (m, 3H), 1.21 (d, *J* = 6.5 Hz, 3H); ¹³C NMR (126 MHz, DMSO) δ 16.54, 23.13, 31.42, 45.49, 50.31, 70.67, 87.24, 161.95; ¹¹B NMR (161 MHz, DMSO) δ 19.64; HRMS (ESI+) *m/z* calcd for [C₈H₁₅O₄N₂B + H]⁺ 215.1198, found 215.1192

(2*S*,3*S*,6*R*,8*aS*)-6-Borono-2-methyl-4-oxohexahydro-2*H*-pyrrolo[2,1-*b*][1,3]oxazin-3-aminium chloride (13b). The product was synthesized from bicycle **13a** according to the general procedure for the deprotection of *N*-Boc protected boronic esters, giving a sticky white solid (52%). *R_f* = does not elute on silica-backed TLC plates; IR (film) cm⁻¹ 3198, 2948, 1653, 1477, 1193, 1138; ¹H NMR (500 MHz, DMSO-*d*₆) δ 8.63 (d, *J* = 4.7 Hz, 3H), 5.40 – 5.19 (m, 1H), 4.28 – 3.79 (m, 2H), 3.21 – 2.87 (m, 1H), 2.24 – 1.98 (m, 1H), 1.98 (s, 3H), 1.39 (d, *J* = 6.3 Hz, 3H); ¹³C NMR (126 MHz, DMSO) δ 18.85, 24.70, 31.99, 46.74, 52.40, 70.24, 83.35, 162.86; ¹¹B NMR (161 MHz, DMSO) δ 20.09; HRMS (ESI+) [C₈H₁₅O₄N₂B + H]⁺ 215.1198, found 215.1190

Supporting Information.

NMR spectra, dose-response curves for each compound (PDF)

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Abbreviations: POP, prolyl oligopeptidase; FAP, fibroblast activation protein- α ; DPP, dipeptidyl peptidase; TLC, thin layer chromatography

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