Self-assembled Structures Formed by Fmoc modified aliphatic amino acids

Bharti Koshti^{a+}, Soumick Naskar^a, Vivekshinh Kshtriya^a, Hanuman Narode^a, Nidhi Gour^{a*}
[a] Department of Chemistry, Indrashil University, Kadi, Mehsana, Gujarat, India; E-mail: gournidhi@gmail.com; nidhi.gour@indrashiluniversity.edu.in

Abstract

Herein, we report the self-assembled structures formed by Fmoc modified aliphatic uncharged single amino acids. The self-assembling properties of (((9H-fluoren-9yl)methoxy)carbonyl)-L-alanine (Fmoc-Ala-OH), (((9H-fluoren-9-yl)methoxy)carbonyl)-Lleucine (Fmoc-Leu-OH), (((9H-fluoren-9-yl)methoxy)carbonyl)-L-isoleucine (Fmoc-Ile-OH), and (((9H-fluoren-9-yl)methoxy)carbonyl)-L-valine (Fmoc-Val-OH) were studied under varying conditions such as concentration and temperature. Fmoc-Ala-OH shows flower-like self-assembled structure at both low and high concentration under room temperature as well as on heating at 70°C. We also studied self-assembly of the modified branched chain amino acids (BCAA) i.e. Fmoc-Leu-OH, Fmoc-Ile-OH, and Fmoc-Val-OH.

Fmoc-Leu-OH forms flower-like morphology at both low and high concentration under room temperature which changes to small tube-like structure on heating. **Fmoc-Ile-OH** on the other hand shows fibres-like self-assembly at lower and higher concentration at room temperature. While, on heating at lower concentration they formed a tube like self-assembled structure and at higher concentration they formed a fibres-like morphology. In the case of **Fmoc-Val-OH** they form a flower-like morphology at lower concentration at room temperature and at higher concentration they formed fibres-like assembly at room temperature. On the other hand, on heating **Fmoc-Val-OH** shows a fibres-like assembly at lower and higher concentration. Once the self-assembled structure of all Fmoc single amino acid characterized through the optical microscopy then our future aims to characterized those self-assembled structure through sophisticated microscopy and spectroscopy techniques and understand the mechanisms of self-assembled structure. Hence, the modified amino acids may pave the way for the design of novel self-assembled architectures which can controllable manipulated to impart desired function..

Keywords

Self-assembly; Fmoc modified single amino acid; aliphatic amino acids; flower; fibers

Introduction

Design of novel nano/macro architecture with the help of self-assembly is a fast growing fied of research due to wide range of applications. Self-assembled architectures based on the amino acids are of particular interest owing to their applications in different fields which include biomedical application,¹⁻³ sensing,⁴ tracking,⁵ fabrication⁶ etc purpose increase due to its biocompatibility,^{7, 8} and stability.⁹ Amino acid based self-assembled structure formed various types of morphology at supramolecular level due to its amphiphilic character.¹⁰ The shape of the self-assembled structure may be the fibers,¹¹⁻¹³ rod,^{14, 15} spherical,^{16, 17} micelles,¹⁸ tube,¹⁹ broomstick and elongated fibers,²⁰ and vesicles²¹ etc. Amino acids based selfassembled structures have also received tremendous attention due to their association with amyloid disease which can open new avenues for understanding the role of single amino aggregation in amyloid disease.^{12, 22} Gazit et al. for the very first time reported the selfassembly of short dipeptide Phe-Phe,¹¹ followed by phenylalanine,¹³ tyrosine,²³ and tryptophan²⁴. In other study done by Wangoo et al. reported the self-assembly of aliphatic single amino acids (Ala, Leu, Ile, and Val).²⁶ However, when the amino acid conjugated with fluorenylmethoxycarbonyl (Fmoc-) group it may get gel like properties due to ability of fluorenylmethoxycarbonyl (Fmoc-) function group form the thick entangled

fibres having ability to trap the water molecules inside the entangled fibres which facilitated material to form the gel like properties.^{27, 28} Previous literature reports suggest that various Fmoc protected amino acids show gel-like properties such as Fmoc-F, Fmoc-M, Fmoc-Y, Fmoc-G, Fmoc-W, and Fmoc-I etc while Fmoc-alanine, Fmoc-Valine, Fmoc-leucine did not form the gel-like²⁷ properties due to ambiguous behaviours. Recently In past extensive work has been reported by the various groups which may include self-assembly Fmoc-L-Lysine in different organic solvents done by Kundu et al.³¹ In other study reported by Panda et al. demonstrated that Fmoc-cysteine forms sphere like self-assembly and its application in drug delivery.³² Gazit et al demonstrated that fluorenyl-methoxy-carbonyl- β , β -diphenyl-Ala-OH (Fmoc-Dip-Ala) to opel gemstone-like structure¹⁷. Another study done by Sato et al reported the self-assembly Fmoc-lysine in DMSO: water mixture³³ and Bai et al reported Fmoc-dipeptide and used in catalytic role as thermolysin.³⁴ In another study did by Gazit et al. co-assembly of Fmoc diphenylalanine and diphenylalanine form the 3D fibers viscous network.³⁵

Our group is interested in studying the self-assembly of single amino acid,^{12, 25, 36, 37} modified single amino acids,^{20, 29, 30} peptides,³⁸⁻⁴³ and heterocyclic compounds.⁴⁴⁻⁴⁸ In this direction, our group has reported self-assembly of aliphatic single amino acid cysteine and methionineto amyloid like structures.¹² Subsequently, we also reported unusual self-assembled structures formed by proline, hydroxy proline and lysine.²⁵ Hence, motivated from our studies on single amino acid based structures we were also interested to assess self-assembly of modified single amino acids. In this context, our group has studied the self-assembled structure formed Fmoc variant of threonine N-(9-Fluorenylmethoxycarbonyl)-O-tert-butyl-L-threonine (Fmoc-Thr(tbu)-OH) and Fmoc variant of serine N-(((9H-fluoren-9-yl)methoxy)carbonyl)-O-(tert-butyl)-L-serine (Fmoc-Ser(tbu)-OH) at various concentration and temperature and reported formation of different morphologies such as sphere, double

sided broom-stick, dumbbell like self-assembled structure.²⁰ In another study, we reported self-assembled structure formation by different CBZ protected aromatic amino acids which include Z-Phe, Z-Trp, and Z-Tyr under varying concentration and temperature.²⁹ Recently, our group also reported the self-assembled structure formed by the modified charge amino acids, Fmoc-Glu(OtBu)-OH, Fmoc-Asp(OtBu)-OH, and Fmoc-Lys(Boc)-OH.³⁰ Herein. this manuscript, we have reported the self-assembly of Fmoc variant of alanine (((9H-fluoren-9-yl)methoxy)carbonyl)-L-alanine (**Fmoc-Ala-OH**), (((9H-fluoren-9-yl)methoxy)carbonyl)-L-leucine (**Fmoc-Leu-OH**), (((9H-fluoren-9-yl)methoxy)carbonyl)-L-isoleucine (**Fmoc-Ile-OH**), (((9H-fluoren-9-yl)methoxy)carbonyl)-L-isoleucine (**Fmoc-Val-OH**) under varying concentration and temperature and controlled morphological changes associated with it.



Figure 1. Morphological changes observed in the self-assembled structures formed by Fmoc-Ala-OH under varying concentration and temperature.

Figure 1 shows the graphical representation of the self-assembled structure formed by **Fmoc-Ala-OH** under varying conditions. We found that **Fmoc-Ala-OH** shows a flower-like assembly at both lower and higher concentration under room temperature condition. While on heating the assemblies were not affected and they formed a flower-like structure in both lower and higher concentration.



Figure 2. Morphological changes observed in the self-assembled structures formed by **Fmoc**-**Leu-OH** under varying concentration and temperature.

Figure 2 shows the graphical representation of self-assembled structure formed by Fmoc variant of Leu i.e. **Fmoc-Leu-OH**. **Fmoc-Leu-OH** shows a flower-like morphology at room temperature in both lower and higher concentration. While on heating the flower-like morphology changed to the small tube-like structure in both lower and higher concentration.



Figure 3. Morphological changes observed in the self-assembled structures formed by **Fmoc-Ile-OH** under varying concentration and temperature.

Figure 3 shows the graphical representation of the self-assembled structure formation by **Fmoc-Ile-OH. Fmoc-Ile-OH** formed a fiber like self-assembled structure at both lower and higher concentration under room temperature condition while on heating at lower

concentration the fibers like structure changes to small tube-like morphology. The fibers formed by **Fmoc-Ile-OH** at higher concentration are not affected by heating.



Figure 4. Morphological changes observed in the self-assembled structures formed by Fmoc-Val-OH under varying concentration and temperature.

Figure 4 shows the graphical representation of self-assembly formation by **Fmoc-Val-OH**. Our study suggests that **Fmoc-Val-OH** shows a flower-like self-assembled structure at lower concentration at room temperature while on heating it forms a fiber-like structure at room temperature. On heating they form the same fiber like morphology at both lower and higher concentration. **Result and Discussion**



Scheme 1: Chemical structure of Fmoc-Ala-OH, Fmoc-Leu-OH, Fmoc-Ile-OH, and Fmoc-Val-OH

The chemical structure of **Fmoc-Ala-OH**, **Fmoc-Leu-OH**, **Fmoc-Ile-OH**, and **Fmoc-Val-OH** is shown in Scheme 1. The self-assembled structures formed by all the non-aromatic modified amino acids were extensively studied by optical microscopy at various concentration and temperature.



Figure 5. Self-assembled structures formed by **Fmoc-Ala-OH** at room temperature (a) Optical microscopy image at 3 mM under 40X; (b) Optical microscopy image at 3 mM under 63X; and (c) Optical microscopy image at 8 mM under 40X; (d) Optical microscopy image at 8 mM under 63X.



Figure 6. Self-assembled structures formed by **Fmoc-Ala-OH** on heating at 70 ⁰C (a) Optical microscopy image at 3 mM under 40X; (b) Optical microscopy image at 3 mM under 63X; and (c) Optical microscopy image at 8 mM under 40X; (d) Optical microscopy image at 8 mM under 63X.



Figure 7. Self-assembled structures formed by **Fmoc-Leu-OH** at room temperature (a) Optical microscopy image at 2 mM under 20X; (b) Optical microscopy image at 2 mM under 63X; (c) Optical microscopy image at 7 mM under 20X; (d) Optical microscopy image at 7 mM under 40X.

Figure 8. Self-assembled structures formed by **Fmoc-Leu-OH** on heating at 70 0 C (a) Optical microscopy image at 2 mM under 40X; (b) Optical microscopy image at 2 mM under 63X; and (c) Optical microscopy image at 7 mM under 40X; (d) Optical microscopy image at 7 mM under 63X.

Figure 9. Self-assembled structures formed by **Fmoc-Ile-OH** at room temperature (a) Optical microscopy image at 3 mM under 20X; (b) Optical microscopy image at 3 mM under 63X; (c) Optical microscopy image at 8 mM under 20X; (d) Optical microscopy image at 8 mM under 40X.

Figure 10. Self-assembled structures formed by **Fmoc-Ile-OH** on heating at 70 0 C (a) Optical microscopy image at 3 mM under 40X; (b) Optical microscopy image at 3 mM under 63X; (c) Optical microscopy image at 8 mM under 20X; (d) Optical microscopy image at 8 mM under 63X

Figure 11. Self-assembled structures formed by **Fmoc-Val-OH** at room temperature (a) Optical microscopy image at 3 mM under 20X; (b) Optical microscopy image at 3 mM under 40X; (c) Optical microscopy image at 9 mM under 20X; (d) Optical microscopy image at 9 mM under 20X.

Figure 12. Self-assembled structures formed by **Fmoc-Val-OH** on heating at 70 0 C (a) Optical microscopy image at 3 mM under 20X; (b) Optical microscopy image at 3 mM under 40X; (c) Optical microscopy image at 9 mM under 20X; (d) Optical microscopy image at 9 mM under 40X.

| Table 1: | Summary of | morphological | transition | observed | by the | modified | single | amino | acid |
|------------|-------------|---------------|------------|----------|--------|----------|--------|-------|------|
| at varving | conditions. | | | | | | | | |

| Modified Amino Acid | Concentration | Morphology | Condition |
|---------------------|---------------|------------|---------------------------------|
| Fmoc-Ala-OH | Lower (3 mM) | Flower | RT |
| Fmoc-Ala-OH | Higher (8 mM) | Flower | RT |
| Fmoc-Ala-OH | Lower (3 mM) | Flower | On heating at 70 ⁰ C |
| Fmoc-Ala-OH | Higher (8 mM) | Flower | On heating at 70 ⁰ C |
| | | | |
| Fmoc-Leu-OH | Lower (2 mM) | Flower | RT |

| Fmoc-Leu-OH | Higher (7 mM) | Flower | RT | |
|---------------|---------------|---------------------|---------------------------------|--|
| | | | | |
| (Fmoc-Leu-OH) | Lower (2 mM) | Tube-like structure | On heating at 70 ⁰ C | |
| (Fmoc-Leu-OH) | Higher (7 mM) | Tube-like structure | On heating at 70 ⁰ C | |
| | | | | |
| Fmoc-Ile-OH | Lower (3 mM) | Fiber | RT | |
| Fmoc-Ile-OH | Higher (3 mM) | Fiber | RT | |
| Fmoc-Ile-OH | Lower (8 mM) | Tube-like structure | On heating at 70 ⁰ C | |
| Fmoc-Ile-OH | Higher (8 mM) | Fiber | On heating at 70 ⁰ C | |
| | | | | |
| Fmoc-Val-OH | Lower (3 mM) | Flower | RT | |
| Fmoc-Val-OH | Higher (9 mM) | Fiber | RT | |
| Fmoc-Val-OH | Lower (3 mM) | Fiber | On heating at 70 ⁰ C | |
| Fmoc-Val-OH | Higher (9 mM) | Fiber | On heating at 70 ⁰ C | |

Conclusion

In conclusion, we assessed the self-assembled structure formation by Fmoc variant of alanine and branched chain amino acids. Our study reveal that **Fmoc-Ala-OH**, **Fmoc-Leu-OH**, **Fmoc-Ile-OH**, and **Fmoc-Val-OH** all assemble to well defined self-assembled structures and also reveal morphological transition as we alter the concentration and temperature.. The results presented here has immense application to design of novel structures via bottom–up approach. There is ever increasing demand to find new bioorganic scaffolds for the design of novel nanoarchitectures through self-assembly. The study of modified single amino acid self-assembly research is important step in this direction due to its good biocompatibility and ease of chemical modification due to which it may have wide range of biomedical applications

Materials and method

General

All chemicals used in these studies were of purity greater than 99%. All the solvents and **Fmoc-Ala-OH**, **Fmoc-Leu-OH**, **Fmoc-Ile-OH**, and **Fmoc-Val-OH** were purchased from commercial suppliers. All the compounds were used without further purification. All the studies were done using distilled solvents. Ultrapure water was used for all the studies.

Optical Microscopy

A 20 mM stock solution of **Fmoc-Ala-OH**, **Fmoc-Leu-OH**, **Fmoc-Ile-OH**, and **Fmoc-Val-OH**, were prepared in 50 % aqueous solution of methanol and heat it with sonication to get a clear solution. The solution was further diluted at 1 to 10 mM concentration by using Milli Q water. A turbid solution has been observed on dilution with water. The self-assembly formed by this amino acid was first assessed by optical microscopy. For optical microscopy a drop casting a 20 μ L solution on a glass slide and dry it at room temperature. The same solutions were heated at 70 $^{\circ}$ C and then immediately drop casting 20 μ L solution of this solution on a clean glass slide. For all the microscopic study always a fresh stock solution and fresh samples has been prepared. The samples were visualized using a Leica DM2500 upright fluorescent microscope at different magnifications.

Corresponding Author

Department of Chemistry, Indrashil University, Mehsana, Gujarat, 382740, India; E-mail: gournidhi@gmail.com; nidhigour.iu@gmail.com; Fax: +91 7930514110.

Funding Sources

The work was supported by the DST SERB extramural research fund (Project No. EMR/2016/003186) and SERB SPG/2021/000521 received by Dr. Nidhi Gour.

Conflicts of interest

There is no conflict of interest to declare.

Acknowledgment

NG, BK and VK greatly acknowledge support from SERB research grant (EMR/2016/003186) and SERB SPG/2021/000521 for funding and fellowships. VK thanks to ICMR for the senior research fellowship No (45/13/2020-/BIO/BMS). BK thanks SHODH for funding and Indrashil University for infrastructure support.

Reference

1. Vong, L. B.; Trinh, N.-T.; Nagasaki, Y., Design of amino acid-based self-assembled nano-drugs for therapeutic applications. *Journal of Controlled Release* **2020**, *326*, 140-149.

2. Pinheiro, L.; Faustino, C., Amino acid-based surfactants for biomedical applications. *Application and Characterization of Surfactants* **2017**.

3. Perween, S.; Chandanshive, B.; Kotamarthi, H. C.; Khushalani, D., Single amino acid based self-assembled structure. *Soft Matter* **2013**, *9* (42), 10141-10145.

4. Mandler, D.; Kraus-Ophir, S., Self-assembled monolayers (SAMs) for electrochemical sensing. *Journal of Solid State Electrochemistry* **2011**, *15* (7-8), 1535.

5. Gu, L.; Li, X.; Jiang, J.; Guo, G.; Wu, H.; Wu, M.; Zhu, H., Stem cell tracking using effective self-assembled peptide-modified superparamagnetic nanoparticles. *Nanoscale* **2018**, *10* (34), 15967-15979.

6. Chakraborty, P.; Gazit, E., Amino Acid Based Self-assembled Nanostructures: Complex Structures from Remarkably Simple Building Blocks. *ChemNanoMat* **2018**, *4* (8), 730-740.

7. Vasantha, T.; Kumar, A.; Attri, P.; Venkatesu, P.; Rama Devi, R., The solubility and stability of amino acids in biocompatible ionic liquids. *Protein and peptide letters* **2014**, *21* (1), 15-24.

8. Pinazo, A.; Pons, R.; Pérez, L.; Infante, M. R., Amino acids as raw material for biocompatible surfactants. *Industrial & Engineering Chemistry Research* **2011**, *50* (9), 4805-4817.

9. Li, Y.; Zou, Q.; Yuan, C.; Li, S.; Xing, R.; Yan, X., Amino acid coordination driven self-assembly for enhancing both the biological stability and tumor accumulation of curcumin. *Angewandte Chemie International Edition* **2018**, *57* (52), 17084-17088.

10. Lombardo, D.; Kiselev, M. A.; Magazù, S.; Calandra, P., Amphiphiles self-assembly: basic concepts and future perspectives of supramolecular approaches. *Advances in Condensed Matter Physics* **2015**, *2015*.

11. Reches, M.; Gazit, E., Casting metal nanowires within discrete self-assembled peptide nanotubes. *Science* **2003**, *300* (5619), 625-627.

12. Gour, N.; Kanth P, C.; Koshti, B.; Kshtriya, V.; Shah, D.; Patel, S.; Agrawal-Rajput, R.; Pandey, M. K., Amyloid-like structures formed by single amino acid self-assemblies of cysteine and methionine. *ACS chemical neuroscience* **2018**, *10* (3), 1230-1239.

13. Adler-Abramovich, L.; Vaks, L.; Carny, O.; Trudler, D.; Magno, A.; Caflisch, A.; Frenkel, D.; Gazit, E., Phenylalanine assembly into toxic fibrils suggests amyloid etiology in phenylketonuria. *Nature chemical biology* **2012**, *8* (8), 701-706.

14. Singh, P.; Brar, S. K.; Bajaj, M.; Narang, N.; Mithu, V. S.; Katare, O. P.; Wangoo, N.; Sharma, R. K., Self-assembly of aromatic α -amino acids into amyloid inspired nano/micro scaled architects. *Materials Science and Engineering: C* **2017**, *72*, 590-600.

15. Aggeli, A.; Nyrkova, I. A.; Bell, M.; Harding, R.; Carrick, L.; McLeish, T. C.; Semenov, A. N.; Boden, N., Hierarchical self-assembly of chiral rod-like molecules as a model for peptide β -sheet tapes, ribbons, fibrils, and fibers. *Proceedings of the National Academy of Sciences* **2001**, *98* (21), 11857-11862.

16. Joshi, K. e. B.; Verma, S., Ditryptophan conjugation triggers conversion of biotin fibers into soft spherical structures. *Angewandte Chemie International Edition* **2008**, *47* (15), 2860-2863.

17. Arnon, Z. A.; Pinotsi, D.; Schmidt, M.; Gilead, S.; Guterman, T.; Sadhanala, A.; Ahmad, S.; Levin, A.; Walther, P.; Kaminski, C. F., Opal-like multicolor appearance of self-assembled photonic array. *ACS applied materials & interfaces* **2018**, *10* (24), 20783-20789.

18. Bachar, M.; Mandelbaum, A.; Portnaya, I.; Perlstein, H.; Even-Chen, S.; Barenholz, Y.; Danino, D., Development and characterization of a novel drug nanocarrier for oral delivery, based on self-assembled β -casein micelles. *Journal of controlled release* **2012**, *160* (2), 164-171.

19. Escuder, B.; Rowan, A. E.; Feiters, M. C.; Nolte, R. J., Enantioselective binding of amino acids and amino alcohols by self-assembled chiral basket-shaped receptors. *Tetrahedron* **2004**, *60* (2), 291-300.

20. Kshtriya, V.; Koshti, B.; Gour, N., Controlled morphological changes in self-assembled structures formed by Fmoc variants of Threonine and Serine. **2021**.

21. Sapala, A. R.; Dhawan, S.; Haridas, V., Vesicles: self-assembly beyond biological lipids. *RSC advances* **2017**, *7* (43), 26608-26624.

22. Stanković, I. M.; Niu, S.; Hall, M. B.; Zarić, S. D., Role of aromatic amino acids in amyloid self-assembly. *International journal of biological macromolecules* **2020**, *156*, 949-959.

23. Zaguri, D.; Kreiser, T.; Shaham-Niv, S.; Gazit, E., Antibodies towards tyrosine amyloid-like fibrils allow toxicity modulation and cellular imaging of the assemblies. *Molecules* **2018**, *23* (6), 1273.

24. Shaham-Niv, S.; Rehak, P.; Vuković, L.; Adler-Abramovich, L.; Král, P.; Gazit, E., Formation of apoptosis-inducing amyloid fibrils by tryptophan. *Israel Journal of Chemistry* **2017**, *57* (7-8), 729-737.

25. Koshti, B. K., Vivekshinh; Singh, Ramesh; Walia, Shanka ; Bhatia, Dhiraj; Joshi, Khashti; Gour, Nidhi, Unusual Aggregates Formed by the Self-assembly of Proline, Hydroxyproline and Lysine. *ACS Chemical Neuroscience* **2021**, (Just accepted).

26. Singh, P.; Pandey, S. K.; Grover, A.; Sharma, R. K.; Wangoo, N., Understanding the selfordering of amino acids into supramolecular architectures: co-assembly-based modulation of phenylalanine nanofibrils. *Materials Chemistry Frontiers* **2021**, *5* (4), 1971-1981.

27. Draper, E. R.; Morris, K. L.; Little, M. A.; Raeburn, J.; Colquhoun, C.; Cross, E. R.; McDonald, T. O.; Serpell, L. C.; Adams, D. J., Hydrogels formed from Fmoc amino acids. *CrystEngComm* **2015**, *17* (42), 8047-8057.

28. Estroff, L. A.; Hamilton, A. D., Water gelation by small organic molecules. *Chemical reviews* **2004**, *104* (3), 1201-1218.

29. Gour, N.; Kshtriya, V.; Koshti, B.; Narode, H.; Naskar, S., Controlled self-assembly of modified aromatic amino acids. **2021**.

30. Nidhi Gour, B. K., Soumick Naskar, Vivekshinh Kshtriya, Hanuman Narode, Controlled aggregation properties of modified single amino acids. *ChemRvix* **2021**.

31. Hashemnejad, S. M.; Huda, M. M.; Rai, N.; Kundu, S., Molecular insights into gelation of difmoc-L-lysine in organic solvent–water mixtures. *ACS omega* **2017**, *2* (5), 1864-1874.

32. Chibh, S.; Katoch, V.; Kour, A.; Khanam, F.; Yadav, A. S.; Singh, M.; Kundu, G. C.; Prakash, B.; Panda, J. J., Continuous flow fabrication of Fmoc-cysteine based nanobowl infused core–shell like microstructures for pH switchable on-demand anti-cancer drug delivery. *Biomaterials Science* **2021**, *9* (3), 942-959.

33. Narang, N.; Sato, T., Liquid-liquid phase separation and self-assembly of a lysine derivative Fmoc-L-lysine in water-DMSO mixtures. *Polymer Journal* **2021**, 1-12.

34. Wang, M.; Zhang, Q.; Jian, H.; Liu, S.; Li, J.; Wang, A.; Dong, Q.; Ren, P.; Li, X.; Bai, S., Role of Thermolysin in Catalytic-Controlled Self-Assembly of Fmoc-Dipeptides. *CCS Chemistry* **2020**, *2* (4), 317-328.

35. Chakraborty, P.; Tang, Y.; Guterman, T.; Arnon, Z. A.; Yao, Y.; Wei, G.; Gazit, E., Co-Assembly between Fmoc Diphenylalanine and Diphenylalanine within a 3D Fibrous Viscous Network Confers Atypical Curvature and Branching. *Angewandte Chemie International Edition* **2020**, *59* (52), 23731-23739.

36. Koshti, B.; Kshtriya, V.; Nardin, C.; Gour, N., Chemical Perspective of the Mechanism of Action of Antiamyloidogenic Compounds Using a Minimalistic Peptide as a Reductionist Model. *ACS Chemical Neuroscience* **2021**.

37. Nidhi Gour, E. G., Metabolite Assemblies: A Surprising Extension to the Amyloid Hypothesis. *Current Opinion in Chemical Biology* **2021**.

38. Kedracki, D.; Filippov, S. K.; Gour, N.; Schlaad, H.; Nardin, C., Formation of DNA-Copolymer Fibrils Through an Amyloid-Like Nucleation Polymerization Mechanism. *Macromolecular Rapid Communications* **2015**, *36* (8), 768-773.

39. Gour, N.; Kedracki, D.; Safir, I.; Ngo, K. X.; Vebert-Nardin, C., Self-assembling DNA–peptide hybrids: morphological consequences of oligonucleotide grafting to a pathogenic amyloid fibrils forming dipeptide. *Chemical Communications* **2012**, *48* (44), 5440-5442.

40. Gour, N.; Verma, S., Bending of peptide nanotubes by focused electron and ion beams. *Soft Matter* **2009**, *5* (9), 1789-1791.

41. Gour, N.; Barman, A. K.; Verma, S., Controlling morphology of peptide-based soft structures by covalent modifications. *Journal of Peptide Science* **2012**, *18* (6), 405-412.

42. Gour, N.; Mondal, S.; Verma, S., Synthesis and self-assembly of a neoglycopeptide: morphological studies and ultrasound-mediated DNA encapsulation. *Journal of Peptide Science* **2011**, *17* (2), 148-153.

43. Abraham, J. N.; Gour, N.; Bolisetty, S.; Mezzenga, R.; Nardin, C., Controlled aggregation of peptide–DNA hybrids into amyloid-like fibrils. *European Polymer Journal* **2015**, *65*, 268-275.

44. Gour, N.; Kshtriya, V.; Gupta, S.; Koshti, B.; Singh, R.; Patel, D.; Joshi, K. B., Synthesis and Aggregation Studies of a Pyridothiazole-Based AIEE Probe and Its Application in Sensing Amyloid Fibrillation. *ACS Applied Bio Materials* **2019**, *2* (10), 4442-4455.

45. Kshtriya, V.; Koshti, B.; Pandey, D. K.; Kharbanda, S.; Kanth P, C.; Singh, D. K.; Bhatia, D.; Gour, N., Sequential and cellular detection of copper and lactic acid by disaggregation and reaggregation of the fluorescent panchromatic fibres of an acylthiourea based sensor. *Soft Matter* **2021**, *17* (16), 4304-4316.

46. Kshtriya, V.; Koshti, B.; Gangrade, A.; Haque, A.; Singh, R.; Joshi, K. B.; Bhatia, D. D.; Gour, N., Self-assembly of a Benzothiazolone Conjugate to Panchromatic Fluorescent Fibres and its Application in Cellular Imaging. *New Journal of Chemistry* **2021**.

47. Kshtriya, V., Koshti, B., Haque, A., Gangrade, A., Singh, R., Joshi, khashti B., Bandyopadhyay, S., Bhatia, D., and Gour, N., Sunflower-like Fluorescent Self-Assembled Morphologies Formed by Pyridothiazole Based Aggregation Induced Emission (AIE) Dye and Its Cell Imaging Applications. ChemRxiv. *ChemRvix* **2021**, Preprint

48. Kshtriya, V., Koshti, B., Haque, A., Gangrade, A., Singh, R., Joshi, K. B., Bandyopadhyay, S., Bhatia, D., and Gour, N, Self-Assembly and Photophysical Studies of an Unusual Red Colored Dye Which Show Green Fluorescence in Cell Imaging. *ChemRvix* **2021**, Preprint