JigCell: A New Environment to Simulate the Dynamics of Memory Formation

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ABSTRACT:

Since their initial discovery, long-term potentiation (LTP), and long-term depression (LTD) are accepted as the main biomolecular mechanism that controls memory acquisition. In doing this, both mechanisms are fairly complex and involve specific triggers and many cascades reactions that cross-talk and communicate with others. Thus, they are very complex. To reveal how these mechanisms operate and instruct the brain to remember and forget, one judicious approach is developing the mathematical models of processes. However, this notion requires some basic knowledge regarding ordinary differential equations and writing codes. To this respect, it can be postulated that tools, which can be utilized rather by everyone, would certainly expedite and facilitate the formulation of such models. With this rationale in mind, we demonstrate that JigCell offers the perfect platform to develop such models of LTP. Our choice for this tool stems from the fact that it is designed to simulate complex biological systems in a modular way. Thus, this manuscript is crafted to illustrate how this model was constructed in the JigCell environment and to give an idea of how this tool works.

Keywords: Memory formation, Long-term potentiation, Mathematical model, JigCell, Calcium, CaMKII.

1. INTRODUCTION

The question of how we code information in our memory has remained a scientific puzzle, that is inherited from one generation to the next one. However, it should not be interpreted as no progress has been made. On the contrary, so much is known from Hebb's early work. Commonly known as Hebb's rule or cell assembly theory, this theory is summarized as "Cells that fire together wire together". With that, Hebb indeed postulates that activation of pre-and postsynaptic cells strengths the connection between neurons [1]. To communicate more, the synapses of neurons are observed to change, as a result of which their synaptic strength is enhanced. This observation is what we presently know as synaptic plasticity, which depicts the capacity of synapsis to strengthen or weaken in time [2]. Presently, our view of memory formation is dominated by long-term changes in synaptic communication. This effect is accepted to be bidirectional; long-term potentiation (LTP) as a long-lasting increase in signal transmission between neurons and long-term depression (LTD), in which case the extent of transmission between neurons is weakened. In summary, LTP and LDP are accepted as the biomolecular mechanism that underlies memory acquisition [3].

The structure of how LTP operates starts up with specific triggers. Under the resting condition, the pores of NMDA (N-methyl-D-aspartate) receptors are occupied with Mg^{2+} ions. Upon depolarization, Mg^{2+} is removed from the pore, which culminates in the influx of Na⁺, K⁺, and Ca²⁺ [4]. Of them, calcium influx triggers a postsynaptic cascade of reactions that alters the density of AMPA receptors at the post-synapsis. In this process, the frequency in calcium dictates the fate of memory formation by impacting the direction on the alteration of AMPA receptors; high-frequency calcium influx results in LTP, whereas low-frequency calcium influx actuates LTD. Thus, calcium influx has a bidirectional role in deciding whether we remember or we forget [5].

There is no need to mention that in reality, the process of how the brain function is far more perplexing than what is described. For this reason, the way to fully grasp and study all biochemical processes postulates composing mathematical models regarding biochemical events taking place during memory formation. With this motivation, many groups (including ours) have developed models that cover either all or a certain fraction of LTP (for instance, see [6-10]).

Conspicuously, such mathematical models could only be formulated by researchers who are familiar with solving ordinary differential equations or writing codes. To this respect, software that can be utilized by everyone without requiring too much background in mathematics and computer science will assuredly expedite progress in this field. For this reason, the JigCell Model Connector was introduced to the scientific community [11]. This software is specially devised to construct molecular network models which are inherently complex. By design, JigCell allows hierarchical model composition; that is to say, a complex network is engineered as a combination of smaller models. In the JigCell environment, these smaller models are named "modules" and each one represents a well-defined fragment of the model. Once the modules are connected through the so-called "interface ports", the overall network is obtained. This way, one can understand how a certain part of the model behaves and interacts with others. From standpoint of user-friendliness, JigCell is very advantageous; the ordinary equations are written through a panel without demanding a code to solve them. Thus, it can be used by anyone.

Given the complexity of the mechanism underlying memory formation, it is reasoned that this software offers the perfect platform to elucidate the dynamics in LTP. To address the applicability of JigCell in this context, a model that covers the Ca2+/CaMKII network is formed as an illustration, with the use of differential equations reported in the literature. Ultimately, this manuscript is crafted to discuss how this model is formed in the JigCell environment and more importantly, to give the reader a feel about the basics of this software.

2. MATERIAL AND METHODS

JigCell and COPASI (analysis environment for JigCell) can be downloaded for free from the link given in [12]. In the JigCell environment, the model designed to simulate the Ca2+/CaMKII network is envisioned to harbor two modules; Module 1 and Module 1. In this model, Module 1 relates to the chemistry of calcium, which serves as the well-established initiation signal of LTP. On the other hand, Module 2 deals with the dynamics of CaMKII, a protein kinase known to activate the following downstream reactions in memory formation.

Before discussing the content of each module, this section of the manuscript envisions providing some insight on how this software works. In general, the JigCell environment offers two rationales to implement rules for dynamic changes. The first one is through the "species" menu; the dynamics of reactants and variables (collectively coined as "species") are given through ordinary differential equations (ODE). To illustrate, this rationale was utilized in Module 1, wherein the calcium influx is delineated by the ODE given in the box "expression" (Figure 1).

Read	tions Species	5 Module quantities	Functions Eve	nts Compartments	Equations Model prope	rties Ports
#	Name	Initial quantity	Туре	Compartment	Expression	Notes
1	Ca	0.01	ODE	cell	-5*Ca+a*(1+Ca1)	
2	Ca1	0.1	ODE	cell	100*Ca2	
3	Ca2	0.1	ODE	cell	-100*Ca1	

Figure 1. A screenshot of the menu "Species".

The second one involves the representation of these dynamic changes in terms of chemical reactions, as in Module 2. This one is more suited to the cases with multiple species reacting in the form of a cascade (such as phosphorylation/dephosphorylation reaction of CaMKII subunits). For this one, the components of each reaction are predefined in the menu "Functions", wherein the left column is utilized to itemize the components of each reaction in the form of reactants ("MOD") or rate-constants ("VAR"). Subsequently, the right column sets the reaction rules that combine the stoichiometric coefficients, and the rate constants with the reactants (Figure 2).

	Reacti	ions Species	Module quantities	Functions	Events	Со	mpartments	Equations	Model properties	Ports			
	# Name						Equation						
	1	Rxn1(MOD A1	,VAR c1,VAR c3)				-c1+c3*A1						
	2	Rxn2(MOD A1	,MOD A2,VAR c1,	/AR c2,VAR	c3)		c1-c3*A1-c2	*A1+2*c3*	A2				
	3	Rxn3(MOD A1	,MOD A2,MOD A3,	VAR c2,VA	R c3)		c2*A1-2*c3*	'A2-1.8*c2'	*A2+3*c3*A3				
	4	Rxn4(MOD A2	2,MOD A3,MOD A4,	VAR c2,VA	R c3)		1.8*c2*A2-3	*c3*A3-2.3	*c2*A3+4*c3*A4				
	5	Rxn5(MOD A3	3,MOD A4,MOD A5,	VAR c2,VA	R c3)		2.3*c2*A3-4	*c3*A4-2.7	'*c2*A4+5*c3*A5				
	6	Rxn6(MOD A4	,MOD A5,MOD A6,	VAR c2,VA	R c3)		2.7*c2*A4-5	*c3*A5-2.8	*c2*A5+6*c3*A6				
	7	Rxn7(MOD A5	5,MOD A6,MOD A7,	VAR c2,VA	R c3)		2.8*c2*A5-6	*c3*A6-2.7	'*c2*A6+7*c3*A7				
	8	Rxn8(MOD A6	5,MOD A7,MOD A8,	VAR c2,VA	R c3)		2.7*c2*A6-7	*c3*A7-2.3	*c2*A7+8*c3*A8				
	9	Rxn9(MOD A7	,MOD A8,MOD A9,	VAR c2,VA	R c3)		2.3*c2*A7-8	*c3*A8-1.8	*c2*A8+9*c3*A9				
	10	Rxn10(MOD A	10,MOD A8,MOD A	9,VAR c2,	/AR c3)		1.8*c2*A8-9	*c3*A9-c2*	*A9+10*c3*A10				
Π	11	Rxn11(MOD A	10,MOD A9,VAR c	2,VAR c3)			c2*A9-c3*10)*A10					

Figure 2. A screenshot of the menu "functions".

With the rate constants and reactants defined once again as "species" (Figure 1), the kinetic law for each reaction is then formulated through the menu "Reactions" (Figure 3).

Reacti	ons Species	Module quantities	Functions	Events	Compartme	ents	Equations	Model properties
#	Name (opt)	Reaction	ĸ	(inetic Ty	pe k	Kineti	c Law	
1	"1"	-> P0; P1	U	ser Defin	ed R	xn1_	1(P1,v1,v3))
2	"2"	-> P1; P1 P2	U	ser Defin	ed R:	xn2_	1(P1,P2,v1	,v2,v3)
3	"3"	-> P2; P1 P2 P3	U	ser Defin	ed R:	xn3_	1(P1,P2,P3	,v2,v3)
4	"4"	-> P3; P2 P3 P4	U	ser Defin	ed R:	xn4_	1(P4,P2,P3	,v2,v3)
5	"5"	-> P4; P3 P4 P5	U	ser Defin	ed R:	xn5_	1(P4,P5,P3	,v2,v3)
6	"6"	-> P5; P4 P5 P6	U	ser Defin	ed R:	xn6_	1(P4,P5,P6	,v2,v3)
7	"7"	-> P6; P5 P6 P7	U	ser Defin	ed R:	xn7_:	1(P5,P6,P7	,v2,v3)
8	"8"	-> P7; P6 P7 P8	U	ser Defin	ed R:	xn8_:	1(P6,P7,P8	,v2,v3)
9	"9"	-> P8; P7 P8 P9	U	ser Defin	ed R:	xn9_	1(P7,P8,P9	,v2,v3)
10	"10"	-> P9; P10 P8 P9) U	ser Defin	ed R:	xn10	_1(P8,P9,P	10,v2,v3)
11	"11"	-> P10; P9 P10	U	ser Defin	ed R	xn11	1(P9,P10,	v2.v3)

Figure 3. A screenshot of the menu "Reactions".

To obtain the model, Module 1 (Figure 4) and Module 2 are connected through Ca port, whereby the chemical information is conveyed from Module 1 to its partner, Module 2 (as an example, a layout of Module 1 and the model are shown in Figure 4 and 5, respectively). Then, the model is exported with the SBML extension and is imported into the COPASI environment. Finally, the simulation is run from the "Time Course" menu, after its parameters are entered (in this paper, they are: duration is 10 s, intervals is 100, and interval size is 0.01 s). COPASI also has an interface to visualize the results of simulations. As a note, the files of this model are available from GitHub) [13].



Figure 4. The layout of Module 1. Note that the inputs and the outputs of Module 1 are marked in red and blue, respectively.



Figure 5. The layout of the model designed to simulate the $Ca^{2+}/CaMKII$ network in LTP.

3. RESULTS AND DISCUSSION

This section will elaborate on the nature of the modules, as well as the differential equation utilized in each one. Then, the results from the simulation will be discussed.

3.1. Module 1: Calcium Module

Serving as the initiation signal for LTP, the calcium module aims at simulating the dynamics of calcium uptake. A general rule, mathematical models solely consider the influx of calcium to post-synaptic neurons as a stimulant to LTP, whilst the roles of calcium from other intracellular sources (such as nuclear calcium) are very much omitted. Hence, calcium (Ca²⁺) in Module 1 stands for synaptic calcium in this manuscript. Of paramount importance is the notion that the model presented hereby doesn't consider any event prior to the calcium entry to post synapsis (for instance, membrane depolarization or displacement of Mg^{2+} from NMDA receptors). Thus, the oscillatory entry of calcium given in Module 1 functions as "the trigger" in this model.

Conventionally, the calcium influx is expressed as an instant elevation, followed by an exponential decay, as tabulated in Equation (1) [14].

$$\left[\operatorname{Ca}^{2+}\right] = \left[\operatorname{Ca}^{2+}\right]_{\operatorname{rest}} + A \sum_{i=1}^{n} \exp\left(-\frac{i}{f\tau}\right)$$
(1)

where $[Ca^{2+}]_{rest}$ is the resting concentration of calcium, A is the amplitude of a single Ca²⁺ pulse, f is the frequency of excitation, τ is the relaxation time of Ca2+ decay, and n is the number of pulses in the tetanic stimuli (i.g., high-frequency sequence of individual stimulations), with frequency varying from 5 Hz to 100 Hz. In constructing Module 1, the formulation of equation (1) in the JigCell environment posed a challenge, as such that this software doesn't recognize summation or exponential function. For this reason, equation (1) was redevised as a set of ordinary differential equations, wherein equation (2) describes the calcium influx, whilst equations (3), and (4) effectuate characteristic periodicity in the equation (2) with a frequency of 100 Hz (therein, variables Ca1 and Ca2 don't have any biological meaning). The kinetics of calcium influx described by equations (2-4) is shown in Figure 5 (parameters in these equations are described in Table 1).

$$\frac{d[\operatorname{Ca}](t)}{dt} = -\tau[\operatorname{Ca}](t) + \alpha \{1 + (\operatorname{Ca}_1)(t)\}$$
(2)

$$\frac{d(\operatorname{Ca}_{1})(t)}{dt} = \Omega(\operatorname{Ca}_{2})(t)$$
(3)

$$\frac{d(\operatorname{Ca}_2)(t)}{dt} = -\Omega(\operatorname{Ca}_1)(t)$$
(4)

Parameters	Description	Value
α	Strength of signal	50 M
Ω	Frequency	100 Hz
τ	Time-constant	5 s ⁻¹

Table 1. Parameters in equations 2-4.

=



Figure 5. The kinetics of calcium influx defined by the set of equations (2-4) (the basal concentration of calcium is 10 nM and the initial values of Ca_1 and Ca_2 are 0.1).

Unsurprisingly, this approach fails to reflect the exponential decay in equation 1. To engineer this behavior, the equation (2) was revised to the equation (5), by expressing the strength of the

signal in the form of a time-dependent variable. In doing this, the prediction was that this way, it would stimulate the calcium influx when initially equal to 50, whereas it would actuate the exponential decay, once set to 0.

$$\frac{d[\operatorname{Ca}](t)}{dt} = -\tau[\operatorname{Ca}](t) + \alpha(t)\{1 + (\operatorname{Ca}_1)(t)\}$$
(5)

$$\frac{d(\operatorname{Ca}_1)(t)}{dt} = \Omega(\operatorname{Ca}_2)(t)$$
(6)

$$\frac{d(\operatorname{Ca}_{2})(t)}{dt} = -\Omega(\operatorname{Ca}_{1})(t)$$
(7)

$$\alpha(t) = \begin{cases} 50 & \text{if } 0 \le t \le 5 \text{ secs,} \\ 0 & \text{otherwise.} \end{cases}$$
(8)

In the JigCell environment, discontinuity in functions is arranged from the menu "events" that stipulates two entries: "actions" to specify the nature of the triggered event and "delay" to indicate the trigger time for the event (Figure 6). With this menu, the variable α was programmed to switch from 50 to 0 after 5 seconds, whereupon it would serve as a turn-off signal for the calcium uptake, initiating the exponential decay in the equation (5) (Figure 7).

React	ions Species	Modul	e quantities	Functions	Events	Compartme	ents	Equations	Model pro	perties	Ports
#	Name (opt)		Trigger exp	ression	Actio	ns	Delay	/	DlyCalc	Notes	
1	Calcium trigge	er	Ca>0.04		a=0		5		\checkmark		
2											

Figure 6. A screenshot of the menu "events" to introduce an exponential decay in the equation (5).



Figure 7. The kinetics of oscillatory calcium influx described by the set of equations (5).

Once completed, Module 1 is of 2-inputs/1-output configuration, wherein Ca1 and Ca2 are the inputs as Ca is the output that stimulates the phosphorylation of Ca2+/calmodulin-dependent protein kinase II (CaMKII) in Module 2 (*vide infra*).

3.2. Module 2: Calmodulin Module

Module 2 is devoted to the chemistry of CaMKII, on the grounds that this kinase is believed to impact the fluctuations of the size and strength of neuronal connection [15]. In the context of LTP, such fluctuations are referred to as "synaptic plasticity", which controls the extent of communication amongst neurons [16,17].

Being a determinant of how we remember, CaMKII is a serine/threonine kinase, which mediates Ca^{2+} -dependent phosphorylation of many neuronal targets. The crystal structure of this enzyme reveals that the holoenzyme is made up of 8-10 of two subunits, 52-kDa α and 60-kDa β . These subunits are arranged in the form of two rings, which pivot between an active and inactive conformation [18]. Under basal conditions, the kinase activity of CaMKII is blocked by the binding of the autoinhibitory domain to the catalytic domain. With the calcium transient, calcium/calmodulin (Ca²⁺/CaM) binds to the autoinhibitory domain and restores the kinase activity of CaMKII by removing it. Later on, this induces autophosphorylation of Thr-286 and phosphorylation of other numerous targets in downstream reactions, which eventually leads up to LTP [19].

Given its significance, many models have been proposed to accurately express and understand how this kinase responds to intracellular calcium [20-23]. Amongst them, the one developed by Zhabotinsky is particularly intriguing as it reported a rather simple one to elucidate the mechanism of autophosphorylation of CaMKII in the presence of a calcium-dependent phosphatase [20]. Therefore, this model was of choice for the design of Module 2.

To make the mechanism of this enzyme more intelligible, Zhabotinsky made certain assumptions. For instance, this model fully ignores any difference between α and β subunits, treating CaMKII as a decamer of a single type of subunits. Then, the activation of subunits by calcium obeys the Hill equation, and the binding of Ca²⁺/CaM and the phosphatase to a subunit is independent of the phosphorylation state of other subunits.

Quite naturally, this manuscript will manifest certain key aspects of this model; for further details, the reader should refer to the original paper. To begin with, CaMKII is portrayed as a single ring of these subunits, in which the cascade of phosphorylation propagates in one direction, as

originally proposed by Hanson and Schulman [24]. In the initiation step of phosphorylation reactions, (Ca2+)4CaM binds to two neighbor subunits. Then, the first subunit phosphorylates the second one in the clockwise direction, as shown by the reactions (9-12).

$$4Ca^{2+} + CaM \rightleftharpoons C \tag{9}$$

$$P_0 + C \rightleftharpoons P_0 C \tag{10}$$

$$P_0C + C \rightleftharpoons P_0C_2 \tag{11}$$

$$P_0C_2 \to P_1C_2 \tag{12}$$

where C represents $(Ca^{2+})_4CaM$ complex. P0 and P1 are unphosphorylated and 1-fold phosphorylated haloenzyme, respectively.

Next is the incorporation of the dephosphatase activity. There are four protein phosphatases known to act upon CaMKII-P [25], however; this model solely deals with PP1, given that this is the only protein that dephosphorylates CaMKII in the postsynaptic densities [25,26]. In this regard, it was demonstrated that the activity of PP1 is regulated by Ca²⁺/CaM via inhibitor 1, calcineurin (CaN), and cAMP-dependent protein kinase (PKA). Of note, inhibitor 1 (I1) is phosphorylated by PKA and dephosphorylated by Calcineurin; once phosphorylated, I1 deactivates PP1 [27,28]. In terms of LPT, I1 is of paramount significance as its role renders this element a regulatory factor that links the calcium influx to synaptic plasticity [29]. Overall, the model of autophosphorylation of CaMKII in the presence of PP1 is given with the following equations (the constants c1-c4 and the parameters therein are defined in Table 2, and Table 3, respectively).

$$\frac{dP_0(t)}{dt} = -v_1 + v_3 P_1 \tag{13}$$

$$\frac{dP_1(t)}{dt} = v_1 - v_3 P_1 - v_2 P_1 + c_1 v_3 P_2 \tag{14}$$

$$\frac{dP_2(t)}{dt} = v_2 P_1 - c_1 v_3 P_2 - c_2 v_2 P_2 + c_3 v_3 P_3$$
(15)

$$\frac{dP_3(t)}{dt} = c_1 v_2 P_2 - c_2 v_3 P_3 - c_3 v_2 P_3 + c_4 v_3 P_4 \tag{16}$$

$$\frac{dP_4(t)}{dt} = c_1 v_2 P_3 - c_2 v_3 P_4 - c_3 v_2 P_4 + c_4 v_3 P_5 \tag{17}$$

$$\frac{dP_5(t)}{dt} = c_1 v_2 P_4 - c_2 v_3 P_5 - c_3 v_2 P_5 + c_4 v_3 P_6 \tag{18}$$

$$\frac{dP_6(t)}{dt} = c_1 v_2 P_5 - c_2 v_3 P_6 - c_3 v_2 P_6 + c_4 v_3 P_7 \tag{19}$$

$$\frac{dP_7(t)}{dt} = c_1 v_2 P_6 - c_2 v_3 P_7 - c_3 v_2 P_7 + c_4 v_3 P_8$$
(20)

$$\frac{dP_8(t)}{dt} = c_1 v_2 P_7 - c_2 v_3 P_8 - c_3 v_2 P_8 + c_4 v_3 P_9$$
(21)

$$\frac{dP_9(t)}{dt} = c_1 v_2 P_8 - c_2 v_3 P_9 - v_2 P_9 + c_3 v_3 P_{10}$$
(22)

$$\frac{dP_{10}(t)}{dt} = v_2 P_9 - c_1 v_3 P_{10} \tag{23}$$

$$\frac{de_p(t)}{dt} = -k_3 I e_p + k_4 (e_{p0} - e_p)$$
(24)

$$\frac{dI(t)}{dt} = -k_3 I e_p + k_4 (e_{p0} - e_p) + v_{PKA} I_0$$

$$-\frac{V_{CaN}([Ca]/K_{H2})^3 I}{1 + ([Ca]/K_{H2})^3}$$
(25)

where the rate constants v1, v2, and v3 are:

$$v_1 = \frac{10k_1([Ca]/K_{H1})^8 P_0}{(1 + ([Ca]/K_{H1})^4)^2}$$
(26)

$$v_{2} = \frac{k_{1}([Ca]/K_{H1})^{4}P_{1}}{(1 + ([Ca]/K_{H1})^{4}}$$
(27)

$$v_{3} = \frac{k_{2}e_{p}}{K_{M} + \sum_{i=1}^{10} iP_{i}}$$
(28)

Equation	c ₁	c ₂	c ₃	c ₄
21	2	-	-	-
22	2	1.8	3	-
23	1.8	3	2.3	4
24	2.3	4	2.7	5
25	2.7	5	2.8	6
26	2.8	6	2.7	7
27	2.7	7	2.3	8
28	2.3	8	1.8	9
29	1.8	9	10	-
30	10	-	-	-

Table 2. The constants c_1 - c_4 in equations (13-23).

3.3. The Model

The results of the simulation indicate that this cascade proceeds as a response to calcium, which concurrently orchestrates phosphorylation and dephosphorylation reactions. In this regard, it is observed that when Ca2+ firstly triggers n-fold phosphorylation reactions of CaMKII subunits, as a result of which unphosphorylated holoenzyme (P0) is fully consumed and evokes the synthesis of holoenzyme with a higher degree of phosphorylation. Mechanistically, the kinetics of each reaction exhibit bell-curve-like behavior in that each signal sequentially reaches its maximum, and then, undergoes "dephosphorylation", forming the cascade of "n-fold dephosphorylation". Thus, this model consists of two series of reactions merged to form a cascade; n-fold phosphorylation reactions ($P_0 \rightarrow P_1 \rightarrow P_2 \dots P_8 \rightarrow P_9$), and n-fold dephosphorylation reactions ($P_{10} \rightarrow P_9 \rightarrow P_8 \dots P_1 \rightarrow P_0$).

From standpoint of LTP, the phosphorylation of subunits and kinase activity in post-synapsis come to mean the signal, which is essential to catalyze the other downstream events in LTP. After fulfilling its role, this signal is shut down through dephosphorylation of holoenzymes, as discussed before. However, it is worth noting that according to this model, the concentration of subunits doesn't fully return to their initial values, indicating partial dephosphorylation, rather than complete dephosphorylation. One may speculate that this stems from the dynamic of phosphatase and inhibitor-1 cycle, which should govern the turn-off signal. To elaborate, calcium accumulating in the post-synapsis yields the concurrent synthesis of free inhibitor 1 (I1, Figure 9) and phosphatase (Ep, Figure 10). This results in a temporary inhibition of phosphatase activity, sustaining the n-fold phosphorylation cascade, as seen in Figure 8. When the calcium influx begins

to decay at t=5 s, however; one would expect that the balance in Ep/ I1 should have favored phosphatase activity, whereby the subunits of CaMKII are thoroughly dephosphorylated. In contrast, it appears that the extent of phosphatase activity tends to decrease beyond t=5 s. Thus, the lack of sufficient phosphatase activity may be responsible for CaMKII subunits preserving their phosphorylated states and not fully turning the kinase activity off at t>5 s.



Figure 8. Kinetics of CaMKII subunit phosphorylation in response to calcium influx (the initial concentration of P_0 is 10 µm whereas that of other subunits are considered to be 0. K_m and KH₂ are 10 µM, 0.6 µM, respectively) [20].



Figure 9. Kinetics of I1 in the response to calcium influx.



Figure 10. Kinetics of E_p in the response to calcium influx (E_{p0} is 0.6 μ M) [20].

Parameter	Description	Value	Unit	Reference
e _k	Total concentration of CaMKII	0.1–30	μΜ	25
ep ₀	Total concentration of protein phosphatase	0.01–1.2	μΜ	20
Io	Concentration of free inhibitor 1	0.0, 0.1	μΜ	20
VCaN	V _{CaN} Activity of calcineurin divided by its Michaelis constant		s ⁻¹	20
VPKA	Activity of PKA divided by its Michaelis constant	1.0	s ⁻¹	20
Км	The Michaelis constant of protein phosphatase	0.4–20	μΜ	20
Кн1	The Ca ²⁺ activation Hill constant of CaMKII	4.0	μΜ	[30,31]
K _{H2}	The Ca ²⁺ activation Hill constant of calcineurin	0.3–1.4	μΜ	32
\mathbf{k}_1	The catalytic constant of autophosphorylation	0.5	s ⁻¹	33
k 2	The catalytic constant of protein phosphatase	2.0	s ⁻¹	20
k 3	k ₃ The association rate constant of the PP1 z I1P complex		μΜ	[20,28]
k4	The dissociation rate constant of the PP1•I1P complex	1×10 ⁻³	s ⁻¹	[20,28]

Table 3. Parameters of Module-2.

4. CONCLUSION

In summary, the results reveal that the model formulated hereby is an avenue to study the $Ca^{2+}/CaMKII$ network in the JigCell environment. Needless to say, this model will serve as a theoretical experimental environment, in which one can alter the parameter and effortlessly elucidate its impact on the process. Of course, the next step should be the validation and fine-tuning of such models through some experimental results; once done, a better understanding of memory formation will be obtained. In long term, more accurate models that are composed this way will aid the scientific community to understand how we learn and but also, how our brain fails us, as in neurodegenerative diseases.

One final remark concerns JigCell and COPASI. To this date, these tools have been utilized to analyze many complex systems (for instance, see [34]). However, their use in the context of LTP has not been precedented, to the best of our knowledge. Thus, this study will expand the repertoire of their applications. Within this scope, the development of new and more elaborative models is currently in progress, in our laboratories (see: https://onuralpturk8.wixsite.com/projectcerebra).

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