

# Syntheses of Hydrocarbazole Derivatives via Brønsted acid-Initiated Diels–Alder Cycloaddition/retro-Michael Addition Cascade Reaction of Azepino[4,5-*b*]indoles and Acrolein

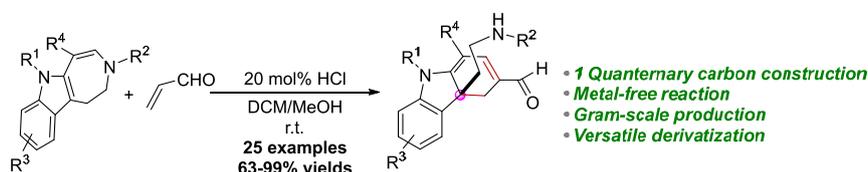
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Supporting Information Placeholder



**ABSTRACT:** A unique approach to hydrocarbazoles bearing an all-carbon quaternary center at C4a position was developed via a Brønsted acid-initiated Diels–Alder cycloaddition/retro-Michael addition cascade process from readily prepared azepino[4,5-*b*]indoles and commercially available acrolein. The method provided a range of hydrocarbazoles in good to excellent yields. The practicality of this transformation was demonstrated by scale-up experiment and various transformations to several hydrocarbazole derivatives.

Hydrocarbazoles with an all-carbon quaternary center at C4a position are the core frameworks existing in a variety of natural indole alkaloids and bioactive molecules.<sup>1</sup> Some of these molecules have attracted much attention from medicinal chemists and synthetic chemists on account of their various bioactivities and highly structural diversity.<sup>2-4</sup> For example, Wang and coworkers have developed a series of lead compounds bearing hydrocarbazole skeletons as resistance-modifying agents (RMAs) against methicillin-resistant *Staphylococcus aureus* (MRSA) that could selectively potentiate many  $\beta$ -lactam antibiotics (Of1)<sup>3a-3d</sup> and as anti-bacterial agent against both methicillin-susceptible *Staphylococcus aureus* (MSSA) and MRSA (Of4)<sup>4</sup>. Recently, Zu and coworkers also reported the enantioselective syntheses of simplified leads of Of1 and their function in potentiating  $\beta$ -lactam antibiotics<sup>3e</sup> (Figure 1).

In the past few years, due to the importance of assembly of the hydrocarbazoles bearing an all-carbon quaternary center at C4a position, many elegant synthetic strategies have been developed on the basis of dearomative cyclization,<sup>2e-2f,5</sup> Heck cyclization,<sup>6</sup> Fischer indolization,<sup>7</sup> aza-Michael addition<sup>8</sup> and cycloaddition of indole derivatives (Scheme 1).<sup>9</sup> Although these efforts have greatly enriched the toolbox for the construction of hydrocarbazoles, the development of an expeditious and efficient strategy for the preparation of compounds bearing hydrocarbazole scaffolds is urgently in need to develop more potent bioactive molecules. Previously we developed a strategy for the syntheses of azepino[4,5-*b*]indoles via cycloisomerization of

tryptamine-ynamides.<sup>10b</sup> The azepino[4,5-*b*]indoles contain diene moiety in their structures therefore they could serve as a diene component for [4 + 2] cycloaddition to generate hydrocarbazoles. In this study, a Diels–Alder cycloaddition/retro-Michael addition cascade strategy was developed aiming for the

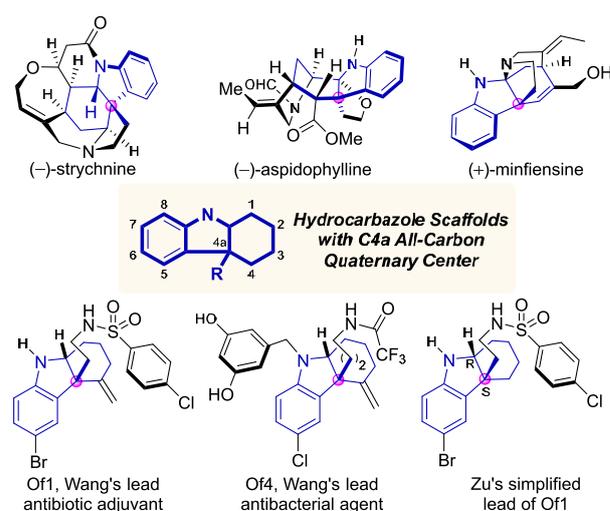
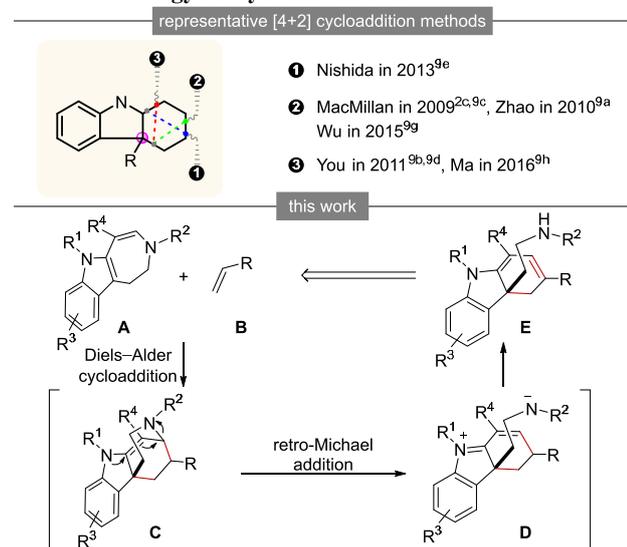


Figure 1. Representative Hydrocarbazole Natural Products and Lead Compounds.

preparation of hydrocarbazole scaffolds as a key molecular motif in natural products and potent RMAs. It was envisioned that a Diels–Alder cycloaddition of facilely prepared azepino[4,5-*b*]indoles A and commercially available dienophiles B would provide the highly congested tetracyclic intermediate C, followed by a sequent retro-Michael addition to release the strain of endocyclic structure to yield hydrocarbazole E after isomerization of intermediate D (Scheme 1).

### Scheme 1. Representative Methods and Our Cascade Cyclization Strategy to Hydrocarbazoles



We began the investigation with azepino[4,5-*b*]indole **1** as the model substrate, which was easily synthesized by following our previous procedures.<sup>10b</sup> To our delight, when azepino[4,5-*b*]indole **1** and acrolein **2** (3 equivalents) were treated with catalytic hydrochloric acid [20 mol%, 1 N HCl (aq)] in dichloromethane (DCM) at room temperature for 10 h, the proposed cascade process happened as expected, affording the desired hydrocarbazole product **3** in a yield of 16% (Table 1, entry 1). The structure of product **3** was determined by NMR spectroscopy and single-crystal X-ray diffraction (Table 1). This promising result prompted us to examine the parameters of the cascade process. First, different solvents such as methanol (MeOH), H<sub>2</sub>O, tetrahydrofuran (THF), acetonitrile (CH<sub>3</sub>CN) and toluene were screened under the catalysis of 20 mol% HCl. However, no obvious improvement in yield was observed (Table 1, entries 2–6). It was rationalized that the low yield might result from the heterogeneity of the reaction in those solvents. To address this issue, the mixed solvent system was considered in the following screening of conditions.<sup>11</sup> Fortunately, a combination of DCM and MeOH in a ratio of 3:1(v/v) was proved to be the best to this transformation (Table 1, entry 7). Next, screening of catalysts based on the optimal solvent system revealed that HCl was superior to other catalysts such as hydrobromic acid (HBr), trifluoroacetic acid (TFA), tribromoacetic acid (TBA), acetic acid (AcOH), *p*-toluenesulfonic acid (PTSA), ytterbium trifluoromethanesulfonate [Yb(OTf)<sub>3</sub>] and zinc chloride (ZnCl<sub>2</sub>) (Table 1, entries 8–14). Blank control experiment without any catalyst gave no product, indicating that a Brønsted acid was essential to this transformation (Table 1, entry 15). Investigation on the amounts of acrolein showed that 3 equivalents of acrolein afforded the most satisfactory result (Table 1, entries 16–17). Decreasing the catalyst loading to 10 mol% led to a lower yield of the desired product and increasing the catalyst loading to 30 mol% gave no obvious improvement in yield (Table 1, entries 18–19).

Overall, the optimal conditions of this cascade transformation were determined as stirring the azepino[4,5-*b*]indoles and acrolein (3 equivalents) under the catalysis of 20 mol% HCl in DCM/MeOH (3:1 v/v) at room temperature for 10 h.

**Table 1. Optimization of Reaction Conditions<sup>a</sup>**

entry	1	2	3	ORTEP of 3
catalyst <sup>b</sup>	solvent	yield (%) <sup>c</sup>		
1	HCl	DCM	16	
2	HCl	MeOH	40	
3	HCl	H <sub>2</sub> O	0 <sup>d</sup>	
4	HCl	THF	10	
5	HCl	CH <sub>3</sub> CN	54	
6	HCl	toluene	4	
7	HCl	DCM/MeOH <sup>e</sup>	98	
8	HBr	DCM/MeOH <sup>e</sup>	53	
9	TFA	DCM/MeOH <sup>e</sup>	82	
10	TBA	DCM/MeOH <sup>e</sup>	0 <sup>d</sup>	
11	AcOH	DCM/MeOH <sup>e</sup>	10	
12	PTSA	DCM/MeOH <sup>e</sup>	9	
13	Yb(OTf) <sub>3</sub>	DCM/MeOH <sup>e</sup>	53 (74 <sup>f</sup> )	
14	ZnCl <sub>2</sub>	DCM/MeOH <sup>e</sup>	7 (51 <sup>f</sup> )	
15	-	DCM/MeOH <sup>e</sup>	0 <sup>d,f</sup>	
16	HCl	DCM/MeOH <sup>e</sup>	42 <sup>g</sup>	
17	HCl	DCM/MeOH <sup>e</sup>	98 <sup>h</sup>	
18	HCl <sup>i</sup>	DCM/MeOH <sup>e</sup>	71	
19	HCl <sup>j</sup>	DCM/MeOH <sup>e</sup>	97	

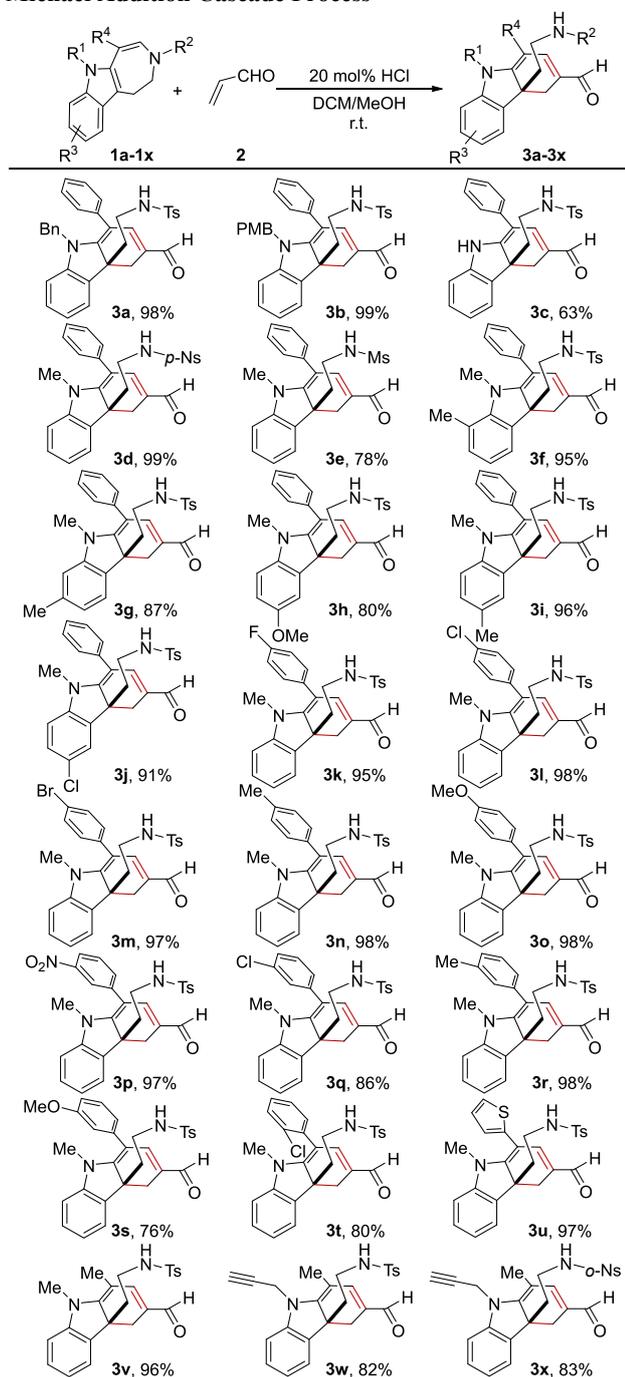
<sup>a</sup>Reaction was run at room temperature with 3 equiv. of acrolein.

<sup>b</sup>20 mol% catalyst was used in specified solvent (0.03 M). <sup>c</sup>Isolated yields. <sup>d</sup>Stirring for 48 h. <sup>e</sup>DCM/MeOH = 3:1 (v/v). <sup>f</sup>Stirring at 50 °C in sealed tube. <sup>g</sup>1 equiv. of acrolein was used. <sup>h</sup>5 equiv. of acrolein was used. <sup>i</sup>10 mol% catalyst was used. <sup>j</sup>30 mol% catalyst was used.

Under the optimal reaction conditions, various azepino[4,5-*b*]indoles were synthesized to examine the scope of this cascade reaction. Investigation on the substitutions of indolic nitrogen revealed that the substrates with Bn and PMB substitutions gave excellent yields, however, the substrate with no protecting group on nitrogen afforded the product with a relatively lower yield probably due to the instability of the product during isolation (Scheme 2, **3a–3c**). The examination of protecting groups on the nitrogen of the azepino core demonstrated that strong electron-withdrawing groups would facilitate the reaction (Scheme 2, **3d–3e**). Furthermore, the substrates with different substituents on the indole ring were examined. It was found that the electronic nature of the substituents had little influence on the yields of this reaction (Scheme 2, **3f–3j**). The scope of substituents on diene motif (R<sup>4</sup>) was also examined. The substrates with different aryl groups underwent smoothly to give the hydrocarbazole products in satisfactory yields (Scheme 2, **3k–3t**). The steric hindrance caused by *meta*-methoxyl and *ortho*-chloro substituents resulted in slightly lower yields of **3s** and **3t**.

This reaction was also extended to the substrates with heteroaryl or alkyl groups on the azepino core, which provided the target molecules in almost quantitative yields (Scheme 2, **3u-3v**). Finally, azepino[4,5-*b*]indoles with alkyl-substituted diene motif were investigated. Different protecting groups on indolic and azepino nitrogens were well-tolerated, thus allowing the facile synthesis of hydrocarbazole products in acceptable yields

**Scheme 2. Generality of Diels-Alder Cycloaddition/Retro-Michael Addition Cascade Process<sup>a,b</sup>**

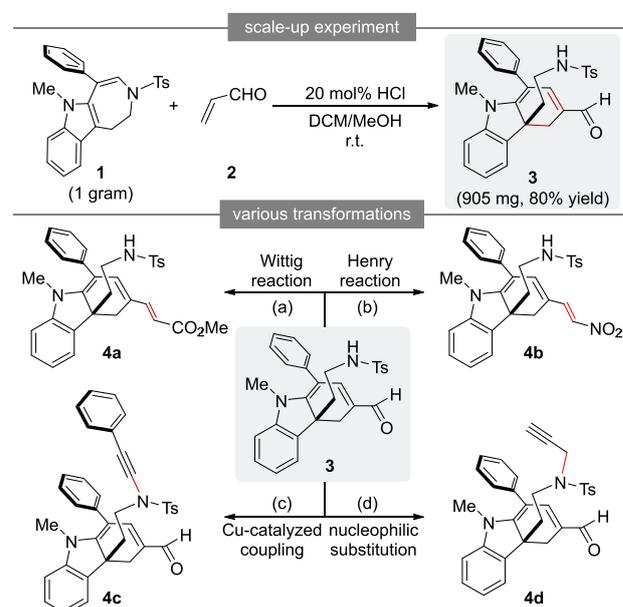


<sup>a</sup>Performed with **2a-2x** (1.0 mmol) and **1** (3.0 mmol) in the presence of a catalytic amount of HCl (1 N HCl, 0.2 mmol) in DCM/MeOH (3:1 v/v, 7 mL) at room temperature. The yields refer to the isolated products. <sup>b</sup>Ts = *p*-toluenesulfonyl, *p*-Ns = 4-nitrobenzenesulfonyl, Ms = methanesulfonyl, *o*-Ns = 2-nitrobenzenesulfonyl, Bn = benzyl, PMB = *p*-methoxybenzyl.

(Scheme 2, **3w-3x**).

To demonstrate the practicality of this robust methodology, scale-up synthesis using substrate **2** was performed, affording the hydrocarbazole product **3** in 80% yield. The following derivatizations were then attempted to explore the synthetic application of the product **3** by modifying the reactive aldehyde and sulfonamide moieties (Scheme 3). The Wittig reaction of product **3** underwent smoothly in toluene at 50 °C, providing the highly conjugated ester derivatives in a yield of 90%. The aldehyde moiety could also be transformed into unsaturated nitro compound by treating with nitromethane under the catalysis of ammonium acetate. In terms of the modification of the sulfonamide, a copper-catalyzed cross-coupling reaction of product **3** with phenylbromoethyne proceeded well to give versatile ynamide scaffold in 88% yield. Under basic condition, a propargyl group was readily introduced to the product **3** via a nucleophilic substitution. Such transformations demonstrated the versatile applications of the methodology.

**Scheme 3. Scale-Up Experiment and Further Derivatization of 3<sup>a</sup>**



<sup>a</sup>Reagent and conditions: (a) Ph<sub>3</sub>PCHCO<sub>2</sub>Me, toluene, 50 °C, 18 h, 90%; (b) NH<sub>4</sub>OAc, CH<sub>3</sub>NO<sub>2</sub>, 100 °C, 15 h, 93%; (c) phenylbromoethyne, CuSO<sub>4</sub>·5H<sub>2</sub>O, Cs<sub>2</sub>CO<sub>3</sub>, 1,10-phenanthroline, toluene, 90 °C, 12 h, 88%; (d) NaH, propargyl bromide, THF, 0~r.t., 91%.

In conclusion, a robust Brønsted acid-initiated Diels-Alder cycloaddition/retro-Michael addition cascade strategy from azepino[4,5-*b*]indoles and acrolein was developed. This methodology provided a facile entry to a series of hydrocarbazole derivatives bearing an all-carbon quaternary center at C4a position with a wide functional group tolerance. The synthetic utility of the current method was exemplified by scale-up experiment and various derivatizations. It is expected that the methodology is of great use in development of bioactive hydrocarbazole analogs as well as total syntheses of complex natural products with hydrocarbazole scaffolds.

**ASSOCIATED CONTENT**

**Supporting Information**

The Supporting Information is available free of charge on the ACS Publications website.

Experimental procedures, X-ray single crystal analyses, compound characterization data (PDF)

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### Author Contributions

All authors have given approval to the final version of the manuscript.

### Notes

The authors declare no competing financial interest.

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