

# Photocatalytic Cycloaddition Reaction of Triarylphosphines with Alkynes Forming Cyclic Phosphonium Salts

Yusuke Masuda,<sup>†</sup> Daichi Ikeshita, and Masahiro Murakami\*

Department of Synthetic Chemistry and Biological Chemistry, Kyoto University, Katsura, Kyoto 615-8510, Japan

<sup>†</sup> Present address: Department of Chemistry, Faculty of Science, Hokkaido University, Kita-ku, Sapporo, Hokkaido 060-0810, Japan.

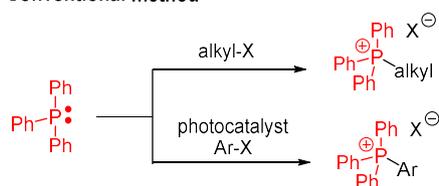
E-mail: murakami@sbchem.kyoto-u.ac.jp

1 Herein reported is a photocatalytic cycloaddition  
2 reaction of triarylphosphines with alkynes. Phosphonium  
3 salts of unique bicyclic structures are synthesized through a  
4 radical pathway under mild reaction conditions. The  
5 phosphonium salts are subjected to the Wittig olefination  
6 reaction to afford structurally interesting phosphine oxides.

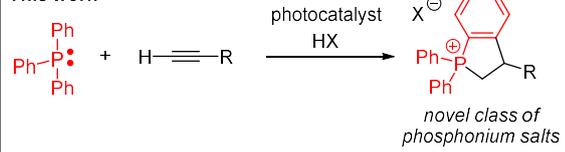
7 **Keywords:** Phosphonium salt • Alkyne • Photoredox  
8 **catalyst**

9 Phosphonium salts belong to an important class of  
10 compounds that are utilized as synthetic reagents,<sup>1</sup> organo-  
11 catalysts,<sup>2</sup> and ionic liquid,<sup>3</sup> among others.<sup>4</sup> The most  
12 conventional method for the synthesis of quaternary  
13 phosphonium salts is given by a nucleophilic substitution  
14 reaction of alkyl halides with tertiary phosphines.<sup>5</sup> An  
15 alternative preparative approach to phosphonium salts via a  
16 radical pathway recently appeared.<sup>6</sup> However, the  
17 phosphonium salts that are accessible by these methods are  
18 limited to those of acyclic structures. It presents a challenge  
19 to develop a facile method to synthesize phosphonium salts  
20 of cyclic structures from readily available starting  
21 substances. Herein reported is a photocatalytic  
22 cycloaddition reaction of triarylphosphines with alkynes.  
23 The phosphonium salts of unique bicyclic structures were  
24 synthesized from those readily available starting materials.  
25

## Conventional method



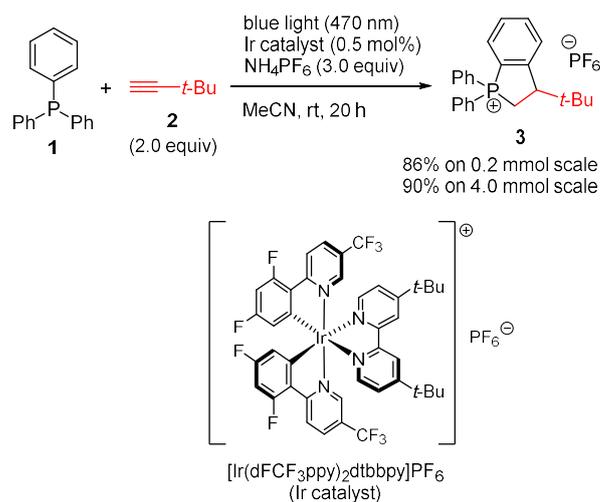
## This work



26  
27 **Figure 1.** Syntheses of phosphonium salts.

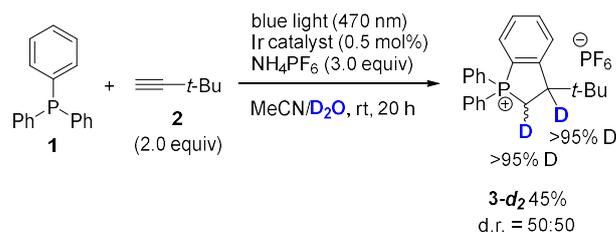
28 We recently reported a three component coupling  
29 reaction of triarylphosphines, alkenes, and water.<sup>7</sup> It was  
30 marked by dramatic skeletal changes involving C–P bond  
31 formation and dearomatization of the phenyl ring. We next  
32 examined an analogous reaction using an alkyne instead of  
33 an alkene; a mixture of triphenylphosphine (**1**), alkyne **2**

34 (2.0 equiv), ammonium hexafluorophosphate (NH<sub>4</sub>PF<sub>6</sub>, 3.0  
35 equiv), and an iridium photocatalyst  
36 ([Ir(dFCF<sub>3</sub>ppy)<sub>2</sub>dtbbpy]PF<sub>6</sub>, 0.5 mol%) in acetonitrile was  
37 irradiated with blue light (470 nm) for 20 h, and the reaction  
38 mixture was subjected to chromatographic purification on  
39 silica gel. To our surprise, the phosphonium salt having a  
40 bicyclic structure **3** was obtained in 86% isolated yield  
41 (Scheme 1). The structure of **3** was confirmed by an X-ray  
42 crystallographic analysis (CCDC 2084056). In a formal  
43 sense, [3 + 2]-type cycloaddition took place to construct the  
44 five-membered ring. Light, NH<sub>4</sub>PF<sub>6</sub>, and the iridium  
45 catalyst were all indispensable for the production of **3**.<sup>8</sup>  
46 When the reaction was scaled up to a 4.0 mmol scale, the  
47 product **3** could be isolated by recrystallization in place of  
48 chromatography, and its yield improved to 90%.  
49



50  
51 **Scheme 1.** Photocatalyzed cyclization of triphenylphosphine (**1**) and  
52 alkyne **2**.

53 Next, the reaction was conducted in the presence of  
54 D<sub>2</sub>O (1 mL). Two deuterium atoms were incorporated on  
55 the two ring carbons, each  $\alpha$  and  $\beta$  to phosphorus (Scheme  
56 2).  
57

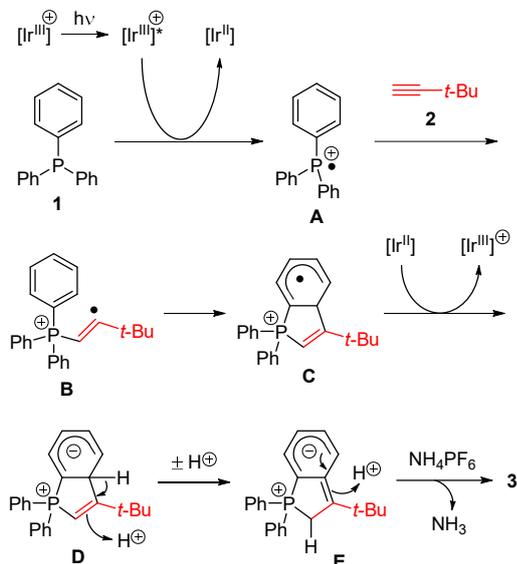


58

1 **Scheme 2.** Reaction in the presence of D<sub>2</sub>O.

2 We propose a reaction pathway shown in Scheme 3 to  
 3 explain the formation of **3** from **1** and **2**. Initially, the  
 4 iridium(III) catalyst absorbs light to get excited.  
 5 Triphenylphosphine (**1**) ( $E_{1/2}^{\text{red}} = +0.98$  V vs SCE in  
 6 CH<sub>3</sub>CN)<sup>9</sup> transfers a single electron to the excited  
 7 iridium(III) catalyst ( $E_{1/2}^{\text{red}}[\text{Ir(III)}^*/\text{Ir(II)}] = +1.21$  V vs SCE  
 8 in CH<sub>3</sub>CN),<sup>10</sup> generating radical cation **A** and an iridium(II)  
 9 species. The phosphine radical cation **A** adds across the C–  
 10 C triple bond of **2** to generate substituted alkenyl radical **B**,  
 11 which adds onto the phenyl ring to form the five-membered  
 12 ring. The resulting cyclohexadienyl radical **C** accepts a  
 13 single electron from the iridium(II) species to become the  
 14 cyclohexadienyl anion **D**, the anionic charge of which is  
 15 delocalized onto the attached cationic phosphorus, thereby  
 16 regenerating the iridium(III) catalyst. The following 1,3-  
 17 prototropic shift extends delocalization even further, giving  
 18 rise to the conjugated alkene **E**. Finally, aromaticity is  
 19 regained upon protonation at the carbon β to the phosphorus  
 20 atom to furnish the bicyclic phosphonium salt **3**.

21



22

23 **Scheme 3.** A proposed reaction pathway.

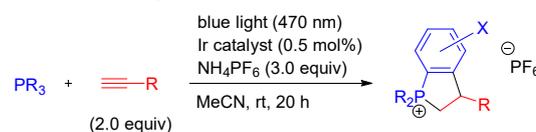
24 Various terminal alkynes were reacted with  
 25 triphenylphosphine (**1**). Propargylic alcohols afforded the  
 26 corresponding phosphonium salts **4-6** in good yields,  
 27 suggesting the tolerance of a hydroxy functionality. The  
 28 silylated phosphonium salts **7-9** were produced from silyl-  
 29 substituted alkynes in yields ranging from 49% to 83%.  
 30 Although internal alkynes such as 3-hexyne also  
 31 participated in the reaction, we failed to isolate the  
 32 corresponding phosphonium salts in a pure form because  
 33 some byproducts were inseparable.

34 Other triarylphosphines than **1** were also examined.  
 35 Functional groups such as fluoro (**11**) and methoxy (**12**)  
 36 groups were tolerated on the aromatic ring of the  
 37 triarylphosphines. Not only triarylphosphines but also

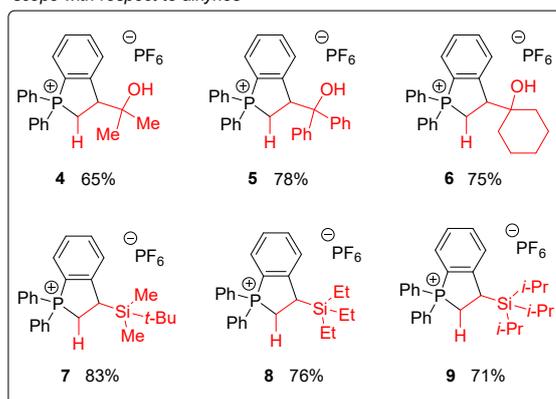
38 methylphenylphosphine participated in the cycloaddition  
 39 reaction with alkyne. The cyclic phosphonium salt **13** was  
 40 produced in 97% yield as a mixture of diastereomers.

41

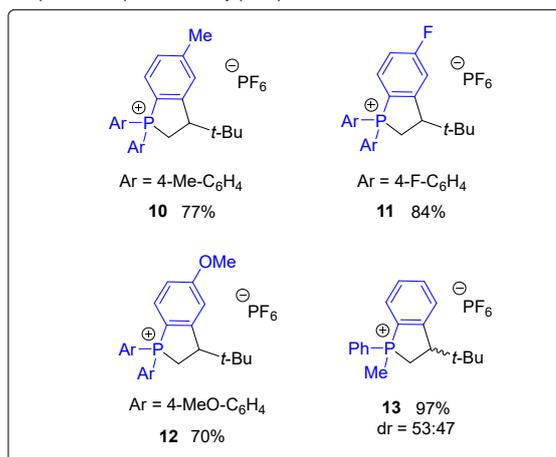
42 **Table 1.** Substrate scope.<sup>a</sup>



scope with respect to alkynes



scope with respect to tertiary phosphines



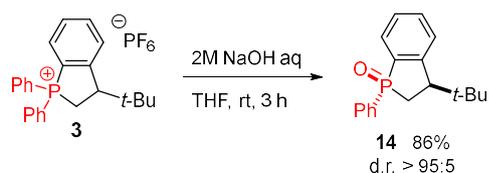
43

44 <sup>a</sup> Reaction conditions: tertiary phosphine (0.20 mmol), alkene (0.40  
 45 mmol, 2.0 equiv), NH<sub>4</sub>PF<sub>6</sub> (0.60 mmol, 3.0 equiv), Ir catalyst (0.001  
 46 mmol, 0.5 mol%), MeCN (3 mL), blue LEDs (470 nm), rt, 20 h.  
 47 Isolated yield.

48

49 It was possible to convert the produced phosphonium  
 50 salts to the corresponding phosphine oxides according to a  
 51 literature procedure;<sup>11</sup> the phosphonium salt **3** was treated  
 52 with NaOH in THF/water at room temperature for 3 h.  
 53 P(III)<sup>+</sup> was oxidized to P(V)=O and the phenyl substituent  
 54 on it was protonated to afford benzophospholane oxide **14** in  
 55 86% yield as a single diastereomer.

56



26  
27  
28  
29  
30  
31  
32  
33  
34  
35  
36  
37  
38

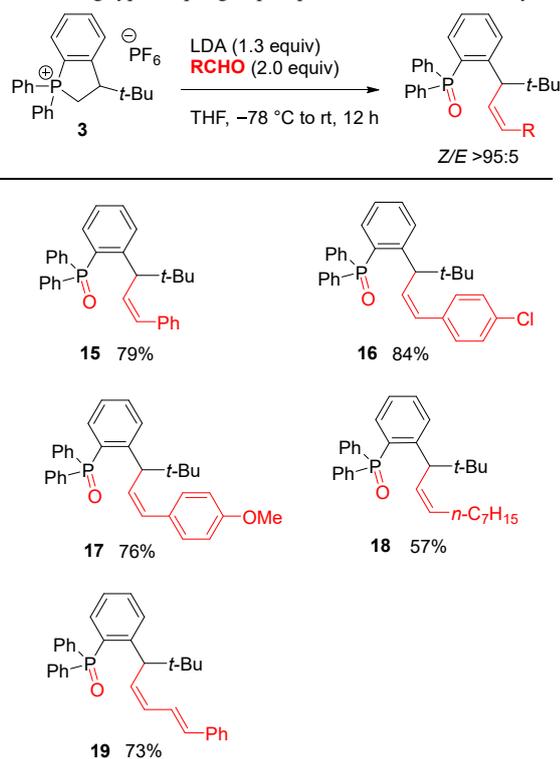
**Scheme 4.** Derivatization of the produced phosphonium salts.

3  
4  
5  
6  
7  
8  
9  
10  
11  
12  
13  
14  
15  
16  
17  
18  
19  
20

The phosphonium salts were used in the Wittig olefination reaction (Table 2). After a solution of the phosphonium salt **3** in tetrahydrofuran (THF) was treated with lithium diisopropylamide (LDA, 1.3 equiv) at  $-78\text{ }^{\circ}\text{C}$  for 1 h, benzaldehyde (2.0 equiv) was added to the generated ylide, and the reaction mixture was stirred at room temperature for 12 h. The corresponding alkenylated phosphine oxide **15** was obtained in 79% yield. A high *Z*-selectivity (*Z/E* > 95/5) was observed, as is often the case with the Wittig reaction of unstable ylides.<sup>12</sup> Not only benzaldehyde but also octanal served as an aldehyde to furnish the alkene **18** in 57% yield. Cinnamyl aldehyde produced 1,3-diene **19** in 73% yield. Of note is that the phosphine oxide moiety remaining in the product is amenable to further transformation.<sup>13</sup>

21  
22  
23  
24  
25

**Table 2.** Wittig-type coupling of phosphonium salt **3** with aldehydes.<sup>a</sup>



26  
27  
28  
29  
30  
31  
32  
33  
34  
35  
36  
37  
38  
39  
40  
41  
42  
43  
44  
45  
46  
47  
48  
49  
50  
51  
52  
53  
54  
55  
56  
57  
58  
59  
60  
61  
62  
63  
64  
65  
66  
67  
68  
69  
70  
71  
72  
73  
74  
75  
76  
77  
78  
79  
80  
81  
82  
83  
84  
85  
86  
87

<sup>a</sup> Reaction conditions: **3** (0.20 mmol), LDA (1.0 M in THF/hexane, 0.26 mmol, 1.3 equiv), THF (3 mL),  $-78\text{ }^{\circ}\text{C}$ , 1 h; then aldehyde (0.40 mmol, 2.0 equiv), rt, 12 h. Isolated yield.

26 In conclusion, the present study discloses a new synthetic  
27 approach to the phosphonium salts of unique cyclic  
28 structures starting from tertiary phosphines and alkynes,  
29 which are both readily available. The produced  
30 phosphonium salts are engaged in the Wittig olefination  
31 reaction to afford the corresponding phosphine oxides of  
32 interesting structures.

34 This work was supported by JSPS KAKENHI Grant  
35 Numbers 19K15562 and 21K14626 (Y.M.).

37 Supporting Information is available on  
38 [http://dx.doi.org/10.1246/cl.\\*\\*\\*\\*\\*](http://dx.doi.org/10.1246/cl.*****).

## 39 References and Notes

- 40 1 a) Modern Carbonyl Olefination—Methods and Applications  
41 (Ed: T. Takeda), Wiley-VCH, Weinheim, **2004**. b) M. Sakamoto,  
42 I. Shimizu, A. Yamamoto, *Chem. Lett.* **1995**, *24*, 1101–1102.  
43 2 For selected reviews, see: a) T. Werner, *Adv. Synth. Catal.* **2009**,  
44 *351*, 1469–1481. b) D. Enders, T. V. Nguyen, *Org. Biomol.*  
45 *Chem.* **2012**, *10*, 5327–5331. c) S. Liu, Y. Kumatabara, S.  
46 Shirakawa, *Green Chem.* **2016**, *18*, 331–341. (d) A. Golandaj, A.  
47 Ahmad, D. Ramjugernath, *Adv. Synth. Catal.* **2017**, *359*, 3676–  
48 3706. For selected examples, see: e) D. Uraguchi, S. Sakaki, T.  
49 Ooi, *J. Am. Chem. Soc.* **2007**, *129*, 12392–12393. f) R. He, C.  
50 Ding, K. Maruoka, *Angew. Chem. Int. Ed.* **2009**, *48*, 4559–4561.  
51 g) Y. Toda, Y. Komiyama, A. Kikuchi, H. Suga, *ACS Catal.*  
52 **2016**, *6*, 6906–6910.  
53 3 a) J. McNulty, J. J. Nair, S. Cheekoori, V. Larichev, A. Capretta,  
54 A. J. Robertson, *Chem. Eur. J.* **2006**, *12*, 9314–9322. b) H. Cao,  
55 L. McNamee, H. Alper, *J. Org. Chem.* **2008**, *73*, 3530–3534. c)  
56 Y. Kunugi, H. Hayakawa, K. Tsunashima, M. Sugiya, *Bull.*  
57 *Chem. Soc. Jpn.* **2007**, *80*, 2473–2475. d) C. G. Cassity, A.  
58 Miriafari, N. Mobarrez, K. J. Strickland, R. A. O'Brien, J. H.  
59 Davis, Jr., *Chem. Commun.* **2013**, *49*, 7590–7592.  
60 4 Selected examples, see: a) D. C. Rideout, A. Bustmante, J. Patel,  
61 *Int. J. Cancer* **1994**, *57*, 247. b) R. A. J. Smith, C. M. Porteous,  
62 A. M. Gane, M. P. Murphy, *Proc. Natl. Acad. Sci. U. S. A.* **2003**,  
63 *100*, 5407. c) K. Tanaka, H. Itoh, A. Nakashima, D. Wójcik, Z.  
64 Urbanczyk-Lipkowska, *Bull. Chem. Soc. Jpn.* **2009**, *82*, 489–493.  
65 d) W. Wan, X. Yang, R. C. Smith, *Chem. Commun.* **2017**, *53*,  
66 252–254. e) A. H. Nahlé, T. J. Harvey, F. C. Walsh, *J. Alloys*  
67 *Compd.* **2018**, *765*, 812–825. f) R. D. Dolewski, M. C. Hilton, A.  
68 McNally, *Synlett*, **2018**, *29*, 8–14.  
69 5 a) T. Migita, T. Nagai, K. Kiuchi, M. Kosugi, *Bull. Chem. Soc.*  
70 *Jpn.* **1983**, *56*, 2869–2870. b) D. Marcoux, A. B. Charette, *Adv.*  
71 *Synth. Catal.* **2008**, *350*, 2967–2974. c) D. Marcoux, A. B.  
72 Charette, *J. Org. Chem.* **2008**, *73*, 590–593. d) E. Rémond, A.  
73 Tessier, F. R. Leroux, J. Bayardon, S. Jugé, *Org. Lett.* **2010**, *12*,  
74 1568–1571. e) W. Huang, C.-H. Zhong, *ACS Omega* **2019**, *4*,  
75 6690–6696.  
76 6 a) A. F. Fearnley, J. An, M. Jackson, P. Lindovska, R. M.  
77 Denton, *Chem. Commun.* **2016**, *52*, 4987–4990. b) D. I.  
78 Bugaenko, A. A. Volkov, M. V. Livantsov, M. A. Yurovskaya,  
79 A. V. Karchava, *Chem. Eur. J.* **2019**, *25*, 12502–12506. c) P.  
80 Beatrice Arockiam, U. Lennert, C. Graf, R. Rothfelder, D. J.  
81 Scott, T. G. Fischer, K. Zeitler, R. Wolf, *Chem. Eur. J.* **2020**, *26*,  
82 16374–16382. d) V. V. Levin, A. D. Dilman, *Chem. Commun.*  
83 **2021**, *57*, 749–752.  
84 7 Y. Masuda, H. Tsuda, M. Murakami, *Angew. Chem. Int. Ed.*  
85 **2021**, *60*, 3551–3555.  
86 8 Other proton sources than  $\text{NH}_4\text{PF}_6$ , such as ammonium chloride  
87 and acetic acid, gave **3** in lower yields.

- 1 9 G. Pandey, D. Pooranchand, U. T. Bhalerao, *Tetrahedron* **1991**,  
2 47, 1745–1752.
- 3 10 M. S. Lowry, J. I. Goldsmith, J. D. Slinker, R. Rohl, R. A. Pascal,  
4 Jr., G. G. Malliaras, S. Bernhard, S. *Chem. Mater.* **2005**, *17*,  
5 5712–5719.
- 6 11 For hydrative dearylation of arylphosphonium salts, see: a) W. E.  
7 McEwen, K. F. Kumli, A. Blade-Font, M. Zanger, C. A.  
8 VanderWerf, *J. Am. Chem. Soc.* **1964**, *86*, 2378–2384. b) J. R.  
9 Corfield, S. Trippett, *J. Chem. Soc. Chem. Commun.* **1970**,  
10 1267–1276. c) A. Schnell, J. C. Tebby, *J. Chem. Soc. Perkin*  
11 *Trans. I* **1977**, 1883–1886.
- 12 12 P. A. Byrne, D. G. Gilheany, *Chem. Soc. Rev.* **2013**, *42*, 6670–  
13 6696.
- 14 13 a) J. M. Muchowski, M. C. Venuti, *J. Org. Chem.* **1981**, *46*, 459–  
15 461. b) I. Yamamoto, S. Tanaka, T. Fujimoto, K. Ohta, *J. Org.*  
16 *Chem.* **1989**, *54*, 747–750.
- 17

**NOTE** The diagram is acceptable in a colored form. Publication of the colored G.A. is free of charge.

For publication, electronic data of the colored G.A. should be submitted. Preferred data format is EPS, PS, CDX, PPT, and TIFF.

If the data of your G.A. is "bit-mapped image" data (not "vector data"), note that its print-resolution should be 300 dpi.

You are requested to put a brief abstract (50-60 words, one paragraph style) with the graphical abstract you provided, so that readers can easily understand what the graphic shows.

Graphical Abstract	
Textual Information	
A brief abstract (required)	Herein reported is a photocatalytic cycloaddition reaction of triarylphosphines with alkynes. A radical pathway induced by a photocatalyst enables the ready synthesis of unique bicyclic phosphonium salts under mild reaction conditions. The produced phosphonium salt serves as a versatile intermediate to access to structurally varied phosphine oxides through the Wittig olefination reaction.
Title(required)	Photocatalytic Cycloaddition Reaction of Triarylphosphines with Alkynes Forming Cyclic Phosphonium Salts
Authors' Names(required)	Yusuke Masuda, Daichi Ikeshita, and Masahiro Murakami*
Graphical Information	
<p>The reaction scheme illustrates the synthesis of a phosphine oxide derivative. It begins with a triarylphosphine (represented by a benzene ring attached to a phosphorus atom bonded to three phenyl groups, Ph<sub>3</sub>P) reacting with an alkyne (represented by a triple bond with an R group). The reaction is catalyzed by light and a photocatalyst, with NH<sub>4</sub>PF<sub>6</sub> as a counterion. This step yields a bicyclic phosphonium salt intermediate, where the phosphorus atom is part of a fused ring system and carries a positive charge, with a PF<sub>6</sub><sup>-</sup> counterion. The second step involves Wittig olefination using 1) LDA and 2) R'CHO, resulting in the final product: a phosphine oxide where the phosphorus atom is double-bonded to an oxygen atom and single-bonded to a carbon atom that is part of an alkene system, with an R' group attached to the alkene.</p>	