

Gold Nanoclusters as Nano-Antibiotic Auranofin Analogues

William Ndugire,¹ N.G. Hasitha Raviranga,¹ Jingzhe Lao,¹ Olof Ramström,*^{1,2} Mingdi Yan*¹

¹Department of Chemistry, University of Massachusetts Lowell, One University Ave., Lowell, MA 01854, USA; ²Department of Chemistry and Biomedical Sciences, Linnaeus University, SE-39182 Kalmar, Sweden

Abstract. Auranofin, a gold(I)-complex with tetraacetylated thioglucose and triethylphosphine ligands, is an FDA-approved drug as an anti-inflammatory aid in the treatment of rheumatoid arthritis. In repurposing auranofin for other diseases, it was found that the drug showed significant activity against Gram-positive bacteria but was inactive against Gram-negative bacteria. Herein, we report the design and synthesis of gold nanoclusters (AuNCs) based on the structural motif of auranofin. Phosphine-capped AuNCs were synthesized and glycosylated, yielding auranofin AuNC analogues with mixed phosphine/thioglucose ligand shells. These AuNCs were active against both Gram-negative and Gram-positive bacteria, including a panel of resistant ESKAPE pathogens. Notably, an auranofin analogue, a mixed-ligand 1.6 nm AuNC (**4b**) was ~4 times more active than auranofin against *Pseudomonas aeruginosa*, while exhibiting 24 times lower toxicity against human A549 cells. The enhanced antibacterial activity of these AuNCs was characterized by a greater uptake of Au by the bacteria compared to Au^I-complexes (20% for AuNC **4b**). Additional factors include increased oxidative stress, moderate inhibition of thioredoxin reductase (TrxR), and DNA damage. Most intriguingly, the AuNCs were not affected by the bacterial outer membrane (OM) barrier or by extracellular proteins. This contrasts with Au^I-complexes like auranofin that are susceptible to protein binding and hindered by the OM barrier.

INTRODUCTION

The staggering rate of antimicrobial resistance is one of the foremost crises facing global public health. A recent report by the Centers for Disease Control and Prevention (CDC) estimated that up to 3 million people suffered from infections caused by antibiotic-resistant bacteria in 2019.¹ Besides the rise in community-acquired infections, the rapid increase in nosocomial infections threatens the safety of in-patient medical procedures. In 2017, up to 32,600 cases of multidrug-resistant (MDR) *Pseudomonas aeruginosa* infection occurred in hospitals in the US, resulting in 2,700 deaths.¹ As a result, new therapies are needed, especially approaches that can offer alternative mechanisms to combat the antimicrobial resistance observed against traditional antibiotics. A convenient strategy that is becoming more widespread is the redirection of non-antibiotic drugs into the antimicrobial pipeline - so-called drug repurposing. Repurposing drugs is associated with reduced regulatory costs and time to market.^{2,3}

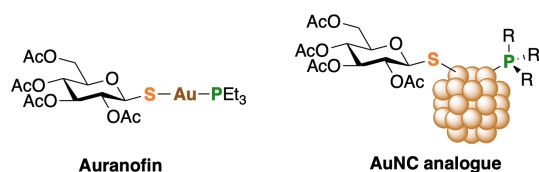


Figure 1: Structures of auranofin and AuNC analogue

Auranofin is an FDA-approved drug previously used as an anti-inflammatory drug in the treatment of rheumatoid arthritis (**Figure 1**). It has since been tested for other diseases, showing antibacterial,^{4–11} antifungal,^{12–14} antiparasitic,^{15,16} antiviral,^{17–21} and antitumor²² activities. The main mode of action (MOA) of auranofin has been suggested to be inhibition of thioredoxin reductase (TrxR).^{23,24} The thioredoxin (Trx) system is critical in maintaining the redox homeostasis by controlling the intracellular disulfide/thiol concentration in both bacteria and mammalian cells.^{25,26} Auranofin has shown high activity against Gram-positive bacteria but is inactive against Gram-negative strains.^{9,10,27} The lack of Gram-negative activity has been attributed to the outer membrane (OM) barrier, leading to reduced drug uptake.²⁷ It has also been suggested that the glutathione/glutaredoxin (GSH/Grx) system present in Gram-negative bacteria counteract the inhibition of the Trx/TrxR system, thereby reducing the oxidative stress level.²⁸

Gold nanoclusters (AuNCs) are a class of gold nanoparticles that contain up to several hundred Au atoms and are 2 nm or less in diameter.^{29,30} Similar to other nanomaterials, AuNCs can provide multivalent presentation of attached ligands within close proximity. The multivalent effect often leads to a marked increase in binding affinity by several orders of magnitude.³¹ This property has been exploited, for example, by conjugating antibiotics such as vancomycin, daptomycin, bacitracin, surfactin, and beta-lactams to AuNCs to enhance their activities.^{32–37} Another common approach is to conjugate charged species, particularly cationic ligands, to AuNCs.^{38–41} The resulting positively-charged AuNCs disrupt the negatively-charged bacterial membrane, leading to leakage of intracellular contents and eventual cell death. In addition, AuNCs have been shown to increase the concentration of reactive oxygen species (ROS) in bacteria and interrupt other metabolic pathways.⁴²

Common AuNCs are generally composed of a thiolate-coordinated ligand shell around the gold core, largely due to the relative ease in preparation and the high stability of thiolated AuNCs.²⁹ In contrast, reports of phosphine-functionalized AuNCs for antibacterial use are rare. However, phosphine-coordinated Au^I-complexes are relatively active and have been proposed as potential candidates as antimalarial, anticancer, and antiviral agents.^{43,44} For example, Jahnke-Dechent and coworkers reported the synthesis of “magic number” Au₅₅ (1.4 nm) and Au₈ (0.8 nm) nanoclusters functionalized with triphenylphosphine monosulfonate (TPPMS).⁴⁵ Both AuNCs were toxic to Gram-positive *S. aureus* and *S. epidermidis* (minimal inhibitory concentration (MIC): 25 μM) but not to Gram-negative *E. coli* or *P. aeruginosa* strains (MIC > 400 μM). However, thiolate-capped AuNCs of similar size (1.9 nm, AuroVist Nanoprobes 1102A) were inactive against all tested strains, suggesting higher activity for AuNCs carrying more labile phosphine ligands. Proposed MOAs include AuNC-mediated oxidative stress and the loss of membrane permeability.⁴⁵ The smaller 0.8 nm AuNCs caused significantly greater bacterial inhibition than the larger 1.4 nm AuNCs.

In this study, we designed AuNCs based on the structural motif of auranofin, consisting of an Au core coordinated with a tetraacetylated 1-thiogluco- and a phosphine ligand. We hypothesized that these AuNCs would harness the properties of both auranofin and AuNCs to afford enhanced antimicrobial activity. In addition, through optimization of the ratio of thiolate/phosphine ligands on the AuNCs, several critical parameters could be controlled and evaluated, including bacterial Au uptake, antimicrobial activity, and mammalian cytotoxicity.

RESULTS

Synthesis of AuNCs

Phosphine-functionalized AuNCs are typically chosen as precursors for the synthesis of

clusters with mixed phosphine/thiolate ligand shells owing to the favorable replacement of phosphine ligands with thiolates.⁴⁶ Of the potential clusters, “Au₅₅” (1.4 nm) was initially considered a candidate due to its demonstrated activity against Gram-positive bacteria, such as *S. aureus* and *S. epidermidis*.⁴⁵ However, the synthesis of this cluster involves the use of pyrophoric diborane gas,⁴⁷ and an improved synthesis protocol developed by Hutchison and coworkers was instead adopted.^{48,49} In this protocol, HAuCl₄ is reduced by NaBH₄ in the presence of PPh₃ in a biphasic water/toluene solvent using tetraoctylammonium bromide (TOAB) as a phase transfer catalyst, yielding ~1.5 nm AuNCs with a molecular formula of Au₁₀₁(PPh₃)₂₁Cl₅ (**1**, **Figure 2a**) as determined by XPS. Since auranofin possesses a PEt₃ ligand, exchange with PEt₃ was also attempted. However, the reaction resulted in a black product that was insoluble in organic or aqueous solvents, while yielding a clear solution in *aqua regia*, indicative of Au⁰ formation.⁴⁸ Excess PEt₃ also resulted in decomposition of cluster **1**. The difficulty of ligand exchange with PEt₃ is consistent with the scarcity of reported small aliphatic phosphine-functionalized clusters,⁵⁰ despite multiple reports of AuNCs possessing aromatic, bidentate, and alicyclic phosphine ligands.⁵¹ More sterically encumbered ligands are furthermore known to provide better stabilization of nanoclusters in solution.^{52,53} Consequently, we designed AuNCs passivated by aromatic phosphines and several candidates were prepared to evaluate the effects of glycosylation, aqueous solubility, and cluster size on the antibacterial activity (**Figure 2**).

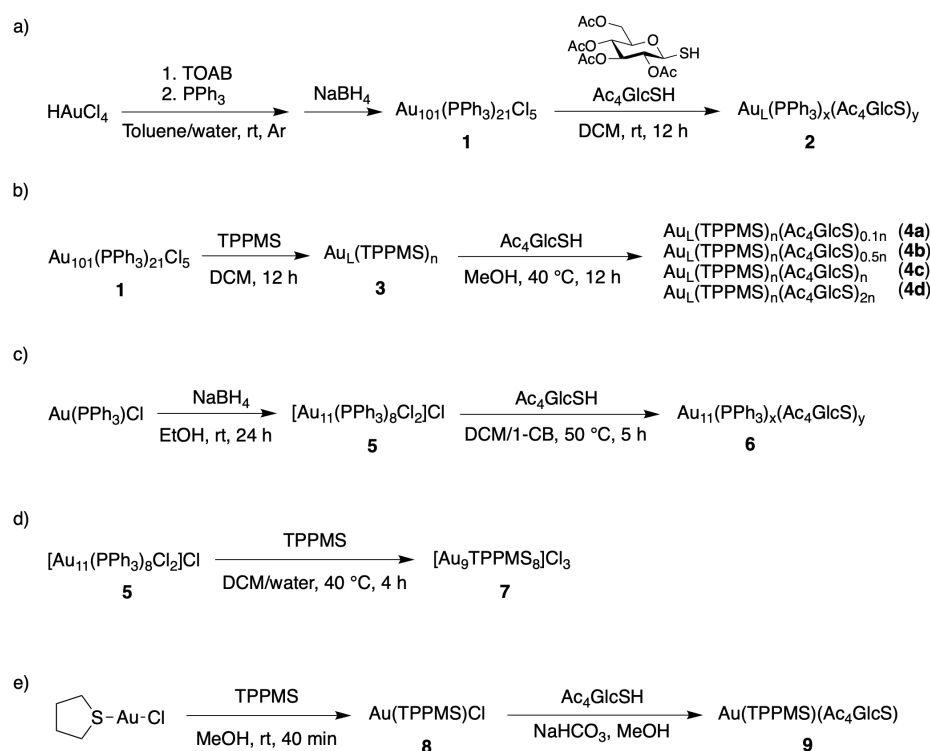


Figure 2: Synthesis of AuNCs and complexes: (a) Au_L(PPh₃)_x(Ac₄GlcS)_y (**2**), (b) Au_L(TPPMS)_n(Ac₄GlcS)_{0.1-2n} (**4a–4d**), (c) Au₁₁(PPh₃)_x(Ac₄GlcS)_y (**6**), (d) Au₉(TPPMS)₈Cl₃ (**7**), (e) Au(TPPMS)(Ac₄GlcS) (**9**). 1-CB = 1-chlorobutane.

Consistent with the literature report,⁴⁸ the ¹H NMR spectrum of cluster **1** showed broadened PPh₃ signals centered at ~7.2 ppm (Figure S15) as a result of the cluster formation and environmental heterogeneity of the ligands.^{54,55} A byproduct of the synthesis, Au(PPh₃)Cl, was observed as sharper signals at 7.51–7.58 ppm and could not be completely removed due to the metastability of the AuNC.⁴⁸

Glycosylation of cluster **1** was performed in dichloromethane (DCM) using 16 equivalents of Ac₄GlcSH, synthesized according to a literature protocol (Figure 2a).⁵⁶ After stirring at ambient temperature for 12 h, complete consumption of Ac₄GlcSH was confirmed. The solution was concentrated and purified by size exclusion chromatography to give cluster **2**, denoted as Au_L(PPh₃)_x(Ac₄GlcS)_y. The presence of the Ac₄GlcS and PPh₃ ligands was supported by the prominent acetyl resonance at ~2 ppm and the PPh₃ signals at ~7.5 ppm in the ¹H NMR spectrum (Figure S16). The byproduct Au(PPh₃)Cl was still present in cluster **2**, although at a lower concentration compared to the starting cluster **1**. Cluster **2** was slightly lighter in color, more soluble in alcohols, and more stable in DCM or CHCl₃ compared to cluster **1**, which rapidly decomposed to Au⁰ within a few hours.⁴⁸

Water-soluble analogues **4a-4d** were synthesized by first replacing PPh₃ in cluster **1** with sodium triphenylphosphine-3-sulfonate (TPPMS), prepared by sulfonation of PPh₃ using oleum (Figure 2b).⁵⁷ A 50-molar excess of TPPMS in a biphasic water/DCM solution resulted in the complete transfer of cluster **1** to the aqueous layer after 12 h at room temperature. Purification by size exclusion chromatography gave cluster **3** as a black powder. No free TPPMS ligand (-6 ppm), its oxide (37 ppm), or the Au^I impurity (32 ppm) was observed by ³¹P NMR. Similar to cluster **1**, the aromatic proton resonances of cluster **3** were significantly broadened in the ¹H NMR spectrum (Figure S17). Cluster **3** proved more stable than cluster **1** and could be heated in aqueous solution with no decomposition to Au⁰. Multiple freeze-drying cycles did not cause any discernible change in the ¹H NMR spectrum either. The improved colloidal stability is a result of electrostatic stabilization by replacing neutral with charged ligands and has also been observed in other nanoparticle systems.⁵⁸

Glycosylation of cluster **3** was carried out using different equivalents of Ac₄GlcSH, resulting in mixed-ligand AuNCs **4a-4d** with different carbohydrate/phosphine ratios. Cluster **3** was mixed with Ac₄GlcSH (3.2, 0.8, 0.2, and 0.05 w/w) and heated at 40 °C in MeOH for 12 h, at which time TLC confirmed that no free Ac₄GlcSH was present. The clusters were purified by size exclusion chromatography to give wispy dark brown solids. The ¹H NMR signals of the ligands were broadened and only the peaks of the acetyl groups were distinguishable at ~2 ppm (Figure S18). The higher the Ac₄GlcSH load, the more the TPPMS was replaced in the product (Figure S19). From the peak integrations of the acetyl and aromatic protons, the ratio of the Ac₄GlcS and TPPMS ligands in AuNCs **4a-4d** was estimated to be 2, 1, 0.5, and 0.1 (Table S1). Increased glycosylation resulted in reduced solubility in water and higher solubility in alcohols.

The size of cluster **4b** was analyzed by TEM (Figure S3). From a total of 561 particles, the particle core diameter was determined to be 1.6 ± 0.3 nm, concurring with literature reports that the ligand exchange process resulted in minimal changes in cluster size.⁴⁹ This was further supported by the UV-Vis absorption spectra of clusters **1**, **3**, and **4b**, showing no significant plasmon bands or absorptions resulting from quantized transitions as expected for smaller AuNCs (Figures S22-S26).⁵⁵

The smaller cluster Au₁₁(Ac₄GlcS)_x(PPh₃)_y (**6**) was synthesized from cluster [Au₁₁(PPh₃)₈Cl₂]Cl (**5**) through ligand exchange (Figure 2c). The parent cluster **5** was accessed using a literature procedure,⁵⁹ yielding a red-colored product that matched the ¹H and ³¹P NMR data reported (Figures S19-S20). Although being PPh₃-capped, this cluster is considerably more stable than cluster **1**, requiring elevated temperature to undergo ligand exchange.⁶⁰ The reduced exchangeability was attributed to the reduction in core size, where the steric hindrance around the core prevented the associative exchange process involving the incoming ligands and the Au core.⁶¹ To obtain a mixed-ligand shell of PPh₃ and Ac₄GlcS, cluster **5** was allowed to react with 11 equivalents of Ac₄GlcSH in DCM/1-chlorobutane at

50 °C, yielding Au₁₁(Ac₄GlcS)_x(PPh₃)_y (**6**) as a brown solid after 5 h.

Ligand exchange at cluster **5** gave AuNC [Au₉(TPPMS)₈]Cl₃ (**7**, **Figure 2d**). Attempts to exchange TPPMS with Ac₄GlcSH in methanol were unsuccessful despite the use of excess amounts of carbohydrate (up to 200 equiv.) and elevated temperature (up to 60 °C). In these reactions, a white precipitate was formed that consisted of TPPMS oxide (37 ppm, ³¹P NMR) and Au(TPPMS)Cl (32 ppm, ³¹P NMR).

An Au^I complex analogue of the clusters, Au(TPPMS)(Ac₄GlcS) (**9**), was synthesized by ligand exchange from the known Au^I complex Au(TPPMS)Cl (**8**, **Figure 2e**), synthesized following a literature protocol.⁶² Base-promoted replacement of the Cl ligand in complex **8** with the Ac₄GlcSH gave cluster **9**. Both complexes were soluble in water.

Antibacterial activity depends on ligand shell and cluster size

The MICs of the prepared clusters and complexes were first evaluated against a panel of antibiotic-resistant ESKAPE pathogens consisting of Gram-positive (*Staphylococcus aureus*, *Enterococcus faecalis*) and Gram-negative bacterial strains (*Klebsiella pneumoniae*, *Acinetobacter baumannii*, *Pseudomonas aeruginosa*, *Enterobacter cloacae*), as well as a Gram-negative strain (*E. coli* ATCC 25922) used as quality control (**Table 1**). The minimal bactericidal concentrations (MBCs) were also obtained for those entities showing MICs of <130 µg/mL [Au]. Consistent with auranofin, the clusters were more active against Gram-positive bacteria (*S. aureus* and *E. faecalis*) than the Gram-negative strains. The larger water-soluble clusters **4b** and **3** were active against Gram-negative strains, including the multidrug-resistant *A. baumannii* (MIC/MBC = 20/40 µg/mL) and *P. aeruginosa* NCTC13437 (MIC/MBC = 20/20 µg/mL). The activity against *P. aeruginosa* was higher than that of auranofin (MIC/MBC = 74/74 µg/mL [Au]).

Table 1: MIC^a (MBC) (µg/mL [Au]^b) of AuNCs and Au^I-complexes against ESKAPE pathogens and *E. coli* ATCC 25922.

	<i>A. baumannii</i> NCTC 13420	<i>P. aeruginosa</i> NCTC 13437	<i>E. cloacae</i> NCTC 13405	<i>K. pneumoniae</i> ATCC 700603	<i>S. aureus</i> USA300 JE2	<i>E. faecalis</i> ATCC 51299	<i>E. coli</i> ATCC 25922
Au ₁₀₁ (PPh ₃) ₂₁ Cl ₅ (1) ^c	50 (50)	>200	>200	>200	3.2 (3.2)	25 (25)	>200
Au ₁₀₁ (PPh ₃) _x (Ac ₄ GlcS) _y (2) ^c	>200	>200	>200	>200	3.2 (3.2)	12 (12)	>200
Au ₁₁ (TPPMS) _n (3) ^d	20 (40)	20 (20)	>160	>160	1.4 (1.4)	5.1 (5.1)	160
Au ₁₁ (TPPMS) _n (Ac ₄ GlcS) _{0.5n} (4b) ^d	20 (41)	20 (20)	>160	>160	1.2 (1.2)	5.1 (5.1)	>160
[Au ₁₁ (PPh ₃) ₈ Cl ₂]Cl (5) ^c	>130	>130	>130	>130	3.9 (3.9)	>130	>130
Au ₁₁ (PPh ₃) _x (Ac ₄ GlcS) _y (6) ^c	>130	>130	>130	>130	3.9 (3.9)	>130	>130
[Au ₉ (TPPMS) ₈]Cl ₃ (7) ^d	>92	>92	>92	>92	1.4 (1.4)	1.4 (1.4)	>92
Auranofin ⁶³	9.3 (9.3)	74 (74)	37 (37)	74 (74)	0.0079 (0.018)	0.039 (0.079)	4.7 (4.7)

^aPerformed at the Antimicrobial Screening Facility, Warwick University, UK; all tests repeated twice; ciprofloxacin and teicoplanin included as quality controls. ^b[Au] in AuNCs determined by inductively coupled plasma mass spectrometry (ICP-MS). ^cDissolved in DMSO. ^dDissolved in water.

The antipseudomonal activity of the water-soluble TPPMS-capped AuNCs and Au^I-complexes was further tested against the laboratory strain *P. aeruginosa* PAO1 and the quality control strain *P. aeruginosa* ATCC 27853. The impact of glycosylation on the antibacterial activity was probed using clusters **4a-4d** and the cluster size effect was evaluated by comparing cluster **3** with Au₉-cluster **7**, and Au^I complex **8**. Cluster **4b**, having a TPPMS:Ac₄GlcS ratio of 1:2, again showed the highest activity with MICs of 20 µg/mL for both *Pseudomonas* strains (**Table 2**). Either a higher or lower degree of glycosylation resulted in lower activity. Comparing the clusters having the same TPPMS ligand, the larger cluster **3** was more active than the smaller Au₉-cluster **7**.

Table 2: MIC^a (μg/mL [Au]^b) of TPPMS-capped AuNCs and Au^I-TPPMS complexes against *P. aeruginosa* and *S. aureus* (MRSA).

	<i>P. aeruginosa</i> PAO1	<i>P. aeruginosa</i> ATCC27853	<i>P. aeruginosa</i> NCTC13437 ^c
Au _L (TPPMS) _n (Ac ₄ GlcS) _{0.1n} (4a)	56	56	-
Au _L (TPPMS) _n (Ac ₄ GlcS) _{0.5n} (4b)	20	20	20
Au _L (TPPMS) _n (Ac ₄ GlcS) _n (4c)	47	93	-
Au _L (TPPMS) _n (Ac ₄ GlcS) _{2n} (4d)	>100	>100	-
Au _L (TPPMS) _n (3)	20	40	20
[Au ₉ (TPPMS) ₈]Cl ₃ (7)	46	>46	>92
Au(TPPMS)Cl (8)	21	40	-
Au(TPPMS)(Ac ₄ GlcS) (9)	>27	>27	-

^aAll compounds dissolved in water; MIC values represent mode of 4 trials; ciprofloxacin and gentamicin included as quality controls. ^b[Au] in AuNCs determined by ICP-MS. ^cPerformed at the Antimicrobial Screening Facility, Warwick University, UK; repeated twice.

Clusters display lower mammalian cell toxicity than auranofin

The effectiveness and safety of a drug can be evaluated by the selectivity index (SI), expressed as the ratio of cytotoxicity and activity. Auranofin has a half-maximal cytotoxic concentration (CC₅₀) of 2.2 ± 0.4 μg/mL against A549 human lung cancer cells (ATCC CCL-185). As auranofin has low activity against *P. aeruginosa*, this leads to a low SI of 0.029 (= CC₅₀/MIC, **Table 3**). Both the glycosylated cluster **4b** (Figure S4) and the non-glycosylated cluster **3** (Figure S5) were less toxic than auranofin towards A549 cells, having a CC₅₀ of 53 ± 7 μg/mL and 42 ± 10 μg/mL, respectively (**Table 3**). The SI of cluster **4b** (SI = 2.6) was slightly higher than for cluster **3** (SI = 2.1), and the two clusters were 90 and 72 times better than auranofin.

Table 3: CC₅₀^a (μg/mL [Au]^b) against A549 cells and MIC against *P. aeruginosa* of Au_LNCs and auranofin.

	A549 ATCC CCL-185	<i>P. aeruginosa</i> NCTC13437	Selectivity index
Au _L (TPPMS) _n (Ac ₄ GlcS) _{0.5n} (4b)	53 ± 7	20	2.6
Au _L (TPPMS) _n (3)	42 ± 10	20	2.1
Auranofin	2.2 ± 0.4	74	0.029

^aMean ± SD of two trials, each done in triplicate. ^b[Au] in AuNCs determined by ICP-MS.

AuNCs show high cellular uptake

The activity of most antibiotics against *P. aeruginosa* and other Gram-negative bacteria is to a large extent determined by their ability to overcome the OM barrier and accumulate inside bacteria.^{64,65} This uptake efficiency could be a source of the observed disparity in bacterial killing between different gold clusters and complexes. To test this, uptake experiments were carried out by analyzing the intracellular gold content using ICP-MS. Bacteria were incubated with the different gold compounds for 4 h, washed with PBS to remove excess compounds, digested with *aqua regia*, and the Au content measured. Candidates include TPPMS-capped cluster **3**, Au₉-cluster **7** and Au^I complex **8** to test the size-dependent uptake, and cluster **4b** for the impact of glycosylation (**Figure 3**). A significantly higher uptake was observed with the larger clusters **4b** (20 ± 3.5%) and **3** (26 ± 1.9%) compared to the smaller cluster **7** (3.7 ± 3.6%) and the Au^I complex **8** (6.3 ± 5.5%).

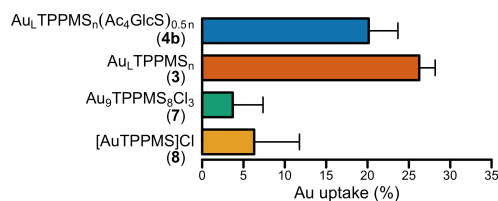


Figure 3: Uptake of Au of AuNCs and Au^I complex by *P. aeruginosa* PAO1. Each data was an average of 3 independent trials, presented as mean \pm SD. %Au uptake is the percent of Au taken up by bacteria vs. amount initially added. All data were adjusted to a final bacteria count of 4.0×10^9 CFU/mL.

The water-soluble TPPMS-capped AuNCs **4b** and **3** were stable, showing no signs of disintegration in D₂O for months. To further evaluate the stability, cluster **4b** and HAuCl₄ were dialyzed in PBS at pH 7.4 and 35 °C across a 3,500 MWCO dialysis membrane. The dialysates were continuously sampled over 26 h and the Au contents were determined by ICP-MS. The results show that less than 0.3% of the gold leached from the AuNC **4b** through the membrane, whereas the control sample reached a maximum level of 97% after 8 h (Figure S4).

Antibacterial activity of Au_LNCs is unaffected by extracellular proteins

It has been reported that the antibacterial activity of auranofin can be antagonized by the presence of extracellular proteins in the growth medium.⁹ Proteins, such as serum albumin, can react with gold complexes, for example through interactions with cystein residues, thereby preventing the complexes to reach their bacterial targets.^{66,67} For instance, the MIC of auranofin against *M. tuberculosis* was reported to increase 20 times in normal versus minimal medium; a result associated with albumin-promoted inactivation.⁹ For this reason, we investigated the antibacterial activity of the Au clusters and complexes in M9 minimal salts medium, a buffered solution of salts supplemented with 0.4% glucose as the carbon source, and compared the results to nutrient-rich cation-adjusted Mueller-Hinton broth (CAMHB). As expected, the Au^I complexes Au(TPPMS)Cl (**8**) and auranofin showed a 4-fold reduction in MIC in the M9 minimal medium compared to that in CAMHB (Table 4). However, the activities of clusters **4b**, **3**, and **7** were unaffected, having the same MICs in both media.

Table 4: MIC^a ($\mu\text{g/mL}$ [Au]^b) of Au clusters and complexes in M9 minimal medium vs. nutrient-rich CAMHB.

	M9	CAMHB	MIC _{CAMHB} /MIC _{M9}
$\text{Au}_L(\text{TPPMS})_n(\text{Ac}_4\text{GlcS})_{0.5n}$ (4b)	20	20	1
$\text{Au}_L(\text{TPPMS})_n$ (3)	20	20	1
$[\text{Au}_9(\text{TPPMS})_8]\text{Cl}_3$ (7)	46	46	1
$[\text{AuTPPMS}]\text{Cl}$ (8)	5.3	21	4
Auranofin	19	74	4

^aEach data repeated twice; ciprofloxacin and gentamicin included as quality controls. ^b[Au] in AuNCs determined by ICP-MS.

AuNCs are unaffected by the OM barrier

Since the OM barrier plays an important role in the cellular uptake and antibacterial activity against Gram-negative bacteria like *P. aeruginosa*, the ability of the clusters to cross the bacterial OM could explain the observed enhancement in the uptake and antimicrobial susceptibility. To shed further light on this, we measured the MICs of the clusters in the presence of colistin, a cationic antibiotic peptide that is believed to bind to

lipopolysaccharides and phospholipids, displace Ca^{2+} and Mg^{2+} , and generate pores in the bacterial OM.^{68,69} Antibiotics that are inactive against Gram-negative bacteria due to the OM barrier can be made active by colistin.²⁷ Colistin was thus added at $0.25 \times \text{MIC}$ ($\text{MIC}_{\text{colistin}} = 4 \mu\text{g/mL}$ against *P. aeruginosa* PAO1) together with the Au clusters or complexes, and the MICs were measured. The MIC of Au complex **8** was improved 64-fold in the presence of colistin (Table 5). A slight 4-fold enhancement was observed for the smaller cluster **7**, whereas no change in MIC was observed with the addition of colistin for cluster **4b**.

Table 5: MIC^a ($\mu\text{g/mL}$ [Au]^b) of Au clusters and complexes against *P. aeruginosa* PAO1 with and without colistin.

	MIC (w/o colistin) ^a	MIC (w/ colistin)	Enhancement
$\text{Au}_L(\text{TPPMS})_n(\text{Ac}_4\text{GlcS})_{0.5n}$ (4b)	20	20	1
$\text{Au}_L(\text{TPPMS})_n$ (3)	20	10/20	1-2
$[\text{Au}_9(\text{TPPMS})_8]\text{Cl}_3$ (7)	46	11	4
$[\text{AuTPPMS}]\text{Cl}$ (8)	21	0.33	64

^aEach data repeated twice; ciprofloxacin and gentamicin included as quality controls. ^b[Au] in AuNCs determined by ICP-MS. ^cEnhancement = MIC (w/o colistin)/MIC (w/ colistin).

AuNCs induce moderate genotoxicity

AuNCs can induce genotoxicity in bacteria, damaging DNA by AuNC-generated ROS species or by direct binding.⁷⁰ Damage to bacterial DNA can lead to double-stranded breaks (DSBs), which can be quantified using DNA repair proteins fused to a fluorescent protein, such as the yellow fluorescent protein (YFP).⁷¹ We used the recombinant *E. coli* SX1220 strain, capable of expressing the YFP-tagged RecN repair protein in response to DSBs and resulting in the appearance of yellow fluorescence in a concentration-dependent fashion.⁷² These YFP-tagged RecN fusion proteins usually appear as concentrated assemblies in bacteria having DSBs.⁷³ In our experiment, *E. coli* SX1220 was incubated with the Au cluster or complex at a sub-MIC concentration ($8 \mu\text{g/mL}$) for 1 h. The cells were washed with PBS, stained with the blue-fluorescent nucleic acid-staining dye 4',6-diamidino-2-phenylindole (DAPI), and then fixed for imaging by laser-scanning confocal microscopy (LSCM). Cells expressing the RecN-YFP fusion protein displayed green emission (overlap between yellow and blue), thereby indicating DSB generation (Figure 4). DSBs were observed for all AuNCs as shown by the green fluorescence emitted, albeit to a low extent compared to the controls, and the images showed mostly DAPI staining with little YFP fluorescence.

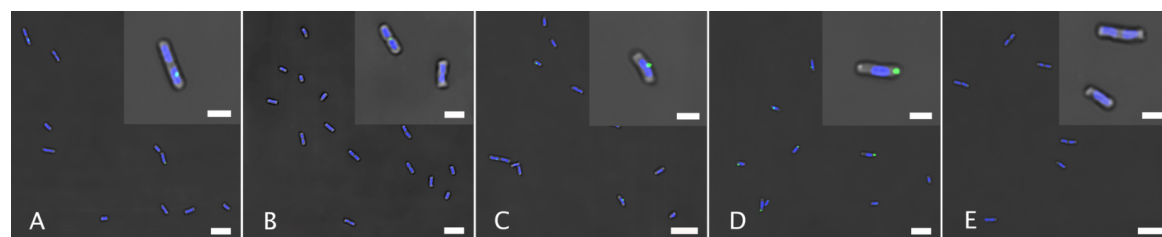


Figure 4: LSCM images of *E. coli* SX1220 treated with (A) $[\text{Au}_9(\text{TPPMS})_8]\text{Cl}_3$ (**7**), (B) $\text{Au}_L(\text{TPPMS})_n(\text{Ac}_4\text{GlcS})_{0.5n}$ (**4b**), (C) $\text{Au}_L(\text{TPPMS})_n$ (**3**), (D) ciprofloxacin, and (E) buffer only as the control. Scale bars: $5 \mu\text{m}$. Insets are enlarged regions of each panel, scale bars: $2 \mu\text{m}$. The blue DAPI-stained bacteria show the cells in a given focal plane. Overlaps with the yellow RecN-YFP fusion proteins gave green, with higher intensity indicating more DSBs.

The relative number of expressed RecN centers per cell was estimated (Figure 6), showing

18 ± 2.6% for cluster **4b**, 33 ± 3.5% for cluster **3**, 22 ± 4.3% for cluster **7**, and 19 ± 3.6% for Au^I complex **8**. To put these results in context, a series of control samples were tested. Bacteria only gave a low RecN expression of 12 ± 9.7%. Colistin, an antibiotic that causes cell death through mechanisms other than direct DNA damage,⁷⁴ gave a similar low RecN expression. Even at a high colistin concentration of 8×MIC (8 µg/mL), the RecN-YFP expression was still low (14 ± 8.9%). On the other hand, ciprofloxacin, an antibiotic known to generate DNA damage by inhibiting bacterial DNA topoisomerase IV and DNA gyrase,^{75–77} gave the most DSBs, with an expression of 138 ± 42%. Hydrogen peroxide was furthermore included to evaluate any potential DSBs caused by oxidative stress.⁷⁷ Cells treated with 30 mM H₂O₂ led to a relatively low RecN-YFP expression of 28 ± 5.4%.

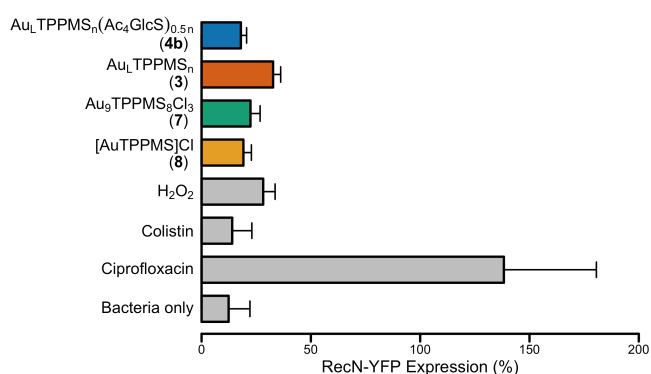


Figure 5: Percent (%) RecN-YFP expression of *E. coli* SX1220 treated with different compounds and controls. DAPI channel assigned to total bacterial cell counts and YFP channel to total RecN centers (cf. Figures S6-S7). Data represent mean ± SD from analyzing 3-6 confocal images. Cumulative cell count per sample: 107-210.

AuNCs are weaker inhibitors of TrxR than auranofin

To assess whether the disruption of intracellular redox homeostasis could be a mode of antibacterial action of AuNCs, inhibition assays were performed using the thioredoxin/thioredoxin-reductase system (Trx/TrxR). TrxR reduces the disulfide bond at the Trx active site in the presence of NADPH, facilitating the reduction of intracellular thiols to maintain the redox balance.⁷⁸ Auranofin and other Au^I complexes are known to inhibit this enzyme, putatively through interaction with the S-Se active site, and thereby suppress the activity of Trx.^{75,76,79} This inhibition then results in unchecked oxidative stress and ultimate cell death. The *in vitro* inhibition assays were conducted using the *E. coli* TrxR in the presence of Ellman's reagent 5,5'-dithiobis(2-nitrobenzoic acid) (DTNB), which upon reduction by Trx generates colored 2-nitro-5-thiobenzoate (λ_{max} : 412 nm). The TrxR inhibition was estimated by following the rate of DTNB reduction spectrophotometrically in the presence of Au clusters and complexes. The MICs of these Au clusters and complexes against *E. coli* ATCC 25922 were furthermore recorded for comparison. Auranofin is a strong inhibitor of *E. coli* TrxR, giving an IC₅₀ of 0.0045 ± 0.006 µg/mL [Au] (**Table 6**). Au complex **8** showed ~12 times weaker inhibition than auranofin. The clusters showed considerably lower degrees of inhibition, where AuNC **4b** (IC₅₀: 2.4 ± 0.4 µg/mL [Au]) showed a >500 times higher IC₅₀-value than auranofin. The AuNCs **3** and **7** displayed slightly higher inhibitory activity compared to cluster **4b**.

Table 6: IC_{50} of Au clusters and complexes against *E. coli* TrxR and MIC against *E. coli* ATCC 25922.

	IC_{50} ($\mu\text{g/mL Au}$) ^a	MIC ($\mu\text{g/mL [Au]}$)
$\text{Au}_L(\text{TPPMS})_n(\text{Ac}_4\text{GlcS})_{0.5n}$ (4b)	2.4 ± 0.4	20
$\text{Au}_L(\text{TPPMS})_n$ (3)	1.2 ± 0.6	40
$[\text{Au}_9(\text{TPPMS})_8]\text{Cl}_3$ (7)	0.69 ± 0.4	>46
$[\text{AuTPPMS}]\text{Cl}$ (8)	0.053 ± 0.02	21
Auranofin	0.0045 ± 0.006	4.7

^aMean \pm SD of two independent measurements.

Intracellular thiol depletion

In addition to the Trx/TrxR system, Gram-negative bacteria also utilize the glutathione (GSH)-glutaredoxin (Grx) system to counteract the buildup of ROS within the cell.⁸⁰ Analogous to Trx/TrxR, this NADPH-dependent process occurs through initial reduction of glutathione reductase (GR), followed by consequential reduction of intracellular GSH, Grx, and potential downstream disulfide reduction.⁸⁰ Therefore, quantification of available thiols in the bacteria can indicate the general oxidative stress the bacteria are under, which includes both systems of thiol-disulfide regulation.^{81,82} To observe this effect, *P. aeruginosa* PAO1 was incubated with the Au clusters and complexes and the amount of free intracellular thiol was measured using a commercially available thiol detection kit. Compared with the thiol concentration in untreated bacteria, clusters **4b** and **3** resulted in ~50% reduction, similar to Au^I-complex **8**. The smaller cluster **7** led to higher thiol levels, resulting in a decrease of only 12% relative to the untreated control.

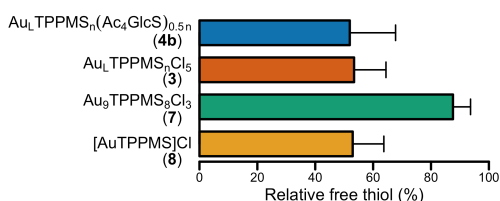


Figure 6: Free thiol concentration in PAO1 upon treating with different Au compounds (9.8 $\mu\text{g/mL [Au]}$), with 100% being untreated bacteria.

DISCUSSION

All Au clusters and complexes tested were more active against Gram-positive bacteria compared to Gram-negative strains, mirroring the general trend observed with auranofin. The exchangeability of the TPPMS-protected Au^INC enabled the preparation of auranofin AuNC analogues with varying levels of glycosylation. While the increase in glycosylation led to reduction in cytotoxicity, the antibacterial activity was less regular with the cluster $\text{Au}_L(\text{TPPMS})_n(\text{Ac}_4\text{GlcS})_{2n}$ (**4b**) showing the best activity. Most notably, the MIC and MBC of cluster **4b** against *P. aeruginosa* at 20 $\mu\text{g/mL [Au]}$ is an improvement over auranofin, which has an MIC of 74 $\mu\text{g/mL [Au]}$ (Table 1). The water-soluble cluster **3** showed a similar activity as cluster **4b**, whereas the water-insoluble clusters **2** and **1** had no activity against Gram-negative bacteria. The reduced activity of the more hydrophobic AuNCs could be due to their lower availability in aqueous solution as is observed with other poorly water-soluble antibiotics.⁴¹

The coordination of the Ac_4GlcSH ligand to the AuNCs, formulating the auranofin analogue AuNCs, had a sizable impact on the antibacterial activity (Table 2). An apparent

optimal ligand ratio was obtained, where the highest activity against *P. aeruginosa* was observed at a Ac₄GlcS/TPPMS ratio of 1:2 (MIC = 20 µg/mL for cluster **4b**). The activity decreased drastically at higher Ac₄GlcS content (MIC >100 µg/mL for cluster **4d**). The MICs of cluster **4b** against the non-MDR *P. aeruginosa* strains PAO1 and ATCC 27853 were similar to the MDR strain NCTC 13437, which shows extensive resistance to β-lactams and aminoglycosides.⁸³ This points toward a killing mechanism of the AuNCs that is different from these two traditional antibiotic classes. Interestingly, Au₉ cluster **7** was less active than non-glycosylated AuNC **3** for all strains tested. The Au^I-complexes also showed ligand-dependent activity, where Au(TPPMS)Cl (**8**) had similar MICs as cluster **3**, while Au(TPPMS)(Ac₄GlcS) (**9**) was more inert to the *P. aeruginosa* strains.

Glycosylation of cluster **3** also led to a slight reduction in cytotoxicity and cluster **4b** displayed a higher CC₅₀-value against A549 cells (53 ± 7.1 µg/mL) compared to AuNC **3** (42 ± 12 µg/mL, **Table 3**). This reduction in cellular toxicity has been postulated to result from a reduction in DNA-AuNC interactions due to the inability of DNA phosphate groups to replace thiols on the AuNCs.⁸⁴ Our genotoxicity results indeed showed lower DSBs for the thiolated cluster **4b** (18%) than for the non-thiolated cluster **3** (33%, **Figure 5**). In addition, the AuNCs were considerably less cytotoxic than auranofin, where cluster **4b** displayed a 90 times higher SI compared to the Au complex (**Table 6**).

The first insight into the AuNC activity was provided by measuring the accumulation of Au in *P. aeruginosa* PAO1 using ICP-MS, showing that the uptake of Au from the larger Au_L-clusters (20% and 26% for **4b** and **3**) was considerably higher than from Au₉-cluster **7** (3.7%) and Au^I-complex **8** (6.3%, **Figure 3**). The uptake results of the clusters correlate well with their MIC values, with the higher Au uptake resulting in higher activity. Thus, Au_L-clusters **4b** and **3** showed both a higher degree of uptake and better activity than the less incorporated Au₉ cluster **7**. The stability test over 26 h pointed against the possibility that the uptake of the Au_L-cluster could be the result of decomposition products into Au^I or small AuNCs (**Figure 3**). However, despite the relatively low uptake, the activity of complex **8** was similar to the AuNC clusters, implying that this complex requires lower accumulation to achieve similar killing.

Auranofin is inactive towards *P. aeruginosa*. One hypothesis to account for this is the OM barrier, supported by the observation that the activity of auranofin improved 1000-fold in the presence of a cationic peptide, polymyxin B.²⁷ For the Au^I complex Au(TPPMS)Cl, the addition of colistin led to a drastic 64-fold improvement in MIC. However, only a 4-fold enhancement was observed for the smaller Au₉-cluster **7**, and no change was observed for clusters **4b** and **3**. These results imply that Au^I complex **8** has a relatively high intracellular activity but is limited by the inability to traverse the outer membrane. Another factor that contributes to the decrease in the activity of the Au^I complex appears to be the interaction with extracellular proteins in the growth medium. Similar to auranofin, the activity of Au^I complex **8** was reduced 4-fold in nutrient-rich broth versus in the minimal medium (**Table 4**). In comparison, AuNCs **3** and **4b** were largely unaffected by the extracellular proteins in the growth medium. This behavior, together with the colistin co-administration data, explains the conflicting results of low accumulation yet moderate activity observed for the Au^I complex. In contrast, while the AuNCs may not be as intracellularly active as the Au^I-complexes, their ability to effectively penetrate the bacterial cell wall results in similar activity. Internalization of AuNCs (~1–2 nm) by bacteria has been reported in antibacterial and photosensitizing studies.^{82,85} However, the actual pathways that allow higher accumulation of larger clusters compared to the smaller Au₉-clusters and Au^I-complexes warrant further investigations.

Assessment of DNA damage that leads to DSBs can determine whether DNA is the site of

drug action or collateral damage of ROS species generated by AuNCs. From the analysis, it was evident that the Au compounds caused DNA damage compared to the negative controls of bacteria only and colistin. However, the degree of DSBs was considerably lower than ciprofloxacin, and more similar to that induced by H_2O_2 (**Figure 5**). This suggests that the Au compounds are less likely to interfere with bacterial DNA directly, and potentially more prone to cause damage through ROS generation. Of the clusters, $Au_L(TPPMS)_n$ (**3**) displayed a higher level of DSBs compared to all other clusters, suggesting that some DNA-interaction could potentially occur. This hypothesis has been proposed for a TPPMS-protected “Au55” cluster of comparable size where the phosphine ligands were replaced with DNA phosphate groups, thereby allowing the AuNC core to insert into the major groove of DNA.^{67,84} This is further supported by lower levels of DSBs by the glycosylated cluster **4b** where the Ac_4GlcS -groups are significantly less likely to undergo exchange by phosphates. The smaller AuNC **7** and Au^I complex **8** both induced intermediate levels of DSBs, most likely ROS-mediated in these cases.

Bacteria possess several mechanisms to respond to oxidative stress, including the thioredoxin and glutaredoxin systems that can mitigate the effects of ROS. The larger AuNCs **4b** and **3** are only moderate inhibitors of TrxR, a key component of the Trx system, more than 500 times less active than auranofin (**Table 6**). The results indicate that inhibition of the thioredoxin system may not be the major antimicrobial mechanism for these AuNCs. We also quantified the amount of free thiols in bacteria exposed to Au clusters and complexes. Such thiols, like glutathione, are used by the GR/Grx system in Gram-negative bacteria to regulate oxidative stress. The thiol depletion assay showed that all Au clusters and complexes except Au_9 -cluster **7** reduced the intracellular thiols by ~50% (**Figure 6**), indicating that, aside from Trx, thiol-dependent redox systems were compromised in the presence of AuNCs.

Considering the similarity in thiol depletion level and the difference in TrxR inhibition, the AuNCs and Au^I complexes appear to affect the bacterial homeostasis by different mechanisms. For Au^I -complexes, like compound **8** and auranofin, the main source of disruption appears to be the strong inhibition of the TrxR/Trx system. Conversely, AuNCs **4b** and **3** displayed low inhibition of TrxR and a higher effect on the GR/Grx system, possibly by direct reaction with GSH and other reduced thiols. This mechanism is supported by the inability of Au_9 -cluster **7** to reduce free thiols. Experimentally, the exchange of phosphine ligands by Ac_4GlcSH was not feasible for this cluster, implying that it is less reactive to free thiols in bacteria. However, this does not exclude the possibility of intracellular AuNC degradation to generate active Au^I species. Overall, these mechanisms point to the effectiveness of AuNCs **4b** and **3** in limiting bacteria adaptability to oxidative stress that results in bactericidal activity.

CONCLUSIONS

Based on the structural motif of the FDA-approved gold drug auranofin, we designed AuNCs functionalized with mixed phosphine and glycosyl thiol ligands, synthesized by ligand exchange of PPh_3 -capped AuNCs with TPPMS followed by Ac_4GlcSH . The exchangeability of the TPPMS-protected AuLNCs enabled the preparation of clusters with increasing levels of glycosylation that were evaluated for their antibacterial activity and cytotoxicity. A collection of water-soluble 1.5 nm AuNCs and Au_9 -clusters were thus prepared, of which the larger Au_L NCs demonstrated broad-spectrum activity against a range of bacteria, including highly resistant ESKAPE pathogens. In addition, the clusters displayed notably lower toxicity against human cells compared to auranofin. While the increase in glycosylation led to a

proportional but gradual reduction in cytotoxicity, the antibacterial activity was less regular. Cluster **4b** showed the highest activity and was less cytotoxic compared to the non-glycosylated counterpart **3**.

Investigations on the MOAs of the Au clusters and complexes against *P. aeruginosa* PAO1 revealed important aspects of the complex antibacterial mechanism of these entities with respect to the uptake, DNA damage, and disruption of redox homeostasis. Overall, the results show that the conversion of Au^I complexes into an AuNC formulation can enhance the activity against bacteria and reduce toxicity towards human cells. The source of this enhancement combines several factors, including the ability of AuNCs to accumulate more effectively in bacteria than the Au^I complexes, to translocate over the OM barrier, and to avoid extracellular protein interactions. The observed antibacterial mechanisms, including DNA damage, ROS generation, and inhibition of bacterial oxidative stress response, appeared multifaceted, showing both similarities and discrepancies compared to the MOAs observed for Au^I complexes including auranofin.

The results from this study demonstrate that drug repurposing can be extended to nanoformulation. The retention of these properties opens up the possibility of this strategy as a formulation route in the drug development process, applied to enhance the efficacy of small coordination complexes as drugs in antimicrobial therapy.

MATERIALS AND METHODS

Materials

All reagents, solvents, and other chemicals were obtained in the highest available purity from commercial suppliers and generally used as received. Chloroform was passed through basic alumina before use. Deionized water was obtained from a Milli-Q ultrapure water purification system. All reactions were monitored by thin layer chromatography (TLC) using plates precoated with silica gel 60 F₂₅₄ (Merck KGaA, Darmstadt, Germany), visualized under a hand-held ultraviolet light device or by staining with a 5% H₂SO₄ solution in ethanol. Sephadex LH-20 (GE Lifesciences) was purchased from VWR (Radnor, PA, USA). TraceCERT HCl, HNO₃, and Au standard for ICP-MS analyses were from MilliporeSigma (St. Louis, MO, USA). ICP-MS measurements were performed on an Agilent 7900 ICP-MS. RecN-YFP images were collected using a Leica SP8 TCS laser scanning confocal microscope. *E. coli* Trx, *E. coli* TrxR and NADPH as well as the Thiol Detection Assay Kit were obtained from Cayman Chemical (Ann Arbor, MI).

ESI-MS spectra and TEM images were acquired using Bruker MicrOTOF II and JEOL JEM-2200FS instruments, respectively, at the University of Massachusetts Amherst. OD and fluorescence data were obtained using a Tecan Infinite PRO microplate reader and a Tecan Spark 10M plate reader. NMR spectra were recorded on Bruker Avance Spectrospin DRX500 (¹H NMR) or JEOL ECZ 400 MHz (³¹P NMR) spectrometers. ¹H NMR signals are referenced to either (residual) solvent peaks or TMS (δ 0.00 ppm). For ³¹P NMR, 85% H₃PO₄ (0.00 ppm) was used as external reference. UV-Vis spectra were obtained using a PerkinElmer Lambda 750 UV-Vis NIR spectrometer.

Bacteria were cultured in either BBL Mueller Hinton Broth (MHB) from Becton, Dickinson and Company (Sparks, MD, USA) or LB broth from MilliporeSigma (St. Louis, MO, USA). Antimicrobial assays were conducted in MHB, cation-adjusted with Mg²⁺ and Ca²⁺. BD Difco Dehydrated Culture Media: M9 Minimal Salts, 5x media was bought from Fisher Scientific (Hampton, NH, USA). *P. aeruginosa* ATCC 27853 and *E. coli* ATCC 25922 were purchased from ATCC (Manassas, VA, USA). Methicillin-resistant *S. aureus* USA300

JE2 was from BEI resources (Manassas, VA, USA). *E. coli* SX1220 was purchased from The Coli Genetic Stock Center (CGSC #12775) at Yale University (New Haven, CT, USA). A549 cells (ATCC CCL-185) and NIH/3T3 cells (ATCC CRL-1658) were purchased from ATCC. Dulbecco's Modified Eagle's Medium (DMEM) was acquired from MilliporeSigma. Iron-fortified Bovine Calf Serum (SAFC Biosciences) and 0.25% trypsin with EDTA (Gibco) were purchased from Fisher Scientific (Hampton, NH, USA). Cells were counted using a Countess automated hemocytometer from Invitrogen (Carlsbad, CA, USA).

AuNC syntheses

Au₁₀₁(PPh₃)₂₁Cl₅ (1). A modified method of Hutchison and coworkers was adopted.⁴⁸ To a biphasic solution (toluene/water 1.3/1, v/v, 115 mL) purged with Ar for 1 h, TOAB (1.6 g, 2.9 mmol) and HAuCl₄•3H₂O (1.0 g, 2.5 mmol) were added. The mixture was stirred for 10 min, after which the yellow color in the aqueous layer disappeared and the organic layer turned red. Triphenylphosphine (2.3 g, 8.9 mmol) was then added, and the reaction was stirred for a further 10 min until the organic phase was white and cloudy. A solution of NaBH₄ (1.4 g, 37 mmol) in water (10 mL) was immediately added, whereupon the mixture turned black. The reaction mixture was stirred under argon for 4 h, after which the reaction mixture was washed with water (3 × 50 mL). The organic layer was separated, and the solvent was evaporated to yield the crude product as a black solid. The crude was dispersed in hexanes by sonication and then filtered through a medium-porosity frit. The solid was then washed successively with: 1) hexanes (200 mL) followed by water (100 mL), 2) hexanes (5 × 100 mL) followed by MeOH/water 2:3, v/v (100 mL), 3) hexanes (5 × 100 mL) followed by saturated sodium nitrite solution (100 mL), 4) hexanes (5 × 100 mL) followed by MeOH/water 2:3, v/v (100 mL), 5) hexanes (5 × 100 mL) followed by saturated sodium nitrite solution (100 mL), and 6) hexanes (5 × 50 mL) followed by MeOH/water 2:3, v/v (100 mL). The washed product was dissolved in chloroform, concentrated (~10 mL), and slowly precipitated with pentane dispensed from a syringe pump at 10 mL/h over 2 h. This process was repeated 3 times, after which the solvent was evaporated to give cluster **1** as a black solid (190 mg, 27% [Au]). ¹H NMR (CDCl₃, 400 MHz): δ 7.2 (br)

Au_L(PPh₃)_x(Ac₄GlcS)_y (2). Cluster **1** (10 mg, 0.39 mmol) and Ac₄GlcSH (2.3 mg, 6.4 mmol) were dissolved in DCM (5.0 mL) and the reaction was stirred at room temperature for 12 h. The reaction mixture was concentrated and purified by size exclusion chromatography (DCM/MeOH 1:5, v/v), after which the free ligand (R_f: 0.4, hexanes/EtOAc 1:1, v/v) disappeared as visualized by TLC. The solvent was evaporated to give cluster **2** as a black solid (4.0 mg, 40% [Au]). ¹H NMR (CDCl₃, 400 MHz): δ 7.2 (br), 2.1 (br). ³¹P NMR (CDCl₃, 162 MHz): Not discernible.

Au_L(TPPMS)_n (3). Cluster **1** (10 mg, 0.39 mmol) was dissolved in DCM (3 mL), to which a solution of TPPMS (7.1 mg, 2.0 mmol) in water (3.0 mL) was added, and the biphasic mixture stirred at room temperature for 12 h. Following separation of the solvent phases, the aqueous layer was washed with DCM (3 × 10 mL) and the organic layer with water (3 × 10 mL), after which the combined aqueous phase was concentrated. Purification by size exclusion chromatography (MeOH/water 4:1, v/v) yielded product **3** as a black solid (7.9 mg, 70% [Au]). ¹H NMR (D₂O, 400 MHz): δ 7.3 (br). ³¹P NMR (CDCl₃, 162 MHz): Not discernible.

Au_L(TPPMS)_x(Ac₄Glc)_y (4a-4d). Cluster **3** (12 mg) was dissolved in MeOH (4 mL) and the solution was heated to 40 °C. Ac₄GlcSH, in a mole ratio of 4, 16, 64, or 256 to the cluster, was dissolved in DCM (0.5 mL) and added to the stirring solution. The reactions were performed at 40 °C for 12 h. The resulting solutions were concentrated and purified by size

exclusion chromatography (MeOH/water 4:1, v/v). Purity was confirmed by the absence of free ligand (R_f : 0.4, hexanes/EtOAc 1:1, v/v) by TLC. After evaporation of the solvent to dryness, products **4a-4d** were obtained as dark brown/black solids, **4a** (7.7 mg, 64% [Au]), **4b** (8.0 mg, 66% [Au]), **4c** (8.5 mg, 70% [Au]), **4d** (9.3 mg, 76% [Au]). ^1H NMR (D_2O , 400 MHz): δ 7.3 (br), 2.1 (br). ^{31}P NMR (CDCl_3 , 162 MHz): Not discernible.

[Au₁₁(PPh₃)₈Cl₂]Cl (5).^{59,86} To a stirred solution of Au(PPh₃)Cl (0.66 g, 1.3 mmol) in DCM (28 mL), NaBH₄ (14 mg, 0.37 mmol) in absolute ethanol (4.2 mL) was quickly added, and the reaction allowed to proceed while stirring at room temperature for 24 h. The crude product was purified by successive slow precipitation using pentanes, followed by column chromatography (DCM/MeOH 20:1, 15:1, 10:1, 5:1, v/v). Following concentration to dryness, product **5** was obtained as a red solid (11 mg, 40% [Au]). ^1H NMR (CDCl_3 , 400 MHz): δ 7.30 (br, 2H), 6.93 (t, 1H, J = 7.4 Hz), 6.68 (t, 2H, J = 7.6 Hz). ^{31}P NMR (CDCl_3 , 162 MHz): δ 53.05.

Au₁₁(PPh₃)_x(Ac₄Glc)_y (6). [Au₁₁(PPh₃)₈Cl₂]Cl (**5**, 4.9 mg, 1.6 μmol) was dissolved in DCM/1-chlorobutane 3:1, v/v (4.0 mL) and heated to 50 °C. To this solution, Ac₄GlcSH (4.5 mg, 12 μmol) was added while stirring, and the solution was heated for 4 h until the solution turned from red to brown. The solvent was removed, and the product was purified by size exclusion chromatography (ethanol), supported by the absence of free ligand (R_f : 0.4, hexanes/EtOAc 1:1, v/v) by TLC. Following concentration to dryness, product **6** was obtained as a brown solid. (3.7 mg, 76%). ^1H NMR (CDCl_3 , 500 MHz): δ 7.6-7.3 (m, 15H), 5.0-5.3 (m, 4H), 4.2-4.5 (m, 2H), 2.0-2.2 (m, 12H). ^{31}P NMR (CDCl_3 , 81 MHz): NMR (δ 44.23)

[Au₉(TPPMS)₈]Cl₃ (7). Au₁₁(PPh₃)₈Cl₂]Cl (**5**, 11 mg, 2.5 μmol) was dissolved in DCM (3.0 mL) and the solution was added to TPPMS (18 mg, 50 μmol) dissolved in water (3 mL), after which the resulting biphasic mixture was stirred vigorously while heating at 40 °C. After 4 h, the red color had completely transferred to the aqueous layer. Following separation of the solvent phases, the aqueous layer was washed with DCM (3 \times 5 mL) and the organic layer with water (3 \times 5 mL), after which the combined aqueous phase was concentrated. Following purification by size exclusion chromatography (MeOH/water = 4:1 v/v), the colored fractions were lyophilized, yielding product **7** as a powdery red solid. (13.3 mg, 91% [Au]). ^1H NMR (D_2O , 400 MHz): δ 7.36 (d, 1H, J = 7.9 Hz), 7.24 (s, 1H), 7.05 (m, 8H), 6.71 (t, 4H, J = 7.7 Hz), 6.36 (t, 1H, J = 7.8 Hz). ^{31}P NMR (D_2O , 162 MHz): δ 57.81.

Determination of MIC and MBC

MICs and MBCs were determined by the broth microdilution method, as described in the Clinical & Laboratory Standards Institute (CLSI) guidelines.⁸⁷ The compounds were dissolved in either DMSO or water as stock solutions. All bacterial strains were cultured in CAMHB up to log phase and then diluted to match 0.5 \times McFarland standard using 0.85% saline. This bacterial suspension was diluted to $\sim 1 \times 10^6$ CFU/mL and then dispensed in aliquots (100 μL) to a 96-well plate containing volumes (100 μL) of two-fold serially diluted compound solutions. After incubation at 35 °C for 18 h, the plates were checked spectrophotometrically for growth (OD_{600}) and/or by incubation with 10% alamarBlue for 2 h followed by fluorimetry. The MIC was the concentration that resulted in an absorbance less than 90% of the control at 600 nm (OD_{600}), showing 90% less absorbance (570 nm) or fluorescence (Ex: 570 nm, Em: 590 nm) in the case of alamarBlue. To determine the MBCs, agar plates were treated with sample solutions (5 μL) from each well and then incubated at 35 °C for 18 h without shaking. The sample concentrations that resulted in no colony formation on agar plates were the MBCs.

MIC determination in minimal medium followed the same protocol as above except that

M9 Minimal Salts (BD Difco) medium was used. The medium was prepared following the manufacturer's instructions and supplemented with 1 mM MgSO₄, 0.5 mM CaCl₂, and 0.4% D-glucose as the carbon source. Bacteria was grown in M9 medium until an OD of ~0.3 and then diluted to match 0.5 × McFarland standard using 0.85% saline.

Determination of CC₅₀ for A549 cells

A549 cells were cultured in DMEM supplemented with 10% fetal bovine serum and 1% each of penicillin/streptomycin, gentamicin sulfate, and amphotericin B. Cells were grown to 80-90% confluency in a humidified chamber kept at 37 °C and 5% CO₂. The confluent cells were detached from the flasks and seeded in a 96-well plate overnight at a concentration of 5,000 cells per well. The medium was removed, and aliquots (80 µL) containing compounds were added to the wells. The cells were incubated with the compounds for 18 h, after which the medium was removed and 10% of alamarBlue solution (100 µL) in DMEM complete culture media was added to the wells. The well plates were incubated for a further 3 h before fluorescence readings were obtained using a Tecan Infinite PRO 200 plate reader (Ex: 575 nm, Em: 590 nm). The CC₅₀ values were determined by analyzing the fluorescence intensity vs. log[Au] by standard nonlinear regression (Origin PRO/Graphpad Prism). Complete sigmoidal curves were observed for the compounds (Figures S5-S6)].

Determination of Au uptake by *P. aeruginosa*

P. aeruginosa PAO1 was cultured to log phase and then adjusted to 4.2 × 0.5 McFarland standard as determined by OD₆₀₀. Volumes of this bacterial suspension (0.95 mL) were combined with solutions of the gold compounds (50 µL, final concentration 8 µg/mL) and were incubated at 35 °C for 4 h without shaking. The cultures were pelleted (3000 × g for 6 min) and washed with PBS (3 × 1 mL). The samples were then digested at room temperature overnight using freshly prepared *aqua regia* (1.0 mL) (HCl/HNO₃ = 4:1 v/v, using TraceCERT acids), followed by heating to 60 °C for 2 h at which point no solids were visible in the solution. The samples were diluted in 2% HNO₃ and analyzed by ICP-MS against a standard series prepared from a TraceCERT Au Standard for ICP. The accumulated Au content was quantified as the percentage of the total Au in the initially added gold compounds. For each experiment, the final bacteria count was established by dilution and plating followed by incubation for 18 h. The final concentration of Au was then normalized to the bacteria counts using 4.0 × 10⁹ CFU/mL as the baseline.

Genotoxicity assay

E. coli SX1220, cultured to a density equal to 0.5 McFarland standard, were incubated with gold compounds at a sub-MIC level (8 µg/ml) for 1 h. The bacteria were then pelleted (3000 × g for 6 min), washed once with PBS (1.0 mL), and then fixed using 4% paraformaldehyde (1.0 mL) at room temperature for 15 min. The pellets were washed with PBS (3 × 1.0 mL) to remove the excess aldehyde and redispersed in PBS (1.0 mL). The samples (2.5 µL) were mounted on cleaned slides using ProLong Glass Antifade Mountant with NucBlue Stain (5.0 µL), covered using cleaned #1.5H glass slips. After being left to cure for at least 24 h, confocal images were acquired using the 405 nm excitation laser line for DAPI and the 488 nm laser line for YFP. The images obtained were processed and analyzed using the Leica Application Suite X (LAS X) software.

Determination of IC₅₀ for TrxR

All solutions were prepared in 50 mM HEPES buffer, pH 7.5, with 1 mM EDTA, and the experiments were conducted using black, clear-bottom 96-well plates. A 250 nM solution of TrxR (10 µL) was combined with 0.067 mM NADPH (15 µL) and 0.067 mM of DTNB solution (15 µL). Aliquots (50 µL) of serially diluted solutions of Au compounds were then added to the wells and the mixtures were incubated at 30 °C for 10 min. The reaction was initiated by the addition of the Trx solution (100 µM, 10 µL), and the absorbance of the reduced DTNB was recorded at 412 nm every minute for 10 min.⁹ The enzyme activity in the presence of the gold compound compared to the reference without inhibitor ($\Delta Abs_{AuNC}/\Delta Abs_{ref}$) vs. log[Au] was used to estimate half-maximal inhibitory concentration IC₅₀ using standard non-linear regression analysis (Origin PRO/Graphpad Prism). Two independent assays were performed and the final IC₅₀ values were averaged.

Quantification of free thiols

P. aeruginosa PAO1 was obtained from a single colony and grown overnight in CAMHB to OD₆₀₀ ~1.0. The culture was diluted to OD₆₀₀ ~0.5 and incubated with 50 µM of Au compound for 5 h. The cultures were then pelleted (7,000 × g for 5 minutes), washed twice with PBS (1.0 mL), and then resuspended in PBS (3 mL). The bacteria were lysed via probe sonication using Vibra-Cell VCX 130 (Sonics) with a pulse of 9 s on and 9 s off at 30% amplitude for 5 min. Thiols were detected and quantified using the Thiol Detection Assay Kit (Cayman Chemicals) at an excitation wavelength of 365 nm and an emission wavelength of 520 nm.

AUTHOR INFORMATION

Corresponding authors

Mingdi Yan; E-mail: Mingdi_Yan@uml.edu

Olof Ramström; E-mail: olof_ramstrom@uml.edu

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KEYWORDS

Gold nanoclusters, auranofin, antimicrobial, *Pseudomonas aeruginosa*, glyconanoparticles

REFERENCES

- (1) U.S. Department of Health and Human Services; CDC. *Antibiotic resistance threats in the United States*; 2019.
- (2) Pushpakom, S.; Iorio, F.; Eyers, P. A.; Escott, K. J.; Hopper, S.; Wells, A.; Doig, A.; Guilleams, T.; Latimer, J.; McNamee, C.; Norris, A.; Sanseau, P.; Cavalla, D.; Pirmohamed, M. *Nat. Rev. Drug Discov.* **2019**, *18* (1), 41–58.

- (3) Brown, D. *Nat. Rev. Drug Discov.* **2015**, *14* (12), 821–832.
- (4) Abutaleb, N. S.; Seleem, M. N. *Sci. Rep.* **2020**, *10* (1).
- (5) Roder, C.; Athan, E. *Drugs R D* **2020**, *20* (3), 209–216.
- (6) AbdelKhalek, A.; Abutaleb, N. S.; Elmagarmid, K. A.; Seleem, M. N. *Sci. Rep.* **2018**, *8* (1), 1–9.
- (7) Thangamani, S.; Mohammad, H.; Abushahba, M. F. N.; Sobreira, T. J. P.; Seleem, M. N. *Int. J. Antimicrob. Agents* **2016**, *47* (3), 195–201.
- (8) Owings, J. P.; McNair, N. N.; Mui, Y. F.; Gustafsson, T. N.; Holmgren, A.; Contel, M.; Goldberg, J. B.; Mead, J. R. *FEMS Microbiol. Lett.* **2016**, *363* (14), fnw148.
- (9) Harbut, M. B.; Vilchèze, C.; Luo, X.; Hensler, M. E.; Guo, H.; Yang, B.; Chatterjee, A. K.; Nizet, V.; Jacobs, W. R.; Schultz, P. G.; Wang, F. *Proc. Natl. Acad. Sci. USA* **2015**, *112* (14), 4453–4458.
- (10) Hokai, Y.; Jurkowicz, B.; Fernández-Gallardo, J.; Zakirkhodjaev, N.; Sanaú, M.; Muth, T. R.; Contel, M. J. *Inorg. Biochem.* **2014**, *138*, 81–88.
- (11) Jackson-Rosario, S.; Cowart, D.; Myers, A.; Tarrien, R.; Levine, R. L.; Scott, R. A.; Self, W. T. *J. Biol. Inorg. Chem.* **2009**, *14* (4), 507–519.
- (12) Rossi, S. A.; Oliveira, H. C. de; Agreda-Mellon, D.; Lucio, J.; Mendes-Giannini, M. J. S.; García-Camero, J. P.; Zaragoza, O. *Antimicrob. Agents Chemother.* **2020**, *64* (4).
- (13) Wiederhold, N. P.; Patterson, T. F.; Srinivasan, A.; Chaturvedi, A. K.; Fothergill, A. W.; Wormley, F. L.; Ramasubramanian, A. K.; Lopez-Ribot, J. L. *Virulence* **2017**, *8* (2), 138–142.
- (14) Thangamani, S.; Maland, M.; Mohammad, H.; Pascuzzi, P. E.; Avramova, L.; Koehler, C. M.; Hazbun, T. R.; Seleem, M. N. *Front. Cell. Infect. Microbiol.* **2017**, *7*, 4.
- (15) May, H. C.; Yu, J.-J.; Guentzel, M. N.; Chambers, J. P.; Cap, A. P.; Arulanandam, B. P. *Front. Microbiol.* **2018**, *9*, 336.
- (16) Sannella, A. R.; Casini, A.; Gabbiani, C.; Messori, L.; Bilia, A. R.; Vincieri, F. F.; Majori, G.; Severini, C. *FEBS Lett.* **2008**, *582* (6), 844–847.
- (17) Gil-Moles, M.; Basu, U.; Büssing, R.; Hoffmeister, H.; Türck, S.; Varchmin, A.; Ott, I. *Chem. Eur. J.* **2020**, *26* (66), 15140–15144.
- (18) Rothan, H. A.; Stone, S.; Natekar, J.; Kumari, P.; Arora, K.; Kumar, M. *Virology* **2020**.
- (19) Stratton, M.; Ramachandran, A.; Camacho, E. J. M.; Patil, S.; Waris, G.; Grice, K. A. *J. Inorg. Biochem.* **2020**, *206*, 111023.
- (20) Lewis, M. G.; DaFonseca, S.; Chomont, N.; Palamara, A. T.; Tardugno, M.; Mai, A.; Collins, M.; Wagner, W. L.; Yalley-Ogunro, J.; Greenhouse, J. *AIDS* **2011**, *25* (11), 1347–1356.
- (21) Fonteh, P. N.; Keter, F. K.; Meyer, D. *Biomaterials* **2010**, *23* (2), 185–196.
- (22) Marzano, C.; Gandin, V.; Folda, A.; Scutari, G.; Bindoli, A.; Rigobello, M. P. *Free Radic. Biol. Med.* **2007**, *42* (6), 872–881.
- (23) Talbot, S.; Nelson, R.; Self, W. T. *Br. J. Pharmacol.* **2008**, *154* (5), 940–948.
- (24) Gromer, S.; Johansson, L.; Bauer, H.; Arscott, L. D.; Rauch, S.; Ballou, D. P.; Williams, C. H.; Schirmer, R. H.; Arner, E. S. J. *Proc. Natl. Acad. Sci. USA* **2003**, *100* (22), 12618–12623.
- (25) Saccoccia, F.; Angelucci, F.; Boumis, G.; Carotti, D.; Desiato, G.; Miele, A.; Bellelli, A. *Curr Protein Pept Sc* **2014**, *15* (6), 621–646.
- (26) Holmgren, A. *J. Biol. Chem.* **1989**, *264* (24), 13963–13966.
- (27) Thangamani, S.; Mohammad, H.; Abushahba, M. F. N.; Sobreira, T. J. P.; Hedrick, V. E.; Paul, L. N.; Seleem, M. N. *Sci. Rep.* **2016**, *6*, 22571–22571.
- (28) Lu, J.; Vlamis-Gardikas, A.; Kandasamy, K.; Zhao, R.; Gustafsson, T. N.; Engstrand, L.; Hoffner, S.; Engman, L.; Holmgren, A. *FASEB J.* **2012**, *27* (4), 1394–1403.
- (29) Jin, R.; Zeng, C.; Zhou, M.; Chen, Y. *Chem. Rev.* **2016**, *116* (18), 10346–10413.
- (30) Yang, X.; Yang, M.; Pang, B.; Vara, M.; Xia, Y. *Chem. Rev.* **2015**, *115* (19), 10410–10488.
- (31) Liyanage, S. H.; Yan, M. *Chem. Commun.* **2020**, *56* (88), 13491–13505.
- (32) Zheng, Y.; Liu, W.; Chen, Y.; Li, C.; Jiang, H.; Wang, X. *J. Colloid Interface Sci.* **2019**, *546*, 1–10.
- (33) Wang, S.; Wang, Y.; Peng, Y.; Yang, X. *ACS Appl. Mater. Interfaces* **2019**, *11* (8), 8461–8469.
- (34) Kalita, S.; Kandimalla, R.; Bhowal, A. C.; Kotoky, J.; Kundu, S. *Sci Rep* **2018**, *8* (1), 5778.

- (35) Li, Q.; Pan, Y.; Chen, T.; Du, Y.; Ge, H.; Zhang, B.; Xie, J.; Yu, H.; Zhu, M. *Nanoscale* **2018**, *10* (21), 10166–10172.
- (36) Liang, J.; Xiong, H.; Wang, W.; Wen, W.; Zhang, X.; Wang, S. *Sens. Actuators B Chem.* **2018**, 255, 2170–2178.
- (37) Chen, W.-Y.; Chang, H.-Y.; Lu, J.-K.; Huang, Y.-C.; Harroun, S. G.; Tseng, Y.-T.; Li, Y.-J.; Huang, C.-C.; Chang, H.-T. *Adv. Funct. Mater.* **2015**, *25* (46), 7189–7199.
- (38) Li, Y.; Zhen, J.; Tian, Q.; Shen, C.; Zhang, L.; Yang, K.; Shang, L. *J. Colloid Interface Sci.* **2020**, 569, 235–243.
- (39) Ravindran Girija, A.; Balasubramanian, S.; Bright, R.; Cowin, A. J.; Goswami, N.; Vasilev, K. *ChemNanoMat* **2019**, *5* (9), 1176–1181.
- (40) Zheng, Y.; Liu, W.; Qin, Z.; Chen, Y.; Jiang, H.; Wang, X. *Bioconjug. Chem.* **2018**, *29* (9), 3094–3103.
- (41) Zheng, K.; Setyawati, M. I.; Leong, D. T.; Xie, J. *ACS Nano* **2017**, *11* (7), 6904–6910.
- (42) Yougbare, S.; Chang, T. K.; Tan, S. H.; Kuo, J. C.; Hsu, P. H.; Su, C. Y.; Kuo, T. R. *Int. J. Mol. Sci.* **2019**, *20* (12), 2924.
- (43) Glišić, B. Đ.; Djuran, M. I. *Dalton Trans.* **2014**, *43* (16), 5950–5969.
- (44) Shaw, C. F. *Chem. Rev.* **1999**, *99* (9), 2589–2600.
- (45) Boda, S. K.; Broda, J.; Schiefer, F.; Weber-Heynemann, J.; Hoss, M.; Simon, U.; Basu, B.; Jahn-Dechent, W. *Small* **2015**, *11* (26), 3183–3193.
- (46) Li, M.-B.; Tian, S.-K.; Wu, Z.; Jin, R. *Chem. Mater.* **2016**, *28* (4), 1022–1025.
- (47) Schmid, G.; Alexander, B. D.; Barthelmes, J.; Mueting, A. M.; Pignolet, L. H. *Inorg. Synth.* **1990**, 214–218.
- (48) Weare, W. W.; Reed, S. M.; Warner, M. G.; Hutchison, J. E. *J. Am. Chem. Soc.* **2000**, *122* (51), 12890–12891.
- (49) Brown, L. O.; Hutchison, J. E. *J. Am. Chem. Soc.* **1997**, *119* (50), 12384–12385.
- (50) Kenzler, S.; Fetzner, F.; Schrenk, C.; Pollard, N.; Frojd, A. R.; Clayborne, A. Z.; Schnepf, A. *Angew. Chem. Int. Ed.* **2019**, *58* (18), 5902–5905.
- (51) Konishi, K. In *Structure and bonding*; Springer International Publishing, 2014; pp 49–86.
- (52) Wijenayaka, L. A.; Ivanov, M. R.; Cheatum, C. M.; Haes, A. J. *J. Phys. Chem. C* **2015**, *119* (18), 10064–10075.
- (53) Schulz, F.; Vossmeier, T.; Bastús, N. G.; Weller, H. *Langmuir* **2013**, *29* (31), 9897–9908.
- (54) Petroski, J.; Chou, M. H.; Creutz, C. *Inorg. Chem.* **2004**, *43* (5), 1597–1599.
- (55) Sharma, R.; Holland, G. P.; Solomon, V. C.; Zimmermann, H.; Schiffenhaus, S.; Amin, S. A.; Buttry, D. A.; Yarger, J. L. *J. Phys. Chem. C* **2009**, *113* (37), 16387–16393.
- (56) Shu, P.; Zeng, J.; Tao, J.; Zhao, Y.; Yao, G.; Wan, Q. *Green Chem.* **2015**, *17* (4), 2545–2551.
- (57) Karschin, A.; Kläui, W.; Peters, W.; Spingler, B. *Eur. J. Inorg. Chem.* **2010**, 2010 (6), 942–946.
- (58) Snee, P. T. *Acc. Chem. Res.* **2018**, *51* (11), 2949–2956.
- (59) McKenzie, L. C.; Zaikova, T. O.; Hutchison, J. E. *J. Am. Chem. Soc.* **2014**, *136* (38), 13426–13435.
- (60) Woehrle, G. H.; Hutchison, J. E. *Inorg. Chem.* **2005**, *44* (18), 6149–6158.
- (61) Caragheorgheopol, A.; Chechik, V. *Phys. Chem. Chem. Phys.* **2008**, *10* (33), 5029.
- (62) Sanz, S.; Jones, L. A.; Mohr, F.; Laguna, M. *Organometallics* **2007**, *26* (4), 952–957.
- (63) Coffey, M. T.; Shaw, C. F.; Eidsness, M. K.; Watkins, J. W.; Elder, R. C. *Inorg. Chem.* **1986**, *25* (3), 333–339.
- (64) Krishnamoorthy, G.; Leus, I. V.; Weeks, J. W.; Wolloscheck, D.; Rybenkov, V. V.; Zgurskaya, H. I. *mBio* **2017**, *8* (5).
- (65) Delcour, A. H. *Biochim. Biophys. Acta Proteins Proteom.* **2009**, *1794* (5), 808–816.
- (66) Schmid, G.; Kreyling, W. G.; Simon, U. *Arch. Toxicol.* **2017**, *91* (9), 3011–3037.
- (67) Tsoli, M.; Kuhn, H.; Brandau, W.; Esche, H.; Schmid, G. *Small* **2005**, *1* (8-9), 841–844.
- (68) Ahmed, M. A. E.-G. E.-S.; Zhong, L.-L.; Shen, C.; Yang, Y.; Doi, Y.; Tian, G.-B. *Emerg. Microbes Infect.* **2020**, *9* (1), 868–885.
- (69) Nation, R. L.; Li, J. *Curr. Opin. Infect. Dis.* **2009**, *22* (6), 535–543.

- (70) Hooper, D. C. *Drugs* **1999**, 58 Suppl 2, 6–10.
- (71) Hong, Y.; Zeng, J.; Wang, X.; Drlica, K.; Zhao, X. *Proc. Natl. Acad. Sci. USA* **2019**, 116 (20), 10064–10071.
- (72) Hong, Y.; Li, L.; Luan, G.; Drlica, K.; Zhao, X. *Nat. Microbiol.* **2017**, 2 (12), 1667–1675.
- (73) Holmgren, A.; Lu, J. *Biochem. Biophys. Res. Commun.* **2010**, 396 (1), 120–124.
- (74) Olaitan, A. O.; Morand, S.; Rolain, J.-M. *Front. Microbiol.* **2014**, 5.
- (75) Yan, K.; Lok, C.-N.; Bierla, K.; Che, C.-M. *Chem. Commun.* **2010**, 46 (41), 7691–7693.
- (76) Epstein, T. D.; Wu, B.; Moulton, K. D.; Yan, M.; Dube, D. H. *ACS Infect. Dis.* **2019**, 5 (10), 1682–1687.
- (77) Fernandes, A. P.; Holmgren, A. *Antioxid. Redox Signal.* **2004**, 6 (1), 63–74.
- (78) Gusarov, I.; Nudler, E. *Proc Natl Acad Sci U S A* **2005**, 102 (39), 13855–13860.
- (79) Wu, B.; Yang, X.; Yan, M. *J. Med. Chem.* **2019**, 62 (17), 7751–7768.
- (80) Woodford, N.; Zhang, J.; Kaufmann, M. E.; Yarde, S.; Tomas, M. del M.; Faris, C.; Vardhan, M. S.; Dawson, S.; Cotterill, S. L.; Livermore, D. M. *J. Antimicrob. Chemother.* **2008**, 62 (6), 1265–1268.
- (81) Leifert, A.; Pan, Y.; Kinkeldey, A.; Schiefer, F.; Setzler, J.; Scheel, O.; Lichtenbeld, H.; Schmid, G.; Wenzel, W.; Jahnen-Dechent, W.; Simon, U. *Proc. Natl. Acad. Sci. USA* **2013**, 110 (20), 8004–8009.
- (82) Zhang, Q.; Yang, M.; Zhu, Y.; Mao, C. *Curr. Med. Chem.* **2018**, 25 (12), 1379–1396.
- (83) Galhardo, R. S.; Almeida, C. E.; Leitão, A. C.; Cabral-Neto, J. B. *J. Bacteriol.* **2000**, 182 (7), 1964–1968.
- (84) Liu, Y.; Meyer-Zaika, W.; Franzka, S.; Schmid, G.; Tsoli, M.; Kuhn, H. *Angew. Chem. Int. Ed.* **2003**, 42 (25), 2853–2857.
- (85) Zheng, K.; Setyawati, M. I.; Leong, D. T.; Xie, J. *Bioact. Mater.* **2021**, 6 (4), 941–950.
- (86) Woehrle, G. H.; Warner, M. G.; Hutchison, J. E. *J. Phys. Chem. B* **2002**, 106 (39), 9979–9981.
- (87) CLSI. *Performance standards for antimicrobial susceptibility testing. CLSI supplement M100*; 2019.