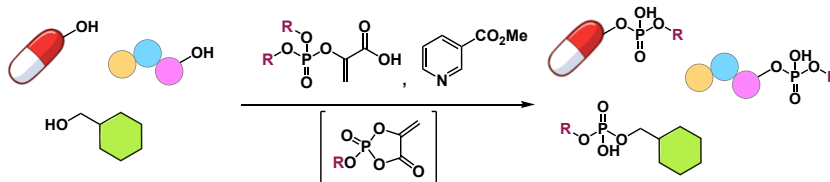


A Single-Step Asymmetric Phosphodiester Synthesis from Alcohols with Phosphoenolpyruvate Phosphodiester

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Single-step phosphodiester synthesis
 ■ Simple operation. ■ Functional group-tolerant.

Abstract Phosphodiesters are important structural motifs observed in a diverse field of molecular science. It is, thus, important to develop a simple and robust way to synthesize them from corresponding alcohols. Here we report a single-step asymmetric phosphodiester synthesis from alcohols with phosphoenolpyruvate phosphodiester as phosphoryl donors. This transformation allows for the use of various functionalized alcohols as substrates, and would be useful for diverse fields including biology and medicine.

Key words Phosphorylation, alcohol, phosphodiester, phosphoenolpyruvate, pentacoordinate phosphorous

Phosphate esters play pivotal roles in the life science.¹ They are divided into three classes; phosphomonoester, phosphodiester, and phosphotriester, each of which prevails among the molecules of life and shows distinct functions.² Phosphodiesters are, especially, fundamental motifs observed in biomolecules. For example, DNA, which is a chain of deoxyribonucleic acids connected through phosphodiester linkages, encodes the genetic information, and a phospholipid, which has two hydrophobic fatty acid chains and one polar phosphodiester head group, separates the interior of the cell from the outside environment. Inspired by those examples in life, a lot of functional molecules containing phosphodiesters are designed and used for various purposes, such as a methyl phosphate as an abiotic lipid analog for liposome-mediated drug delivery system,³ a “PC-inverse” choline phosphate designed to strongly interact with natural phosphatidylcholine in the plasma membrane,⁴ lower alkyl phosphates as enzyme inhibitors through transition-state mimicry⁵ or increasing binding affinity to a target enzyme,⁶ a methyl phosphate metabolically stabilizing a nucleoside analog,⁷ and lower alkyl phosphate prodrugs to increase water solubility of molecules (Figure 1).⁸

A straightforward way to access phosphodiesters is via phosphorylation of alcohol starting materials. There are mainly

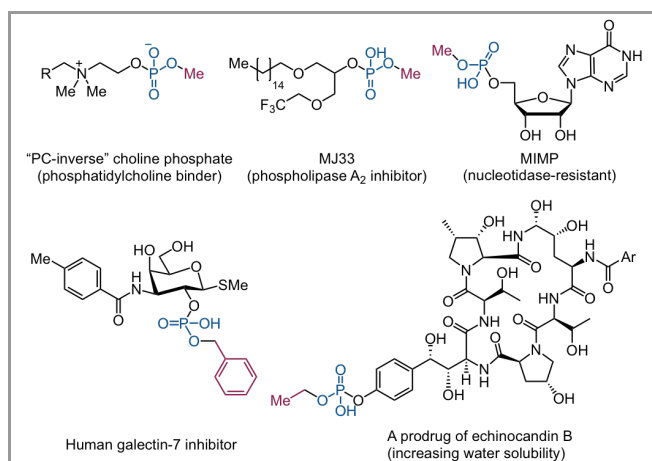
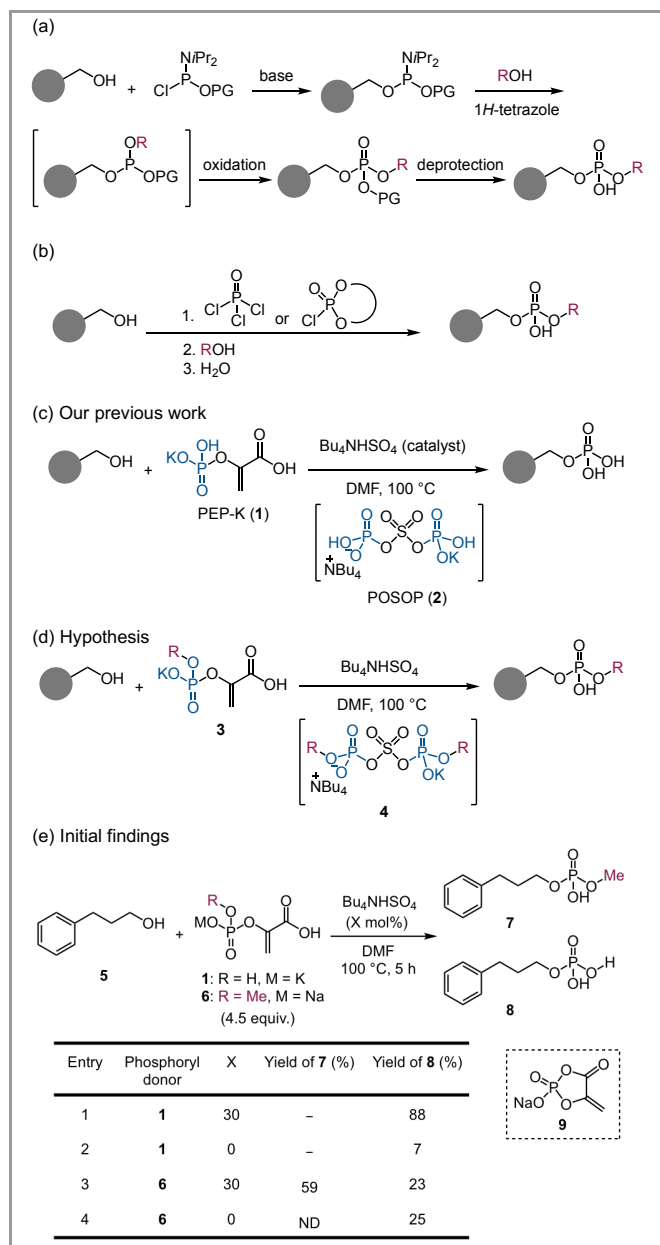


Figure 1 Representative bioactive phosphodiesters.

two ways to synthesizing asymmetric phosphodiesters from alcohols: (1) the formation of trivalent phosphite followed by oxidation, and (2) the reaction with a pentavalent phosphoryl donor (Scheme 1a,b). In a typical example of the first approach, two different alcohols are sequentially reacted with a trivalent chlorophosphoramidite reagent, the product of which is then oxidized to the protected pentavalent phosphate triester.⁹ Finally, the removal of the protecting group under acidic, reductive, or basic conditions (PG = alkyl, benzyl, or cyanoethyl, respectively) produces the desired phosphodiester. This method is quite robust, but requires four steps and is sometimes incompatible with substrates, which are sensitive to oxidation. In a typical example of the second approach, phosphoryl chloride and its derivatives are used as phosphoryl donors (Scheme 1b).¹⁰ Although reactions of two different alcohols and water with phosphoryl chloride can directly provide phosphodiesters, it is often difficult to control the reaction, resulting in complex mixtures with various substitution patterns. Although several masked phosphoryl donors with attenuated reactivity were developed to circumvent this issue,¹¹ multiple steps are still

required to synthesize asymmetric phosphodiester. Thus, an alternative method for phosphodiester synthesis with a simple operation and with high functional group tolerance is still in high demand. Here we report a phosphodiester synthesis from alcohols using phosphoenolpyruvate phosphodiester as phosphoryl donors. This transformation provides asymmetric phosphodiester from various functionalized alcohols in a single-step reaction.



Scheme 1 Phosphate ester synthesis from alcohols. ^aA representative P(III) chemistry for phosphodiester synthesis. ^bA representative P(V) chemistry for phosphodiester synthesis. ^cPhosphomonoester synthesis using phosphoenolpyruvate monopotassium salt (**1**) and tetrabutylammonium hydrogen sulfate (ref. 12). ^dInitial hypothesis for phosphodiester synthesis with phosphoenolpyruvate phosphomonoester (**3**). ^eInitial findings in phosphodiester synthesis using monomethyl phosphoenolpyruvate sodium salt (**6**). ND denotes not detected.

We previously reported a catalytic chemoselective phosphorylation of alcohols affording phosphomonoesters in a

single step with tetrabutylammonium hydrogen sulfate (TBAHS) and phosphoenolpyruvate monopotassium salt (PEP-K: **1**) as a catalyst and a phosphoryl donor, respectively (Scheme 1c and 1e, entry 1).¹² In this reaction, a mixed anhydride species, POSOP (**2**), which is catalytically generated from two molecules of phosphoenolpyruvate and one molecule of hydrogen sulfate ion, serves as a reactive phosphorylation species. We hypothesized that if we use phosphoenolpyruvate phosphomonoester (**3**) as a starting phosphoryl donor, an analogous phosphorylation species (**4**) would be formed, affording an asymmetric phosphodiester (Scheme 1d).

Based on this hypothesis, we reacted 3-phenyl-1-propanol (**5**) with monomethyl phosphoenolpyruvate sodium salt (**6**)¹³ in the presence of 30 mol% of TBAHS in DMF at 100 °C for 5 hours (Scheme 1e). On the contrary to our hypothesis, however, phosphodiester **7** was produced in 59% yield concomitant with phosphomonoester **8** obtained in 23% yield (entry 3). Intriguingly, the formation of phosphomonoester **8** was not TBAHS-dependent, and **8** was obtained in 25% yield by simply reacting substrate **5** with phosphoryl donor **6** at 100 °C in the absence of TBAHS (Scheme 1e, entry 4). Since a brief optimization of the reaction conditions of entry 3 for suppressing the generation of **8** did not result in success, and PEP-K (**1**) afforded the phosphomonoester **8** only in 7% yield in the absence of TBAHS on the contrary to **6** (entry 2 vs. entry 4), we turned our attention to this interesting catalyst-independent high reactivity of **6** and decided to explore the reactivity of phosphoenolpyruvate phosphoesters.

Clark and Kirby reported that phosphoenolpyruvate phosphoesters were hydrolyzed rapidly through an intermediate of cyclic acyl phosphate (e.g., **9** in Scheme 1e), which is generated via the intramolecular nucleophilic attack of the carboxylic acid moiety to the phosphorous atom followed by the elimination of the alcohol.¹⁴ The formation of phosphomonoester **8** with **6** in Scheme 1e could be explained by the attack of the substrate hydroxy group to the cyclic acyl phosphate **9** followed by elimination of the enol oxygen and methanolysis of the resulting anhydride. Higher reactivity of **6** than **1** in the absence of TBAHS (i.e., Scheme 1e, entry 2 vs. entry 4) was likely due to smaller negative charge at the phosphate group, thus facilitating the formation of putative intermediate **9**. Based on this idea, we envisioned that phosphoenolpyruvate phosphodiester would serve as a phosphoryl donor for the synthesis of asymmetric phosphodiester of various alcohols.

First, we reacted **5** with three equivalents of dimethyl phosphoenolpyruvate (**10**), which was prepared just before the phosphorylation reaction with Perkow reaction using bromopyruvic acid and trimethyl phosphite, in acetonitrile at 60 °C for 12 hours without any additive (Table 1, entry 1). The substrate **5** was nearly consumed, and methyl phosphate **7** was obtained in 9% yield as we expected. Since the product phosphate diester could be unstable in the protonated state under acidic conditions on heating, we examined addition of one equivalent of basic additives to neutralize the product phosphate in the reaction (entries 2–4). Among examined, pyridine worked the best to afford **7** in 67% yield (entry 4). Next, we examined several substituted pyridine derivatives (entries 5–7), and found that methyl nicotinate (**11**) increased the yield and **7** was obtained in 73% yield (entry 7). The methoxycarbonyl group at

the 3-position of the pyridine may attenuate nucleophilicity to suppress undesired reactions while maintaining its ability to neutralize the product in the reaction. The reaction completed within 6 hours to afford **7** in 72% yield (entry 8). Neither increasing the amount of phosphoryl donor **10** to 4.5 equivalents (entry 9) nor raising temperature to 80 °C (entry 10) improved the yield, while conducting the reaction at room temperature significantly decreased the yield and the most of the substrate was recovered unchanged (entry 11).

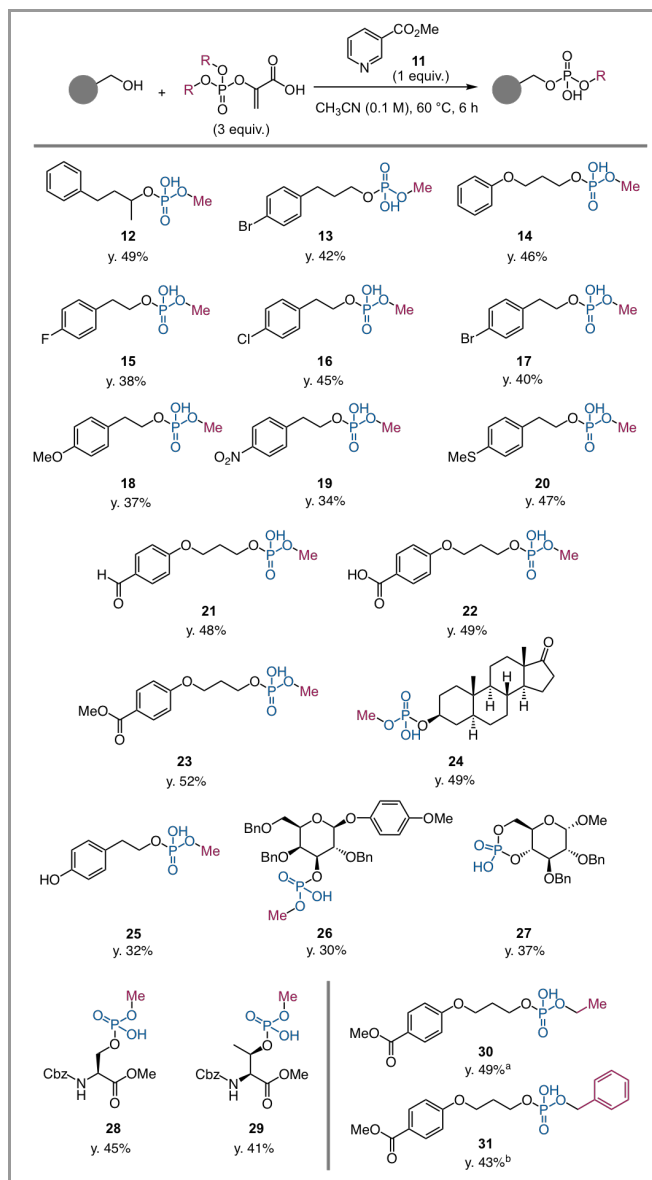
Table 1 Optimization of the Reaction Conditions

Entry	X	Additive	Time (h)	Temp. (°C)	Yield (%) ^a
1	3	—	12	60	9
2	3	NEt ₃	12	60	51
3	3	K ₂ HPO ₄	12	60	64
4	3	Pyridine	12	60	67
5	3	2,6-Lutidine	12	60	54
6	3	3-Methoxypyridine	12	60	42
7	3	Methyl nicotinate (11)	12	60	73
8	3	Methyl nicotinate (11)	6	60	72
9	4.5	Methyl nicotinate (11)	6	60	70
10	3	Methyl nicotinate (11)	6	80	71
11	3	Methyl nicotinate (11)	6	r.t.	20

^a Determined by ¹H NMR.

Having identified the optimized reaction conditions, we next investigated the substrate scope of this transformation. As shown in Scheme 2, a secondary alcohol is a competent substrate and underwent the phosphorylation to afford **12** in 49% yield. Substrates with a range of functional groups including fluoro- (**15**: y. 38%), chloro- (**16**, y. 45%), bromo- (**13**: y. 42%, **17**: y. 40%), alkoxy (**14**: y. 46%, **18**: 37%), nitro (**19**: y. 34%), and thiomethyl (**20**: y. 47%)-substituted phenyl groups afforded the product phosphodiester in moderate yields. It was noteworthy that the phosphodiester **20** was directly obtained considering that the thiomethyl group is prone to oxidation in a phosphoramidite chemistry (Scheme 1a). The reaction conditions were mild, and styrene byproducts derived from the β -elimination of the product phosphates were not observed in the transformation to **15–20**. An aldehyde (**21**: y. 48%), a carboxylic acid (**22**: y. 49%), an ester (**23**: y. 52%) and a ketone (**24**: 49%) were accommodated in this transformation. A chemoselective phosphorylation of an alcohol in the presence of a phenol was possible, and **25** was obtained in 32% yield, where we did not observe phenol-phosphorylated products in the crude mixture. An acid-labile glycosidic linkage was tolerated, and a phosphorylated carbohydrate (**26**) was obtained in 30% yield. Interestingly, when a glucose derivative with its 4,6-hydroxy groups unprotected, was used as a substrate, cyclic phosphate (**27**: y. 37%) was obtained, which could be generated via first methyl phosphate formation either on 4- or 6-hydroxy group followed by an intramolecular attack of the remaining hydroxy group to the methyl phosphate. Amino acid serine (**28**: y. 45%)

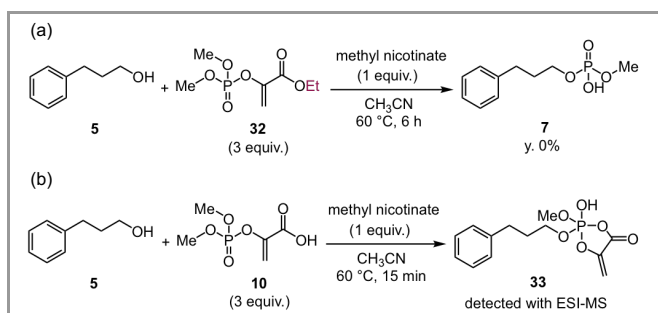
and threonine (**29**: y. 41%) were also good substrates for this transformation. Finally, synthesis of phosphate esters other than methyl were possible, and ethyl (**30**: y. 49%) and benzyl (**31**: y. 43%) phosphates were obtained in a single step reaction from an alcohol starting material (For the preparation of diethyl/dibenzyl phosphoenolpyruvate, see Supporting Information). Although the yield was moderate in most of the cases due to concomitant formation of small amounts of several unidentified byproducts, which could be mostly removed through the acid-base extraction (see Supporting Information), the concise operation to directly access phosphodiester is an advantage over existing methods (e.g., Scheme 1a,1b).



Scheme 2 Substrate scopes. The yields are isolated yield. ^a6 equivalents of the phosphoryl donor, 2 equivalents of methyl nicotinate, 80 °C, 13 hours. ^b13 hours.

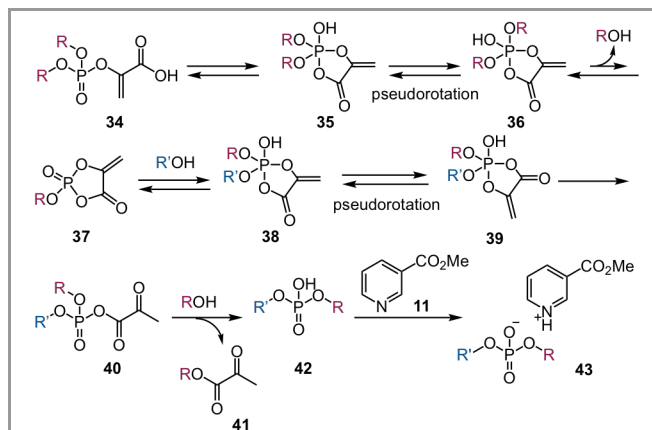
To get insights into the reaction mechanisms, we did several investigations. First, we conducted a reaction using a carboxy ester derivative of phosphoenolpyruvate phosphodiester as a phosphoryl donor. When **5** was treated with 3 equivalents of phosphoenolpyruvate triester **32** and one equivalent of methyl

nicotinate (**11**) under the optimized conditions, phosphorylation product **7** was not obtained at all, and the starting material **5** and **32** were recovered (Scheme 3a). This result is consistent with our idea that cyclic acyl phosphate, which results from intramolecular attack of the carboxylic acid moiety to the phosphorous atom of phosphoenolpyruvate phosphodiester, is an actual phosphorylation species. Second, to detect reaction intermediates, we conducted mass spectrometry analysis of the reaction mixture. When we analyzed the reaction mixture of alcohol **5** and dimethyl phosphoenolpyruvate **10** before the completion of the reaction (15 min) with electrospray ionization mass spectrometry (ESI-MS), we observed a species of m/z 301.1, which corresponds to the adduct of alcohol **5** to the cyclic acyl phosphate generated from **10**, such as **33** (Scheme 3b and Figure S1). This result also indicates that the cyclic acyl phosphate is the phosphorylating species.



Scheme 3 Investigations of the Reaction Mechanisms. ^aA control reaction using carboxy-ethylated phosphoryl donor **32**. ^bDetection of a reaction intermediate with ESI-MS.

Based on these results, a proposed reaction mechanism is shown in Scheme 4. Phosphoenolpyruvate phosphodiester **34** is in equilibrium with cyclized pentacoordinate phosphorous species **35**.¹⁵ After pseudorotation to place the alkoxy group into an apical position (**36**), one of the alcohol group (ROH) is released to generate a reactive cyclic acyl phosphate **37** in a reversible manner.¹⁴ The substrate alcohol (R'OH), then, attacks **37** to generate another pentacoordinate phosphorous species **38**.¹⁶ These elimination and addition of alcohols are all reversible. When **38** undergoes pseudorotation to place the enol oxygen into the apical position (**39**), the enol group can be irreversibly eliminated to generate acyclic acyl phosphate **40**. Acyclic acyl phosphate are known to prefer nucleophilic attack at the carbonyl carbon to at the phosphorous atom, serving as a good acylating agent.¹⁷ Thus, **40** reacts with the leaving group alcohol (ROH) to generate the protonated form of phosphodiester (**42**) and pyruvate ester of the alcohol (**41**). Methyl nicotinate stabilizes the product phosphodiester **42** by neutralizing to generate pyridinium phosphate salt **43** in the reaction mixture at 60 °C, which in turn provides protonated **42** after acidic work-up.



Scheme 4 A proposed mechanism of the phosphodiester synthesis with phosphoenolpyruvate phosphodiester

In conclusion, we developed a single step asymmetric phosphodiester synthesis from alcohols and phosphoenolpyruvate phosphodiester. This reaction provides a variety of functionalized asymmetric phosphodiester in a short and concise procedure, which would be useful in diverse field including biology and medicine.

Funding Information

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Supporting Information for

A Single-Step Asymmetric Phosphodiester Synthesis from Alcohols with Phosphoenolpyruvate Phosphodiester

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Materials and Methods

General

NMR spectra were recorded on JEOL ECS400 (400 MHz for ^1H NMR, 100 MHz for ^{13}C NMR and 159 MHz for ^{31}P NMR) spectrometers. Chemical shifts were reported in ppm on the δ scale relative to residual CHCl_3 ($\delta = 7.24$ for ^1H NMR and $\delta = 77.0$ for ^{13}C NMR), CHD_2OD ($\delta = 3.31$ for ^1H NMR and $\delta = 49.0$ for ^{13}C NMR), or CHD_2CN ($\delta = 1.94$ for ^1H NMR and $\delta = 118.2$ for ^{13}C NMR) as an internal reference, and 85% H_3PO_4 aq. ($\delta = 0$ for ^{31}P NMR) as an external reference, respectively.

Preparative HPLC was conducted by using a JASCO HPLC system equipped with a UV-2075 spectrometer, PU-4086 pumps, a DG-4580 degasser, and an MX-2080-32 mixer, and a CO-4065 column oven. General eluent was linear gradient of MeCN in 0.1% TFA (LC-MS grade) aq. (10% acetonitrile for 5 min, followed by a linear gradient of 10–100% acetonitrile over 60 min. YMC-Triart C18, 254 nm), unless otherwise stated. C18 reverse phase column was used at 40 $^\circ\text{C}$, unless otherwise stated.

MQ means distilled water purified with a Millipore Milli-Q water purification system (Merck K. Ga. Co., Darmstadt, Germany).

LC-MS analysis was conducted by using an Agilent Technologies LC-MS (ESI) system equipped with a 1260 Infinity High Performance Degasser, an Agilent 1260 Infinity Binary Pump, a 1260 Infinity Standard Autosampler, a 1290 Infinity Thermostatted Column Compartment, a 1260 Infinity Variable Wavelength Detector, and an Agilent 6120 Single Quadrupole LC-MS or Shimadzu LCMS-2020. Retention times (Rt/min) were recorded using a gradient elution method of 2–90% B over 9 min, where solution A consisted of water (buffered with 0.1% HCO_2H) and solution B consisted of acetonitrile (LC-MS grade) unless otherwise stated. C18 reverse phase column (2.0 \times 50 mm; YMC-Triart C18; YMC Co., Ltd.) was used at 40 $^\circ\text{C}$ at a flow rate of 0.2 mL/min. The eluent was monitored by absorbance at 230 nm, unless otherwise stated.

Reactions were carried out in dry solvents under an argon atmosphere, unless otherwise stated. Other reagents were used as received from commercial sources (Aldrich, TCI, or FUJIFILM Wako), unless otherwise stated.

Preparation of dimethyl phosphoenolpyruvate (10)

Dimethyl phosphoenolpyruvate was prepared according to the literature.¹ A heat-gun dried and argon-flushed test tube equipped with a magnetic stirrer bar was charged with 3-bromopyruvic acid (1.0 equiv.) and diethyl ether (1.66 M). Then, a solution of trimethyl phosphite (1.0 equiv.) in diethyl ether (1.66 M) was added dropwise at a gently reflux rate, and the reaction mixture was stirred at r.t. After 1 h, the solvent was removed under high vacuum, and the resulting dimethyl phosphoenolpyruvate was used without further purification.

General procedures for phosphorylation of alcohols

Procedure S (Standard)

To the freshly-prepared dimethyl phosphoenolpyruvate (300 μ mol, 3.0 equiv.), were added sequentially acetonitrile (0.10 M), substrate (100 μ mol, 1.0 equiv.), and methyl nicotinate (13.7 mg, 100 μ mol, 1.0 equiv.). The mixture was warmed to 60 °C and stirred for 6 h. Then the reaction mixture was cooled to r.t. and concentrated. After adding saturated Na₂CO₃ aq., the mixture was washed three times with diethyl ether. The aqueous layer was acidified with concentrated HCl to pH 1, and extracted five times with dichloromethane. The combined organic layers were dried over with Na₂SO₄, filtered and concentrated to afford crude product, which was purified by preparative HPLC to afford the corresponding phosphorylated product.

Procedure A

To the freshly-prepared dimethyl phosphoenolpyruvate (300 μ mol, 3.0 equiv.), were added sequentially acetonitrile (1.0 mL, 0.10 M), substrate (100 μ mol, 1.0 equiv.), and methyl nicotinate (13.7 mg, 100 μ mol, 1.0 equiv.). The mixture was warmed to 60 °C and stirred for 6 h. Then the reaction mixture was cooled to r.t. and concentrated. After adding 1N HCl, the mixture was extracted five times with dichloromethane. The combined organic layers were dried over with Na₂SO₄, filtered and concentrated to afford crude product, which was purified by preparative HPLC to afford the corresponding phosphorylated product.

Procedure B

To the freshly-prepared dimethyl phosphoenolpyruvate (3.0 equiv.), were added sequentially acetonitrile (1.0 mL, 0.10 M), substrate (100 μ mol, 1.0 equiv.), and methyl nicotinate (13.7 mg, 100 μ mol, 1.0 equiv.). The mixture was warmed to 60 °C and stirred for 6 h. Then the reaction mixture was cooled to r.t., concentrated and purified by preparative HPLC to give the corresponding phosphorylated product.

Procedure C

A heat-gun dried and argon-flushed test tube equipped with a magnetic stirrer bar was charged with 3-bromopyruvic acid (600 μ mol, 6.0 equiv.) and diethyl ether (1.66M). Then, a solution of the triethyl phosphite (600 μ mol, 6.0 equiv.) in diethyl ether (1.66M) was added dropwise at a gently reflux rate, and the reaction mixture was stirred at r.t. for 2 h. The solvent was removed under high vacuum, and the resulting diethyl

phosphoenolpyruvate was used without further purification.

To the freshly-prepared diethyl phosphoenolpyruvate (6.0 equiv.), were added sequentially acetonitrile (1.0 mL, 0.10 M), substrate (100 μ mol, 1.0 equiv.), and methyl nicotinate (200 μ mol, 2.0 equiv.). The mixture was warmed to 80 °C and stirred for 13 h. Then the reaction mixture was cooled to r.t. and concentrated. After adding saturated Na_2CO_3 aq, the mixture was washed three times with diethyl ether. The aqueous layer was acidified with concentrated HCl to pH 1, and extracted five times with dichloromethane. The combined organic layers were dried over Na_2SO_4 , filtered and concentrated to afford crude product, which was purified by preparative HPLC to afford the corresponding phosphorylated product.

Procedure D

A heat-gun dried and argon-flushed test tube equipped with a magnetic stirrer bar was charged with 3-bromopyruvic acid (3.0 equiv.) and diethyl ether (1.66M). Then, a solution of the tribenzyl phosphite (3.0 equiv.) in diethyl ether (1.66M) was added dropwise at a gently reflux rate, and the reaction mixture was stirred at r.t. for 2 h. The solvent was removed under high vacuum, and the resulting residue was washed with anhydrous hexane three times. Remained hexane was removed under high vacuum, and the resulting dibenzyl phosphoenolpyruvate was used without further purification.

To the freshly-prepared dibenzyl phosphoenolpyruvate (3.0 equiv.), were added sequentially acetonitrile (1.0 mL, 0.10 M), substrate (100 μ mol, 1.0 equiv.), and methyl nicotinate (13.7 mg, 100 μ mol, 1.0 equiv.). The mixture was warmed to 60 °C and stirred for 13 h. Then the reaction mixture was cooled to r.t. and concentrated. After adding saturated Na_2CO_3 aq, the reaction mixture was washed three times with diethyl ether. The aqueous layer was acidified with concentrated HCl to pH 1, and extracted five times with dichloromethane. The combined organic layers were dried over Na_2SO_4 , filtered and concentrated to afford crude product, which was purified by preparative HPLC to afford the corresponding phosphorylated product.

Reaction of **32** with 3-phenyl-1-propanol (Scheme 3a)

A heat-gun dried and argon-flushed test tube equipped with a magnetic stirrer bar was charged with ethyl 3-bromo-2-oxopropanoate (58.5 mg, 300 μ mol, 3.0 equiv.) and diethyl ether (181 μ L). Then, a solution of trimethyl phosphite (37.2 mg, 300 μ mol, 3.0 equiv.) in diethyl ether (181 μ L) was added dropwise at a gently reflux rate, and the reaction mixture was stirred at r.t. for 1 h. The solvent was removed under high vacuum, and the resulting **32** was used without further purification.

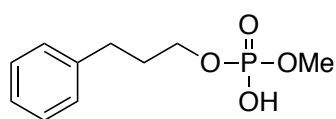
To the freshly-prepared **32**, were added sequentially acetonitrile (1.0 mL, 0.10 M), 3-phenyl-1-propanol (13.2 mg, 100 μ mol, 1.0 equiv.), and methyl nicotinate (13.7 mg, 100 μ mol, 1.0 equiv.). The mixture was warmed to 60 °C and stirred for 6 h. Then the reaction mixture was cooled to r.t. and concentrated. After diluting adding

saturated Na_2CO_3 aq, the mixture was extracted three times with diethyl ether (organic layer 1). The aqueous layer was acidified with concentrated HCl to pH 1, and extracted five times with dichloromethane (organic layer 2). The two organic layers (1 and 2) were dried over Na_2SO_4 , filtered, concentrated, and analyzed with ^1H NMR, respectively.

Detection of 33 with LC-MS (Scheme 3b).

To the freshly-prepared dimethyl phosphoenolpyruvate (3.0 equiv.), were added sequentially acetonitrile (1.0 mL, 0.10 M), 3-phenyl-1-propanol (13.2 mg, 100 μmol , 1.0 equiv.), and methyl nicotinate (13.7 mg, 100 μmol , 1.0 equiv.). The mixture was warmed to 60 $^\circ\text{C}$ and stirred for 15 min. Then 1.0 μL of the reaction mixture was diluted with 99.0 μL acetonitrile, and was analyzed by LC-MS (ESI-MS) using a gradient elution method of 2 – 90% acetonitrile over 9 min.

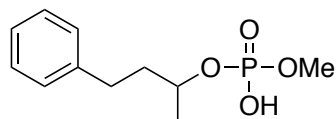
Characterization of phosphorylated products



Methyl 3-phenylpropyl hydrogen phosphate (7):

General procedure S (9.64 mg, 0.0418 mmol, 42% isolated yield)

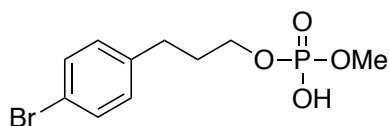
^1H NMR (CDCl_3 , 400 MHz) δ 7.34-7.31 (m, 2H), 7.25-7.22 (m, 3H), 4.08 (dt, J = 6.3 Hz, 6.4 Hz, 2H), 3.78 (d, J = 11.4 Hz, 3H), 2.76 (t, J = 7.9 Hz, 2H), 2.07-2.00 (m, 2H); ^{13}C NMR (CDCl_3 , 100 MHz) δ 141.0, 128.5, 128.4, 126.0, 66.8 (d, J = 5.7 Hz), 53.9 (d, J = 5.7 Hz), 31.8 (d, J = 7.6 Hz), 31.6; ^{31}P NMR (CDCl_3 , 159 MHz) δ 1.9; ESI-MS m/z 229.1 $[\text{M}-\text{H}]^-$; HRMS calcd for $[\text{C}_{10}\text{H}_{14}\text{O}_4\text{P}]^-$ requires m/z 229.0635; found 229.0635.



Methyl (4-phenylbutan-2-yl) hydrogen phosphate (12):

General procedure S (11.9 mg, 0.0488 mmol, 49% yield)

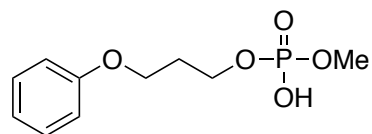
^1H NMR (CDCl_3 , 400 MHz) δ 7.34-7.30 (m, 2H), 7.24-7.20 (m, 3H), 4.56-4.49 (m, 1H), 3.78 (d, J = 11.4 Hz, 3H), 2.84-2.77 (m, 1H), 2.74-2.66 (m, 1H), 2.06-1.97 (m, 1H), 1.91-1.82 (m, 1H), 1.41 (d, J = 6.4 Hz, 3H); ^{13}C NMR (CDCl_3 , 100 MHz) δ 141.5, 128.4, 128.4, 125.9, 75.5 (d, J = 5.7 Hz), 53.8 (d, J = 5.7 Hz), 39.2 (d, J = 6.7 Hz), 31.4, 21.5 (d, J = 2.9 Hz); ^{31}P NMR (CDCl_3 , 159 MHz) δ 1.3; ESI-MS m/z 243.1 $[\text{M}-\text{H}]^-$; HRMS calcd for $[\text{C}_{11}\text{H}_{16}\text{O}_4\text{P}]^-$ requires m/z 243.0792; found 243.0792.



Methyl 3-(4-bromophenyl)propyl hydrogen phosphate (13):

General procedure S (13.0 mg, 0.0421 mmol, 42% yield)

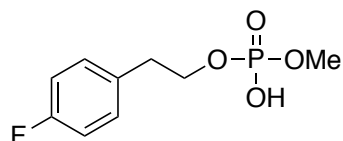
^1H NMR (CDCl_3 , 400 MHz) δ 7.37 (d, J = 8.5 Hz, 2H), 7.04 (d, J = 8.5 Hz, 2H), 3.99 (dt, J = 6.6 Hz, 6.6 Hz, 2H), 3.72 (d, J = 11.4 Hz, 3H), 2.65 (t, J = 7.8 Hz, 2H), 1.97-1.90 (m, 2H); ^{13}C NMR (CDCl_3 , 100 MHz) δ 139.9, 131.5, 130.2, 119.8, 66.5 (d, J = 5.7 Hz), 54.0 (d, J = 5.7 Hz), 31.6 (d, J = 7.6 Hz), 31.0; ^{31}P NMR (CDCl_3 , 159 MHz) δ 2.0; ESI-MS m/z 307.0 $[\text{M}-\text{H}]^-$; HRMS calcd for $[\text{C}_{10}\text{H}_{13}\text{BrO}_4\text{P}]^-$ requires m/z 306.9740; found 306.9740.



Methyl (3-phenoxypropyl) hydrogen phosphate (14):

General procedure S (11.3 mg, 0.0459 mmol, 46% yield)

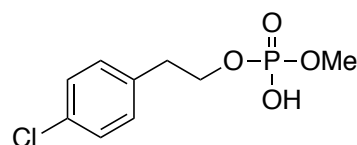
^1H NMR (CDCl_3 , 400 MHz) δ 7.26-7.23 (dd, J = 7.3 Hz, 8.9 Hz, 2H), 6.91 (t, J = 7.3 Hz, 1H), 6.86 (d, J = 8.9 Hz, 2H), 4.19 (dt, J = 6.4 Hz, 6.4 Hz, 2H), 4.04 (t, J = 6.0 Hz, 2H), 3.67 (d, J = 11.4 Hz, 3H), 2.14-2.08 (m, 2H); ^{13}C NMR (CDCl_3 , 100 MHz) δ 158.7, 129.4, 120.8, 114.4, 64.2 (d, J = 5.7 Hz), 63.5, 53.9 (d, J = 5.7 Hz), 30.1 (d, J = 7.6 Hz); ^{31}P NMR (CD_3OD , 159 MHz) δ 0.2; ESI-MS m/z 245.1 $[\text{M}-\text{H}]^-$; HRMS calcd for $[\text{C}_{10}\text{H}_{14}\text{O}_5\text{P}]^-$ requires m/z 245.0584; found 245.0584.



Methyl (4-fluorophenethyl) hydrogen phosphate (15):

General procedure S (8.94 mg, 0.0382 mmol, 38% yield)

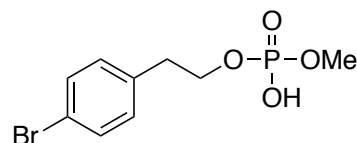
^1H NMR (CDCl_3 , 400 MHz) δ 7.15 (dd, $J = 5.5$ Hz, 8.8 Hz, 2H), 6.95 (dd, $J = 8.8$ Hz, 8.8 Hz, 2H), 4.14 (dt, $J = 7.2$ Hz, 7.2 Hz, 2H), 3.60 (d, $J = 11.4$ Hz, 3H), 2.92 (t, $J = 7.2$ Hz, 2H); ^{13}C NMR (CDCl_3 , 100 MHz) δ 161.7 (d, $J = 248.9$ Hz), 132.8 (d, $J = 2.9$ Hz), 130.4 (d, $J = 7.6$ Hz), 115.3 (d, $J = 21.9$ Hz), 67.7 (d, $J = 6.7$ Hz), 53.9 (d, $J = 6.7$ Hz), 35.8 (d, $J = 6.7$ Hz); ^{31}P NMR (CDCl_3 , 159 MHz) δ 1.5; ESI-MS m/z 233.1 $[\text{M}-\text{H}]^-$; HRMS calcd for $[\text{C}_9\text{H}_{11}\text{FO}_4\text{P}]^-$ requires m/z 233.0385; found 233.0385.



Methyl (4-chlorophenethyl) hydrogen phosphate (16):

General procedure S (11.3 mg, 0.0451 mmol, 45% yield)

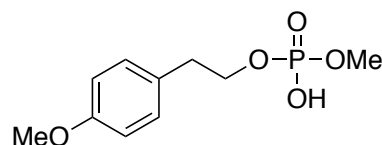
^1H NMR (CDCl_3 , 400 MHz) δ 7.24 (d, $J = 8.7$ Hz, 2H), 7.13 (d, $J = 8.7$ Hz, 2H), 4.14 (dt, $J = 6.9$ Hz, 6.9 Hz, 2H), 3.61 (d, $J = 11.4$ Hz, 3H), 2.92 (t, $J = 6.9$ Hz, 2H); ^{13}C NMR (CDCl_3 , 100 MHz) δ 135.6, 132.5, 130.3, 128.6, 67.5 (d, $J = 5.7$ Hz), 53.9 (d, $J = 5.7$ Hz), 35.9 (d, $J = 6.7$ Hz); ^{31}P NMR (CDCl_3 , 159 MHz) δ 1.4; ESI-MS m/z 249.1 $[\text{M}-\text{H}]^-$; HRMS calcd for $[\text{C}_9\text{H}_{11}\text{ClO}_4\text{P}]^-$ requires m/z 249.0089; found 249.0089.



Methyl (4-bromophenethyl) hydrogen phosphate (17):

General procedure S (11.9 mg, 0.0403 mmol, 40% yield)

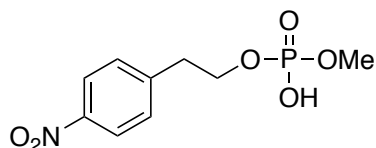
^1H NMR (CDCl_3 , 400 MHz) δ 7.33 (d, $J = 8.5$ Hz, 2H), 7.02 (d, $J = 8.5$ Hz, 2H), 4.07 (dt, $J = 6.9$ Hz, 6.9 Hz, 2H), 3.54 (d, $J = 11.0$ Hz, 2H), 2.84 (t, $J = 6.9$ Hz, 2H); ^{13}C NMR (CDCl_3 , 100 MHz) δ 136.2, 131.5, 130.7, 120.5, 67.3 (d, $J = 5.7$ Hz), 53.9 (d, $J = 5.7$ Hz), 36.0 (d, $J = 7.6$ Hz); ^{31}P NMR (CDCl_3 , 159 MHz) δ 1.2; ESI-MS m/z 294.1 $[\text{M}-\text{H}]^-$; HRMS calcd for $[\text{C}_9\text{H}_{11}\text{BrO}_4\text{P}]^-$ requires m/z 292.9584; found 292.9584.



Methyl (4-methoxyphenethyl) hydrogen phosphate (18):

General procedure S (9.08 mg, 0.0368 mmol, 37% yield)

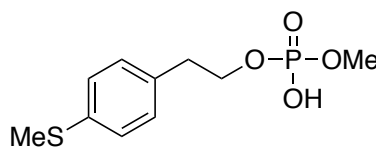
^1H NMR (CDCl_3 , 400 MHz) δ 7.10 (d, $J = 8.7$ Hz, 2H), 6.80 (d, $J = 8.7$ Hz, 2H), 4.12 (dt, $J = 7.1$ Hz, 7.1 Hz, 2H), 3.75 (s, 3H), 3.60 (d, $J = 11.0$, 3H), 2.89 (t, $J = 7.1$ Hz, 2H); ^{13}C NMR (CDCl_3 , 100 MHz) δ 158.3, 130.0, 129.2, 113.9, 68.0 (d, $J = 5.7$ Hz), 55.2, 53.8 (d, $J = 4.8$ Hz), 35.8 (d, $J = 6.7$ Hz); ^{31}P NMR (CDCl_3 , 159 MHz) δ 1.6; ESI-MS m/z 245.1 $[\text{M}-\text{H}]^-$; HRMS calcd for $[\text{C}_{10}\text{H}_{14}\text{O}_5\text{P}]^-$ requires m/z 245.0584; found 245.0584.



Methyl (4-nitrophenethyl) hydrogen phosphate (19):

General procedure S (8.98 mg, 0.0344 mmol, 34% yield)

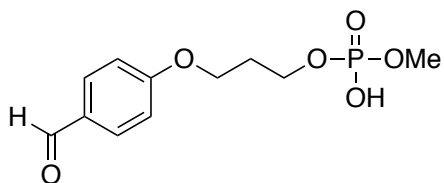
^1H NMR (CDCl_3 , 400 MHz) δ 8.08 (d, J = 8.5 Hz, 2H), 7.32 (d, J = 8.5 Hz, 2H), 4.16 (dt, J = 6.9 Hz, 6.9 Hz, 2H), 3.55, (d, J = 11.4 Hz, 3H), 3.00 (t, J = 6.9 Hz, 2H); ^{13}C NMR (CDCl_3 , 100 MHz) δ 146.9, 145.0, 129.9, 123.7, 66.8 (d, J = 5.7 Hz), 54.0 (d, J = 5.7 Hz), 36.3 (d, J = 7.6 Hz); ^{31}P NMR (CDCl_3 , 159 MHz) δ 1.2; ESI-MS m/z 260.0 $[\text{M}-\text{H}]^-$; HRMS calcd for $[\text{C}_9\text{H}_{11}\text{NO}_6\text{P}]^-$ requires m/z 260.0330; found 260.0330.



Methyl (4-(methylthio)phenethyl) hydrogen phosphate (20):

General procedure S (12.4 mg, 0.0474 mmol, 47% yield)

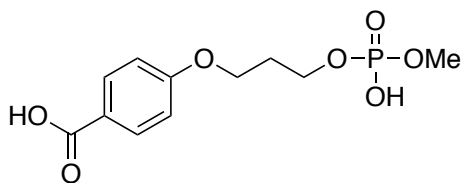
^1H NMR (CDCl_3 , 400 MHz) δ 7.17 (d, J = 8.0 Hz, 2H), 7.12 (d, J = 8.0 Hz, 2H), 4.17 (dt, J = 7.3 Hz, 7.3 Hz, 2H), 3.63 (d, J = 11.4 Hz, 3H), 2.93 (t, J = 4.7 Hz, 2H), 2.44 (s, 3H); ^{13}C NMR (CDCl_3 , 100 MHz) δ 136.6, 133.9, 129.5, 126.9, 67.8 (d, J = 6.7 Hz), 54.0 (d, J = 5.7 Hz), 36.0 (d, J = 7.6 Hz), 16.0; ^{31}P NMR (CDCl_3 , 159 MHz) δ 1.5; ESI-MS m/z 261.1 $[\text{M}-\text{H}]^-$; HRMS calcd for $[\text{C}_{10}\text{H}_{14}\text{O}_4\text{PS}]^-$ requires m/z 261.0356; found 261.0356.



3-(4-Formylphenoxy)propyl methyl hydrogen phosphate (21):

General procedure S (13.2 mg, 0.0482 mmol, 48% yield)

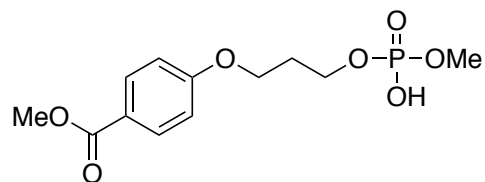
^1H NMR (CDCl_3 , 400 MHz) δ 9.85 (s, 1H), 7.80 (d, J = 8.5 Hz, 2H), 6.97 (d, J = 8.5 Hz, 2H), 4.24-4.19 (m, 2H), 4.14 (t, J = 5.7 Hz, 2H), 3.69 (d, J = 11.0 Hz, 3H), 2.17-2.14 (m, 2H); ^{13}C NMR (CDCl_3 , 100 MHz) δ 190.9, 163.7, 132.0, 130.0, 114.7, 64.0 (d, J = 4.8 Hz), 64.0, 54.1 (d, J = 3.8 Hz), 29.8 (d, J = 6.7 Hz); ^{31}P NMR (CDCl_3 , 159 MHz) δ 1.6; ESI-MS m/z 273.1 $[\text{M}-\text{H}]^-$; HRMS calcd for $[\text{C}_{11}\text{H}_{14}\text{O}_6\text{P}]^-$ requires m/z 273.0534; found 273.0534.



4-(3-((Hydroxy(methoxy)phosphoryl)oxy)propoxy)benzoic acid (22):

General procedure S (14.2 mg, 0.0489 mmol, 49% yield)

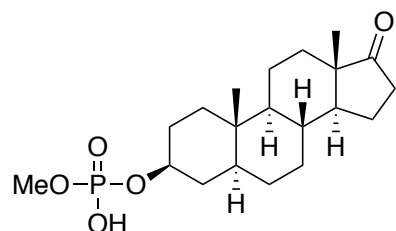
^1H NMR (CD_3OD , 400 MHz) δ 7.87 (d, J = 8.9 Hz, 2H), 6.90 (d, 8.9 Hz, 2H), 4.11-4.07 (m, 4H), 3.57 (d, J = 11.4 Hz, 3H), 2.09-2.03 (m, 2H); ^{13}C NMR (CD_3OD , 100 MHz) δ 169.8, 164.3, 132.9, 124.1, 115.2, 65.2, 64.7 (d, J = 5.7 Hz), 54.2 (d, J = 5.7 Hz), 31.2 (d, J = 6.7 Hz); ^{31}P NMR (CD_3OD , 159 MHz) δ 0.7; ESI-MS m/z 289.1 $[\text{M}-\text{H}]^-$; HRMS calcd for $[\text{C}_{11}\text{H}_{14}\text{O}_7\text{P}]^-$ requires m/z 289.0483; found 289.0483.



Methyl 4-(3-((hydroxy(methoxy)phosphoryl)oxy)propoxy)benzoate (23):

General procedure S (15.7 mg, 0.0516 mmol, 52% yield)

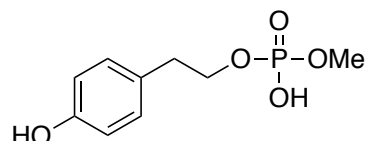
^1H NMR (CDCl_3 , 400 MHz) δ 7.95 (d, J = 8.7 Hz, 2H), 6.88 (d, J = 8.7 Hz, 2H), 4.21 (dt, J = 6.4 Hz, 6.4 Hz, 2H), 4.10 (t, J = 6.4 Hz, 2H), 3.86 (s, 3H), 3.68 (d, J = 11.4 Hz, 3H), 2.17-2.11 (m, 2H); ^{13}C NMR (CDCl_3 , 100 MHz) δ 166.9, 162.4, 131.6, 122.7, 114.0, 64.1 (d, J = 4.8 Hz), 63.7, 54.0 (d, J = 5.7 Hz), 51.9, 29.9 (d, J = 6.7 Hz); ^{31}P NMR (CDCl_3 , 159 MHz) δ 1.6; ESI-MS m/z 303.1 $[\text{M}-\text{H}]^-$; HRMS calcd for $[\text{C}_{12}\text{H}_{16}\text{O}_7\text{P}]^-$ requires m/z 303.0639; found 303.0639.



(3S,5S,8R,9S,10S,13S,14S)-10,13-Dimethyl-17-oxohexadecahydro-1H-cyclopenta[a]phenanthren-3-yl methyl hydrogen phosphate (24):

General procedure S (19.0 mg, 0.0494 mmol, 49% yield)

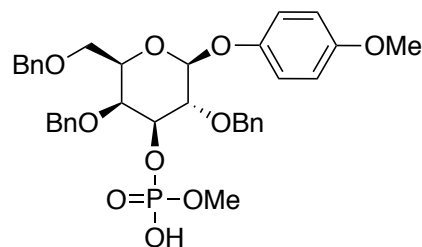
^1H NMR (CDCl_3 , 400 MHz) δ 4.24-4.21 (m, 1H), 3.70 (d, J = 11.0 Hz, 3H), 2.41 (dd, J = 9.2 Hz, 19.6 Hz, 1H), 2.08-1.85 (m, 3H), 1.78-1.71 (m, 4H), 1.62-1.58 (m, 2H), 1.53-1.41 (m, 3H), 1.33-1.08 (m, 6H), 1.00-0.89 (m, 2H), 0.83 (s, 3H), 0.81 (s, 3H), 0.68-0.62 (m, 1H); ^{13}C NMR (CDCl_3 , 100 MHz) δ 221.2, 78.1 (d, J = 5.7 Hz), 54.3, 53.9 (d, J = 5.7 Hz), 51.3, 47.7, 44.6, 36.7, 35.8, 35.6 (d, J = 3.8 Hz), 35.4, 34.9, 31.5, 30.8, 29.1 (d, J = 4.8 Hz), 28.2, 21.7, 20.4, 13.8, 12.2; ^{31}P NMR (CD_3OD , 159 MHz) δ 1.2; ESI-MS m/z 383.3 $[\text{M}-\text{H}]^-$; HRMS calcd for $[\text{C}_{20}\text{H}_{32}\text{O}_5\text{P}]^-$ requires m/z 383.1993; found 383.1993.



4-Hydroxyphenethyl methyl hydrogen phosphate (25):

General procedure S (7.49 mg, 0.0323 mmol, 32% yield)

^1H NMR (CD_3OD , 400 MHz) δ 6.97 (d, J = 8.7 Hz, 2H), 6.62 (d, J = 8.7 Hz, 2H), 4.01 (dt, J = 7.1 Hz, 7.1 Hz, 2H), 3.49 (d, J = 11.0 Hz, 3H), 2.77 (t, J = 7.1 Hz, 2H); ^{13}C NMR (CD_3OD , 100 MHz) δ 157.2, 131.1, 129.6, 116.2, 69.2 (d, J = 5.7 Hz), 54.2 (d, J = 5.7 Hz), 36.9 (d, J = 6.7 Hz); ^{31}P NMR (CD_3OD , 159 MHz) δ 0.4; ESI-MS m/z 231.1 $[\text{M}-\text{H}]^-$; HRMS calcd for $[\text{C}_9\text{H}_{12}\text{O}_5\text{P}]^-$ requires m/z 231.0428; found 231.0428.

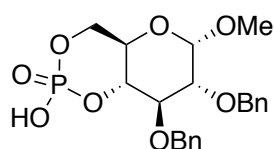


(2R,3S,4S,5R,6S)-3,5-Bis(benzyloxy)-2-((benzyloxy)methyl)-6-(4-methoxyphenoxy)tetrahydro-2H-pyran-4-yl methyl hydrogen phosphate (26):

General procedure A (19.8 mg, 0.0304 mmol, 30% yield)

^1H NMR (CDCl_3 , 400 MHz) δ 7.35-7.18 (m, 15H), 6.92 (d, J = 7.8, 2H), 6.71 (d, J = 7.8, 2H), 4.92 (d, J = 11.4 Hz, 1H), 4.88 (d, J = 11.9 Hz, 1H), 4.78-4.74 (m, 2H), 4.53 (d, J = 11.9 Hz), 4.28-4.29 (m, 3H), 4.03 (d, J = 2.7 Hz, 1H), 3.98 (dd, J = 8.7 Hz, 8.7

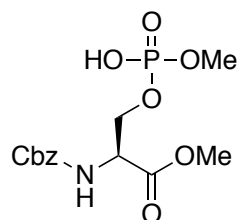
Hz, 1H), 3.70 (s, 3H), 3.56-3.49 (m, 5H), 3.43-3.41 (m, 1H); ^{13}C NMR (CDCl_3 , 100 MHz) δ 155.2, 151.3, 138.1, 137.9, 137.8, 128.4, 128.3, 128.3, 128.2, 127.8, 127.7, 127.7, 118.4, 114.4, 102.6, 79.7 (d, $J = 5.7$ Hz), 77.4 (d, $J = 6.7$ Hz), 77.2, 75.1 (d, $J = 7.6$ Hz), 74.7, 73.5, 73.2, 68.6, 55.6, 54.3 (d, $J = 6.7$ Hz); ^{31}P NMR (CDCl_3 , 159 MHz) δ 1.2; ESI-MS m/z 325.2 $[\text{M}-2\text{H}]^{2-}$; HRMS calcd for $[\text{C}_{35}\text{H}_{38}\text{O}_{10}\text{P}]^-$ requires m/z 649.2208; found 649.2208.



(4aR,6S,7R,8R,8aR)-7,8-Bis(benzyloxy)-2-hydroxy-6-methoxyhexahydropyrano[3,2-d][1,3,2]dioxaphosphinine 2-oxide (27):

General procedure A (16.1 mg, 0.0368 mmol, 37% yield)

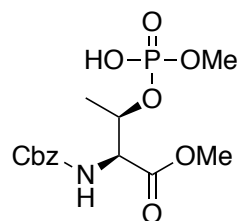
^1H NMR (CDCl_3 , 400 MHz) δ 7.33-7.19 (m, 10H), 4.82 (d, $J = 11.4$ Hz, 1H), 4.76 (d, $J = 12.4$ Hz, 1H), 4.70 (d, $J = 11.4$ Hz, 1H), 4.56 (d, $J = 12.4$ Hz, 1H), 4.46 (d, $J = 3.2$ Hz, 1H), 4.23-4.10 (m, 3H), 3.97-3.90 (m, 2H), 3.41-3.39 (m, 1H), 3.34 (s, 3H); ^{13}C NMR (CDCl_3 , 100 MHz) δ 138.2, 137.8, 128.5, 128.3, 128.1, 128.1, 128.0, 127.7, 98.9, 80.5 (d, $J = 5.7$ Hz), 78.6 (d, $J = 8.6$ Hz), 78.4, 75.5, 74.0, 68.4 ($J = 6.8$ Hz), 62.3 (d, $J = 5.7$ Hz), 55.7; ^{31}P NMR (CDCl_3 , 159 MHz) δ -3.8; ESI-MS m/z 435.1 $[\text{M}-\text{H}]^-$; HRMS calcd for $[\text{C}_{21}\text{H}_{24}\text{O}_8\text{P}]^-$ requires m/z 435.1214; found 435.1214.



Methyl N-((benzyloxy)carbonyl)-O-(hydroxy(methoxy)phosphoryl)-L-serinate (28):

General procedure B (15.4 mg, 0.0445 mmol, 45% yield)

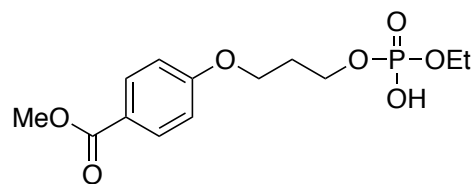
^1H NMR (CD_3OD , 400 MHz) δ 7.37-7.28 (m, 5H), 5.12 (s, 2H), 4.54-4.52 (m, 1H), 4.30-4.24 (m, 2H), 3.75 (s, 3H), 3.65 (d, $J = 11.4$ Hz, 3H); ^{13}C NMR (CD_3OD , 100 MHz) δ 171.2, 158.4, 138.1, 129.5, 129.1, 128.9, 67.8, 67.2 (d, $J = 5.7$ Hz), 55.9 (d, $J = 8.6$ Hz), 54.4 (d, $J = 5.7$ Hz), 53.1; ^{31}P NMR (CD_3OD , 159 MHz) δ 0.2; ESI-MS m/z 346.1 $[\text{M}-\text{H}]^-$; HRMS calcd for $[\text{C}_{13}\text{H}_{17}\text{NO}_8\text{P}]^-$ requires m/z 346.0697; found 346.0697.



Methyl N-((benzyloxy)carbonyl)-O-(hydroxy(methoxy)phosphoryl)-L-threoninate (29):

General procedure B (15.0 mg, 0.0414 mmol, 41% yield)

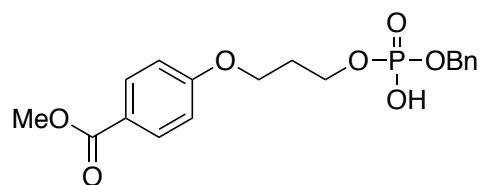
^1H NMR (CDCl_3 , 400 MHz) δ 7.34-7.29 (m, 5H), 5.69 (d, $J = 9.2$ Hz, 1H), 5.14-5.08 (m, 2H), 4.96-4.93 (m, 1H), 4.47-4.44 (m, 1H), 3.74 (s, 3H), 3.67 (d, $J = 11.4$ Hz, 3H), 1.38 (d, $J = 6.0$ Hz, 3H); ^{13}C NMR (CDCl_3 , 100 MHz) δ 170.0, 156.6, 136.0, 128.5, 128.2, 128.0, 75.3 (d, $J = 4.8$ Hz), 67.3, 58.3 (d, $J = 6.7$ Hz), 54.3 (d, $J = 6.5$ Hz), 52.8, 18.4 (d, $J = 1.9$ Hz); ^{31}P NMR (CDCl_3 , 159 MHz) δ 0.2; ESI-MS m/z 360.1 $[\text{M}-\text{H}]^-$; HRMS calcd for $[\text{C}_{14}\text{H}_{19}\text{NO}_8\text{P}]^-$ requires m/z 360.0854; found 360.0854.



Methyl 4-(3-((hydroxy(ethoxy)phosphoryl)oxy)propoxy)benzoate (30):

General procedure C (15.5 mg, 0.0487 mmol, 49% yield)

^1H NMR (CDCl_3 , 400 MHz) δ 7.94 (d, J = 8.2 Hz, 2H), 6.87 (d, J = 8.2 Hz, 2H), 4.18 (dt, J = 6.4 Hz, 6.4 Hz, 2H), 4.08 (t, J = 6.0 Hz, 2H), 4.04-4.00 (m, 2H), 3.85 (s, 3H), 2.15-2.09 (m, 2H), 1.26 (t, J = 7.3 Hz, 3H); ^{13}C NMR (CDCl_3 , 100 MHz) δ 166.8, 162.4, 131.5, 122.6, 114.0, 63.9 (d, J = 5.7 Hz), 63.8, 63.7 (d, J = 5.7 Hz), 51.9, 29.9 (d, J = 7.6 Hz), 16.0 (d, J = 6.7 Hz); ^{31}P NMR (CDCl_3 , 159 MHz) δ 0.5; ESI-MS m/z 317.2 $[\text{M}-\text{H}]^-$; HRMS calcd for $[\text{C}_{13}\text{H}_{18}\text{O}_7\text{P}]^-$ requires m/z 317.0796; found 317.0796.



Methyl 4-(3-((hydroxy(benzyloxy)phosphoryl)oxy)propoxy)benzoate (31):

General procedure D (16.2 mg, 0.0426 mmol, 43% yield)

^1H NMR (CDCl_3 , 400 MHz) δ 7.88 (d, J = 7.3 Hz, 2H), 7.24-7.17 (m, 5H), 6.77 (d, J = 7.3 Hz, 2H), 4.94 (d, J = 9.6 Hz, 2H), 4.10 (dt, J = 6.3 Hz, 6.3 Hz, 2H), 3.95 (t, J = 6.2 Hz, 2H), 3.82 (s, 3H), 2.03-1.97 (m, 2H); ^{13}C NMR (CDCl_3 , 100 MHz) δ 166.8, 162.4, 135.7 (d, J = 7.6 Hz), 131.5, 128.5, 128.4, 127.7, 122.6, 114.0, 69.0 (d, J = 5.7 Hz), 64.1 (d, J = 5.7 Hz), 63.8, 51.8, 29.8 (d, J = 7.6 Hz); ^{31}P NMR (CDCl_3 , 159 MHz) δ 0.6; ESI-MS m/z 379.1 $[\text{M}-\text{H}]^-$; HRMS calcd for $[\text{C}_{18}\text{H}_{20}\text{O}_7\text{P}]^-$ requires m/z 379.0952; found 379.0952.

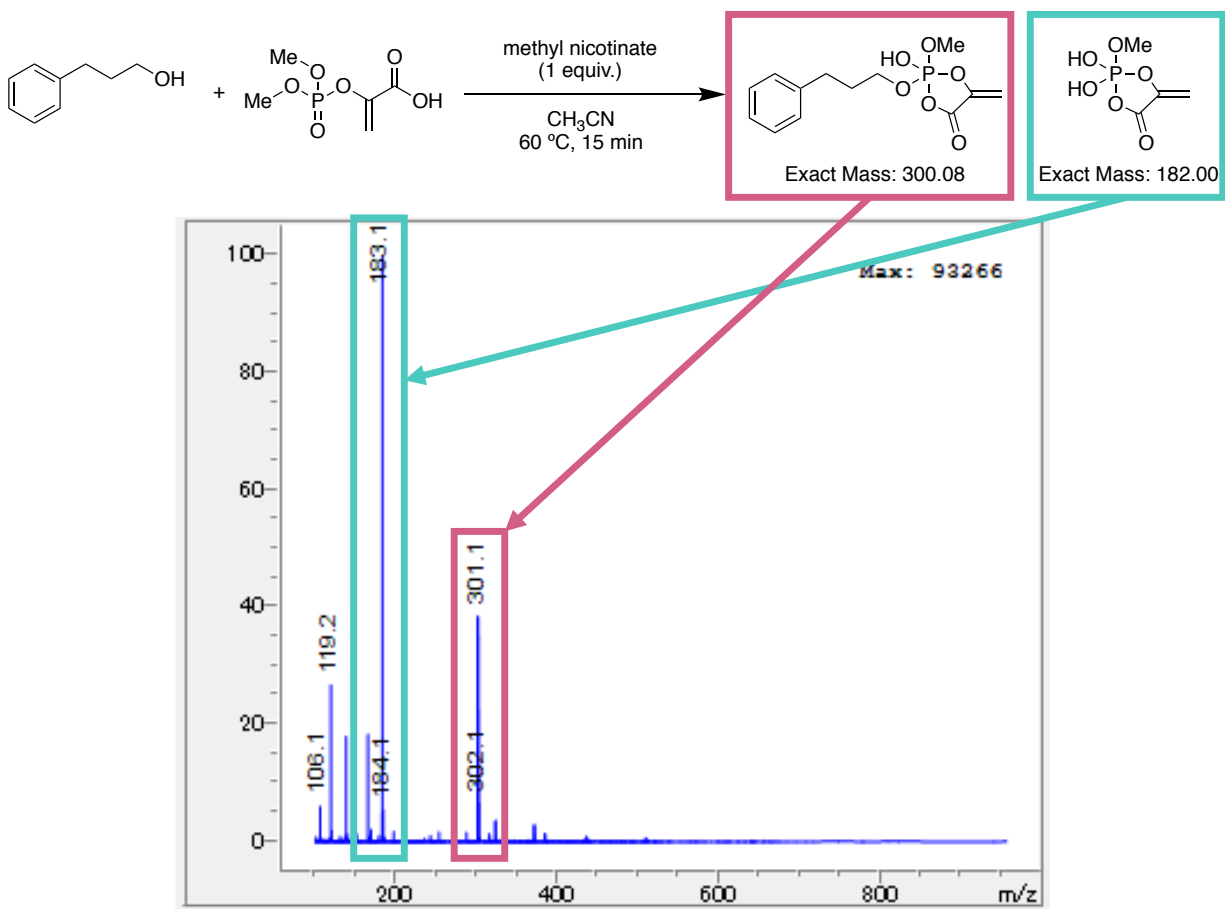
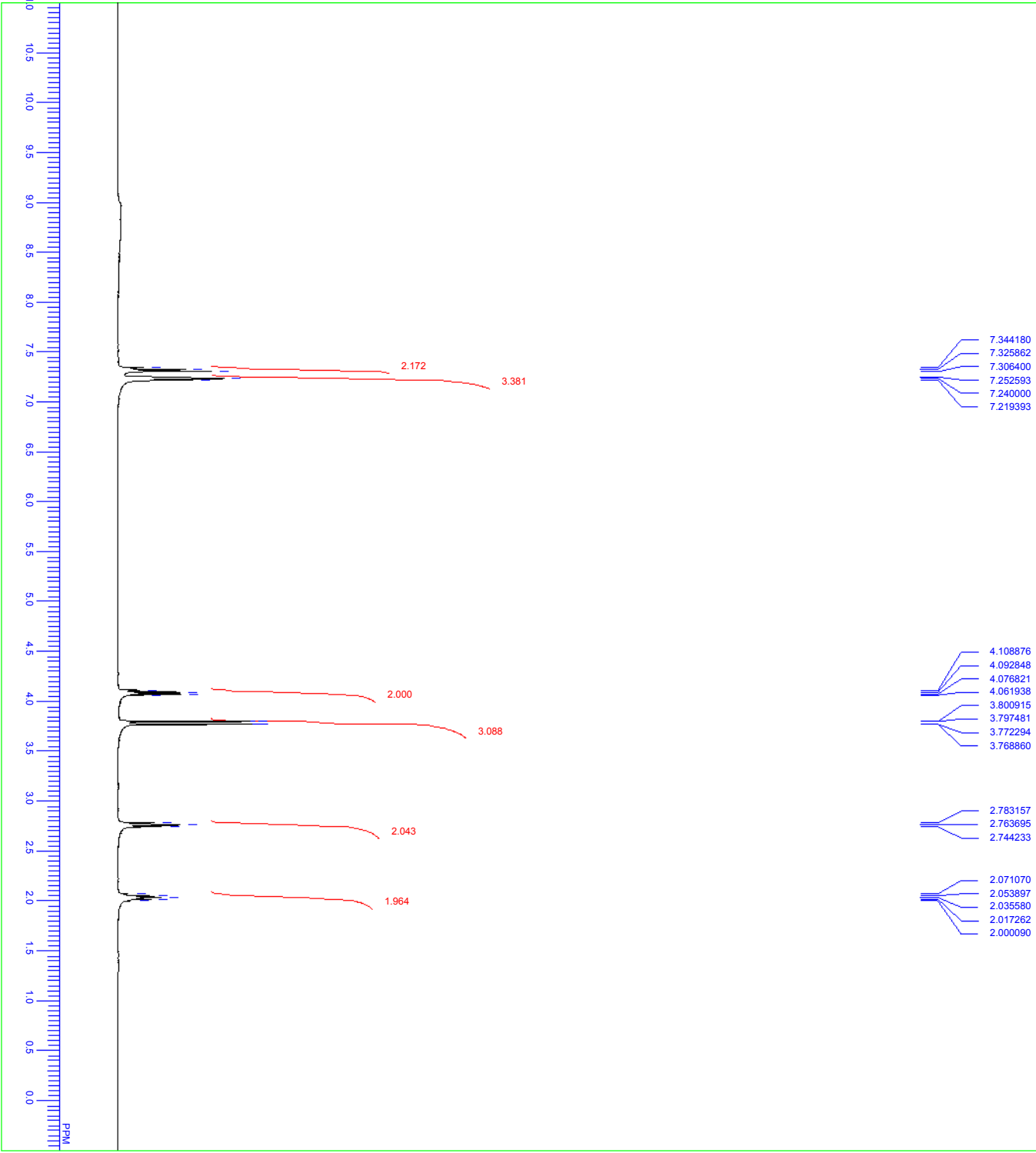


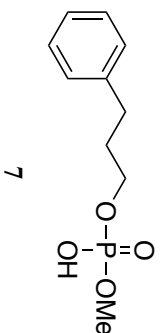
Figure S1. A MS spectrum after the reaction between 3-phenyl-1-propanol and dimethyl phosphoenolpyruvate at 60°C for 15 min. The MS spectrum showed peaks of m/z 301.1 and 183.1, which correspond to pentacoordinate phosphorus intermediates shown in the scheme.

References

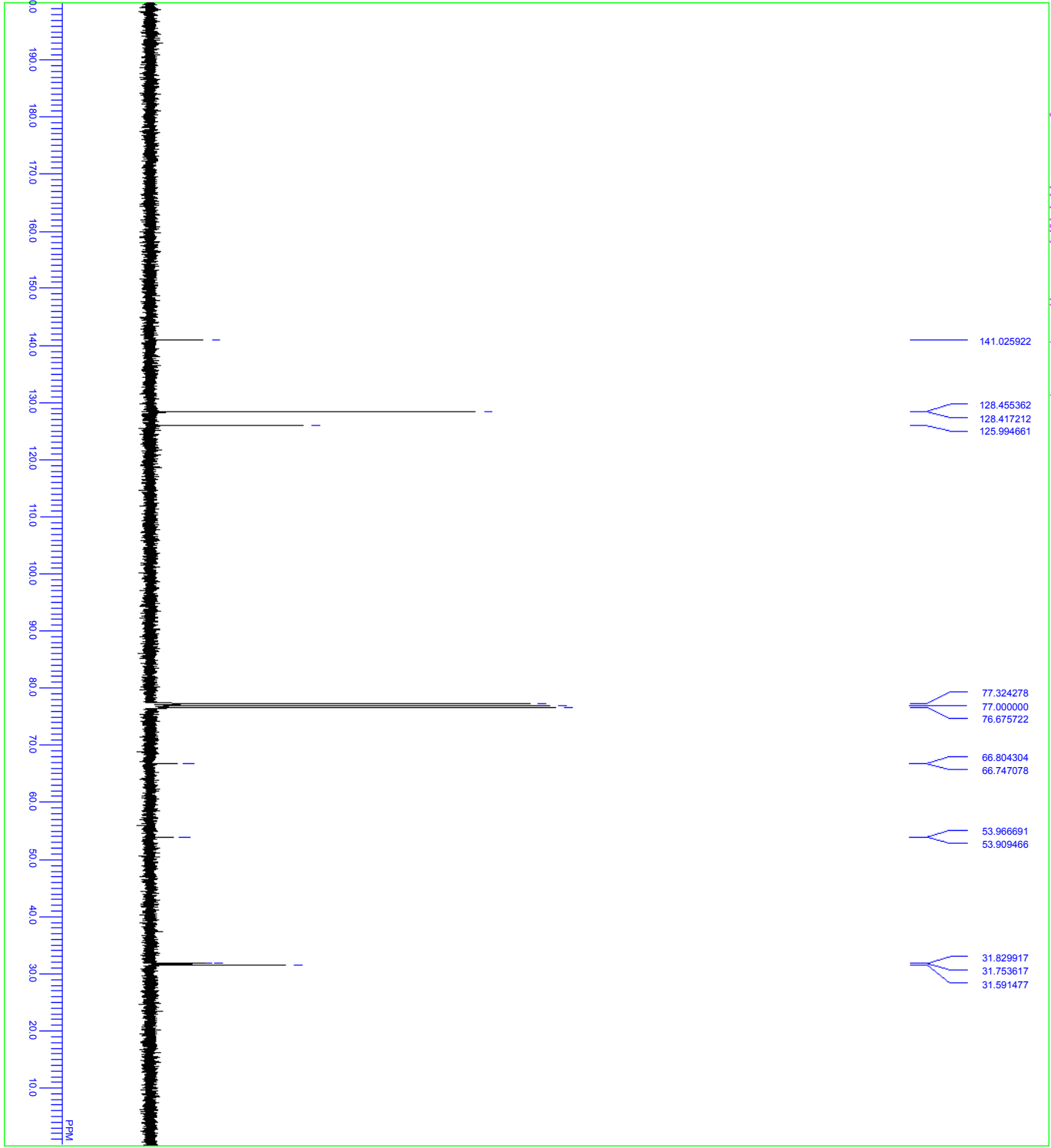
1. Bernard L. Hirschbein; Francois P. Mazenod; and George M. Whitesides; Synthesis of Phosphoenolpyruvate and Its Use in Adenosine Triphosphate Cofactor Regeneration, *J. Org. Chem.* **1982**, 47, 3765-3766.



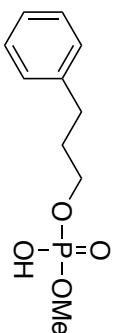
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OBFN 3.34 Hz
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FREQU 5878.90 Hz
SCANS 8
ACQTM 2.2295 sec
PD 6.0000 sec
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CTEMP 20.7 c
SLUNT CDCL3
EXREF 7.24 ppm
BF 0.12 Hz
RGAIN 38



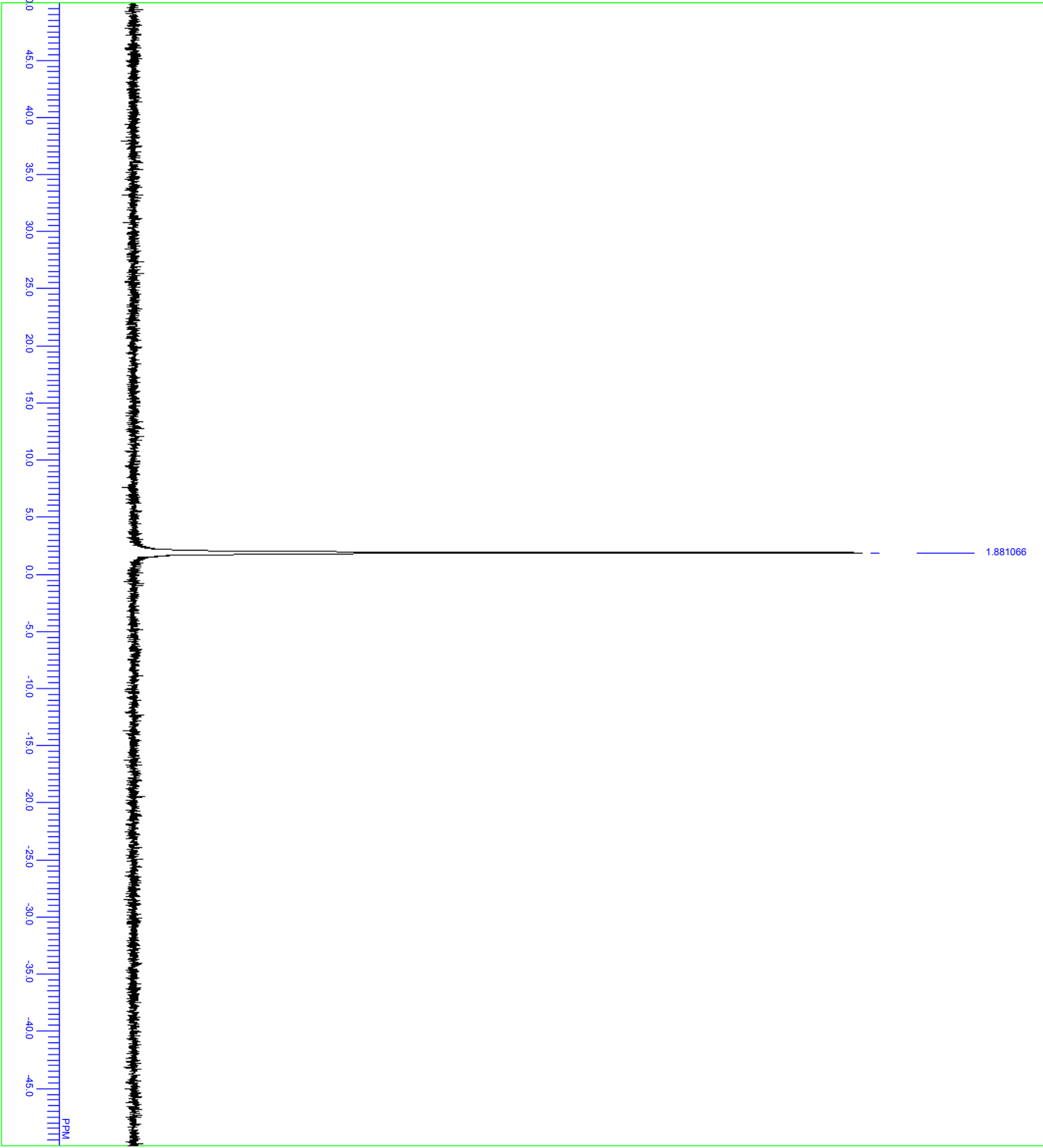
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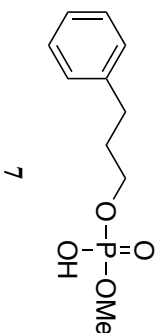
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PD	2.0000 sec
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RGAIN	60

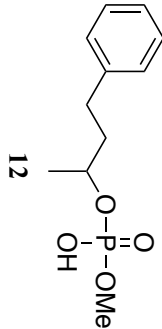
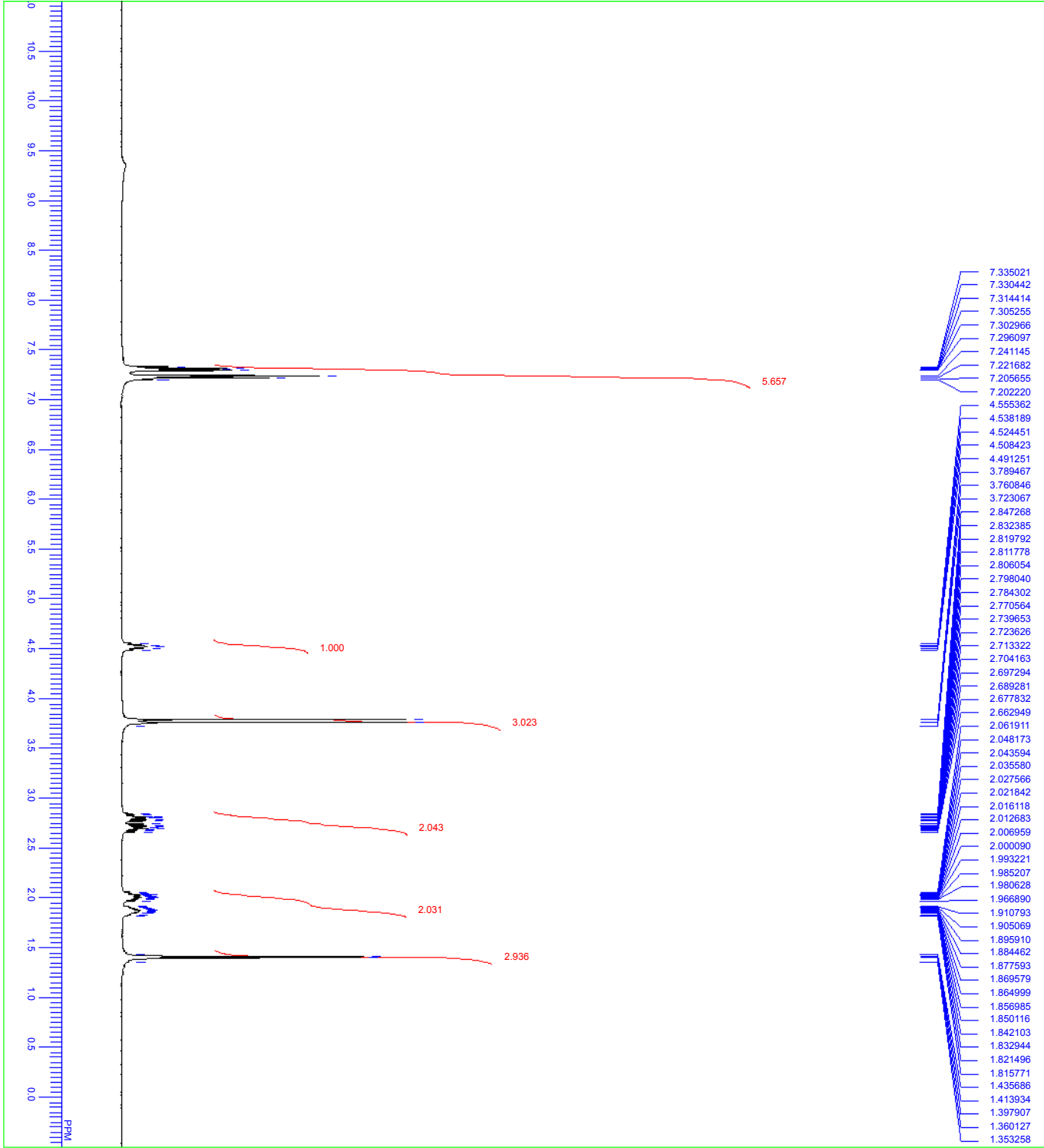


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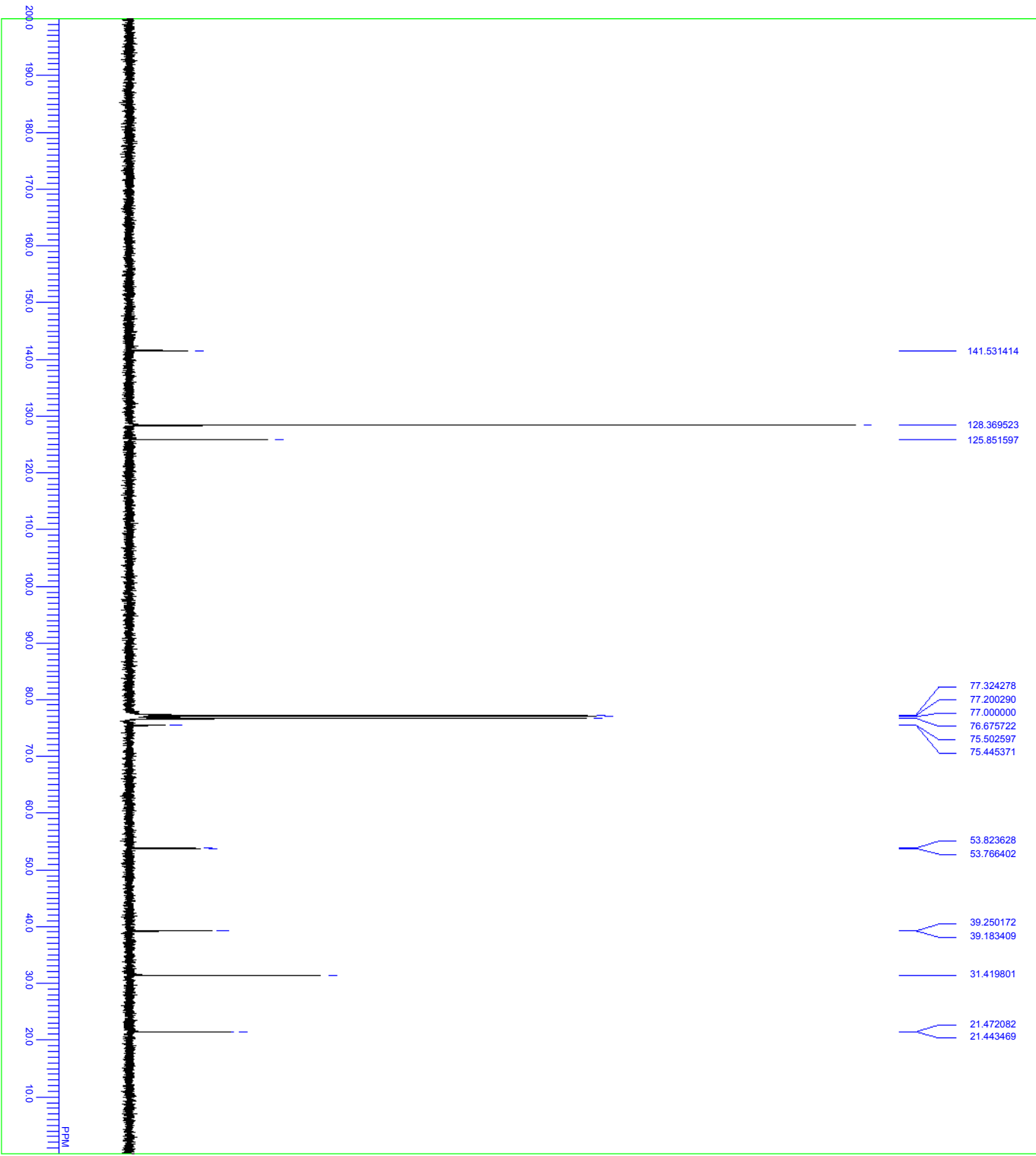


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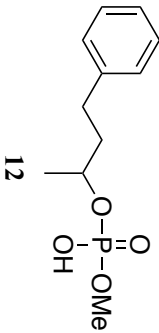


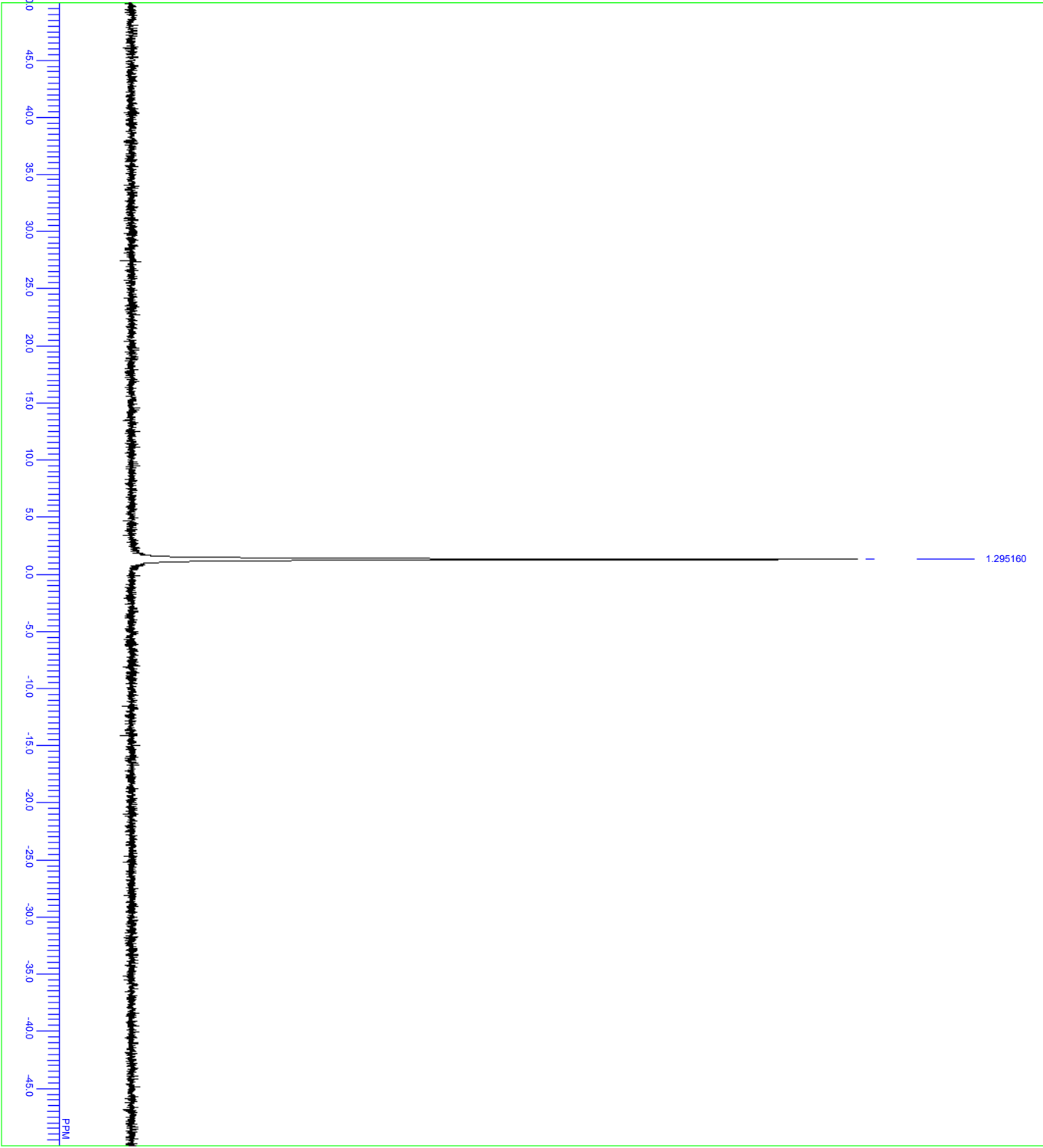


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OBFIN 3.34 Hz
POINT 131.07
FREQU 507.830 Hz
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AQCTM 2.2295 sec
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RGAIN 36

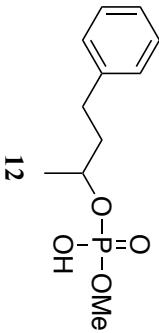


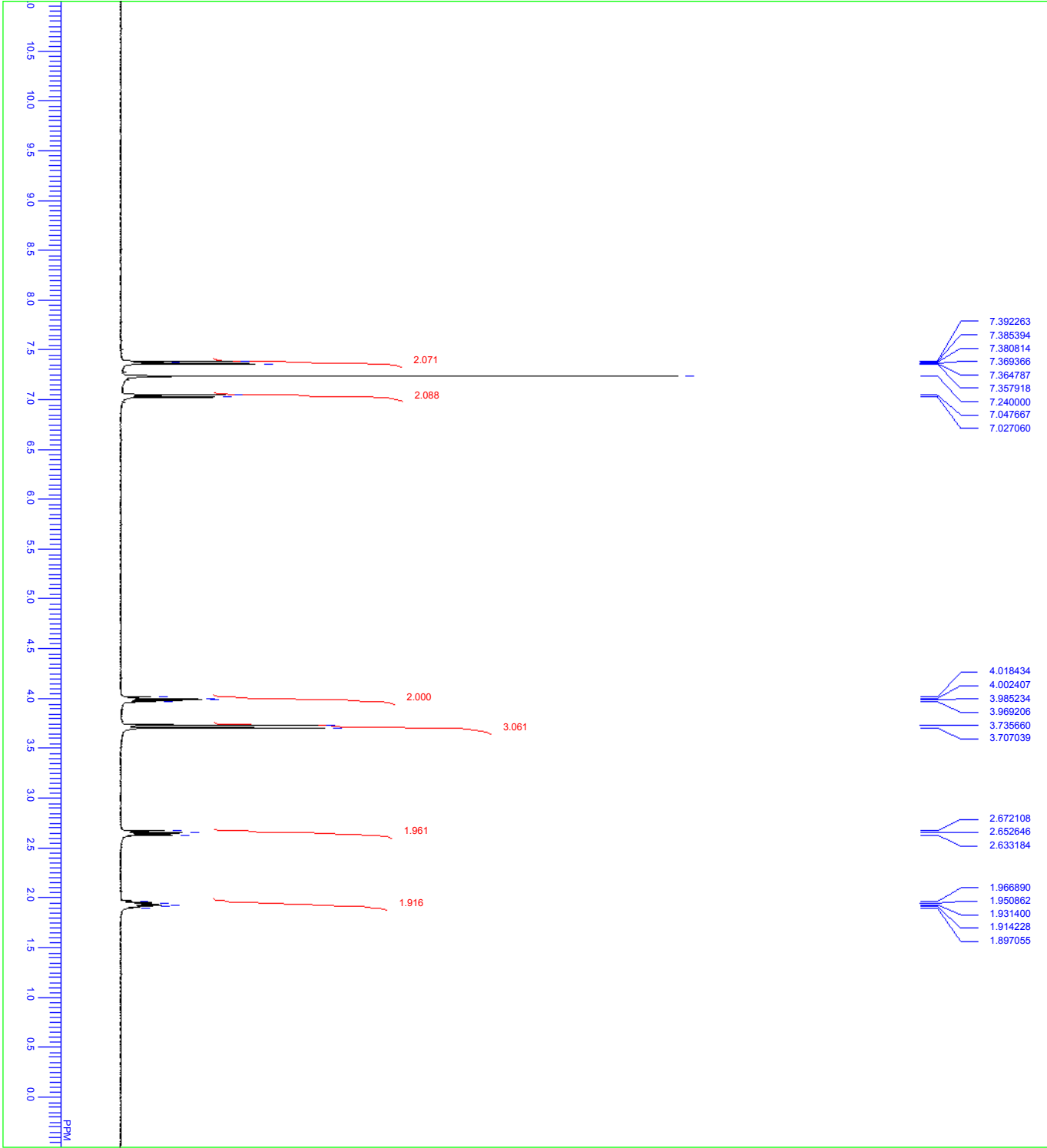
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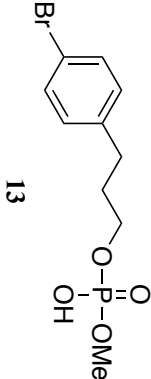


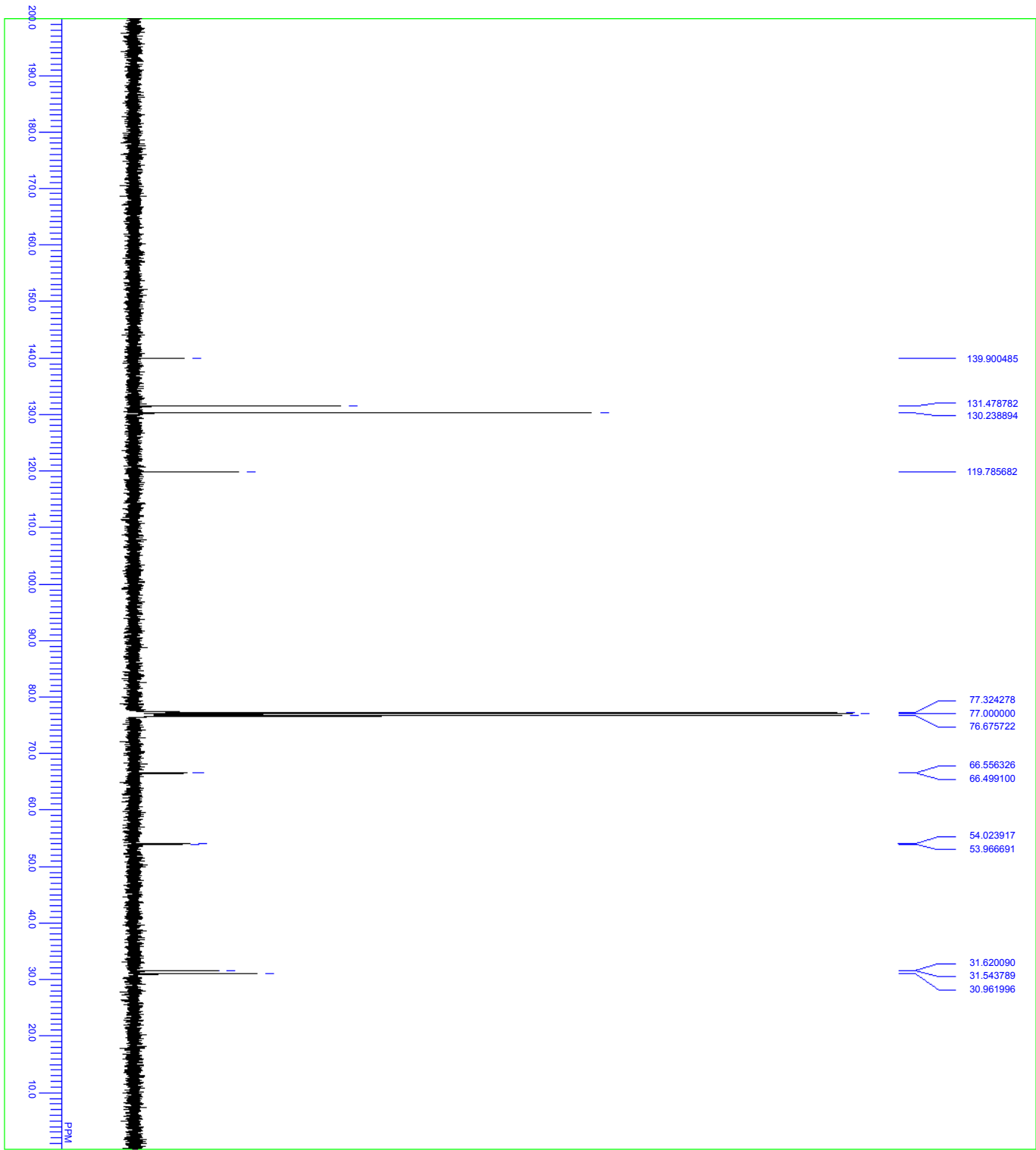
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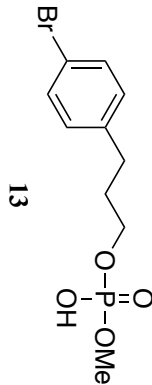


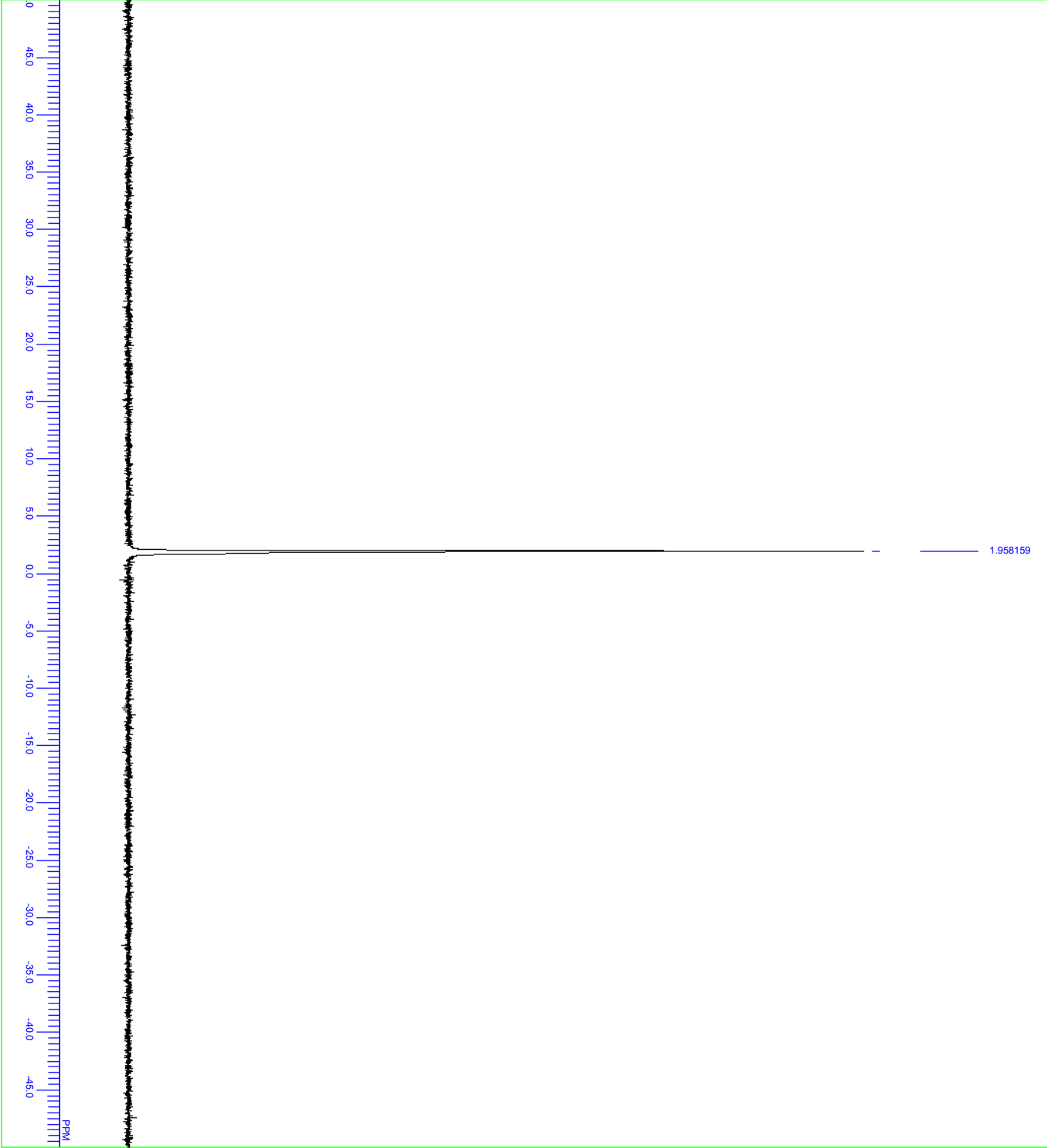
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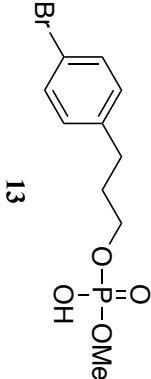


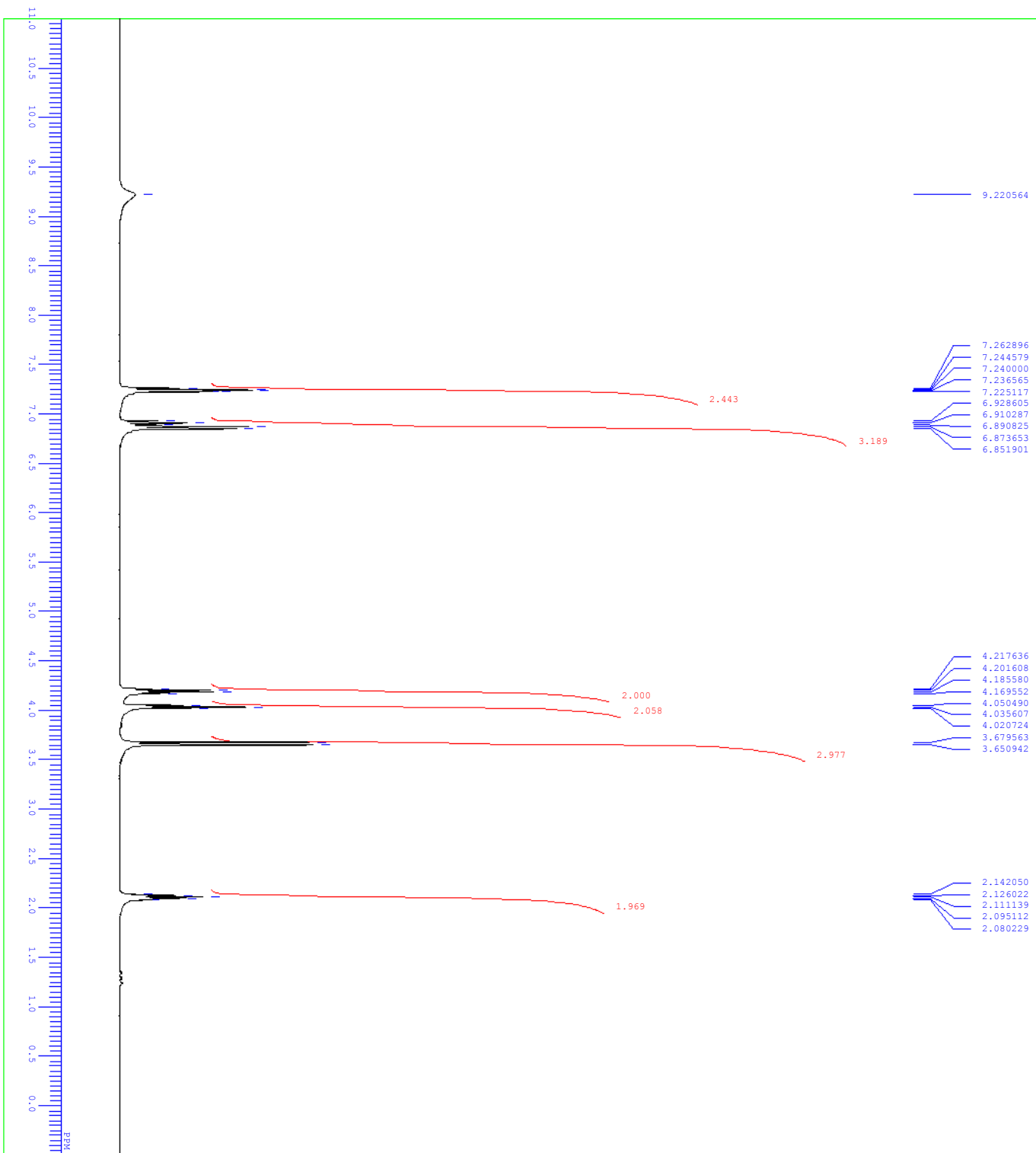
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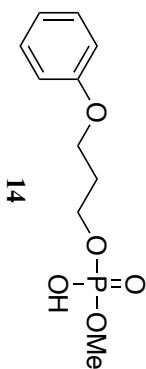


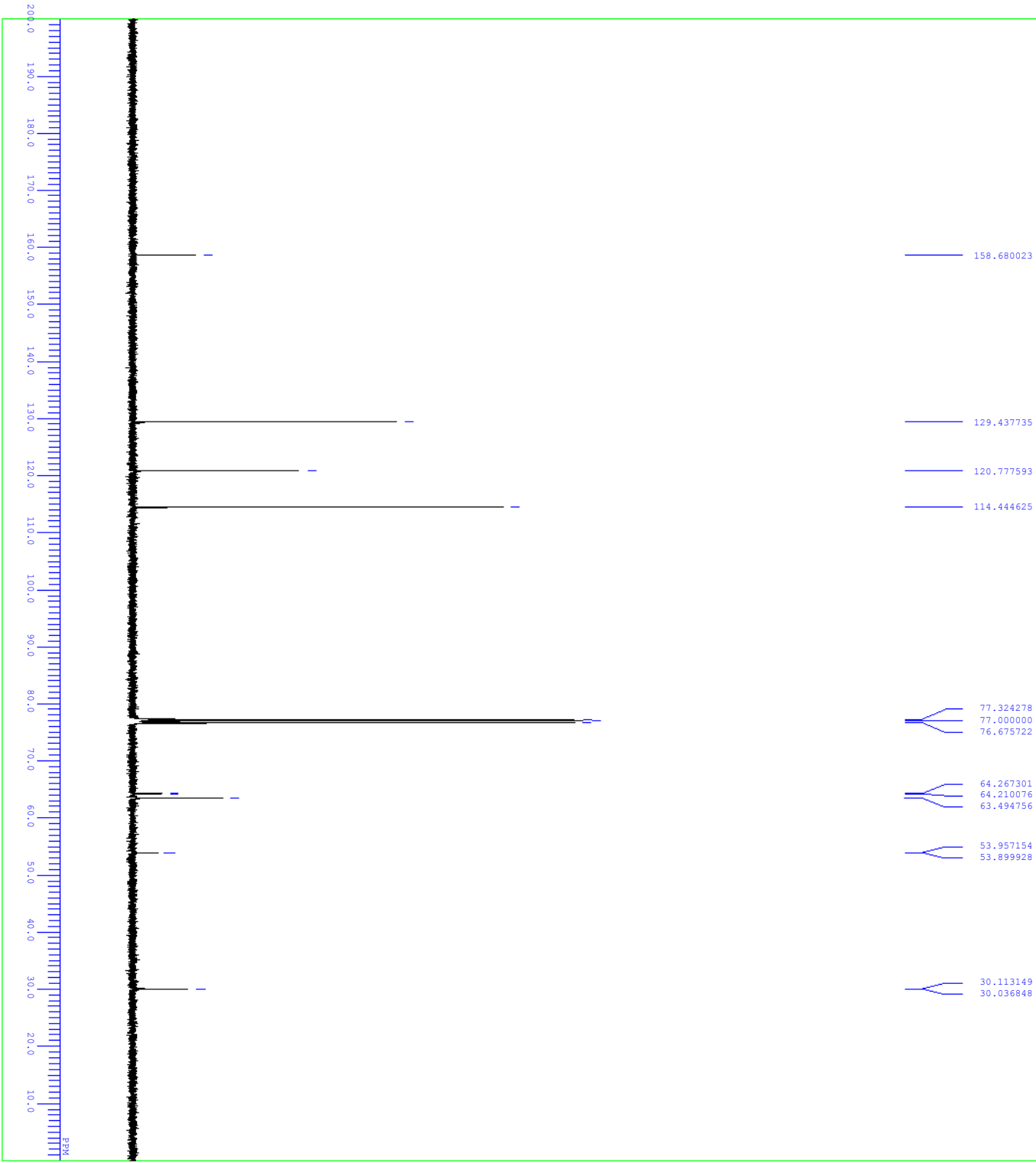
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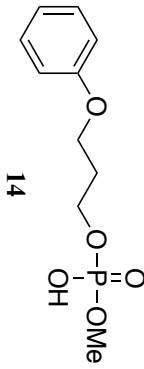


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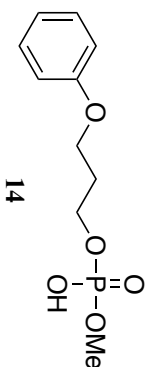


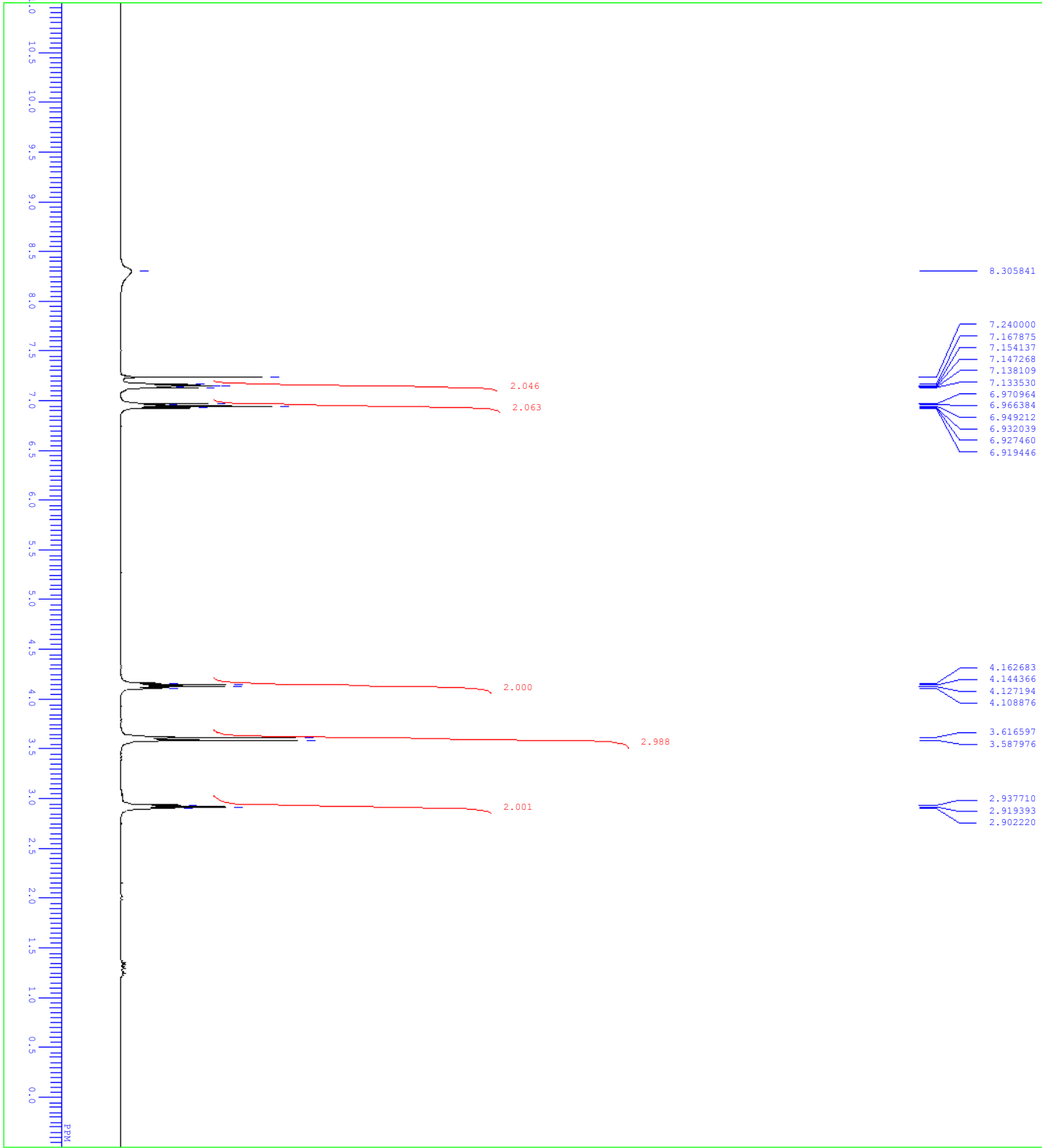


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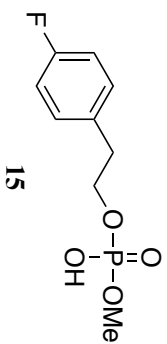


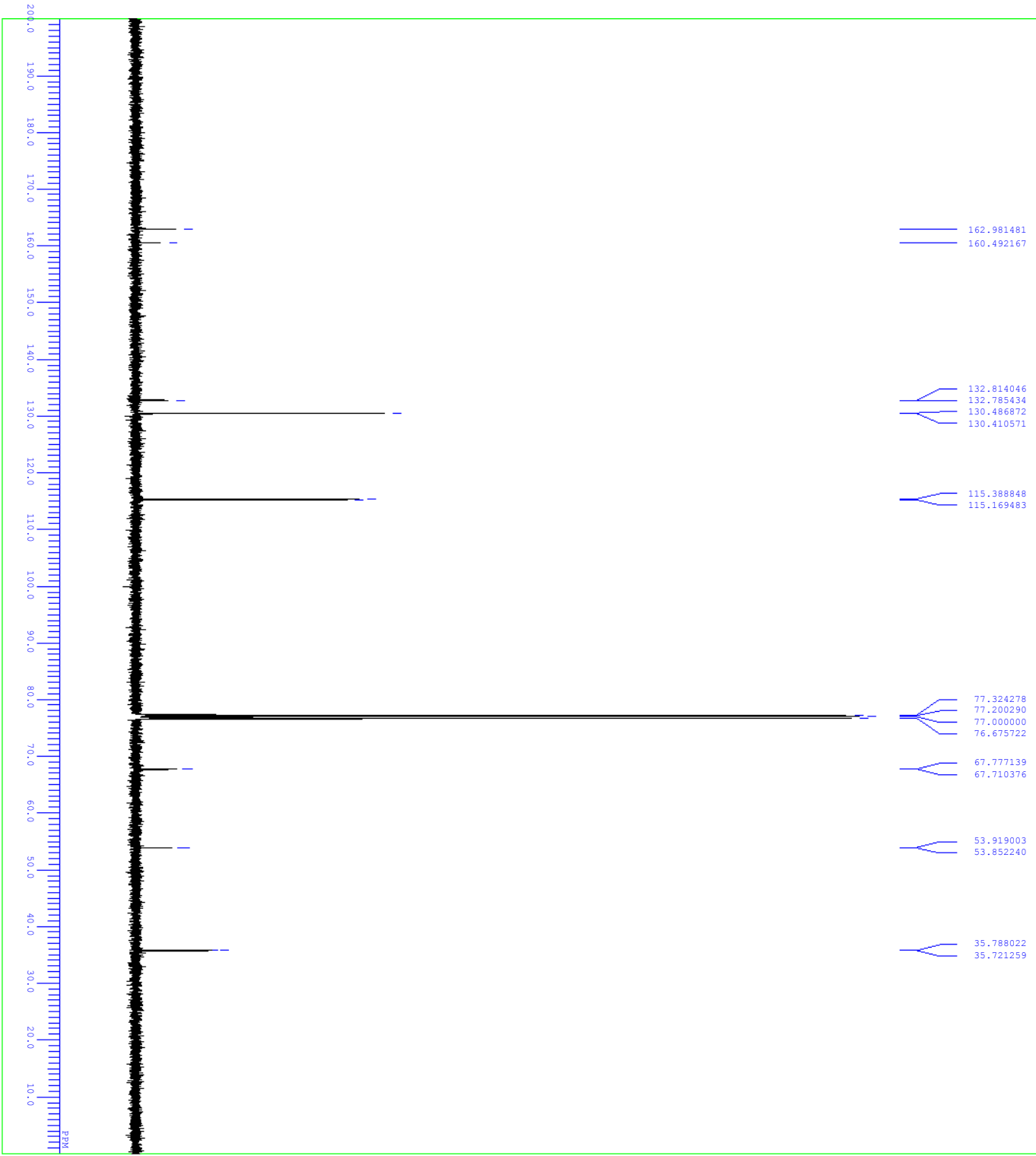
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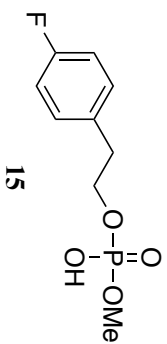


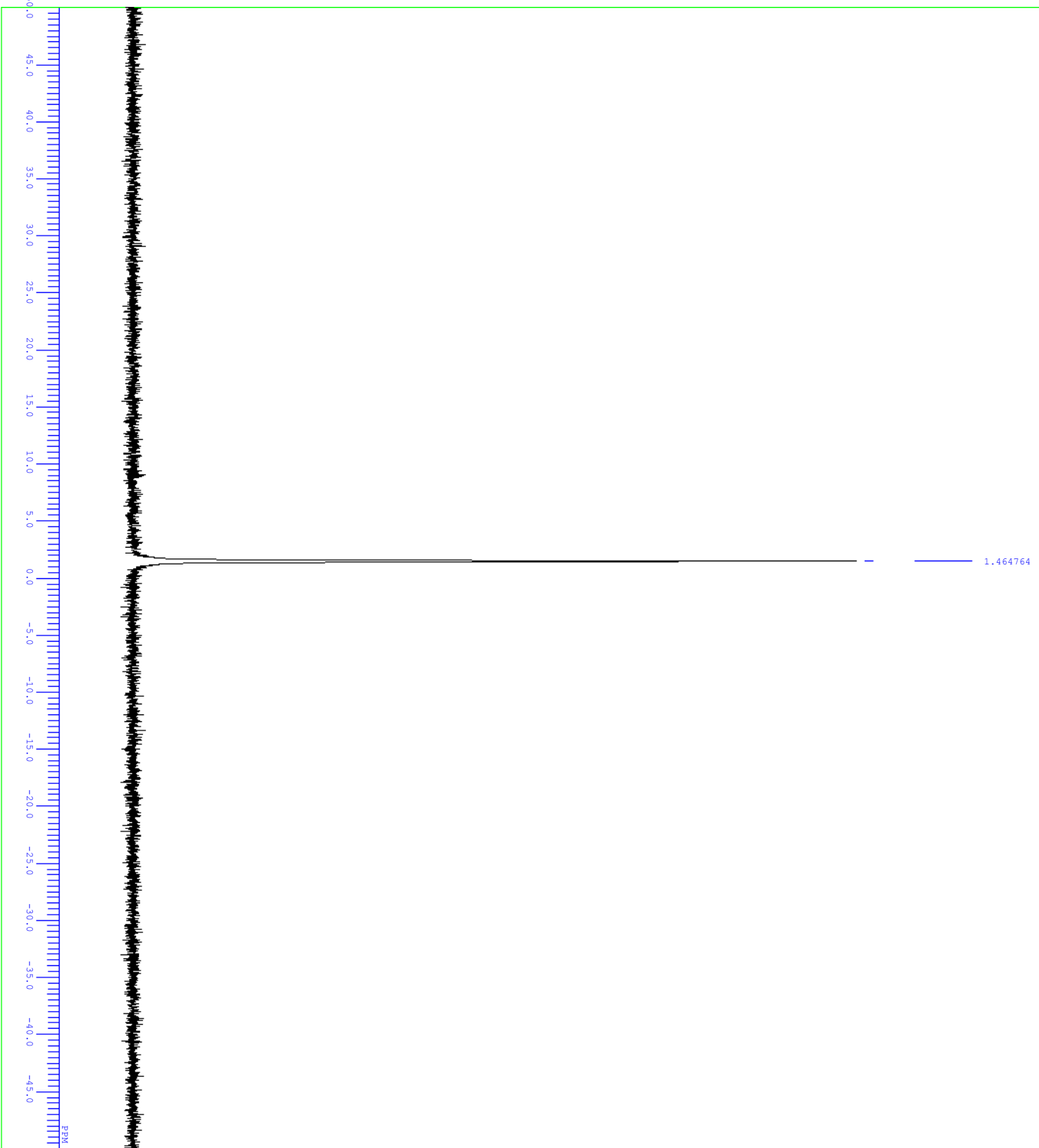
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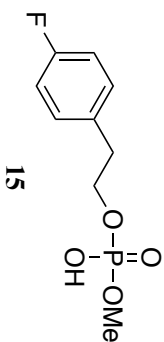


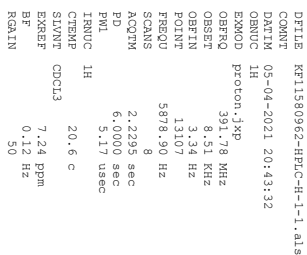
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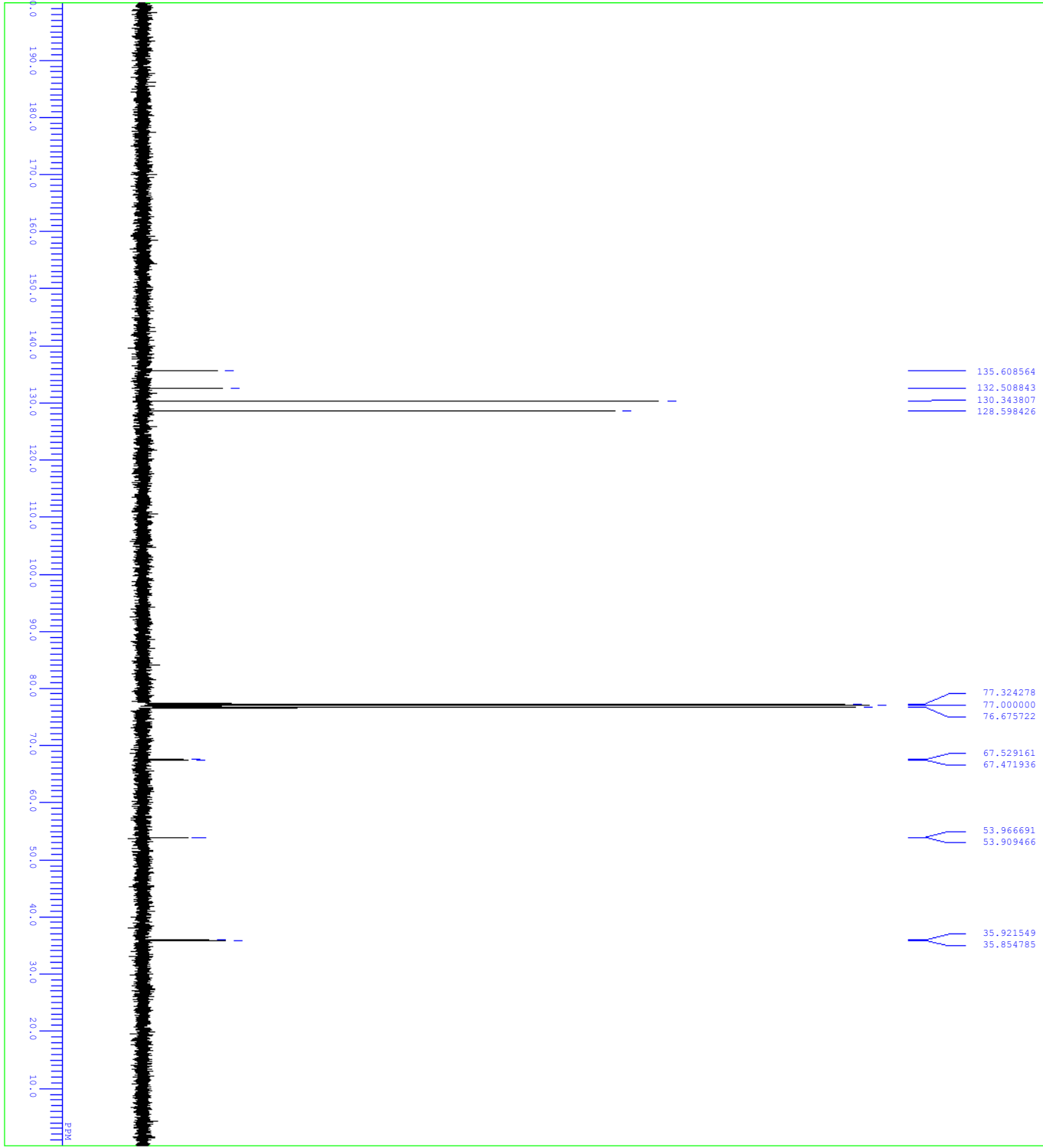




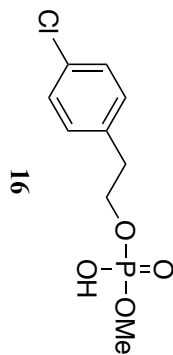
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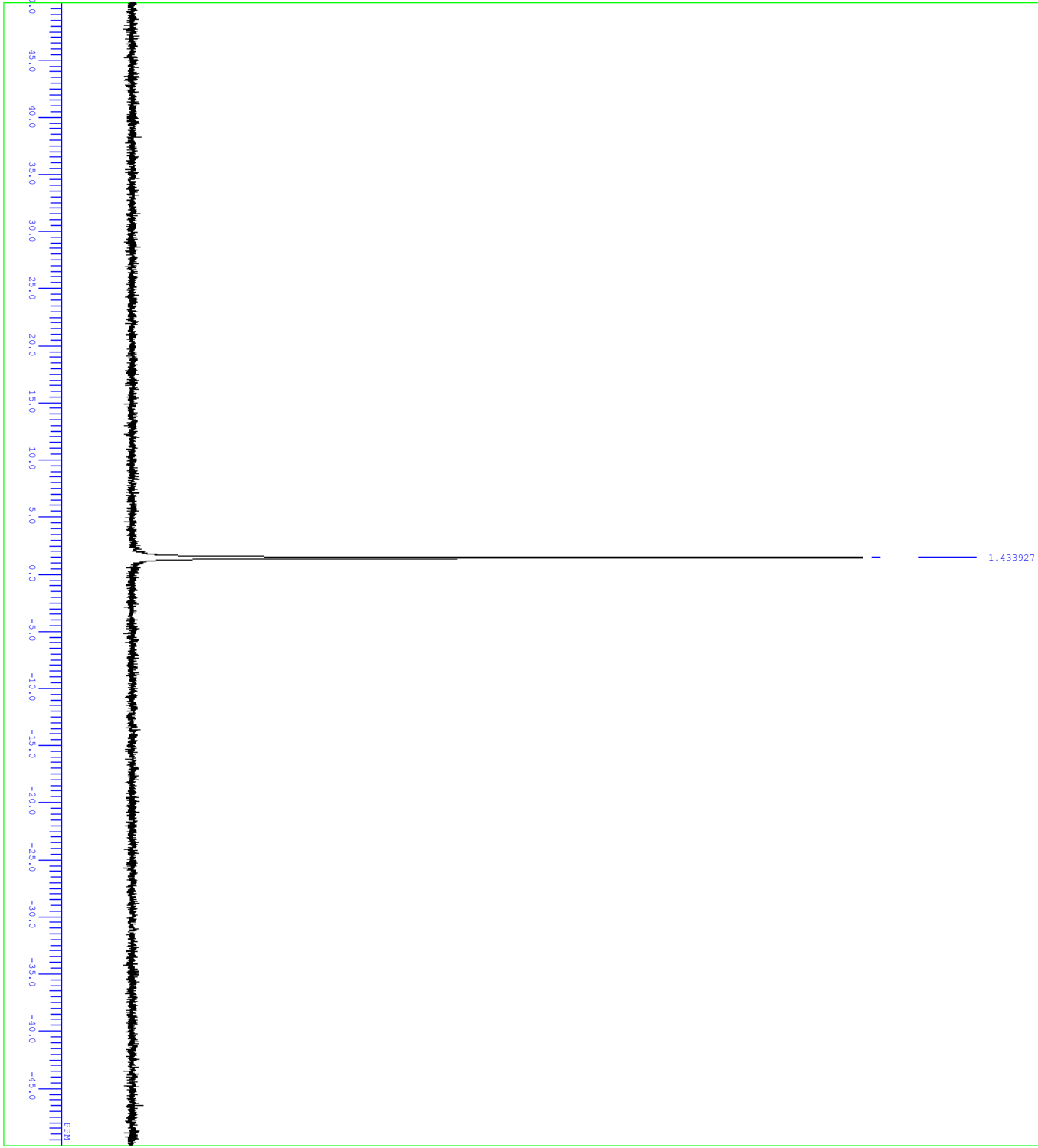




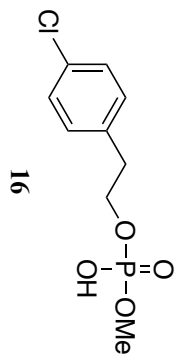


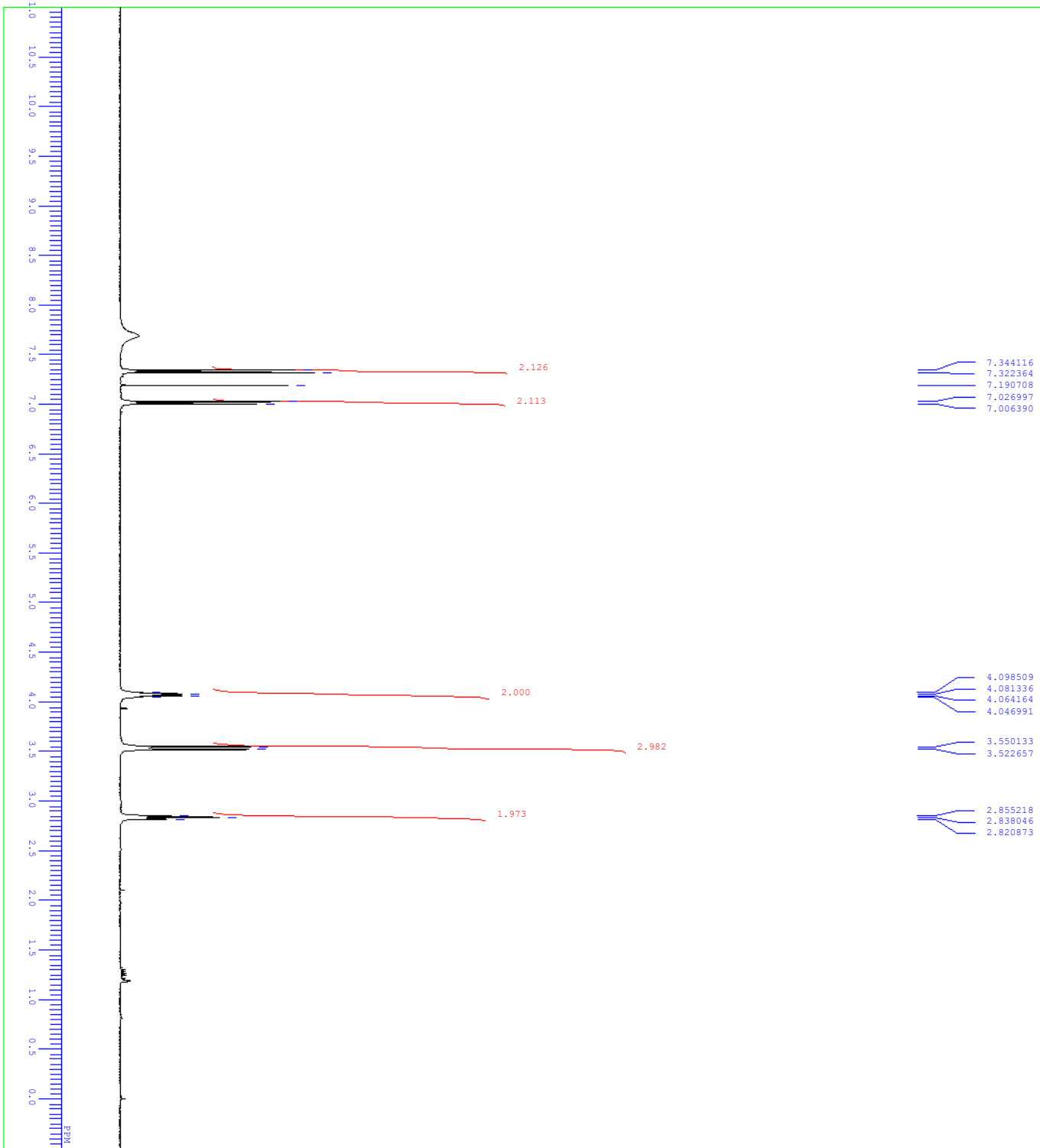
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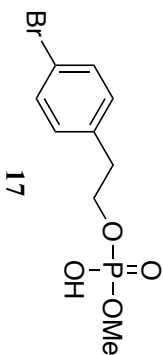


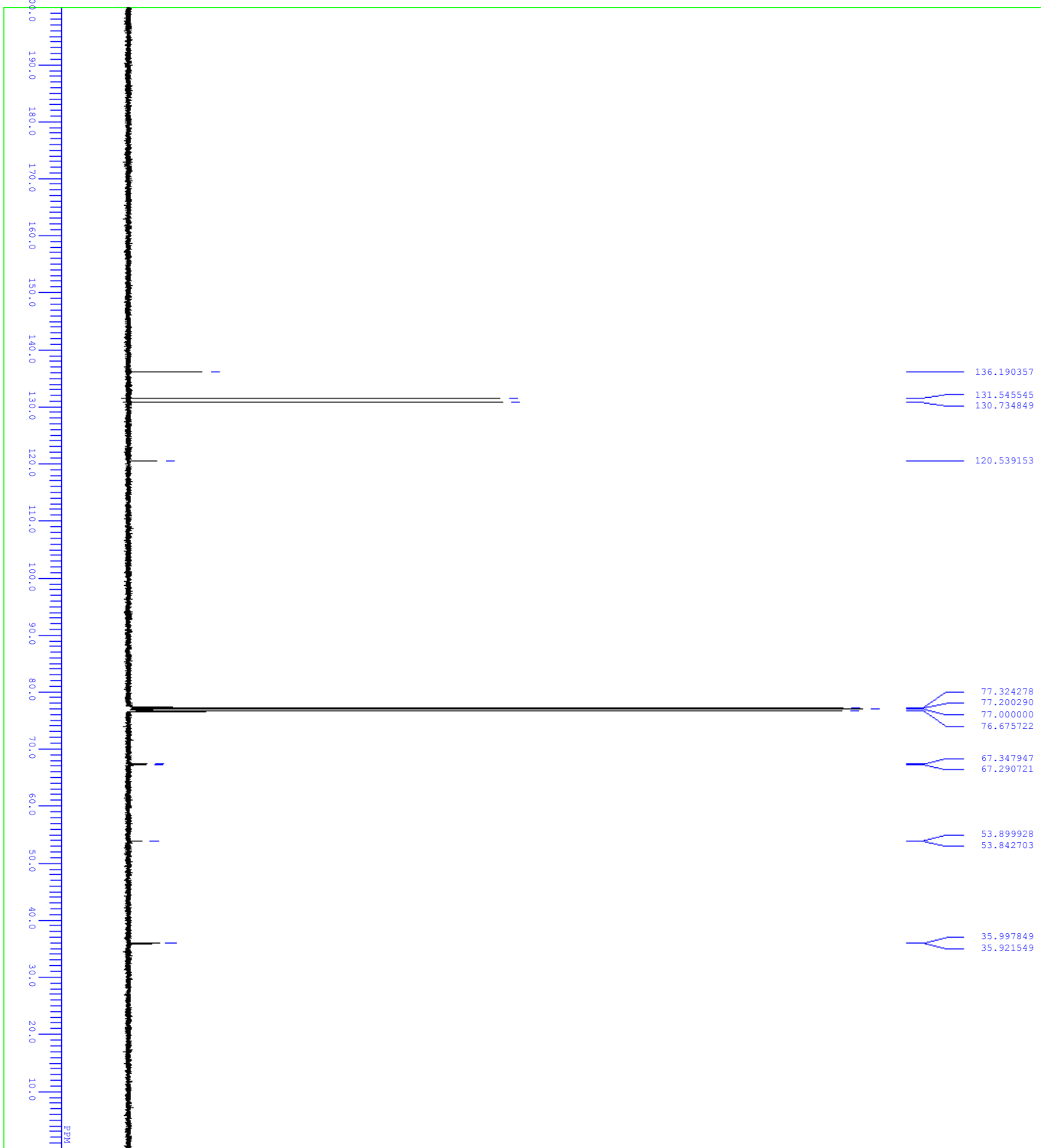
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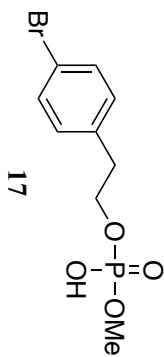


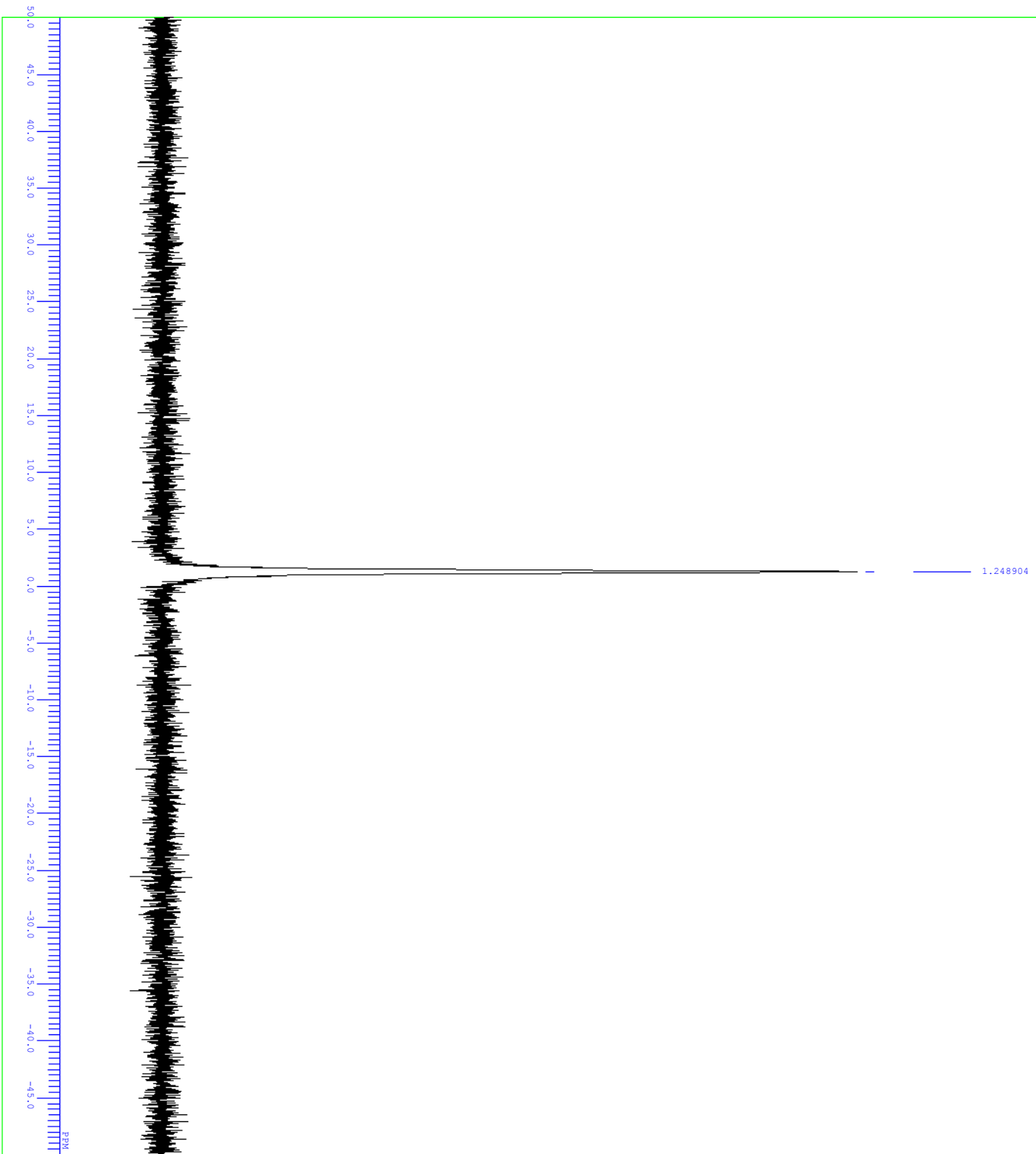
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 RGAIN 42



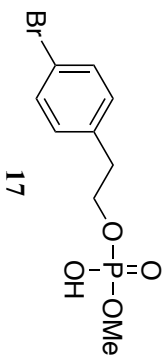


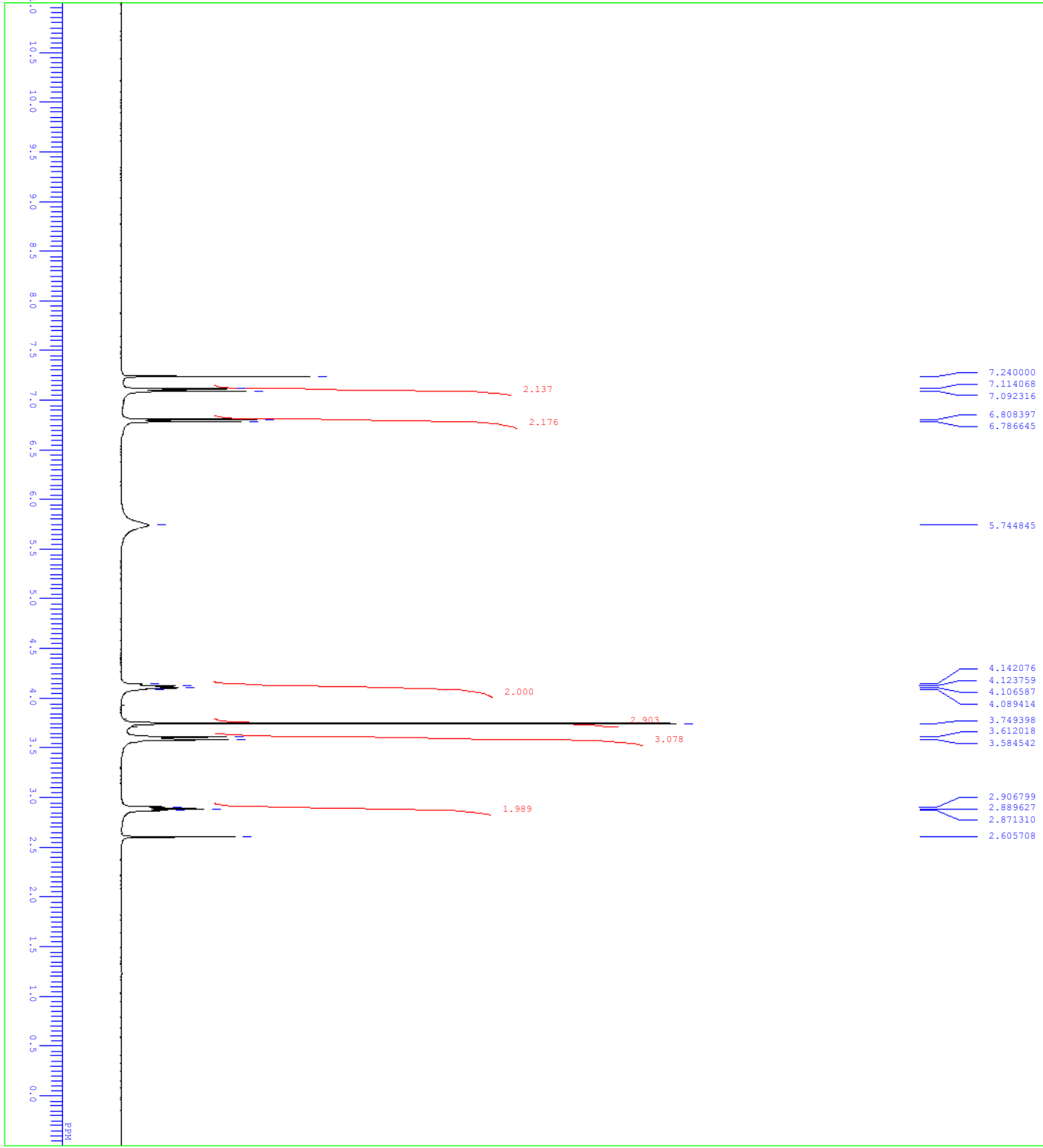
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 RGAIN 60



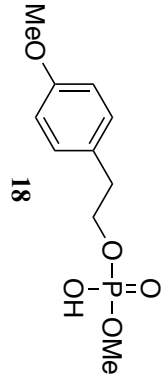


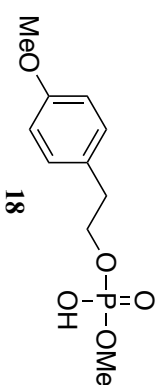
DTITLE KF11580963-HPLC-P-1-1.als
 COUNT 31P
 DATEIM 05-04-2021 21:38:16
 EXMOC 31P
 EXMOC carbon-13
 OBSRQ 158.59 MHz
 OBSRQ 175.92 MHz
 OBSRQ 9.23 MHz
 POINT 26214 Hz
 FREQU 64102.56 Hz
 SCANS 201
 ACQTM 0.4089 sec
 PD 2.0000 sec
 PUL 4.90 usec
 IRNUC 1H
 CTEMP 20.8 C
 SLVNT CDCL3
 EXREF 0.00 ppm
 BF 0.12 Hz
 RGAIN 56

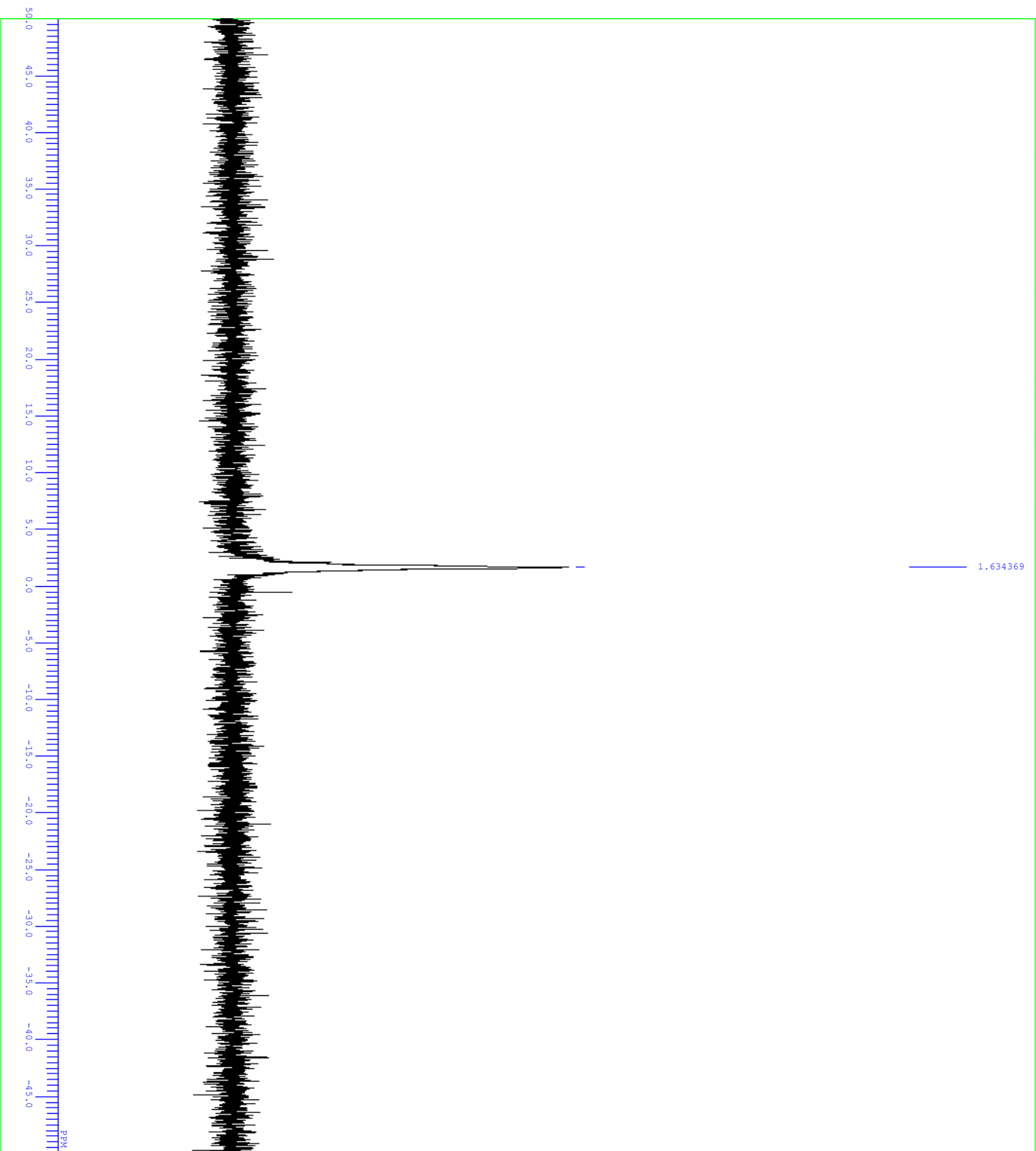




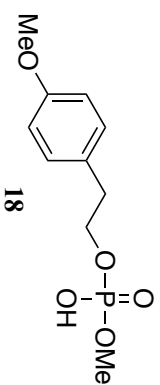
DFILE KF11570953-HPLC-H-3-1-1.a1s
CONTC 13-04-2021 17:18:20
NAME 18
EXPNO 391.79 MHz
PROCNO 1
F2 8.43 kHz
F3 8.34 kHz
POINT 13107
FREQ 5878.90 Hz
SCANS 24
AQ 2.2295 sec
PD 6.0000 sec
PUL 5.17 usec
IRNUC 1H
CTEMP 20.3 C
SOLVT CDCL3
EXREF 7.24 ppm
BF 0.12 Hz
RGAIN 44

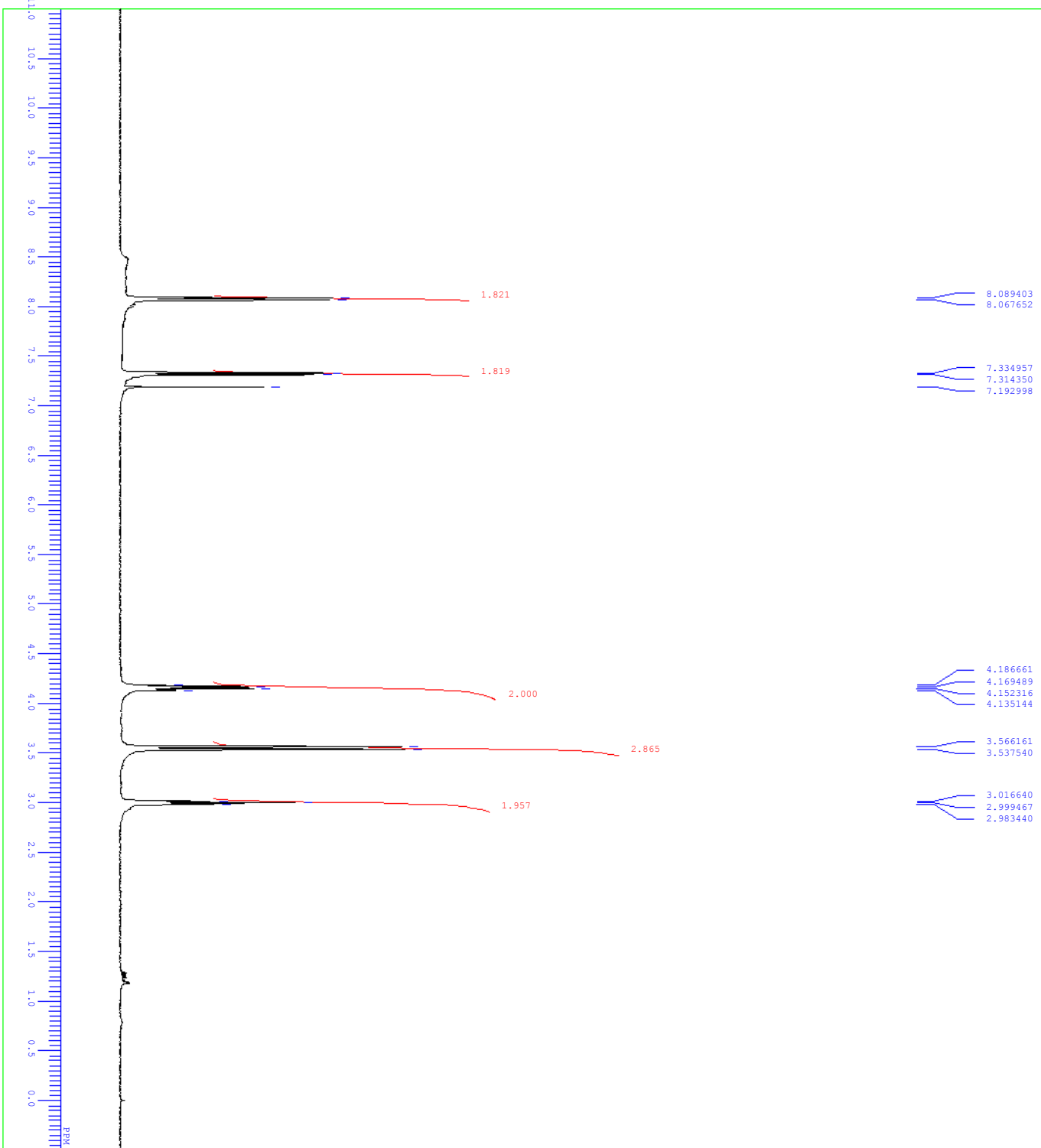




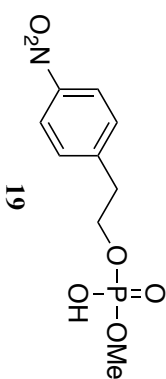


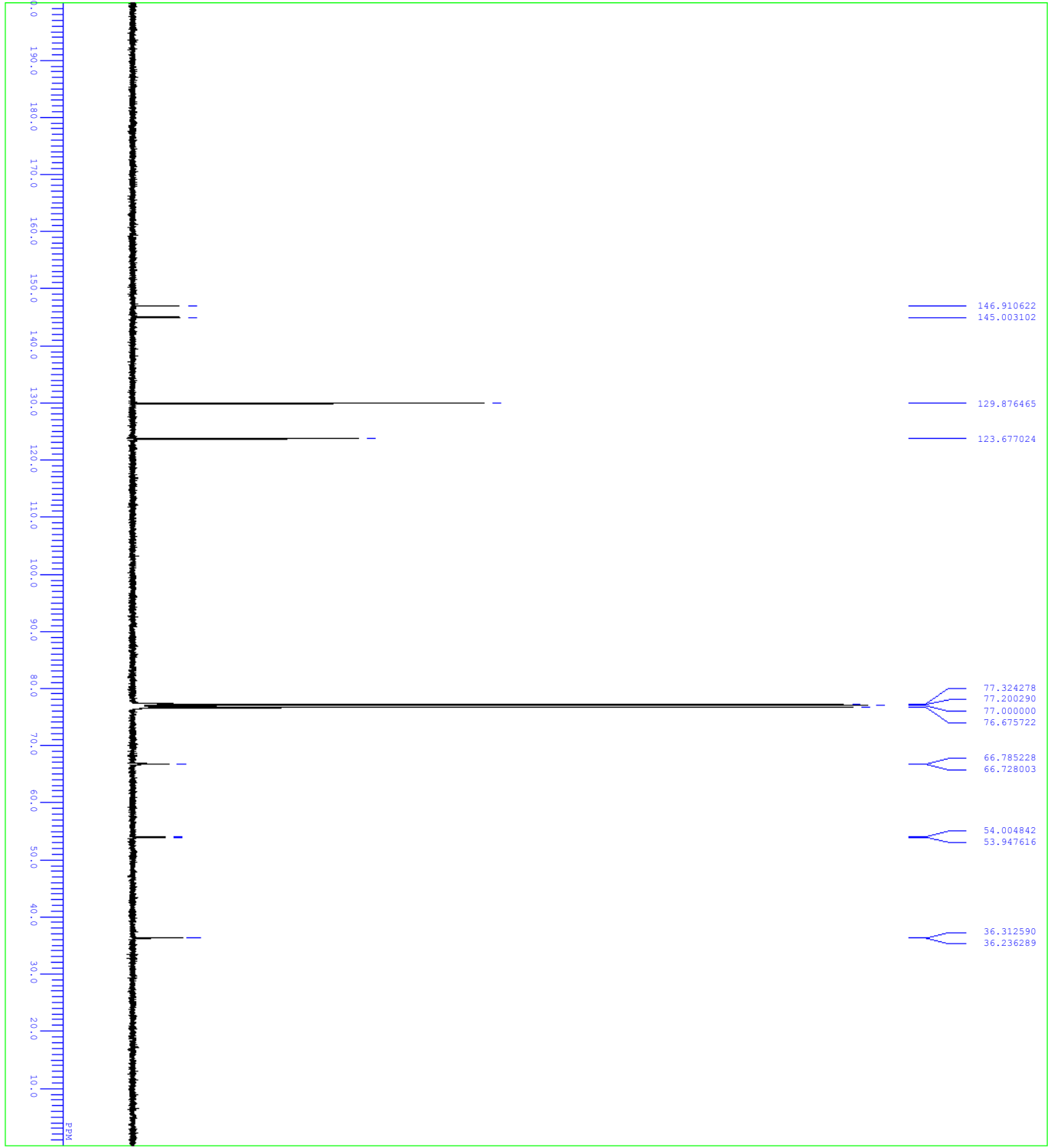
DF1LE KF11570953-HPLC-P-3-1-1.a1a
 COMNT
 DATIM 13-04-2021 17:08:00
 OBNUC 31P
 EXMOD carbon-13xp
 OBFREQ 158.59 MHz
 OBSSET 7.99 KHz
 OBFIN 9.23 Hz
 POINT 26214
 FREQU 64102.56 Hz
 SCANS 163
 ACQTM 0.4089 sec
 PU 2.0000 sec
 RL 4.90 usec
 TRNUC 1H
 CTMP 21.0 C
 SIVNT CHCL3
 EXREF 0.00 ppm
 BF 0.12 Hz
 RGAIN 56



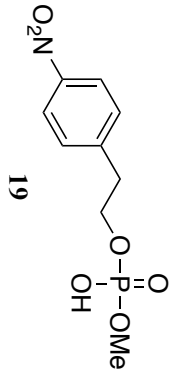


DFILE KF11560945-HPLC-H-1-1.a1a
 COUNT
 DATIM 03-04-2021 13:49:12
 EXMNUC 1H
 EXMNUC proton-1xp
 OBSFREQ 391.778 MHz
 OBSSET 8.51 KHz
 OBSFREQ 3.34 Hz
 POINT 13107
 FREQ 5878.90 Hz
 SCANS 8
 ACQTM 2.2295 sec
 DEL 6.000 sec
 PU 5.17 usec
 TBNUC 1H
 CTMP 20.6 C
 SIVNT CHCL3
 EXREF 0.00 ppm
 BF 0.12 Hz
 RGAIN 42





DPFILE KF11560945-HPLC-C-1-1.a1a
COMPT 03-04-2021 14:00:16
DATIM 03-04-2021 14:00:16
OBNUC 13C
EXMOD carbon-13p
OBSFQ 98.52 MHz
OBSST 4.64 KHz
OBSFT 8.74 Hz
POINT 26214
FREQU 24630.54 Hz
SCANS 3370
ACQTM 1.0643 sec
PUL 2.0000 sec
PR 3.12 usec
IRNUC 1H
CTEMP 20.9 C
SIVNT CHCL3
EXREF 77.00 ppm
BF 0.12 Hz
RGAIN 60



DATE	TIME	COMMENT
03	04	2001 12-EG-22

OBNUC	31P
-------	-----

EXMOD carbon.jxp

OBDRNG	400.00 MHz
OBSET	7.99 KHz

OBFIN 9.23 Hz

POINT 20214
FRECH 64102 56 H7

SCANS 129

ACQIM	0.4089 sec
PD	0.0000

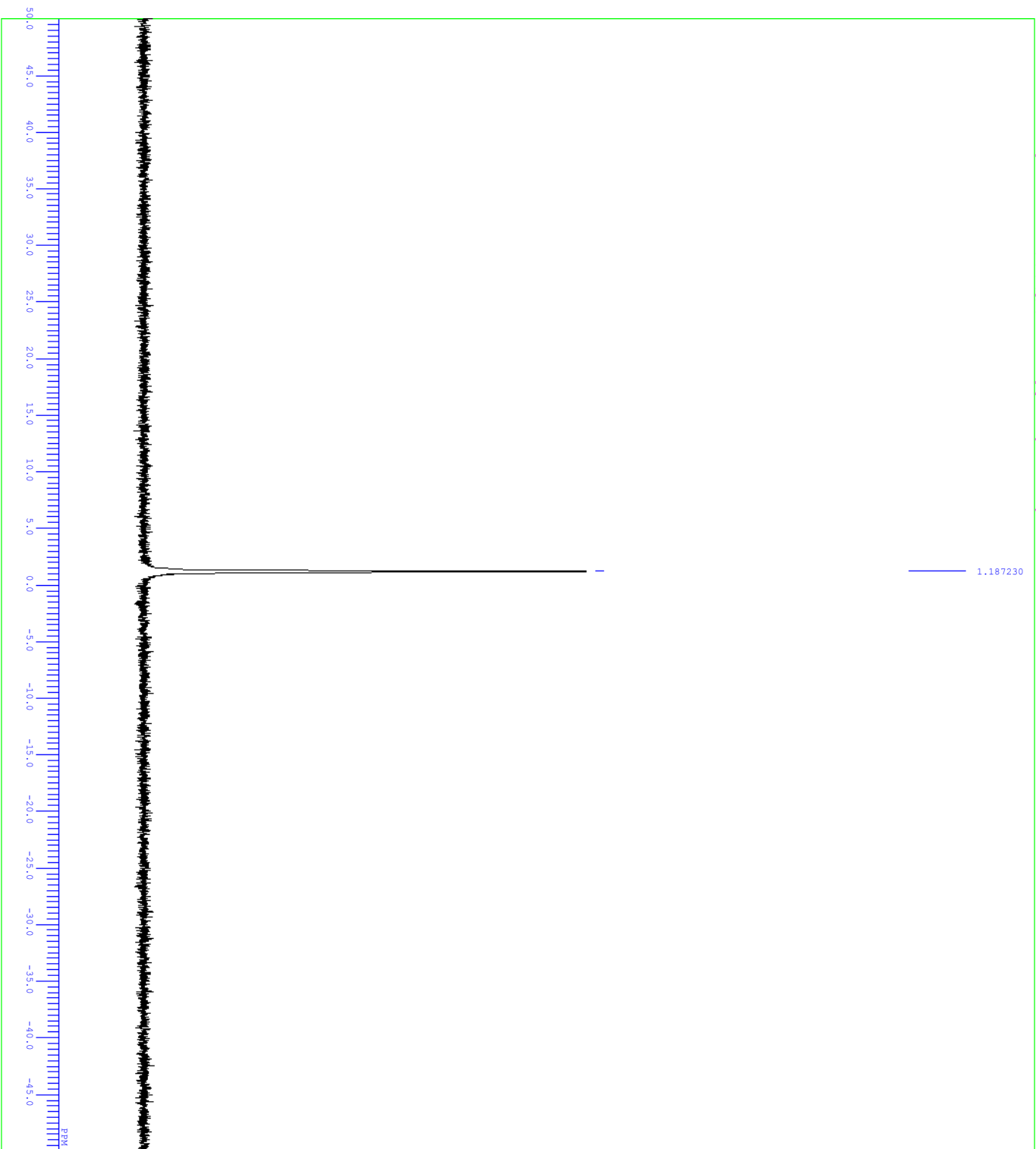
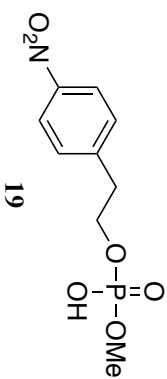
PMI	4.80 use
PMI	4.80 use

IRNUC	IH
000	000

CALVIN	200000
SIVNT	CDCL3

EXREF 0.00 ppm

BE	0.12 HZ
BGAIN	56





4.17986
4.16154
4.14322
4.12490



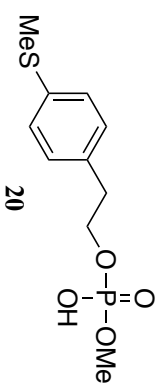
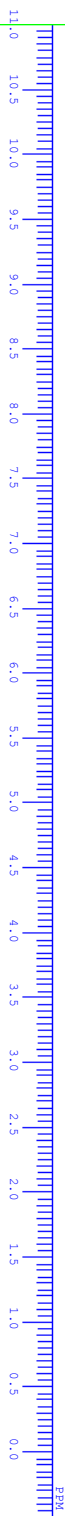
3.64178
3.61316



2.93885
2.92054
2.90336



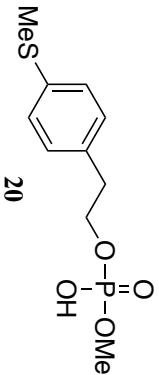
2.43627
2.40880

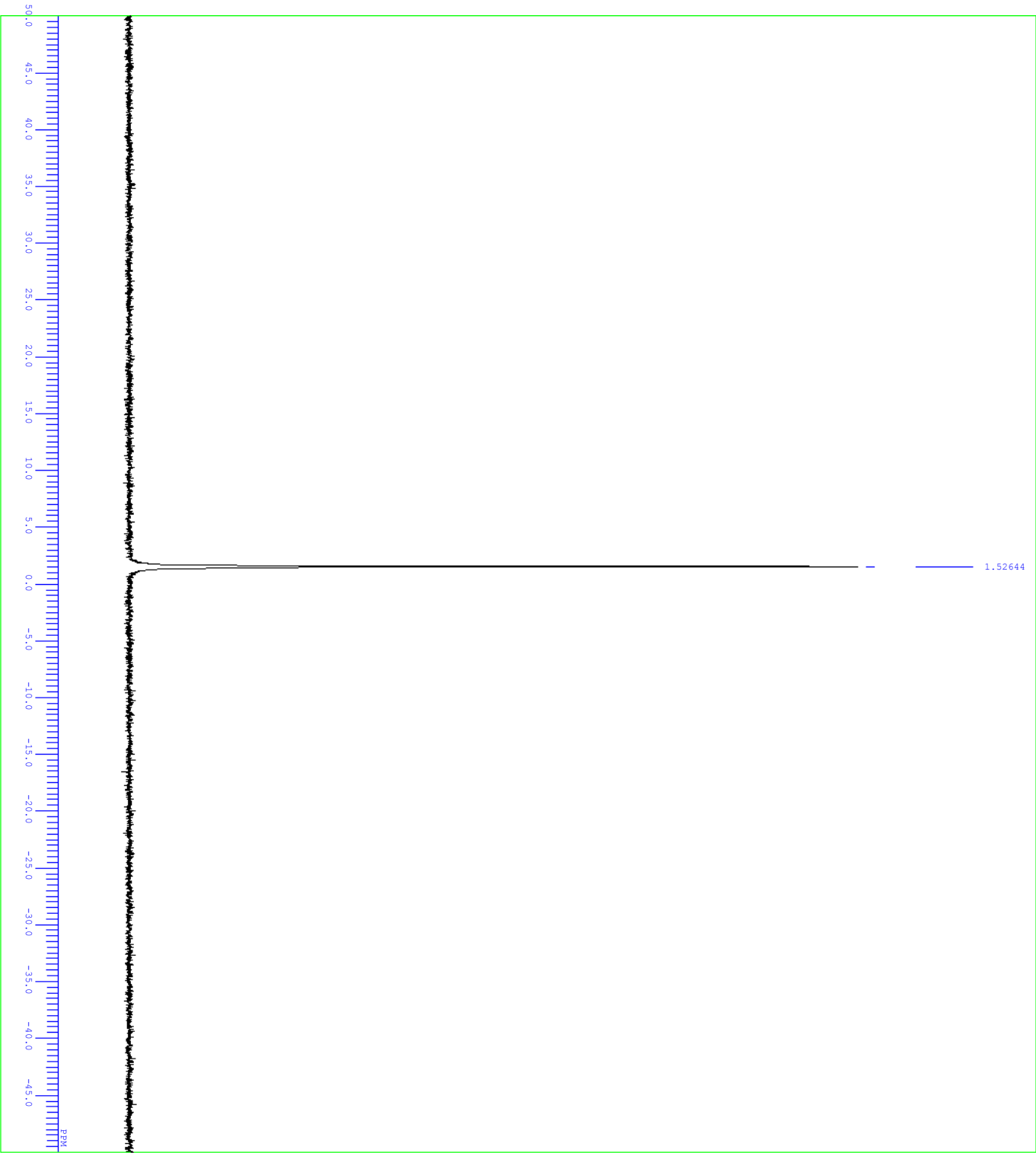


CONT	RN1741043-HPC-B-1-i_als
CCNT	
DATM	29-04-2021 14:14:42
ONUC	1h
EYMO	proton.jpg
OBFO	331.8 MHz
OBSST	8.51 KHz
ORBN	3.34 Hz
POINT	13107
FREQO	5878.90 Hz
ACQMS	8
SCANS	2.2295 sec
FW	6.0000 sec
PD	5.17 usec
CTEM	1h
RNUC	20.7 c
CHCL3	
SIVNT	7.24 ppm
EXREF	1.20 Hz
B ⁺	
GAIN	38

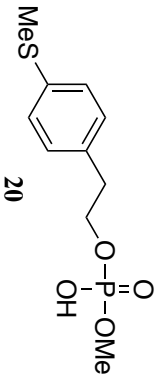


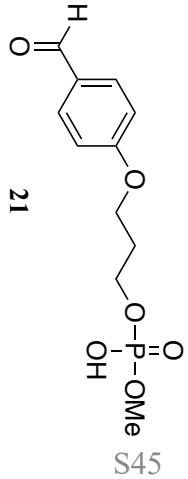
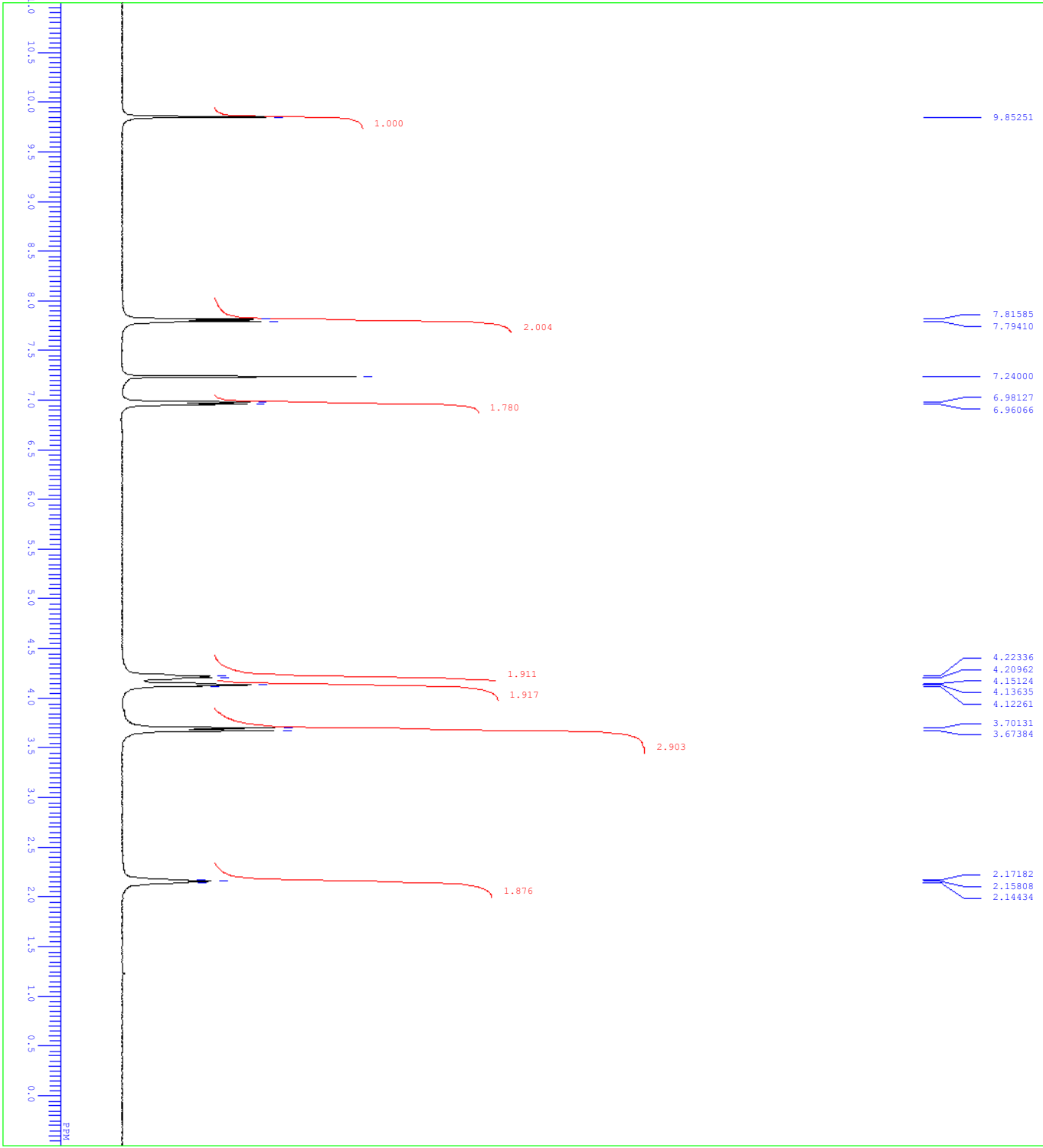
DFILE KF11741043-HPLC-C-1-1.a1a
COMNT
DATIM 29-04-2021 14:23:19
EXMNUC 13C
EXMOD carbon-13xp
OBFREQ 98.52 MHz
OBSSET 4.64 KHz
OBSFIN 8.774 Hz
POINT 26214
FREQU 24630.34 Hz
SCANS 1073
RGAIN 1.000 sec
PUL 2.000 sec
PRG 3.112 usec
IRNUC 1H
CTEMP 21.0 C
SIVNT CHCL3
EXREF 77.00 ppm
BF 1.20 Hz
RGAIN 60



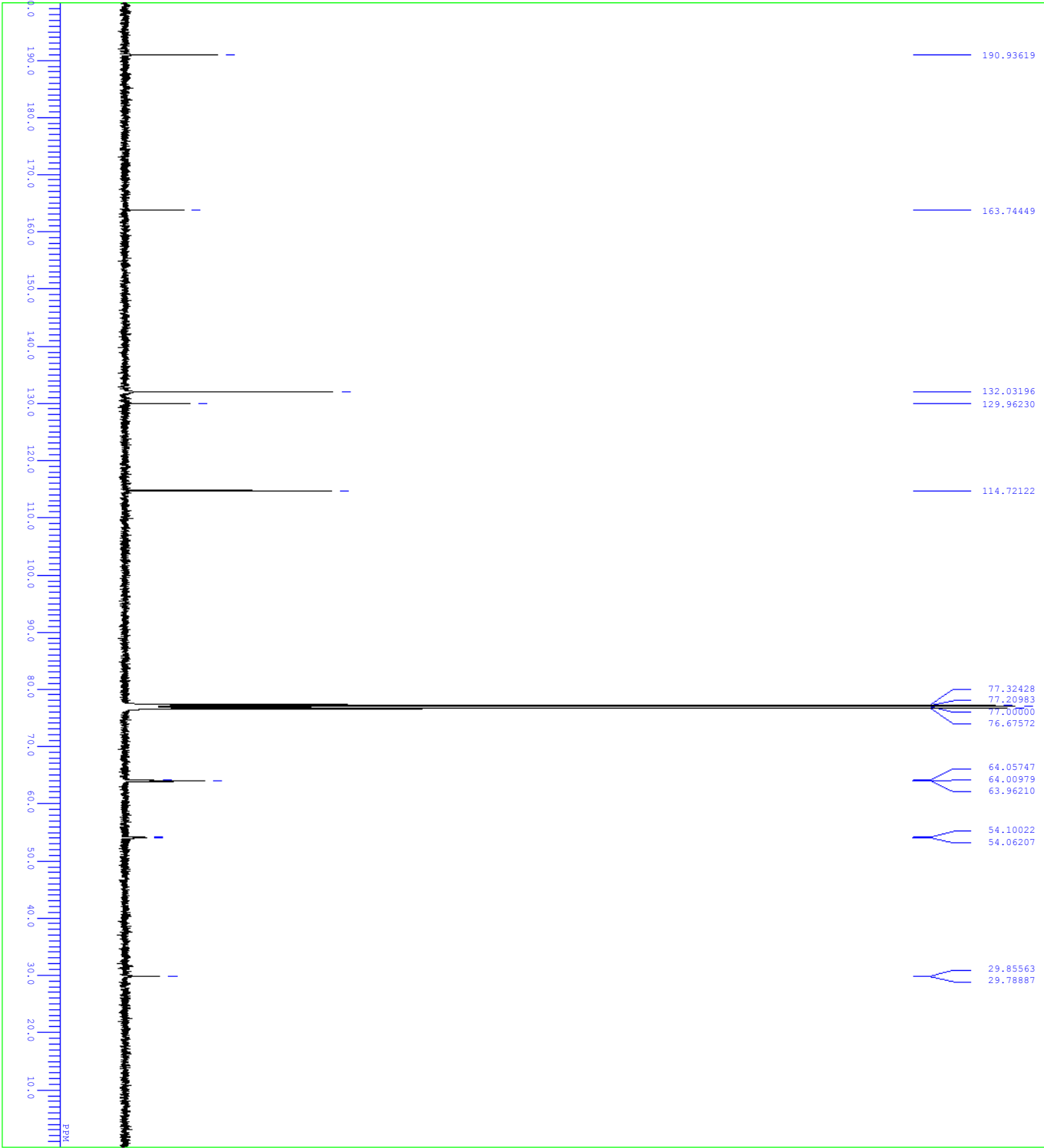


DPFILE FF11741043-HPLC-P-1-1.als
COMNT
DATIM 29-04-2021 14:18:19
ORNUC 31P
EXMOD carbon-13XP
OBFREQ 158.59 MHz
OBSSET 7.99 KHz
OBFIN 9.23 Hz
NUC1 31P
PREFOU 64102.14 Hz
SCANS 164
ACQTM 0.4089 sec
PD 2.0000 sec
PWL 4.80 usec
IRNUC 1H
CTEMP 20.9 C
SLVNT CDCL3
EXREF 0.00 ppm
BF 1.20 Hz
RGAIN 56

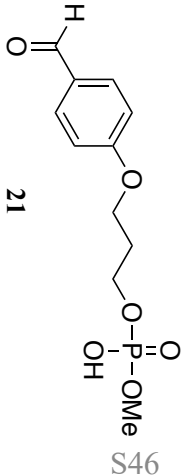


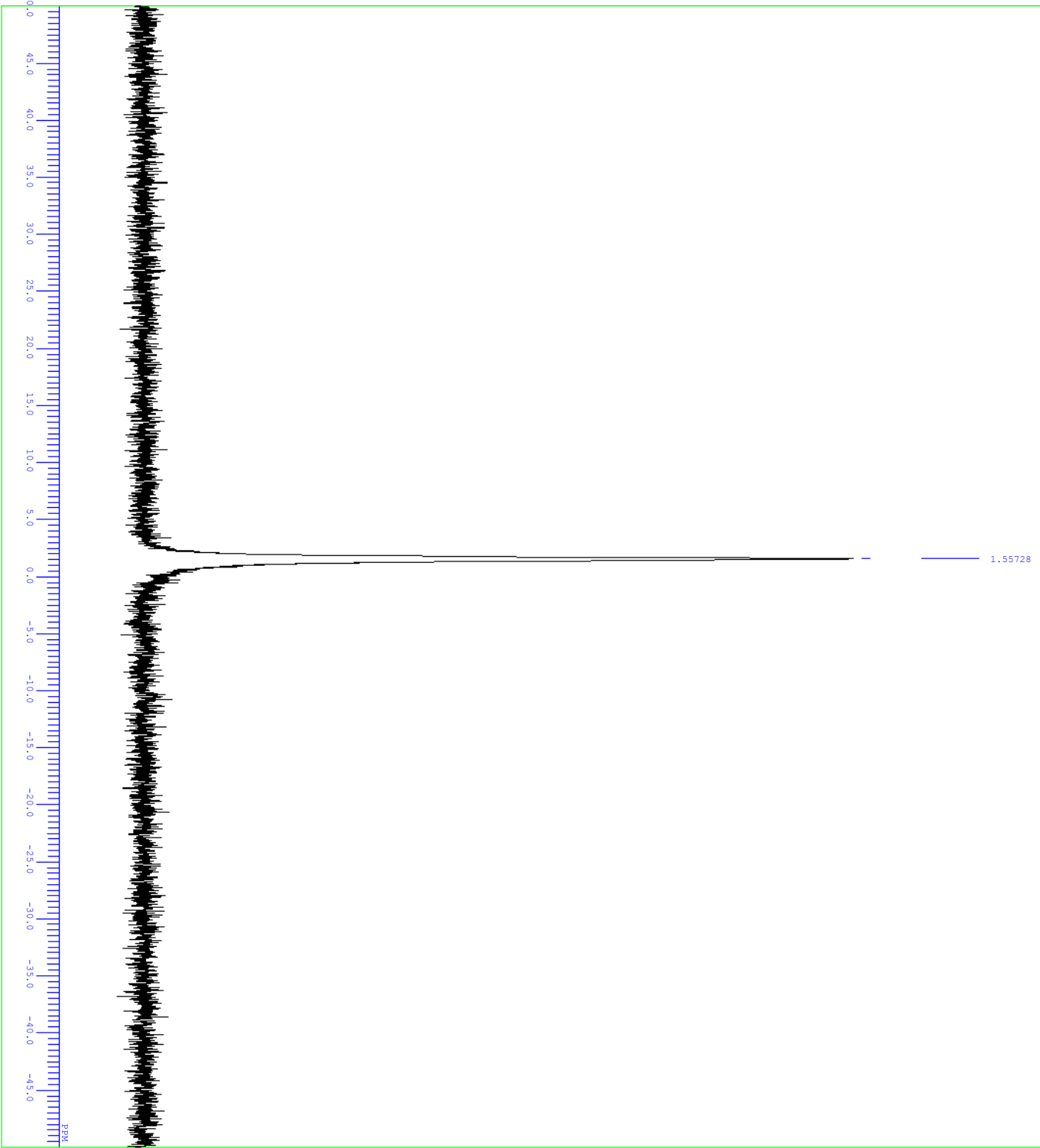


DPFILE FF11741044-HPLC-H-3-1-1-1a1a
CONNT 29-04-2021 19:11:15
OBNUC 1H
EXMOD proton-j3p
OBSRQ 391.78 MHz
OBSST 8.51 KHz
OBSIN 3.34 Hz
OBSIT 13104 Hz
PREFOU 5878.100 Hz
SCANS 8
ACQTM 2.2295 sec
PD 6.0000 sec
PWL 5.17 usec
IRNUC 1H
CTEMP 20.8 C
SIVNT CDCL3
EXREF 7.24 ppm
BF 1.20 Hz
RGAIN 46

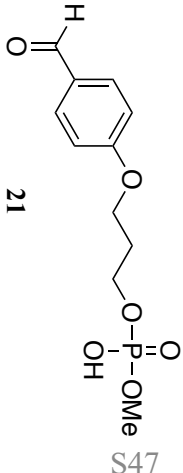


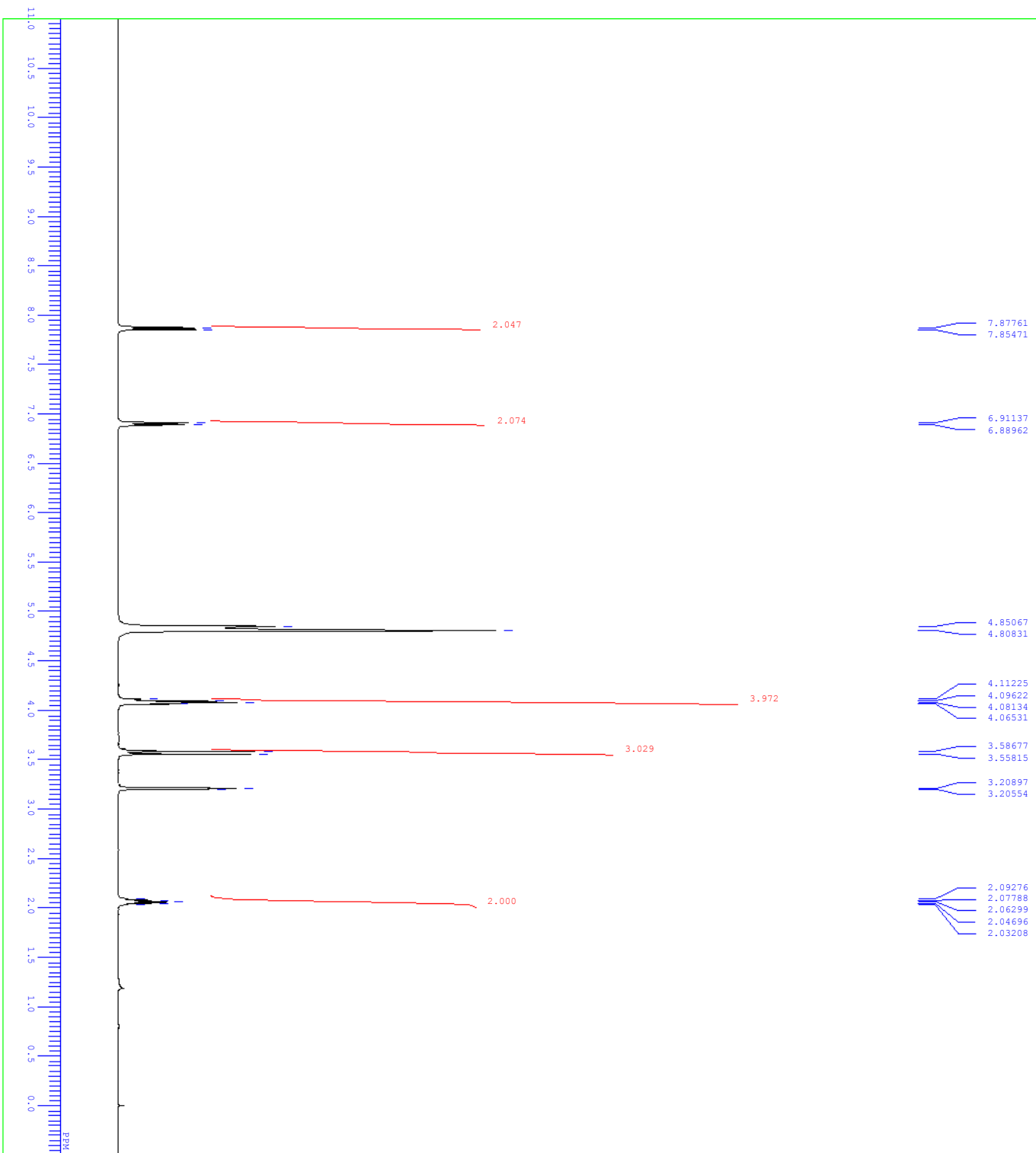
DTITLE KF11741044-HPLC-C-1-1.s
COUNT 29-04-2021 15:24:14
ORNUC 13C
EXMRQ carbon-13P
ORNUC 98.52 MHz
ORNUC 8.42 kHz
ORNUC 8.74 Hz
ORNUC 2621.4 Hz
FREQOU 24630.54 Hz
SCANS 1040
ACQTM 1.0643 sec
PD 2.0000 sec
PWL 3.12 usec
IRNUC 1H
CTEMP 21.0 C
SIVNT CDCL3
EXREF 77.00 ppm
BF 1.20 Hz
RGAIN 60



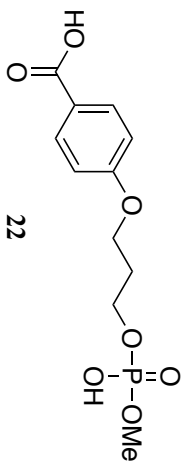


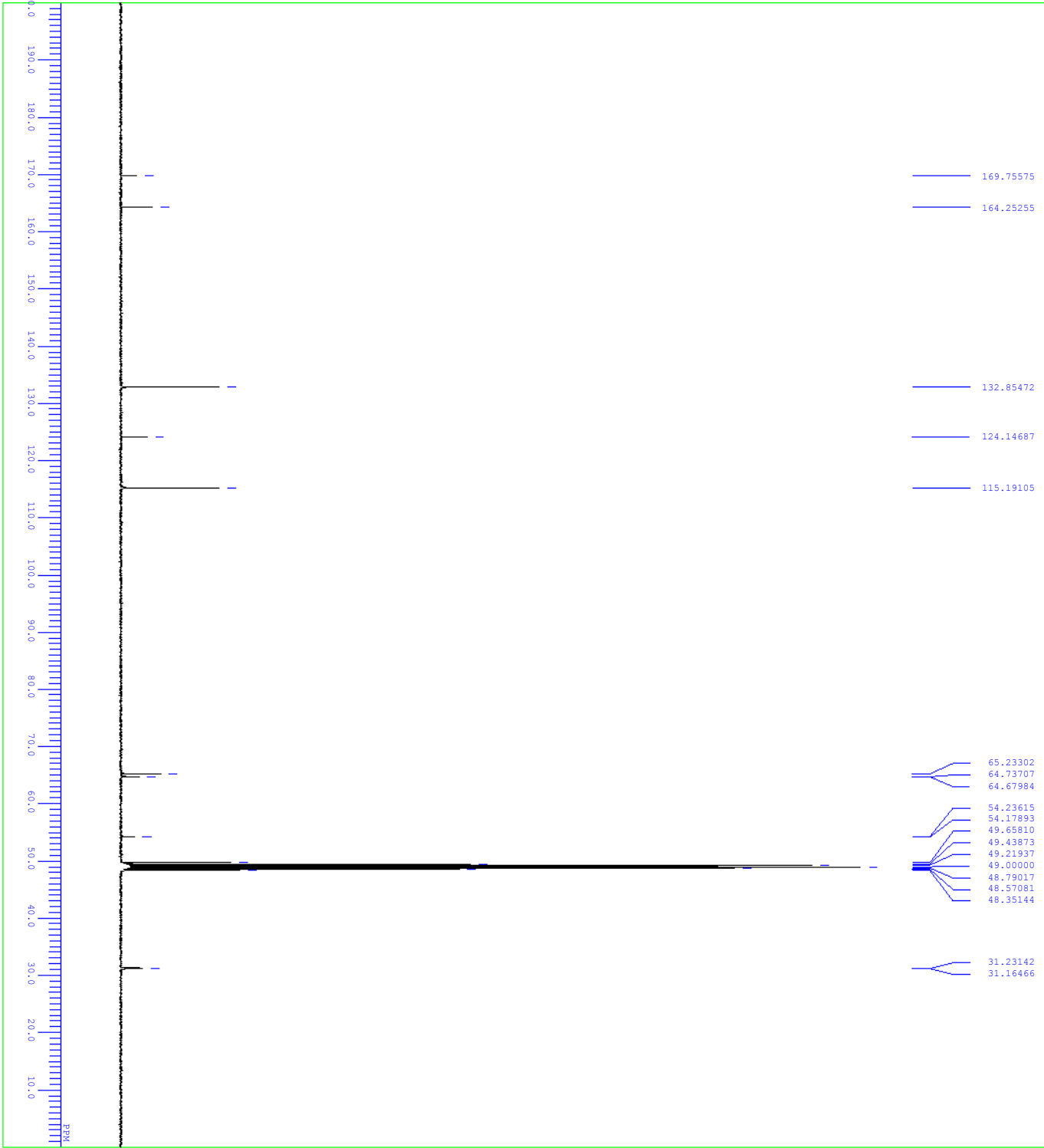
DFILE KF11741044-HPLC-P-1-1.sls
CONTC 29-04-2021 16:21:60
ORNUC 31P
EXMRQ carbon-13P
OBSRQ 158.59 MHz
OBSF 127.49 MHz
OBSR 9.23 KHz
OBSPT 26214 Hz
FREQU 64102.56 Hz
SCANS 125
ACQTM 0.4089 sec
PD 2.0000 sec
PWL 4.90 usec
IRNUC 1H
CTEMP 20.9 C
SLVNT CDCL3
EXREF 0.00 ppm
BF 1.20 Hz
RGAIN 56



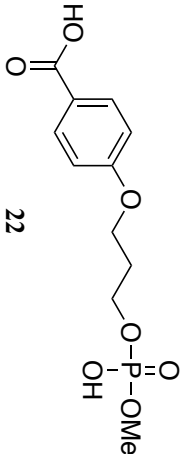


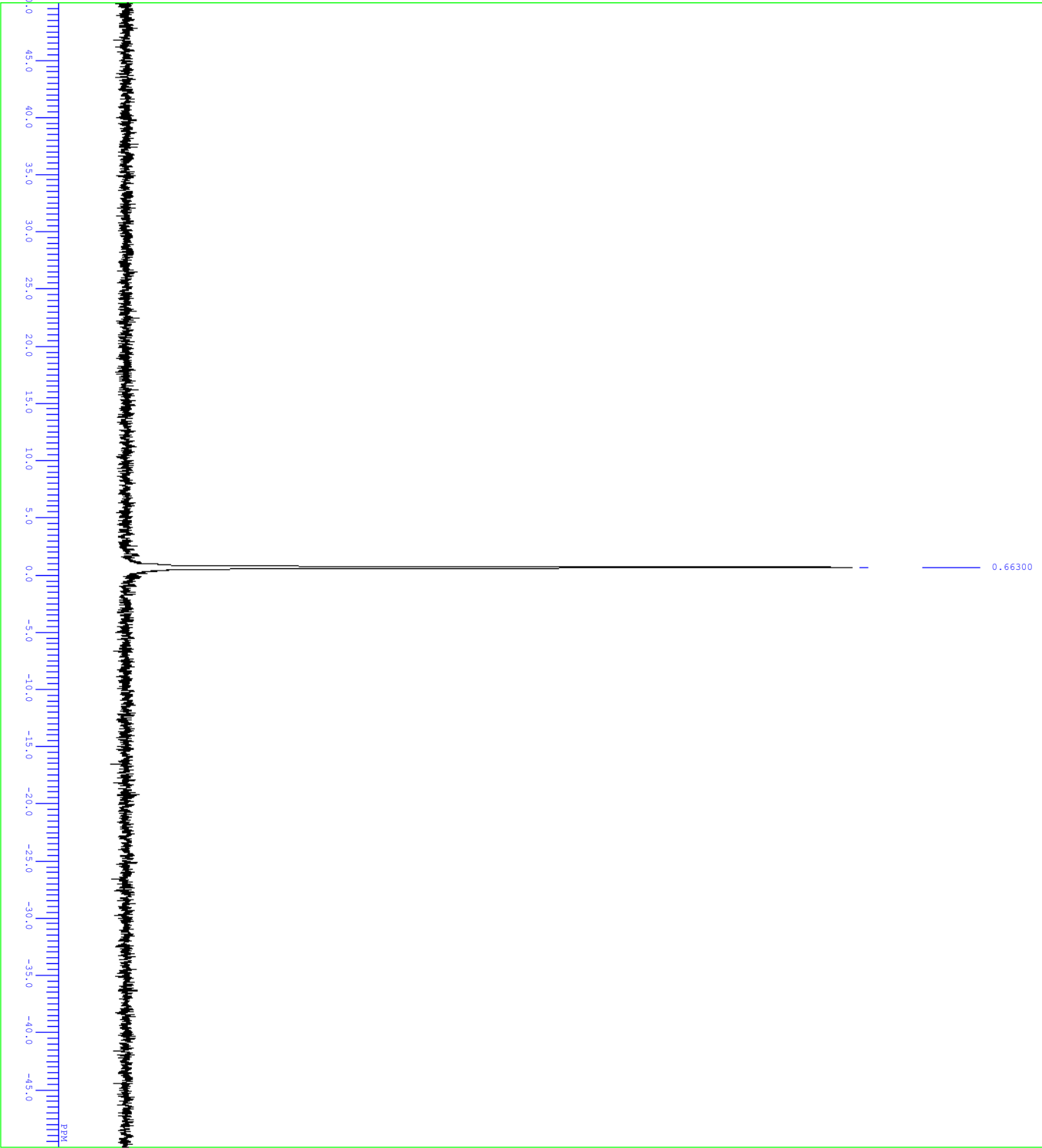
DFILE FF11741045-HPLC-1-H-1-1.a1a
 COMNT
 DATIM 29-04-2021 19:36:19
 OBNUC 1H
 EXMOD proton-j3p2
 OBRFQ 391.78 MHz
 OBRFQ 8.51 KHz
 OBRFQ 1.334 Hz
 PREOU 5878.100 Hz
 SCANS 8
 ACQTM 2.2295 sec
 PD 6.0000 sec
 PUL 5.17 usec
 TRNUC 1H
 CTEMP 20.8 C
 SLVNT CD3OD
 EXREF 0.00 ppm
 BF 1.20 Hz
 RGAIN 38



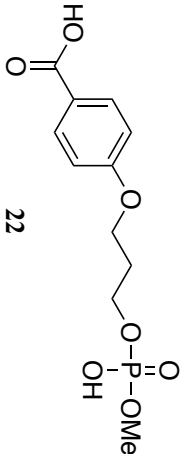


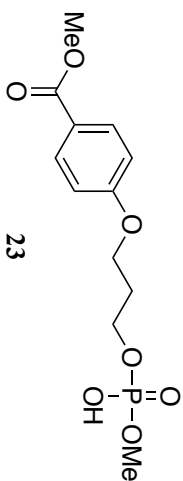
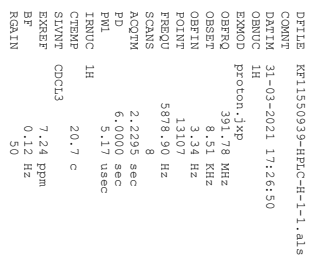
DPFILE FF11741045-HPLC-1-C-1-1.a1a
COMNT
DATIM 29-04-2021 19:47:45
IRNUC 13C
EXMOD carbon-13p2
OBFREQ 98.92 MHz
OBSSET 4.64 KHz
OBSFID 8.774 Hz
PREFID 2627.4 Hz
PREFOU 246315.4 Hz
SCANS 1.0643 sec
ACQTM 2.0000 sec
PD 3.12 usec
PWL
IRNUC 1H
CTEMP 21.0 C
SOLVT CD3OD
EXREF 49.00 ppm
BF 1.20 Hz
RGAIN 60

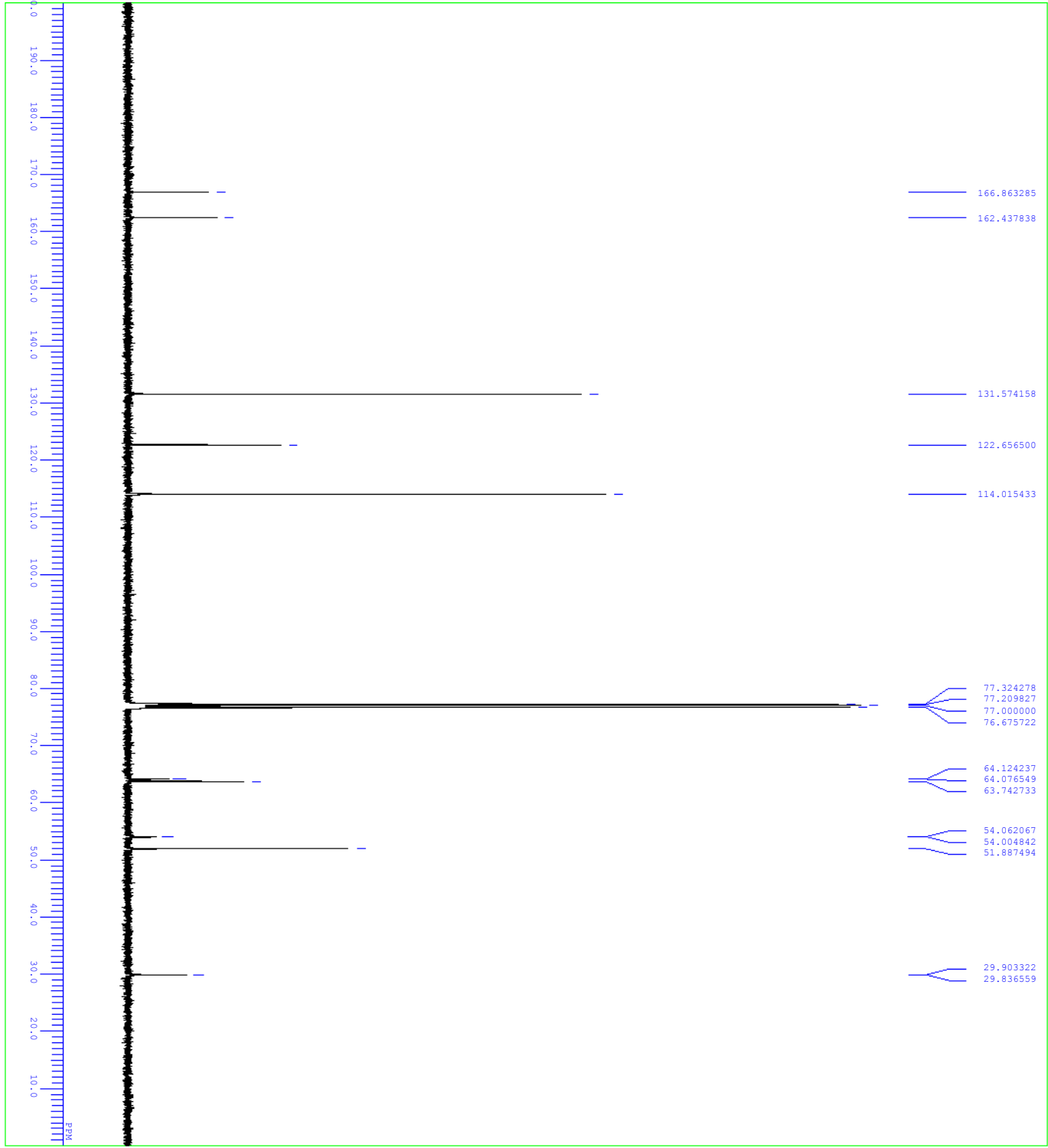




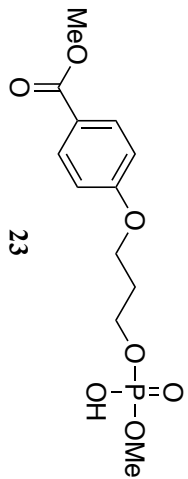
DPFILE FF1741045-HPLC-1-P-1-1.a1a
COMNT
DATIM 29-04-2021 19:39:44
ORNUC 31P
EXMOD carbon-13XP
OBSFREQ 158.59 MHz
OBSSET 7.99 KHz
OBSFID 9.23 Hz
P1 2.6214
P2 64102.14 Hz
P3 1150
SCANS 0.4089 sec
ACQTM 2.0000 sec
PD 4.80 usec
PWL
IRNUC 1H
CTEMP 21.0 C
SLVNT CD3OD
EXREF 0.00 ppm
B1 1.20 Hz
RGAIN 56

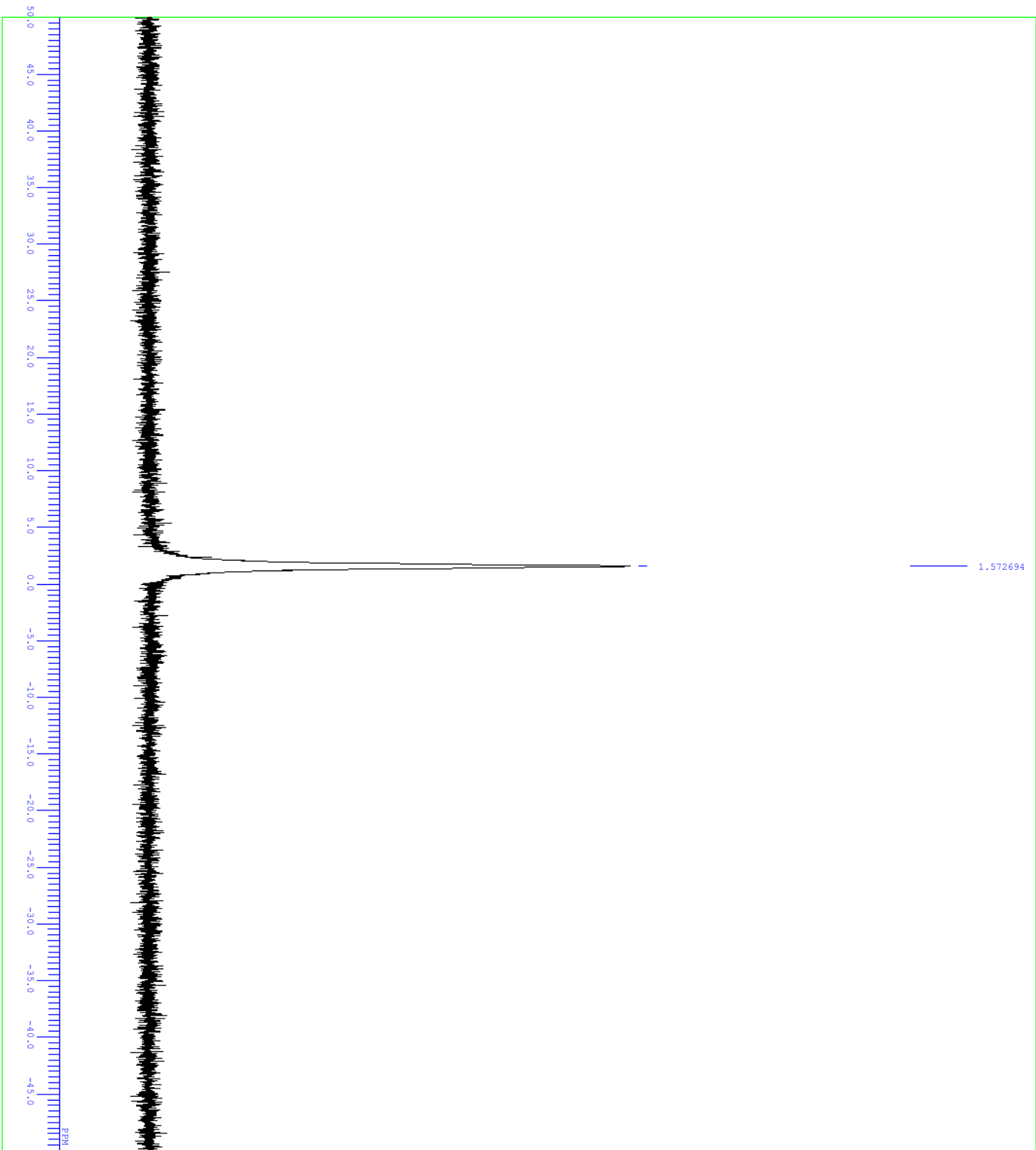






DPFILE KF1150939-HPLC-C-1-1.a.s
COMNT
DATIM 31-03-2021 17:52:46
OBNUC 13C
EXMOD carbon-13p
OBFREQ 98.52 MHz
OBSSET 4.64 KHz
FOFINT 8.74 Hz
FRESQU 24630.54 Hz
SCANS 2823
ACQTM 1.0843 sec
PUL 2.0000 sec
PULPRG 3.12 usec
IRNUC 1H
CTEMP 21.0 C
SOLVENT CHCL3
EXREF BF
RGAIN 60





DFFILE RF1150939-HPLC-P-1-1.als
COMNT
DATIM 31-03-2021 17:38:17
OBNUC 31P
EXMOD carbon-13p
OBFREQ 158.59 MHz
OBSSET 7.99 KHz
OBSINT 9.23 Hz
FOLINT 2621.4 Hz
FREQD 64102.14 Hz
SCANS 1100
SCAND 0.4089 sec
PCQTM 2.0000 sec
PD 4.80 usec
PWL
IRNUC 1H
CTEMP 20.9 C
SIVNT CDCL3
EXREF 0.00 ppm
BF 0.12 Hz
RGAIN 56

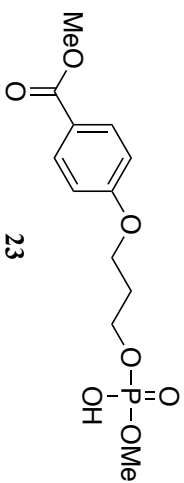
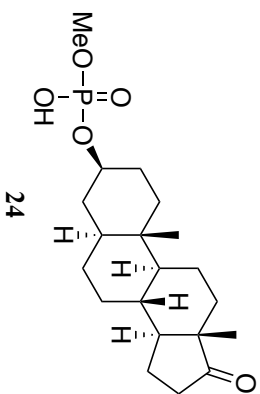
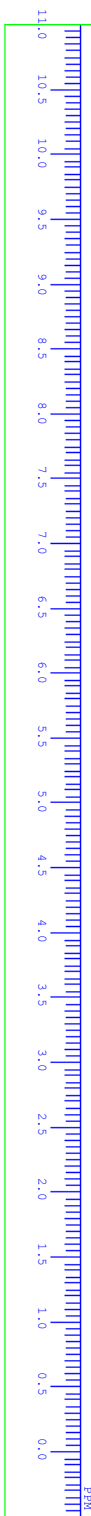


Figure 1: Spatial distribution of the first 100 eigenvalues of the Laplace operator on a rectangular domain.



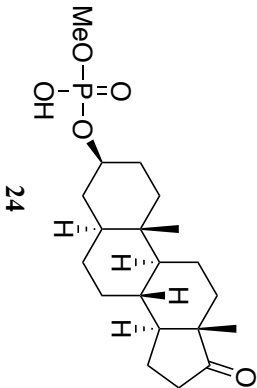
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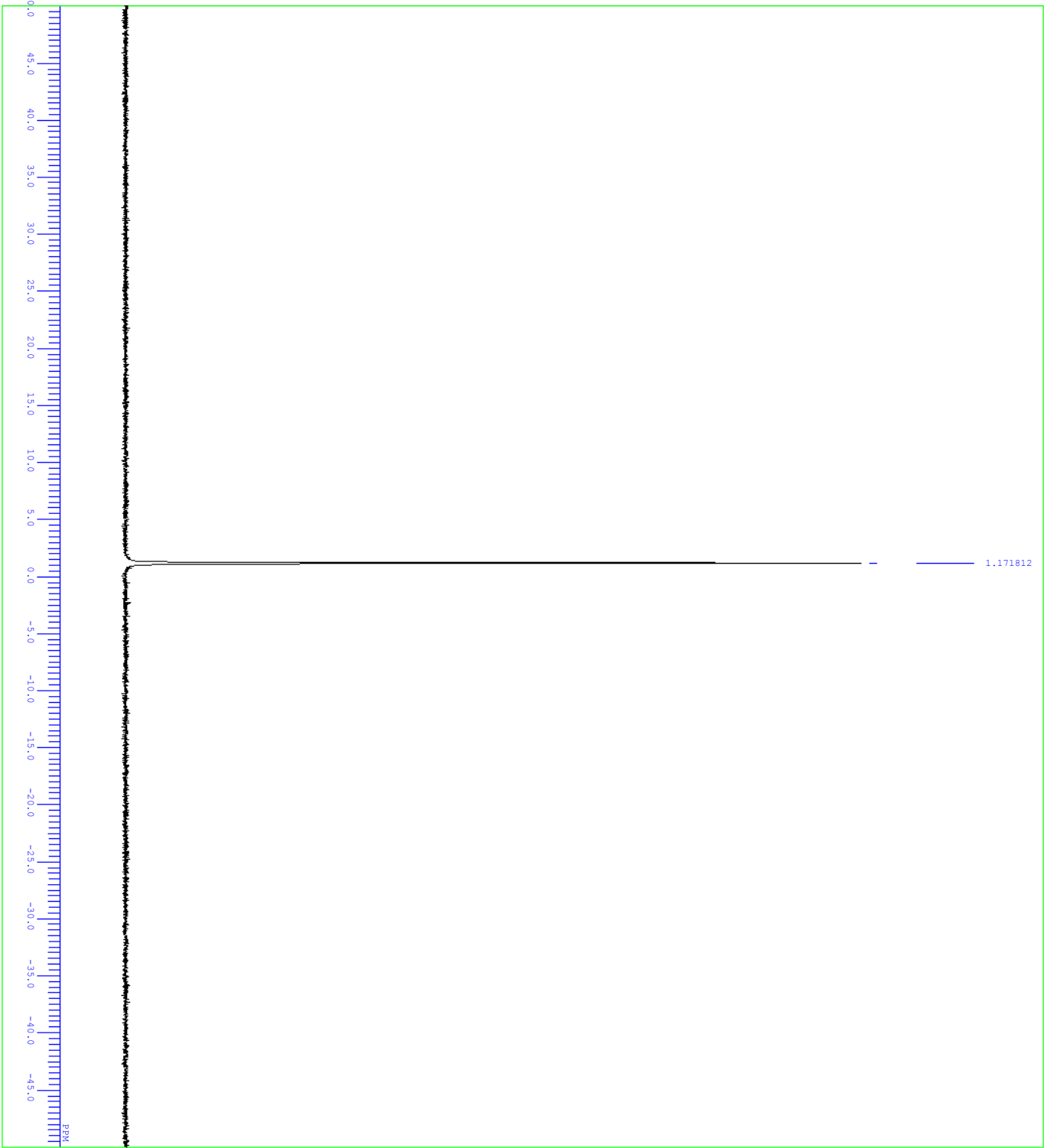
DEFNAME      N1163098+9BC3-N-1-1.1a
CONTENT
COMMENT
DATE          17-04-2021 18:49:48
DNAME         1H
OBSNUM        1H
EXMOD         proton_xjp
OBSFO         331.78 MHz
OBSSET        8.51 KHz
OBSIN         3.34 KHz
POINT         13107
FREQO         5898.90 KHz
SCANS         8
ACQIM         2.2295 sec
PD            6.0000 sec
PWI          5.117 usec
CTEMP         1H
SLANT         20.7 c
CDCL3         7.24 ppm
EXREF         0.12 Hz
GAIN          34

```

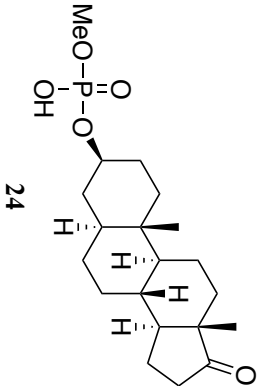


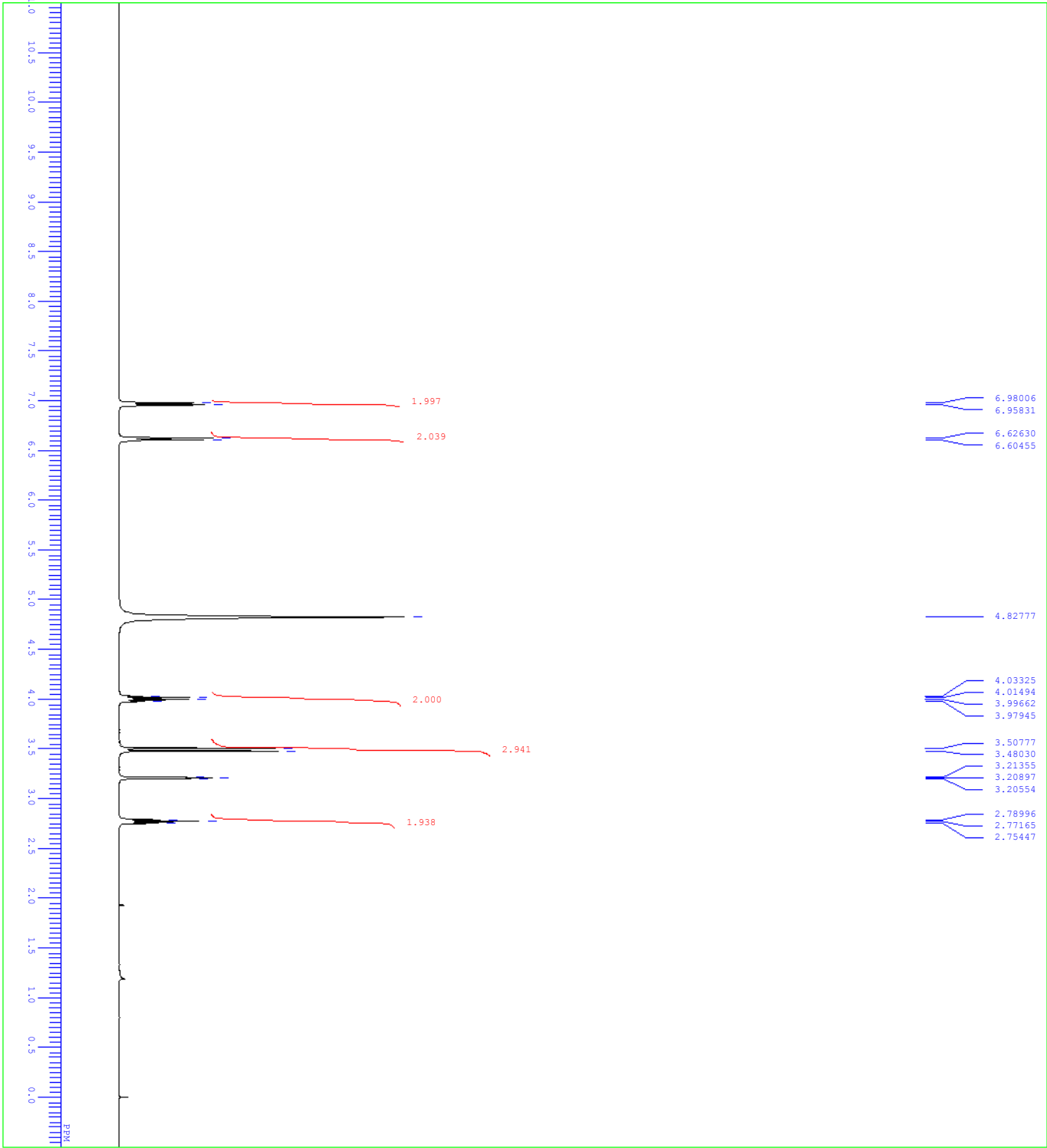
DFILE F11630998-HPLC-C-2-1-1.a1s
COMET 17-04-2021 18:53:04
NAME 13C
EXPNO 320
PROCNO 1
F2 - 98.52 MHz
F1 - 4.64 kHz
FREQ 8.774 Hz
P1 2621.4 Hz
P2 24630.54 Hz
P3 1742 Hz
P4 1.0643 sec
P5 2.0000 sec
P6 3.112 usec
IRNUC 1H
IRNUC 1H
CTEMP 20.9 C
SOLVENT CDCl3
SOLVENT 77.00 ppm
BF 0.12 Hz
RGAIN 60



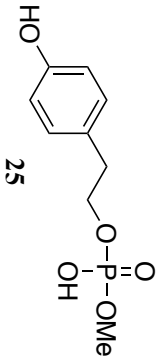


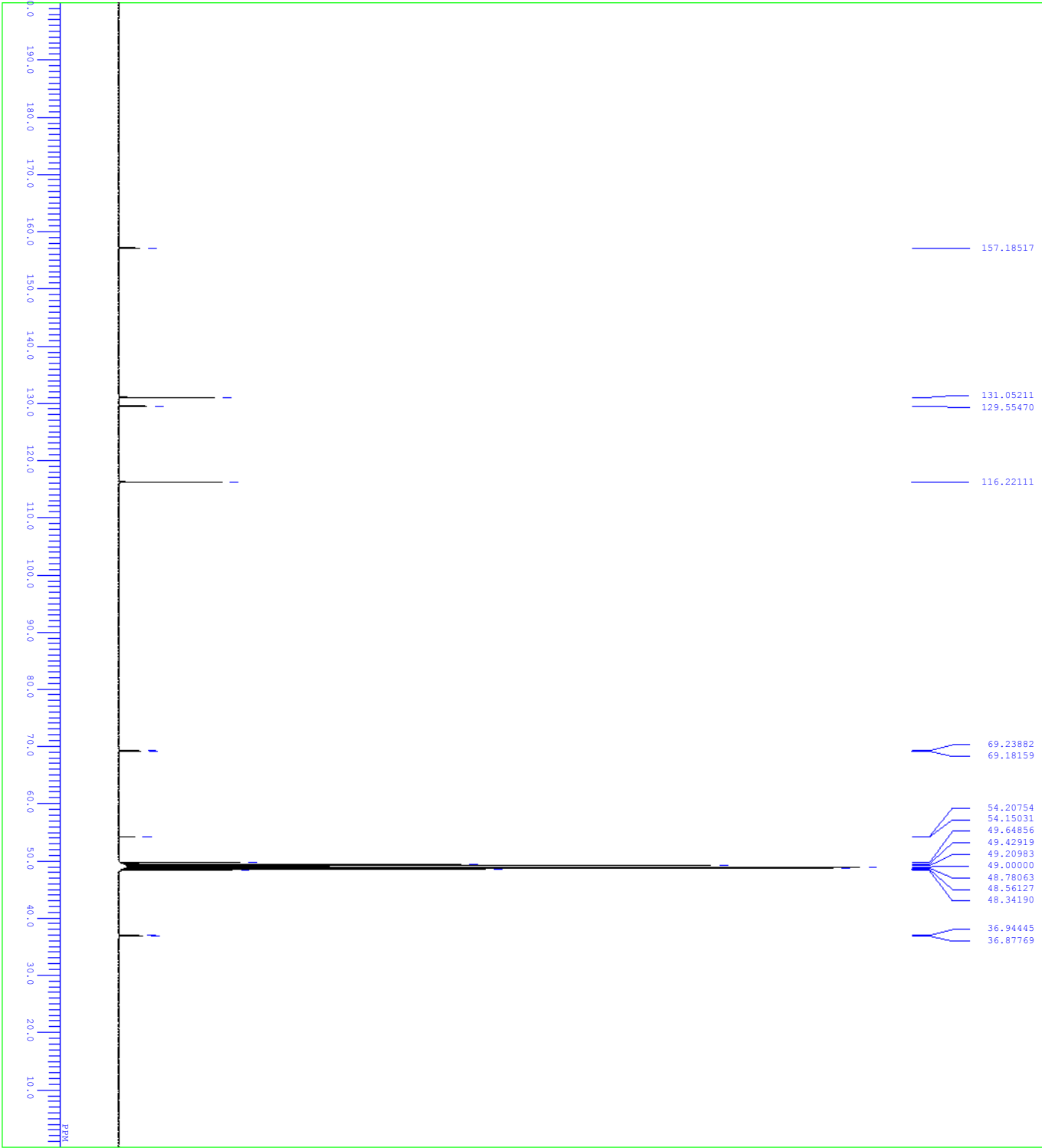
DFILE KF11630998-HPLC-P-1-1.als
COMET 17-04-2021 12:31:19
OBNUC 31P
EXMOD carbon-13xp
OBFREQ 158.59 MHz
OBSSET 7.99 KHz
OBFIN 9.23 Hz
POINT 26214
FREQU 64102.56 Hz
SCANS 129
ACQTM 0.4089 sec
PUL 2.0000 sec
PR 4.90 usec
IRNUC 1H
CTEMP 20.8 C
SOLVT CHCL3
EXREF 0.00 ppm
BF 0.12 Hz
RGAIN 56



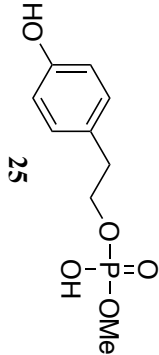


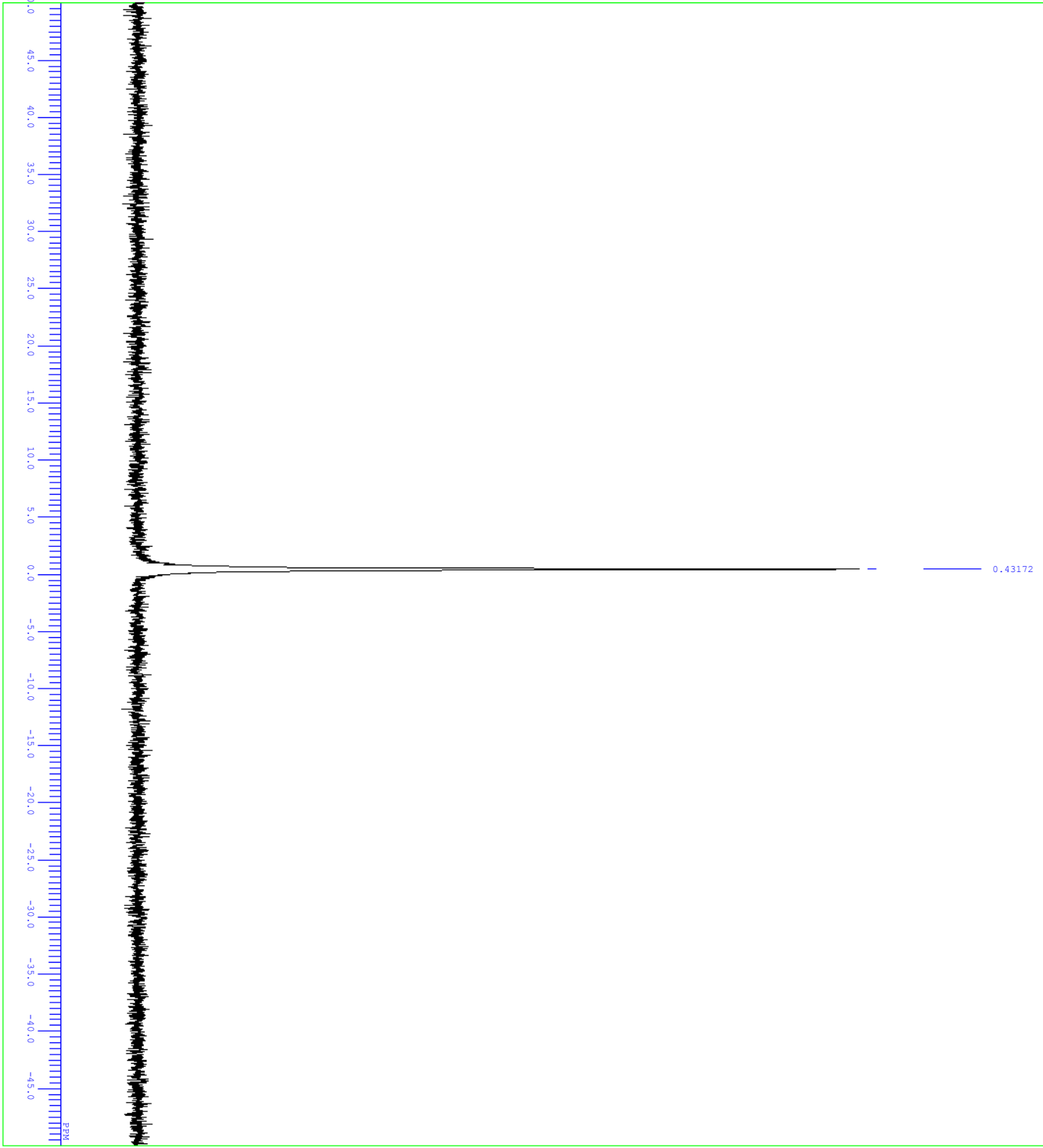
DFILE KF11741042-HPLC2-H-1-1.a1s
CONNT
DATIM 30-04-2021 00:29:35
EXMOD proton.jxp
OBSFQ 391.78 MHz
OBSST 8.51 KHz
OBSIN 13107
POINT 5878.90 Hz
SCANS 8
ACQTM 2.2295 sec
PUL 6.917 sec
IRNUC 1H
CTEMP 20.8 C
SIVNT
EXREF
BF 1.20 Hz
RGAIN 38



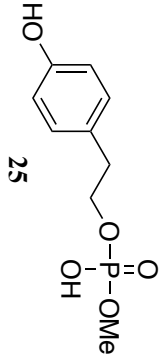


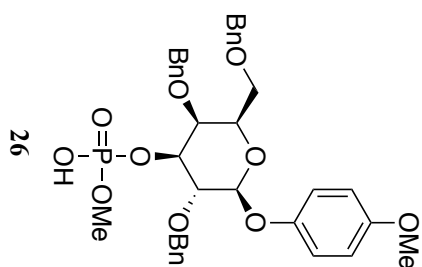
DTITLE FF11741042-HPLC2-C-1-1.a1s
COUNT 30
DATEIN 30-04-2021 00:40:36
EXMOD 13C
EXMOD carbon-13p
OBSFQ 98.52 MHz
OBSRT 4.64 kHz
OBSIN 8.271 Hz
POINT 26314 Hz
FREQD 24630.54 Hz
SCANS 12741
ACQTM 1.0643 sec
PD 2.0000 sec
PUL 3.12 usec
IRNUC 1H
CTEMP 21.0 C
SLVNT CD3OD
EXREF 49.00 ppm
BF 1.20 Hz
RGAIN 60



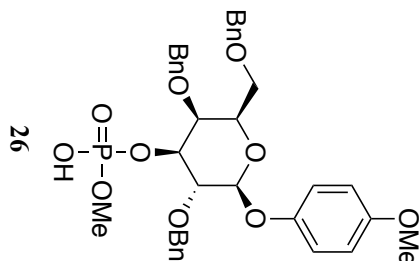
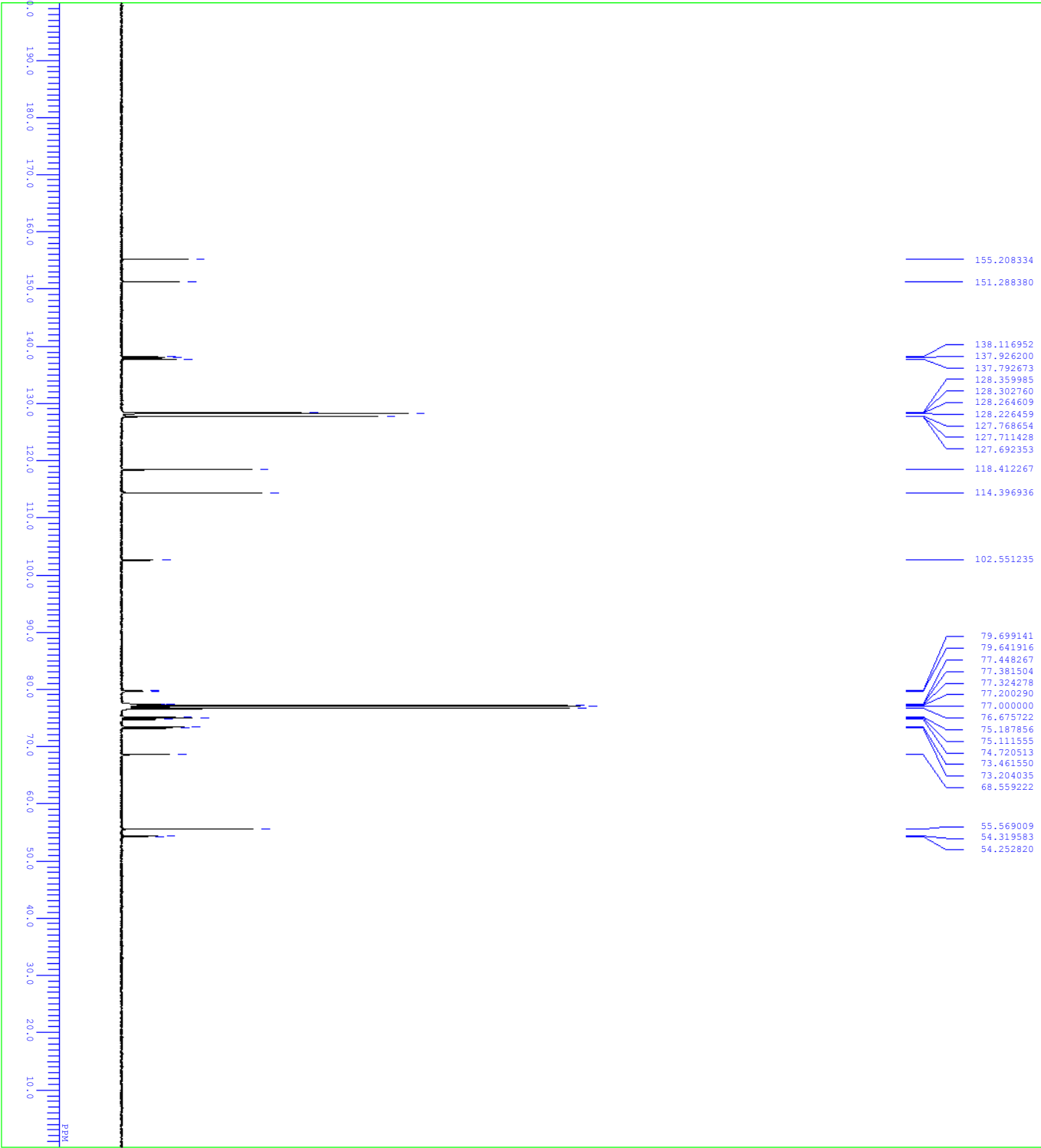


DFFILE FF11741042-HPLC2-P-1-1.a1s
COMINT
DATIM 30-04-2021 00:32:58
OBNUC 31P
EXMOD carbon-13xp
OBFREQ 158.59 MHz
OBSSET 7.99 kHz
OBSIN 9.23 Hz
COLINT 262.4 Hz
FREQ 64102.14 Hz
SCANS 1129
XCOPIH 0.4089 sec
PD 2.0000 sec
PWL 4.80 usec
IRNUC 1H
CTEMP 21.1 C
SIVNT CD3OD
EXREF 0.00 ppm
BF 1.20 Hz
RGAIN 56

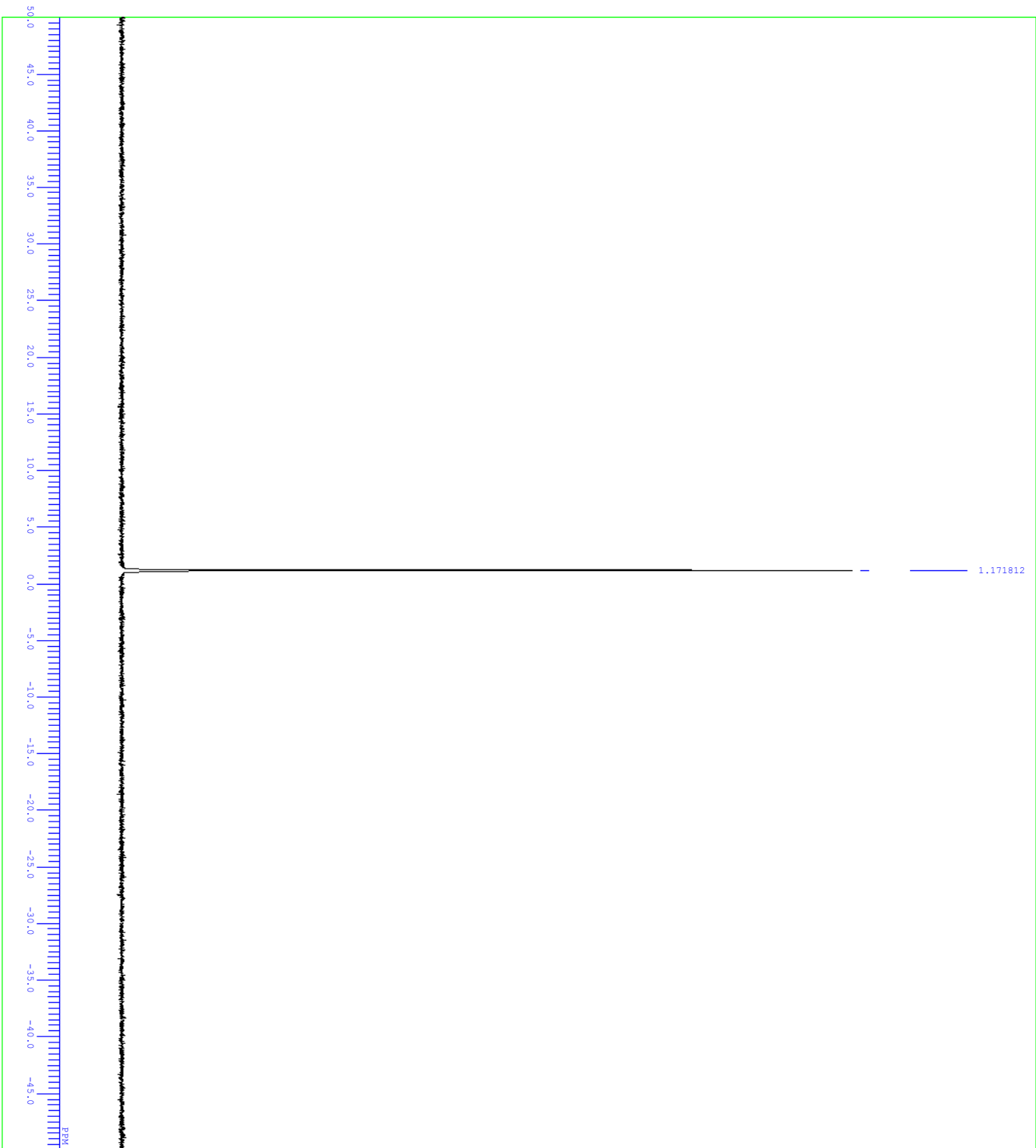




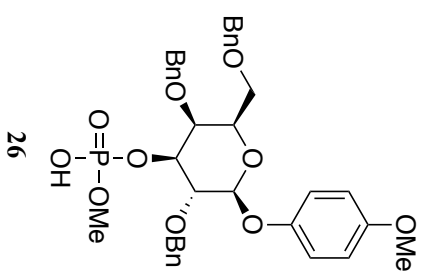
CONF	RN16163100I-HLCC-B-7-I_1.alis
DATIM	15-04-2021 22:38:20
ONBUC	IH
EXMOD	proton.jxp
ORBRQ	331.78 MHz
OBSEF	8.51 KHz
ORBIN	3.34 Hz
POINT	1.307
FREQU	5878.90 Hz
ACQMS	
SCANS	2,225 sec
PW	6,000 sec
FMI	5.17 usec
CTRUC	IH
CTEMP	20.8 C
SLOVT	CHCL3
EXREF	7.24 ppm
B ¹ H	0.12 Hz
GAIN	34

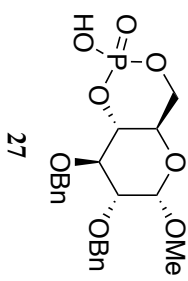
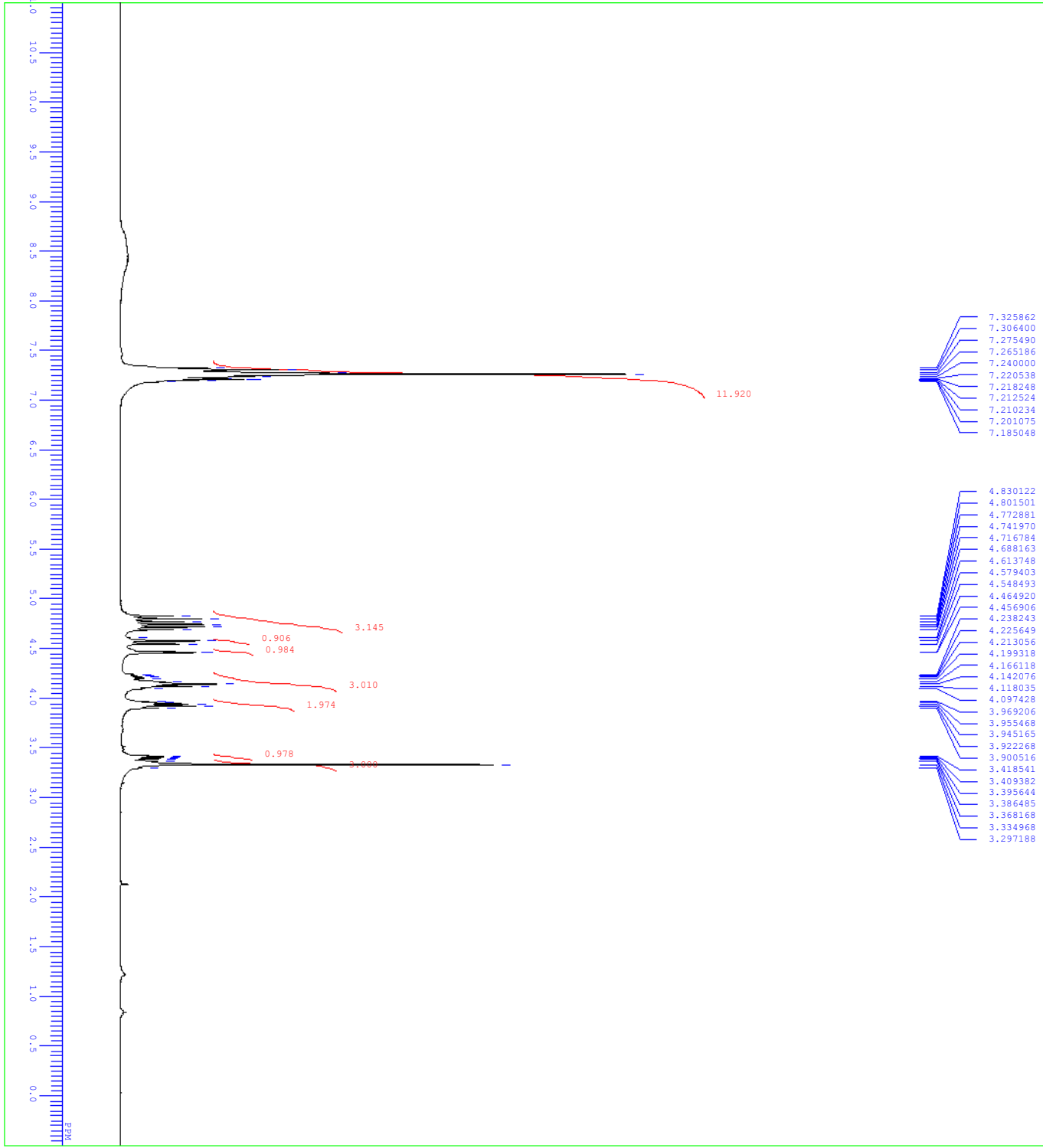


FF1631001-HPLC-C-1-1-1.sls
COMET
DATE 15-04-2021 22:48:12
EXNO 13C
EXNOF carbon-13
OBSF 98.52 MHz
PULPROG zgpg30
NUC1 13C
NUC2 13C
POINT 2621.4 Hz
FREQ 24630.54 Hz
SCANS 13306
ACQTM 1.0643 sec
PD 2.0000 sec
PUL 3.12 usec
IRNUC 1H
CTEMP 20.5 C
SOLVENT CDCL3
EXREF 77.00 ppm
BF 0.12 Hz
RGAIN 60

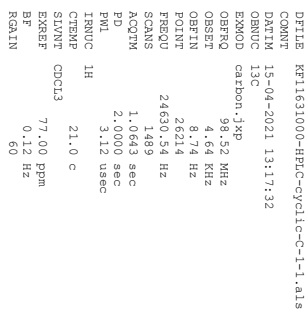


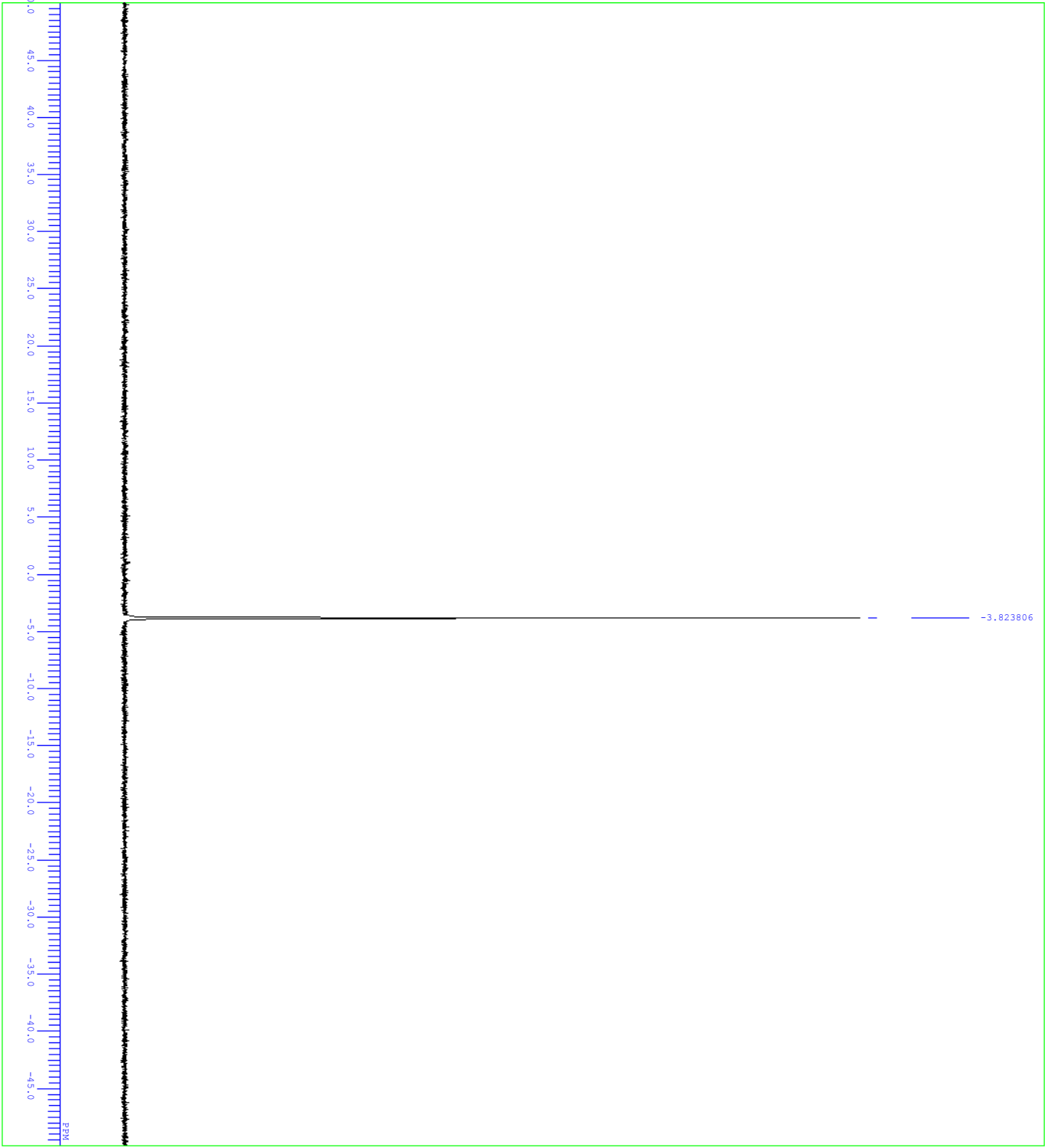
DFILE FF11631001-HPLC-P-1-1.ms
 COMMENT
 DATIM 15-04-2021 22:41:43
 OBNUC 31P
 EXMOD carbon-13p
 OBFREQ 158.59 MHz
 OBSSET 7.99 kHz
 OFFSET 9.23 Hz
 POINT 26214
 FREQ 64102.14 Hz
 SCANS 1100
 ACQTM 0.4099 sec
 PD 2.0000 sec
 PUL 4.80 usec
 TRNUC 1H
 CTMP 21.0 C
 SLVNT CDCL3
 EXREF 0.00 ppm
 BF 0.12 Hz
 RGAIN 56



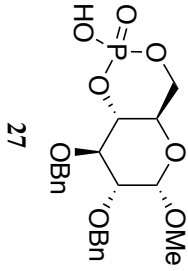


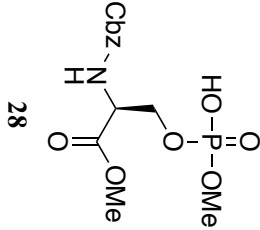
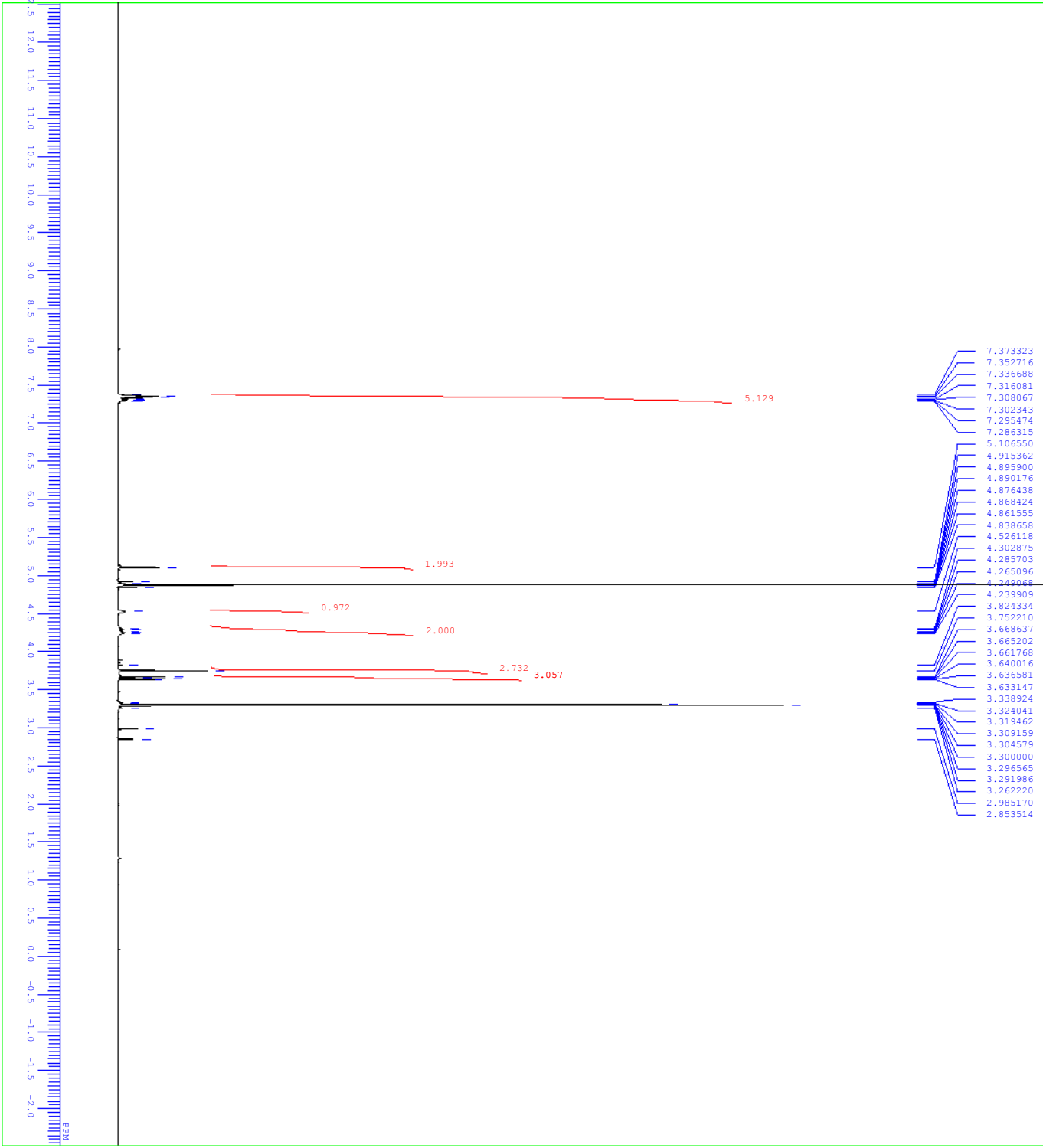
DTITLE F11631000-HPLC-Cyclic-H-1-1.a1s
COUNT 15-04-2021 13:08:35
ORNUC 1H
EXMRQ proton-3XPT
ORNUC 391.719 MHz
ORNUC 8.53 KHz
ORNUC 3.24 KHz
POINT 13107
FREQU 5878.90 Hz
SCANS 8
ACQTM 2.2295 sec
PD 6.0000 sec
PWL 5.17 usec
IRNUC 1H
CTEMP 20.8 C
SOLVT CDCL3
SOLVT 7.24 ppm
EXREF BF
RGAIN 0.12 Hz
36



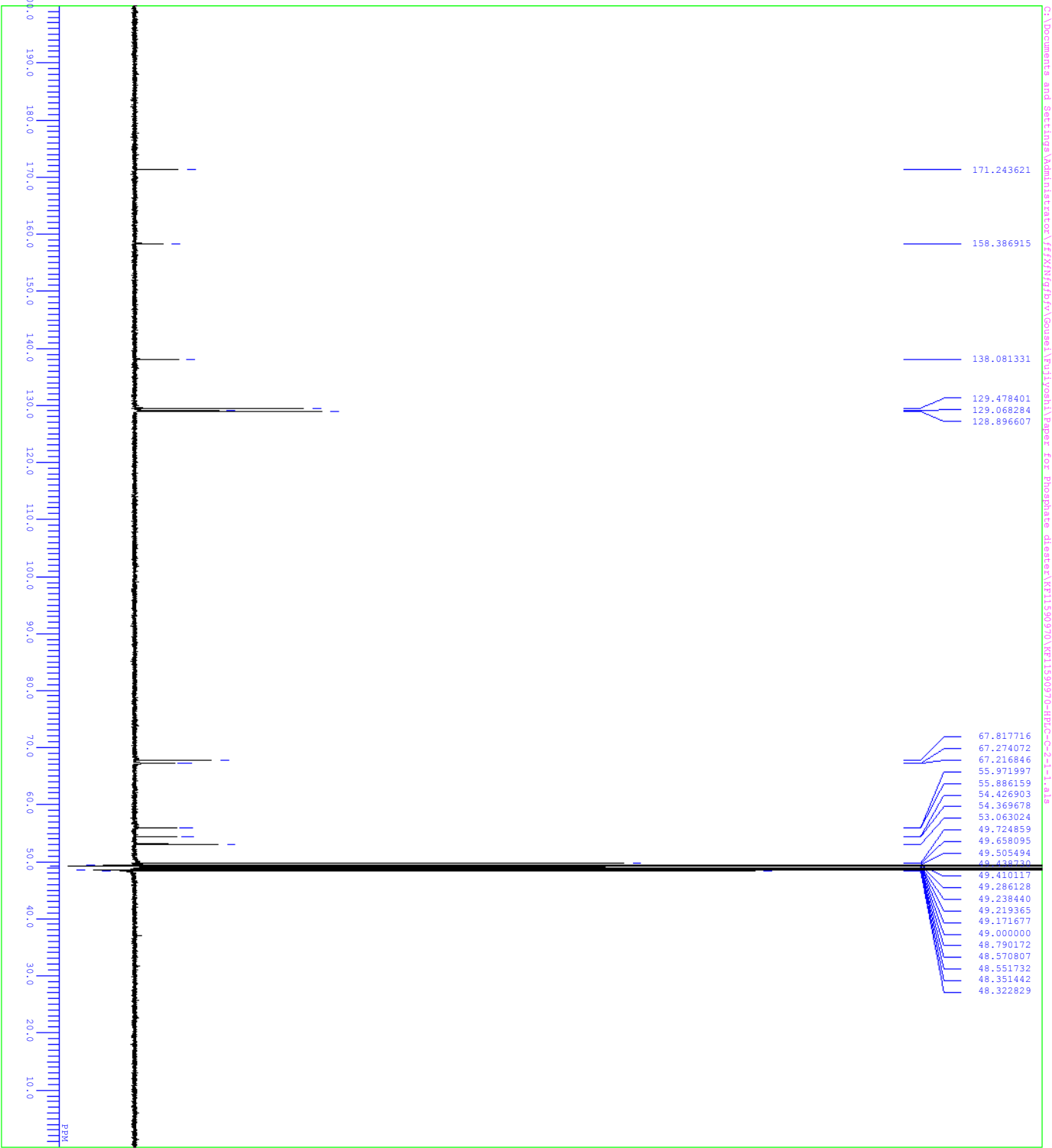


DFILE KF11631000-HPLC-cyclic-P-1-1.a1s
COMNT
DATIM 15-04-2021 13:11:57
OBNUC 31P
EXMOD carbon-13p
OBFREQ 158.59 MHz
OBSSET 7.99 KHz
OBFIN 9.23 Hz
POINT 26214
FREQU 64102.56 Hz
SCANS 101
ACQTM 0.4089 sec
PUL 2.0000 sec
PR 4.80 usec
IRNUC 1H
CTEMP 20.9 C
SIVNT CHCL3
EXREF 0.00 ppm
BF 0.12 Hz
RGAIN 56

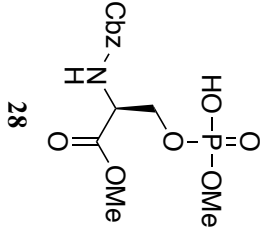


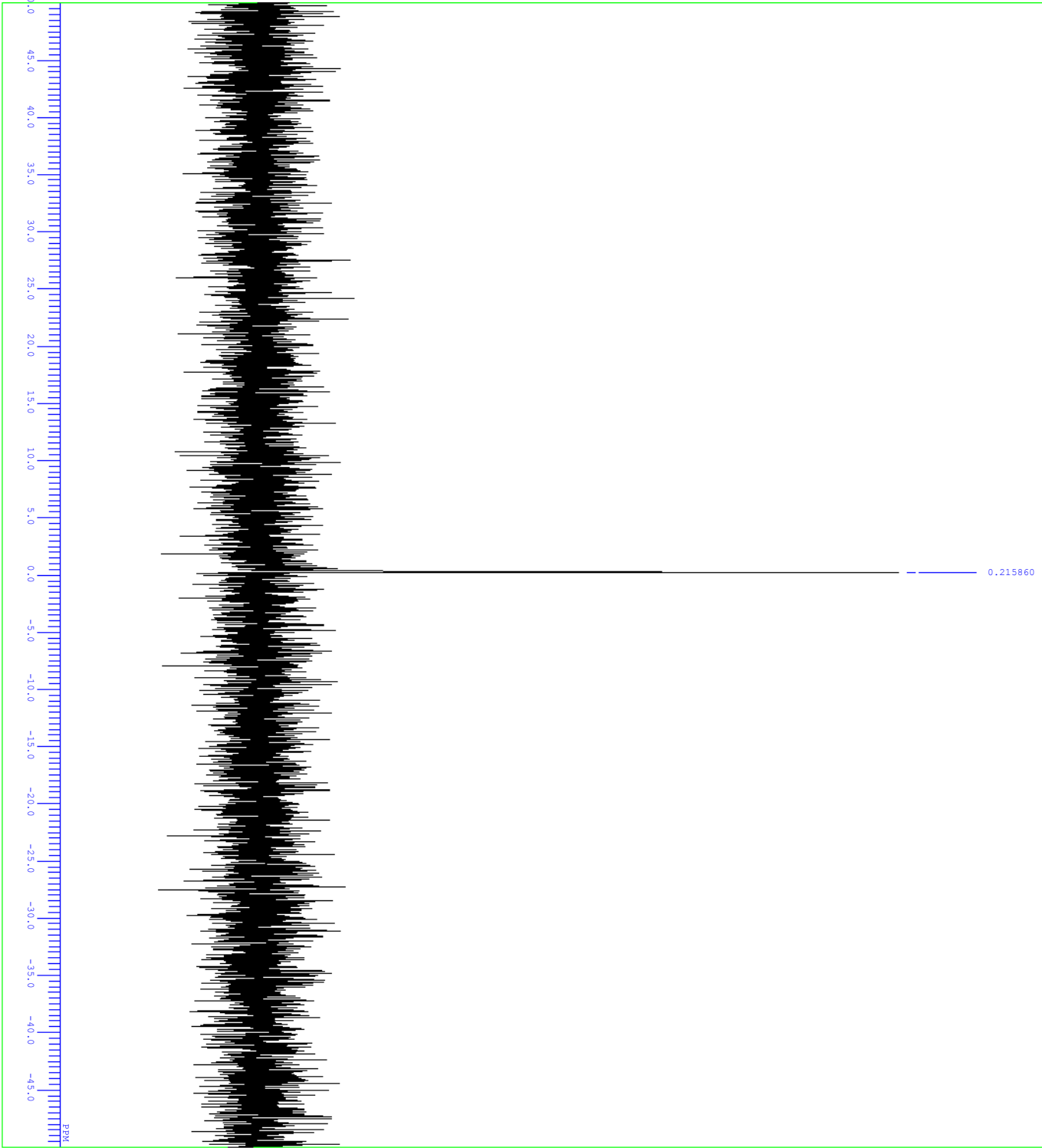


FF1150970-H-3-1-1.a1s
COMET
DATE 24-04-2021 19:34:48
NAME
EXPNO 1H
PROCNO 1
PROCPS 391.78 MHz
OBSFREQ 391.78 MHz
OBSSET 8.51 KHz
PULPROG zgpg30
PC 1.3107
F2 - 5878.10 Hz
SCANS 8
AQ 2.2295 sec
RG 6.0000 sec
PD 5.17 usec
IRNUC 1H
CTEMP 21.0 C
SOLVENT CD3OD
EXREF BF 3.30 ppm
RGAIN 0.12 Hz
50

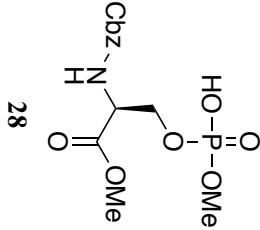


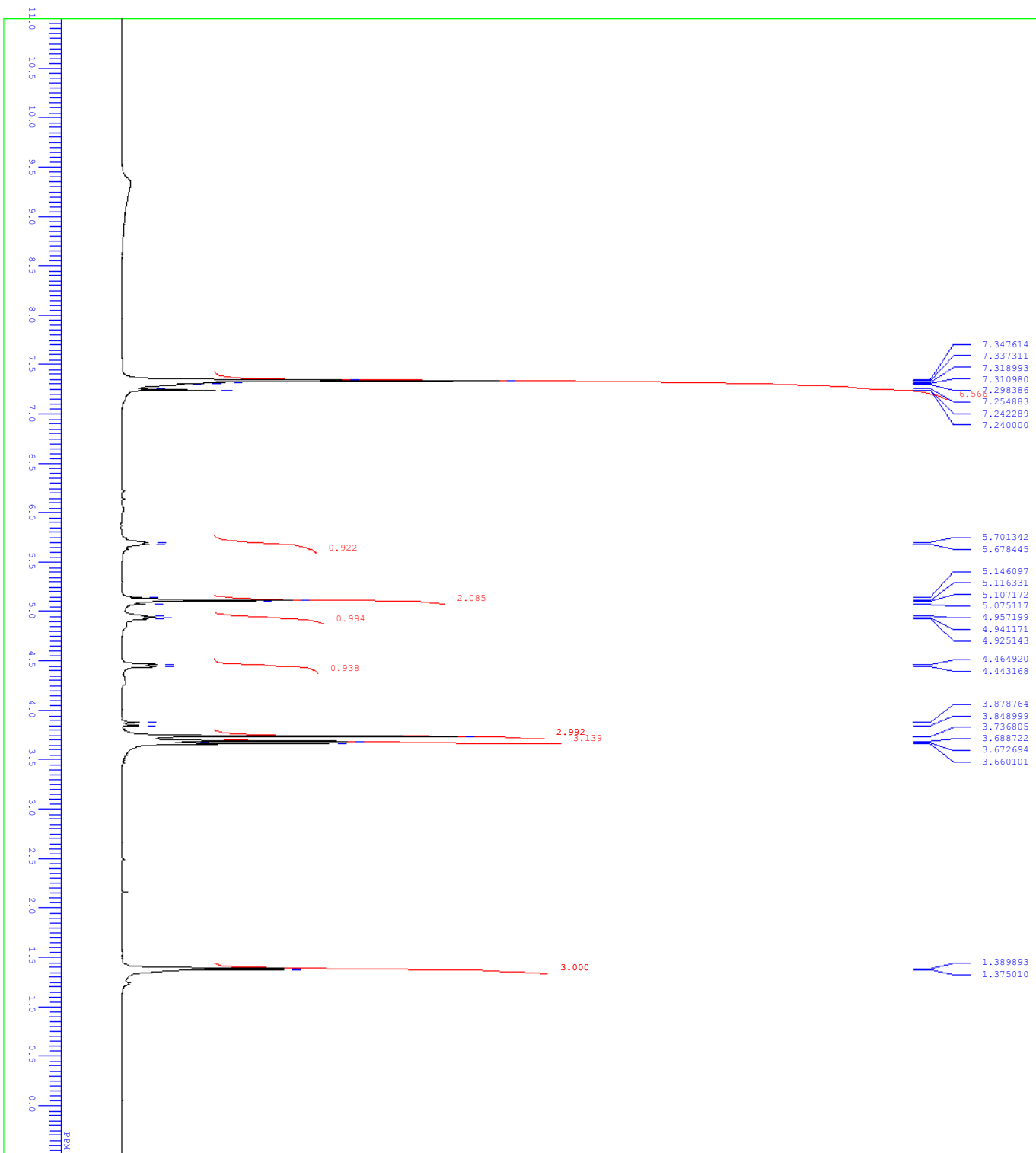
DFILE KF11590970-HPLC-C-2-1-1.a1s
 COMMENT 09-04-2021 23:43:29
 ORIGIN 13C
 EXPRO 98.52 MHz
 OBSRO carbon-13
 OBSSET 98.52 MHz
 OBSFIN 8.774 Hz
 POINT 2621.4 Hz
 FREQOU 24630.54 Hz
 SCANS 10000
 ACQTM 1.0643 sec
 PD 2.0000 sec
 PUL 3.12 usec
 IRNUC 1H
 CTEMP 20.7 c
 SLVNT CD3OD
 EXREF 49.00 ppm
 BF 0.12 Hz
 RGAIN 60



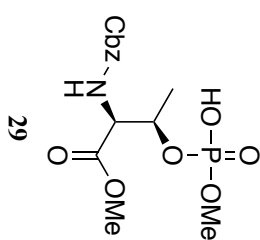


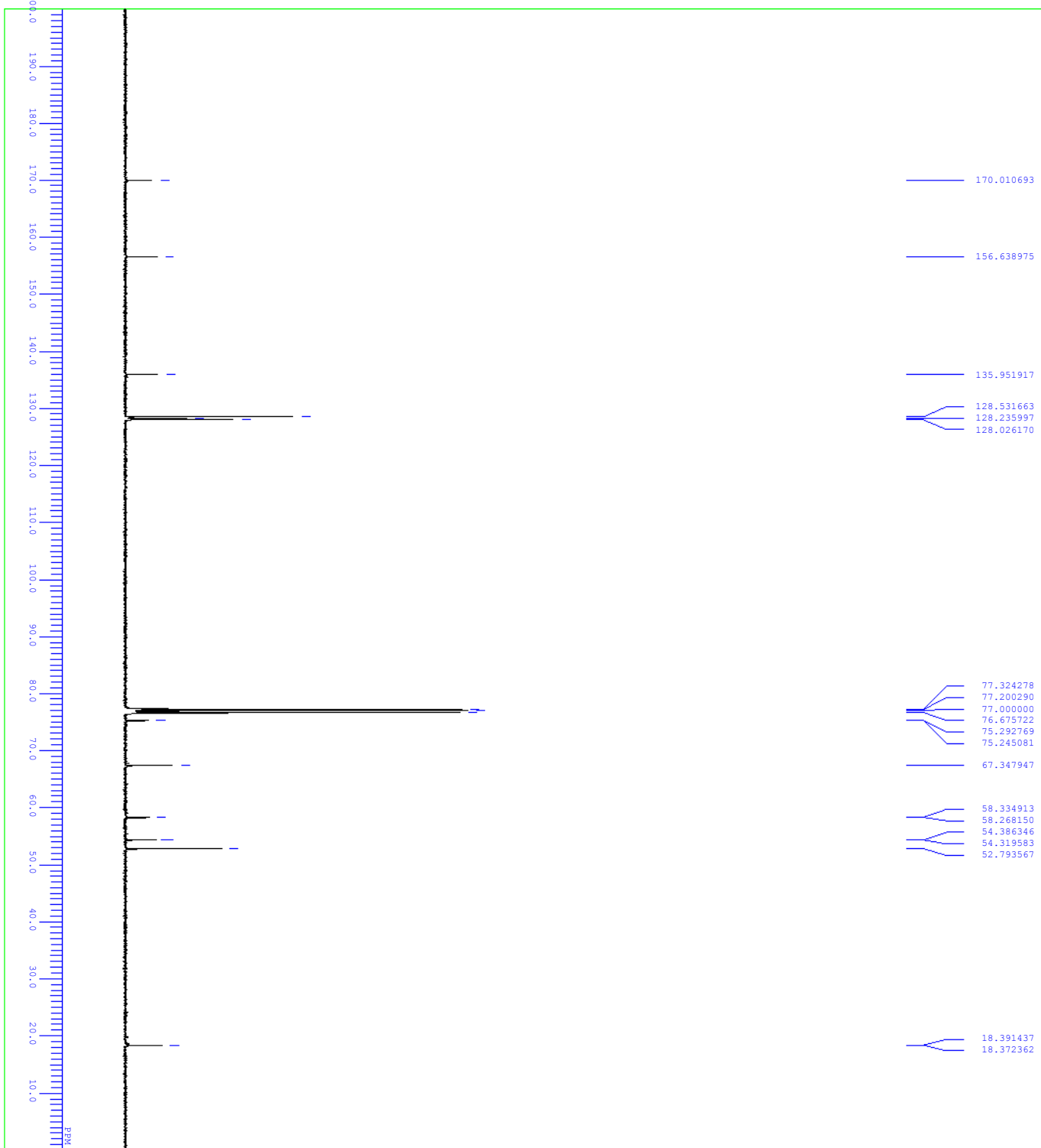
DTITLE FF115970-P-3-1-1.a1s
COMNT
DATEM 24-04-2021 19:38:01
EXNUC 31P
EXPROD carbon-³XP
OBSFQ 158.59 MHz
OBSRZ 177.99 kHz
OBSRZ 9.23 kHz
POINT 26214 Hz
FREQOU 64102.56 Hz
SCANS 128
ACQTM 0.4089 sec
PD 2.0000 sec
PWL 4.90 usec
IRNUC 1H
CTEMP 21.1 C
SLVNT CD3OD
EXREF 0.00 ppm
BF 0.12 Hz
RGAIN 56



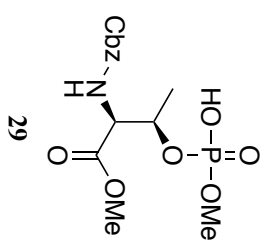


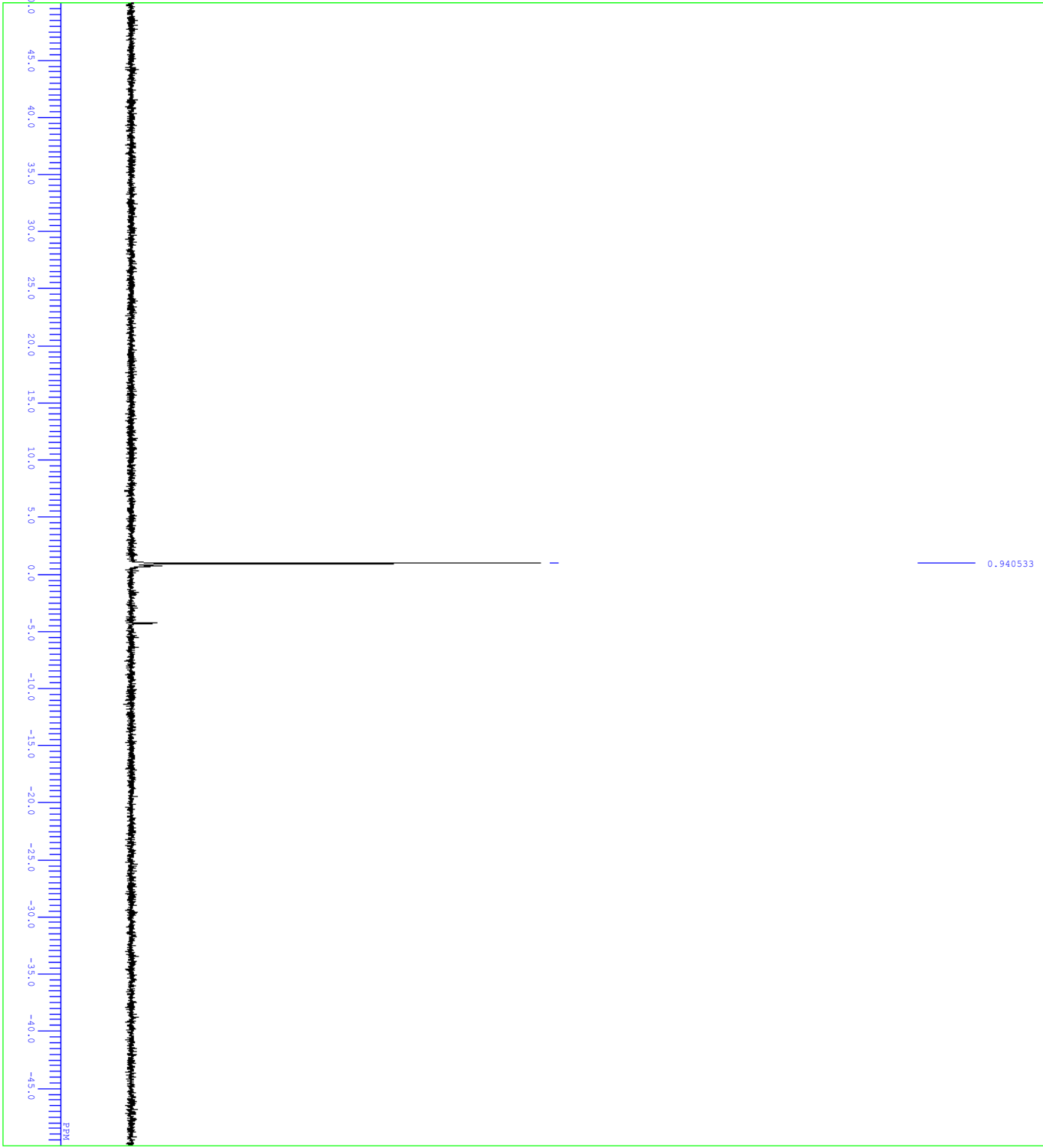
DTITLE FN15971-HPLC-H-3-1-1.a1s
 COUNT 05-04-2021 01:50:51
 ORNUC 1H
 EXMRQ proton-3K79 MHz
 OBSRQ 391.779 MHz
 OBRF 8.53 KHz
 OBRIN 3.24 KHz
 POINT 13107
 FREQU 5878.90 Hz
 SCANS 8
 ACQTM 2.2295 sec
 PD 6.0000 sec
 PUL 5.17 usec
 IRNUC 1H
 CTEMP 20.8 c
 SIVNT CDCL3
 EXREF 7.24 ppm
 BF 0.12 Hz
 RGAIN 36



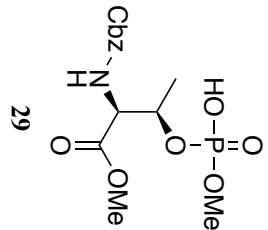


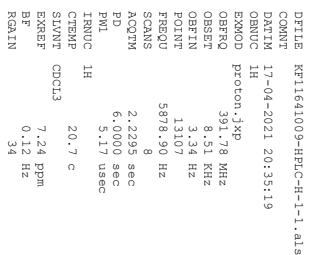
DTITLE KF11590971-HPLC-C-1-1.a1s
 COUNT 150
 DATEIN 09-04-2021 02:02:55
 ORNUC 13C
 EXMOD carbon-13
 OBSAQ 98.52 MHz
 OBSFQ 144.44 MHz
 OBSIN 8.774 Hz
 POINT 2621.4 Hz
 FREOU 24630.54 Hz
 SCANS 10000
 ACQTM 1.0643 sec
 PD 2.0000 sec
 PUL 3.12 usec
 IRNUC 1H
 CTEMP 20.9 C
 SIVNT CDCL3
 EXREF 77.00 ppm
 BF 0.12 Hz
 RGAIN 60

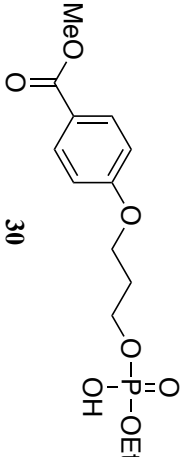
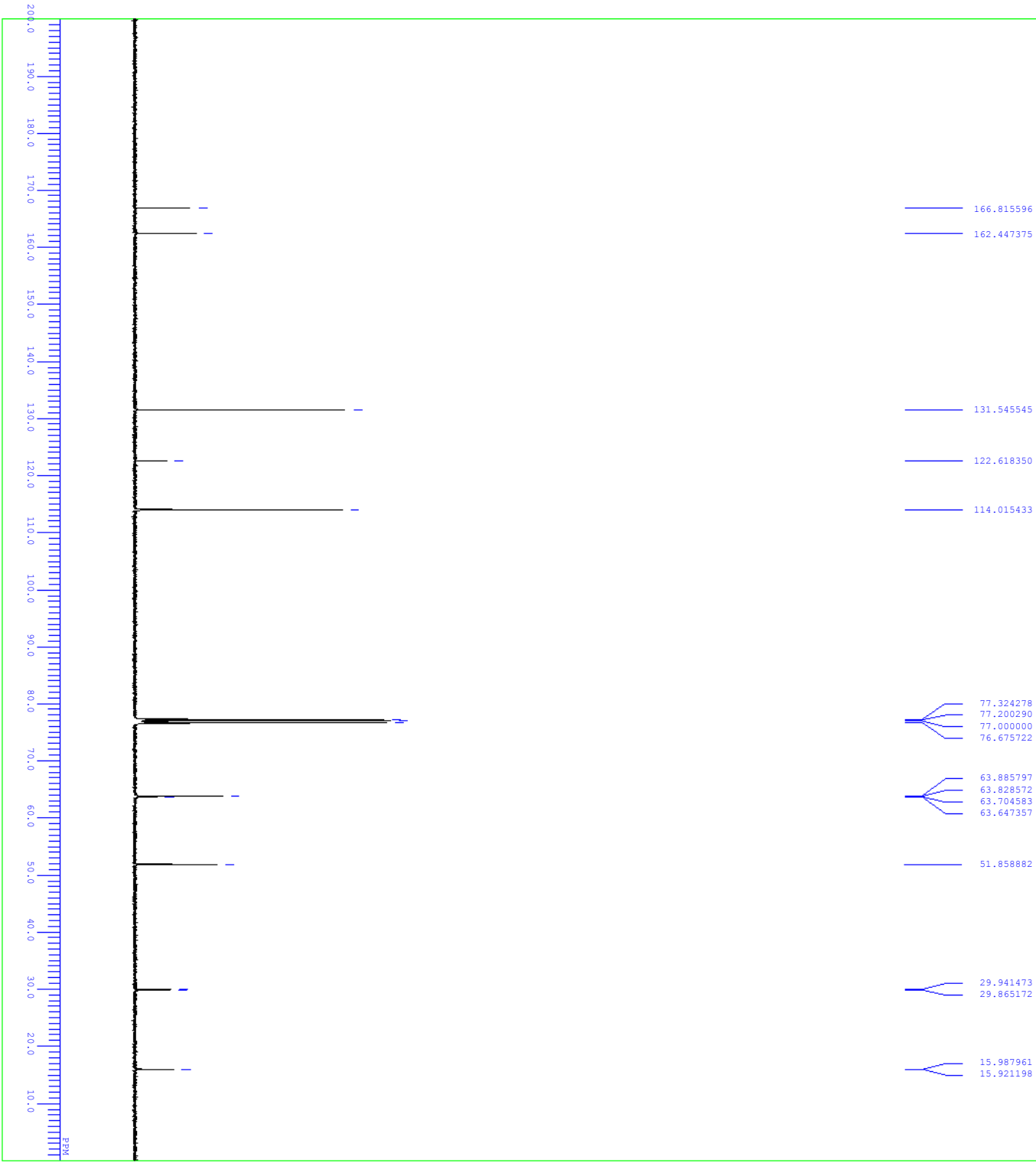




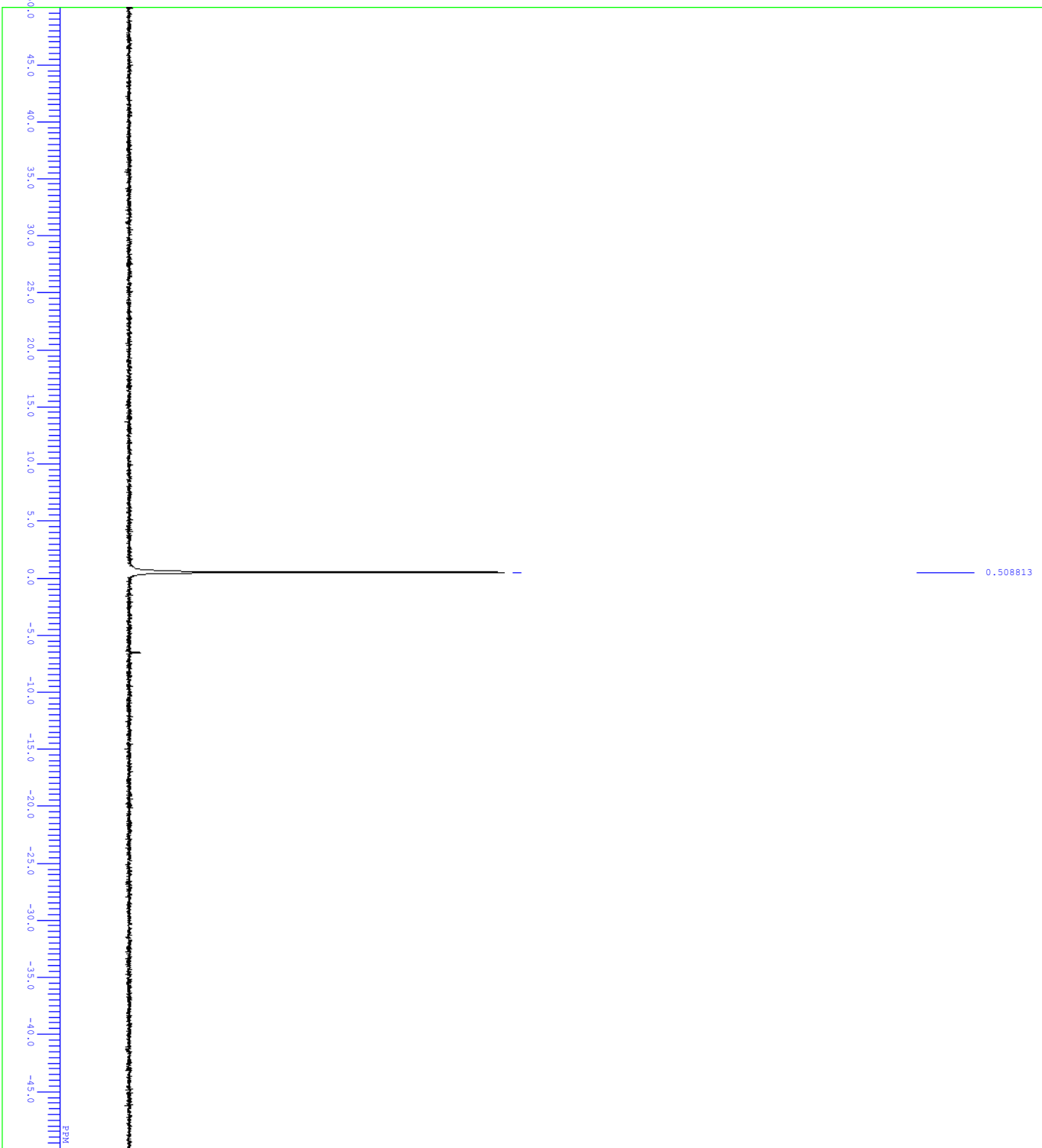
DPFILE KF11590971-HPLC-P-4-1-1.a1a
COMNT
DATIM 24-04-2021 15:38:08
ORNUC 31P
EXMOD carbon-13XP
OBSFREQ 158.59 MHz
OBSSET 7.99 KHz
OBSFID 9.23 Hz
OBSFID 2.62 Hz
FREQUS 64102.56 Hz
SCANS 41
ACQTM 0.4089 sec
PD 2.0000 sec
PWL 4.80 usec
IRNUC 1H
CTEMP 21.0 C
SLVNT CDCL3
EXREF 0.00 ppm
BF 0.12 Hz
RGAIN 56



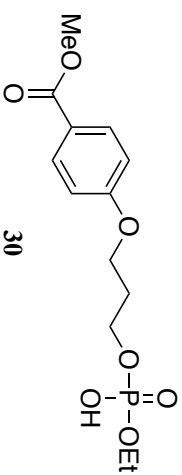


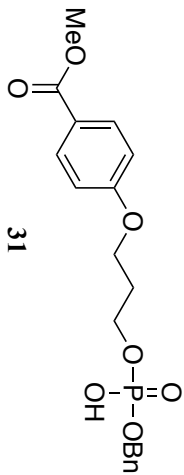
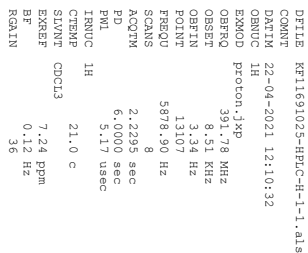


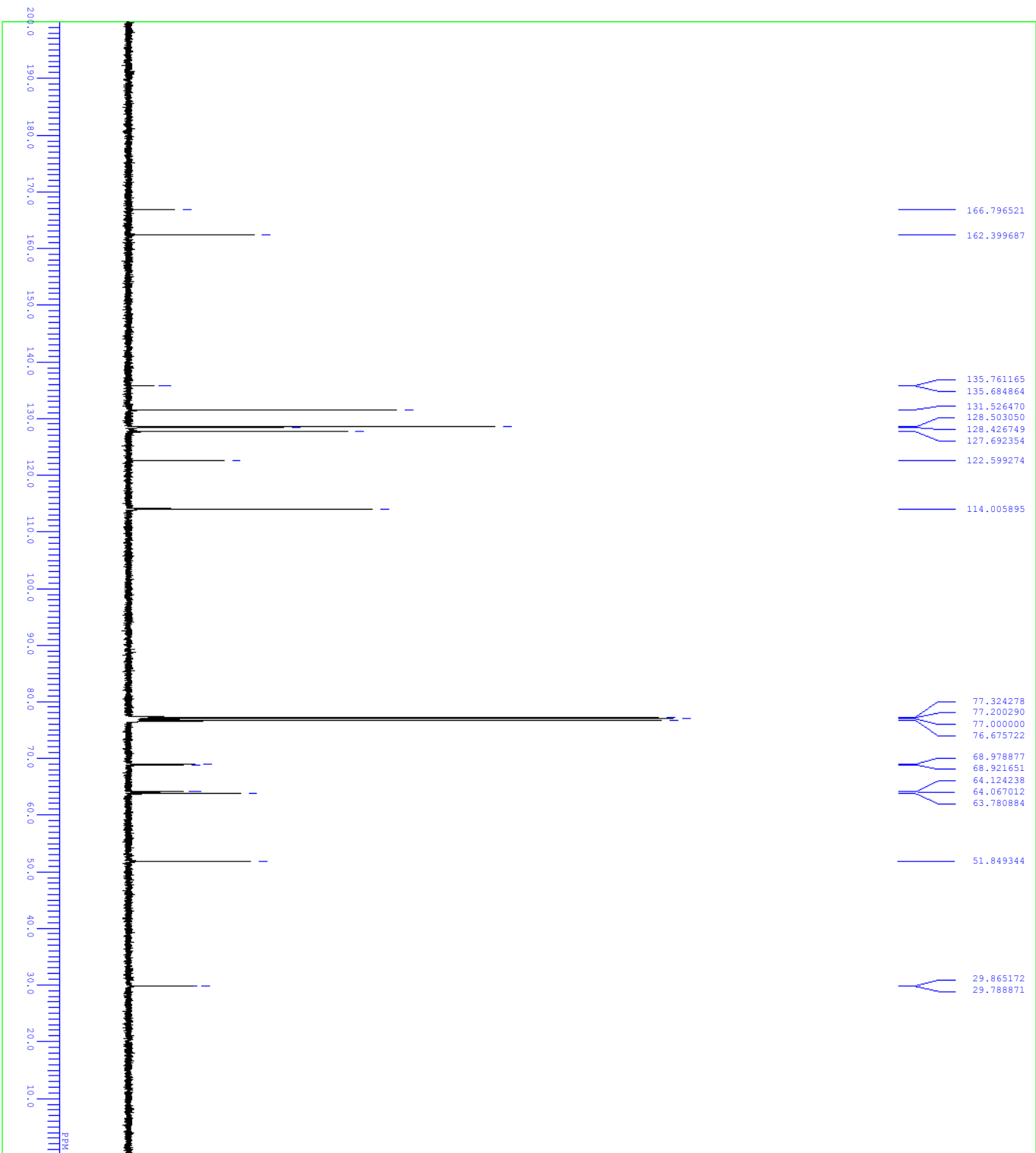
DFFILE FF1641009-HPLC-C-1-1-.als
COMENT
DATIM 17-04-2021 20:36:59
OBNUC 13C
EXMOD carbon-13p
OBFREQ 98.52 MHz
OBSSET 4.64 KHz
OBSIN 8.774 Hz
FOLINT 26214 Hz
FRESNO 24632453 Hz
SCANS 1.0643 sec
ACQTM 2.0000 sec
PD 3.12 usec
PWL
IRNUC 1H
CTEMP 20.8 C
SIVNT CDCL3
EXREF 77.00 ppm
BF 0.12 Hz
RGAIN 60



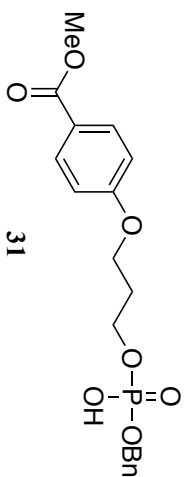
DFILE FF1641009-HPLC-P-1-1.als
 COMNT
 DATIM 17-04-2021 20:28:58
 OBNUC 31P
 EXMOD carbon-13P
 OBFREQ 158.59 MHz
 OBSSET 7.99 KHz
 OFFSET 9.23 Hz
 FOLINT 2621.4 Hz
 FREQD 64102.4 Hz
 SCANS 180
 SCANS 0.4089 sec
 PCQTM 2.0000 sec
 PD 4.80 usec
 PUL
 ITNUC 1H
 CTMP 20.8 C
 SLVNT CDCL3
 EXREF 0.00 ppm
 BF 0.12 Hz
 RGAIN 56

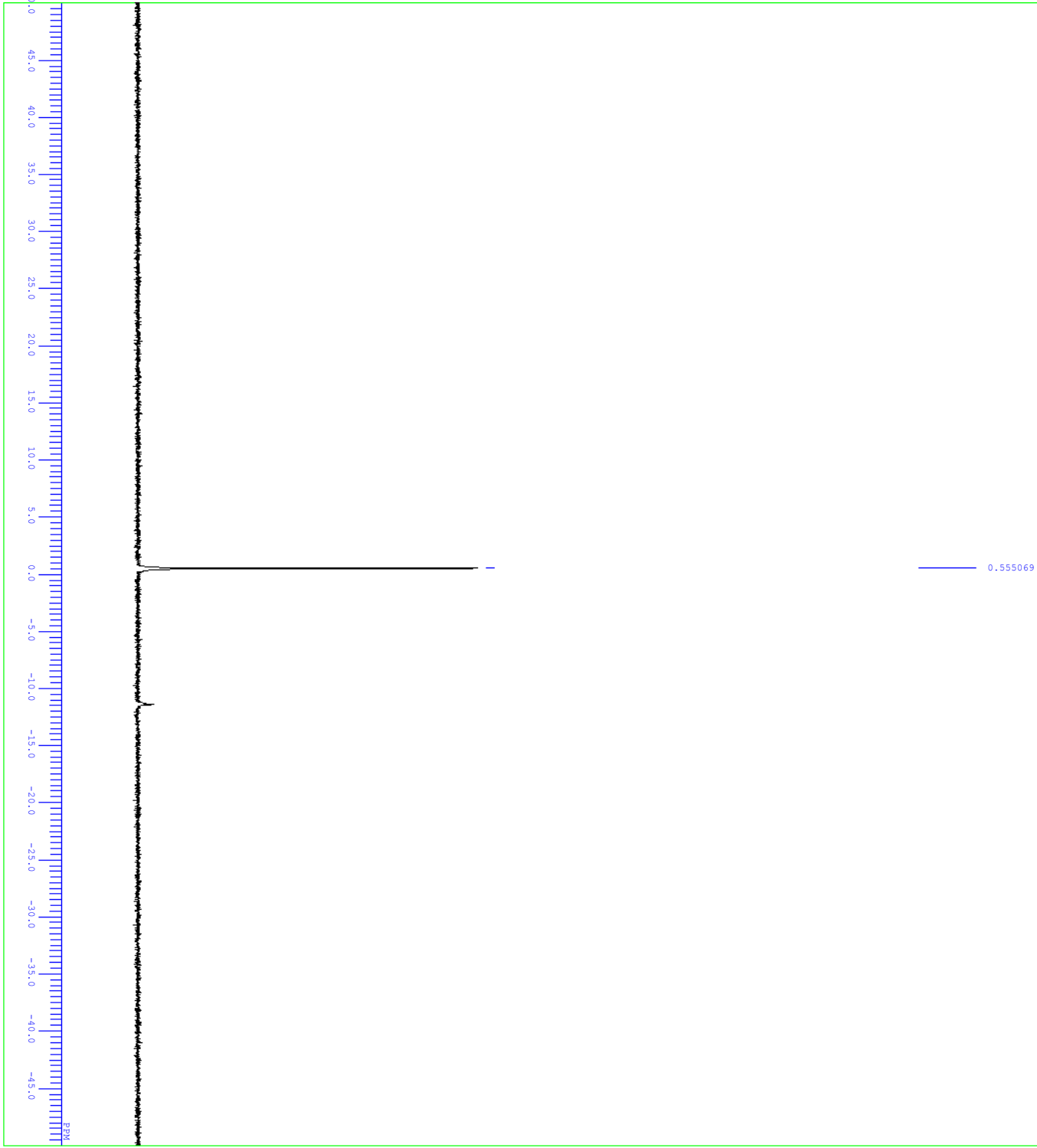






DFILE FF11691025-HPLC-O-1-1-.als
 COMMENT 22-04-2021 12:25:43
 DATE 13C
 EXMOD carbon-13p
 OBSERVE 98.52 MHz
 OFFSET 4.64 kHz
 OBSERVE 8.714 Hz
 F2 262.14 Hz
 F1 24630.54 Hz
 SCANS 11563
 AQC 1.0643 sec
 ACQ 2.0000 sec
 PD 3.112 usec
 IRNUC 1H
 CTMP 21.1 C
 SLVNT CDCl3
 EXREF 77.00 ppm
 BF 0.12 Hz
 RGAIN 60





DPF1E FT1691025-HFC-P-1-1.a1s
COMET
DATEIM 22-04-2021 12:16:03
OBNNUC 31P
EXMOD carbon-13xp
OBFREQ 158.159 MHz
OBSSET 7.99 KHz
OBFIN 9.23 Hz
POINT 26214
FREQU 64102.56 Hz
SCANS 76
ACQIM
PD 0.4089 sec
2.0000 sec
FID
FID 4.80 usec
IRNUC 1H
STEHP 21.2 c
SOLUT CDCl3
EXREF 0.00 ppm
BP 0.12 Hz
RGAIN 56

