

Analytic Gradients for Complete and Restricted Active Space Second-order Perturbation Theory within the Diagonal Approximation

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The computational cost of analytic derivatives in multireference perturbation theory is strongly affected by the size of the active space employed in the reference self-consistent field calculation. To overcome previous limits on active space size, the analytic gradients of single-state complete and restricted active space second-order perturbation theory within the diagonal approximation (CASPT2-D and RASPT2-D) have been developed and implemented in a local version of OpenMolcas. Similar to previous implementations of CASPT2, the RASPT2 implementation employs the Lagrangian or Z-vector method. The numerical results show that restricted active spaces with up to 20 electrons in 20 orbitals can now be employed for geometry optimizations.

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I. INTRODUCTION

Accurate and efficient quantum chemical approaches are indispensable for predicting electronic and geometrical properties. For this purpose, electron correlation must be considered to the greatest extent possible, ideally with a low computational cost. One useful and well-known approach to account for electron correlation is multiconfiguration self-consistent field (MCSCF) calculations followed by electron excitation, namely multireference (MR) or post-MCSCF treatments, to consider dynamic and static (nondynamic) electron correlation in a balanced way. MR methods include MR coupled-cluster¹⁻³ and configuration interaction⁴ approaches. However, MR perturbation theory (MRPT) may be the most balanced approach in terms of computational cost and accuracy. The most well-known MRPT is probably the complete active space second-order perturbation theory (CASPT2).⁵⁻⁷ Other MRPTs such as (extended)⁸ multiconfiguration quasi-degenerate second-order perturbation theory [(X)MCQDPT2],⁹ n -electron valence state second-order perturbation theory (NEVPT2),¹⁰⁻¹² generalized van Vleck second-order perturbation theory (GVVPT2),¹³ and retaining the excitation degree perturbation theory (REPT)¹⁴ have also been employed for various tasks.

Recent efforts to develop analytic derivatives of MRPTs¹⁵ have made it possible to efficiently compute properties. Earlier studies reported such implementations in MOLPRO,¹⁶ BAGEL,¹⁷ GAMESS-US,¹⁸ and TeraChem¹⁹ for different MRPT methods. However, the size of the active space in these past works was rather limited. To the best of our knowledge, the largest active space in terms of the number of determinants [or configuration state functions (CSFs)] applied in analytic derivatives corresponds to 12 electrons in 12 orbitals (12e,12o)²⁰ or (12e,11o).^{21,22} In the former calculation, the number of determinants was 853,776 in GAMESS-US.

This severe limitation on active space size is due to the full configuration interaction (CI), or complete active space (CAS), treatment in the active space, and the limitation is even more severe for subsequent post-CASSCF approaches. The formal computational cost of CAS treatment grows as a factorial, and higher-order reduced density matrices are needed in post-CASSCF calculations if the internal contraction scheme is applied. The situation is worse for analytic derivatives of MRPTs because one has to contract terms that formally scale as $N_{\text{CSF}}N_{\text{act}}^6$ to $N_{\text{CSF}}N_{\text{act}}^8$ (a power of eight for NEVPT2), where N_{CSF} and N_{act} are the number of CSFs (or determinants) and the number of active orbitals, respectively. Thus, one must carefully develop a computationally and memory efficient algorithm²¹ to allow larger active spaces.

One solution to this factorial growth is to employ restricted active space (RAS)²³ SCF references followed by perturbation theory (i.e., RASPT2).^{24,25} In RASSCF, the active space is first partitioned into three subspaces: RAS1, RAS2, and RAS3. The RAS2 space is treated as the CAS space, and full CI is performed. In RAS1 and RAS3, the numbers of holes and electrons, respectively, are limited by setting maximum values. As a result, the number of CSFs or determinants for the RAS is much smaller than that for the corresponding CAS. Using this approach, an initial RASPT2 study was conducted with an active space as large as (28e,32o).²⁵ Similar approximate CAS treatments for MRPTs have been developed for MCQDPT, including the quasi-CAS QDPT,²⁶ general MCQDPT,²⁷ and occupation-restricted multiple active space (ORMAS) PT²⁸ methods. One important difference between these QDPT-based methods and RASPT2 is the inclusion of the fully internal excitations (perturbative two-electron excitations within the active space): to date, developed RASPT2 methods do not consider these excitations, primarily because of the complications resulting from the internal contraction. Another famous CAS-based approach combines the density-matrix renormalization group (DMRG) with MRPT (DMRG-CASPT2).²⁹

This paper describes the development of analytic gradients for fully internally contracted single-state CASPT2-D and RASPT2-D. Here, the suffix “-D” indicates the diagonal approximation, which neglects the off-diagonal matrix blocks of the zeroth-order Hamiltonian.⁶ Different from the “full” CASPT2 and RASPT2 methods,⁷ this approximation allows the evaluation of energy without an iterative procedure. Thus, the approximate CASPT2-D and RASPT2-D methods are computationally advantageous and allow many terms to be omitted, facilitating their implementation without greatly affecting the results, as demonstrated in Ref. 7. The developed method is applied to typical π -rich molecules, and calculations are performed with active spaces containing up to 20 electrons in 20 orbitals.

II. METHODS

CASSCF and CASPT2 are the special cases of RASSCF and RASPT2, respectively. Thus, here, we focus on RASSCF and RASPT2 as general cases. In this section, p , q , r , and s refer to general orbitals, i and j refer to inactive (doubly occupied) orbitals, t and u refer to active orbitals, a and b refer to secondary (virtual) orbitals, and ϕ and χ refer to internally contracted bases.

This section does not provide the full details of algorithm development; rather, it provides an outline of the algorithm and highlights a few differences between the analytic derivatives of

CASPT2 and RASPT2 because many equations are equivalent to those presented in earlier studies for CASPT2. Readers interested in the details should refer to, for instance, Refs. 16, 17, 19, and 30.

A. RASPT2 Energy

In RASPT2, the zeroth-order Hamiltonian is defined by

$$\hat{H}^{(0)} = \hat{P}\hat{F}\hat{P} + \hat{Q}\hat{F}\hat{Q}, \quad (1)$$

where $\hat{P} = |0\rangle\langle 0|$ ($|0\rangle$ is a reference state obtained by SCF) is the projector on the reference space, and \hat{Q} is the complementary projector. The Fock operator is defined by

$$\hat{F} = \sum_{pq} f_{pq} \hat{E}_{pq}, \quad (2)$$

where \hat{E} is the one-electron spin-averaged excitation operator, and f_{pq} is the Fock matrix:

$$f_{pq} = h_{pq} + \sum_{rs} \left((pq|rs) - \frac{1}{2}(pr|qs) \right) D_{rs}^{\text{SA}} \quad (3)$$

with the state-averaged one-electron density matrix D_{rs}^{SA} . Analyzing the structure of the Fock operator [Eq. (2)] shows that it consists of 25 blocks generated by the combinations of inactive, active (RAS1, RAS2, and RAS3), and external orbitals for RASPT2 (or nine for CASPT2). In the diagonal approximation, the off-diagonal elements are formally put into the perturbation operator, and the Fock operator is simply defined by

$$\hat{F} \rightarrow \hat{F}^{\text{D}} = \sum_p f_{pp} \hat{E}_{pp} \quad (4)$$

with pseudo-canonical orbitals (separately diagonalized in the inactive, RAS1, RAS2, RAS3, and secondary orbital spaces). Historically, CASPT2 with this approximation was reported in 1990,⁶ and the “full” CASPT2 with the off-diagonal blocks was reported in 1992.⁷ The RASPT2-D energy is usually obtained by direct summation (similar to partially contracted NEVPT2¹⁰), while the “full” RASPT2 energy is usually obtained by solving a large linear equation.

The second-order perturbation energy E^{PT2} is then obtained as a minimum of the Hylleraas functional

$$E^{\text{PT2}} \equiv E_2 = 2\langle \Psi^{(1)} | \hat{H} | \Psi^{(0)} \rangle + \langle \Psi^{(1)} | \hat{H}^{(0)} - E^{(0)} | \Psi^{(1)} \rangle. \quad (5)$$

For simplicity, the level shift, either real³¹ or imaginary,³² is not included. The zeroth-order wavefunction $|\Psi^{(0)}\rangle$ is obtained at the SCF level, and the first-order wavefunction $|\Psi^{(1)}\rangle$ is generally defined by operating two-electron excitations:²⁵

$$|\Psi^{(1)}\rangle = \sum_{pqrs} T_{pqrs} \hat{E}_{pqrs} |\Psi^{(0)}\rangle. \quad (6)$$

In Eq. 6, T_{pqrs} is the amplitude of the excitation, which is obtained by solving the amplitude equation

$$\frac{1}{2} \frac{\partial E_2}{\partial T_{pqrs}} = \langle \Phi_{pqrs} | \hat{H} | \Psi^{(0)} \rangle + \langle \Phi_{pqrs} | \hat{H}^{(0)} - E^{(0)} | \Psi^{(1)} \rangle = 0, \quad (7)$$

where $|\Phi_{pqrs}\rangle$ is the doubly excited configuration. This equation is solved either directly or iteratively in RASPT2-D or RASPT2, respectively. The RASPT2 energy is finally obtained as a sum of the RASSCF and PT2 energies: $E^{\text{RASPT2}} = E^{\text{RASSCF}} + E^{\text{PT2}}$. Strictly speaking, above equations must be formulated with indices of the internally contracted basis and distinguish singlet and triplet excitations, making this method more complicated. For further discussion, readers should refer to, for instance, Refs. 6, 7, and 33.

B. First-order Derivatives

Since the RASPT2 energies are not variational with respect to changes in wavefunction parameters, regardless of the diagonal approximation, analytic derivatives of the energies are needed to evaluation of the response (derivative) of the wavefunction parameters. As in earlier studies, the first-order derivatives of the RASPT2-D energies in this study were evaluated using the Lagrangian approach.³⁴ At the first order, the equations derived using this approach are essentially equivalent to those derived by the Z-vector method.³⁵

First, we define the Lagrangian, which can be written as a sum of the RASSCF and the PT2 Lagrangians:

$$\mathcal{L}^{\text{RASPT2}} = \mathcal{L}^{\text{RASSCF}} + \mathcal{L}^{\text{PT2}}. \quad (8)$$

The RASSCF Lagrangian $\mathcal{L}^{\text{RASSCF}}$ is defined as the sum of the RASSCF energy (E^{RASSCF}) and the constraint conditions imposed when solving the RASSCF equation:

$$\begin{aligned} \mathcal{L}^{\text{RASSCF}} = & E^{\text{RASSCF}} + \frac{1}{2} \text{Tr} \left[\mathbf{Z} \left(\mathbf{A} - \mathbf{A}^\dagger \right) \right] - \frac{1}{2} \text{Tr} [\mathbf{X} (\mathbf{S} - \mathbf{I})] \\ & + \sum_N \omega_N \left[\sum_I z_{I,N} \langle I | \hat{H} - E_N^{\text{ref}} | N \rangle - \frac{1}{2} x_N (\langle N | N \rangle - 1) \right], \end{aligned} \quad (9)$$

where the second term is the generalized Brillouin condition with the orbital gradient \mathbf{A} , the third term is the requirement for the orthonormalization of molecular orbitals (MOs) with the overlap matrix \mathbf{S} in the MO basis, and the fourth term is the CI condition. In the fourth term, ω_N is the weight in state averaging, E_N^{ref} is the RASSCF energy relevant to CI for state N , and I is the CSF index. The Lagrangian multipliers (\mathbf{Z} , \mathbf{X} , \mathbf{z} , and \mathbf{x}) are determined by solving the following simultaneous equation, which is usually referred to as the Z-vector:³⁵

$$\begin{cases} \frac{\partial \mathcal{L}^{\text{RASSCF}}}{\partial \kappa_{pq}} = 0 \\ \frac{\partial \mathcal{L}^{\text{RASSCF}}}{\partial c_{I,N}} = 0 \end{cases}, \quad (10)$$

where κ_{pq} is the orbital rotation parameter, and $c_{I,N}$ is the CI coefficient. Details of the Z-vector or coupled-perturbed MCSCF equation can be found, for instance, in Refs. 36 and 37. Once the Lagrangian multipliers are determined, one can compute the gradient of the energy as the partial derivative of the Lagrangian:

$$\frac{dE^{\text{RASSCF}}}{d\alpha} = \frac{\partial \mathcal{L}^{\text{RASSCF}}}{\partial \alpha}, \quad (11)$$

where α is the derivative parameter (nuclear coordinates). If the single state is employed, the first-order derivatives can be computed without solving the response equation.

The RASPT2 Lagrangian can be written as it follows:

$$\begin{aligned} \mathcal{L}^{\text{PT2}} = & \sum_s \mathcal{L}_s^{\text{PT2}} \\ & + \sum_{pqrs} \lambda_{pqrs} \left(\langle \Phi_{pqrs} | \hat{H} | \Psi^{(0)} \rangle + \langle \Phi_{pqrs} | \hat{H}^{(0)} - E^{(0)} | \Psi^{(1)} \rangle \right) \\ & + \sum_i^{\text{core inactive}} \sum_j Z_{ij}^c f_{ij}. \end{aligned} \quad (12)$$

The second term in Eq. (12) is the variational condition of the amplitude [Eq. (7)] with multiplier λ_{pqrs} , and the third term is required for the frozen core approximation. $\mathcal{L}_s^{\text{PT2}}$ is the Lagrangian for the excitation class s , which corresponds to Eqs. (1a), (1b), ..., and (1h) in Ref. 6 and is also defined by

$$\mathcal{L}_s^{\text{PT2}} = E_s^{\text{PT2}} - \sum_{\phi\chi} \xi_{\phi\chi} (\Lambda_{\phi\chi}^S - \mathbf{I}), \quad (13)$$

where E_s^{PT2} is the perturbation energy from the excitation class s ($E^{\text{PT2}} = \sum_s E_s^{\text{PT2}}$), and $\Lambda_{\phi\chi}^S$ corresponds to Eq. (16) in Ref. 7 and is due to the orthonormalized overlap matrix in the internally contracted basis. An additional Lagrangian multiplier $\xi_{\phi\chi}$ is introduced, and it can also be computed without iteration.

Since the Z-vector equation has already been implemented in OpenMolcas both without³⁷ and with³⁸ the density-fitting approximation, the main task of this work is to construct the RASPT2-D part of the “source term”³⁹ of the Z-vector equation:

$$Y_{pq} := \frac{\partial \mathcal{L}^{\text{PT2}}}{\partial \kappa_{pq}} \quad (14)$$

$$y_{I,N} := \frac{\partial \mathcal{L}^{\text{PT2}}}{\partial c_{I,N}}, \quad (15)$$

which are roughly equivalent to the “orbital Lagrangian” and the “configuration Lagrangian,” respectively, in the author’s previous studies.^{18,20} Once all the Lagrangian multipliers are determined by solving the Z-vector equation, as in the case of the state-averaged RASSCF [Eq. (10) but where $\mathcal{L}^{\text{RASSCF}}$ is replaced with $\mathcal{L}^{\text{RASPT2}}$], the derivative of the RASPT2 energy can be evaluated by

$$\frac{dE^{\text{RASPT2}}}{d\alpha} = \frac{\partial \mathcal{L}^{\text{RASPT2}}}{\partial \alpha}. \quad (16)$$

Z_{ij}^c can be easily computed as⁴⁰

$$Z_{ij}^c = -\frac{1}{2} \frac{Y_{ij} - Y_{ji}}{f_{ii} - f_{jj}}. \quad (17)$$

As stated earlier, most of the above equations are equivalent to those derived in previous developments of CASPT2.^{16,17,19,30} One major difference between the CAS and RAS references is the definition of the independent orbital rotations. Independent orbital rotations change the (electronic) energy. When the CAS reference is employed, the independent orbital rotation consists of three subspaces: inactive–active, inactive–external, and active–external blocks. In contrast, for the RAS reference, we need to consider three additional blocks: RAS1–RAS2, RAS1–RAS3, and RAS2–RAS3 blocks. The orbital rotation parameters in these additional blocks are optimized by solving the Z-vector equation. The difference is schematically explained in Fig. 1. The area with diagonal lines in white squares represents the independent orbital rotations for CASPT2(-D). For RASPT2(-D), three additional blocks (gray areas) are considered. The other blank areas are either dependent or redundant (diagonal or off-diagonal blocks, respectively) orbital rotations. The density in the dependent part is constructed with the non-canonical approach^{41,42} because RASPT2(-D) is an invariant theory with respect to rotations within each orbital block. An alternative approach for the active part (iterative) is outlined in Ref. 19.

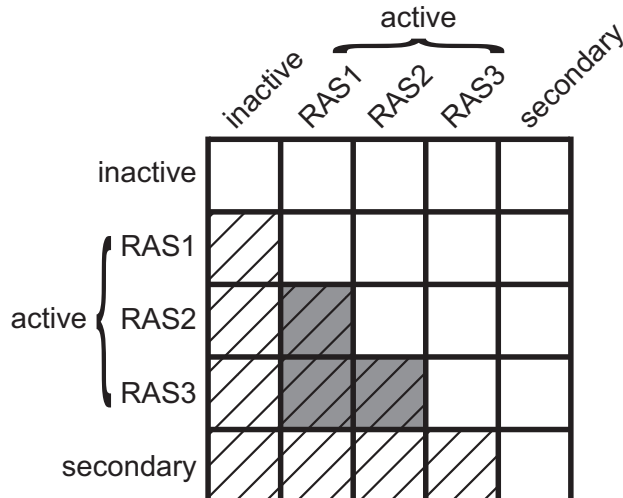


FIG. 1. Independent orbital rotations for CASPT2(-D) and RASPT2(-D).

C. Implementations

The above equations, particularly Eqs. (14) and (15), were implemented in a local version of OpenMolcas.^{43,44} Although these equations are formulated based on the Lagrangian method, the actual implementation closely follows that of the equations derived by direct differentiation, as in Refs. 18 and 20. The present implementation can employ the frozen core approximation (i.e., the chemically inert orbitals are neglected) and consider the real and imaginary level shift. Electron-repulsion (two-electron) integrals can be evaluated conventionally or with the density-fitting approximation³⁸ but not with the Cholesky decomposition. The diagonal preconditioning for the active-active rotations in the Z-vector has not been implemented; thus, it was implemented according to Eq. (C.12e) in Ref. 45.

If several states are averaged in the reference SCF calculation, the Fock operator may be defined using either the state-averaged or (unrelaxed) state-specific density matrices and molecular orbitals by canonicalizing the Fock matrix. By default, OpenMolcas employs the state-specific density matrix (except for XMS-CASPT2); however, the present implementation for analytic gradients is limited to the state-averaged density matrix [Eq. (3)]. The effect of this difference, including the XMS extension using the state-specific density matrix, has been well investigated by Park.⁴⁶

III. COMPUTATIONAL DETAILS

All calculations were performed with a locally modified version of OpenMolcas.^{43,44} The frozen core approximation was always employed in the perturbation calculation. The cc-pVDZ basis set^{47,48} and the density-fitting approximation with the corresponding RI fitting auxiliary basis set taken from the EMSL basis set exchange^{49–51} were employed unless otherwise noted. The ionization potential–electron affinity shift⁵² was set to zero. No symmetry constraints were applied. A single node of a six-core E5-2643 v4 3.40-GHz processor with 48 GB of random-access memory was used to perform all calculations and measure all computational times.

It is useful to define the RAS. In this study, the nomenclature in Ref. 25 is employed: $(ie,jo)/(ke,lo)/m$, where i and j are the numbers of electrons and orbitals in the entire active space (RAS1+RAS2+RAS3), respectively; k and l are those in the RAS2 space, respectively; and m is the maximum number of electrons excited from RAS1 or into RAS3. Different from the nomenclature in Ref. 25, the definition of the RAS2 space is explicitly written, even if k and l are zero.

First, the accuracy of the implemented gradient is discussed using *trans*-1,3-butadiene optimized at the Hartree–Fock/cc-pVDZ level of theory. The two-point stencil method with a displacement parameter of 10^{-3} was employed to evaluate the numerical gradients. The active space consisted of four electrons in four π orbitals (4e,4o). The lowest two states were averaged in all calculations.

Second, CASPT2-D and RASPT2-D with various active spaces were applied to *trans*-1,3,5,7,9,11-dodecahexaene (C₁₂H₁₄). The active space consisted of 12 electrons in 12 π orbitals (12e,12o) in all calculations. The lowest two states were averaged with an imaginary shift of 0.2i.

Finally, the developed methodology, RASPT2-D, was applied to dibenzopentalene derivatives **2a** and **4a** in Ref. 53. As these two molecules have 16 and 20 electrons in the corresponding π orbitals, respectively, only RASPT2-D calculations were performed, although it would be possible to employ small CASs. RAS(16e,16o)/(4e,4o)/2 and RAS(20e,20o)/(4e,4o)/2 were employed for **2a** and **4a**, respectively. The lowest three states were averaged with an imaginary shift of 0.2i.

The optimized coordinates of the *trans*-1,3,5,7,9,11-dodecahexaene and dibenzopentalene derivatives are provided in the supplementary material.

TABLE I. Maximum Differences (MAXDs) and Root-Mean-Square^a Differences (RMSDs) Between Analytic and Numerical Gradients in a.u./bohr for the Ground State of *trans*-1,3-Butadiene at the CASPT2-D or RASPT2-D Level of Theory.

Active space	MAXD	RMSD
CAS(4e,4o) ^b	4.10×10^{-6}	2.22×10^{-6}
CAS(4e,4o)	5.17×10^{-7}	3.14×10^{-7}
RAS(4e,4o)/(0e,0o)/2	4.42×10^{-6}	2.65×10^{-6}
RAS(4e,4o)/(0e,0o)/2 ^c	4.39×10^{-6}	2.63×10^{-6}
RAS(4e,4o)/(2e,2o)/1	2.87×10^{-6}	1.65×10^{-6}
RAS(4e,4o)/(2e,2o)/1 ^c	2.85×10^{-6}	1.65×10^{-6}

^a Forces perpendicular to the planar axis are not included in RMSD

^b Without the density-fitting approximation

^c With an imaginary shift of 0.2i

IV. RESULTS AND DISCUSSION

A. Accuracy of the Implemented Gradient

First, the accuracies of the implementations with CASPT2-D and RASPT2-D gradients are briefly presented by comparing the analytic and numerical (with a two-point stencil) gradients. Table I shows that the difference between the gradients is at most 1.0×10^{-5} a.u./bohr, indicating the implemented gradients are sufficiently accurate. Here, three active spaces were employed: CAS(4e,4o), RAS(4e,4o)/(0e,0o)/2, and RAS(4e,4o)/(2e,2o)/1. The differences with the different active spaces are sufficiently small, indicating that any active space may be employed so long as the reference RASSCF calculation converges. The use of the imaginary shift technique does not degrade the accuracy. Although analytic gradients with the real level shift can be evaluated with the present implementation, they were not employed in this study. The accuracy for S_1 is similar to that for S_0 .

Note that it is not possible to perform gradient calculations analytically using, for instance, RAS(4e,4o)/(2e,2o)/2 and RAS(4e,4o)/(0e,0o)/4, which span the same variational space as CAS. In such a case, some of the orbital rotation parameters are linearly dependent on the configuration parameters; thus, the Z-vector equation does not converge.

TABLE II. Deviations in Vertical Excitation Energies at the S_0 minimum (E^{vert} in eV), Adiabatic Excitation Energies (E^{AEE} in eV), and Maximum Absolute Deviations of C–C Bond Lengths (Max $|\Delta d_{\text{C-C}}^{S_0}|$ and Max $|\Delta d_{\text{C-C}}^{S_1}|$ in Å) with Different Partitionings of the Active Space (12e,12o) for CASPT2-D and RASPT2-D Calculations.

Partitioning	N^{CSF}	E^{vert}	E^{AEE}	Max $ \Delta d_{\text{C-C}}^{S_0} $	Max $ \Delta d_{\text{C-C}}^{S_1} $	t^{CI} ^a
CAS(12e,12o) ^b	226 512	(3.387)	(2.620)	—	—	462
CAS(12e,12o)	226 512	−0.002	0.000	2.10×10^{-4}	1.30×10^{-4}	460
RAS(12e,12o)/(0e,0o)/2	703	1.154	1.108	5.05×10^{-3}	6.61×10^{-3}	1
RAS(12e,12o)/(0e,0o)/4	28 278	0.031	0.039	9.80×10^{-4}	1.41×10^{-3}	41
RAS(12e,12o)/(0e,0o)/6	147 042	0.033	0.033	1.18×10^{-3}	9.08×10^{-4}	289
RAS(12e,12o)/(2e,2o)/2	2 028	0.780	0.252	4.72×10^{-2}	8.23×10^{-3}	3
RAS(12e,12o)/(4e,4o)/2	8 860	0.073	0.075	2.37×10^{-3}	2.56×10^{-3}	16
RAS(12e,12o)/(6e,6o)/2	36 148	0.037	0.037	1.29×10^{-3}	1.47×10^{-3}	62
RAS(12e,12o)/(8e,8o)/2	115 548	0.016	0.018	3.80×10^{-4}	6.70×10^{-4}	237

^a Wall time for evaluating the partial derivative with respect to the CI coefficient (in seconds).

^b Without the density-fitting approximation; E^{vert} and E^{AEE} are shown for reference.

B. Performance of RASPT2-D

The performance of RASPT2-D against CASPT2-D was evaluated using *trans*-1,3,5,7,9,11-dodecahexaene ($\text{C}_{12}\text{H}_{14}$). Geometry optimizations were performed with various partitionings of the active space, and the vertical excitation energies (E^{vert}) at the S_0 minimum, adiabatic excitation energies (E^{AEE}), and maximum deviations of the C–C bond lengths (as an indicator of the geometrical difference) were calculated. Note that the adiabatic excitation energy is the difference between the energies at the S_1 and S_0 minima, and the zero-point vibrational energy (ZPVE) correction was not considered.

Table II shows that the density-fitting approximation does not sacrifice the accuracy, as observed in Ref. 54. In terms of E^{vert} and E^{AEE} , the deviation is less than 0.002 eV, indicating negligible degradation. The difference in bond length is also negligible; the maximum deviation is only 2.10×10^{-4} Å.

The number of CSFs (N^{CSF}) is greatly reduced by employing RAS references. The compu-

tational cost for evaluating the partial derivative of the energy with respect to the CI coefficient [t^{CI} ; Eq. (15)] is almost proportional to N^{CSF} . Although the use of a very small number of CSFs, such as RAS(12e,12o)/(0e,0o)/2 or RAS(12e,12o)/(2e,2o)/2, is not appropriate, other RASs give reasonable agreement with the reference CAS result; the deviations in excitation energies and C–C bond lengths are less than 0.1 eV and 5.0×10^{-3} Å, respectively. Thus, either more than two electron excitations from RAS1 and into RAS3 or moderate RAS2 spaces should be employed to reasonably reproduce the corresponding CAS result.

The convergence of the SCF and Z-vector equation with the RASs is not as smooth as in the case with CAS; however, analytic derivatives can be efficiently evaluated with larger active spaces.

C. Calculations with Larger Active Spaces

Using the developed method, it is possible to perform geometry optimizations with larger active spaces. As a demonstration, RASSCF and RASPT2-D calculations were performed for dibenzopentalene derivatives⁵³ **2a** and **4a** (Fig. 2) with active spaces of RAS(16e,16o)/(4e,4o)/2 and RAS(20e,20o)/(4e,4o)/2, respectively. The initial structures were taken from Ref. 53 and optimized at the RB3LYP-D3/6-311G* level of theory. The adiabatic singlet–triplet energy gaps ($\Delta E_{\text{ST}} := \Delta E_{\text{S}} - E_{\text{T}}$ without ZPVE corrections) were then computed. The numerical results are summarized in Table III. The gaps computed with spin-flip noncollinear time-dependent density functional theory (SF-NC-TDDFT/6-311G(d) with the PBE50 functional) in Ref. 53 and at the UB3LYP/6-31G(d,p) and UBLYP/6-31G(d,p) levels of theory in Ref. 55 are also tabulated.

The numbers of CSFs for the RAS(20e,20o)/(4e,4o)/2 partitioning were 110 292 and 184 863 for the singlet and triplet states, respectively. Although these numbers are smaller than those for CAS(12e,12o) reported in the previous subsection, the required computational cost was much greater (approximately $t^{\text{CI}} = 7700$ s for the triplet state) due to the large number of active orbitals. Still, these calculations were doable on a single computer node. A comparison of RASSCF and RASPT2-D indicates that RASSCF clearly overestimates the stability of the singlet state by over 5 kcal/mol. The final energy gap predicted with RASPT2-D is similar to that for SF-NC-TDDFT/6-311G(d), although the RASSCF and RASPT2-D calculations predicted the singlet state to be more stable than the triplet state for **2a**, in agreement with the UBLYP/6-31G(d,p) calculations. However, the energy gap is very small, making it difficult to determine the ground state.

In Ref. 53, the open-shell character is discussed with the diradical character, which was ob-

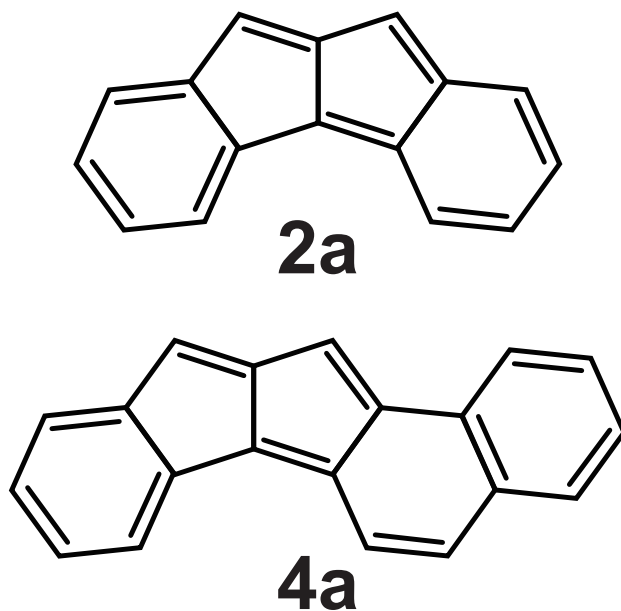


FIG. 2. Molecular structures of **2a** and **4a**.

TABLE III. Adiabatic Singlet–Triplet Energy Gaps (ΔE_{ST}) in kcal/mol at the RASSCF, RASPT2-D, RASPT2, SF-NC-TDDFT,⁵³ UB3LYP, and UBLYP⁵⁵ Levels of Theory.

	2a	4a
RASSCF	−6.85	−15.33
RASPT2-D	−1.17	−8.71
RASPT2 ^a	−1.35	−8.90
SF-NC-TDDFT ⁵³	2.48	−8.20
UB3LYP ⁵⁵	0.23	—
UBLYP ⁵⁵	−1.50	—

^a Single-point energy calculation for the optimized geometry at RASPT2-D

tained from the occupation number of the lowest unoccupied natural orbitals.⁵³ For multiconfiguration methods, the open-shell character may be discussed based on the weight of the configuration. For the ground states of **2a** and **4a**, the contributions of the closed-shell configuration are 32% and 37%, respectively. Although the closed-shell contribution is slightly greater for **4a**, the difference is small, and both derivatives have strong multiconfiguration character.

The presented result for ΔE_{ST} is not a definitive prediction. These RASPT2-D calculations are missing the off-diagonal elements in the zeroth-order Hamiltonian. Still, single-point energy calculations at the “full” RASPT2 level for the geometry optimized at the RASPT2-D level indicate that the degradation introduced by the diagonal approximation is not significant (only 0.2 kcal/mol; see Table III). In addition, considering the dependences of the computed properties on the size of the active space, the accuracy may not be satisfactory with the present definition of the RAS. Based on the error analysis presented in Table II, the present numerical result may deviate from the CAS(20e,20o) result by a few kcal/mol. Nevertheless, the present numerical result demonstrates that geometry optimizations with the (20e,20o) active space can be performed using MRPT, and the predicted gap is rather reasonable.

V. CONCLUSIONS

Analytic gradients of the single-state CASPT2-D and RASPT2-D methods were developed and implemented in a local version of OpenMolcas. Similar to previous CASPT2 developments,^{16,17,19} the responses of the wavefunction parameters were evaluated by solving one linear Z-vector equation. The correctness of the implementation was shown by comparison with numerical gradients. The performance of RASPT2-D against CASPT2-D was also investigated, demonstrating that a sensible partitioning of the active space in RASPT2-D reproduces the corresponding CASPT2-D result well. The present implementation can use active spaces with large sizes up to (20e,20o); however, the RAS2 space and the maximum number of holes and electrons cannot be large. In the future, the inclusion of the off-diagonal blocks of the zeroth-order Hamiltonian⁷ and (extended)⁵⁶ multistate extension⁵⁷ will be considered, including conical intersection search as in Refs. 20, 22, 39, and 58. Moreover, algorithmic improvement is essential for practical use. The present implementation cannot use a large number of atomic orbitals (~ 400).

SUPPLEMENTARY MATERIAL

See the supplementary material for the optimized coordinates of *trans*-1,3,5,7,9,11-dodecahexaene and dibenzopentalene derivatives.

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REFERENCES

- ¹M. Musiał, A. Perera, and R. J. Bartlett, “Multireference coupled-cluster theory: The easy way,” *J. Chem. Phys.* **134**, 114108 (2011).
- ²D. I. Lyakh, M. Musiał, V. F. Lotrich, and R. J. Bartlett, “Multireference nature of chemistry: The coupled-cluster view,” *Chem. Rev.* **112**, 182–243 (2012).
- ³A. Köhn, M. Hanauer, L. A. Mück, T.-C. Jagau, and J. Gauss, “State-specific multireference coupled-cluster theory,” *WIREs Comput. Mol. Sci.* **3**, 176–197 (2013).
- ⁴P. G. Szalay, T. Müller, G. Gidofalvi, H. Lischka, and R. Shepard, “Multiconfiguration self-consistent field and multireference configuration interaction methods and applications,” *Chem. Rev.* **112**, 108–181 (2012).
- ⁵B. O. Roos, P. Linse, P. E. Siegbahn, and M. R. Blomberg, “A simple method for the evaluation of the second-order-perturbation energy from external double-excitations with a CASSCF reference wavefunction,” *Chem. Phys.* **66**, 197 – 207 (1982).
- ⁶K. Andersson, P.-Å. Malmqvist, B. O. Roos, A. J. Sadlej, and K. Wolinski, “Second-order perturbation theory with a CASSCF reference function,” *J. Phys. Chem.* **94**, 5483–5488 (1990).
- ⁷K. Andersson, P.-Å. Malmqvist, and B. O. Roos, “Second-order perturbation theory with a complete active space self-consistent field reference function,” *J. Chem. Phys.* **96**, 1218–1226 (1992).
- ⁸A. A. Granovsky, “Extended multi-configuration quasi-degenerate perturbation theory: The new approach to multi-state multi-reference perturbation theory,” *J. Chem. Phys.* **134**, 214113 (2011).
- ⁹H. Nakano, “Quasidegenerate perturbation theory with multiconfigurational self-consistent-field reference functions,” *J. Chem. Phys.* **99**, 7983–7992 (1993).
- ¹⁰C. Angeli, R. Cimiraglia, S. Evangelisti, T. Leininger, and J.-P. Malrieu, “Introduction of n -electron valence states for multireference perturbation theory,” *J. Chem. Phys.* **114**, 10252–10264 (2001).

- ¹¹C. Angeli, R. Cimiraglia, and J.-P. Malrieu, “ n -electron valence state perturbation theory: a fast implementation of the strongly contracted variant,” *Chem. Phys. Lett.* **350**, 297–305 (2001).
- ¹²C. Angeli, R. Cimiraglia, and J.-P. Malrieu, “ n -electron valence state perturbation theory: A spinless formulation and an efficient implementation of the strongly contracted and of the partially contracted variants,” *J. Chem. Phys.* **117**, 9138–9153 (2002).
- ¹³Y. G. Khait, J. Song, and M. R. Hoffmann, “Explication and revision of generalized van vleck perturbation theory for molecular electronic structure,” *J. Chem. Phys.* **117**, 4133–4145 (2002).
- ¹⁴R. F. Fink, “Two new unitary-invariant and size-consistent perturbation theoretical approaches to the electron correlation energy,” *Chem. Phys. Lett.* **428**, 461–466 (2006).
- ¹⁵J. W. Park, R. Al-Saadon, M. K. MacLeod, T. Shiozaki, and B. Vlaisavljevich, “Multireference electron correlation methods: Journeys along potential energy surfaces,” *Chem. Rev.* **120**, 5878–5909 (2020).
- ¹⁶P. Celani and H.-J. Werner, “Analytical energy gradients for internally contracted second-order multireference perturbation theory,” *J. Chem. Phys.* **119**, 5044–5057 (2003).
- ¹⁷M. K. MacLeod and T. Shiozaki, “Communication: Automatic code generation enables nuclear gradient computations for fully internally contracted multireference theory,” *J. Chem. Phys.* **142**, 051103 (2015).
- ¹⁸Y. Nishimoto, “Time-dependent long-range-corrected density-functional tight-binding method combined with the polarizable continuum model,” *J. Phys. Chem. A* **123**, 5649–5659 (2019).
- ¹⁹C. Song, J. B. Neaton, and T. J. Martínez, “Reduced scaling formulation of CASPT2 analytical gradients using the supporting subspace method,” *J. Chem. Phys.* **154**, 014103 (2021).
- ²⁰Y. Nishimoto, “Locating conical intersections using the quasidegenerate partially and strongly contracted NEVPT2 methods,” *Chem. Phys. Lett.* **744**, 137219 (2020).
- ²¹J. W. Park and T. Shiozaki, “On-the-fly CASPT2 surface-hopping dynamics,” *J. Chem. Theory Comput.* **13**, 3676–3683 (2017).
- ²²J. W. Park, “Analytical gradient theory for quasidegenerate n -electron valence state perturbation theory (QD-NEVPT2),” *J. Chem. Theory Comput.* **16**, 326–339 (2020).
- ²³P.-Å. Malmqvist, A. Rendell, and B. O. Roos, “The restricted active space self-consistent-field method, implemented with a split graph unitary group approach,” *J. Phys. Chem.* **94**, 5477–5482 (1990).
- ²⁴P. Celani and H.-J. Werner, “Multireference perturbation theory for large restricted and selected active space reference wave functions,” *J. Chem. Phys.* **112**, 5546–5557 (2000).

- ²⁵P.-Å. Malmqvist, K. Pierloot, A. R. M. Shahi, C. J. Cramer, and L. Gagliardi, “The restricted active space followed by second-order perturbation theory method: Theory and application to the study of CuO₂ and Cu₂O₂ systems,” *J. Chem. Phys.* **128**, 204109 (2008).
- ²⁶H. Nakano, J. Nakatani, and K. Hirao, “Second-order quasi-degenerate perturbation theory with quasi-complete active space self-consistent field reference functions,” *J. Chem. Phys.* **114**, 1133–1141 (2001).
- ²⁷H. Nakano, R. Uchiyama, and K. Hirao, “Quasi-degenerate perturbation theory with general multiconfiguration self-consistent field reference functions,” *J. Comput. Chem.* **23**, 1166–1175 (2002).
- ²⁸L. Roskop and M. S. Gordon, “Quasi-degenerate second-order perturbation theory for occupation restricted multiple active space self-consistent field reference functions,” *J. Chem. Phys.* **135**, 044101 (2011).
- ²⁹Y. Kurashige and T. Yanai, “Second-order perturbation theory with a density matrix renormalization group self-consistent field reference function: Theory and application to the study of chromium dimer,” *J. Chem. Phys.* **135**, 094104 (2011).
- ³⁰J. W. Park, R. Al-Saadon, N. E. Strand, and T. Shiozaki, “Imaginary shift in CASPT2 nuclear gradient and derivative coupling theory,” *J. Chem. Theory Comput.* **15**, 4088–4098 (2019).
- ³¹B. O. Roos and K. Andersson, “Multiconfigurational perturbation theory with level shift — the Cr₂ potential revisited,” *Chem. Phys. Lett.* **245**, 215–223 (1995).
- ³²N. Forsberg and P.-Å. Malmqvist, “Multiconfiguration perturbation theory with imaginary level shift,” *Chem. Phys. Lett.* **274**, 196–204 (1997).
- ³³F. Menezes, D. Kats, and H.-J. Werner, “Local complete active space second-order perturbation theory using pair natural orbitals (PNO-CASPT2),” *J. Chem. Phys.* **145**, 124115 (2016).
- ³⁴T. Helgaker, S. Coriani, P. Jørgensen, K. Kristensen, J. Olsen, and K. Ruud, “Recent advances in wave function-based methods of molecular-property calculations,” *Chem. Rev.* **112**, 543–631 (2012).
- ³⁵N. C. Handy and H. F. Schaefer III, “On the evaluation of analytic energy derivatives for correlated wave functions,” *J. Chem. Phys.* **81**, 5031–5033 (1984).
- ³⁶K. L. Bak, J. Boatz, and J. Simons, “First-order geometrical response equations for state-averaged multiconfigurational self-consistent field (SA-MCSCF) wave functions,” *Int. J. Quantum Chem.* **40**, 361–378 (1991).

- ³⁷J. Stålring, A. Bernhardsson, and R. Lindh, “Analytical gradients of a state average MCSCF state and a state average diagnostic,” *Mol. Phys.* **99**, 103–114 (2001).
- ³⁸I. Fdez. Galván, M. G. Delcey, T. B. Pedersen, F. Aquilante, and R. Lindh, “Analytical state-average complete-active-space self-consistent field nonadiabatic coupling vectors: Implementation with density-fitted two-electron integrals and application to conical intersections,” *J. Chem. Theory Comput.* **12**, 3636–3653 (2016).
- ³⁹J. W. Park and T. Shiozaki, “Analytical derivative coupling for multistate CASPT2 theory,” *J. Chem. Theory Comput.* **13**, 2561–2570 (2017).
- ⁴⁰J. W. Park, “Analytical gradient theory for strongly contracted (SC) and partially contracted (PC) n-electron valence state perturbation theory (NEVPT2),” *J. Chem. Theory Comput.* **15**, 5417–5425 (2019).
- ⁴¹R. Moccia, “Variable bases in SCF MO calculations,” *Chem. Phys. Lett.* **5**, 260–264 (1970).
- ⁴²N. Handy, R. Amos, J. Gaw, J. Rice, and E. Simandiras, “The elimination of singularities in derivative calculations,” *Chem. Phys. Lett.* **120**, 151–158 (1985).
- ⁴³I. Fdez. Galván, M. Vacher, A. Alavi, C. Angeli, F. Aquilante, J. Autschbach, J. J. Bao, S. I. Bokarev, N. A. Bogdanov, R. K. Carlson, L. F. Chibotaru, J. Creutzberg, N. Dattani, M. G. Delcey, S. S. Dong, A. Dreuw, L. Freitag, L. M. Frutos, L. Gagliardi, F. Gendron, A. Giussani, L. González, G. Grell, M. Guo, C. E. Hoyer, M. Johansson, S. Keller, S. Knecht, G. Kovačević, E. Källman, G. Li Manni, M. Lundberg, Y. Ma, S. Mai, J. P. Malhado, P.-Å. Malmqvist, P. Marquetand, S. A. Mewes, J. Norell, M. Olivucci, M. Oppel, Q. M. Phung, K. Pierloot, F. Plasser, M. Reiher, A. M. Sand, I. Schapiro, P. Sharma, C. J. Stein, L. K. Sørensen, D. G. Truhlar, M. Ugandi, L. Ungur, A. Valentini, S. Vancoillie, V. Veryazov, O. Weser, T. A. Wesolowski, P.-O. Widmark, S. Wouters, A. Zech, J. P. Zobel, and R. Lindh, “OpenMolcas: From source code to insight,” *J. Chem. Theory Comput.* **15**, 5925–5964 (2019).
- ⁴⁴F. Aquilante, J. Autschbach, A. Baiardi, S. Battaglia, V. A. Borin, L. F. Chibotaru, I. Conti, L. De Vico, M. Delcey, I. Fdez. Galván, N. Ferré, L. Freitag, M. Garavelli, X. Gong, S. Knecht, E. D. Larsson, R. Lindh, M. Lundberg, P. Å. Malmqvist, A. Nenov, J. Norell, M. Odelius, M. Olivucci, T. B. Pedersen, L. Pedraza-González, Q. M. Phung, K. Pierloot, M. Reiher, I. Schapiro, J. Segarra-Martí, F. Segatta, L. Seijo, S. Sen, D.-C. Sergentu, C. J. Stein, L. Ungur, M. Vacher, A. Valentini, and V. Veryazov, “Modern quantum chemistry with [Open]Molcas,” *J. Chem. Phys.* **152**, 214117 (2020).

- ⁴⁵J. Olsen, D. L. Yeager, and P. Jørgensen, "Optimization and characterization of a multiconfigurational self-consistent field (MCSCF) state," in *Advances in Chemical Physics* (John Wiley & Sons, Ltd, 1983) pp. 1–176.
- ⁴⁶J. W. Park, "Single-state single-reference and multistate multireference zeroth-order hamiltonians in MS-CASPT2 and conical intersections," *J. Chem. Theory Comput.* **15**, 3960–3973 (2019).
- ⁴⁷T. H. Dunning, "Gaussian basis sets for use in correlated molecular calculations. I. The atoms boron through neon and hydrogen," *J. Chem. Phys.* **90**, 1007–1023 (1989).
- ⁴⁸R. A. Kendall, T. H. Dunning, and R. J. Harrison, "Electron affinities of the first-row atoms revisited. Systematic basis sets and wave functions," *J. Chem. Phys.* **96**, 6796–6806 (1992).
- ⁴⁹D. Feller, "The role of databases in support of computational chemistry calculations," *J. Comput. Chem.* **17**, 1571–1586 (1996).
- ⁵⁰K. L. Schuchardt, B. T. Didier, T. Elsethagen, L. Sun, V. Gurumoorthi, J. Chase, J. Li, and T. L. Windus, "Basis set exchange: A community database for computational sciences," *J. Chem. Inf. Model.* **47**, 1045–1052 (2007).
- ⁵¹B. P. Pritchard, D. Altarawy, B. Didier, T. D. Gibson, and T. L. Windus, "New basis set exchange: An open, up-to-date resource for the molecular sciences community," *J. Chem. Inf. Model.* **59**, 4814–4820 (2019).
- ⁵²G. Ghigo, B. O. Roos, and P.-Å. Malmqvist, "A modified definition of the zeroth-order hamiltonian in multiconfigurational perturbation theory (CASPT2)," *Chem. Phys. Lett.* **396**, 142 – 149 (2004).
- ⁵³A. Konishi, Y. Okada, R. Kishi, M. Nakano, and M. Yasuda, "Enhancement of antiaromatic character via additional benzoannulation into dibenzo[a,f]pentalene: Syntheses and properties of benzo[a]naphtho[2,1-f]pentalene and dinaphtho[2,1-a,f]pentalene," *J. Am. Chem. Soc.* **141**, 560–571 (2019).
- ⁵⁴W. Győrffy, T. Shiozaki, G. Knizia, and H.-J. Werner, "Analytical energy gradients for second-order multireference perturbation theory using density fitting," *J. Chem. Phys.* **138**, 104104 (2013).
- ⁵⁵A. Konishi, Y. Okada, M. Nakano, K. Sugisaki, K. Sato, T. Takui, and M. Yasuda, "Synthesis and characterization of dibenzo[a,f]pentalene: Harmonization of the antiaromatic and singlet biradical character," *J. Am. Chem. Soc.* **139**, 15284–15287 (2017).
- ⁵⁶T. Shiozaki, W. Győrffy, P. Celani, and H.-J. Werner, "Communication: Extended multi-state complete active space second-order perturbation theory: Energy and nuclear gradients," *J. Chem.*

- Phys. **135**, 081106 (2011).
- ⁵⁷J. Finley, P.-Å. Malmqvist, B. O. Roos, and L. Serrano-Andrés, “The multi-state CASPT2 method,” Chem. Phys. Lett. **288**, 299–306 (1998).
- ⁵⁸J. W. Park, “Analytical first-order derivatives of second-order extended multiconfiguration quasi-degenerate perturbation theory (XMCQDPT2): Implementation and application,” J. Chem. Theory Comput. **16**, 5562–5571 (2020).