

Biosynthetic Crossover of 5-Lipoxygenase and Cyclooxygenase-2 Yields 5-Hydroxy-PGE₂ and 5-Hydroxy-PGD₂

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Running title: Discovery of 5-hydroxy-prostaglandins

Abbreviations:

COX	cyclooxygenase
ESI	electrospray ionization
HETE	hydroxy-eicosatetraenoic acid
HHT	hydroxy-heptadecatrienoic acid
4-HNE	4-hydroxy-nonenal
H-PGDS	hematopoietic prostaglandin D synthase
LC-MS	liquid chromatography-mass spectrometry
LOX	lipoxygenase
LPS	lipopolysaccharide
MDA	malondialdehyde
PG	prostaglandin

Abstract

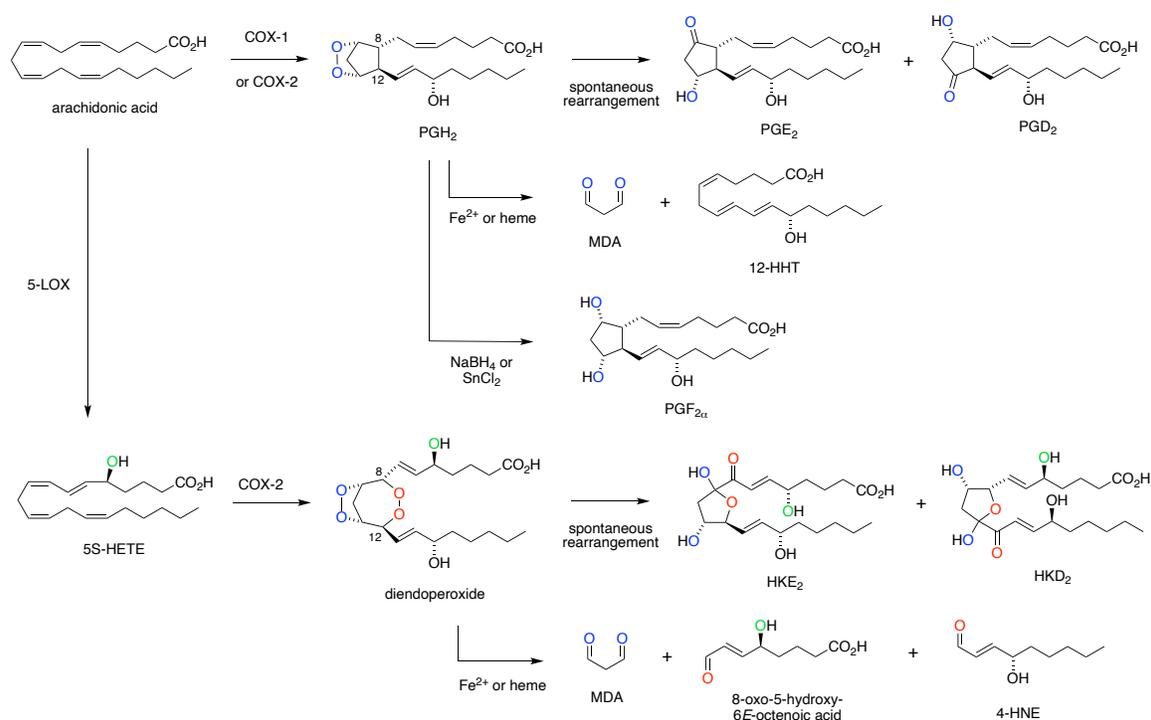
Biosynthetic crossover of 5-lipoxygenase (5-LOX) and cyclooxygenase-2 (COX-2) enzymatic activities is a productive pathway to convert arachidonic acid into unique eicosanoids. Here we show that COX-2 catalysis with 5-LOX derived 5-hydroxy-eicosatetraenoic acid yields the endoperoxide 5-hydroxy-PGH₂ that spontaneously rearranges to 5-OH-PGE₂ and 5-OH-PGD₂, the 5-hydroxy analogs of arachidonic acid derived PGE₂ and PGD₂. The endoperoxide was identified via its predicted degradation product, 5,12-dihydroxy-heptadecatri-6*E*,8*E*,10*E*-enoic acid, and by SnCl₂-mediated reduction to 5-OH-PGF_{2α}. Both 5-OH-PGE₂ and 5-OH-PGD₂ were unstable and degraded rapidly upon treatment with weak base. The instability hampered detection in biologic samples which was overcome by in situ reduction using NaBH₄ to yield the corresponding stable 5-OH-PGF₂ diastereomers and enabled detection of 5-OH-PGF_{2α} in activated primary human leukocytes. 5-OH-PGE₂ and 5-OH-PGD₂ were unable to activate EP and DP prostanoid receptors suggesting their bioactivity is distinct from PGE₂ and PGD₂.

1. Introduction

The inducible form of cyclooxygenase (COX), COX-2, has a larger active site (1) resulting in a wider substrate specificity than COX-1, the isoform that reacts preferably with arachidonic acid (2,3). Among the unique substrates for COX-2 is 5-hydroxy-eicosatetraenoic acid (5-HETE), a product of the reaction of 5-lipoxygenase (5-LOX) with arachidonic acid (4). The reaction of COX-2 with 5-HETE yields the highly oxygenated hemiketal (HK) eicosanoids HKE_2 and HKD_2 that result from triple oxygenation of the substrate (5) while arachidonic acid is reacted with two molecules of oxygen when forming prostaglandins (Scheme 1) (6,7). Reaction of COX-2 with 5-HETE and other isoform-specific substrates like fatty acid amides (anandamide) (8) and esters (2-arachidonyl-glycerol) (9) suggests that the resulting products serve specific biologic functions that are distinct from those of traditional prostaglandins (3,10).

Scheme 1

Biosynthesis and transformation of endoperoxides in the reaction of cyclooxygenases with arachidonic acid and 5-HETE



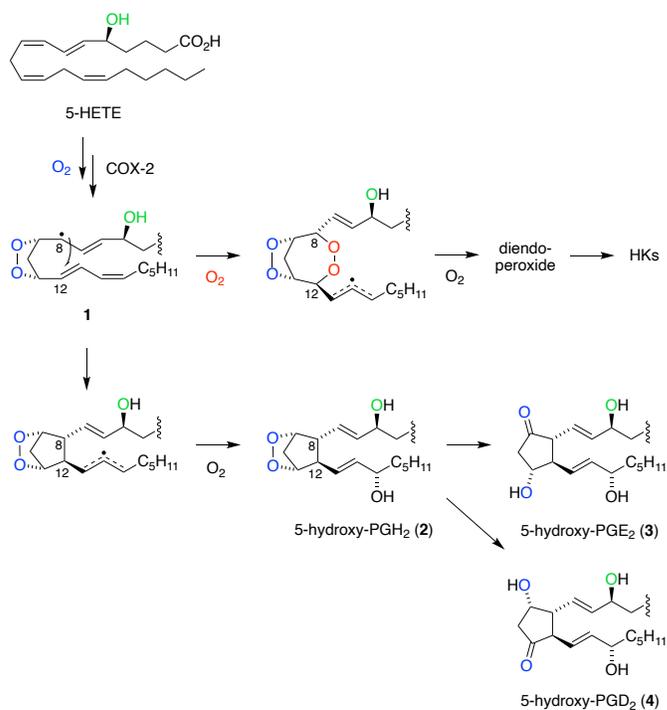
The main catalytic product of the reaction of COX-2 with 5-HETE is a diendoperoxide as a precursor to the two HK eicosanoids (Scheme 1) (4,5). The diendoperoxide features two fused 5- and 7-membered rings with peroxides linking carbons 9 and 11 as well as 8 and 12, respectively. Spontaneous opening of both peroxides in a reaction equivalent to the non-enzymatic rearrangement of PGH₂ to PGE₂ and PGD₂ (11) followed by forming the hemiketal ring moiety yields HKE₂ and HKD₂ (Scheme 1) (5). Alternatively, the diendoperoxide degrades to the aldehyde fragments malondialdehyde (MDA), 4-hydroxy-2E-nonenal (4-HNE), and 8-oxo-5-hydroxy-6E-octenoic acid in the presence of

ferrous iron or excess heme (12), again in a reaction equivalent to the degradation of PGH₂ to MDA and 12-hydroxy-heptadecatrienoic acid (12-HHT) (11). Other, less abundant products resulting from the COX-2 reaction with 5-HETE are 5,11- and 5,15-dihETE (13), the 5-hydroxy analogues of the minor AA/COX products, 11- and 15-HETE (11,14).

A critical intermediate in COX-2 transformation of 5-HETE is the C-8 carbon radical of the 9,11-endoperoxide **1** (Scheme 2). During prostaglandin formation from arachidonic acid the C-8 radical reacts with C-12 to yield the five-membered prostanoid ring. During formation of the diendoperoxide from 5-HETE, the C-8 radical reacts with O₂ followed by reaction of the peroxy radical with C-12, thus expanding the prostanoid ring to include a peroxide linking carbons 8 and 12 (4). The final product in either case is formed by stereospecific oxygenation and reduction at C-15. Direct addition of C-8 to C-12 would predict the formation of 5-hydroxy-analogue **2** of PGH₂ as a precursor to novel prostanoids (Scheme 2). Here, we describe the conversion of 5-HETE to 5-OH-PGH₂ by COX-2 yielding 5-OH-PGE₂ and -PGD₂ as novel products of the biosynthetic crossover of 5-LOX and COX-2 and their formation in human leukocytes that have been activated to express both enzymes.

Scheme 2

Reaction of the C-8 carbinyl radical intermediate **1 in COX-2 catalysis with 5-HETE determines the formation of hemiketals (HK) or 5-hydroxy-prostaglandins as enzymatic products**



2. Results and Discussion

Novel products in the reaction of COX-2 with 5-HETE

Negative ion LC-MS analysis of products formed by the reaction of recombinant human COX-2 with 5-HETE revealed two products **3** and **4** with m/z 367.3, equivalent to a molecular weight of 368.5 g/mol (Fig. 1). The increase of 16 mass units compared to prostaglandins like PGD₂ and PGE₂ (MW 352.5 g/mol) suggested that the novel products represent 5-hydroxy analogs of PGD₂ and PGE₂. Products **3** and **4** eluted at a retention time similar to the hemiketal products HKD₂ and HKE₂ (m/z 399.3; MW 400.5 g/mol) while unreacted 5-HETE substrate eluted at 5.1 min retention time (Fig. 1).

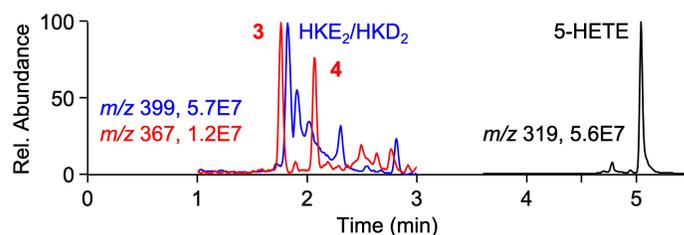


Figure 1. LC-MS detection of HKE₂ and HKD₂ (m/z 399 ion trace) and novel products **3** and **4** (m/z 367) in a reaction of recombinant human COX-2 with 5-HETE substrate (m/z 319).

Comparison of the ion intensities in negative ion mode MS1 analyses, albeit an imperfect measure of relative intensities, suggested that **3** and **4** were about 20% of the abundance of HKE₂ and HKD₂, the major products formed in the reaction of COX-2 with 5-HETE *in vitro* as determined using radiolabeled [¹⁴C]5-HETE substrate (5).

Identification of 5-OH-PGD₂ and 5-OH-PGE₂

Product **4** was isolated from a large-scale reaction of 5-HETE with COX-2 conducted in the presence of hematopoietic prostaglandin D synthase (H-PGDS). H-PGDS was included in the enzymatic reaction to facilitate transformation of the predicted COX-2 catalytic product, 5-OH-PGH₂ **2**, to 5-OH-PGD₂. Homo- and heteronuclear 1D- and 2D-NMR analyses of product **4** showed signals for a 5-membered prostanoid ring with 9-hydroxyl and 11-keto moieties and the side chains at C-8 and C-12 in *trans* configuration. The presence of two allylic hydroxy groups, representing the 5-OH-6*E*-ene and 15-OH-13*E*-ene moieties, confirmed the identification of **4** as 5-OH-PGD₂. Incubations of 5-HETE with COX-2 did not afford a sufficient amount for NMR analysis of **3** which was presumed to be 5-OH-PGE₂. As an alternative approach, a standard of 5-OH-PGE₂ was independently prepared by chemical synthesis starting from PGE₂ as described below. Comparison to the synthetic standard confirmed identification of **3** as 5-OH-PGE₂.

Synthesis of 5-OH-PGD₂ and 5-OH-PGE₂

5-OH-PGD₂ and 5-OH-PGE₂ were synthesized by singlet oxidation of PGD₂ and PGE₂, respectively. Reaction of PGD₂ in methanol with methylene blue and visible light followed by reduction (PPh₃) yielded 5*S*-OH-PGD₂ **4** and 5*R*-OH-PGD₂ **5** as well as the 6-OH-4*E*-isomers (**6**, **7**) (Supporting Information, Fig. S1A). Singlet oxidation of PGE₂ gave two major products **3** and **8**, identified as 5*S*- and 5*R*-OH-PGE₂, respectively, with

insignificant formation of the 6-OH isomers (Supporting Information, Fig. S1B). The absolute configuration of C-5 of the synthesized 5-OH-PGs (**3**, **4**, **5**, and **8**) was determined by LC-MS analysis of diastereomeric cleavage products formed by methylation (CH_2N_2), derivatization with (-)-menthyl chloroformate, and oxidative ozonolysis (15) (Supporting Information, Fig. S2). Authentic standards for the assignment of the configuration of (-)-menthoxy carbonyl 2*S*- and 2*R*-hydroxy-1,6-hexanedioic acid-6-methyl ester diastereomers were obtained starting with synthetic 5*S*- and 5*R*-HETE, respectively, for which the absolute configuration was known (16).

Reduction and degradation of 5-OH-PGH₂

The structural analogy of 5-OH-PGD₂ and 5-OH-PGE₂ with PGD₂ and PGE₂ suggested the former were formed by spontaneous rearrangement of an endoperoxide precursor, predicted to be 5-OH-PGH₂ **2**, as the actual product of the COX-2 reaction with 5-HETE. 5-OH-PGH₂ was predicted to be too unstable to attempt isolation and direct structural identification. In order to test whether endoperoxide **2** was the product of COX-2 catalysis with 5-HETE, the enzymatic reaction was treated with SnCl₂ as a mild reducing agent for the endoperoxide group (17). LC-MS analysis showed formation of 5-OH-PGF_{2α} **9** as a reduction product with a molecular ion at *m/z* 369 that was absent without SnCl₂ treatment (Fig. 2A). Product **9** co-eluted with 5-OH-PGF_{2α} that was formed by NaBH₄ reduction of synthetic 5-OH-PGD₂ as described below.

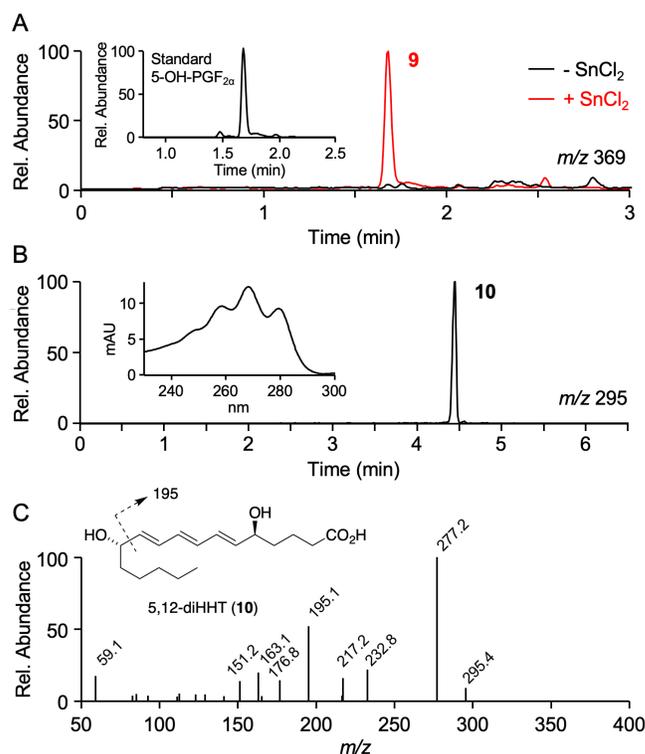


Figure 2. Indirect identification of 5-OH-PGH₂. (A) Treatment with SnCl₂ of a 5-min reaction of COX-2 with 5-HETE generated product **9** with *m/z* 369 that had the same retention time as a standard of 5-OH-PGF_{2α} formed by reduction (NaBH₄) of 5-OH-PGD₂ (inset). (B) Treatment of the enzymatic reaction with hematin yielded product **10**

identified as 5,12-dihydroxy-heptadecatri-6*E*,8*E*,10*E*-enoic acid (5,12-diHHT) due to its UV spectrum characteristic of a conjugated triene in *E,E,E* configuration (inset) and (C) fragment ions derived from m/z 295.4 (**10**) in the MS² spectrum.

Further evidence for 5-OH-PGH₂ as the product of COX-2 catalysis with 5-HETE was obtained from its degradation in the presence of hematin, similar to the transformation of PGH₂ to 12-HHT (18). Recombinant human COX-2 was incubated with 5-HETE for 5 min followed by addition of hematin to induce degradation. The products were extracted and analyzed by RP-HPLC with diode array detection and by LC-MS. The RP-HPLC chromatogram showed elution of product **10** with a UV chromophore typical of a conjugated triene with λ_{max} at 268 nm and diagnostic shoulders at 260 and 280 nm, with the deeply defined bathochromic shoulder indicating *E,E,E*-configuration of the conjugated triene (Fig. 2B) (19,20). Product **10** was isolated using RP-HPLC and analyzed by LC-MS operated in the negative ion mode which gave a molecular ion at m/z 295.4, indicating an MW of 296.4 g/mol (Fig. 2C). UV spectrum, MW, and the MS fragment ions indicated that product **10** was 5,12-dihydroxy-heptadecatri-6*E*,8*E*,10*E*-enoic acid (5,12-diHHT).

Stability of 5-OH-PGD₂ and 5-OH-PGE₂

Treatment of PGE₂ and PGD₂ with base induces dehydration of the hydroxyl groups at C-11 or C-9, respectively, resulting in formation of the respective enone prostanoids (21,22). Base treatment dehydrates PGE₂ (λ_{max} 205 nm) to the enone PGA₂ (λ_{max} 217 nm) (23) that further rearranges to the dienone PGB₂ with λ_{max} at 278 nm (Fig. 3A) (24). Likewise, PGD₂ dehydrates to PGJ₂ that rearranges to Δ^{12} -PGJ₂ and undergoes a second dehydration to yield 15-deoxy- $\Delta^{12,14}$ -PGJ₂ (22,25).

When base treatment is conducted in a UV/Vis cuvette, the dehydration can be followed by observing the change of the chromophores over time. For example, treatment of PGE₂ with NaOH (200 mM) in Fig. 3B illustrates the time course of formation of PGB₂. To test the stability of 5-OH-PGE₂ a 10-fold lower concentration of base was used (20 mM NaOH) and the scan frequency was doubled to 1/min. Under these conditions, the first scan recorded immediately after addition of 5-OH-PGE₂ to the cuvette indicated complete conversion to a product with a dienone-like chromophore at UV295 nm (Fig. 3C). The dienone-like chromophore decreased rapidly to yield products with less distinct absorbance. This indicated that 5-OH-PGE₂ was much more sensitive to base-induced dehydration than PGE₂, and that further transformation yielded several undefined products. When 5-OH-PGE₂ was incubated in a buffer of pH 8 the time course of formation of the UV295 nm chromophore, taken as a measure of the initial dehydration, indicated a half-life of \approx 12 min (Fig. 3D). When 5-OH-PGD₂ was treated with base (20 mM NaOH, scan rate 1/min) a similar change of chromophores was observed (Fig. 3E), suggesting a comparable half-life.

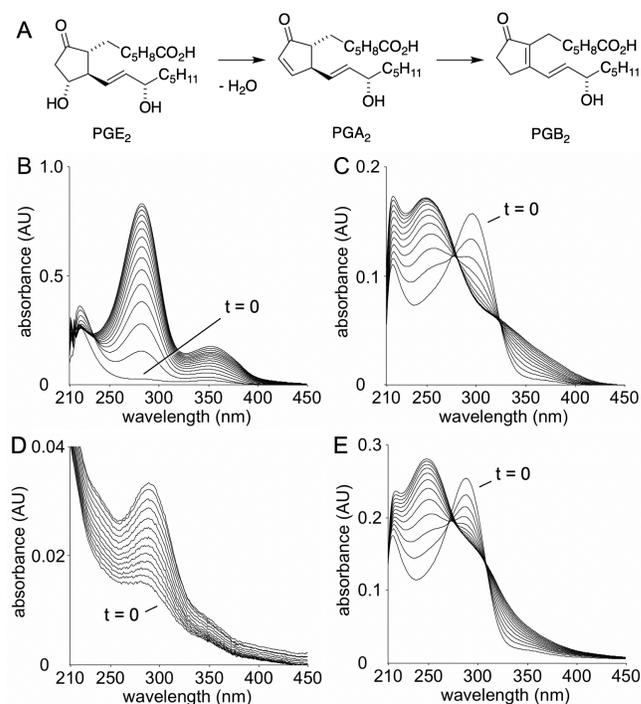


Figure 3. Base-catalyzed dehydration of prostaglandins. (A) Dehydration of PGE₂ yields PGA₂ that rearranges to PGB₂. (B) Time-course of dehydration of PGE₂ in NaOH (200 mM) illustrates formation of PGA₂ (λ_{max} 217 nm) and PGB₂ (λ_{max} at 278 nm). UV/Vis spectra were acquired every 2 min. (C) Degradation of 5-OH-PGE₂ in 10-times dilute NaOH (20 mM) indicating complete conversion to a product with λ_{max} 295 nm at the time of the first scan. (D) Degradation of 5-OH-PGE₂ in Tris-HCl (50 mM, pH 8). (E) Degradation of 5-OH-PGD₂ in NaOH (20 mM), with $t = 0$ min indicating conversion to a product with λ_{max} 295 nm. The UV/Vis scanning rate was 1/min in panels C-E.

In situ reduction of 5-OH-PGs

The chemical instability was predicted to impede analysis of 5-OH-PG biosynthesis in a biologic environment due to their rapid transformation to uncharacterized products. To minimize loss of 5-OH-PGs during cell incubations, extraction, and analysis a method was developed for in situ reduction to the 5-OH-PGF₂ analogues. Reduction of the dehydration-prone cyclopentanones and remaining endoperoxide was predicted to increase stability as is observed for PGF_{2 α} and F₂-isoprostanes relative to cyclopentanone PGs (26). Recombinant COX-2 was incubated with 5-HETE for 5 min followed by addition of NaBH₄ in order to achieve reduction of the immediate products to 5-OH-PGF₂ diastereomers. Analysis using LC-MS gave two products with the expected m/z 369 molecular ion, eluting at 1.81 min (**11**) and 1.90 min (**9**) retention time, respectively (Fig. 4A). It was assumed that **11** and **9** were diastereomers of 5-OH-PGF₂ but the configuration of the hydroxy groups on the prostanoid ring was unclear, i.e., whether it was α or β . Reduction of synthetic 5-OH-PGE₂ gave **11** and **9** in a ratio of $\approx 75:25$ (Fig. 4B). Reduction of synthetic 5-OH-PGD₂ gave **9** as the overwhelming product together with a second isomer **12** (Fig. 4C) (27). These analyses suggested that the α and β

diastereomers were not formed in a 1:1 ratio in the reduction of the carbonyl by NaBH₄. It should be noted that reduction of 5-OH-PGE₂ and 5-OH-PGD₂ will only yield three diastereomers since both precursors will form the F_{2α} diastereomer whereas only the other diastereomer is unique. Comparison to the PGF₂ diastereomers formed by reduction of PGD₂ and PGE₂ (Supporting Information, Fig. S3) enabled to identify product **11** as 5-OH-PGF_{2β} and **9** as 5-OH-PGF_{2α}. The third diastereomer, 5-OH-11β-PGF_{2α} **12**, was formed only at low abundance by reduction of 5-OH-PGD₂.

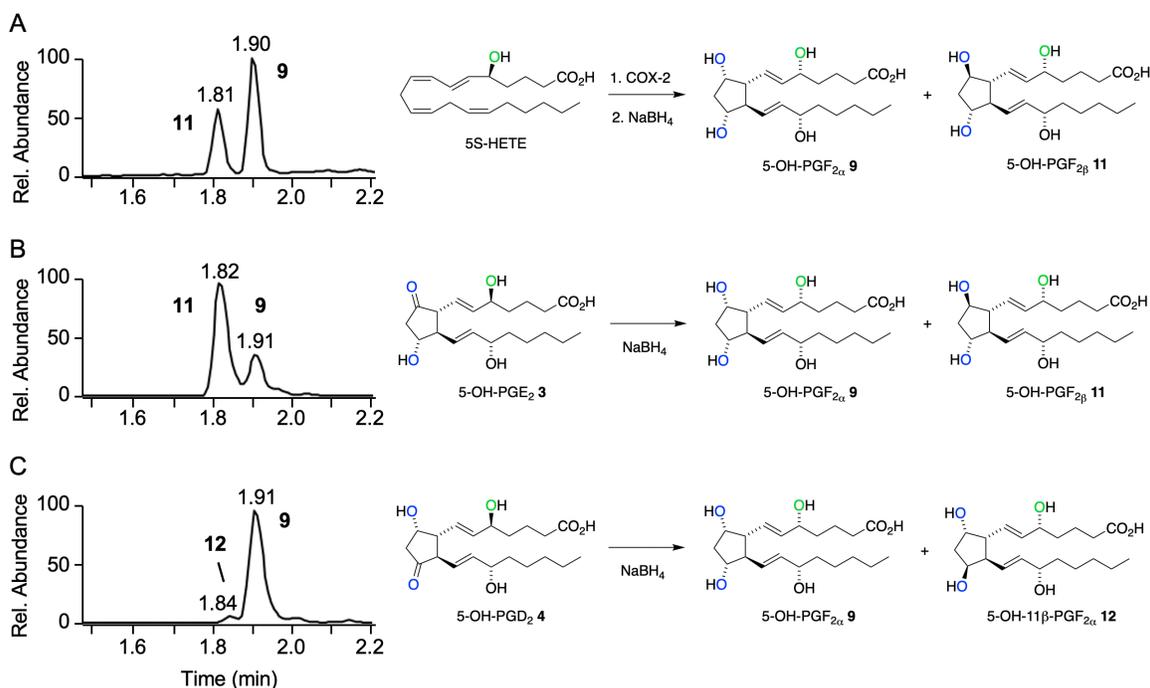


Figure 4. 5-OH-PGF₂ diastereomers formed by NaBH₄ reduction of (A) a 5-min reaction of 5-HETE with COX-2, (B) synthetic 5-OH-PGD₂, and (C) synthetic 5-OH-PGE₂. The ion trace for *m/z* 369 from LC-MS analysis of the reactions is shown. Products in (A) are expected to be mainly derived from reduction of 5-OH-PGH₂.

Reduction of the diendoperoxide and hemiketals in a crude reaction of 5-HETE with COX-2 as well as reduction of HKE₂ gave a single peak (RT 1.69 min) with *m/z* 403, representing 5,8,9,11,12,15-hexahydroxy-eicosadienoic acid **13** (Supporting Information, Fig. S4). Detection of a single peak indicated that fewer than the expected diastereomers were formed and/or that the diastereomers were not resolved chromatographically. This matter was not further investigated.

Formation of 5-OH-PGs by human leukocytes

Biosynthetic cross-over of 5-LOX and COX-2 activities in leukocyte mixtures isolated from normal human volunteers and stimulated *ex vivo* with LPS and A23187 yielded HKE₂, HKD₂, as well as 5,11- and 5,15-diHETEs (28,29). The use of inhibitors confirmed the role of the two enzymes in biosynthesis (28,29). To test whether human leukocytes also form 5-OH-PGs the cells were activated and then treated *in situ* using NaBH₄ to

reduce unstable 5-OH-PGs to the 5-OH-PGF₂ diastereomers prior to extraction and LC-MS analysis.

Freshly isolated human leukocytes stimulated with LPS and A23187 showed formation of 5-OH-PGF_{2α} along with 5,8,9,11,12,15-hexahydroxy-eicosadienoic acid **13**, the reduction product of HKE₂, HKD₂, and their diendoperoxide precursor, respectively (Fig. 5). 5-OH-PGE₂ and 5-OH-PGD₂ could not be detected, even when samples were not treated with NaBH₄ (Fig. 5). The absence of 5-OH-PGF_{2β} indicated that the leukocyte mixtures preferentially formed 5-OH-PGD₂ over 5-OH-PGE₂ or that 5-OH-PGH₂ was the most abundant product at the time when NaBH₄ was added. As expected, HKE₂ and HKD₂ were absent or markedly decreased after NaBH₄ reduction.

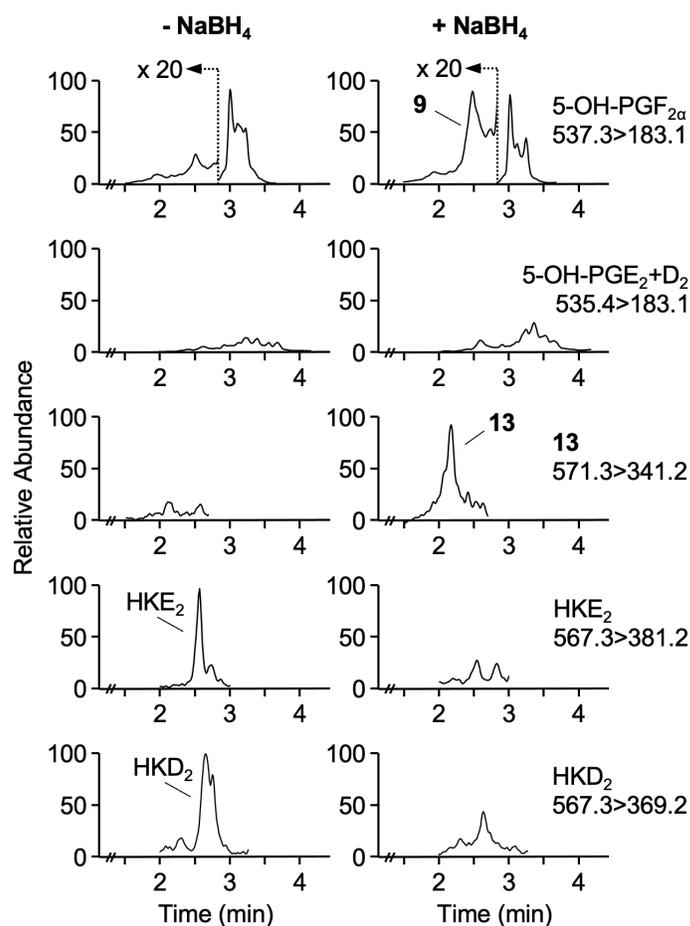


Figure 5. Formation of 5-OH-PGs by activated human leukocytes. Cells were treated with LPS for 45 min and then with A23187 for 15 min, followed by reduction or not with NaBH₄, extracted, derivatized with AMPP, and analyzed using LC-MS in the positive ion SRM mode. Ion traces for the detection of 5-OH-PGF_{2α}, 5-OH-PGD₂ and 5-OH-PGE₂, hexahydroxy-eicosadienoic acid **13**, HKE₂, and HKD₂ are shown.

Receptor activation

The understanding of the biological role of the 5-LOX/COX-2 crossover eicosanoids is still in its infancy. A factor holding back progress is the difficulty to prepare HKs by chemical or enzymatic synthesis (30,31). The ready synthesis, albeit on a small scale, of 5-OH-PGE₂ and 5-OH-PGD₂ starting from PGE₂ and PGD₂, respectively, enabled to test whether they activate EP and DP prostanoid receptors. Prostanoid receptors appeared a logical target due to the close structural similarity between prostaglandins and their 5-OH-analogues. However, neither 5-OH-PGE₂ nor 5-OH-PGD₂ activated the EP (EP1-EP4), DP1, FP, IP, or TP prostanoid receptors (Supporting Information, Figs. S5 and S6). This outcome was consistent with the absence of a 5-hydroxy modification in the structures of the known prostanoid analogues that are agonists of these receptors (<https://www.guidetopharmacology.org/GRAC/FamilyDisplayForward?familyId=58>; accessed March 2021).

3. Conclusion

Biosynthesis of 5-OH-PGH₂ from 5-HETE follows the known radical and oxygenation reactions catalyzed in the COX reaction with arachidonic acid (6). The 9,11-endoperoxide-8-carbinyl radical is a crucial catalytic intermediate (Scheme 2). Reaction of the carbon radical with oxygen or not prior to addition to the 12*E* double bond determines the outcome of the enzymatic reaction, i.e., formation of the diendoperoxide versus 5-OH-PGH₂ as the precursors to HKs and 5-OH-PGs, respectively. Using the purified enzyme *in vitro*, HKs are about 10-fold more abundant than 5-OH-PGs, indicating that oxygenation of the C-8 radical is favored over direct reaction with C-12. Nevertheless, it is interesting to speculate on a mechanism how the enzyme might control the ratio between the products. Control may be accomplished, for example, via regulating the concentration of O₂ or through an allosteric effect of the non-catalytic on the catalytic subunit of the functional COX-2 heterodimer (33,34). In the latter scenario, binding of non-substrate fatty acids like palmitate to the allosteric subunit of COX-2 increases efficiency of the catalytic subunit with arachidonic acid by more than 2-fold (35,36). Subunit allosteric effects also account for substrate-selective inhibitor potency. For example, the NSAIDs ibuprofen and mefenamic acid inhibited COX-2 oxygenation of 2-AG with orders of magnitude greater potency than oxygenation of arachidonic acid (37,38). This testifies to regulation of catalysis through interaction between the subunits of the COX dimer, and similar regulatory events may control the reactivity of the C-8 radical and thus the ratio of HKs versus 5-OH-PGs formed in COX-2 catalysis with 5-HETE.

Formation of 5-OH-PGs is biologically relevant beyond the mechanistic insight into COX-2 catalysis since it occurred in response to *ex vivo* stimulation of primary leukocytes isolated from normal human volunteers. Biological effects of the novel prostanoids await to be uncovered but the lack of activation of EP and DP prostanoid receptors by 5-OH-PGs suggests they exhibit bioactivity that is unique and different from traditional prostaglandins.

Associated Content

The Supporting Information is available free of charge at .

Experimental procedures, SI figures, and NMR data for new compounds

Notes

The authors declare no competing financial interest exists.

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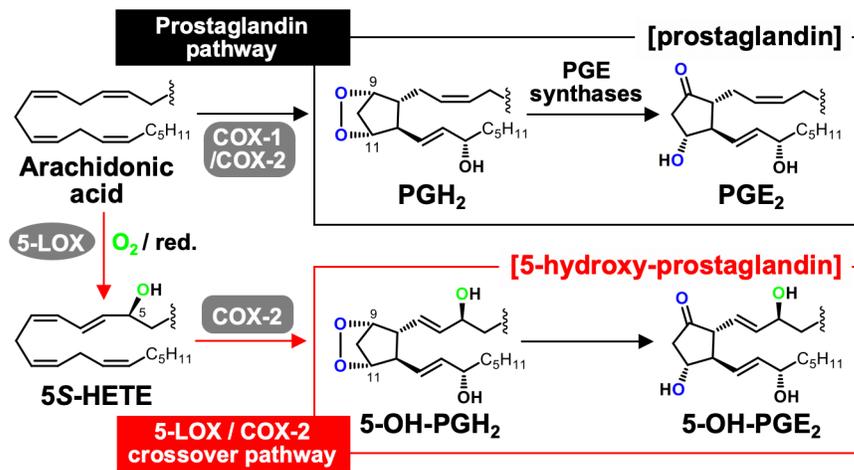
References

1. Blobaum, A. L., and Marnett, L. J. (2007) Structural and functional basis of cyclooxygenase inhibition. *J. Med. Chem.* **50**, 1425-1441
2. Smith, W. L., DeWitt, D. L., and Garavito, R. M. (2000) Cyclooxygenases: Structural, cellular, and molecular biology. *Annu. Rev. Biochem.* **69**, 145-182
3. Rouzer, C. A., and Marnett, L. J. (2008) Non-redundant functions of cyclooxygenases: oxygenation of endocannabinoids. *J. Biol. Chem.* **283**, 8065-8069
4. Schneider, C., Boeglin, W. E., Yin, H., Stec, D. F., and Voehler, M. (2006) Convergent oxygenation of arachidonic acid by 5-lipoxygenase and cyclooxygenase-2. *J. Am. Chem. Soc.* **128**, 720-721
5. Griesser, M., Suzuki, T., Tejera, N., Mont, S., Boeglin, W. E., Pozzi, A., and Schneider, C. (2011) Biosynthesis of hemiketal eicosanoids by cross-over of the 5-lipoxygenase and cyclooxygenase-2 pathways. *Proc. Natl. Acad. Sci. U.S.A.* **108**, 6945-6950
6. van der Donk, W. A., Tsai, A. L., and Kulmacz, R. J. (2002) The cyclooxygenase reaction mechanism. *Biochemistry* **41**, 15451-15458
7. Schneider, C., Pratt, D. A., Porter, N. A., and Brash, A. R. (2007) Control of oxygenation in lipoxygenase and cyclooxygenase catalysis. *Chem. Biol.* **14**, 473-488
8. Yu, M., Ives, D., and Ramesha, C. S. (1997) Synthesis of prostaglandin E₂ ethanolamide from anandamide by cyclooxygenase-2. *J. Biol. Chem.* **272**, 21181-21186
9. Kozak, K. R., Rowlinson, S. W., and Marnett, L. J. (2000) Oxygenation of the endocannabinoid, 2-arachidonylglycerol, to glyceryl prostaglandins by cyclooxygenase-2. *J. Biol. Chem.* **275**, 33744-33749

10. Hermanson, D. J., Gamble-George, J. C., Marnett, L. J., and Patel, S. (2014) Substrate-selective COX-2 inhibition as a novel strategy for therapeutic endocannabinoid augmentation. *Trends Pharmacol. Sci.* **35**, 358-367
11. Hamberg, M., and Samuelsson, B. (1967) Oxygenation of unsaturated fatty acids by the vesicular gland of sheep. *J. Biol. Chem.* **242**, 5344-5354
12. Griesser, M., Boeglin, W. E., Suzuki, T., and Schneider, C. (2009) Convergence of the 5-LOX and COX-2 pathways. Heme-catalyzed cleavage of the 5S-HETE-derived di-endoperoxide into aldehyde fragments. *J. Lipid Res.* **50**, 2455-2462
13. Mulugeta, S., Suzuki, T., Tejera Hernandez, N., Griesser, M., Boeglin, W. E., and Schneider, C. (2010) Identification and absolute configuration of dihydroxy-arachidonic acids formed by oxygenation of 5S-HETE by native and aspirin-acetylated COX-2. *J. Lipid Res.* **51**, 575-585
14. Baer, A. N., Costello, P. B., and Green, F. A. (1991) Stereospecificity of the products of the fatty acid oxygenases derived from psoriatic scales. *J. Lipid Res.* **32**, 341-347
15. Hamberg, M. (1983) A novel transformation of 13-*L*_s-hydroperoxy-9,11-octadecadienoic acid. *Biochim. Biophys. Acta* **752**, 191-197
16. Schneider, C., Yu, Z., Boeglin, W. E., Zheng, Y., and Brash, A. R. (2007) Enantiomeric separation of hydroxy and hydroperoxy eicosanoids by chiral column chromatography. *Methods Enzymol.* **433**, 145-157
17. Valmsen, K., Järving, I., Boeglin, W. E., Varvas, K., Koljak, R., Pehk, T., Brash, A. R., and Samel, N. (2001) The origin of 15*R*-prostaglandins in the Caribbean coral *Plexaura homomalla*: Molecular cloning and expression of a novel cyclooxygenase. *Proc. Natl. Acad. Sci. U.S.A.* **98**, 7700-7705
18. Hamberg, M., and Samuelsson, B. (1974) Prostaglandin endoperoxides. Novel transformations of arachidonic acid in human platelets. *Proc. Natl. Acad. Sci. U.S.A.* **71**, 3400-3404
19. Borgeat, P., and Samuelsson, B. (1979) Metabolism of arachidonic acid in polymorphonuclear leukocytes. Structural analysis of novel hydroxylated compounds. *J. Biol. Chem.* **254**, 7865-7869
20. Maas, R. L., Brash, A. R., and Oates, J. A. (1981) A second pathway of leukotriene biosynthesis in porcine leukocytes. *Proc. Natl. Acad. Sci. U.S.A.* **78**, 5523-5527
21. Polet, H., and Levine, L. (1975) Metabolism of prostaglandins E, A, and C in serum. *J. Biol. Chem.* **250**, 351-357
22. Fitzpatrick, F. A., and Wynalda, M. A. (1983) Albumin-catalyzed metabolism of prostaglandin D₂. Identification of products formed in vitro. *J. Biol. Chem.* **258**, 11713-11718
23. Pike, J. E., Lincoln, F. H., and Schneide.Wp. (1969) Prostanoid acid chemistry. *J. Org. Chem.* **34**, 3552-&
24. Floyd, M. B., Schaub, R. E., Siuta, G. J., Skotnicki, J. S., Grudzinskas, C. V., Weiss, M. J., Dessy, F., and VanHumbecck, L. (1980) Prostaglandins and congeners. 22. Synthesis of 11-substituted derivatives of 11-deoxyprostaglandins E₁ and E₂. Potential bronchodilators. *J. Med. Chem.* **23**, 903-913
25. Kikawa, Y., Narumiya, S., Fukushima, M., Wakatsuka, H., and Hayaishi, O. (1984) 9-Deoxy-delta 9, delta 12-13,14-dihydroprostaglandin D₂, a metabolite of

- prostaglandin D2 formed in human plasma. *Proc. Natl. Acad. Sci. U.S.A.* **81**, 1317-1321
26. Morrow, J. D. (2000) The isoprostanes: their quantification as an index of oxidant stress status in vivo. *Drug Metab. Rev.* **32**, 377-385
 27. Hamberg, M., and Samuelsson, B. (1973) Detection and isolation of an endoperoxide intermediate in prostaglandin biosynthesis. *Proc. Natl. Acad. Sci. U.S.A.* **70**, 899-903
 28. Tejera, N., Boeglin, W. E., Suzuki, T., and Schneider, C. (2012) COX-2-dependent and -independent biosynthesis of dihydroxy-arachidonic acids in activated human leukocytes. *J. Lipid Res.* **53**, 87-94
 29. Gimenez-Bastida, J. A., Shibata, T., Uchida, K., and Schneider, C. (2017) Roles of 5-lipoxygenase and cyclooxygenase-2 in the biosynthesis of hemiketals E2 and D2 by activated human leukocytes. *FASEB J.* **31**, 1867-1878
 30. Gimenez-Bastida, J. A., Suzuki, T., Sprinkel, K. C., Boeglin, W. E., and Schneider, C. (2017) Biomimetic synthesis of hemiketal eicosanoids for biological testing. *Prostaglandins Other Lipid Mediat.* **132**, 41-46
 31. Boer, R. E., Gimenez-Bastida, J. A., Boutaud, O., Jana, S., Schneider, C., and Sulikowski, G. A. (2018) Total synthesis and biological activity of the arachidonic acid metabolite Hemiketal E2. *Org. Lett.* **20**, 4020-4022
 32. Ungrin, M. D., Carriere, M. C., Denis, D., Lamontagne, S., Sawyer, N., Stocco, R., Tremblay, N., Metters, K. M., and Abramovitz, M. (2001) Key structural features of prostaglandin E(2) and prostanoid analogs involved in binding and activation of the human EP(1) prostanoid receptor. *Mol. Pharmacol.* **59**, 1446-1456
 33. Yuan, C., Rieke, C. J., Rimon, G., Wingerd, B. A., and Smith, W. L. (2006) Partnering between monomers of cyclooxygenase-2 homodimers. *Proc. Natl. Acad. Sci. U.S.A.* **103**, 6142-6147
 34. Smith, W. L., and Malkowski, M. G. (2019) Interactions of fatty acids, nonsteroidal anti-inflammatory drugs, and coxibs with the catalytic and allosteric subunits of cyclooxygenases-1 and -2. *J. Biol. Chem.* **294**, 1697-1705
 35. Dong, L., Vecchio, A. J., Sharma, N. P., Jurban, B. J., Malkowski, M. G., and Smith, W. L. (2011) Human cyclooxygenase-2 is a sequence homodimer that functions as a conformational heterodimer. *J. Biol. Chem.* **286**, 19035-19046
 36. Dong, L., Zou, H., Yuan, C., Hong, Y. H., Kuklev, D. V., and Smith, W. L. (2016) Different fatty acids compete with arachidonic acid for binding to the allosteric or catalytic subunits of cyclooxygenases to regulate prostanoid synthesis. *J. Biol. Chem.* **291**, 4069-4078
 37. Prusakiewicz, J. J., Duggan, K. C., Rouzer, C. A., and Marnett, L. J. (2009) Differential sensitivity and mechanism of inhibition of COX-2 oxygenation of arachidonic acid and 2-arachidonoylglycerol by ibuprofen and mefenamic acid. *Biochemistry* **48**, 7353-7355
 38. Duggan, K. C., Hermanson, D. J., Musee, J., Prusakiewicz, J. J., Scheib, J. L., Carter, B. D., Banerjee, S., Oates, J. A., and Marnett, L. J. (2011) (R)-Profens are substrate-selective inhibitors of endocannabinoid oxygenation by COX-2. *Nat. Chem. Biol.* **7**, 803-809

TOC Graphic



Synopsis

New prostaglandin analogues are formed in the reaction of cyclooxygenase-2 with 5-HETE, a 5-lipoxygenase metabolite of arachidonic acid, and identified in activated human leukocytes.