Closing the yellow gap with Eu- and Tb-doped GaN: one luminescent host resulting in three colours

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Abstract:

Gallium nitride (GaN) is a key material when it comes to light-emitting diodes (LEDs) and has pushed the LED revolution in lighting and displays.

The investigations of Nakamura and coworkers, being awarded the Nobel prize in 2014, made p-doping of GaN easily accessible by a post-growth thermal annealing treatment and triggered the invention of highly efficient blue-emitting diodes.

The concept of down-conversion of a GaN-based blue LED offers also the possibility to provide efficient generation of monochromatic, high-color purity light resulting in a highly efficient warm-white all-nitride phosphor-converted light emitting diode (pc-LED).

Although the down conversion of blue light from InGaN LEDs has become a dominant technique for producing white light, there are still some technical challenges, e.g. the immiscibility of GaN and InN and the lattice mismatch between the substrate and InGaN, that have to be overcome.

Here we demonstrate the doping of bulk GaN with europium, terbium and the combination of both resulting in intriguing luminescence properties, pushing the role of GaN:Eu,Tb as a chief component in future light emitting diodes.
This colour tuning proves that one luminescence host can provide three colours (red, green and orange) and that even the so called "yellow gap" could be closed with a III-nitride.

By using one material for all colours, it will be possible to overcome the technical challenges in building up LED devices, which will open up new capabilities for modern highly efficient phosphors.

Main

Gallium nitride (GaN) has set the benchmark in the last decades when it comes to light-emitting diodes (LEDs). As a key material, it has driven the LED revolution in lighting and displays. The lighting industry with its various market segments, including automotive lighting, indoor and outdoor lighting, medical applications, lifestyle products and vertical farming has arisen from these fundamental discoveries enabled by the exploration of light's interaction with matter.[1,2] The first light emission from a solid-state material driven by an electric current was reported in 1907 by *H.J. Round*[3] and this outreaching discovery paved the way of electroluminescence and highly efficient phosphors in modern LED technology. This investigation triggered further studies on the optoelectronic processes taking place in semiconductors based on the recombination of electric charges.[4,5] The first GaAsP LEDs was reported by Craford *et al.*[6,7] in 1971 and 1972, but many technical challenges still were to overcome. The development of metal-organic vapour phase epitaxy (MOVPE) technique led the foundation for all future milestones to come. The final breakthrough came from the investigations of Nakamura and coworkers[8–10], making p-doping of GaN and its ternary alloys (InGaN, AlGaN) easily accessible by a post-growth thermal annealing treatment and launched the invention of highly efficient blue-emitting diodes.

In 2014, the Nobel Prize in Physics was awarded to Isamu Akasaki, Hiroshi Amano and Shuji Nakamura for "the invention of efficient blue light-emitting diodes which has enabled bright and energy-saving white light sources".[11]

White light from LEDs can be build up combining a red, green and blue emitting semiconductor or by phosphors down-converting the emission of short wavelength emitting GaN/InGaN LEDs.[12–14]

One of the challenges here is the comparatively low external quantum efficiency (EQE) of the green-emitting semiconductors, because of the lattice mismatch between the substrate and InGaN, causing high defect density, immiscibility of GaN and InN[15] or chemical interaction with packaging materials. As such, there is an urgent necessity of developing a solid-state material with a very narrow emission bandwidth in the green spectral region.[14]

The concept of down conversion of a GaN-based blue LED offers also the possibility to provide efficient generation of monochromatic, high-color purity light resulting in a highly efficient warm-white all-nitride phosphor-converted light emitting diode (pc-LED). Eu^{2+}-doped nitridosilicates and oxonitridosilicates emerged as important host lattices for phosphor-converted light-emitting diodes (pc-LEDs) due to their very high chemical and thermal stability, their very high quantum efficiency of the luminescence process and their very low thermal

quenching.[13,16–26] Namely $M_2Si_5N_8:Eu^{2+}$ and $MSi_2O_2N_2:Eu^{2+}$ (M = alkaline earth) have been employed as highly effective red–orange (2-5-8) and yellow–green (1-2-2-2) phosphors, respectively.

Even the so called "yellow gap", neither III-nitrides nor III-phosphides had been able to close, could be bridged with $(Ba,Sr)_2Si_5N_8:Eu^{2+}$.[14]

Just recently, new Mg-nitridosilicates ($Ca[Mg_3SiN_4]:Ce^{3+}$, $Sr[Mg_3SiN_4]:Eu^{2+}$, $Eu[Mg_3SiN]$[27,28]), Mg-nitridoaluminates ($M[Mg_2Al_2N_4]:Eu^{2+}$ with M = Ca,Sr,Ba[29] and $Sr[LiAl_3N_4]:Eu^{2+}$[30]) and $Sr_2[BeAl_3N_5]:Eu^{2+}$[31] emerged as next generation red phosphor materials with superior luminescence properties.[2] Many attempts concerning MOVPE-grown GaN, GaN nanowires/ nanocrystals, GaN quantum dots doped with Eu or in some cases with Tb, can be found in literature.[32–39] (see supplement) All these investigations consist only of monolayers or 2D materials having very few nanometers of thickness.

To cover the colour range from blue, over green and orange to red (ca. 450 to 650 nm), a combination of InGaN und AlGa(In)P is necessary. Several challenges have to be addressed here. The decrease of the external quantum efficiency versus emission wavelength around 560 nm, is termed the "yellow gap".[14] The immiscibility of GaN and InN leads to a reduction in performance of InGaN-based LEDs with higher InN mole fractions.[14] The lattice mismatch between GaN, InN, GaAlN, AlGaP, different thermal expansion coefficients and the variations of the In/Al content decrease the luminescence performance tremendously.

Down conversion of blue light from InGaN LEDs by suitable color converters, especially phosphors, has become a dominant technique for producing white light, but still some technical challenges have to be overcome. One point here are structural misfits of thin films because of the different host materials. Using one bulk material (GaN) for all colours (blue, red, green, orange) could enable therefore the next step of modern high performance GaN LEDs.

Here, we demonstrate the doping of bulk GaN with europium and terbium and the combination of both results in intriguing luminescence properties of all three doped compounds. This renders GaN:Eu,Tb[40] as a prospective chief component in future light emitting diodes (LEDs).

By using various synthetic approaches (e.g. low temperature urea-based method) we succeeded to co-dope GaN (Chempur 99,999%) with MCl_3 x 6 H_2O or $M(NO_3)_3$ x 5 H_2O (M= Eu, Tb) as substitutes (see Figure 1). For a comparison of the powder diffraction pattern of $GaN:Eu^{3+}$ and $GaN:Tb^{3+}$ see supplement Figure S1. SEM EDX measurements of GaN confirmed the atomic ratio of Ga:N of 1:2 and an europium and terbium content of 3-5% was found, but no oxygen within the detection limit.
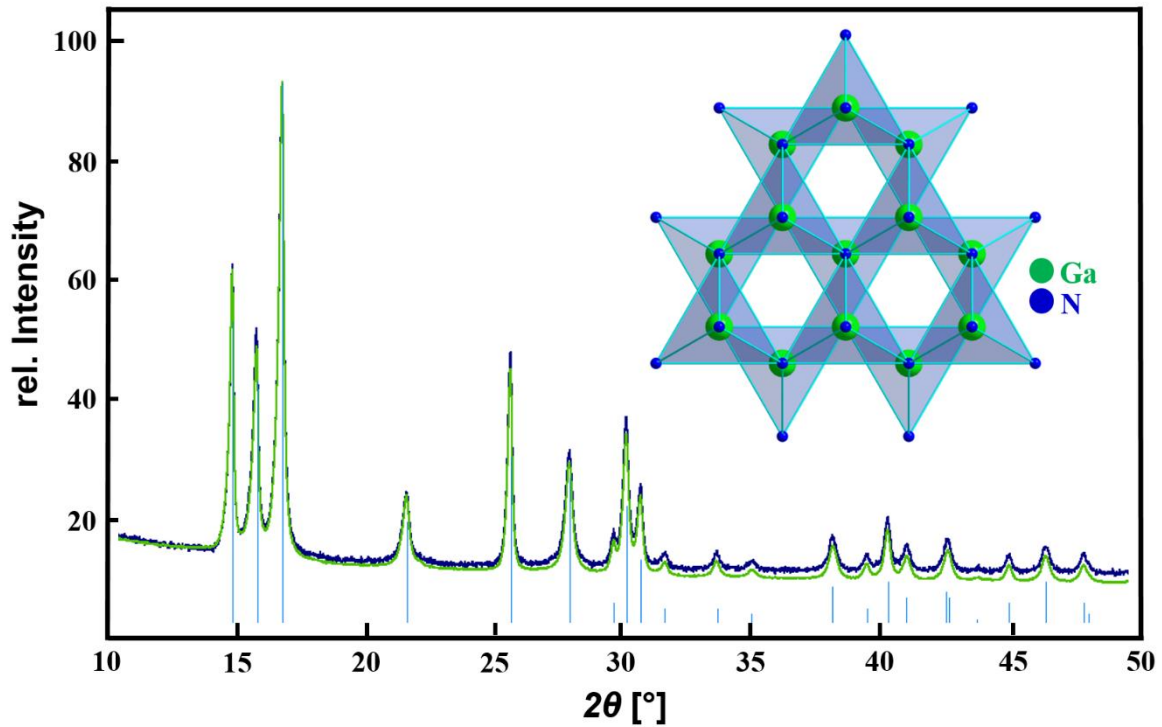
Figure 1: X-ray powder diffraction patterns of undoped GaN (blue), GaN:Tb (green) and GaN ICSD [50-07920] (bright blue), (λ = 0.709026 Å), the inset shows the characteristic motif of the underlying GaN structure.

Luminescence properties

Eu^{3+} is considered as one of the most important activator ions with red emission corresponding to the transition $^5D_0$–^7F_J (J = 1–6). The green emission of Tb^{3+} is due to the transition between the emitting states of ^5D_J and the excited states of ^7F_J. The main intense green emission is attributed to the transition of $^5D_4$–$^7F_5$, which is located at approx. 544 nm. A relevant point here is that normally the doping ion is inserted during the main synthesis and not afterwards. In general, doping is performed with Eu^{2+} and only one activator ion is used. Consequently, one host can only provide one colour. Since different aspects come into play when we think about mixing colours within one host, such an approach for colour tuning offers tremendous opportunities for highly efficient phosphors.

In this work, we doped GaN with Eu^{3+} and Tb^{3+} as activator ions, with each individually and with both simultaneously. Figure 2c shows the CIE 1931 diagram and the colours of doped GaN, proving that a colour tuning of one host with different activator ions and their combination is possible. Having a closer look at the CIE and the basics of colour mixing, it is clear that the combination of green and red leads to orange. (see Figure 2d) This effect could be proven in the meantime for other doped nitrides and carbodiimides as well.[40,41]
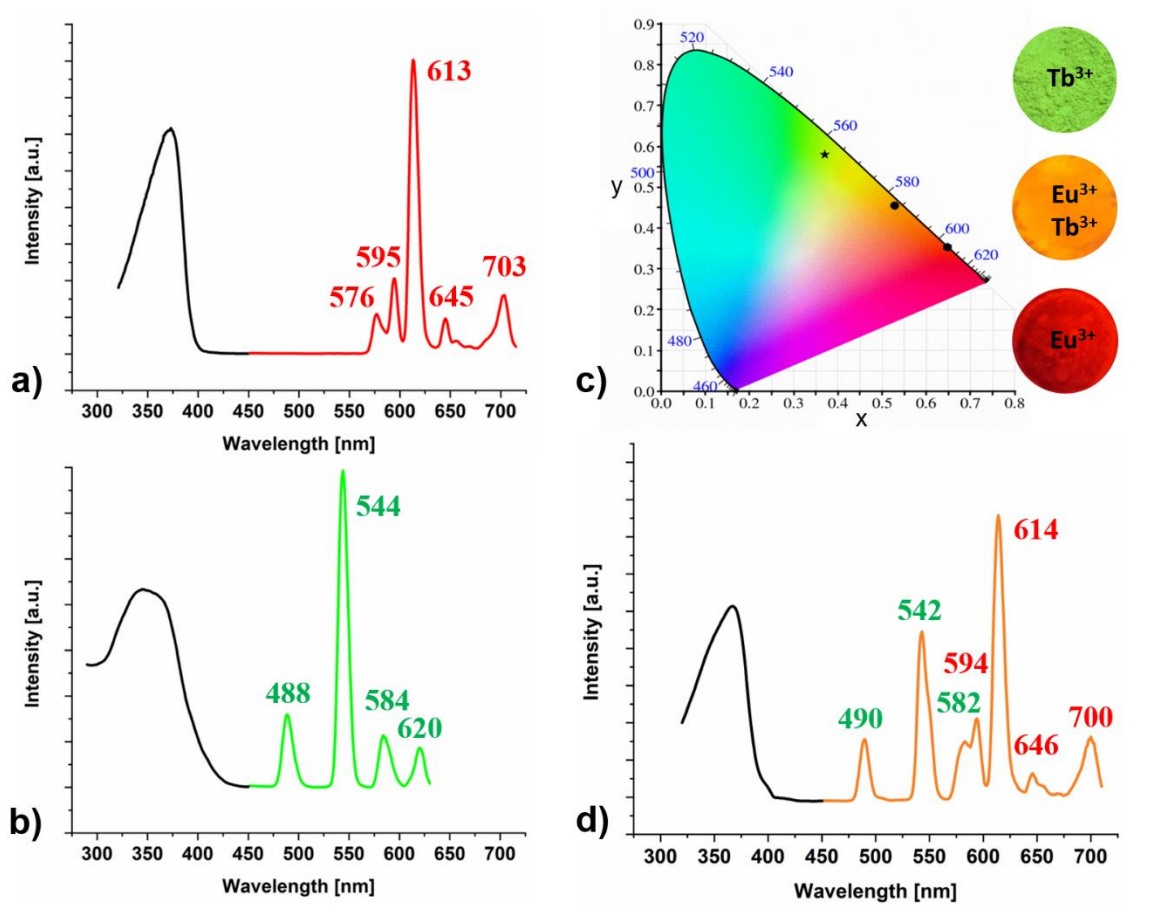
Figure 2: Excitation (black) and emission spectra (coloured) of GaN doped with a) Eu^{3+} (red, ●), b) Tb^{3+} (green, ★) and d) Eu^{3+} and Tb^{3+} (orange, ●), c) CIE 1931 diagram of GaN doped with Tb^{3+}, Eu^{3+}, Tb^{3+} and Eu^{3+}.

GaN: Eu^{3+} shows strong characteristic emission peaks of Eu^{3+} within the region of 570–700 nm, indicating an energy transfer between the GaN host and the Eu^{3+} ions. The peaks associated with the intra-4f shell transitions ($^5D_0$ - ^7F_J) of the Eu^{3+} ions, are the peaks at 576, 595, 613 and 645 nm, wherein the one at 613 nm is the strongest (see Figure 2a). Peng *et al.*[39] assumed according to extended x-ray absorption fine structure measurements, that the Eu^{3+} ion substitutes the Ga site, which is in accordance with our assumption.

The excitation (monitored at 544 nm) and emission spectra of Tb^{3+} doped GaN are represented in Figure 2b. The excitation spectrum exhibits a broad and intense band in the range from 290 nm to 380 nm with a peak at around 322 nm. This broad band is attributed to $4f^8$–$4f^75d^1$ transition of the Tb^{3+} ions. The strongest emission peak is at 544 nm with a **F**ull **W**idth at **H**alf **M**aximum (FWHM) ~12nm corresponding to the $^5D_4 \rightarrow {}^7F_5$ transition, while the peaks at 488 nm, 584 nm and 620 nm, respectively, originate from the $^5D_4 \rightarrow {}^7F_6$, $^5D_4 \rightarrow {}^7F_4$ and

$^5D_4 \rightarrow {}^7F_3$ transitions of the Tb^{3+} ions. Until now, green emitting phospors could only be realized as oxynitrides (e.g. β-SiAlON:Eu^{2+}[42]) or oxonitridosilicates (e.g. $SrSi_2O_2N_2$:Eu^{2+}[43,44]). The here presented luminescence behavior of GaN:Tb^{3+} shows that green emitting phosphors[40] can be also achieved with purely nitridic compounds.

The energy level distributions of Tb^{3+} and Eu^{3+} have a large overlap and their energy transfer has been proven to be very effective.[45] The blue-green light of the Tb^{3+} transition ($^5D_4 \rightarrow {}^7F_{6,5}$) is emitted by polychromatic relaxation and the energy is transferred to the $^5D_1$ and $^5D_0$ levels of the Eu^{3+} by cross relaxation. The Eu^{3+} ions absorb the energy from Tb^{3+} and emit therefore orange light.

Figure 2d proves that it is possible to insert two different activator ions in one host showing the typical bands of Eu^{3+} as well as the ones of Tb^{3+} in one spectrum resulting in a saturated orange body colour. However a close look reveals that this is not only a superposition of the Eu^{3+}-and the Tb^{3+}-spectrum of GaN, because this orange spectrum differs clearly concerning peak form, intensity and wavelength. Here it becomes evident, that this orange colour is only possible by mixing the ions on an atomic scale and cannot be realized by a mixture of particles of the red and the green doped GaN.

To prove this GaN:Eu and GaN:Tb (same molar ratio Eu:Tb as in GaN:Eu,Tb) have been mixed in a mortar. It becomes obvious that this does not result in an orange luminescent GaN. Instead the red GaN:Eu is dominant in the mixture, resulting in the same red colour coordinates of just GaN:Eu. See Figure S10.

The FWHM point out the same values as in the single dopings. The line widths of the emission spectra (FWHM) of nitride-based LEDs vary typically between 20-35 nm and of phosphide-based ones between 15-25 nm[12], however here we range from 8-12 nm for GaN:Eu^{3+},Tb^{3+} which is extremely narrow banded.

The CIE values of amber emitting Phosphor GaN:Eu^{3+}/Tb^{3+} x,y = 0.528, 0.454 (see Figure 2c) are very similar to those of $(Ba,Sr)_2Si_5N_8$:Eu^{2+}[14](x,y = 0.579, 0.416) and are lying therefore well within the "amber box" of the SAE specifications.[14] This 2-5-8 nitridosilicate phosphor has been considered as an important breakthrough for bridging the "yellow gap"[14], and is established nowadays in numerous optoelectronic applications.

To elucidate if the Eu,Tb ions really have been inserted into the GaN structure and the luminescence does not result from the respective doping compounds a comparison of the luminescence spectra of GaN:Eu^{3+} and $EuCl_3$ x 6 H_2O is shown in Figure S1. As optical and luminescence spectra are highly sensitive to structural deformation of the nearest environment

of rare-earth (RE) ions, it is clearly evident that the spectrum has changed due to the insertion of the Eu cation into the structure of GaN. Furthermore, very detailed comparisons of several Eu- and Tb-doped GaN thin films and possible side products have been carried out (see supplement Figures S2-9). Here it becomes obvious that some luminescence spectra seem very similar, but when taking a closer look, they clearly differ. According to De Boer et al.[33], sample growth and doping conditions play as well an important role in influencing the PL (photoluminescence) spectra.

Mott-Schottky (MS) measurements

Electrochemical Impedance Spectroscopy (EIS) is an appropriate tool to study ion diffusion and to resolve the chemical identity of the charge carriers by the use of blocking electrodes. Mott-Schottky (MS) measurements are a very sensitive technique to probe changes in the electronic band structure, i.e. charge carrier density, type of semiconducting behaviour and band edge positions, upon doping. For the elucidation of doping effect on the electronic properties of GaN, Mott-Schottky (MS) measurements were conducted for the bare and RE-doped samples. They were performed in a 0.1 M potassium phosphate electrolyte at an applied frequency of 10 Hz. As shown in Figure 3, all of the acquired curves show a positive slope corresponding to the characteristic of n-type semiconductors.[46] In addition, a smaller slope for all curves can be attributed to increased charge carrier density for GaN upon doping with the rare earth metals.[47,48] The extrapolation of the curves allows to estimate the flat band potential as the corresponding conduction band edge of n-type semiconductor. The obtained conduction band edges reveal that RE-doped GaN yield a significant negative shift, indicating RE doping can be applied to modify the (photo)electrochemical properties of GaN.[48–50]
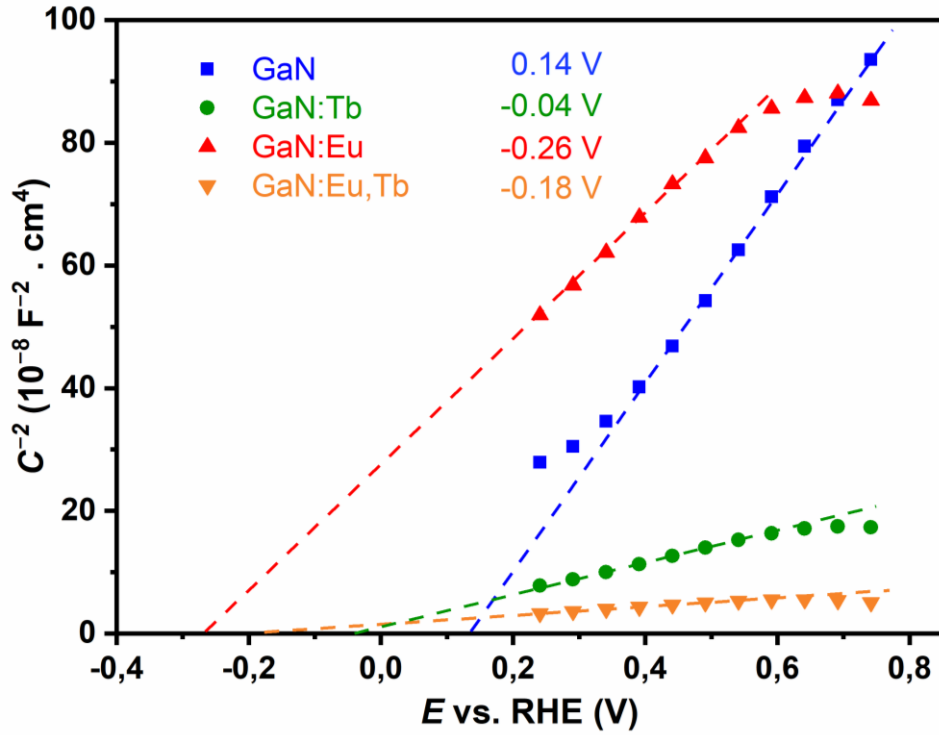
Figure 3: Mott-Schottky (MS) analysis of Electrochemical Impedance Spectroscopy (EIS) measurements of a) undoped GaN (blue), GaN:Tb^{3+} (green), GaN:Eu^{3+} (red) and GaN:Eu^{3+},Tb^{3+}(orange); (Measurements were performed in a 0.1 M potassium phosphate electrolyte at an applied frequency of 10 Hz). The extrapolated curves at y = 0 correspond to the conduction band edges, because the materials exhibit a positive slope that is characteristic for an n-type semiconductor. The determined flatband potentials are provided in the inset.

Conclusions and Outlook

We demonstrated the doping of bulk GaN with europium and terbium and the combination of both resulting in intriguing luminescence properties of all three doped compounds (red, green, orange), rendering GaN:Eu,Tb as an prospective component in future light emitting diodes (LEDs). Our results highlight the opportunities for controlling functionality and luminescence properties of modern energy-efficient white light-emitting diodes and energy-efficient power electronic devices. Especially the closing of the "yellow gap" is a big step foreward.

The here presented methodology of doping bulk GaN applied may be generalized to create different dimensions of device architectures for LEDs, as the possibility to modify their charge transport properties by introducing dopant atoms has turned out for some time to be crucial for the performance of inorganic LEDs.

Experimental Section

Doping of GaN:

The doping of GaN (Chempur 99,999%) was realised by a successive combustion synthesis. Briefly, the respective metal chlorides ($MCl_3 \cdot 6\,H_2O$ or $M(NO_3)_3 \cdot 5H_2O$ (M= Eu, Tb), NH_4NO_3, Urea and H_2O were added to the sample. This mixture was put into an oven at 400-600° for 10 min. The doping, which is about 3-5%, has been checked via EDX measurements.

X-ray diffraction

Powder Diffraction: X-ray diffraction experiments on powder samples of GaN were performed on a STOE STADI P powder diffractometer in Debye-Scherrer geometry with Ge(111)-monochromatized Mo-$K\alpha_1$ radiation ($\lambda = 0.709026$ Å). The sample was enclosed in a glass capillary of 0.3 mm diameter.

EDX measurements

SEM was performed on a Zeiss Merlin microscope and for EDX we used a Quantax 400 system from Bruker.

Mott-Schottky (MS) measurements

Electrodes fabrication

GaN and the corresponding rare earth metal (RE, RE= Eu^{3+}, Tb^{3+} and Eu^{3+}/Tb^{3+}) doped semiconductor electrodes were prepared by electrophoretic deposition. Fluorine doped tin oxide (FTO) glass (2.2 mm thick, Sigma-Aldrich) was used as the substrate after sequentially ultrasonic cleaning with dilute nitric acid, acetone and ethanol for 15 min. The dispersion was prepared by mixing 5 mg iodine and 20 mg sample with 20 ml acetone, followed by treatment with ultrasounds. The electrodes were obtained after depositing the dispersed powder at 30 V and drying under ambient atmosphere.

Mott-Schottky (MS) measurements

The MS measurements were performed in an electrochemical cell using a potentiostat (Gamry instruments) operating in a three-electrode setup. The deposited samples on FTO, a 1 M Ag/AgCl electrode and a platinum wire were used as a working electrode, a reference electrode and a counter electrode, respectively. All MS data were recorded *vs.* $E_{1\,M\,Ag/AgCl}$ (V), which was subsequently converted with respect to E_{RHE} (V) according to the formula: E_{RHE} (V) = 0.235 + $E_{1\,M\,Ag/AgCl}$ + [0.059 × pH] (V) at 25 °C.

Luminescence

The luminescence spectra and quantum yield measurements were performed on a Fluorolog®-3 Horiba Jobin Yvon equipped with a TBX detector picosecond photon detection device and a 450 W xenon lamp.

Acknowledgement

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