

# **Traceless Click-Assisted Native Chemical Ligation Enabled by Protecting Dibenzocyclooctyne from Acid-Mediated Rearrangement with Copper(I)**

Patrick W. Erickson<sup>††</sup>, James M. Fulcher<sup>††§</sup>, Michael S. Kay<sup>†\*</sup>

<sup>†</sup>Department of Biochemistry, University of Utah School of Medicine, 15 North Medical Drive East, Room 4100, Salt Lake City, Utah 84112-5650, United States

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## ABSTRACT

Chemoselective ligation reactions, such as native chemical ligation (NCL), enable the assembly of synthetic peptides into proteins. However, the scope of proteins accessible to total chemical synthesis is limited by ligation efficiency. Sterically hindered thioesters and poorly soluble peptides can undergo incomplete ligations, leading to challenging purifications with low yields. This work describes a new method, Click-Assisted NCL (CAN), which overcomes these barriers. In CAN, peptides are modified with traceless “helping hand” lysine linkers that enable addition of dibenzocyclooctyne (DBCO) and azide handles for strain-promoted alkyne-azide cycloaddition (SPAAC) reactions. This cycloaddition templates the peptides to increase their effective concentration and greatly accelerate ligation kinetics. After ligation, mild hydroxylamine treatment tracelessly removes the linkers to afford the native ligated peptide. Although DBCO is incompatible with standard Fmoc solid-phase peptide synthesis (SPPS) due to an acid-mediated rearrangement that occurs during peptide cleavage, we demonstrate that copper(I) protects DBCO from this side reaction, enabling direct production of DBCO-containing synthetic peptides. Excitingly, low concentrations of triazole-linked model peptides reacted ~1,200-fold faster than predicted for non-templated control ligations, which also accumulated many side products due to the long reaction time. Using the *E. coli* ribosomal subunit L32 as a model protein, we further demonstrate that the SPAAC, ligation, desulfurization, and linker cleavage steps can be performed in a one-pot fashion. CAN will be useful for overcoming ligation challenges to expand the reach of chemical protein synthesis.

## INTRODUCTION

The combination of solid-phase peptide synthesis (SPPS)<sup>1</sup> with chemoselective ligation reactions enables the chemical synthesis of uniquely modified proteins. Of the chemoselective ligation strategies available, the Kent group's native chemical ligation (NCL) has proven to be highly robust and is widely used in the chemical synthesis of proteins.<sup>2-3</sup> With NCL, peptide segments bearing an N-terminal Cys and C-terminal thioester can be chemoselectively ligated to generate synthetic proteins. NCL has been extended beyond Cys junctions via free-radical desulfurization, a robust technique that allows the more common Ala, as well as other amino acids amenable to thiol surrogates, to be used as ligation junctions.<sup>4-5</sup> NCL has produced many useful synthetic proteins,<sup>6-7</sup> including targets for mirror-image drug discovery,<sup>8-12</sup> functional native and mirror-image enzymes,<sup>13-18</sup> mirror-image proteins for racemic crystallography,<sup>19-22</sup> and proteins containing complex post-translational modifications.<sup>23-26</sup>

However, NCL has several challenges that limit its scope in chemical protein synthesis (CPS).<sup>7,27</sup> First, NCL requires high peptide concentrations, typically  $\geq 1$  mM, at neutral pH for efficient ligation. Unfortunately, peptide segments are often too insoluble to reach these ideal concentrations, even in denaturing conditions.<sup>28</sup> While many groups have developed strategies to increase peptide solubility,<sup>28-31</sup> these do not overcome the inherent issue of requiring high concentrations.<sup>32</sup> Additionally, the availability of suitable Cys or Ala ligation junctions is limited for many synthesis projects, forcing the selection of suboptimal segments for NCL.<sup>11, 33</sup> Although alternative thiol-containing amino acids, such as penicillamine (thiol derivative of Val), can be used as junctions to increase access to alternative ligation strategies, these often suffer from

slow ligation rates.<sup>34-36</sup> In cases where limited suitable junctions exist, the use of sterically hindered thioesters may be required (e.g., Thr, Ile, and Val).<sup>36-38</sup> Even under ideal NCL conditions, sterically hindered thioesters suffer from long reaction times that often result in loss of peptide because of competing side reactions, creating complex HPLC purifications that further lower yields. The NCL rate is also dependent on the type of thioester used.<sup>39</sup> Most commonly, 4-mercaptophenylacetic acid (MPAA) is used for NCL due to its favorable transthioesterification kinetics.<sup>40</sup> However, MPAA has several disadvantages, including strong near-UV absorbance and potential co-elution with products in RP-HPLC. Several alternatives can be used in place of MPAA to avoid co-elution with peptide, though at the cost of slower NCL kinetics.<sup>41-43</sup>

Templated ligation strategies can overcome these major barriers by increasing the proximity of the reactive partners, the C-terminal thioester and the N-terminal thiol, thereby increasing their effective concentration (**Scheme 1A**). Templated ligations occur in an intramolecular fashion, converting reaction kinetics from 2<sup>nd</sup> to 1<sup>st</sup> order. An early example of such a proximity-driven reaction was demonstrated with Kemp's thiol capture.<sup>44-45</sup> Effectively, templated ligations are macrocyclization reactions, a strategy commonly found in the synthesis of cyclic natural products and other compounds.<sup>46-49</sup> Importantly, application of this strategy can improve the efficiency of slow thioesters and enable ligations at dramatically lower peptide concentrations, especially helpful for poorly soluble peptides.

Several templated ligation strategies have been developed, but specific caveats currently limit the utility of each strategy for CPS. Protein-templated ligations were the first demonstration of templated NCL, as peptides containing specific secondary

structures can be brought into close proximity via their non-covalent interactions.<sup>50-57</sup> While protein-templated ligation works well in a variety of contexts (e.g., self-replicating peptides, conformationally assisted protein ligations, and labeling of cell membrane proteins), the method is infeasible for most CPS applications, as peptides do not usually self-associate, and denaturants are typically required for peptide solubilization. Nucleic acids and peptide nucleic acids (PNAs) have also been used for templated ligations, as peptides conjugated with DNA or PNA can be brought into proximity via specific base-pairing interactions.<sup>58-63</sup> Nucleic acid-templated ligations complete quickly with only nM-pM template, but DNA/PNA linkers cannot be removed from ligated product. In addition, this method has only been used in non-denaturing buffers, predominantly with short peptide sequences (<20 residues) that do not fully represent the diversity of amino acid functional groups. Recently, several traceless templated ligation strategies have been developed. One such method uses peptides functionalized with UV-cleavable linkers attached to either PNA<sup>64</sup> or DNA<sup>65</sup> handles to enable templated ligation at low peptide concentrations. This method is not yet robust enough for routine CPS applications, as the linker syntheses are intricate, and the required UV irradiation and copper-catalyzed click steps may cause oxidative damage.<sup>66-68</sup> The Bode group has also designed a traceless templated ligation using streptavidin-binding linkers, which cleave upon amide bond formation.<sup>69</sup> Although this strategy is promising, complex linker synthesis is currently needed, and the methodology has not yet been demonstrated on peptides.

### **A New Strategy for Traceless Templated Ligations**

We propose that the following fundamental elements are required for a traceless templated ligation strategy to be broadly applicable in CPS:

- 1) Simple and robust bioorthogonal linker synthesis
- 2) Bioorthogonal chemistry can be performed in typical peptide-solubilizing denaturants (e.g., 6 M guanidine)
- 3) Bioorthogonal linkers are stable to commonly encountered CPS conditions (e.g., resin cleavage, ligation, and desulfurization)
- 4) Removal of bioorthogonal linkers is chemoselective, traceless, and efficient
- 5) Flexible positioning of bioorthogonal linkers within peptide sequences (e.g., placement not restricted to termini or infrequent amino acids)

Here, we describe Click-Assisted NCL (CAN), which meets all these criteria by combining a traceless “helping hand” Lys linker (Ddap)<sup>70</sup> with the bioorthogonal strained-click pair dibenzocyclooctyne (DBCO) and azide (**Scheme 1B**). In the CAN strategy, peptide segments are first coupled with a Ddap linker on a Lys sidechain on-resin during SPPS before functionalization with either DBCO or azide. After resin cleavage and HPLC purification, the DBCO/azide-bearing peptides are combined for SPAAC. The resultant triazole linkage effectively templates the C-terminal thioester and N-terminal thiol for ligation, increasing their effective concentration and allowing NCL to proceed with high efficiency. After ligation, mild hydroxylamine treatment chemoselectively removes the templating linkers,<sup>70</sup> affording the native ligated peptide. Using model peptides, we show that the CAN strategy is compatible with all canonical

amino acids (although Met is slightly more prone to oxidation). CAN also greatly accelerates ligation in the context of sterically hindered thioesters. Furthermore, these demonstrations use methyl thioglycolate (MTG) as a slower, relative to MPAA, but desulfurization-compatible thiol additive for NCL.<sup>71</sup> Finally, we illustrate the utility of CAN for CPS by performing a strain-promoted click reaction, NCL, desulfurization, and linker cleavage to produce the *E. coli* 50S ribosomal subunit L32 in one pot.

## RESULTS AND DISCUSSION

### Synthesis of DBCO Peptides via Fmoc-SPPS

In order for CAN to be a robust CPS technique, peptides need to be templated together through a bioorthogonal reaction that will not cause unintended side reactions during conjugation. For this purpose, we chose to use strain-promoted alkyne-azide cycloaddition (SPAAC)<sup>72-73</sup> over copper-catalyzed azide-alkyne cycloaddition (CuAAC)<sup>74-75</sup> because SPAAC avoids potential copper-catalyzed oxidation during the click reaction.<sup>67-68</sup> In addition, we wanted to perform one-pot templating and NCL and expected that the presence of copper within NCL conditions may cause side-product formation (e.g., during *in situ* thioester formation<sup>76</sup>).

The first obstacle to developing CAN was generating a robust, SPPS-compatible method for installing azide and DBCO cycloaddition partners. While azide-containing peptides can routinely be synthesized via SPPS,<sup>77</sup> DBCO peptides have only been synthesized using post-cleavage couplings, such as thiol-maleimide<sup>78-80</sup> or amine-NHS ester.<sup>81</sup> These post-cleavage workarounds are necessary because DBCO undergoes an inactivating rearrangement under the strongly acidic conditions, 95% trifluoroacetic acid (TFA), used for peptide deprotection and cleavage from the resin.<sup>79-80, 82-83</sup> This rearrangement has been demonstrated in DBCO and related biarylazacyclooctynones and is thought to result from an acid-catalyzed 5-*endo-dig* cycloisomerization.<sup>83-84</sup> Longer peptides are unlikely to have a single primary amine or thiol for post-cleavage couplings, so DBCO needs to be incorporated during Fmoc-SPPS. While DBCO can withstand lower TFA concentrations (<30%),<sup>82, 85-86</sup> most commonly used protecting groups and resins require higher TFA for efficient removal. We set out to investigate the

stability of DBCO in standard TFA cleavage conditions using the model peptides H-C(*S*tBu)GK(DBCO)ENTWY-R (R = NH<sub>2</sub> **1a**, OH **1b**, or NHHN<sub>2</sub> **1c**; see **Table S1** for all peptide sequences) and Ac-RRRYSTEVEK(DBCO)NV-NHHN<sub>2</sub> **2a** (DBCO coupled to Lys side chains and Cys protected with *S*tBu to prevent potential thiol-yne reactions).<sup>87</sup> Our initial attempts to synthesize **1a** via Fmoc-SPPS using a standard TFA peptide cleavage cocktail failed due to formation of an undetermined DBCO-related degradation product (**Figure S1**). **2a** was found to have the correct molecular weight via LC/MS analysis (**Figure S2**). However, the DBCO moiety was unreactive when incubated with 6-azidohexanoic acid (**Figure S3**), suggesting it had undergone acid-catalyzed 5-*endo-dig* cycloisomerization. We did observe formation of the triazole through SPAAC before resin cleavage (**Figure S4**; see **Table S2** for symbols used to denote common modifications such as triazole), indicating that DBCO degraded during cleavage and not during on-resin coupling.

The Hosoya group previously reported that a copper(I) salt, tetrakis(acetonitrile)copper(I) tetrafluoroborate or (MeCN)<sub>4</sub>CuBF<sub>4</sub>, could reversibly protect cyclooctynes from reacting with azides.<sup>88-89</sup> This strategy builds on the finding that various metals can form complexes with cycloalkyne rings,<sup>90</sup> which led us to examine copper as an additive to prevent the acid-mediated 5-*endo-dig* cycloisomerization of DBCO. Guided by the methodology reported by the Hosoya group, initial attempts at protecting peptides **1a** and **2a** were performed by treating peptide resins with ~3 equiv. (MeCN)<sub>4</sub>CuBF<sub>4</sub> in dimethylformamide (DMF) for 1 h. The resin was then washed with DMF and dichloromethane (DCM) before cleavage in the standard cocktail at rt for 3 h. Standard ether precipitation and air drying produced the

crude DBCO peptides. As envisioned, a transient DBCO-Cu complex was protected from acid-promoted ring rearrangement, and the resultant DBCO peptide product **1a** was able to react with excess 6-azidohexanoic acid (**Figure S5**). However, we discovered that  $(\text{MeCN})_4\text{CuBF}_4$  caused premature cleavage of peptides assembled on 2-chlorotrityl chloride resin. Additionally, the Cu(I) salt reacted with the C-terminal hydrazide-containing peptide **2a** to form undesired side products (**Figure S6**). Switching the solvents used to dissolve  $(\text{MeCN})_4\text{CuBF}_4$  (DCM, N-methylpyrrolidone, methanol, and MeCN) did not overcome these side reactions (**Figures S7-11**).

To address this issue, we next tested the direct addition of ~1.5 equiv.  $(\text{MeCN})_4\text{CuBF}_4$  to the peptide resin as a dry powder, immediately followed by addition of the TFA cleavage cocktail and incubation at rt for 3 h (**Figure 1**). Again, standard ether precipitation and air drying afforded the crude DBCO peptide **1a**. Incubation of **1a** with 6-azidohexanoic acid demonstrated a successful SPAAC (**Figures 1 and S12**). This simplified protection method was also successful in producing peptide **1b** with reactive DBCO (**Figure S13**). However, while this method successfully generated hydrazide-containing **2a** and **1c**, side reactions occurred when attempting to react crude **2a** or **1c** with 6-azidohexanoic acid (**Figures S14-15**). Copper test strips indicated that Cu was only partially removed during ether precipitation and was likely causing the undesired side reactions during SPAAC. Gratifyingly, HPLC purification appeared to remove the remaining Cu, as pure **1c** did not contain Cu (copper test strips), and the peptide successfully coupled with 6-azidohexanoic acid to afford the triazole with minimal side products (**Figures S16-19**). Inductively coupled plasma mass spectrometry (ICP-MS) confirmed removal of Cu, as several HPLC-purified DBCO

peptides contained the same background Cu levels as peptides that were never exposed to the Cu(I) salt (**Table S3**).

### **Robustness of the DBCO Protection Method**

Concerned with the potential of Cu(I) salts to oxidize certain amino acids, particularly Met and His,<sup>67</sup> we examined the compatibility of the DBCO protection strategy using the more diverse model peptide H-C(S*t*Bu)DEAFGHK(DBCO)LMNPQRSTVWYK-NH<sub>2</sub> **3a** (and without DBCO, **3b**), which contains every canonical amino acid. We found that thioanisole and 1,2-ethanedithiol, two commonly used TFA cleavage scavengers that prevent Cys alkylation, diminished the protective properties of the Cu(I) additive with **3a** (**Figures S20-25**). Furthermore, we noted that ammonium iodide (NH<sub>4</sub>I), a cleavage additive used to reverse Met oxidation,<sup>91-92</sup> was also capable of completely removing the S*t*Bu-protecting group on **3a** and **3b** (**Figures S26-28**). When **3b** was cleaved in the presence or absence of ~3 equiv. (MeCN)<sub>4</sub>CuBF<sub>4</sub>, the analytical HPLC and LC/MS chromatograms were similar (**Figures 2A** and **S29-30**). However, the presumed oxidation side-product (+16 Da) increased from 6% to 12.5% in Cu(I)-treated **3b** (**Figure 2A**). Using trypsin digestion and high-resolution LC/MS/MS on **3b** ± Cu(I) during cleavage, Met was identified as the oxidation-prone residue (**Figure 2B**). Furthermore, when the Cu(I) salt was used in the cleavage of model peptides H-C(S*t*Bu)GKENTWYX-NH<sub>2</sub> **4a** (X=H) and **4b** (X=M), oxidation was only observed in the Met-containing peptide **4b** (**Figures S31-32**). While increased Met oxidation is a potential limitation of the DBCO protection method, all other canonical amino acids are

fully compatible. In addition, Met is one of the least abundant amino acids and can typically be substituted with an isosteric norleucine (Nle) residue without affecting protein activity and folding.<sup>42, 93-94</sup>

We next investigated the ideal concentration of the  $(\text{MeCN})_4\text{CuBF}_4$  additive. Considering the objective of these experiments was to determine the amount of active DBCO following cleavage across a range of  $(\text{MeCN})_4\text{CuBF}_4$  equivalents, the analysis was simplified by synthesizing **3a** with a Met-to-Nle substitution to give oxidation-resistant **3c** (**Figure 2C**). **3c** was incubated with the indicated amounts of  $(\text{MeCN})_4\text{CuBF}_4$  in **Figure 2C** through standard cleavage conditions for 3 h before ether precipitation and air drying. Crude **3c** was then dissolved in HPLC buffer (0.1% TFA, 18% MeCN) and coupled with excess 6-azidohexanoic acid for 1 h at 37°C to determine the amount of reactive DBCO present (**Figures 2C** and **S33-46**). DBCO protection was observed to increase until 5 equiv.  $(\text{MeCN})_4\text{CuBF}_4$ . Surprisingly, 10 and 50 equiv. increased the amount of DBCO that failed to click with azide. This loss of reactivity may be attributed to excessive Cu(I) salts that are incompletely removed following ether precipitation, which could form a complex with the DBCO peptide, thus precluding efficient cycloaddition reactivity (**Figures S39** and **S46**). Based on these optimizations, the use of 5 equiv. of  $(\text{MeCN})_4\text{CuBF}_4$  was determined to be ideal for preventing the rearrangement of DBCO-containing peptides through standard cleavage conditions.

### Click-Assisted NCL with Model Peptides

With a DBCO protection strategy in hand, we next investigated Click-Assisted NCL (CAN). We substituted the commonly used 4-mercaptophenylacetic acid (MPAA)

thiol for methyl thioglycolate (MTG) during NCL, as MTG elutes in the void during RP-HPLC, avoiding co-elution with peptide. Additionally, MTG is compatible with one-pot desulfurizations.<sup>71</sup> Although MTG has slower ligation kinetics compared to MPAA,<sup>71</sup> we hypothesized that the effective concentration enhancement of the CAN strategy would render the choice of thiol additive less impactful. The non-templated control peptides STEVE (**5a**) and KENT (**6a**) were synthesized without SPAAC linkers to determine their in-solution NCL rate under representative conditions (**Figure 3A**). **5a** was also designed with a C-terminal Val thioester, since they are known to be among the slowest in NCL due to steric hindrance.<sup>37-38</sup> After 48 h, the ligation reaction was incomplete and several side products had formed, with only 29% of the total HPLC peak area corresponding to ligated product **7** (**Figures 3B** and **S47-52**).

As a proof-of-concept for CAN, we then synthesized model peptides **5b** and **6b**, each functionalized with the traceless Lys linker Ddap,<sup>70</sup> a single Lys (to enhance solubility), and azide or DBCO as described in **Scheme 1B**. Equal amounts of purified **5b** and **6b** were dissolved together in 6 M GnHCl at pH 3 to undergo SPAAC (**Figures 4** and **S53-58**). Importantly, to the best of our knowledge, this is the first demonstration of SPAAC in denaturing conditions using guanidine. HPLC analysis revealed the presence of two triazole cycloadduct peptides (both referred to as **8**). In this instance, the two distinguishable isobaric peaks are attributed to the two anticipated triazole regioisomers (**Figures 4** and **S57**). Using comparable conditions to the non-templated reaction, the C-terminal hydrazide was converted to an MTG thioester *in situ* prior to performing CAN (0.5 mM **8**, **Figure 4**). Excitingly, the ligation to form **9** was complete at 4 h (**Figures 4** and **S59-60**), and the relative peak area of **9** from this CAN reaction was much higher

(85%) compared to the non-templated control NCL (29%), attributable in part to significantly diminished side products (**Figure 4**). We then cleaved the Ddap linkers of **9** in one pot via the addition of hydroxylamine to ~1 M at pH 6.75 for 2 h (**Figures 4** and **S61**). The native, ligated product without Ddap linkers was easily purified by HPLC to afford **7** (**Figures 4** and **S62**). CAN with **8** was then repeated for a more precise kinetic analysis (**Figures 5** and **S63-65**).

To investigate the impact of linker length on the CAN reaction rate, peptides **5c** and **6c** were each functionalized with Ddap, a single Lys, and a PEG<sub>8</sub> linker before addition of their respective SPAAC partners. The PEG<sub>8</sub> spacers double the total extended distance between the triazole-linked peptides. Using the same conditions as with **5b** and **6b**, the PEG<sub>8</sub> peptides first underwent SPAAC to give triazole-linked **10** (2.5 mM both peptides at 37°C for 2 h, **Figures S66-71**) before performing CAN at 0.5 mM to give ligated product **11** (**Figures S72-73**). This CAN with PEG<sub>8</sub> spacers was ~3-fold slower than the non-PEGylated CAN (**Figure 5**).

One of the more attractive uses of CAN, and templated ligations in general, is the ability to perform ligations at lower peptide concentrations (i.e., the triazole-linkage renders the NCL an intramolecular reaction). To validate this property, we repeated CAN with **8** and **10** ( $\pm$ PEG<sub>8</sub>) at a 10-fold lower concentration (0.05 mM). We observed that a significant amount of the N-terminal Cys peptide was being capped (initially +74 Da, **Figures S74-77**). The mass of this side product suggested that modification was the result of acylation from an MTG acyl donor. This acylation occurs negligibly in the 0.5 mM CAN reactions, indicating that acylation occurs on a small amount of peptide that only becomes problematic at lower peptide concentrations. Therefore, we surmised

that reducing the MTG concentration would reduce this side reaction. A ten-fold reduction in MTG concentration without proportional reduction of TCEP concentration, however, led to significant Cys desulfurization (**Figures S78-80**). Fortunately, a proportionate reduction of TCEP concentration eliminated this desulfurization. Using these conditions, CAN kinetics were similar for both the 0.05 and 0.5 mM reactions (**Figures 5** and **S81-84**). Compared to the non-templated NCL (0.4 mM **5a** and 0.5 mM **6a**), the 0.05 mM **8** CAN reached completion ~12x faster (4 h for CAN vs. 48 h for non-templated NCL). Based on the 2<sup>nd</sup>-order rate law, a 10-fold reduction in concentration of both peptides in the non-templated NCL would slow the reaction 100-fold, which is too slow to measure and would likely be dominated by side reactions. Therefore, the 0.05 mM **8** CAN reached completion ~1,200-fold faster than expected for the non-templated NCL at the same concentration.

### **Click-Assisted NCL of the *E. coli* 50S ribosomal subunit L32**

To demonstrate the general utility of CAN in CPS, we pursued a representative protein target – the *E. coli* 50S ribosomal subunit L32 (**Figure 6A**). L32 is an ideal CAN choice for several reasons. First, the 56-residue L32 contains a reasonably placed Thr-Ala ligation junction (**Figure 6A**). Similar to the Val thioesters used above, Thr thioesters are sterically hindered and exhibit slow NCL kinetics.<sup>37-38</sup> Furthermore, L32 lacks any native Cys residues, so desulfurization must be performed after NCL to produce native L32. This protein allows us to test four reactions (SPAAC, CAN, desulfurization, and Ddap linker cleavage) in a one-pot fashion (**Scheme 2**). Finally, L32-C (45-57) conveniently bears a Lys residue near its N-terminus, and L32-N (2-44,

removed initiator Met) contains Lys residues both proximal and distal to the C-terminal thioester, allowing us to investigate the impact of Ddap linker placement on CAN reaction rate (**Scheme 2**). Following the same procedure for the model peptide control reactions, we first performed a representative control NCL between peptide **12a** (L32-N 2-44) and **13a** (L32-C 45-57) that was nearly complete after 48 h (**Figures 6B, 6C, and S85-90**), with 70% of the peak area corresponding to ligated **14a** (**Figure 6C**). Notably, a significant amount of **12a** was lost due to thioester hydrolysis.

To investigate CAN for L32, L32-N (2-44) was functionalized with Ddap followed by a single Lys and 6-azidohexanoic acid at Lys<sup>37</sup> (closer distance, **12b**, L32-close) or Lys<sup>12</sup> (farther distance, **12c**, L32-far) (**Scheme 2**). Nle was substituted for the sole Met<sup>15</sup>, as cleavage cocktails containing NH<sub>4</sub>I additive were found to cause azide degradation (**Figure S91**), and we desired to avoid Met oxidation during peptide cleavage. L32-C (45-57) was functionalized at Lys<sup>53</sup> with Ddap followed by a single Lys and DBCO-C6 to give **13b**. We first investigated CAN of L32-close by clicking **12b** with **13b** in 6 M GnHCl at pH 3 to give **15** (**Figures 7 and S92-97**), which has a distance of 15 AA between Lys Ddap linkers. CAN was then performed using the same ligation conditions as the non-templated L32 ligation (0.3 mM of triazole-linked **15**, **Figure 7**). **15** ligated to completion within 2 h, much faster than the 48 h needed for the non-templated ligation. Furthermore, 88% of the peak area corresponded to ligated product **16** (**Figures 7 and S98-99**). Finally, we performed desulfurization to give **17** and Ddap linker cleavage to produce full-length L32 **14b** in a one-pot synthesis (**Figures 7 and S100-102**).

To determine how placing Ddap linkers further from ligation junctions affects CAN rate, we investigated CAN of L32-far by performing SPAAC with **12c** and **13b** in 6 M GnHCl at pH 3 (1.7 mM both peptides at 37°C for 4 h, **Figures S103-106**). L32-far CAN was then tested with 0.3 mM of the triazole-linked **18** (distance of 40 AA between Ddap linkers) under the same ligation conditions as L32-close. The L32-far CAN was complete after 6 h to give peptide **19**, compared to the 48 h non-templated ligation (**Figures 8** and **S107-108**). The 1<sup>st</sup> order rate constant for L32-close (98-membered macrocycle) was 4.5-fold faster than for L32-far (173-membered macrocycle) (**Figure 8**).

The synthesis of L32 using CAN highlights some of the key features of this technique. First, implementing CAN into a synthesis project does not add extra intermediate HPLC purifications. Second, the distance between linker sites on the two peptides is quite flexible – the L32 ligation was separated by 40 AA yet still achieved a significant improvement in ligation rate relative to the non-templated NCL. Finally, CAN dramatically increases the throughput of CPS, as CAN enabled full-length, L32 to be easily HPLC purified in ~25 h (including SPAAC, ligation, desulfurization, and linker removal), whereas the non-templated L32 NCL requires 48 h to reach completion and a subsequent desulfurization step.

## CONCLUSIONS

For the first time, DBCO peptides were directly synthesized via Fmoc-SPPS by using  $(\text{MeCN})_4\text{CuBF}_4$  to protect DBCO from acid-mediated rearrangement during standard acid cleavage. This DBCO protection strategy was shown to be robust, as DBCO peptides with various C-terminal functional groups were successfully prepared on different resins. Furthermore, all canonical amino acids were compatible, although Met is slightly oxidation-prone. This DBCO protection method enabled development of Click-Assisted NCL (CAN), which is a traceless templated ligation strategy that can be used to dramatically increase NCL efficiency. Using the STEVE and KENT model peptides, CAN was estimated to be approximately 1,200-fold faster than a non-templated NCL at 0.05 mM. CAN also improved yield and reduced side products compared to non-templated NCL. The utility of CAN in CPS was also demonstrated through the preparation of the *E. coli* 50S ribosomal subunit L32. Compared to non-templated NCL, CAN enabled a faster ligation with one-pot SPAAC, ligation, desulfurization, and linker removal. Finally, the CAN rate was shown to modestly drop with template distance (i.e., increased macrocycle size), maintaining significant NCL rate enhancements even with distant attachment points – up to 40 AA separation demonstrated.

This new method to directly synthesize DBCO peptides via Fmoc-SPPS will be a useful tool for many peptide chemists. For example, peptide and protein nanoparticle conjugates have exciting biomedical applications (e.g., drug delivery and *in vivo* imaging), but there is a need to develop additional methods enabling site-specific incorporation of peptides onto nanoparticles.<sup>95</sup> DBCO peptides can be easily conjugated

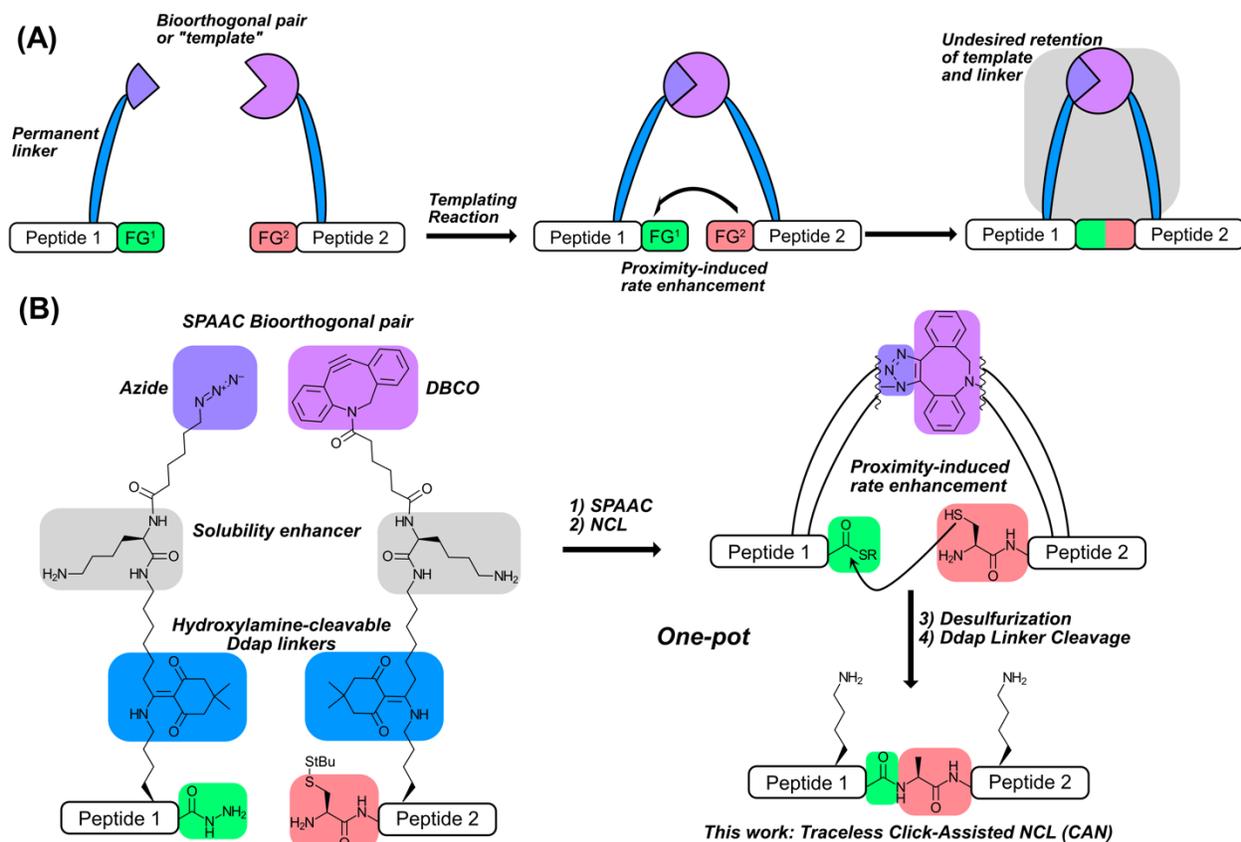
to azide-functionalized nanoparticles,<sup>82</sup> and this method will enable larger and more sequence-diverse DBCO peptides to be incorporated. Additionally, several methods exist to incorporate azides into proteins via engineered tRNA synthetases or Met replacement with azide-containing analogs,<sup>96-100</sup> enabling DBCO peptides to tag recombinant azido-proteins for various applications (e.g., proteome labeling and imaging). Although cyclooctynes can be genetically encoded into proteins,<sup>98, 100-104</sup> these methods are more challenging to implement compared to azide.<sup>98</sup> The finding that (MeCN)<sub>4</sub>CuBF<sub>4</sub> prevents acid-catalyzed 5-*endo-dig* cycloisomerization in DBCO, and perhaps other alkyne-derived isomerization reactions,<sup>105</sup> is an important observation. For example, several groups have reported synthetic limitations when working with various strained alkynes due to the 5-*endo-dig* cycloisomerization,<sup>82-86</sup> and protecting these strained alkynes with (MeCN)<sub>4</sub>CuBF<sub>4</sub> should increase the number of synthetic strategies available. This (MeCN)<sub>4</sub>CuBF<sub>4</sub> protection may also be effective in protecting DBCO peptides after resin cleavage and ether precipitation, as suggested by the 10 and 50 equiv. (MeCN)<sub>4</sub>CuBF<sub>4</sub> data shown in **Figure 2C**. The larger quantities of (MeCN)<sub>4</sub>CuBF<sub>4</sub> in the cleavage solution appear to allow DBCO to remain protected after resin cleavage, as 6-azidohexanoic acid did not efficiently react with DBCO in aqueous conditions. Liberation of copper from the DBCO peptides could be accomplished with aqueous ammonia, EDTA, or metal scavengers, as reported by the Hosoya group.<sup>88-89</sup> As other metal salts have been demonstrated to complex with cycloalkyne rings,<sup>90, 106</sup> these may also be able to protect against acid-mediated degradation.

We expect CAN to find use as a tool in CPS for efficiently ligating peptides suffering from sluggish thioesters and/or poor solubility. Interestingly, since lower-

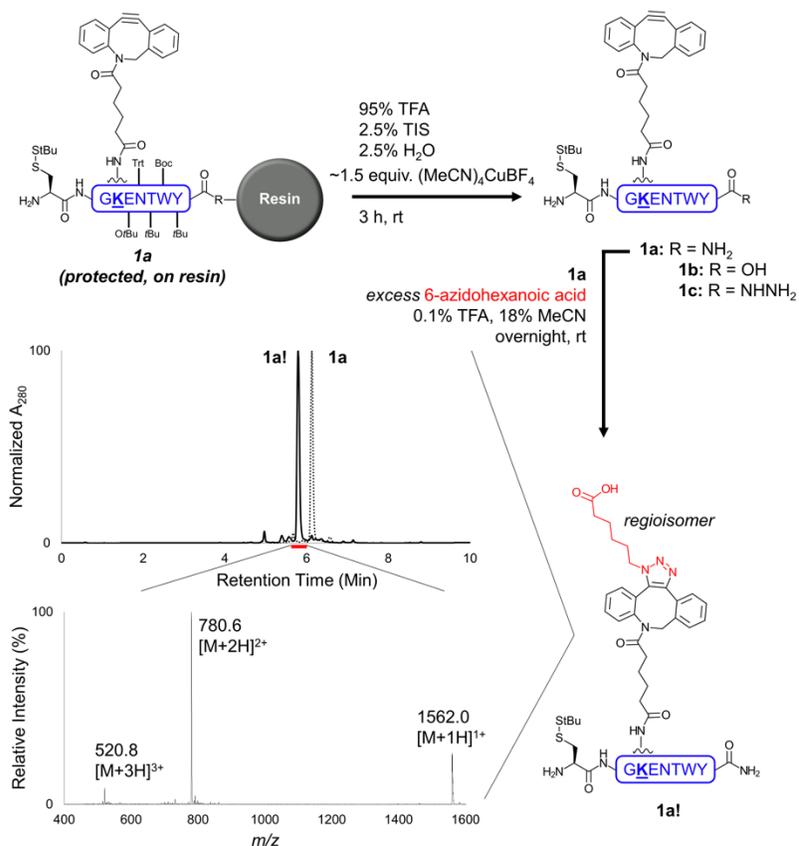
concentration CAN is expected to maintain its ligation rate, the relative benefit of CAN over solution NCL will increase at further reduced peptide concentrations (e.g., compared to NCL at 500 nM, STEVE/KENT CAN is predicted to be ~12,000,000-fold faster). The current CAN methodology is expected to be applicable to many CPS strategies, especially since the reaction rate enhancement is relatively insensitive to linker attachment points. For example, 64% of all possible Cys and Ala ligation junctions in the *E. coli* 50S and 30S ribosomal proteins are CAN-compatible (Lys within 20 AA's of the junction on both the N- and C-terminal peptides). CAN will also be particularly useful for sluggish ligations using unnaturally thiolated amino acids that enable NCL outside of Cys, such as thiolated valine.<sup>34-35</sup> Furthermore, CAN should be compatible with other peptide linker strategies,<sup>30-31</sup> due to the simplicity of preparing both azide and DBCO peptides.

Future improvements to the CAN methodology have the potential to dramatically streamline CPS projects. For example, CAN currently requires relatively high peptide concentrations to efficiently perform SPAAC (typically >50  $\mu\text{M}$ )<sup>32</sup> before NCL. Although we limited our demonstration to aqueous conditions with denaturant, it is worth noting that SPAAC is much more versatile than NCL since SPAAC can be performed in solvent mixtures more conducive for solubilizing peptides, such as MeCN/H<sub>2</sub>O that is compatible with being directly lyophilized.<sup>107-108</sup> Furthermore, unlike NCL, SPAAC does not suffer from competition with thioester hydrolysis. Future iterations of CAN may employ faster bioorthogonal chemistries for linking the peptides, as long as the chemistries are SPPS compatible. In fact, such bioorthogonal reactions that are also mutually orthogonal to DBCO and azide SPAAC would potentially enable multiple

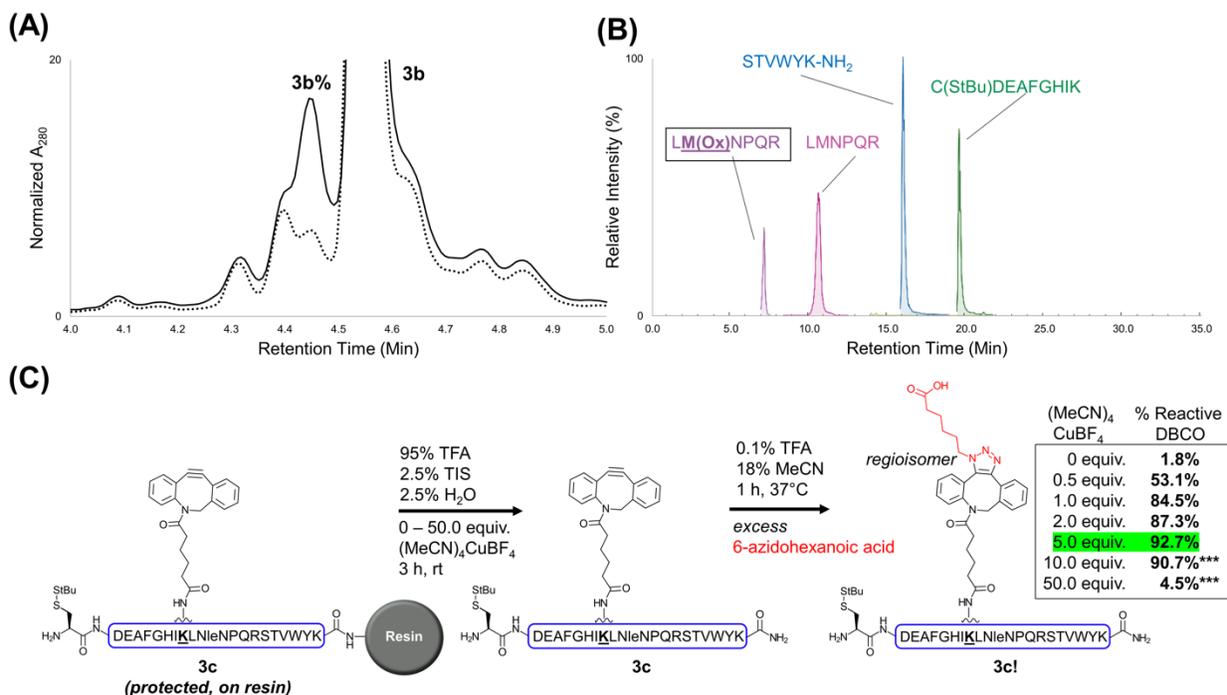
ligations to be performed in one pot via a CAN strategy. Such a combination would greatly increase yields in CPS by eliminating the need for intermediate HPLC purifications after each ligation. Finally, CAN parameters can be incorporated into an improved version of our “Automated Ligator” (Aligator) program,<sup>109</sup> to predict segments that will benefit most from CAN.



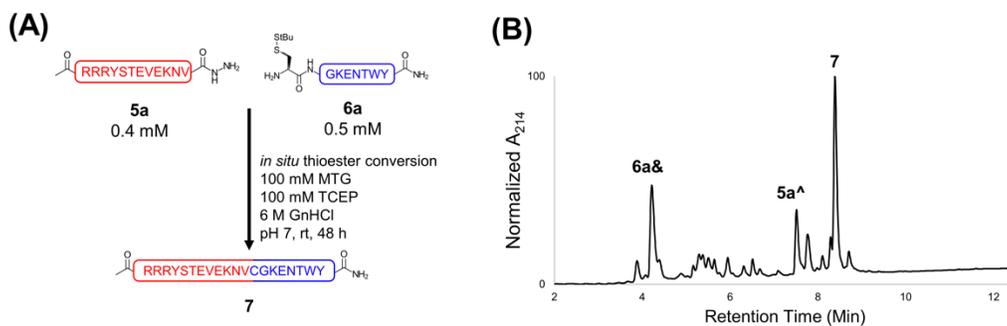
**Scheme 1:** Overview of templated peptide ligation strategies. (A) Peptides functionalized with bioorthogonal groups are templated together, increasing the effective concentration of two functional groups (FG<sup>1</sup> and FG<sup>2</sup>). While this strategy increases reaction efficiency, the bioorthogonal groups remain bound to the ligated peptide, leaving “scars” that are unacceptable for most CPS applications. (B) With Click-Assisted NCL (CAN), peptides functionalized with traceless Lys linkers (Ddap) and either DBCO or azide are first templated together via SPAAC. These templated peptides then undergo a proximity-enhanced NCL, desulfurization (if desired), and gentle hydroxylamine treatment to cleave the Ddap linker and tracelessly afford the native peptide. Importantly, all steps can be performed in one pot.



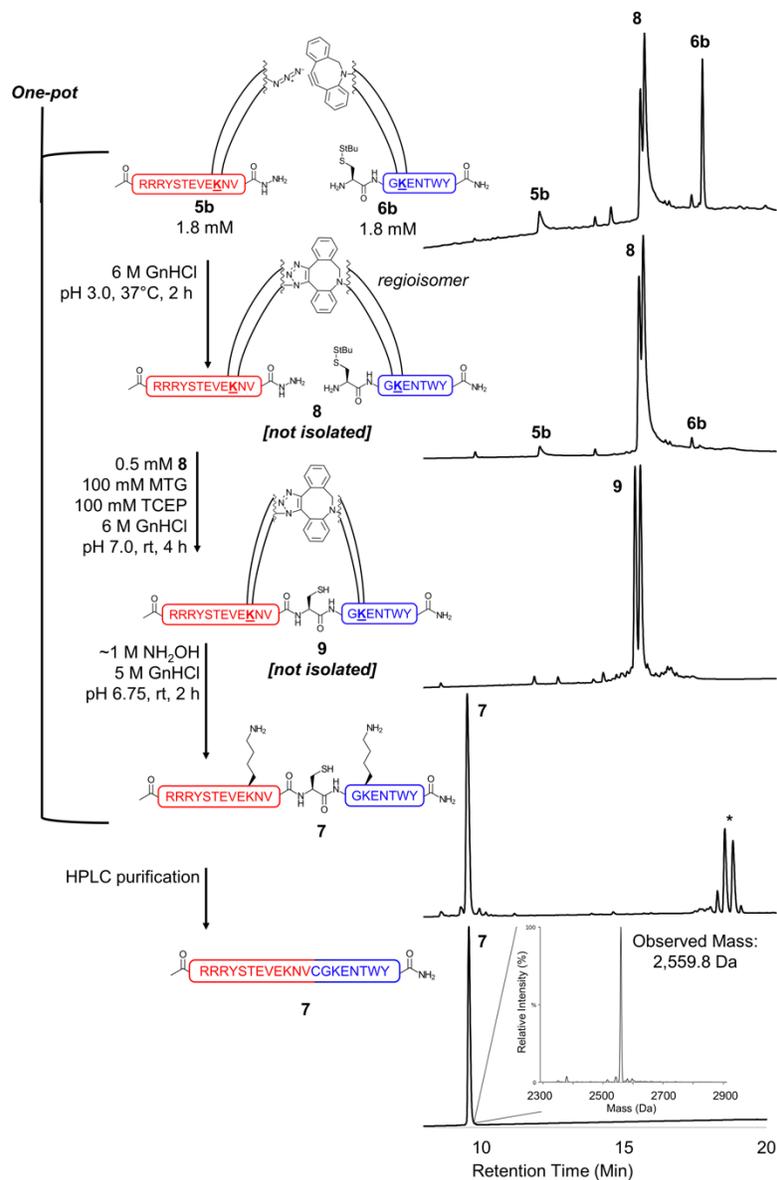
**Figure 1:** Protection of DBCO peptide **1a** from acid-mediated degradation using (MeCN)<sub>4</sub>CuBF<sub>4</sub> in standard peptide cleavage conditions. This protection strategy generated a relatively clean crude peptide (dashed line in the LC/MS chromatogram) corresponding to the expected mass of **1a** (observed mass: 1,401.9 Da, calculated average mass: 1,402.6 Da, **Figure S12**). This crude peptide successfully underwent SPAAC when incubated with excess 6-azidohexanoic acid, as the expected triazole formed (solid line in the LC/MS chromatogram, **1a!**; observed mass: 1,559.9 Da, calculated average mass: 1,559.8 Da). LC/MS Method C was used.



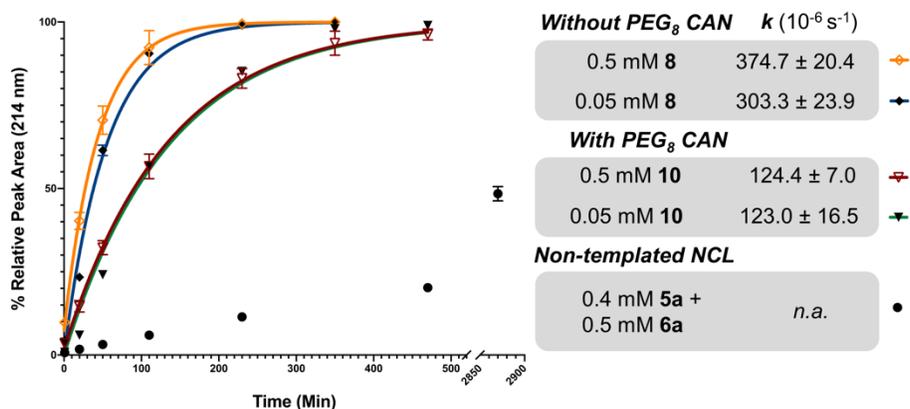
**Figure 2:** Assessment of the  $(\text{MeCN})_4\text{CuBF}_4$  DBCO protection strategy's robustness. Peptide **3b** (all canonical residues represented) was used to determine the strategy's oxidation potential, and peptide **3c** (Met-to-Nle variant of **3b**) was used to establish effective amounts of Cu(I) salt to use for DBCO protection. (A) Overlaid LC/MS chromatogram showing **3b** cleaved with (solid line) or without (dashed line)  $\sim 3$  equiv.  $(\text{MeCN})_4\text{CuBF}_4$ . Integration of the peak areas suggested oxidation (**3b%**) increased from 6% to 12.5% with Cu(I) salt addition to the standard cleavage cocktail. LC/MS Method C was used, and the individual chromatograms are shown in **Figure S29**. (B) Trypsin digestion and LC/MS/MS analysis of **3b** revealed Met as the oxidation-susceptible residue. The **3b** sequence is color-coded by trypsin-digested fragments and matches the appropriate ion signal identified through MS/MS analysis. LM(Ox)NPQR (far left peak, purple) indicates peptide containing oxidized Met. (C) Cleavage of **3c** with different equiv. of  $(\text{MeCN})_4\text{CuBF}_4$  indicated that 5.0 equiv. provides the most effective DBCO protection. After performing SPAAC on crude **3c**,  $A_{280}$  peak areas of reacted (**3c!**) and unreacted peptide were integrated to calculate % reactive DBCO. \*\*\*Indicates higher equiv. of  $(\text{MeCN})_4\text{CuBF}_4$  that were found to complex with the DBCO peptide, preventing reaction with azide after cleavage. Although the % reactive DBCO is lower, this decrease is not due to 5-endo-dig-cycloisomerization. LC/MS chromatograms of crude **3c** (before and after SPAAC) are shown in **Figures S33-46**.



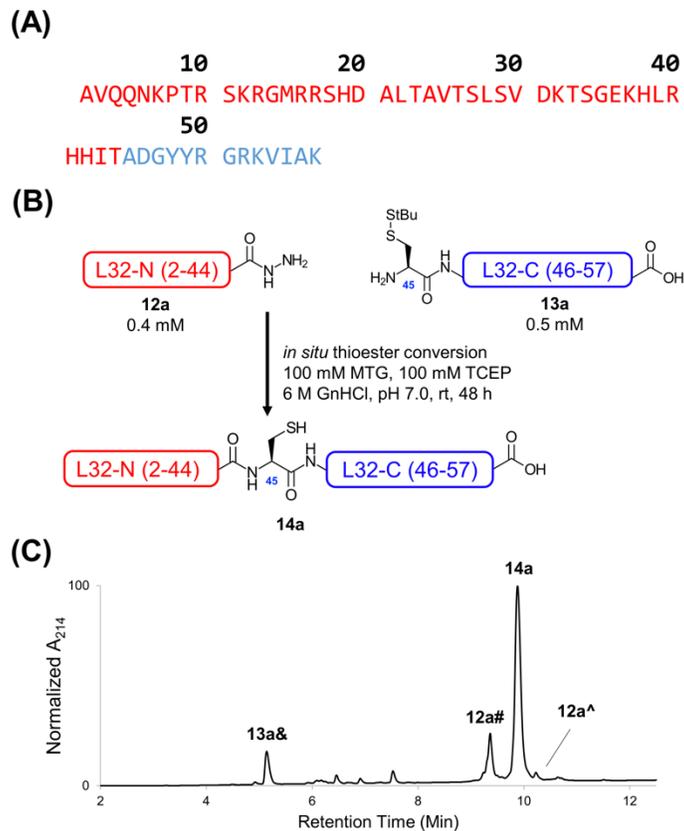
**Figure 3:** Non-templated native chemical ligation between purified **5a** and **6a**. (A) Simplified schematic of the ligation. Note that MTG was allowed to form the thioester for 10 min prior to TCEP addition, as shown in the more detailed scheme (**Scheme S1**). (B) HPLC chromatogram of the ligation endpoint (48 h). **6a&** refers to *StBu*-deprotected **6a**, and **5a^** refers to **5a** containing a C-terminal MTG thioester. After 48 h (from MTG addition), the reaction was incomplete, and numerous side products were observed. Analytical Method B was used. HPLC and LC/MS chromatograms of additional time points for this ligation are shown in **Figures S51-52**.



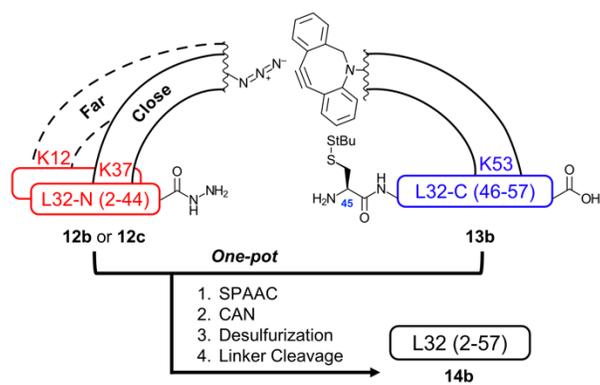
**Figure 4:** One-pot SPAAC, CAN, and linker cleavage. **5b** and **6b** were first templated together via 2 h SPAAC to produce triazole-linked **8**. CAN was then performed on **8** to efficiently generate **9** after only 4 h. A 2 h linker cleavage of **9** resulted in native **7**, which was easily HPLC purified. The HPLC chromatograms show, in descending order, 0 and 2 h SPAAC, 4 h CAN (after MTG addition), 2 h linker cleavage, and purified **7**. The double peaks observed after SPAAC and CAN are expected, as the triazole forms two regioisomers. \* indicates cleaved Ddap linker peaks. There are four Ddap linker peaks due to the triazole regioisomer and a second nucleophilic addition via hydroxylamine on the Ddap dimedone ring. Note that unreacted STEVE and KENT peptides observed after linker cleavage do not stick to the column with the HPLC gradient used here (**Figure S61** shows these unreacted peptides). The mass spectrum of purified **7** shows the correct full-length STEVEKENT peptide was generated (observed mass: 2,559.8 Da, calculated average mass: 2,559.9 Da). A more detailed one-pot reaction schematic is shown in **Scheme S2**. Analytical Method C and LC/MS Method C were used. HPLC and LC/MS chromatograms of additional time points are shown in **Figures S57-62**.



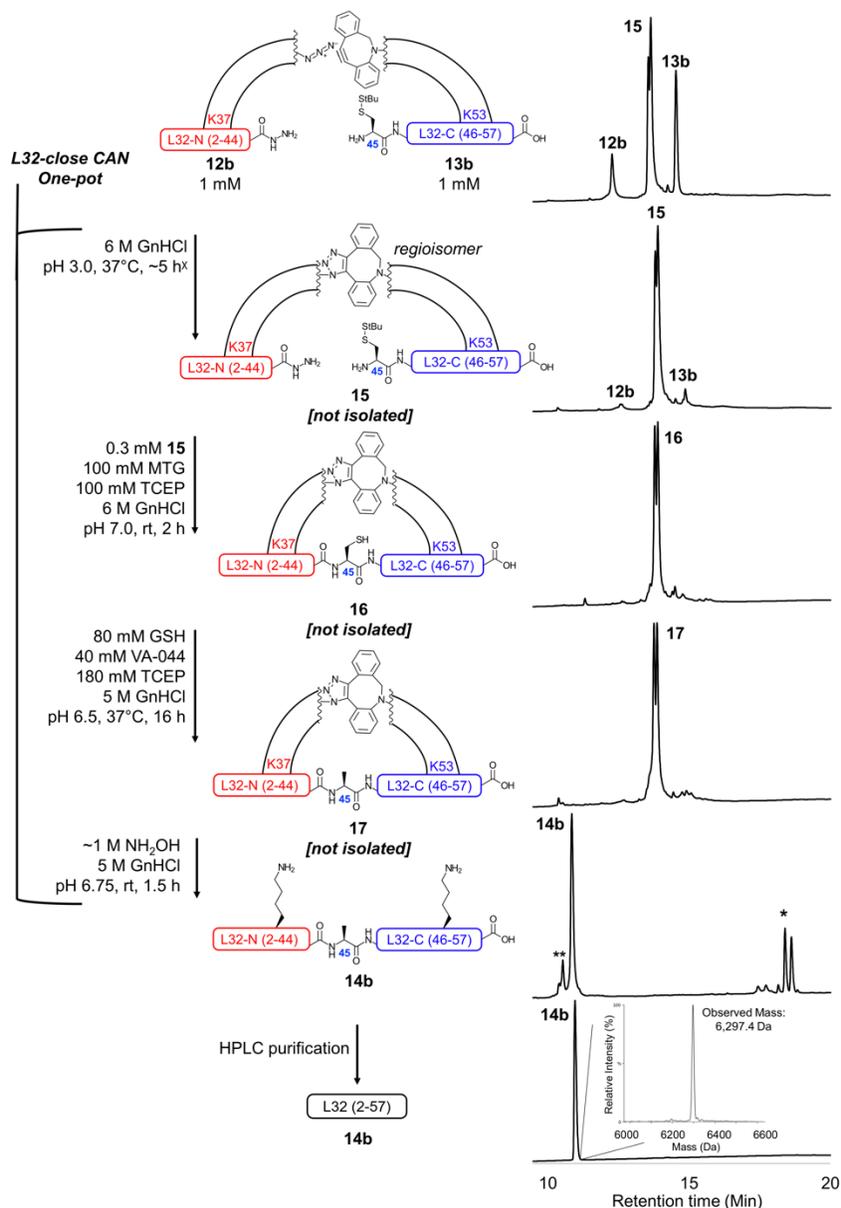
**Figure 5:** Comparison of non-templated NCL and CAN reaction rates. NCL or CAN was performed using the STEVE and KENT model peptides (**5a**, **6a**, **8**, and **10**). The relative integrated peak area of ligated products are shown at various time points relative to TCEP addition (1, 20, 50, 110, 230, 350, 470, 1430, and 2870 min). CAN ligations were fit to first-order rate constants. Note that the calculated rate constant for the 0.05 mM **8** CAN is slightly lower than expected due to the *t*Bu protecting group being removed more slowly at the 10-fold lower TCEP concentration used. Importantly, the **8** CAN is complete after 4 h at both peptide concentrations. For the non-templated NCL, peak areas of both product (**7**) and STEVE reactant (**5a**) were calculated using estimated extinction coefficients, as described in the supplemental methods. Each data point represents the average of two independent experiments, with the error bars corresponding to standard deviation.



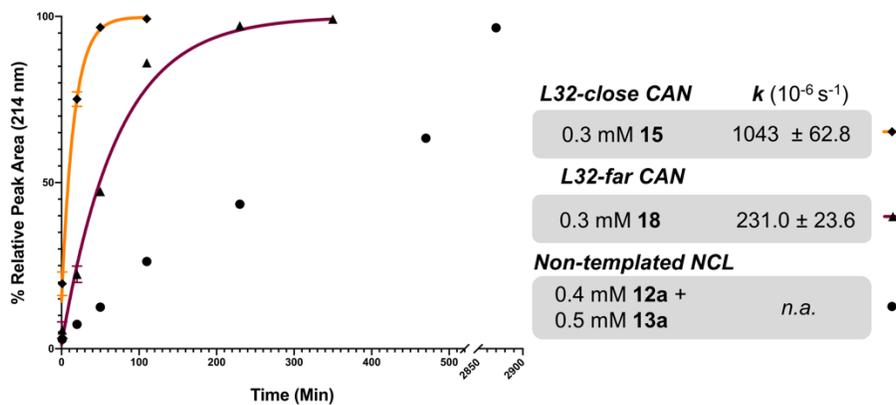
**Figure 6:** Non-templated native chemical ligation of *E. coli* 50S ribosomal subunit L32 (**14a**). (A) Sequence of L32 with L32-N (2-44, red) and L32-C (45-57, blue). The cleaved initiator Met was excluded. (B) Simplified schematic of the non-templated NCL between L32-N (**12a**) and L32-C (**13a**). Note that MTG was allowed to form the thioester for 10 min prior to TCEP addition, as shown in the more detailed scheme (**Scheme S3**). (C) HPLC chromatogram of the ligation endpoint. The ligation was considered complete after 48 h (after MTG addition), as nearly all of the reactive L32-N peptide (**12a<sup>^</sup>**) was depleted. However, a significant amount of L32-N peptide underwent hydrolysis (**12a<sup>#</sup>**) instead of ligation. **13a&** refers to *St*Bu-deprotected **13a**. Analytical Method B was used. HPLC and LC/MS chromatograms of more time points are shown in **Figures S89-90**.



**Scheme 2:** One-pot SPAAC, CAN, desulfurization, and linker cleavage assembly strategy for L32-far (dotted line) and L32-close (solid line), showing location of Ddap attachment points at specified Lys residues.



**Figure 7:** One-pot SPAAC, CAN, desulfurization, and linker cleavage of **12b** and **13b** to generate full-length L32-Nle (**14b**). **12b** and **13b** were first templated together via ~5 h SPAAC to produce triazole-linked **15**. The <sup>x</sup> in the reaction scheme indicates that additional **12b** was added to the reaction after 4 h, as a significant amount of unreacted **13b** was observed (see more details in **Scheme S4**). CAN was then performed on **15** to efficiently generate **16** after only 2 h. Desulfurization was then performed for 16 h to produce **17**. A 1.5 h linker cleavage of **17** resulted in **14b**, which was easily HPLC purified. The HPLC chromatograms show, in descending order, 0 h and ~5 h SPAAC, 2 h CAN (after MTG addition), 16 h desulfurization, 1.5 h linker cleavage, and purified **14b**. The double peaks within the SPAAC, CAN, and desulfurization chromatograms are expected, as the triazole forms two regioisomers. \* indicates cleaved Ddap linker peaks. There are more Ddap linker peaks in this chromatogram compared to the STEVEKENT CAN desulfurization chromatogram (**Figure 4**) due to Ddap hydrolysis occurring during desulfurization (**Figure S100**). \*\* indicates hydrolyzed L32-N and desulfurized L32-C. The mass spectrum of purified **14b** indicates that the correct full-length L32-Nle was generated (observed mass: 6,297.4 Da, calculated average mass: 6,297.1 Da). Analytical Method C and LC/MS Method C were used. HPLC and LC/MS chromatograms of additional time points taken for the SPAAC, CAN, desulfurization, linker cleavage, and purification are shown in **Figures S96-102**.



**Figure 8:** Comparison of non-templated NCL and CAN reaction rates using L32 peptides (**12a**, **13a**, **15**, and **18**). Relative peak area of the ligated product is shown at various time points (relative to TCEP addition). CAN ligations were fit to first-order rate constants. For the non-templated NCL, peak areas of both product (**14a**) and L32-N reactant (**12a<sup>^</sup>**) were calculated using estimated extinction coefficients, as described in the supplemental methods. Each data point represents the average of two independent experiments, with the error bars showing standard deviation.

## **ASSOCIATED CONTENT**

Materials, methods, supplemental tables, supplemental schemes, and supplemental figures are available in the supplementary information document.

## AUTHOR INFORMATION

### Corresponding Author

\*kay@biochem.utah.edu

### Present Address

§Biological Sciences Division, Pacific Northwest National Laboratory, Richland, WA, USA  
99354

### Author Contributions

‡These authors contributed equally. The manuscript was written through contributions of all authors. All authors have given approval to the final version of the manuscript.

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### Notes

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## ABBREVIATIONS

NCL, native chemical ligation

Ddap, N-Fmoc-2-(7-amino-1-hydroxyheptylidene)-5,5-dimethylcyclohexane-1,3-dione

EDTA, Ethylenediaminetetraacetic acid

Fmoc, fluorenylmethoxycarbonyl

GnHCl, guanidine hydrochloride

SPPS, solid-phase peptide synthesis

CAN, click-assisted native chemical ligation

DBCO, Dibenzocyclooctyne

ICP-MS, inductively coupled plasma mass spectrometry

RP-HPLC, reverse-phase high-performance liquid chromatography

LC/MS, liquid chromatography with mass spectrometry

TFA, trifluoroacetic acid

TIS, triisopropylsilane

AA, amino acid

MeCN, acetonitrile

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