Masked Bisketenes as Diels-Alder Dienes

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ABSTRACT: 2,5-Bis(*tert*-butyldimethylsilyloxy)furans are established as masked vicinal bisketenes for application as dienes in the Diels–Alder reaction. Cycloaddition with olefinic dienophiles, under exceptionally mild conditions, enables convergent access to highly substituted *para*-hydroquinones in unprotected form via a one-pot Diels–Alder/ring-opening/tautomerization sequence. The synthesis of *para*-benzoquinones from acetylenic dienophiles, including benzynes, is also demonstrated, and 2,5-bis(*tert*-butyldimethylsilyloxy)pyrroles are established as competent dienes for the synthesis of *para*-iminoquinones. Application in natural product synthesis enables gram-scale access to the neuroprotective agent (\pm)-indanostatin.

The *para*-hydroquinone (*p*-HQ) motif is embedded within a diverse array of pharmaceuticals,¹ natural products,^{2,3} synthetic reagents,⁴ and ligand families⁵ (**Figure 1a**). Traditional approaches towards the synthesis of highly substituted derivatives typically involve multi-step functionalization sequences, initiated from either an aromatic in a lower oxidation state, or a simple protected *p*-HQ (**Figure 1b**).⁶ Selective redox state adjustment^{7,8} or protecting group removal⁹ can be challenging to achieve in these contexts, highlighting the need for more convergent, non-oxidative strategies that deliver *p*-HQs in unprotected form. Although a small number of such strategies have been developed, most notably the Hauser–Kraus reaction of annulated furanones,¹⁰ and Moore rearrangement of γ hydroxy-cyclobutenones,¹¹ the harsh conditions necessary for substrate preparation and/or cyclization limits generality (e.g. strong base, organolithium nucleophiles, high temperatures).

p-HQs can be retrosynthetically disconnected via a Diels-Alder (DA) transform through their bis-keto tautomer to reveal vicinal bisketenes as synthons (Figure 1c). The pronounced instability of this motif,¹² coupled with the propensity of ketenes to undergo [2+2] cycloadditions with olefins,¹³ necessitates the development of a masked equivalent to realize the proposed strategy. We hypothesized that an appropriately 2,5difunctionalized furan may enable direct access to p-HQs unconditions der redox-neutral via а DA/ringopening/tautomerization sequence.14 The feasibility of such an approach was demonstrated by Chan and coworkers in 1980 though DA reactions of simple 2,5bis(trimethylsilyloxy)furans, followed by NaF promoted aromatization.15 The authors described 2,5bis(trimethylsilyloxy)furans as "very sensitive to moisture and air", noting that the rate of autoxidation increased with additional substitution on the furan backbone.¹⁶ Consequently it appears the challenges associated with substrate stability, combined with a proclivity for users to employ strong acids to promote cycloadduct unmasking,17 have discouraged efforts to further develop this strategy.^{18,19}



Figure 1. Motivation for reaction development. PG = protect-ing group, LG = leaving group.

In an attempt to address these limitations we initiated reaction development with a systematic stability $study^{20}$ of pertinent 2,5-difunctionalized furans (**Figure 2a**; see Supporting Information for additional details). In alignment with the findings of Chan, 2,5-bis(trimethylsilyloxy)furan (1) rapidly decomposed when employing standard handling techniques. Assessment of related candidates revealed diene stability scales with silyl group size (compare 1–3), arriving at 2,5bis(*tert*-butyldimethylsilyloxy)furan (3) as a robust and conveniently accessed candidate. Introduction of alternate leaving groups were also explored. Methoxy-derivative 4, prepared in 3 steps from commercial materials, exhibited a comparable stability profile relative to 3. Finally, introduction of bromine (**5**) proved beneficial, resulting in the most stable of all substrates prepared.

The DA reactivities of furans 1–5 were benchmarked in PhMe using *N*-methylmaleimide (NMM) as dienophile



Figure 2. Development of a convenient vicinal bisketene equivalent. LG = leaving group. ${}^{a}0.025M$ (CD₃)₂CO/D₂O mixture (9:1) stirred for 1 h. b Compressed air bubbled through a 0.025M CDCl₃ solution for 10 minutes. ^cDilute C₆D₆ solution heated at 120 °C in a microwave reactor for 2 h. ${}^{d}0.10M$ TFA solution in CDCl₃ stirred for 1 minute. ^e1.0 mL of a 0.025M CDCl₃ solution stirred with untreated silica (250 mg) for 2 h. ${}^{f}0.20M$ CDCl₃/CD₃OD mixture (1:4) stirred for 1 h.

(Figure 2b). Bis(silyloxy)furans 1, 2 and 3 all reached completion within ~150 minutes, exhibiting only a small rate retardation with increased silyl group size. Methoxy-derivative 4 proved slightly more reactive, whereas bromide 5 performed the worst, with less than 50% conversion observed after 250 minutes. Overall, bis(*tert*-butyldimethylsilyloxy)furan (3) strikes the best balance between ease of synthesis, stability, and DA reactivity, and as such we elected to focus exclusively on its continued development.

Studies into the ring opening/tautomerization of DA adducts **6-endo** and **6-exo** revealed that simply stirring in MeOH promotes the desired aromatization event, avoiding the need for addition of an acid, or fluoride source (**Figure 2c**). Moreover, conducting the DA reaction in MeOH enabled the synthesis of *p*-HQ **7** in 88% isolated yield, directly from furan **3**, via a one-pot DA/ring-opening/tautomerization cascade.

Having identified a suitable masked vicinal bisketene motif we investigated the scope of p-HQ formation (**Table 1**). With respect to diene structure the reaction appears general, and a library of furans (prepared directly from the corresponding cyclic anhydrides, see Supporting Information) were readily converted into p-HQs upon reaction with NMM in MeOH at 65 °C. In the case of mono-functionalized derivatives, methyl (**8**), allyl (**9**), and phenyl (**10**) functionality was well tolerated. More highly substituted 5 and 6-membered fused bicyclic furans also reacted smoothly (**11–15**), including steroidinspired pentacyclic derivative **14**. The chemoselectivity of desilylation was highlighted through preparation of silyl enol

Table 1. Scope of *p*-HQ formation



^{*a*}DA at 23 °C. ^{*b*}Aromatization achieved by addition of TFA at 23 °C. ^{*c*}DA in PhMe at 23 °C, aromatization achieved by addition of H₂SO₄ at 23 °C. ^{*d*}80 °C. ^{*e*}60 °C. ^{*f*}111 °C. ^{*g*}50 °C. ^{*h*}DA neat at 23 °C. ^{*i*}Combined yield of isomers. ^{*j*}DA neat at 55 °C.

ether 15 in 95% yield, and notably, no evidence for olefin migration was observed for all substrates screened. With more sterically demanding silvlated intermediates, such as for indole 16 and biaryl 17, the rate of MeOH promoted aromatization drops significantly. In these cases p-HQ formation can be expedited through addition of an acid additive. Extension to a one-pot double DA/double aromatization strategy provided the BiPhePhos ligand scaffold^{5a} in 53% yield (17), creating opportunities for a highly convergent approach to a library of ligand derivatives. With less reactive dienophiles the DA reaction is best conducted in an aprotic solvent (or neat) followed by addition of MeOH. Under these conditions, sulfone (18), ketone (19), ester (20) and aldehyde (21) activating groups were all tolerated. Finally, orientational selectivity of the cycloaddition event was demonstrated through coupling of unsymmetrical mono-functionalized dienes with methyl acrylate (22 and 23, up to 6:1 in favor of the expected *para*-isomer).

Exploratory studies from Chan and coworkers toward the synthesis of para-benzoquinones (p-BQ) from 2,5bis(trimethylsilyloxy)furans were hampered by competitive p-HQ formation.¹⁵ Optimistic that the 2,5-bis(tertbutyldimethylsilyloxy)furan motif may be more well suited to p-BQ formation we probed reactivity with acetylenic dienophiles (Table 2). Indeed, cycloaddition of several representative derivatives with dimethyl acetylenedicarboxylate (DMAD), followed by addition of TFA, provided p-BQs 24-26 in good to excellent yields (without any evidence for p-HQ formation). Encouraged by these results we evaluated benzynes as coupling partners. ortho-Iodotriflates were identified as suitable precursors, enabling low temperature benzyne generation in the presence of our furans. In contrast to the DMAD-derived cycloadducts, TFA addition is unnecessary, and intermediate unmasking can be achieved via a simple acidic workup. A brief diene screen revealed that increased steric bulk on the furan backbone resulted in improved yields (27–29). With respect to benzyne substitution, electron rich substrates react exceptionally well (30 and 31), however a significant reduction in yield was observed with incorporation

Table 2. Scope of *p*-BQ formation



DMAD scope:

THF, 66 °C then TFA, 0 °C

Aryne scope:

^aDetermined by ¹H NMR. ^bCombined yield of isomers.

of trifluoromethyl functionality (32). 1-Naphthyne is also a competent dienophile (33), as too are more complex steroidderived benzynes (34). Although orientational selectivity²¹ of the cycloaddition is not especially high in the case of 35 and 36 (up to 2:1), isomer separation can be achieved via standard flash chromatography, allowing ready access to the methoxy-naphthalene-1,4-dione fragment present in both doxo- and daunorubicin.¹

The generality of our approach is highlighted through extension to 2,5-bis(tert-butyldimethylsilyloxy)pyrroles as synthetic equivalents of vicinal ketenimine-ketenes (Figure 3a). While these pyrrole derivatives are less stable than their corresponding furan congeners (see Supporting Information), they exhibit comparable reactivity with arynes.²² Thus, reaction of aryl- and alkylated pyrroles 37 and 38 with benzyne provided para-iminoquinones (p-IQs) 39 and 40, both in 61% yield. Notably, p-IQs feature within natural products,²³ and derivatives have been demonstrated as versatile intermediates for the synthesis of complex alkaloids,²⁴ and BINOL derivatives.²⁵ Moreover, they serve as convenient precursors to the biologically relevant para-aminophenol motif (e.g. paracetamol)²⁶ via sodium dithionite promoted reduction (41 and 42). Further experiments demonstrated regioselective ring opening of unsymmetrical cycloadducts can be achieved (Figure 3b). In this example, a two-step reductive coupling of pyrrole 37 with 1naphthyne (generated from ortho-iodotriflate 43) provided phenanthrene 44 in 75% yield.

We became interested in the development of a crossconjugated extension to further expand the synthetic potential of our methodology (**Figure 4a**).²⁷ In principle the simplest candidate, vinylfuran **45**, would enable a diene-transmissive double DA sequence, which if conducted with two different dienophiles leads to even more highly functionalized aromatics. The success of such a strategy is reliant upon both the inherent diene site-selectivity of **45** (cyclic versus semicyclic), as well as the relative rates of the first and second cycloadditions, both of which are difficult to infer from the



Figure 3. Bis(*tert*-butyldimethylsilyloxy)pyrroles as masked vicinal ketenimine-ketenes. PMP = para-methoxyphenyl. ^aDetermined by ¹H NMR.

diverging reactivity reported for related cross-conjugated furans.²⁸ Fortuitously, NMM reacted with complete selectivity for the (desired) cyclic diene site of 45.²⁹ Moreover, the second DA reaction is markedly slower than the first, allowing not only a change in dienophile, but bifurcation of reactivity: addition of MeOH promoted aromatization to generate vinyl-substituted *p*-HQ **46**, whereas addition of 4-phenyl-1,2,4-triazole-3,5-dione (PTAD) enabled a hetero-Diels–Alder reaction. Although subsequent deprotection and aromatization can be achieved in excellent yield via in situ addition of TFA, in this particular case purification is simplified by a stepwise approach, yielding tetracyclic *p*-HQ **47** in 80% yield over two-steps from **45**.

Finally, to demonstrate scalability of p-HQ formation, as well as the applicability of our methodology in natural product synthesis, we pursued the total synthesis of the neuroprotective agent (\pm) -indanostatin^{3b} (Figure 4b). The marked base envisioned sensitivity of our dienophile, 1.3cyclopentenedione (49), has precluded its application as a coupling partner in conceptually related anionic (formal) DA strategies towards p-HQs.³⁰ Pleasingly, methylfuran 48 (synthesized in one step from methylsuccinic anhydride) reacted smoothly with 49 in MeOH at room temperature, providing gram-scale quantities of p-HQ 50 in 93% yield, and without the need for chromatographic purification. A subsequent onepot oxidation/alkylation procedure (via the intermediate trione hydrate) yielded over one gram of (\pm) -indanostatin in only three steps total (cf. 9 steps for the previous total synthesis³⁰).

In conclusion, we have established the 2,5-bis(*tert*butyldimethylsilyloxy)furan motif as a general vicinal bisketene equivalent for application as a diene in the DA reaction. A mild, user-friendly protocol provides access to a variety of highly substituted *p*-HQs by reaction with olefinic dienophiles. In turn, employing acetylenic dienophiles provides efficient access to *p*-BQs, or *p*-IQs when 2,5-bis(*tert*butyldimethylsilyloxy)pyrroles are utilized as dienes. While



Figure 4. (a) Extension to a cross-conjugated derivative, and (b) application of *p*-HQ formation in natural product synthesis.

one can imagine employing more robust silvloxy substituents in attempts to further tweak furan stability, and indeed we anticipate in some cases this may provide improved results. generally speaking it appears the tert-butyldimethylsilyl functionality strikes an excellent balance between diene stabilization and ease of intermediate unmasking. In essence, the methodology disclosed herein represents a transform-based strategy³¹ in which two C-C bonds of several para-quinone ring systems are made simultaneously via a cycloaddition event. This approach avoids the need for lengthy sequences of redox manipulations and/or functional group interconversions, both of which are commonplace in more classical structuregoal strategies.³¹ As such, we believe the methodology developed herein has the potential to permit more step-economic syntheses of *para*-quinones, and a proof-of-principle application has been demonstrated through a gram-scale three-step total synthesis of (\pm) -indanostatin.

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Notes

The authors declare no competing financial interest.

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