A graphenic and potentiometric sensor for monitoring the growth of bacterial biofilms

Noemi Poma¹, Federico Vivaldi^{1,2*}, Andrea Bonini¹, Pietro Salvo², Bernardo Melai¹, Daria Bottai³, Arianna Tavanti³ and Fabio Di Francesco¹

- ¹ Department of Chemistry and Industrial Chemistry University of Pisa, Italy; n.pomasajama@studenti.unipi.it (N.P); andrea.bonini@phd.unipi.it (A.B); melai@dcci.unipi.it (B.M); fabio.difrancesco@unipi.it (F.D.F).
- ² Institute of Clinical Physiology- National Research Council, Italy; pietro.salvo@ifc.cnr.it
- ³ Department of Biology –University of Pisa, Italy; daria.bottai@unipi.it (D.B), arianna.tavanti@unipi.it (A.T)
- * Correspondence: federicomaria.vivaldi@phd.unipi.it

Abstract: Biofilm development on surfaces represents one of the most challenging issues in the medical field. The irreversibility of this process, and the characteristics of the biofilm, make biofilm associated infections difficult to treat and prone to relapse. Thanks to the short response time, low-cost, easy fabrication and potential for miniaturization, electrochemical sensors are expected to play an important role in monitoring biofilm formation. Here, we describe the use of a potentiometric sensor based on functionalized reduced graphene oxide for monitoring the formation and growth of biofilms from three bacterial species (*Escherichia coli, Pseudomonas aeruginosa* and *Staphylococcus aureus*) in liquid and semisolid culture media. The Open Circuit Potential (OCP) between an Ag/AgCl pseudo reference electrode and a working electrode, both screen-printed on a flexible polyethylene terephthalate film, was measured during the growth of bacteria. The OCP decreased over time as bacteria grew on the sensor surface exhibiting a negative correlation.

Keywords: Biofilm growth monitoring; *Escherichia coli*; *Pseudomonas aeruginosa*; *Staphylococcus aureus*; functionalized reduced graphene oxide; potentiometric sensor.

1. Introduction

Microbial biofilm formation on medical devices and tissues represents an important and still challenging issue [1,2]. Just in the medical field, according to the National Institute of Health (NIH), biofilms are responsible for around 80% of human microbial infections, considering both device and non-device related diseases [3,4]. Among them, central venous catheter associated infections [5,6], prosthetic joint infections [7,8], osteomyelitis [9], chronic wounds [10–12], lung infection in cystic fibrosis [13] are some common examples. Moreover, biofilm associated infections (BIs) are difficult to treat and prone to relapse, as microorganisms within a biofilm present a highly reduced susceptibility to the host immune system and to antimicrobial compounds (about 1000 times higher if compared to planktonic bacteria) [14,15]. Biofilm formation is typically an irreversible process: once the biofilm is formed it cannot be easily eradicated, and in some cases the removal of the entire medical device or of the infected tissue is required [16]. Biofilm development is indeed a major cause of medical device malfunctioning, wound treatment failure [17], localized and systemic infections [18] accounting for a significant increase in patients' morbidity and mortality, with an important economic burden to the health system [15]. Just in the United States, the annual cost associated to blood stream catheter infections is estimated to range between \$300 million and \$2.3 billion [19].

Biofilms in BIs are associated to a plethora of different microbial species, but *Staphylococcus epidermidis*, *S. aureus*, *Pseudomonas aeruginosa* and *Escherichia coli* are the bacterial species most frequently involved [20], whose occurrence depends on the microorganism itself, the substrate characteristics and the availability of nutrients [4,6]. Clinical diagnosis of BIs is complex, as it is difficult to discriminate from acute infections associated to planktonic bacteria [21]. Currently, culture-based, molecular approaches and microscopy techniques are commonly used [22]. However, they are time-consuming, require specialized personnel and, more importantly, are not able to monitor biofilm development *in vivo* [21]. As a consequence, a sample biopsy or the device excision are often necessary [21]. A number of techniques such as, Peptide Nucleic Acid-Fluorescence *in situ* hybridization (PNA-FISH) [23,24], Desorption-Electro-Spray-Ionization Mass Spectrometry (DESI-

MS) [25], Matrix Assisted Laser Desorption/Ionization Time of Flight Mass Spectrometry (MALDI-TOF-MS) [26,27] and others have also been adapted to study biofilm formation, but they are mainly restricted to its characterization, and do not fulfill the urgent need for a real time monitoring and *in situ* biofilm detection [28]. In this context, the use of sensing devices as monitoring systems represents a worthy alternative, since they may continuously provide information to the health-care personnel. Such devices would indeed contribute to improve patients' quality of life, as an early warning may reduce the adverse effects linked to the progression of the infection and consequently reduce both the hospital stay time and treatment costs.

So far, biofilm sensing systems mainly based on optical and electrochemical transduction have been proposed. Optical devices encompassing optic fiber [29,30], surface plasmon [31], fluorescence methods [32] and others are based on the measurement of the increase in the layer thickness, the metabolic activity, the biofilm chemical composition and the presence of single chemical species [33]. Devices to monitor biofilm growth in situ have been designed, but their application has been mainly focused in the environmental and industrial area. Strathmann et al. described the development of a non-destructive optical sensor, with a limit of detection of 10⁶ bacterial cells/cm², aimed to be integrated in water pipes. This device used fluorescence and light scattering to distinguish abiotic deposits and biofilm, whose presence was estimated from the autofluorescence of two indicators, namely tryptophan (biomass content) and NADH (biological activity) [34]. The development of a colorimetric smart dressing prototype to assess bacterial colonization in wounds has been described. The device consisted of an hydrated agarose film containing 5,6-carboxyfluorescein dye vesicles released in the presence of virulence factors produced when P. aeruginosa, S. aureus and Enterococcus faecalis grew as a biofilm in an ex-vivo model [35]. The fluorescence response of this system was obtained using a microplate reader, making this approach not suitable for a continuous monitoring.

Electrochemical sensors are promising for their short response time, low-cost, simple fabrication, high sensitivity and possibility of miniaturization [36]. Most electrochemical sensors described in literature measure changes of impedance, which are attributed to biofilm growth and lead to variations in the capacitance, in the solution resistance or in both. Muñoz-Berbel et al. detected the biofilm growth of *P. aeruginosa* on a gold-chip working electrode, where the Constant Phase Element (CPE) decreased as the number of bacteria on the electrode surface increased [37]. In another paper, the monitoring of biofilm growth using gold-interdigitated electrodes in 96-well plates and Petri dishes was associated to a change in the solution resistance reflecting the whole metabolic activity of the biofilm [38]. Huiszoon et al. described a flexible impedance-based sensor capable of monitoring the growth of E. coli in the lumen of a urinary catheter [39]. Voltammetry has also been used to detect the bacterial colonization of a surface, expressed as a change in the electroactive area. Kang et al. described the use of the Cyclic Voltammetry (CV) and the redox couple ferrocyanide-ferricyanide to calculate the surface coverage and individuate adhered P. aeruginosa PAO1 [40]. The detection of electroactive species such as pyocyanin, a phenazine naturally produced by P. aeruginosa, has also been considered. Bellin et al. described how to use Square Wave Voltammetry (SWV) to quantify the phenazines produced by P. aeruginosa with a detection limit of 2.5 µM [41]. However, this approach is restricted to the detection of this bacterium.

Potentiometric sensors for biofilm detection and characterization have also been developed; they present many advantages if compared to other electrochemical techniques, as they do not need an external source of power and are easy to use. Here, the open circuit potential (OCP) between a working and a reference electrode is measured to assess the analyte concentration/activity. The OCP transduction has long been used in the study of electrochemically active biofilms in marine ecosystems, where a change in the OCP may be associated to the biofilm biomass [42,43]. Potentiometric needle microsensors are also used to study biofilms without perturbing the system. Guimerà et al. developed an ion-selective microsensor for monitoring pH gradients within a biofilm [44], but fragility and need of constant calibration limit the use of needle microsensors to carefully performed research studies.

The present work describes the development of a potentiometric sensor capable of monitoring bacterial colonization and biofilm formation in liquid and semisolid culture media of three bacterial species commonly associated to the development of Bis, namely *Escherichia coli*, *Pseudomonas*

aeruginosa and Staphylococcus aureus. This device exploited the chemical and electrical properties a graphene derivative as a sensing layer. The OCP changes in the presence of bacteria were continuously and efficiently recorded to allow a rapid sensing of the growing biofilm biomass. Flexibility of the sensing board makes it suitable to challenging applications related to human health, e.g. monitoring of biofilm development in medical devices (i.e. lumen of catheters) and in chronic wounds.

2. Materials and Methods

2.1. Sensing board

The sensor was fabricated using the screen-printed electrodes described by Salvo et al. [45]. The board consisted of a flexible support of polyethylene terephthalate (PET, 125 μ m thick) with silver signal tracks. This contained four graphite working electrodes (WEs, layer thickness 16 μ m, diameter 2 mm) and an Ag/AgCl pseudo reference electrode (RE, layer thickness 20 μ m, diameter 1 mm). The board was totally passivated with a polyester resin (30 μ m thick), except for the WEs and RE areas.

2.2. Graphene oxide reduction and functionalization

Graphene oxide (GO) reduction and functionalization were performed as previously described [46,47]. Briefly, a 4 mL GO solution 4 mg/mL (C: 49-56%; O: 42-50%, H: 0-1%; N: 0-1%; S: 2-4%) (Graphenea), was reduced with 0.3108 g (0.45M) of L-ascorbic acid (LAA, A7506) (Sigma Aldrich). The solution was then kept at 80°C for 4h until a black dispersion was obtained. Once at room temperature, 0.1645 g (0.3M) of 4-aminobenzoic acid (100536) (Sigma Aldrich) and 0.1655 g (0.6M) sodium nitrite (372109) (Carlo Erba) were added and the dispersion stirred at 80°C for 12 h. The dispersion was then cooled to room temperature and filtered through a polyamide membrane filter Whatman® (WHA10414012, pore size 0.2 μ m, membrane diameter 47 mm), and then rinsed three times with deionized water. Finally, the solid was dispersed in 8 mL of deionized water and a final concentration of about 1 mg/mL was obtained.

2.3. Sensor fabrication

Each working electrode was electrochemically cleaned before the sensor fabrication. A cyclic voltammetry (5 cycles, step 0.009 V, scan rate 0.1 V/s) range -0.6 V to 0.6 V vs pseudo Ag/AgCl reference electrode in a 5 M NaCl solution with a platinum electrode as auxiliary electrode.

After the cleaning procedure, the WEs were previously functionalized by the electrodeposition of 4-aminobenzoic acid diazonium salts. Cyclic voltammetry was performed (5 cycles, step 0.009 V, scan rate 0.1 V/s) range [0.3, -1.4] V vs pseudo Ag/AgCl reference electrode in an aqueous solution of 4-aminobenzoic acid 0.01M and 0.02 M of sodium nitrite. Eventually, 1 μ L of the functionalized rGO dispersion was manually drop casted onto the WEs with a micro pipette (Gilson). The whole device was then dried at 110°C for 3h.

A Nafion® membrane was used to prevent any degradation phenomena of the Ag/AgCl pseudo reference. Thus 0.5 μ L of Nafion® 5% 117 (70160) (Sigma Aldrich) was manually casted onto the pseudo reference electrode surface. Nafion was then fixed by curing at 110°C for 30 min.

2.4. Bacterial strains and culture conditions

A panel of Gram-negative and Gram-positive bacterial species represented by *Escherichia coli* (ATCC 25922), *Pseudomonas aeruginosa* (ATCC 15442) and *Staphylococcus aureus* (ATCC 12600) were used for the evaluation of the sensor's ability to monitor bacterial growth in both liquid and semisolid Luria Bertani (LB) culture media (Sigma Aldrich). LB was used as this should limit the influence of confounding factors linked to different selective media.

Strains were stored at -20 °C and -80 °C in LB supplemented with 40% glycerol (v/v). Bacteria were grown on LB agar and incubated overnight (ON) at 37 °C when necessary. Bacterial pre-

inoculum was prepared in 5 mL of fresh LB broth, from a single colony and incubated at 37 °C under agitation at 220 rpm ON.

Bacteria were grown in LB broth at 37°C for liquid cultures, and on LB semisolid agar (0.75% agar) at 37 °C. For the enumeration of Colony Forming Units (CFU), the bacterial suspensions were serially diluted in phosphate buffered saline (PBS) 1x and seeded on LB agar plates.

2.5. Biofilm formation assessment

Bacteria were evaluated for their ability to form biofilm in vitro. Briefly, for each species, a liquid overnight bacterial culture was 1:100 diluted in fresh LB broth, the bacterial suspension was then inoculated in polystyrene flat-bottom 96-well microtiter plates (100 μ L/well) and grown for 24 hours at 37°C. After incubation, non-adhered cells were removed by washing twice with 1× PBS. Biofilm cellular density was measured as the optical density (OD, λ =490nm) using an automated plate reader (Microplate Reader Model 550, Bio-Rad, Milan, Italy). The background optical density was subtracted from values measured in each well, four replicates were performed.

Biofilm formation on the sensor substrate was also verified. The board was placed in an LB broth solution, inoculated with a bacterial suspension, and incubated at 37°C under constant agitation (220 rpm). At different time points during bacterial growth, the board was removed, washed twice in PBS 1x and its surface gently scrapped to detach all adhered cells. Bacterial suspensions were then diluted in PBS 1x, grown on LB agar and incubated at 37°C for 24 h for the quantification of the CFUs. One experiment was performed for each strain.

2.6. Assessment of sensor ability to monitor bacterial growth in liquid cultures

A bacterial pre-inoculum, prepared as already described, was diluted in 100 mL of fresh LB broth to match a standardized optical density OD (λ = 600nm) corresponding to 10⁶ CFU/mL. Then, a sterilized sensor board was positioned vertically and incubated at 37 °C under agitation at 220 rpm in an orbital shaker. Microorganisms were grown for 6 hours at 37 °C, while the OCP between the WE and the RE was continuously monitored (every 15 minutes) using a potentiostat (Palmsens4, Palmsens). The bacterial culture OD (λ = 600nm) was measured hourly. Sterilized liquid LB culture media was also added as negative control. Three independent experiments, with four replicates each, were performed.

The bacterial generation time, i.e. the time required for a bacterial population to double in number, was determined from the growth curves and approximated from the OD doubling time observed during the exponential phase. To determine if the registered OCP value was due to the growth of bacterial cells on the sensor surface or the bulk culture, a representative *E. coli* ATCC 25922 culture aliquot was filtrated using a 0.22 μ m membrane filter and the OCP measured using a sterilized sensor. Three independent experiments were performed, each with four replicates.

2.7. Assessment of sensor ability to monitor bacterial growth on semisolid culture media

Previous experiments were replicated using an LB semisolid culture media. A standardized bacterial culture corresponding to $\sim 2x10^7$ CFU/mL was prepared and plated on a LB semisolid agar plate (0.75% agar). A sensor board was then placed in direct contact with the agar surface and incubated at 37 °C for 6 hours. A control plate containing the sensor where no bacteria was inoculated was also prepared. The OCP was continuously registered (every 15 minutes), while the bacterial concentration on the agar surface was determined hourly. This procedure included the cutting of a 1cm^2 agar fragment, the transfer into a 50 mL tube containing 1 mL PBS 1x and 2.0 mm glass beads, and vortexing to break the agar and release the cells in suspension. Then the sample was serially diluted using PBS 1x, plated on LB agar and incubated ON at 37°C. Results were expressed as CFU/cm². Three independent experiments were performed, each with four replicates. A negative control was also included.

Bacterial generation time on semisolid LB was determined from the growth curves and approximated from the CFU doubling time.

2.8. Data analysis

Data were expressed as means \pm SD. Correlation between the bacterial growth in liquid (OD λ =600nm) and semisolid (CFU/cm²) culture media with respect to the sensor output (mV) was analyzed with the non-parametric Spearman correlation test using the GraphPad Instat software (version 6.05 for Windows, La Jolla, CA, United States). The significance level was set to *p*<0.05.

3. Results

3.1. Graphene oxide reduction and functionalization

Characterization of graphene oxide reduced with L-ascorbic acid and functionalized with 4amminobenzoic acid (rGO-AB) has been described by Bonini et al. [46].

3.2. Biofilm formation

The biofilm biomass production was first evaluated by the 96 well plate standard assay. This confirmed that all strains included in the study were able to form biofilm in vitro in LB culture media (Figure 1A). Moreover, all strains colonized the sensor surface, as shown in Figure 1B for the *S. aureus* ATCC 12600 that was chosen as the representative strain (Figure 1B).



Figure 1. (a) Biofilm formation of the three reference strains, assessed by the 96 well plate assay and (b) biofilm formation on the sensor surface by *S. aureus* ATCC 12600

3.3. Monitoring of bacterial growth on in liquid cultures media

The OCP changed over time, shifting towards a negative potential value as bacteria grew. For all the bacteria here tested, a significant negative correlation was observed (Figure 2). Using this system, it was possible to distinguish the different phases of bacterial growth: the lag phase, characterized by a reduced growth where the potential remained quite stable, followed by a significant reduction of the OCP value corresponding to the exponential growth phase where bacteria actively proliferate, finally reaching a plateau equivalent to the stationary phase, where bacteria stop replicating as the nutrients of the media had been consumed. This last phase was more evident for *E. coli*, which entered the stationary phase earlier and grew faster than the other two bacterial species. The estimated generation times for *E. coli*, *P. aeruginosa*, and *S. aureus* were equal to 33, 42 and 51 minutes, respectively. As expected, the registered OCP in the negative control did not change over time (data not shown).



Figure 2. (a) OCP and OD values during bacterial growth in LB liquid cultures for *E. coli* ATCC 25922; (c) *P. aeruginosa* ATCC 15442; (e) *S. aureus* ATCC 12600. Sensor response (OCP) versus bacterial growth (OD): (b) *E. coli* ATCC 25922; (d) *P. aeruginosa* 15442 and (f) *S. aureus* ATCC 12600.

A further experiment was carried out to test if the registered OCP value mainly accounted for the bacterial growth on the electrode surface. The OCP measured in the filtered bacterial culture of *E. coli* ATCC 25922 remained stable over time: it did not follow the growth curve and did not show any significant correlation with the bacterial culture OD (r=-0.04, P= 0.96) (Figure 3). Therefore, this system was able to monitor a localized change directly related to the biofilm growth on the electrode surface.



Figure 3. *E. coli* ATCC 25922 growth monitoring in LB liquid culture media, measured as the OCP (filtered and non-filtered solution) and the optical density (non-filtered solution).

3.4. Monitoring of bacterial growth on semi-solid culture media

As in previous experiments, the sensor response on the semi-solid agar surface was shown to change concomitantly to the bacterial growth expressed in CFU/cm²; indeed, as the bacterial concentration increased, the OCP value decreased exhibiting a significant negative correlation in all cases (Figure 4). The lag and logarithmic phases were observed in the OCP curve for *E. coli* and *S. aureus*, but not for *P. aeruginosa*, whose sensor response linearly decreased from the first measurement. During the monitoring time (6 hours), all bacteria entered the logarithmic phase, except *E. coli*, which grew faster and reached the early stationary phase in accordance to the calculated generation times: 36, 46, and 55 minutes for *E. coli*, *P. aeruginosa* and *S. aureus*, respectively. In general, these values were higher than those registered from bacteria growing in the LB liquid media.

The OCP value in the negative control did not change over time (data not shown), which demonstrated its association to the processes occurring during bacterial growth.



Figure 4. (a) OCP values measured during bacterial growth on LB semisolid culture media: (a) *E. coli* ATCC 25922; (c) *P. aeruginosa* 15442; (e) *S. aureus* ATCC 12600. Sensor response (OCP) versus bacterial growth (OD): (b) *E. coli* ATCC 25922; (d) *P. aeruginosa* 15442 and (f) *S. aureus* ATCC 12600.

4. Discussion

Biofilms are dynamic structures, constantly changing over time [48,49]. Their important role in the onset of infections or sepsis urged the scientific community to develop new technologies for an early detection. Sensors susceptible to changes at the electrode surface and surrounding environment are amongst the most interesting tools described in literature for this aim due to the promise of a continuous monitoring. The sensor device described in this paper used rGO-AB as the sensing material and OCP measurement as transduction method; its performances were assessed with bacteria growing both in liquid and semisolid LB culture media.

rGO-AB was selected as the sensing material because rGO is highly conductive and known to promote the exchange of electrons with bacteria [50], which is convenient in combination with OCP measurements to detect bacterial growth. The presence of a limited number of residual defects upon the chemical reduction of graphene oxide does not impair the electrical conductivity. At the same time, defects provide reactive moieties such as the hydroxyl (OH) and carboxyl (COOH) groups that foster bacterial adhesion and electron transfer [51,52,53]. Furthermore, rGO can be easily

functionalized to improve sensor performance in different matrices [52,53]; in this case, the functionalization with 4-aminobenzoic acid was chosen to increase the number of carboxylic groups on the electrode surface and improve the sensitivity of the device [46].

Due to the extremely low currents flowing between working and reference electrode, OCP measurement is an almost ideal electrochemical transduction technique for biofilm monitoring due to its negligible effects on bacteria and electrode surface [54]. OCP results from contributions of different processes taking place at the electrode/solution interface, such as the accumulation of charge and the spontaneous and simultaneously occurring reactions between the working electrode and compounds in the surrounding environment, which establish an intermediate potential between cathodic and anodic reactions [55]. This study showed a negative correlation between bacterial growth on the working electrode surface and OCP, with a significant reduction of this latter during the exponential phase. Previous studies demonstrated that the direction of the potential change depends on the microorganism, the material where the biofilm develops and the environment [42,56]. Here, the OCP shift towards more negative values as the bacterial concentration increases may be explained by the interaction of a variety of factors. An increase in the cell biomass may account for a larger number of negative charges on the working electrode surface. Both Gram-positive and Gramnegative bacteria, although different in composition, have a negative net charge on the cell wall, which contains carboxyl, amide, hydroxyl, phosphate and carbohydrate related moieties [58,59]. The presence of negatively charged biomolecules such as DNA, proteins and others present in the extracellular polymeric substance may also contribute to the accumulation of charges [58]. Indeed, a closer interaction of DNA and negatively charged biomolecules with the electrode surface was associated to a decrease in the OCP value in an aptamer based biosensor detecting the prostate specific antigen (PSA) [59,60], which was also negatively correlated to the amount of added antigen [60].

Another factor to be considered is the direct and constant release of products of bacterial metabolism. Bacterial physiology is dependent on the bacterial species and the culture medium composition; in the present study, we used a non-selective LB medium containing tryptone and yeast extract as a source of oligopeptides, aminoacids, sugars, lipids, vitamins, and carbohydrates [61]. As proposed in previous studies using *E. coli* as a model organism, catabolizable aminoacids are the principal carbon source in this culture media, and are sequentially consumed [61,62]. During their metabolism, these are deaminated by membrane associated proteins such as aminoacid dehydrogenases and transaminases, so that ammonia (NH₃) is produced; ammonia is then protonated to ammonium (NH₄⁺) with the consequent formation of hydroxide ions, which may further contribute to the charge accumulation on the WE [61,63]. However, it should also be considered that sessile cells embedded within the biofilm community vary greatly and that only a few of different bacterial subpopulations may be metabolically active [48]. Variations in the OCP have also been attributed to changes in the environment properties due to bacterial growth. For example, the depletion of oxygen on a steel working electrode surface associated to the growth of *Pseudomonas fragi* and *E. coli* in LB culture media decreased the OCP [64].

Microbial respiration and the associated redox processes in electroactive biofilms (EABs) could also modify OCP. Microorganisms within EABs exchange electrons with the electrodes thanks to bacterial surface components (e.g. outer membrane proteins) or redox mediators [65–67]. Pocaznoi et al. found that the growth of a consortium of soil microorganisms forming a biofilm on the surface of a carbon cloth electrode decreased the OCP, and this behavior was explained by the presence of an EAB and the depletion of oxygen at the electrode surface, resulting in a slow oxidation of the surface [68].

E. coli and *P. aeruginosa*, two of the bacterial strains used in this study, have been reported as electrochemically active [69]; their main electron transfer mechanism is associated to the use of electron mediators: metabolites for *E. coli* [70], and phenazine derivatives for *P. aeruginosa* [71]. Little is known about the electrochemical activity for *S. aureus*. In 2010, Cournet and colleagues studied the catalysis of the electrochemical reduction of oxygen by different bacteria including *S. aureus*, which demonstrated not to have this ability [72]. However, other aforementioned processes may contribute to the OCP variation.

Bacterial generation time is not constant among microorganisms, as it depends on the microorganism and the characteristics of the surrounding environment (e.g. temperature, availability of nutrients). Under the experimental conditions here used, it was estimated that bacteria growing in LB liquid culture media grew faster than their counterparts in semisolid culture. In general, generation times were higher for *S. aureus* and lower for *E. coli*. This was reflected in the sensor response, which stabilized once *E. coli* reached the stationary phase.

Interestingly, although the growth rate of bacteria was lower in semisolid culture media, a sudden decrease of the OCP was observed, so in some cases the lag phase was not well differentiated. It should be considered that prior to biofilm formation, planktonic cells have first to get close to the surface. In liquid media for example, this process is mediated by active (i.e. bacterial swimming) and passive (i.e. precipitation and convection-diffusion) movements [73,74]. In our study, bacteria in the semisolid medium were in direct contact with the working electrode surface, so this physical proximity likely allowed cells to circumvent this step and made it somehow easier for them to colonize the working electrode. On the contrary, the constant movement due to stirring and the vertical position of the sensor may have delayed the adhesion in liquid LB. Moreover, bacteria are just weakly attached to the surface during the early stages of biofilm formation [75], so that the initially suspended bacteria may have attached and detached from the surface several times before that a small fraction could irreversibly adhere and start replicating. This may account for the longer lag phase shown by the OCP in the liquid culture.

Our device is not selective, but biofilms are typically complex polymicrobial structures including different species whose composition may change over time, so we wonder if selectivity is really an issue or if the lack of selectivity may be considered an advantage in many applications. Considering the rapid and worrisome development of biofilms leading to BIs, this sensor could be exactly what needed to monitor the overall bacterial burden in wounds or to provide an early alert of microbial colonization in catheters, avoiding the undesired effects related to the progression of an infection.

5. Conclusions

This study described the development of a flexible, low-cost, biocompatible graphenic sensor allowing a real-time monitoring of the growth of common Gram-negative and Gram-positive bacteria involved in the development of BIs. The OCP used as the transduction technique efficiently detected biofilm growth localized on the sensor surface in liquid and semisolid culture media, and exhibited a significant negative correlation with respect to the bacterial growth. Biofilms are complex in nature, so the interaction of a variety of factors at the electrode surface may determine the OCP change over time. It was hypothesized that processes like the accumulation of a net negative charge, bacterial metabolism and respiration may account for the OCP decrease.

The assessment of bacterial growth and the OCP response in presence of inhibitors of metabolism, or the use disrupting compounds such as antimicrobial peptides in preformed biofilms, will certainly give in future further information about the processes causing the OCP variations. In addition, the use of other electrochemical techniques such as impedance spectroscopy may allow a deeper characterization of this system.

The sensor presented in this study may be potentially applied as a bacterial growth monitoring tool. Thanks to its flexibility, it could be embedded in smart dressings or catheters for a timely detection of biofilms formed by bacterial species of clinical importance.

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