# Rh(III)-catalyzed direct access to 2,3-substituted $\beta$ -*N*-glycosyl indoles through C-H activation/annulation coupling of $\beta$ -*N*-aryl glycosides with substituted internal alkynes

Guangkuan Zhao,<sup>a</sup> Mingxiang Zhu,<sup>a</sup> Olivier Provot,<sup>a</sup> Mouad Alami,<sup>a</sup> and Samir Messaoudi<sup>a\*</sup>

<sup>a</sup> BioCIS, Univ. Paris-Sud, CNRS, University Paris-Saclay, Châtenay-Malabry, France





 $\beta$ -*N*-glycosyl indoles are of high importance in medicinal chemistry and commonly found in many compounds of practical importance, ranging from natural compounds to pharmaceutical agents<sup>1</sup> (Figure 1). While these derivatives clearly hold a great potential in medicinal chemistry, relatively little attention has been devoted to their syntheses (Figure 1-B), since the stereoselective induction of a nitrogen indole scaffold at the anomeric position remains as a particularly difficult task. From a synthetic point-ofview, 2,3-substituted  $\beta$ -*N*-glycosyl indoles 3 (Figure 1) were prepared through multi-steps syntheses by treating indoline derivatives with sugars lactols<sup>ii,2</sup> followed by (*i*) oxidation of the indoline into indole and (ii) functionalization of the C-2 and C-3 positions (Figure 2, Path A). The synthesis of 2,3-substituted  $\beta$ -*N*-glycosyl indoles is also possible *via* a β-glycosylation of indoles through a SN<sub>2</sub> Mitsunobu reaction<sup>1k, 1l, 3</sup> followed by the functionalization of the indole nucleus (Figure 2, Path B). However, the reaction necessitate the use of a well-defined  $\alpha$ -sugar lactols which are difficult to synthesize through multisteps sequences.4 Moreover, a mixture of  $\alpha$ - and  $\beta$ -anomers was obtained in most cases. Another way to prepare stereoselectively  $\beta$ -Nglycosyl indoles is the use of Danishefsky  $\alpha$ -1,2-anhydro sugars (Figure 2, Path C).<sup>5</sup> While this method is efficient, beside the necessary of C2- and C3 functionalization,



Figure 1. Heteroaryl *N*-glycoside-based bioactive molecules

access to  $\alpha$ -1,2-anhydrosugars (epoxides) remain no trivial. Indeed, their synthesis from epoxidation of the corresponding glycals is inherently relies on the use of the Murray's reagent<sup>6</sup> (DMDO) which is an instable volatile peroxide not easy to prepare and manipulate. Despite these advances, the absence of a general and predictable



**Figure 2** Strategies to access to 2,3-substituted  $\beta$ -*N*-glycosyl indoles

method for the direct synthesis of 2,3-substituted  $\beta$ -*N*-glycosyl indoles with a minimum of steps remains a major gap in glycochemistry preventing greater investigation of the biology and applications of these compounds.

Recently, our group reported an efficient protocol for the synthesis of  $\beta$ -*N*-aryl glycosides *via* a copper-catalyzed Chan-Lam-Evans N-arylation of aryl boronic acids.7 As part of our continued efforts to functionalize sugars under transition-metal catalysis to access complex glycosides,<sup>8</sup> we envisioned whether  $\beta$ -*N*-aryl glycosides could be utilized as building blocks in the synthesis of  $\beta$ -*N*-glycosyl indoles through a transition-metal-catalyzed activation/annulation reaction in the presence of various alkynes (Figure 1-C). This modular strategy is conceptually attractive in terms of diversifying the N-glycosyl indoles frameworks with the aim to identify novel scaffolds of biological interest. In this work, we showed for the first time, that  $\beta$ -*N*-aryl glycosides and alkynes were successfuly joined together through a C-H activation/annulation process9 to afford in a single step stereoselectively, a variety of substituted  $\beta$ -N-glycosyl indoles (Figure 1-C).

To achieve successfully our goal, initial investigations focused on identifying optimal conditions for the coupling of  $\beta$ -*N*-phenyl glucopyranoside **1a** with 1,2-diphenylethyne **2a** as models study (Table 1). In preliminary experiments, the C–H activation/annulation reaction was examined under various conditions in the presence of different catalysts such as Ru,<sup>10</sup> Rh,<sup>11</sup> Ni,<sup>12</sup> and Pd,<sup>13</sup> but unfortunately,  $\beta$ -*N*-glycosyl indole **3a** was not detected under these conditions.

Further, inspired by the recent work described by Zhu<sup>14a</sup> and Fan<sup>14b</sup> whose reported the cyclization of *N*nitrosoanilines under Rh(III)-catalysis, we evaluated the influence of a nitroso-substituent on the phenyl glucopyranoside nitrogen atom (compound **1b**). When we used  $\beta$ -*N*-nitrosophenyl glucopyranoside **1b** with [(Cp\*RhCl<sub>2</sub>)<sub>2</sub>] (5 mol%), AgSbF<sub>6</sub> (20 mol%), and diphenylacetylene (**2a**, 2

Table 1 Survey of reaction conditions for the C-H activation/annulation of 1a,b with  $2a^a$ 

|       | AcO<br>AcO<br>R =<br>R = | H<br>N-R<br>OAc<br>H, 1a<br>NO, 1b | Ph Cat.<br>Ph $ad$<br>Ph $ad$<br>Cat. = [RuC<br>Cat2. = RuC<br>Cat3. = [Cp+1<br>Cat4. = Ni(Cl<br>Cat5. = Pd(F | (5 mol%)<br>ditives<br>solvent<br>e (h), T (°C)<br><sup>3</sup><br>RhCl <sub>2</sub><br>OD <sub>2</sub><br>Ph <sub>3</sub> ) <sub>4</sub> |              | Ph<br>Ph<br>POAc<br>3a |  |
|-------|--------------------------|------------------------------------|---|---|--------------|------------------------|--|
| entry | R                        | Cat.                               | additive  | solvent   | Temp<br>(°C) | Time<br>(h)            | yield<br><b>3a</b><br>(%) <sup>b</sup> |
| 1     | Н                        | Cat,                               | AgNO <sub>2</sub>   | t-AmOH  | 120          | 12                     | 0                                      |
| 2     | Н                        | Cat                                | Ag <sub>3</sub> PO <sub>4</sub>   | t-AmOH  | 120          | 12                     | 0                                      |
| 3     | Н                        | Catı                               | MeCO₂Ag   | t-AmOH  | 120          | 12                     | 0                                      |
| 4     | Н                        | Cat                                | CF <sub>3</sub> CO <sub>2</sub> Ag  | t-AmOH  | 120          | 12                     | 0                                      |
| 5     | Н                        | Cat <sub>1</sub>                   | AgSbF <sub>6</sub>  | t-AmOH  | 120          | 12                     | 0                                      |
| 6     | Н                        | Cat <sub>1</sub>                   | $AgSbF_6$   | dioxane   | 120          | 12                     | 0                                      |
| 7     | Н                        | Cat <sub>1</sub>                   | $AgSbF_6$   | DCE   | 120          | 12                     | 0                                      |
| 8     | Н                        | Cat <sub>1</sub>                   | $AgSbF_6$   | Acetone   | 120          | 12                     | 0                                      |
| 9     | Н                        | Cat <sub>1</sub>                   | $AgSbF_6$   | DMF   | 120          | 12                     | 0                                      |
| 10    | Н                        | Catı                               | $AgSbF_6$   | Toluene   | 120          | 12                     | 0                                      |
| 11    | Н                        | Cat <sub>2</sub>                   | $AgSbF_6$   | t-AmOH  | 120          | 12                     | 0                                      |
| 12    | Н                        | Cat <sub>3</sub>                   | $AgSbF_6$   | t-AmOH  | 120          | 12                     | 0                                      |
| 13    | Н                        | Cat <sub>4</sub>                   | $AgSbF_6$   | t-AmOH  | 120          | 12                     | 0                                      |
| 14    | Н                        | Cat <sub>5</sub>                   | $AgSbF_6$   | t-AmOH  | 120          | 12                     | 0                                      |
| 18    | NO                       | Cat <sub>3</sub>                   | $AgSbF_6$   | DCE   | 100          | 3                      | 38                                     |
| 19    | NO                       | Cat <sub>3</sub>                   | $AgSbF_6$   | DCE   | 120          | 3                      | 25                                     |
| 20    | NO                       | Cat <sub>3</sub>                   | $AgSbF_6$   | DCE   | rt           | 15                     | trace                                  |
| 21    | NO                       | Cat <sub>3</sub>                   | $AgSbF_6$   | DCE   | 60           | 15                     | 31                                     |
| 22    | NO                       | Cat <sub>3</sub>                   | AgSbF <sub>6</sub>  | DCE   | 100          | 5                      | 55 <sup>°</sup>                        |
| 23    | NO                       | Cat <sub>3</sub>                   | $AgSbF_6$   | DCE   | 90           | 5                      | 70                                     |
| 24    | NO                       | Cat <sub>3</sub>                   |   | DCE   | 90           | 5                      | 0                                      |
| 25    | NO                       |                                    | AgSbF <sub>6</sub>  | DCE   | 90           | 5                      | 0                                      |
| 26    | NO                       | Cat <sub>3</sub>                   | AgSbF <sub>6</sub>  | DCE   | 90           | 5                      | $72^d$                                 |

<sup>*a*</sup> Reactions were conducted with substrate **1a** or **1b** (0.10 mmol), alkyne (0.2 mmol), catalyst, additive, and solvent (1.0 mL). <sup>*b*</sup> Yield of isolated product **3a**. <sup>*c*</sup> **1b** was completely consumed. <sup>*d*</sup> 0.4 mmol of **2a** were used.

equiv.) in 1,2-dichloroethane (DCE) at 100 °C for 3 h, we obtained *β*-*N*-glucosyl indole **3a** ( $J_{i,2}$  = 9.3 Hz) in 38% yield (entry 18). After screening several parameters, we finally found that the C-H annulation between **1b** and **2a** occured smoothly with 70% yield in the presence of [(Cp\*RhCl2)2] (5 mol%), AgSbF<sub>6</sub> (20 mol%), in DCE at 90 °C for 5 h without adding any external oxidant (Table 1, entry 23). It should be noted that the Rh-catalyst and AgSbF<sub>6</sub> were necessary to achieve this transformation since no reaction occur when the coupling was conducted in the absence of [(Cp\*RhCl<sub>2</sub>)<sub>2</sub>] or AgSbF<sub>6</sub> (entries 24 and 25).

Motivated by these results, we next explored the scope of the coupling reaction of **1b** with a variety of internal alkynes and we are gratifyingly pleased with the generality of this method. Various internal alkynes reacted smoothly to afford the desired 2,3-disubstituted *N*-glucosyl indoles **3a-i** in satisfactory yields. A variety of symmetric diaryl alkynes were efficiently converted into the corresponding products and electron-donating substituents on the aromatic rings had a positive electronic effect on yield than electronwithdrawing substituents (Table 2, **3a-d**). The coupling reaction with 1,2-di(thiophen-2-yl)ethyne was also efficient to afford **3e** in a moderate yield. Moreover, unsymmetrically substituted alkynes **2f-i** were converted with variable yields and moderate regioselectivities (**3f-i**).

In a further set of experiments, we investigated the scope and generality of the method with respect to  $\beta$ -*N*-aryl glycosides **1j-r**. As depicted in Table 3, the protocol tolerated different  $\beta$ -*N*-aryl glucosides and a multitude of 5- and

**Scheme 1** Scope of alkynes coupling with tetraacetyl  $\beta$ -*N*-nitrosophenyl glucopyranoside **1b** 



Reaction conditions: reactions were performed in a flame dried re-sealable Schlenk tube using 1b (0.30 mmol), alkynes 2 (2 equiv), [RhCp\*Cl2]2 (5 mol%), AgSbF6 (20 mol%), in 1,2-DCE (0.1 M) at 90 °C for 5 h. b Yield of isolated product 3.

6-substituted indoles **3j-o** were readily prepared using this reaction. 5-Substituted  $\beta$ -N-aryl glycosides bearing elec tron-donating or electron-withdrawing groups afforded the corresponding 5-substituted  $\beta$ -N glucosyl indoles 3j**m** in acceptable yields. In addition, meta-substitution was tolerated furnishing 6-disubstituted  $\beta$ -N-glucosyl indoles 3n and 30 in 58% and 37% yields, respectively. Interestingly, this cross-coupling tolerated the presence of C-halogen bonds (e.g., F, Cl, Br) which offers a platform for further metal-catalyzed cross coupling reactions (compounds 3l, 3m and 3n). Moreover, the C-H activation/annulation process is not limited to  $\beta$ -*N*-aryl glucosides but also works successfully with  $\beta$ -*N*-aryl galactosides **1p**,**q** and the peracetylated β-D-disaccharide **1r** derived from D-βcellobiose octaacetate, however only 25% isolated yield of the disaccharide **3r** was obtained probably due to its intrinsic instability of **3r**. Of note, the stereochemistry of the  $1 \rightarrow 4$ ' glycosidic bond remained intact. It is noteworthy that the coupling of unprotected  $\beta$ -*N*-nitrosophenyl glucoside with 1,2-diphenylethyne 2a under the above experimental conditions failed, as only starting materials were recovered unchanged.

With substantial amounts of **3l** in hand (Table 2), we focused our attention on demonstrating whether our method could be employed for molecular diversity. As shown in Scheme 1,  $\beta$ -*N*-glucosyl indole 4, which is an analogue of compound **A**, a highly promising cytotoxic and antitubulin agent developed in our group,<sup>15</sup> was easily

Scheme 2 Scope of  $\beta$ -*N*-nitroso-aryl glycosides **1j-r** coupling with **2a** 



Reaction conditions: reaction were performed in a flame dried re-sealable Schlenk tube using **1j-r** (0.30 mmol), alkyne **2a** (2 equiv),  $[RhCp^*Cl_2]_2$  (5 mol%), AgSbF<sub>6</sub> (20 mol%), in 1,2-DCE (0.1 M) at 90 °C for 5 h. <sup>*b*</sup> Yield of isolated product **3**. <sup>*c*</sup> The reaction was achieved at 80 °C

prepared *via* a Pd-catalyzed coupling reaction of **3l** with 3,4,5-trimethoxy-*N*-methylaniline. Compound **5** was obtained from **4** by deprotection of the acetates groups under Zemplen's conditions using a catalytic amount of potassium carbonate in methanol.<sup>16</sup>

Scheme 1 Application of this methodology the synthesis of bioactive compounds 4 and 5



The *in vitro* activity of derivatives 4 and 5 was evaluated by their growth-inhibitory potency against HCT-116 cancer cells (human colon carcinoma) at the concentration of 10<sup>-6</sup> M. The quantification of cells survival in this cell line was established by using MTT assays after 72 h of exposure. We found that analogues 4 and 5 displayed a mod-

erate effect in the growth of HCT-116 (81% and 70% survival, respectively) compared to the reference compound A (IC<sub>50</sub> = 7 nM) (Scheme 1).

In conclusion, we successfully developed an efficient and practical method based on Rh(III)-catalyzed C-H activation/annulation process of various  $\beta$ -*N*-nitroso-aryl glycosides with alkynes. The protocol exhibited a broad substrate scope with respect to the coupling partners, thus providing an attractive access to a large molecular diversity of 2,3-disubstituted *N*-glycosyl indoles 3. This protocol developed is stereoretentive, functional-group tolerant, and proceeds in good yields. We believe that this methodology will find broad applications in organic synthetic chemistry as well as in combinatorial and pharmaceutical sciences.

# ASSOCIATED CONTENT

## **Supporting Information**

The Supporting Information is available free of charge on the ACS Publications website.

Experimental procedures, spectroscopic data and NMR spectra of new compounds.

## **AUTHOR INFORMATION**

#### **Corresponding Author**

\*E-mail: samir.messaoudi@u-psud.fr

#### Author Contributions

The manuscript was written through contributions of all authors. / All authors have given approval to the final version of the manuscript.

## Notes

The authors declare no competing financial interest.

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