

Preparation and crystal structure of 1,2-bis(2,5-dimethylthiophen-3-yl)ethane-1,2-dione

Paweł Pakulski, Dawid Pinkowicz

Jagiellonian University, Faculty of Chemistry, Gronostajowa 2, 30-387 Kraków, Poland dawid.pinkowicz@uj.edu.pl





Preparation and crystal structure of 1,2-bis(2,5-dimethylthiophen-3-yl)ethane-1,2-dione

Paweł Pakulski, Dawid Pinkowicz

Jagiellonian University, Faculty of Chemistry, Gronostajowa 2, 30-387 Kraków, Poland dawid.pinkowicz@uj.edu.pl

Introduction

1,2-bis(2,5-dimethylthiophen-3-yl)ethane-1,2-dione is a convenient starting material for the synthesis of various photoactive organic molecules that belong to the group of diarylethenes (dae).[1] Diarylethenes exhibit a reversible photocyclisation-induced photochromism, photoswitching behavior and memory effects.[1,2] Greater stabilization of the closed-form photocyclisation product, leading in general to improved photochromic properties, can be achieved by the modification of the diarylethene backbone – bis(2,5-dimethylthophen-3-yl) perfuorocyclopentene is perhaps the most recognizable example of such an approach[3].

The title compound can be obtained in a Friedel-Crafts acylation of 1,2-dimethylthiophene with oxalyl chloride in the presence of metal chlorides such as $AlCl_3$ or $TiCl_4.[4]$ Herein, we report the use of $ZrCl_4$ as the catalyst, leading to slightly higher yield. We also report for the first time the crystal structure of 1,2-bis(2,5-dimethylthiophen-3-yl)ethane-1,2-dione. Crystals suitable for the single-crystal X-ray diffraction (sc-XRD) experiments were obtained by a careful evaporation of a diethyl ether solution of the title compound.

Results and discussion

1,2-bis(2,5-dimethylthiophen-3-yl)ethane-1,2-dione

crystalizes in a monoclinic C2/c space group. The details of the sc-XRD experiment can be found in the cif file deposited with the CSD database (deposition number CCDC 1895209, DOI: 10.5517/ccdc.csd.cc21m3sl). The asymmetric unit consisting of one and a half of 1,2-bis(2,5-dimethylthiophen-3-yl)ethane-1,2-dione molecules is presented in Figure 1. Both molecules exhibit a *trans* geometry with the O=C-C=O torsion angle of 121° and 124°, respectively. Selected bond lengths are summarized in Table 1 and correspond well with those of thiophene moiety or carbonyl groups.



Figure 1. ORTEP representation of the asymmetric unit of 1,2bis(2,5-dimethylthiophen-3-yl)ethane-1,2-dione with atom labeling scheme

Table 1. Selected bond	lengths in	1,2-bis(2,5-dimethylthiophen-3-
yl)ethane-1,2-dione		

Bond type	Length / Å	
C1–C1	1.540	
C8–C15	1.542	
C1=O1	1.224	
C8=O2	1.223	
C15=O3	1.227	
S1-C3	1.720	
S1-C6	1.733	
S2-C10	1.721	
S2C13	1.735	
S3-C17	1.715	
S3–C20	1.734	



Figure 2. π - π (top) and CH/ π (bottom) contacts in the crystal structure of 1,2-bis(2,5-dimethylthiophen-3-yl)ethane-1,2-dione



chemical shift / ppm

Figure 3. ¹H NMR spectrum of 1,2-bis(2,5-dimethylthiophen-3-yl)ethane-1,2-dione in CDCl₃ at room temperature

Crystal packing of the title compound is defined by weak π - π and CH/ π intermolecular contacts. Each molecule forms two π - π contacts with two adjacent molecules. These contacts involve the thiophene rings and the average C-C distance is ca. 3.8 Å. Each two molecules form also two CH/ π interactions between the thiophene ring carbon atoms and the methyl group carbon atom of the other molecule with the C-C distance of ca. 3.9 Å (Figure 2). Both interactions must be quite weak if we consider the shortest CH/ π distances to be around 2.8-3.1 Å and the shortes π / π contacts around 2.6-2.8 Å.[5]

Experimental

2,5-dimethylthiophene (0.85 ml, 7.50 mmol, Aldrich) was dissolved in 15 ml of dichloromethane (DCM) and chilled to -20°C. A solution of oxalyl chloride (0.225 ml, 2.85 mmol, Aldrich) in 15 ml of DCM was added and the reaction flask was flushed with argon. Zirconium(IV) chloride (1.00 g, 4.27 mmol) was then added in one portion and the flask was flushed with argon again. The reaction was carried out at -20 to 0°C for 75 minutes. It was poured on 20 ml of ice and the organic phase was separated. The aqueous phase was extracted with three 50 ml portions of chloroform. The combined organic extracts were washed with 20 ml of water, saturated sodium carbonate solution and water again, dried over anhydrous magnesium sulfate and evaporated to a red oil. The crude product was purified by column chromatography (SiO₂, petroleum ether 60:90/ethyl acetate/chloroform 20:2:1) to afford the title compound as a bright yellow oil (358 mg) in nearly 50% yield. The purity and identity of the compound was confirmed by ¹H NMR in

CDCl₃: δ [ppm] 6.92 (m, 2H) 2.73 (s, 6H) 2.39 (m, 6H) (Figure 3).

If the oil is dissolved in ca. 10 ml of diethyl ether and slowly evaporated, yellow prism crystals suitable for singlecrystal X-ray diffraction are obtained.

Acknowledments

This research was funded by the Polish National Science Centre within the Sonata Bis project no. 2016/22/E/ST5/00055.

Authors information

Paweł Pakulski https://orcid.org/0000-0001-6996-085X Dawid Pinkowicz https://orcid.org/0000-0002-9958-3116

References

- 1 Irie, M., Chem Rev. 2000, 100, 1685-1716
- 2 Kobatake, S., Kuma, S., Irie, M., J. Phys. Org. Chem., 2007, 20, 960-967
- 3 Patel, D. G. D, Walton, I. M., Cox, J. M., Gleason, C. J., Butzer, D. R., Benedict, J. B., *Chem. Commun.*, **2014**, *50*, 2653-2656
- 4 Belen'kii, L. I., Shirinyan, V. Z., Gromova, G. P., Kolotaev, A. V., Strelenko, Y. A., Tandura, S. N., Shumskii, A. N., Krayushkin, M. M., *Chemistry of Heterocyclic Compounds*, 2003, 39, 1570–1579.
- 5 Nishio, M., CrystEngComm, 2004, 6, 130-158.